

Supporting Information

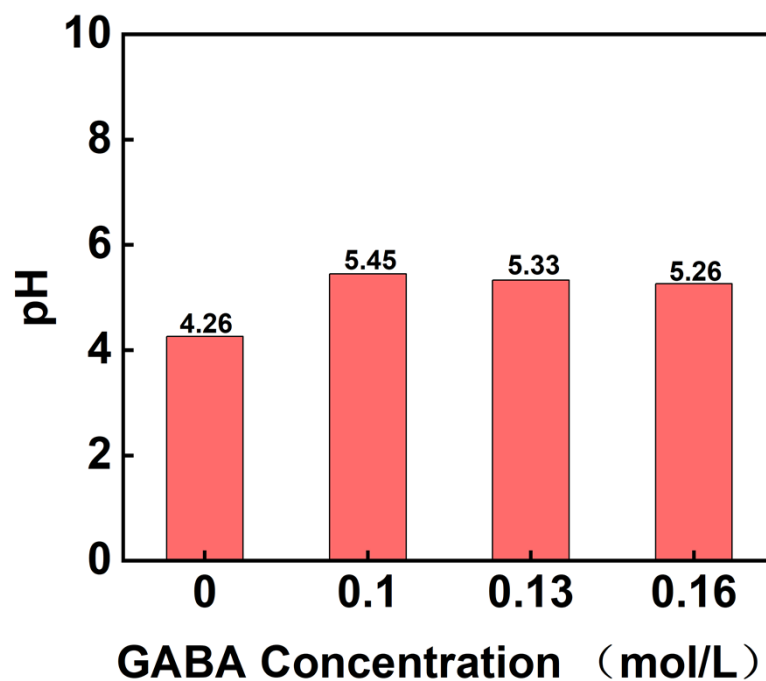


Fig. S1. The pH of PE solutions adding with different concentrations of GABA.

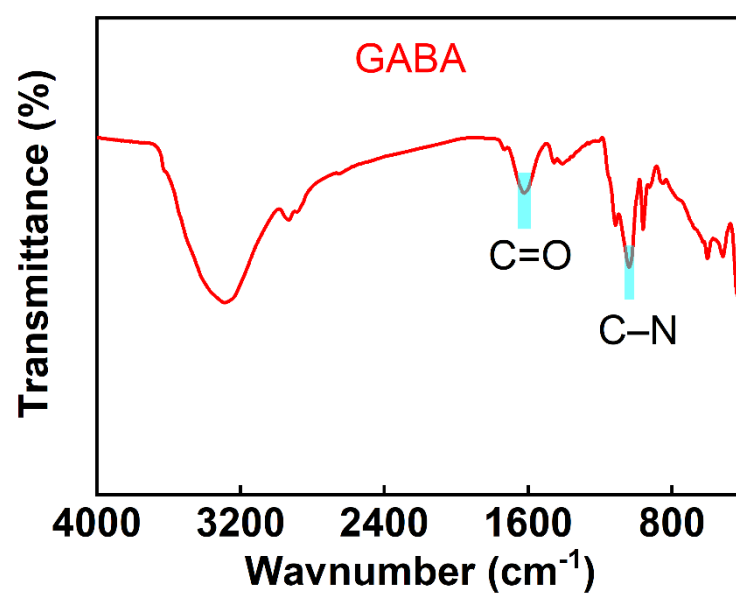


Fig. S2. FTIR spectroscopy of GABA powder.

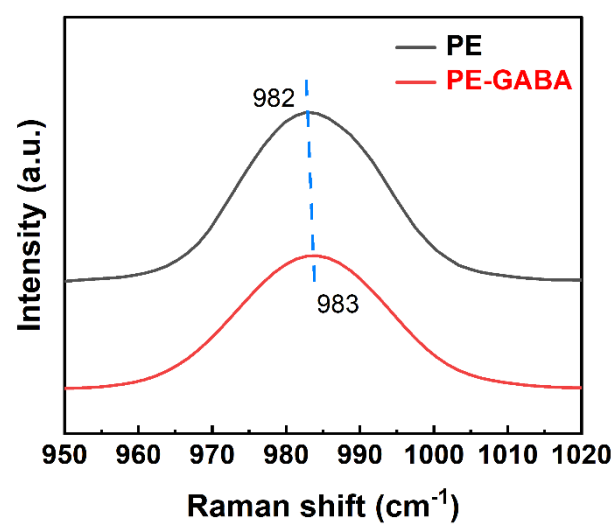


Fig. S3. Raman spectra of PE and PE-GABA.

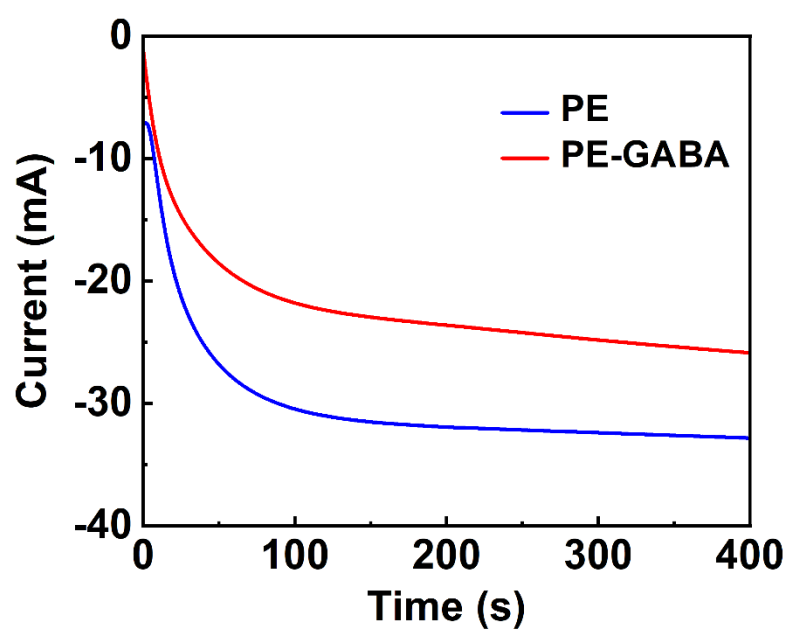


Fig. S4. CA curves of PE and PE-GABA.

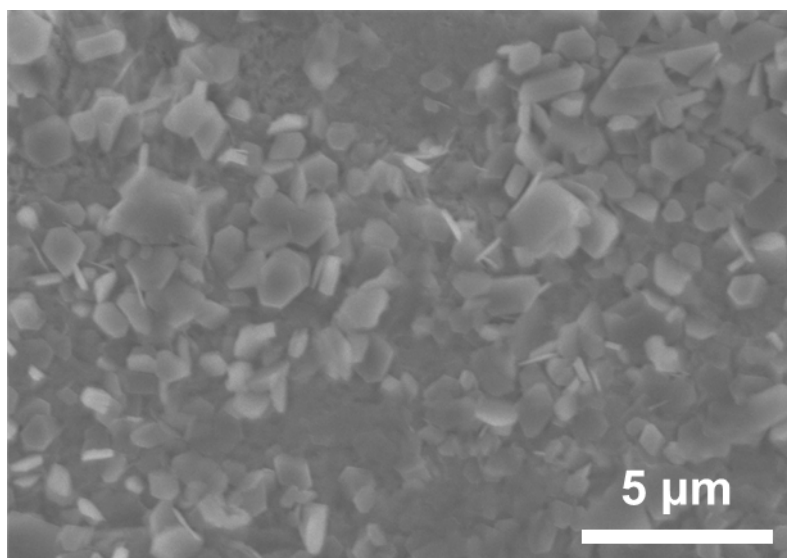


Fig. S5. SEM images of Zn electrode plated with 3 mAh cm⁻² in (a) pristine ZnSO₄ electrolyte.

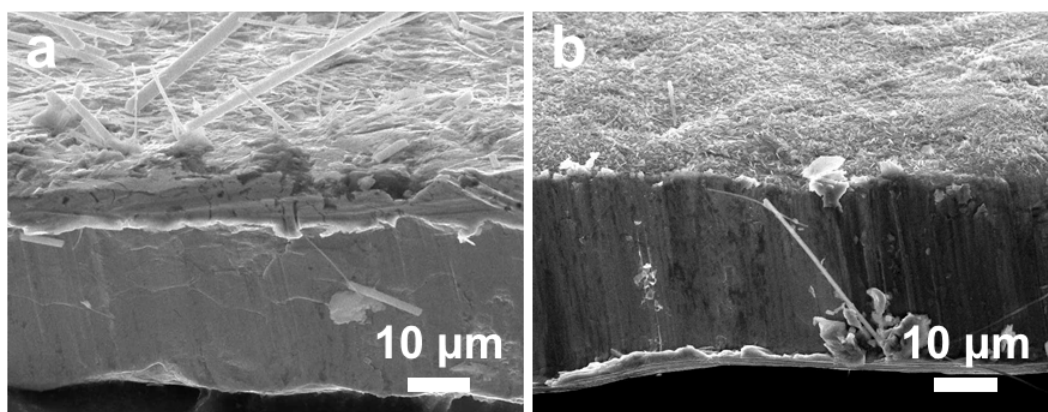


Fig. S6. Cross-sectional SEM images of cycled Zn electrode in (a) PE-GABA and (b) PE.

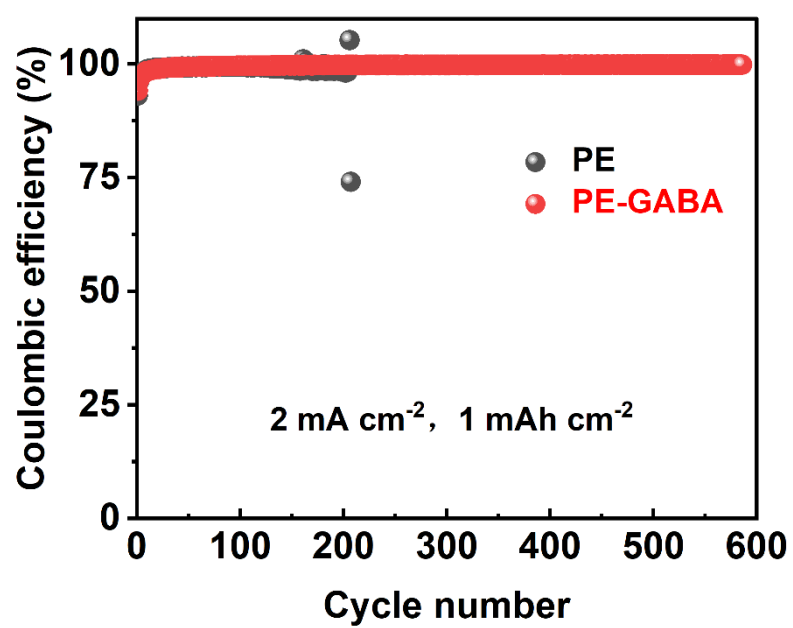


Fig. S7. CE curves at 2 mA cm⁻² in PE and PE-GABA.

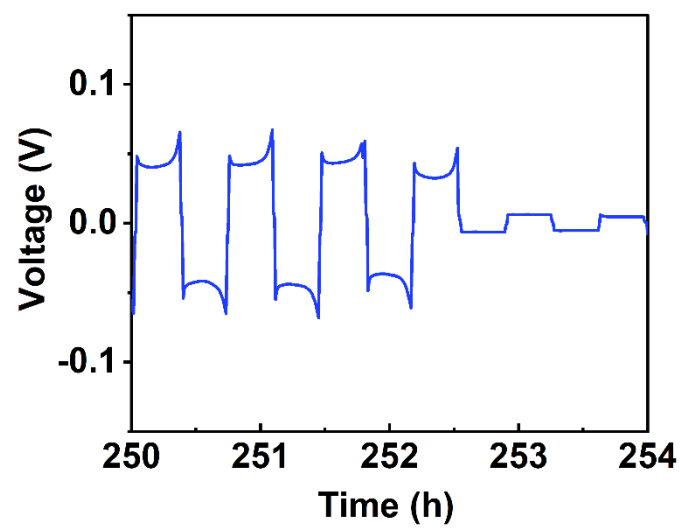


Fig. S8. Specific cycling behavior of the cell using PE.

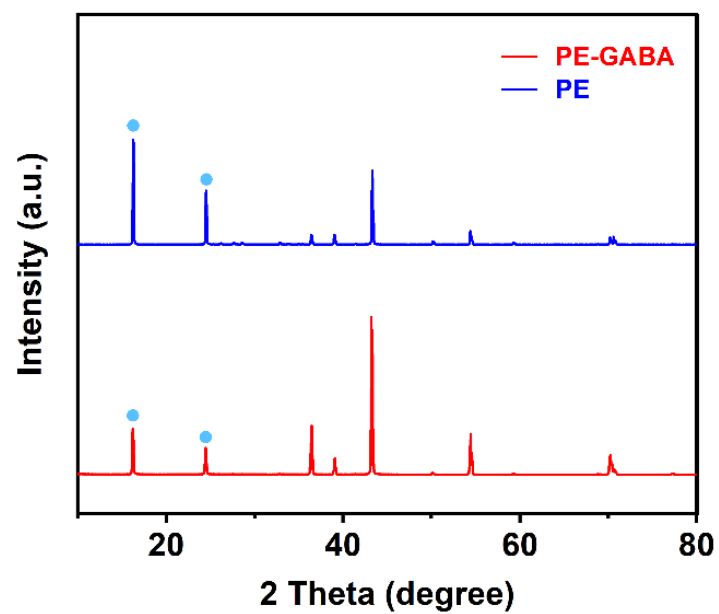


Fig. S9. XRD patterns of Zn electrodes worked at 3 mA cm⁻² after 200 cycles in PE and PE-GABA.

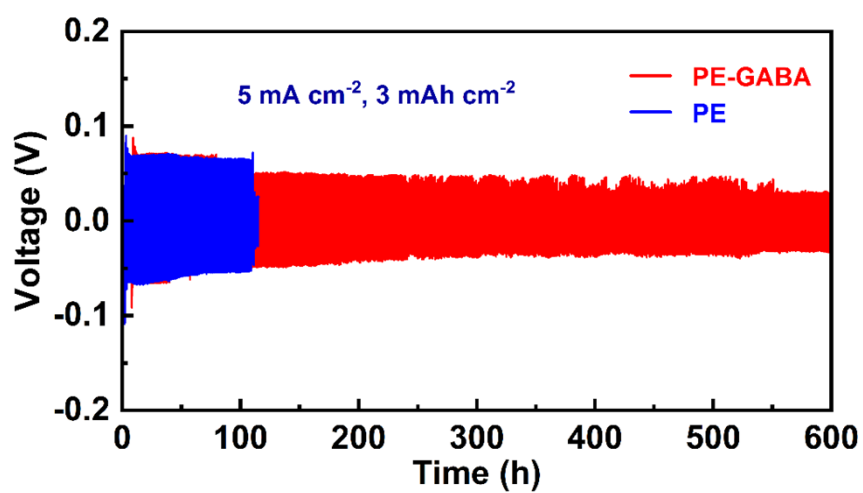


Fig. S10. Long-term cycling performances of cells at 5 mA cm^{-2} and 3 mAh cm^{-2} .

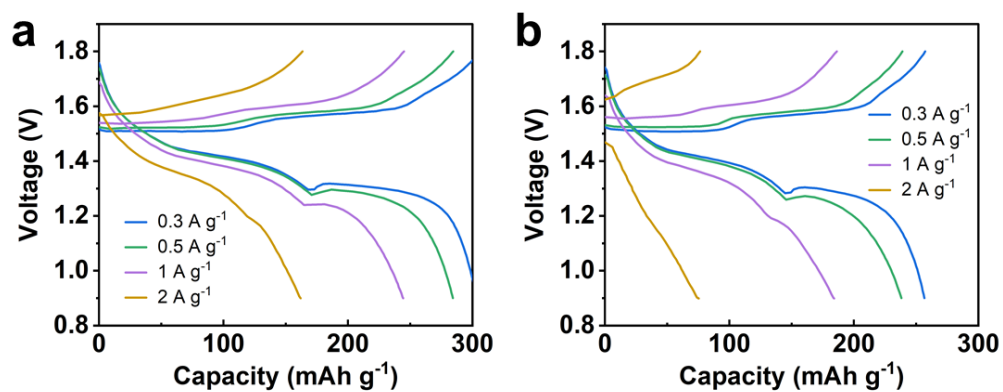


Fig. S11. Capacity-voltage plots of full cells with (a) PE-GABA and (b) PE at various current densities.

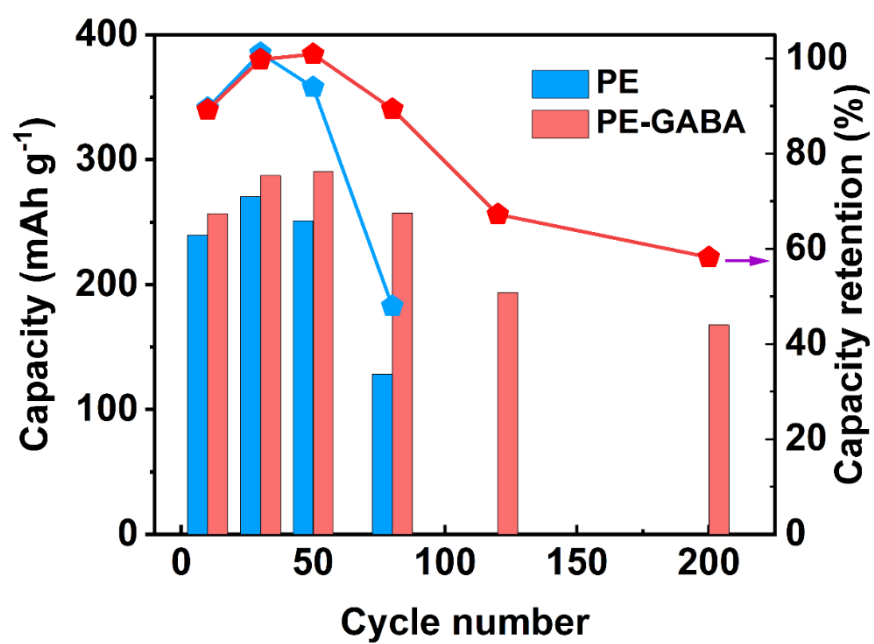


Fig. S12. Capacity delivery and retention ratio of full cells at different cycles.