

Supplementary Information

Modulation of Acid-Basic and Metal-Carrier Interactions in CO₂-assisted Propane Dehydrogenation over Ga₂La_x/ZSM-5 Catalysts

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Text S1. Catalyst characterization

X-ray diffraction (XRD) patterns of as-prepared samples were recorded on an X-ray diffractometer (D/Max-2200, Rigaku, Japan) using Cu K α irradiation ($k = 0.154$ nm), operating at 30 mA and 40 kV. The structural properties (specific surface area and pore size distribution) of the catalyst were determined by N₂ adsorption-desorption experiments, which were performed on a NOVA 4200e instrument at -196 °C. The scanning electron microscopy (SEM, ZEISS Sigma 360, Germany) was used to detect morphologies. NH₃/CO₂ temperature programmed desorption (NH₃/CO₂-TPD) profiles were conducted to monitor the intensity of the desorbed using a highly sensitive thermal conductivity detector (TCD). Before performing NH₃/CO₂-TPD measurements, 0.05g catalyst was pretreated at 450 °C for 30 min and cooled down to 100 °C under a He purge. Then 10vol % NH₃/CO₂ (He as the carrier gas, 30ml/min) was introduced into the reactor for 2 h. The aim was to saturate the acidic sites of the measured samples. Finally, the NH₃/CO₂-TPD analysis was performed at 30 ml/min He flows with the temperature rising from 100 to 800 °C. X-ray photoelectron spectroscopy (XPS) of samples was recorded on an ESCALAB 250XI spectrometer (XPS, Thermo Scientific K-Alpha, USA) using an Al K α X-ray source. Thermogravimetric Analysis (TGA, NETZSCHSTA2500, Germany) for measurement of carbon deposition in reaction materials. The spent samples were heated in the air at a ramp rate of 10 °C/min from room temperature up to 800 °C. Diffuse reflectance infrared Fourier transform spectroscopy (*in situ* DRIFTS) was recorded on a Nicolet iS50 FT-IR (Thermo Fisher, America) using an in-situ cell. To further verify the intermediate, the reaction pathway, and the reaction mechanism for CO₂-PDH. Each sample was loaded and pretreated for 30 minutes at 550 °C under a stream of nitrogen (30 mL/min) and a DRIFTS spectrum was collected as background. C₃H₈: CO₂: N₂ (1:1:4) was then ramped up from 30 °C to 550 °C at a flow rate of 30 mL/min. And the DRIFTS spectra were recorded automatically every 30 s by OMNIC™ Series Software.

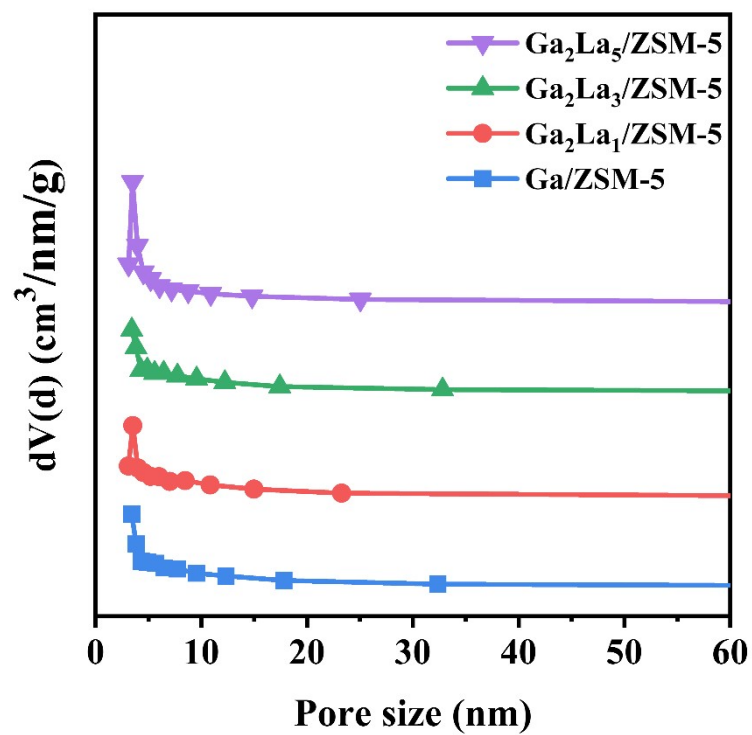


Fig. S1. Pore size distribution curves of Ga/ZSM-5 and Ga₂La_x/ZSM-5 (x = 1, 3, 5) catalysts.

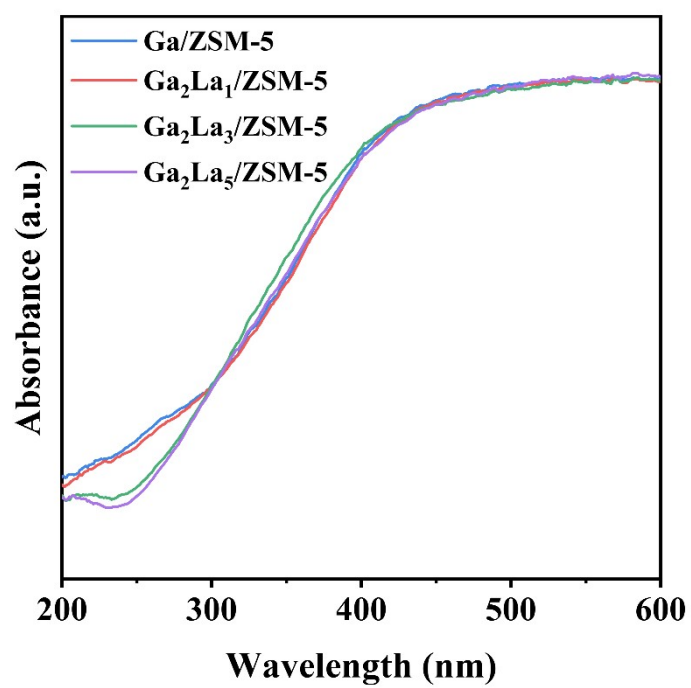


Fig. S2. The UV-vis DRS of Ga/ZSM-5 and Ga₂La_x/ZSM-5 (x = 1, 3, 5) catalysts.

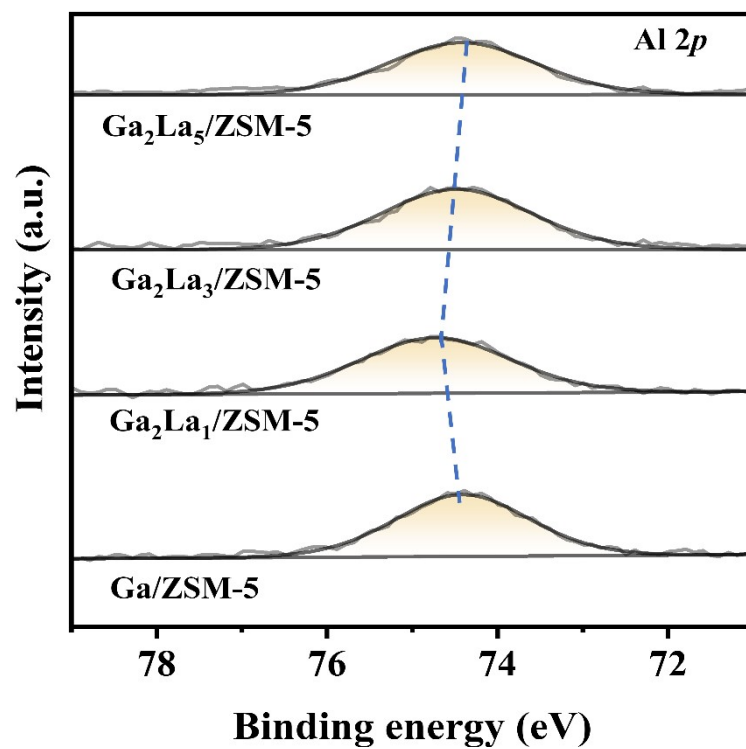


Fig. S3. XPS spectra of Al 2p for Ga/ZSM-5 and Ga₂La_x/ZSM-5 (x = 1, 3, 5) catalysts.

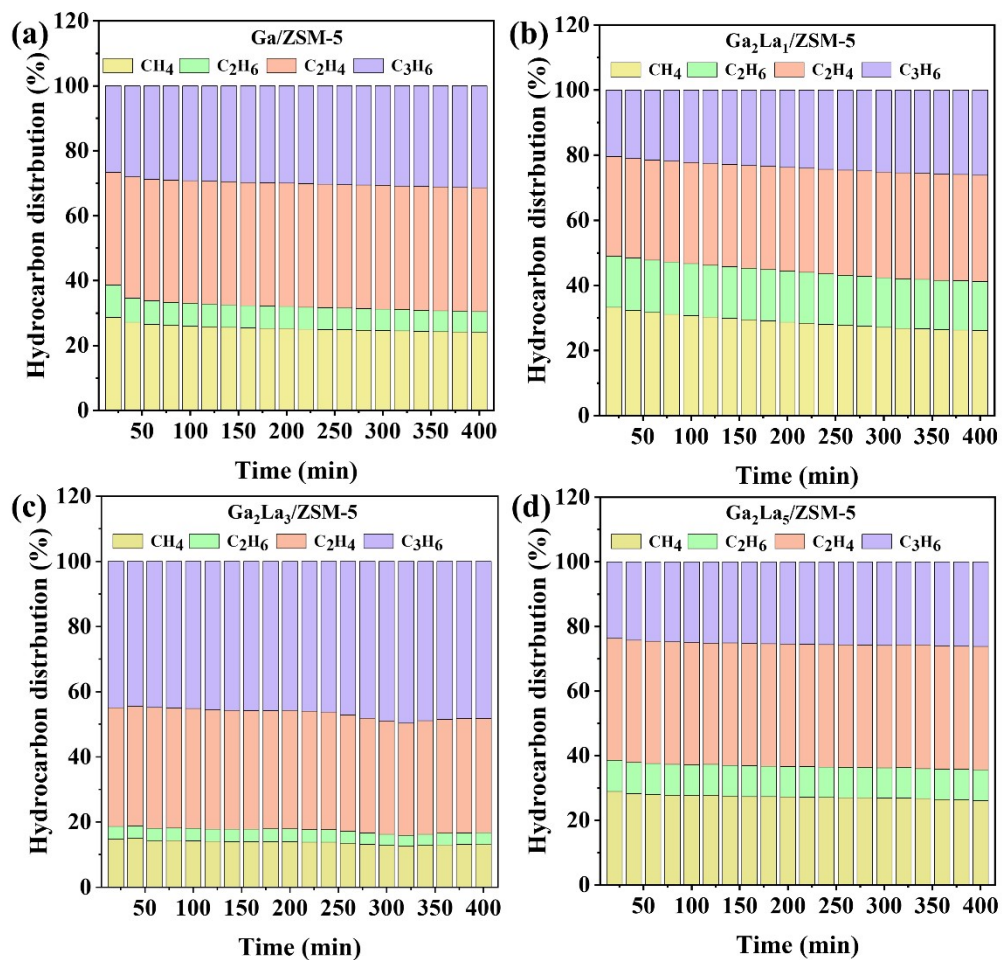


Fig. S4. Hydrocarbon distribution of Ga/ZSM-5 and Ga₂La_x/ZSM-5 (x = 1, 3, 5) catalysts

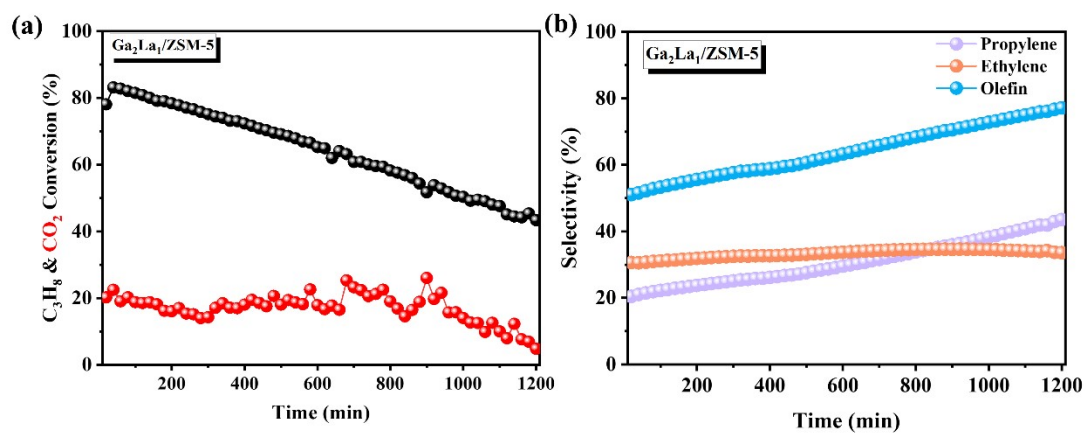


Fig. S5. Long-term stability test for CO₂-ODHP conversion (a) and selectivity (b) over Ga₂La₁/ZSM-5 catalyst.

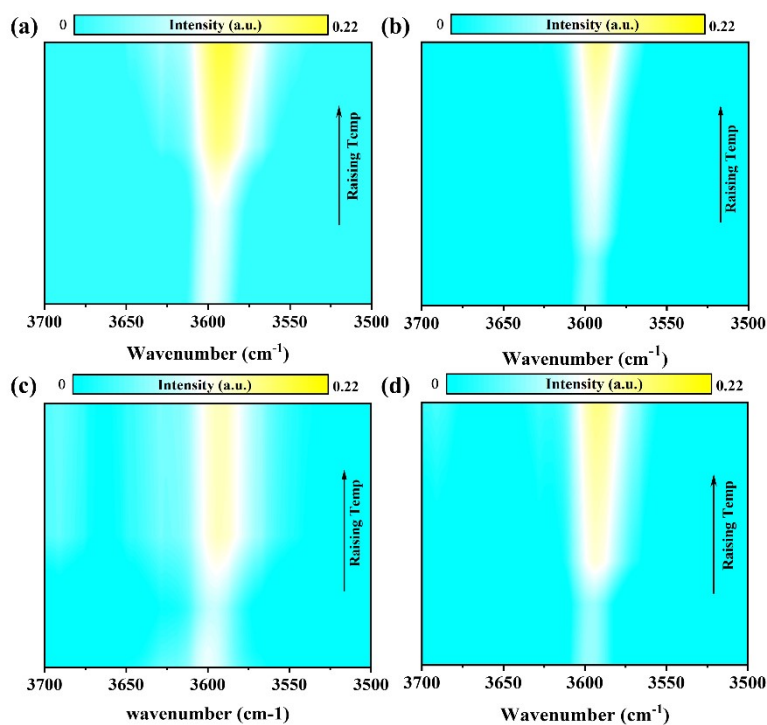


Fig. S6. Color-mapping spectra in the 3700–3500 cm^{-1} range of Si-OH-Ga on the Ga/ZSM-5 (a), Ga_2La_1 /ZSM-5 (b), Ga_2La_3 /ZSM-5 (c) and Ga_2La_5 /ZSM-5 catalysts (d).

Note: The data in the plots are from *in situ* infrared. The Si-OH-Ga groups in the region of 3700–3500 cm^{-1} in the *in situ* DRIFTS were selected for the contour mapping. The color gradation is from cyan to yellow, and the darker the color, the stronger the signal strength of Si-OH-Ga. The intensity of Si-OH-Ga groups of the four catalysts was visualized as a reflection of the change in the intensity of the Si-OH-Ga groups with temperature.

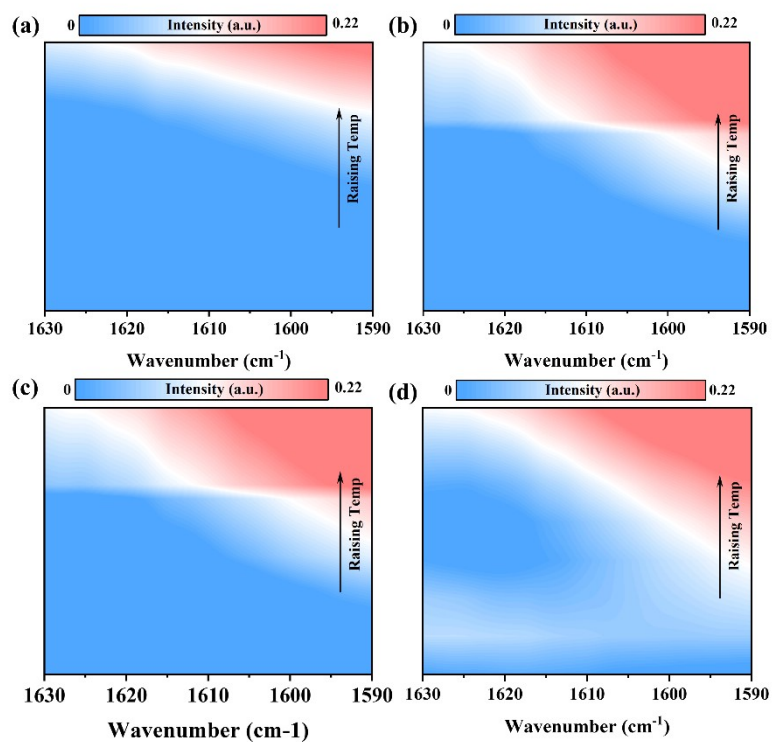


Fig. S7. Color-mapping spectra in the 1630-1590 cm⁻¹ range of HCOO* on the Ga/ZSM-5 (a), Ga₂La₁/ZSM-5 (b), Ga₂La₃/ZSM-5 (c) and Ga₂La₅/ZSM-5 catalysts (d).

Note: The data in the plots are from *in situ* infrared. The HCOO* groups in the region of 1630-1590 cm⁻¹ in the *in situ* DRIFTS were selected for the contour mapping. The color gradation is from blue to red, and the darker the color, the stronger the signal strength of HCOO*. The intensity of HCOO* groups of the four catalysts was visualized as a reflection of the change in the intensity of the HCOO* groups with temperature.

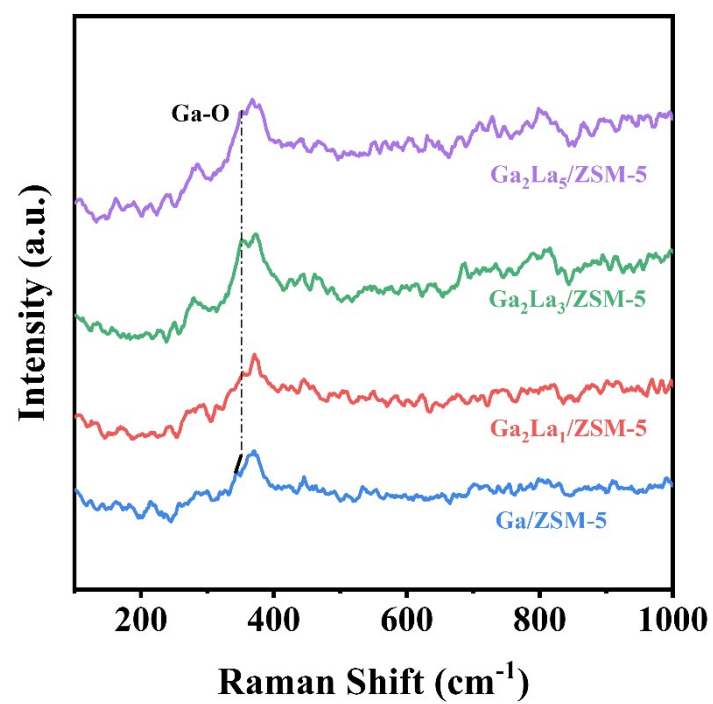


Fig. S8. Raman spectra of $\text{Ga}/\text{ZSM-5}$ and $\text{Ga}_2\text{La}_x/\text{ZSM-5}$ ($x = 1, 3, 5$) catalysts

Table S1. Comparison of Ga₂La₁/ZSM-5 catalyst performance with previously reported catalysts.

Catalyst	m/g	T/°C	C ₃ H ₈ /CO ₂ / N ₂	WHSV ^[a]	X /C ₃ H ₈	S /C ₃ H ₆	Stability	Ref
5GaN/NaZSM-5(470)	0.2	600	1 : 2 : 7	9000	45%	87.8%	48h, 17%	1
5GaN/2000-Fe-Silicalite-1	0.2	600	1.5 : 3 : 25.5	9000	55%	70%	20h, 28%	2
5GaN/HS-1	0.2	600	1 : 2 : 7	9000	67%	58%	40h, 20%	3
5.2V-MSNS	0.2	600	1 : 4 : 4	4500	58%	89%	24h, 18%	4
6.8V-MCM-41	0.2	600	1 : 4 : 4	4500	58%	90%	2h, 30%	5
Fe ₃ Ni-CeO ₂	0.1	550	1 : 1 : 2	24000	4%	58.2%	14h, %	6
PtSn _{1.5} /SiO ₂	0.2	550	1 : 1 : 20	3000	50%	96%	20h,38%	7
2CrO _x /Silicalite-1	0.5	550	4 : 20 : 1	3000	43%	85%	2h, 35%	8
5%Cr/Ce _{0.2} Zr _{0.8} O ₂	0.2	600	1 : 2 : 37	6000	56%	79.4%	6h, 28%	9
3.4V/In	0.1	500	1 : 1 : 27	18000	2.5%	90%	/	10
PtCoIn/CeO ₂	0.1	550	1 : 1 : 2	12000	50%	95%	20h, 47%	11
HEI/CeO ₂	0.1	600	1 : 1 : 2	12000	25%	90%	50h,19%	12
ZnFe ₂ O _x /S-1	0.2	580	1 : 1 : 4	7200	38%	95%	110h , 25%	13
5Fe-5V-Al ₂ O ₃	0.2	600	15 : 15 : 70	15000	43%	80%	1h, 80%	14
Ga/Na-SSZ-39(9)	0.5	600	1 : 1 : 18	7200	39%	63%	2h, 39%	15
Cr-CNFp	0.5	600	1 : 2	3600	20%	56.2%	/	16
Pt1Co1-SiBeta	0.1	550	1 : 1 : 3	12000	51.8%	91.8%	6h, 35%	17
MoC _x /Si-1	0.1	550	1 : 1 : 2	12000	13%	94%	20h, 5%	18
PtZn/Co@S-1	0.1	550	1 : 1 : 4	2000	59.7%	94.9%	60h, 40%	19
Ga ₂ La ₁ /ZSM-5	0.2	580	1 : 1 : 4	7200	80.31%	30%	20h, 43%	This work

^[a]WHSV: mL/g⁻¹·h⁻¹

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