

CuNi-PTC Metal-Organic Framework: Unveiling Pseudocapacitive Energy Storage and Water Splitting Capabilities

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Supporting Information

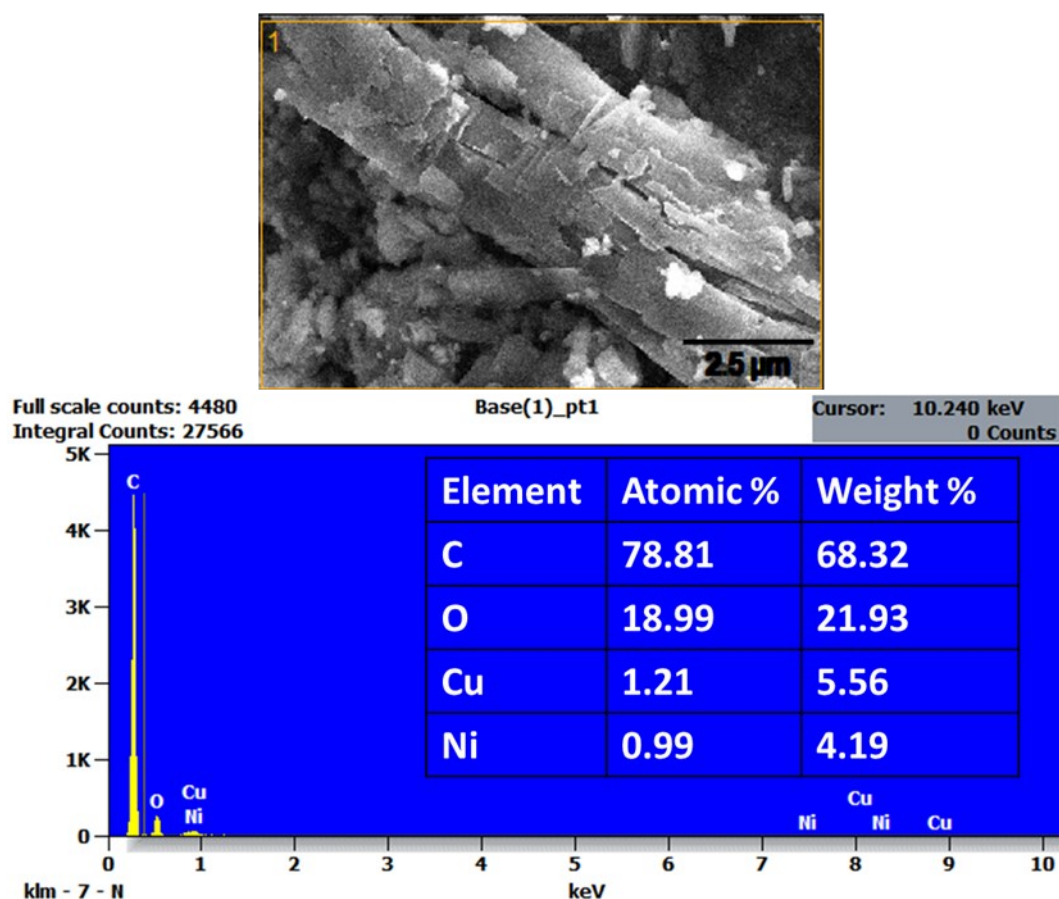


Figure S1. EDS spectrum of CuNi-PTC

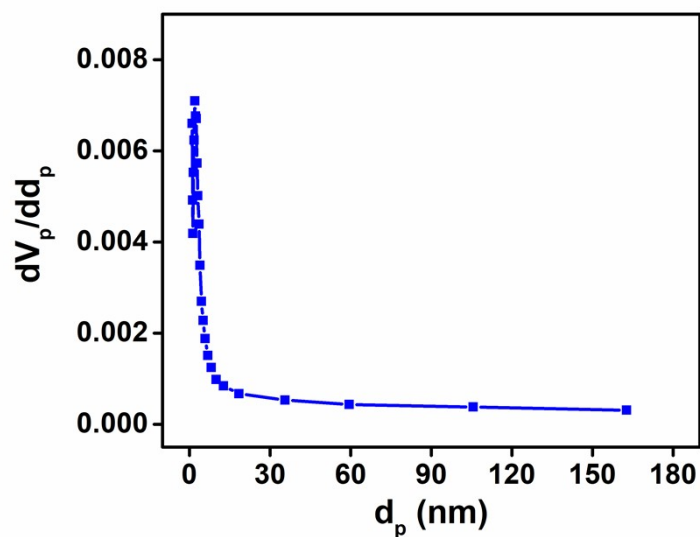


Figure S2. BJH pore size distribution of CuNi-PTC

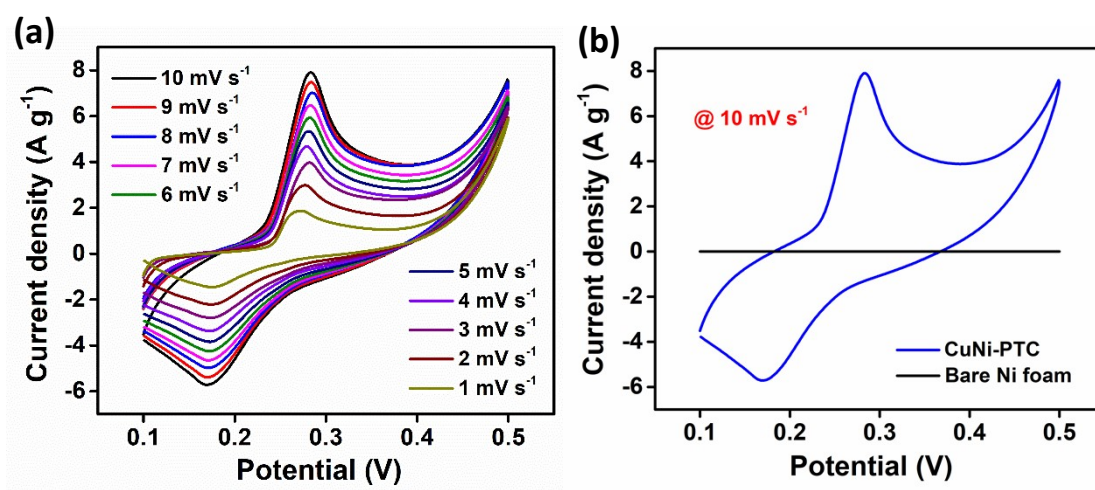


Figure S3. CV curves of (a) CuNi-PTC with scan rates varying from 10 mV s⁻¹ to 1 mV s⁻¹
(b) CuNi-PTC and bare Ni foam at 10 mV s⁻¹.

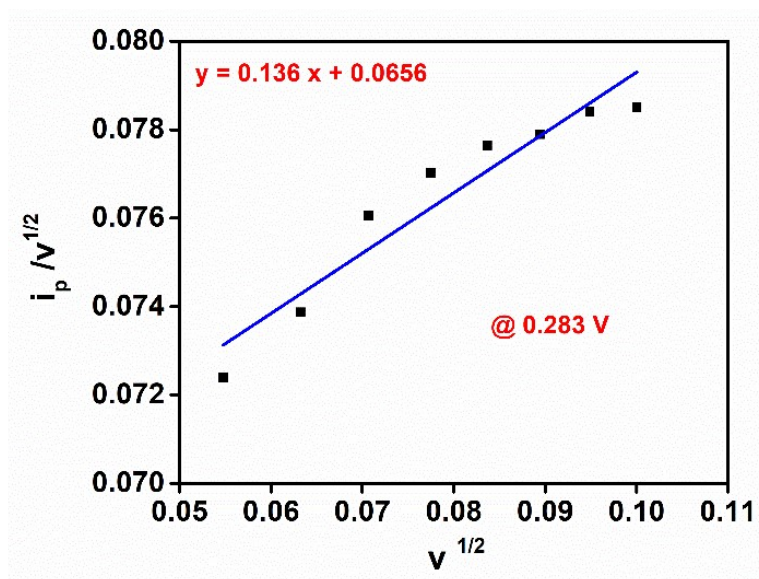


Figure S4. Linear plot of $i_p/v^{1/2}$ vs $v^{1/2}$.

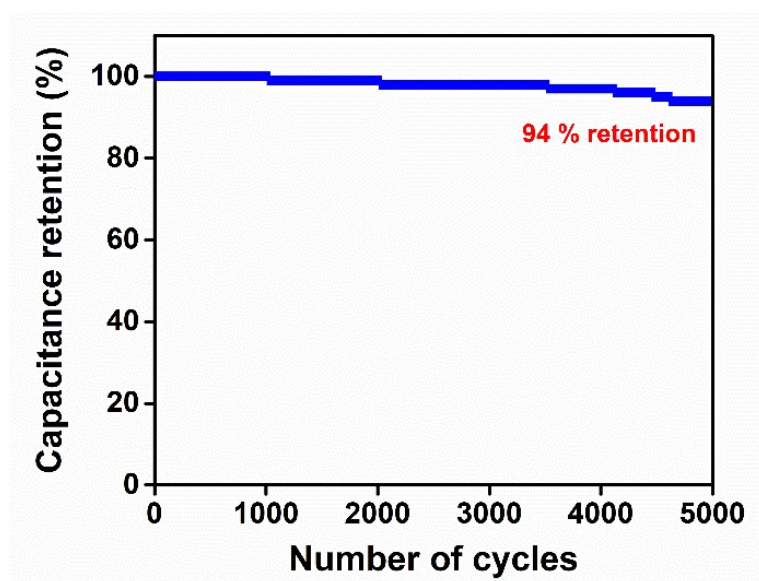


Figure S5. Stability test of CuNi-PTC.

Table S1. Summary of literature reports based on Ni-based bimetallic MOFs for supercapacitors

Sl. No	MOF	Specific capacitance	Capacitance retention	Reference
1.	Fe/Ni-BDC	1190.88 F/g @1 mV/s	93.7 % after 5000 cycles	¹
2.	Co/Ni-MOF-2:1	610 F/g @0.5 A/g	95.5 % after 5000 cycles	²
3.	Zn-doped Ni-MOF	1620 F/g @0.25 A/g	91 % after 3000 cycles	³
4.	CoNi-ZIF@N-CNT-2	1118 F/g @1 A/g	72.5 % after 10000 cycles	⁴
5.	Mn/Ni-MOF@MWCNTs	793.6 F/g @1 A/g	78.3 % after 2000 cycles	⁵
6.	Co/Ni-MOF	2608 F/g @1 A/g	88 % after 5000 cycles	⁶
7.	Zn-Ni MOF	466.5 F/g @0.5 A/g	44 % after 2500 cycles	⁷
8.	CuNi-PTC	1066.24 F/g @1 A/g	94 % after 5000 cycles	This work

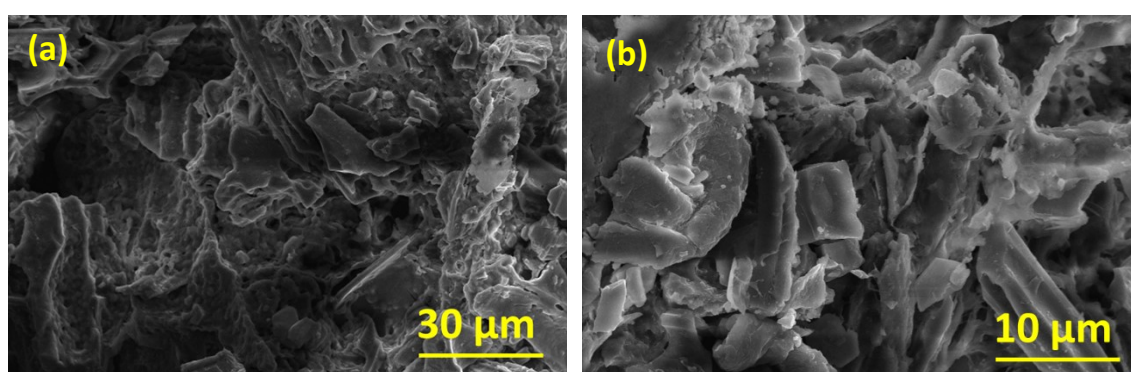


Figure S6. FESEM images of CuNi-PTC after 5000 charge-discharge cycles.

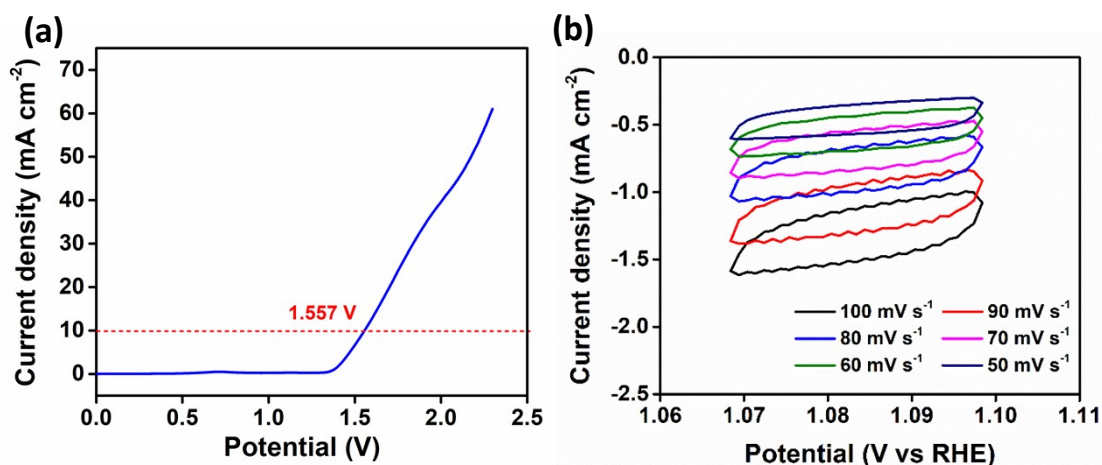


Figure S7. (a) LSV curve for overall water splitting (b) Non faradic cyclic voltammograms with scan rates varying from 100 mV s^{-1} to 50 mV s^{-1} .

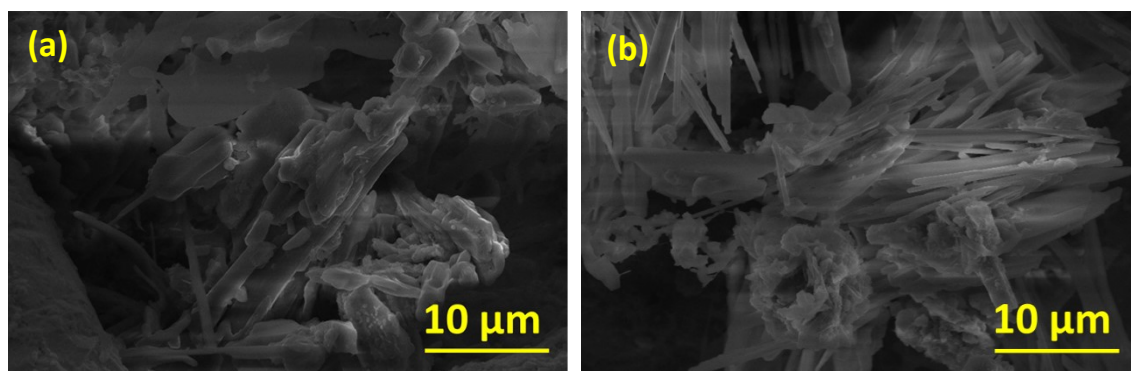


Figure S8. FESEM images of CuNi-PTC after Bulk electrolysis for 2 h.

References

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