Supplementary Information

A novel hollow CuS-CuO nanocube core and NiFe-LDH nanosheet

shell electrocatalyst for oxygen evolution reaction applications

Runmiao Chen,^a Long Ma,^{ad} Qingdan Hui,^a Pengjing Yin,^a Pengpeng Du,^c Qiufang Liu,^c Yan Yan,^c Qi Xue,^{*ab} Yuan Dang^{ab} and Yuanzhen Zhou^{*ab}

- ^{b.} Engineering Research Center of Low-Carbon Energy Efficient Utilization, Universities of Shaanxi Province, Xi'an University of Architecture and Technology, Xi'an, 710055, China.
- ^c Shaanxi Coal Chemical Industry Technology Research Institute Co. Ltd, Xi'an, 710100, China.
- ^{*d.*} Sinopec Great Wall Energy & Chemical (Ningxia) Co., Ltd., Ningxia 750411, China Corresponding author.
- *E-mail address: xueqi@xauat.edu.cn(*Qi Xue*), zhouyuanzhen@xauat.edu.cn(*Yuanzhen Zhou*)

^{*a.*} School of Chemistry and Chemical Engineering, Xi'an University of Architecture and Technology, Xi'an, 710055, China.

Supplementary Figures



Figure S1. Schematic illustration of the synthesis of CuS-CuO@NiFe-LDH.



Figure S2. EDX spectrums of CuS-CuO@NiFe-LDH.



Figure S3. XRD (a) and FT-IR (b) of Cu₂O.



Figure S4. Pore size distribution of the NiFe-LDH, CuS-CuO and CuS-CuO@NiFe-LDH.



Figure S5. The corresponding Tafel slopes and overpotential at 10 mA·cm⁻². (Samples 1-4 correspond in turn to Cu₂O, CuS-CuO, NiFe-LDH and CuS-CuO@NiFe-LDH, respectively.)



Figure S6. CV curves of (A) Cu₂O, (B) CuS-CuO, (C) NiFe-LDH and (D) CuS-CuO@NiFe-LDH at different scan rates from 5 to 100 mV s⁻¹.

Supplementary Tables

element	wt%
0	22.53
S	20.47
Fe	6.01
Ni	10.73
Cu	40.26

Table S1 Mass ratio of element in CuS-CuO@NiFe-LDH

Table S2 Performance comparison of representative OER electrocatalysts at a current density of 10 mA cm⁻² under 1 M KOH solution.

HER Catalysts	Overpotential (mV)	Tafel slope (mV dec ⁻¹)	Stability (h)	Ref.
CuS-CuO@NiFe-LDH	285	47.65	15	This work
CuNiP@Cu	318	100	20	1
Ni-Fe NP/N-C	330	56.7	10	2
FeCoNi-N-rGO	440	124	200 (ŋ ₅)	3
Fe-CuS/CuO/CS	340	31	10	4
Ni@3FCCO	369	69	18	5
Sn-Ni ₃ S ₂ @NF	321	-	110 (ŋ ₁₀₀)	6
Ni SAs@S/N-CMF	285	50.8	60	7
CoNiFe LDH	194	49	10	8
(Ni, Fe)S ₂ @MoS ₂	270	101.22	24	9
Ni@CN	307	-	10	10
(FeCoNiCuZn)O	323	64.5	50	11
Cu _{0.98} Dy _{0.02} CS/CuS	303	-	12 (ŋ ₅₀)	12

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