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Supplementary Information

Study on fluorescence properties both in solutions and the solid states of three *N,O*-chelated difluoroboron compounds and their application in latent fingerprint imaging and ink-free writing

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Figure S10 HRMS of compound 3-BF₂.





10⁻⁵ mol/L), respectively.



Figure S13 (a) Emission spectra of three compounds in THF/H₂O mixtures with different f_w (c = 1 × 10⁻⁵ mol/L, excited at 390 nm), and (b) plots of emission peak intensity versus f_w . Inset: photos in different f_w under 365 nm irradiation.



Figure S14 Absorption spectra of three compounds in THF/H₂O mixtures with different f_w (c = 1 × 10⁻⁵ mol/L), respectively.



Figure S15 (a) Normalized emission spectra (λ_{ex} = 390 nm) and (b) XRD patterns of compound 2-BF₂ in different solid states. Inset: photos of luminescence colors change under a 365 nm UV lamp irradiation.



Figure S16 (a) Normalized emission spectra (λ_{ex} = 390 nm) and (b) XRD patterns of compound **3-BF**₂ in different solid states. Inset: photos of luminescence colors change under a 365 nm UV lamp irradiation.







Figure S18 The τ of compound 2-BF₂ in (a) as-synthesized and (b) grinding states.



Figure S19 The τ of compound 3-BF₂ in (a) as-synthesized and (b) grinding states.



Figure S20 The Φ_f of compound **1-BF**₂ in as-synthesized (up) and grinding (down) states.



Figure S21 The Φ_f of compound 2-BF₂ in as-synthesized (up) and grinding (down) states.



Figure S22 The Φ_f of compound **3-BF**₂ in as-synthesized (up) and grinding (down) states.



Figure S23 The DSC of compound $1-BF_2$ in as-synthesized (up) and grinding (down) states.









Figure S26 Photographs of LFPs undergoing different aging times with $1-BF_2$ and $3-BF_2$ contrast agents on tinfoil (365 nm UV light)

		1		-/			0	
		<i>n</i> -hexane	PhCl	CH ₂ Cl ₂	THF	EtOAc	CH₃CN	CH₃OH
	UV (λ _{abs} , nm)	282, 377	285, 384	287, 379	289, 391	286, 386	286, 383	288, 387
1 05	PL (λ _{em} , nm)	444	481	484	510	502	546	538
1-DF2	Δν _{st} (10 ³ , cm ⁻ 1) ^a	4.0027	502516	5.7241	5.9676	5.9864	7.7946	7.2524
	UV (λ _{abs} , nm)	312, 395	315, 397	313, 394	311, 387	309, 385	308, 383	310, 386
2 DE	PL (λ _{em} , nm)	500	504	513	504	507	508	509
2-DF ₂	Δν _{st} (10 ³ , cm ⁻ 1) ^a	5.3161	5.3476	5.8875	5.9985	6.2502	6.4246	6.2604
	UV (λ _{abs} , nm)	312, 396	315, 396	314, 394	311, 386	309, 385	309, 384	310, 387
2 DE	PL (λ _{em} , nm)	496	508	513	504	501	509	509
3-BF2	$\Delta v_{\rm st} (10^3, {\rm cm}^3)^a$	501552	505675	508875	6.0655	6.0139	6.3953	6.1934
	Δf^{b}	-0.006	0.1425	0.2185	0.1906	0.2	0.3055	0.3085

Table S1 Photophysical data of compounds 1-BF₂, 2-BF₂ and 3-BF₂ in various organic solvents.

 $^{a}\Delta v_{st} = \Delta v_{ICT, abs} - \Delta v_{em}$

 ${}^{b}\Delta f$ referred to solvent polarity parameters, it was calculated as follows:

$$\Delta f = \frac{\varepsilon - 1}{2\varepsilon + 1} - \frac{n^2 - 1}{2n^2 + 1}$$

where ε was the static dielectric constant, *n* was the optical refractive index of the solvent.

Table S2 The emission	wavelengths and	intensities of	compounds	1-BF ₂ , 2-BF ₂	and 3-BF2 in	mixtures
of DMSO/water.						

	<i>f</i> _w (%)	0	10	20	30	40	50	60	70	80	90	100
1.05	Wavelength (nm)	557	542	547	541	536	534	522	485	464	464	497
1-DF2	Intensity (a.u.)	18	13	18	19	17	16	13	15	82	275	782
2 05	Wavelength (nm)	511	512	513	515	516	506	502	502	502	502	504
2-DF ₂	Intensity (a.u.)	149	153	155	146	106	92	114	135	84	119	60
3-BF ₂	Wavelength (nm)	509	512	514	517	516	490	489	504	498	504	504
	Intensity (a.u.)	707	914	872	811	492	78	157	658	546	478	337

Computational details

Kohn-Sham density functional theory (DFT) has been employed to optimize the ground state geometries of the investigated complexes at the B3LYP/6-31G(d, p) level. All the optimized geometries were tested to be local minima by frequency calculations at the same level. To get insight into the photophysical properties of the investigated complexes, time-dependent density functional theory (TD-DFT) calculations at the CAM-B3LYP¹/6-31G(d, p) have been performed. The effect of the solvent was considered in all DFT and TD-DFT calculations utilizing the integral equation formalism polarized continuum model (IEF-PCM) with the dichloromethane as solvent which has been employed in the experiment. All the DFT and TD-DFT calculations were performed using the Gaussian 16 software suit.²

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Car	tesian coordi	inates at th	e IEF-PCM-	Н	7.289534	0.243817	2.176965
B3L	YP/6-31G(d. p)) level.		Н	8.093346	0.176322	0.608947
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15				Н	-8.104336	-2.196783	0.371248
	F _			F	-1.553089	3.543997	-1.161401
1-D	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	1 875485	-0 370272	F	-1.646670	3.209021	1.101135
c	0 996746	2 495362	-0.410978				
N	-0 161313	1 821034	-0 203705	81			
C	-0 146662	0 473821	0.026344	2-B	F ₂		
C	1 091874	-0 193889	0.087232	С	7.839646	-0.589895	-1.915389
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Н	-8.093658	2.795977	1.766080	Н	5.459511	-4.283612	0.049267
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F	-2.850607	-2.804134	-1.239927	Н	5.682520	-3.793862	-1.626483
				Н	7.086136	1.622152	0.786803
Car	tesian coordii	nates at the	TD-IEF-PCM-	Н	7.314601	0.383872	2.021153
CA	M-B3LYP/6-310	G(d, p) level.		Н	8.080956	0.197731	0.445676
				Н	-7.101020	-3.702243	0.429637
45				Н	-8.050183	-2.295329	0.126567
1-B	F ₂			F	-1.562433	3.649557	-0.983628
С	2.194706	1.877878	-0.374636	F	-1.667471	3.088432	1.224186
С	0.972733	2.497287	-0.404587				

81				Н	8.914350	-0.933343	-1.906655
2-B	F ₂			Н	7.324290	-2.307255	-3.204007
С	7.860793	-0.925024	-1.654808	Н	5.572889	0.402196	0.447035
С	6.966199	-1.687591	-2.385492	Н	3.462676	-0.143440	-0.430949
Ν	5.648076	-1.708520	-2.174018	Н	6.885564	2.345867	0.378390
С	5.189221	-0.962163	-1.175310	Н	10.642174	2.908274	2.342740
С	6.004952	-0.160946	-0.372775	Н	9.878472	-0.721784	0.214870
С	7.374345	-0.125378	-0.614838	Н	1.020261	-3.702909	-0.432933
С	8.279113	0.716254	0.205195	Н	-1.383007	-3.286694	-0.057142
Ν	3.803865	-0.942371	-0.943701	Н	-0.731800	0.900455	-0.843854
С	2.927354	-2.000787	-1.079425	Н	1.613320	0.460383	-1.266101
С	1.494026	-1.657940	-0.846379	Н	12.499962	1.334581	1.790180
0	3.278046	-3.138369	-1.346902	Н	11.734207	0.023369	2.684202
С	7.868028	1.973523	0.652589	Н	12.150112	-0.203670	0.986110
С	8.710639	2.775446	1.418295	н	7.188679	4.228442	1.903606
С	9.977873	2.292031	1.741843	н	8.678952	4.395292	2.844633
С	10.415203	1.039254	1.314384	Н	8.633839	4.915045	1.161966
С	9.556260	0.262737	0.540069	Н	-2.680515	0.742840	-1.456669
С	0.620461	-2.698580	-0.510178	н	-8.243775	1.381008	0.437829
С	-0.719629	-2.465771	-0.291151	Н	-4.605763	2.192539	-1.691808
С	-1.246749	-1.162856	-0.419848	н	-8.162322	-2.947782	1.894703
С	-0.369977	-0.117148	-0.770588	Н	-8.093817	-2.322708	0.241470
С	0.973408	-0.367251	-0.978278	Н	-6.600851	-2.787628	1.076209
С	11.773631	0.522346	1.710160	н	-7.381444	-1.538325	3.842035
С	8.277126	4.149231	1.858783	Н	-6.722871	0.081706	3.577384
Ν	-2.602550	-0.942577	-0.209897	Н	-5.784735	-1.346174	3.103201
С	-3.217658	0.142135	-0.735931	н	-9.466098	-0.348156	1.038725
С	-4.560458	0.480159	-0.435259	Н	-9.429081	-1.028987	2.664788
С	-5.323543	-0.330533	0.484965	Н	-8.909348	0.636298	2.407455
С	-6.674162	-0.015204	0.811908	Н	-6.826068	4.772295	-2.668451
С	-7.222685	1.114029	0.210277	Н	-5.393931	4.304172	-1.752212
С	-6.508767	1.926446	-0.693744	Н	-5.954802	3.303332	-3.106666
С	-5.176267	1.584879	-1.001421	Н	-8.935265	3.534139	-2.542129
0	-4.749095	-1.371564	1.018408	Н	-9.154852	2.152700	-1.463273
С	-7.473368	-0.896344	1.774418	Н	-8.146501	1.991303	-2.908193
С	-7.579636	-2.327554	1.207029	Н	-6.782418	4.442453	0.398166
С	-6.788193	-0.925905	3.156415	Н	-8.139128	4.964207	-0.613550
С	-8.899877	-0.367826	1.974129	Н	-8.342273	3.614439	0.507390
С	-7.195310	3.140054	-1.309019	F	-3.615967	-3.130943	0.041724
С	-6.280303	3.917183	-2.262408				
С	-8.433070	2.669082	-2.099593	75			
С	-7.641276	4.091888	-0.180496	3-BF	2		
В	-3.404363	-1.929944	0.696213	С	8.483414	-0.344088	-1.319904
F	-2.743015	-2.106419	1.889455	С	7.654953	-1.047499	-2.176964

Ν	6.330003	-1.135329	-2.038248	Н	11.791313	1.010556	2.241752
С	5.794288	-0.522675	-0.988424	Н	10.405785	-0.279336	0.665481
С	6.539004	0.209260	-0.059922	Н	1.706432	-3.466054	-0.780528
С	7.915825	0.316179	-0.224780	Н	-0.719855	-3.176131	-0.429508
С	8.746083	1.089683	0.729774	Н	-0.198997	1.094202	-0.680633
Ν	4.400465	-0.576493	-0.827286	Н	2.172698	0.785237	-1.079042
С	3.569846	-1.634373	-1.141879	Н	-2.130285	0.939313	-1.375086
С	2.117572	-1.370875	-0.923096	Н	-7.724144	1.209742	0.521572
0	3.971710	-2.713722	-1.543737	Н	-4.100974	2.344247	-1.482688
С	8.266485	2.274326	1.296214	Н	-7.536935	-3.244646	1.506535
С	9.046119	2.999235	2.188413	Н	-7.480532	-2.444127	-0.069677
С	10.316955	2.549879	2.531158	Н	-5.978324	-2.960856	0.716617
С	10.803754	1.371852	1.974516	Н	-6.790898	-2.033918	3.597567
С	10.025689	0.648598	1.079662	Н	-6.170055	-0.380154	3.509771
С	1.273276	-2.473160	-0.746444	Н	-5.198695	-1.727038	2.888540
С	-0.079602	-2.312136	-0.541377	Н	-8.902562	-0.602395	0.931835
С	-0.648626	-1.020891	-0.523694	Н	-8.848757	-1.451335	2.476244
С	0.198035	0.087831	-0.719985	Н	-8.370415	0.244201	2.399054
С	1.554754	-0.091308	-0.914285	Н	-6.391571	4.945896	-2.204884
Ν	-2.015400	-0.871477	-0.322383	Н	-4.945248	4.421357	-1.343605
С	-2.658030	0.245638	-0.735272	Н	-5.487643	3.550112	-2.791864
С	-4.013618	0.508460	-0.417426	Н	-8.469812	3.658859	-2.199944
С	-4.759780	-0.420371	0.399345	Н	-8.658287	2.170898	-1.267359
С	-6.119113	-0.177229	0.751094	Н	-7.653711	2.177110	-2.724070
С	-6.695700	0.995961	0.272045	Н	-6.321706	4.302590	0.812785
С	-6.000965	1.919781	-0.534568	Н	-7.698006	4.894001	-0.132007
С	-4.657715	1.650003	-0.866725	Н	-7.860506	3.429734	0.842797
0	-4.161687	-1.501186	0.814629	F	-2.146563	-2.283882	1.619137
С	-6.896894	-1.176590	1.610212	F	-2.968759	-3.100543	-0.350992
С	-6.967975	-2.540598	0.891909				
С	-6.211757	-1.338129	2.983164				
С	-8.336071	-0.707535	1.861346				
С	-6.716966	3.175095	-1.018700				
С	-5.824230	4.067051	-1.889065				
С	-7.949183	2.763057	-1.849603				
С	-7.176462	3.993586	0.205087				
В	-2.798756	-1.981997	0.446160				
Н	9.547392	-0.289260	-1.516208				
Н	8.075581	-1.558218	-3.039722				
Н	6.046480	0.662147	0.793450				
Н	4.006282	0.137613	-0.233580				
Н	7.284517	2.643098	1.018592				
Н	8.661665	3.921104	2.612203				
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