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Selective hydrogenation of 5-hydroxymethylfurfural to biofuel 2,5-

dimethylfuran over CuNi/ZrO₂-SBA-15 catalyst

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Preparation of the Carrier SBA-15

First, 4 g of P123 was dissolved as a structural directing agent in a beaker containing 120 mL of (2 M) hydrochloric acid and 30 mL of deionized water. The mixture was continuously stirred in a water bath at 40 °C until P123 was completely dissolved. After 4 h, 9.6 g of tetraethyl orthosilicate (TEOS) was added dropwise, acting as the silicon source, with stirring maintained for a further 12 h. Then, 2 g of glucose was slowly added to the mixture, and the resulting solution was continuously stirred at 40 °C for another 12 h. Thereafter, the mixture was transferred into a Teflon-lined reactor and subjected to a 24-h crystallization process at 120 °C. Then, the reactor was removed and cooled to room temperature. The resulting white solid was filtered and washed with deionized water on multiple occasions until the pH of the solution reached a neutral state, thereby removing the residual hydrochloric acid from the mixture. Then, the filtered sample was subjected to an overnight drying process in an oven maintained at 80 °C. Finally, the sample was heated in a muffle furnace at a rate of 3 °C/min to 500 °C and maintained for 6 h. The resulting sample was identified as the carrier SBA-15.

*Preparation of ZrO*₂

 ZrO_2 was prepared as follows: A solution of 2.0 mol/L zirconium oxynitrate was prepared by diluting nitric acid. Then, a solution of 4.0 mol/L sodium hydroxide was slowly added to this solution. The mixture was continuously stirred at room temperature for 30 min. The resulting mixture was filtered and washed with deionized water multiple times until the pH reached approximately 7. Then, the samples were dried at 110 °C for 12 h. Finally, the sample was heated in a muffle furnace at a rate of 15 °C/min to 500 °C and maintained for 6 h, resulting in the formation of ZrO_2 . *Preparation of ZrO_2*/SBA-15

The ZrO₂/SBA-15 was prepared via an equivalent-volume impregnation method. The typical procedure involved the addition of 0.8 g of SBA-15 and 0.2 g of ZrO₂ to a beaker, which was then continuously stirred at 60 °C for 2 h and subsequently at 70 °C for 2 h. Thereafter, the mixture was dried overnight in an oven at 80 °C. Subsequently, the sample was reduced in a muffle furnace at a rate of 2 °C /min to 350 °C and maintained for 5 h under a H₂ flow velocity of 60 mL/min.



Fig. S1 Schematic illustration of the synthesis of CuNi/ZrO₂-SBA-15 catalyst



Fig. S2. Effect of temperature and time on the hydrogenation of HMF, (a) 150 °C, (b) 160 °C, (c) 170 °C, (d) 180 °C, (e) 190 °C, (f) 200 °C. Reaction conditions: 50 mg HMF, 116 μL isopropanol, 25 mg catalyst, 2 MPa H₂.



Fig. S3 XPS spectra of Si 2p for CuNi/ZrO₂-SBA-15 catalyst



Fig. S4 (a) XRD patterns of fresh and spent CuNi/ZrO₂-SBA-15, (b-d) TEM images of spent CuNi/ZrO₂-SBA-15.

Catalyst	Condition	HMF	DMF	Reference
		conversion	selectivity	
				Abhijit D.
Fe-Pd/C	150 °C, 20 bar H ₂ , 2 h	100%	88%	Talpade et al,
				2019
CoNCx/NiFeO	180 °C, 1 MPa H ₂ , 6 h	99.8%	94.3%	Xia et al, 2022
Ni/TMC	180 °C, 3 MPa H ₂ , 3 h	100%	97.6%	Huang et al, 2025
Co ₂ Ni ₁ @N ₁ C-	150 °C, 1 MPa H ₂ , 3 h	100%	98.9%	Liao et al, 2025
700				
CuNi/ZrO ₂ -	180 °C, 2 MPa H ₂ , 3 h	100%	100%	This work
SBA-15				

Table S1 Comparasion of HMF conversion and DMF selectivity of catalyst

Reference

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