

Supporting Information

A mild exfoliation synthesis of two-dimensional layered V_2C MXene for high performance lithium ion batteries

Yanan Xu^{a,*}, Li Li^{b*}

^a*Faculty Of Education, Tianjin Normal University, Tianjin 300387 (P. R. China)*

^b*School of Materials Science and Engineering, University of Jinan, Jinan 250022, Shandong,
China*

*Corresponding authors at: School of Materials Science and Engineering, University of Jinan,

Jinan 250022, PR China.

Tel.: +86 15621888686

E-mail addresses: xuyanan2016@126.com (Y. Xu), mse_lil@ujn.edu.cn (L. Li)

Table S1 Comparison of reversible capacity of different pure MXene

Sample	Electrolyte	Current Density	Reversible Capacity (mAh g ⁻¹)	Ref.
Ti ₂ CT _x	1.0 M LiPF ₆	0.04 C	225	S1
V ₂ CT _x	1.0 M LiPF ₆	1 C	260	S2
Nb ₂ CT _x	1.0 M LiPF ₆	1 C	170	S2
Ti ₃ C ₂ T _x	1.0 M LiPF ₆	0.1 C	178.5	S3
Nb ₄ C ₃ T _x	1.0 M LiPF ₆	0.1 A g ⁻¹	380	S4
Hf ₃ C ₂ T _x	1.0 M LiPF ₆	0.2 A g ⁻¹	146	S5
V ₂ C (HCl+LiF)	1.0 M LiPF ₆	0.1 A g ⁻¹	397.6	this work
V ₂ C (HCl+LiF-3d)	1.0 M LiPF ₆	0.1 A g ⁻¹	402.9	this work

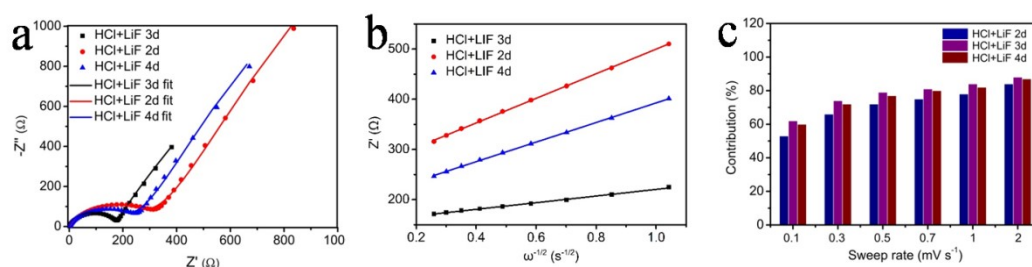


Fig. S1 (a) Nyquist plots (b) Z' - $\omega^{-1/2}$ relationship curve and (c) capacitance contribution ratio of V₂C materials etched during different time with etchant of HCl+LiF

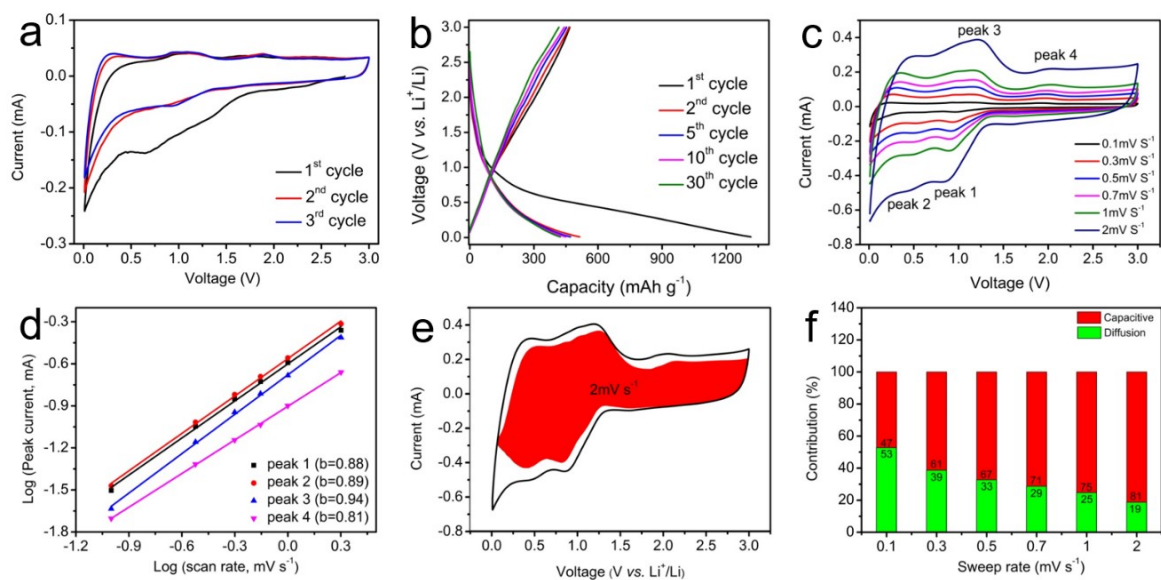


Fig.S2 Test curves of V_2C (HF) electrode: (a) CV curves at 200 mAh g^{-1} , (b) constant current charge-discharge curves, (c) CV curves at various sweep speeds, (d) the peak current-sweep velocity relation curve and, (e, f) capacitance contribution ratio at sweep speed of 2 mV s^{-1} and at various sweep rates, respectively.

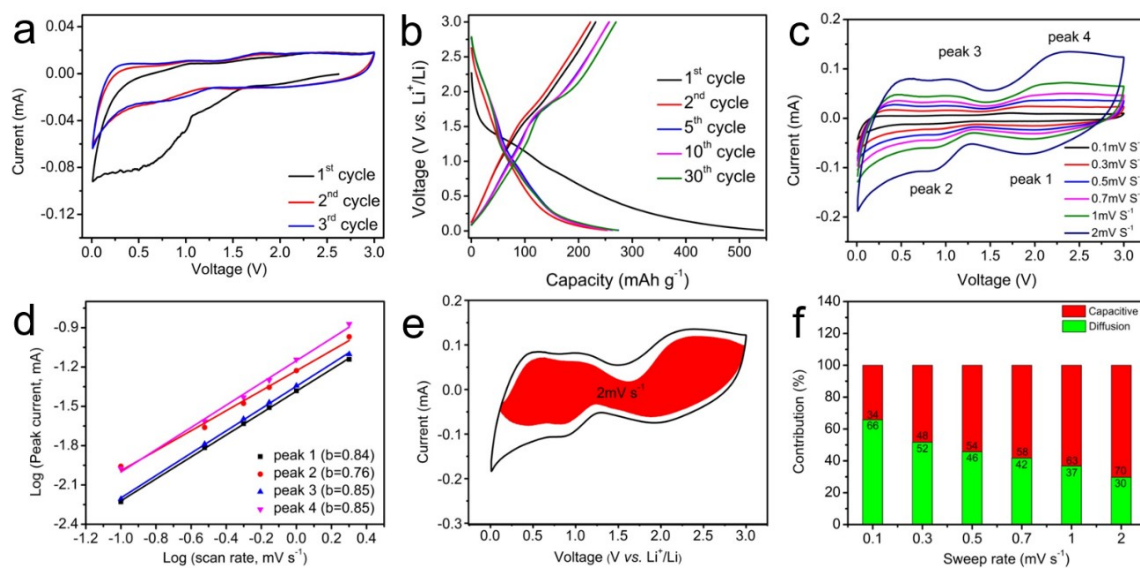


Fig.S3 Test curves of V_2C (HF+ H_2O_2) electrode: (a) CV curves at 200 mAh g^{-1} , (b) constant current charge-discharge curves, (c) CV curves at various sweep speeds, (d) the peak current-sweep velocity relation curve and (e, f) capacitance contribution ratio at sweep speed of 2 mV s^{-1} and at various sweep rates, respectively.

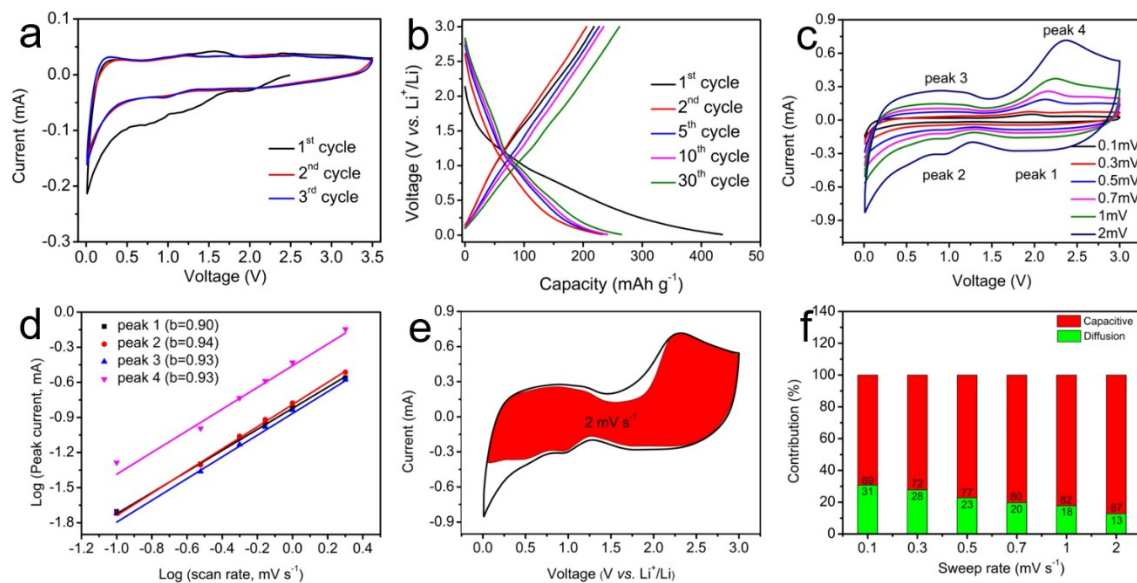


Fig.S4 Test curves of V_2C (HCl+LiF-4d) electrode: (a) CV curves at 200 mAh g^{-1} , (b) constant current charge-discharge curves, (c) CV curves at various sweep speeds, (d) the peak current-sweep velocity relation curve and (e, f) capacitance contribution ratio at sweep speed of 2 mV s^{-1} and at various sweep rates, respectively.

Supplementary References

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