

Electronic Supplementary Material (ESI)

Metallo-Hydrazones of Pyridine–Thiophene Derivatives: Structure–Function Insight and Anticancer Evaluation of Co(III), Ni(II), Cu(II), and Zn(II) Complexes

Ram Nayan Gautam^a, Alok Shukla^b, Ankesh Kumar Jaiswal^b, Anupam Singh^a, Ajay Vishwakarma^a, Rajesh Kumar^c, L.B.

Prasad^a, R. J. Butcher^d, M. K. Bharty^{a*}

^aDepartment of Chemistry, Banaras Hindu University, Varanasi-221005, India.

^bDepartment of Biosciences and Bioengineering, IIT Bombay-400076, India.

^cDepartment of Pharmacology, Maharaja Agrasen School of Pharmacy, Maharaja Agrasen University, Solan, Himachal Pradesh – 174103, India.

^dDepartment of Chemistry, Howard University, 525 College Street NW, Washington, DC 20059, USA.

*Corresponding author E-Mail: manoj_vns2005@yahoo.co.in; mkbharty@bhu.ac.in

Contents

1. IR spectra.....	S2-S3
2. NMR spectra.....	S4
3. Mass Spectrometry.....	S5-S7
4. Crystallographic Appendix.....	S7-S13
5. Molecular Docking.....	S14-S20

1. I.R. Spectra

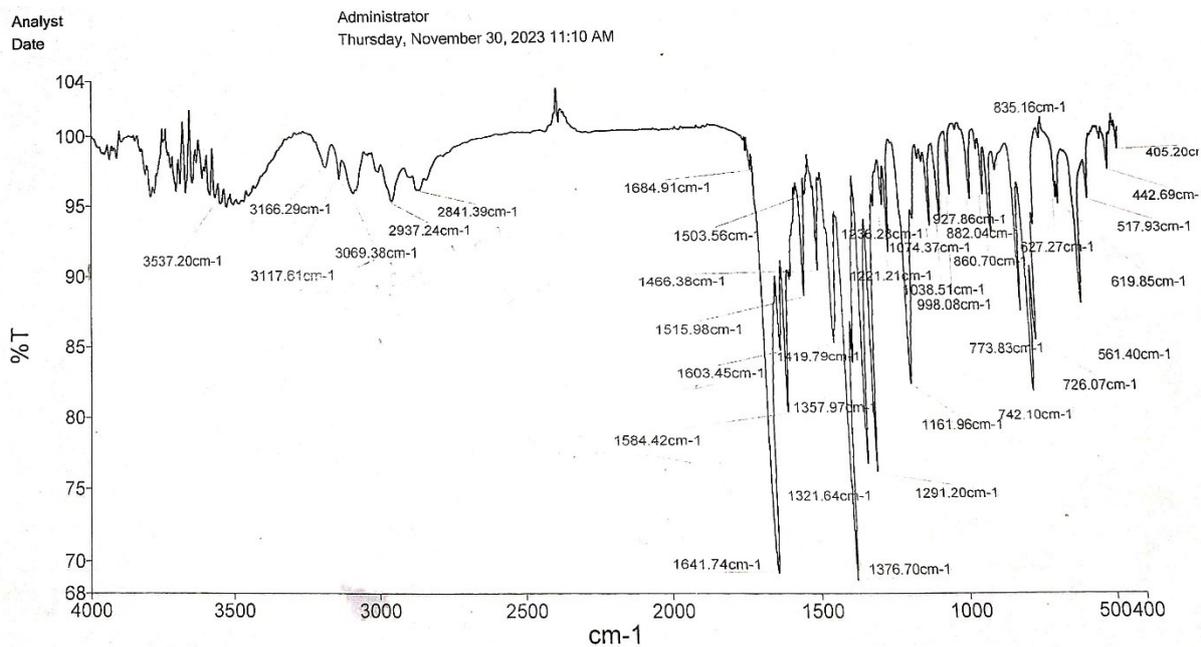


Fig. S1. IR spectrum of ligand Hptch.

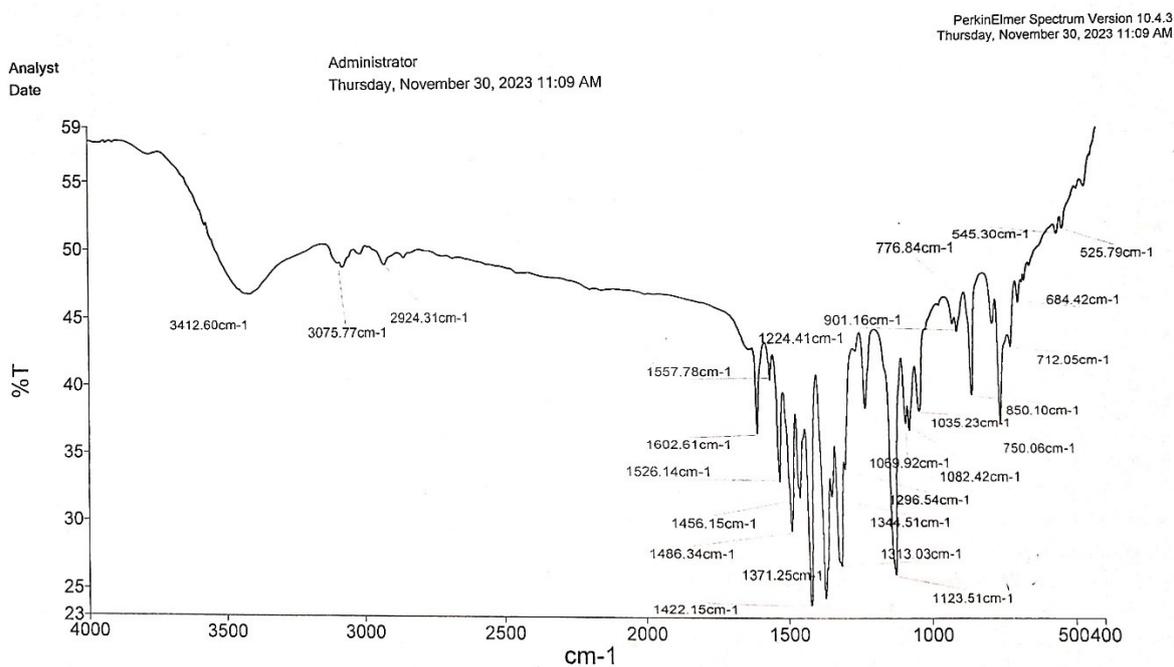


Fig. S2. IR spectrum of complex 1.

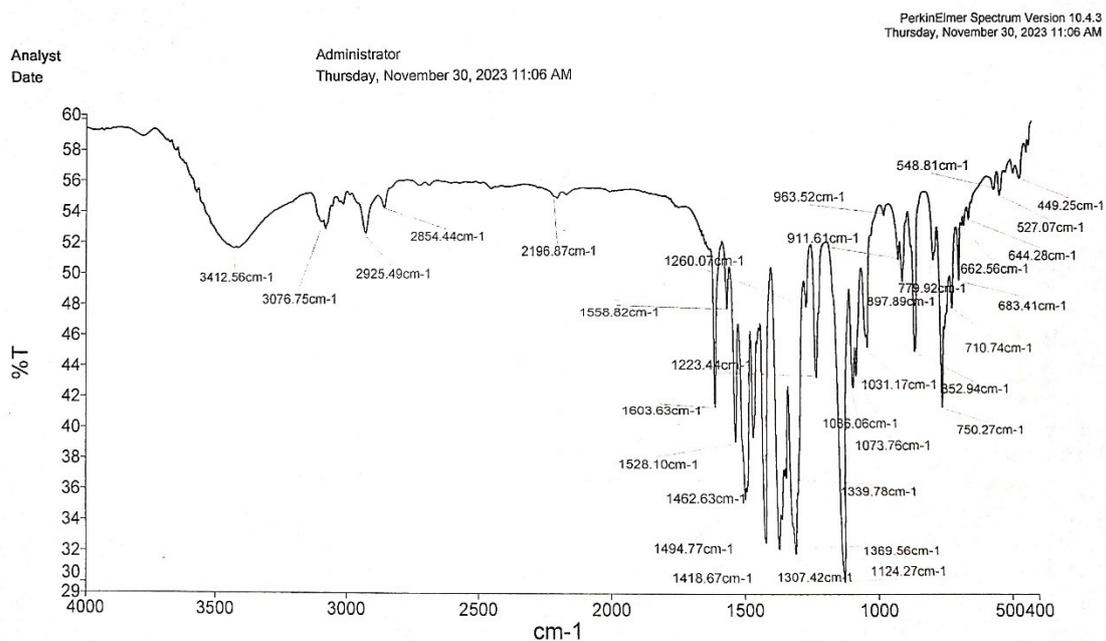


Fig. S3. IR spectrum of complex 2.

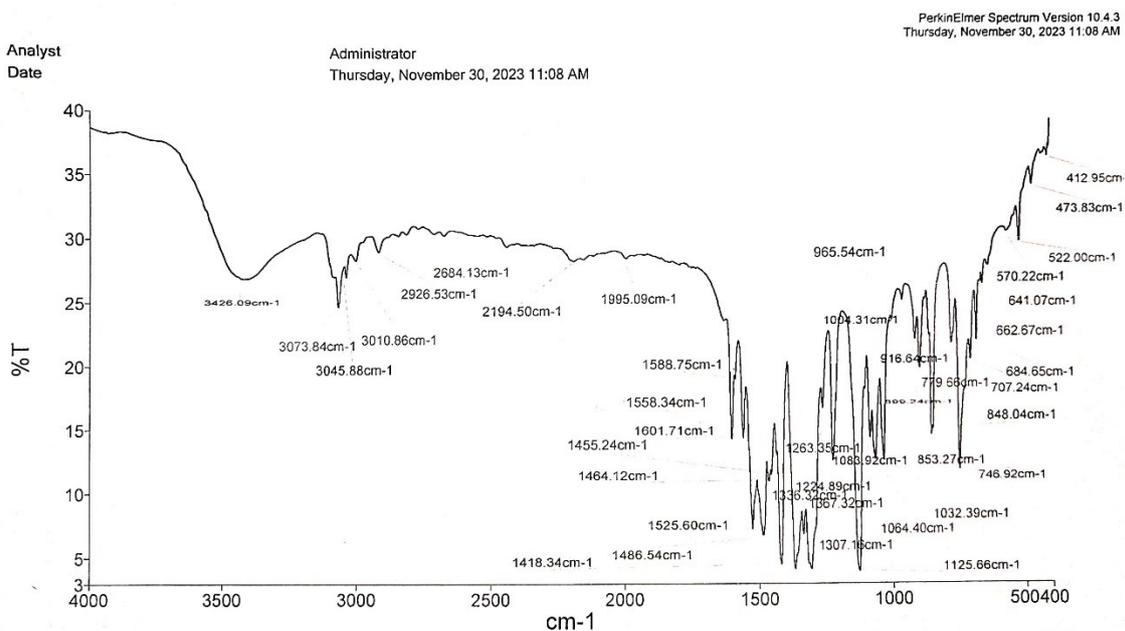


Fig. S4. IR spectrum of complex 3.

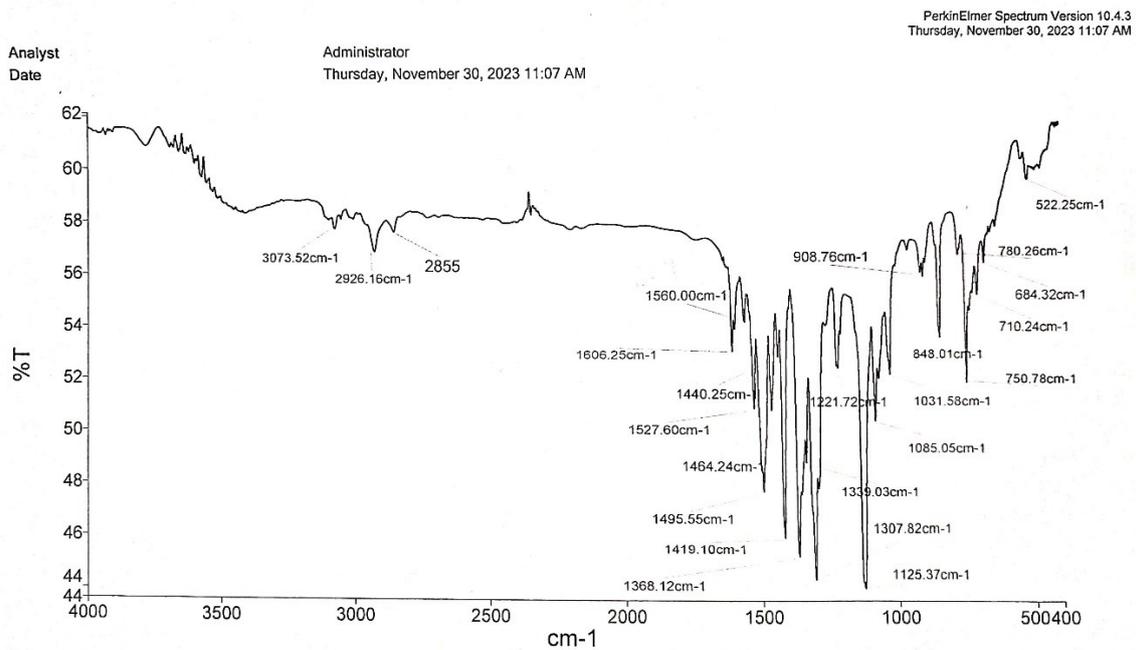


Fig. S5. IR spectrum of complex 4.

2. NMR spectra

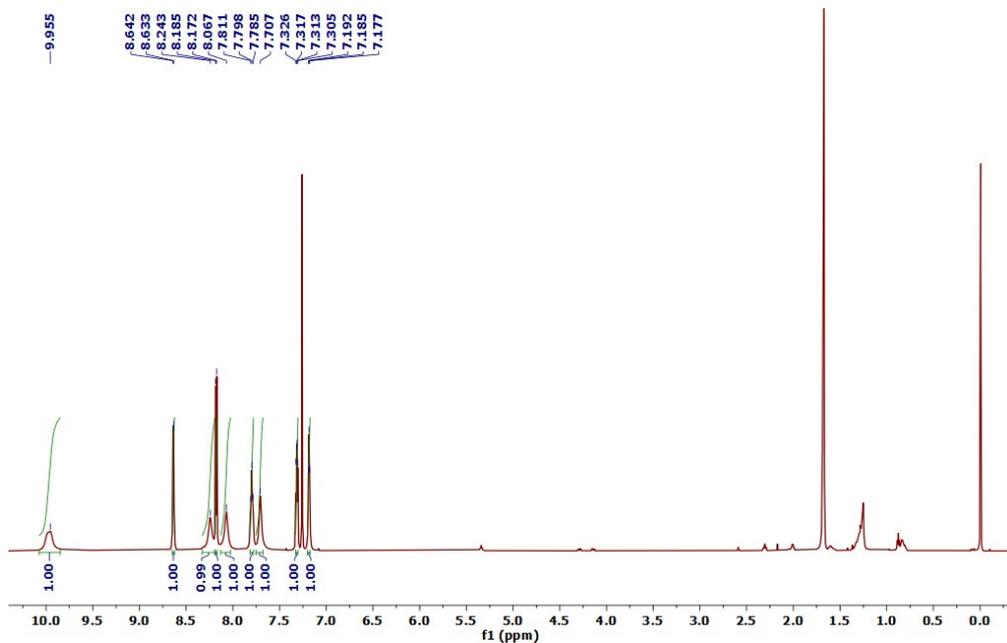


Fig. S6. ¹H NMR spectrum of ligand Hptch.

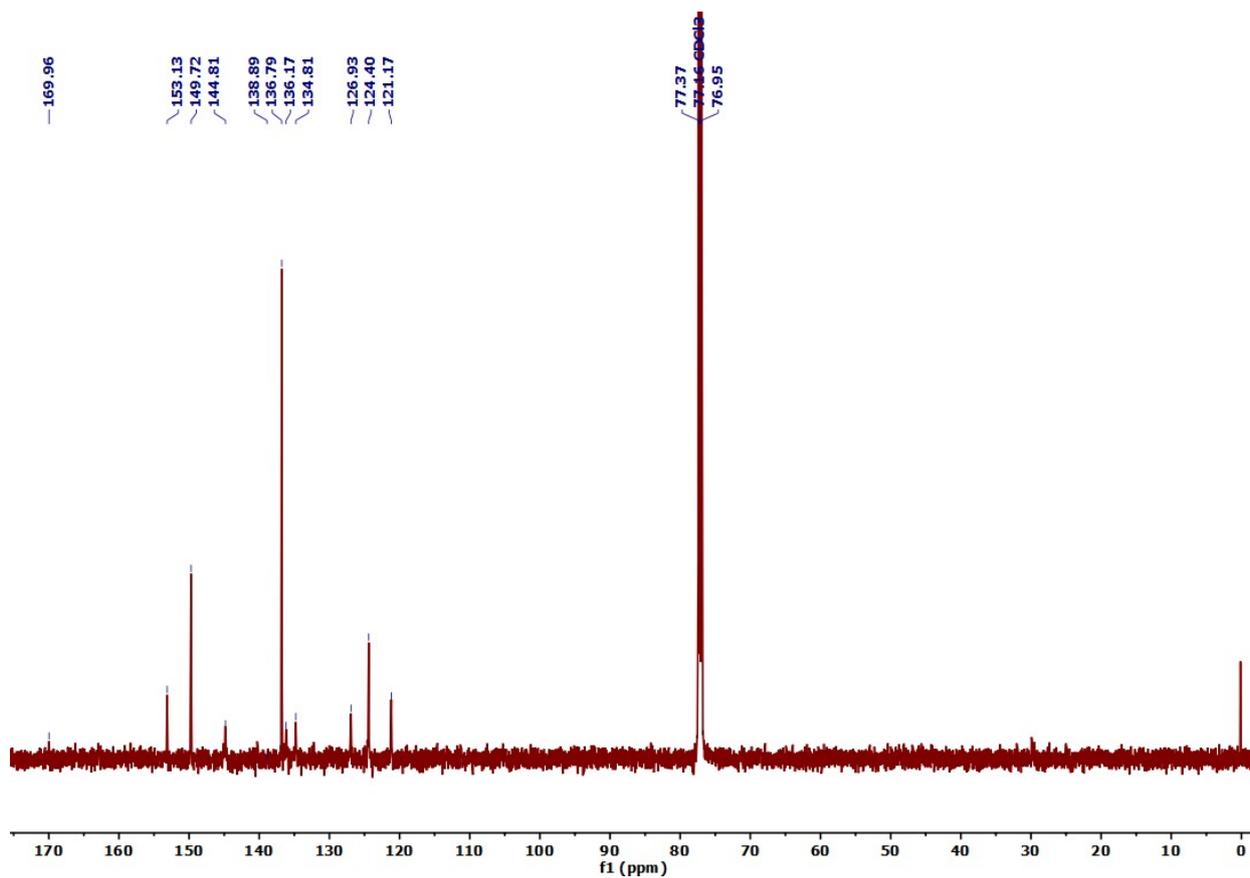


Fig. S7. ^{13}C NMR spectrum of ligand **Hptch**.

3. Mass Spectrometry

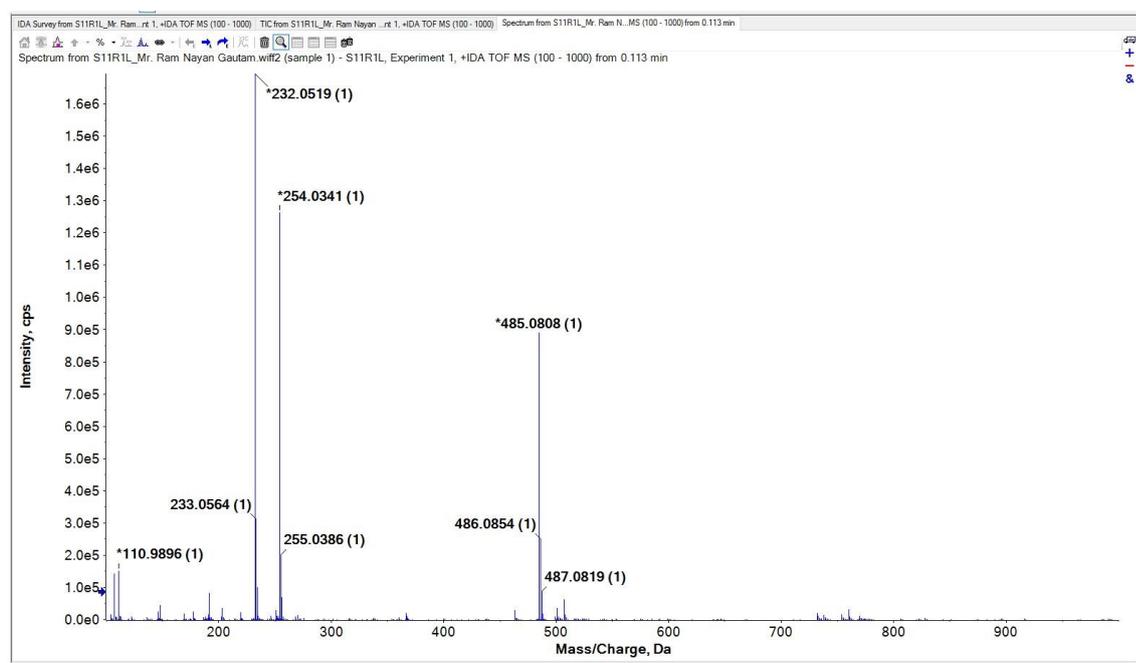


Fig. S8. HR-MS spectrum of ligand **Hptch**.

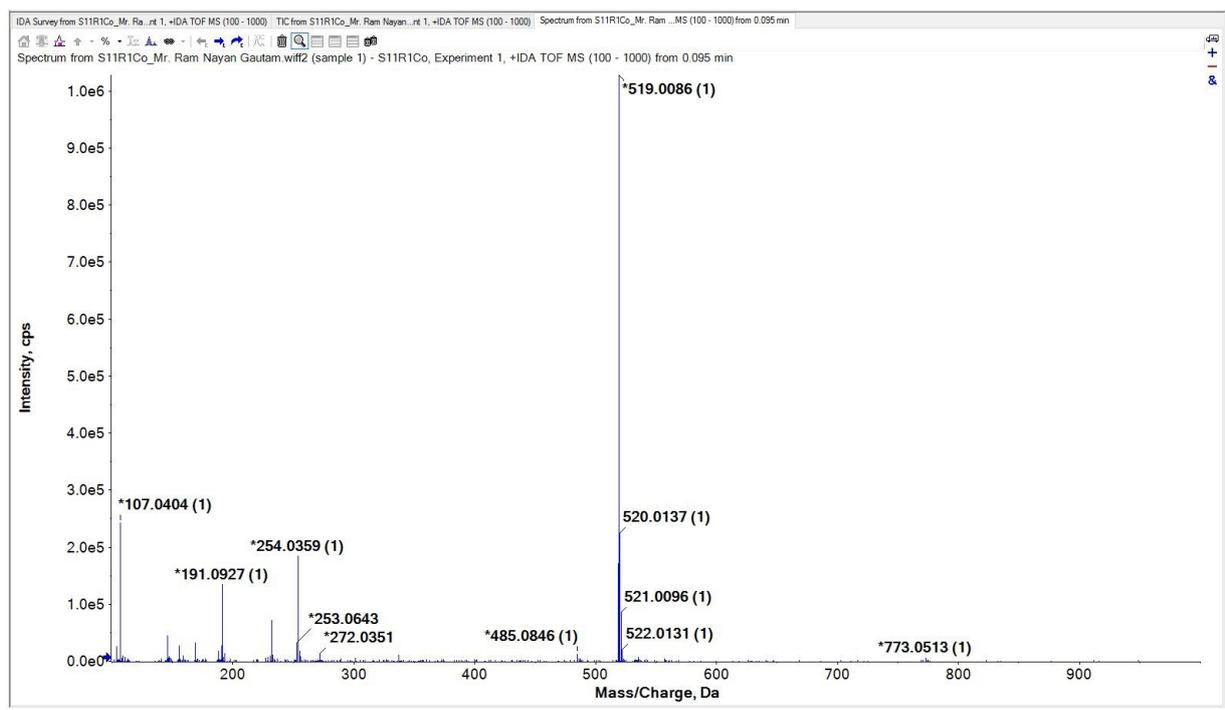


Fig. S9. HR-MS spectrum of complex 1.

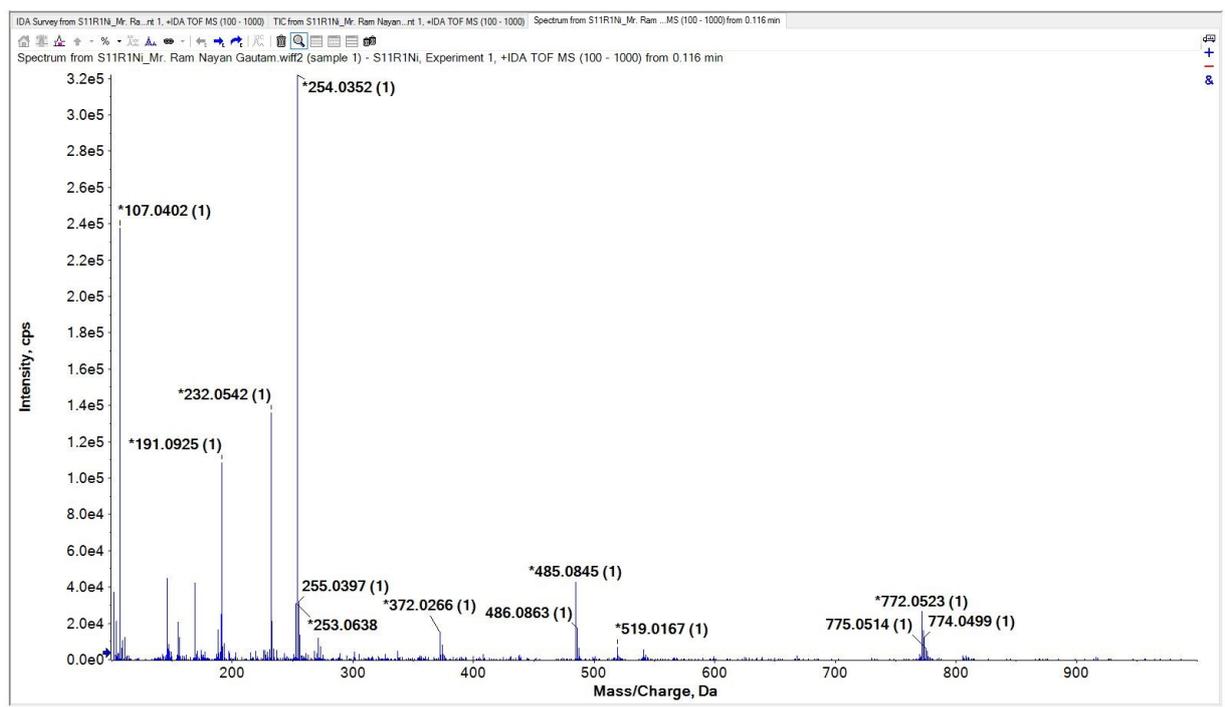


Fig. S10. HR-MS spectrum of complex 2.

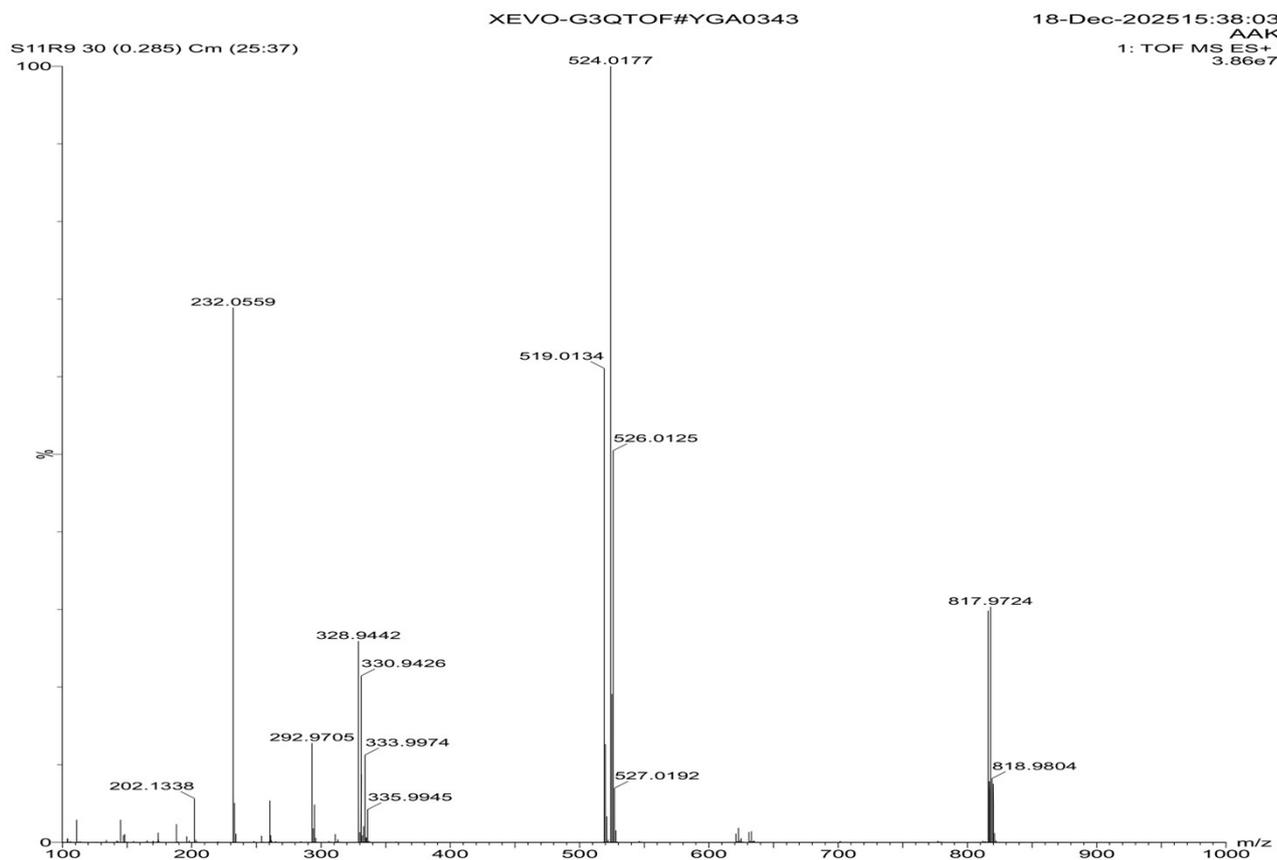


Fig. S11. HR-MS spectrum of complex 3.

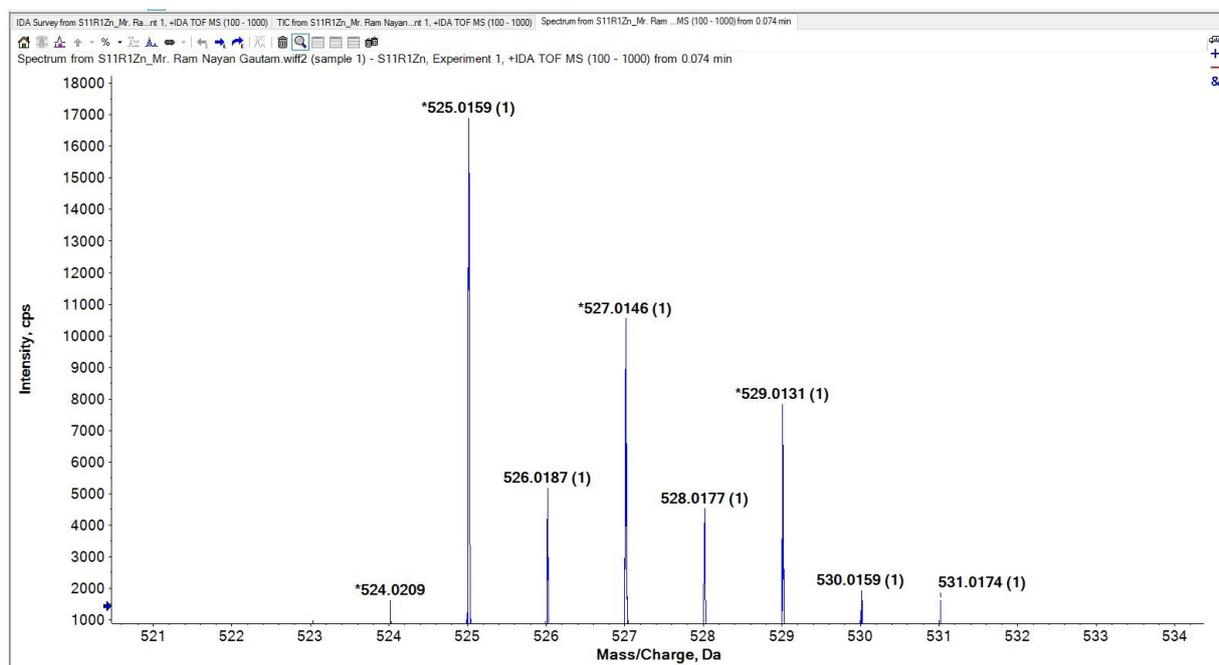


Fig. S12. HR-MS spectrum of complex 4.

4. Crystallographic data

Table S1. Bond lengths [Å] and angles [°] for **Hptch**

Bond lengths [Å]		Bond angles [°]	
S(1)-C(5)	1.699(4)	C(5)-S(1)-C(2)	91.5(2)
S(1)-C(2)	1.726(4)	C(6)-N(2)-N(1)	115.6(3)
N(2)-C(6)	1.271(5)	C(1)-N(1)-N(2)	122.7(3)
N(2)-N(1)	1.374(4)	C(1)-N(1)-H(1)	122(3)
O(1)-C(1)	1.220(4)	N(2)-N(1)-H(1)	115(3)
N(1)-C(1)	1.351(5)	C(7)-N(3)-C(11)	117.9(3)
N(3)-C(7)	1.342(5)	O(1)-C(1)-N(1)	119.4(3)
N(3)-C(11)	1.343(6)	O(1)-C(1)-C(2)	120.4(4)

Table S2. Hydrogen bonds for **Hptch** [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(5)-H(5)...O(1)#1	0.93	2.54	3.161(6)	124.5
N(1)-H(1)...N(3)#2	0.91(4)	2.13(4)	3.037(5)	177(4)

Symmetry transformations used to generate equivalent atoms: #1 -x+3/2,y+1/2,z+1/2 #2 -x+1,-y+1,z-1/2

Table S3. Bond lengths [Å] and angles [°] for **Complex 1**

Bond lengths [Å]		Bond angles [°]	
Co(1)-O(1)	1.916(4)	N(2)-Co(1)-N(5)	176.3(3)
Co(1)-N(2)	1.842(6)	N(5)-Co(1)-O(2)	82.0(2)
Co(1)-N(1)	1.943(6)	N(5)-Co(1)-N(4)	82.2(3)
Co(1)-O(2)	1.925(5)	N(5)-Co(1)-O(1)	96.3(2)
Co(1)-N(5)	1.856(6)	N(2)-Co(1)-N(1)	82.6(3)
Co(1)-N(4)	1.943(6)	O(1)-Co(1)-N(4)	90.4(2)
N(3)-C(7)	1.326(9)	O(2)-Co(1)-N(1)	91.4(2)
O(1)-C(7)	1.290(8)	N(2)-Co(1)-N(4)	100.9(3)
N(2)-N(3)	1.373(9)	N(1)-Co(1)-N(4)	91.5(3)

Table S4. Bond lengths [Å] and angles [°] for **Complex 2**

Bond lengths [Å]		Bond angles [°]	
Ni(1)-O(1)	2.113(3)	N(2)-Ni(1)-N(5)	175.02(13)
Ni(1)-N(2)	1.999(3)	N(5)-Ni(1)-O(2)	76.52(11)
Ni(1)-N(1)	2.105(3)	N(5)-Ni(1)-N(4)	78.16(11)
Ni(1)-O(2)	2.093(3)	N(2)-Ni(1)-O(1)	76.28(11)
Ni(1)-N(5)	1.993(3)	N(2)-Ni(1)-N(1)	78.32(12)
Ni(1)-N(4)	2.131(3)	O(1)-Ni(1)-N(4)	92.48(11)
N(3)-C(7)	1.339(5)	O(2)-Ni(1)-O(1)	93.46(11)
O1-C7	1.268(4)	C(7)-O(1)-Ni(1)	109.7(2)
N2-N3	1.368(4)	C(6)-N(2)-Ni(1)	118.3(3)

Table S5. Bond lengths [Å] and angles [°] for **Complex 3**

Bond lengths [Å]		Bond angles [°]	
Cu(01)-O(1)	2.195(2)	N(2) -Cu(01)- N(5)	177.91(12)
Cu(01)-N(2)	1.962(3)	N(5)-Cu(01)-O(2)	77.17(11)
Cu(01)-N(3)	2.319(3)	N(5)-Cu(01)-N(6)	78.53(11)
Cu(01)-O(2)	2.111(2)	N(2)-Cu(01)-O(1)	76.02(10)
Cu(01)-N(5)	1.945(3)	N(2)-Cu(01)-N(3)	76.20(11)
Cu(01)-N(6)	2.150(3)	N(6)-Cu(01)-O(1)	90.13(10)
N(1)-C(1)	1.337(4)	O(2)-Cu(01)-N(3)	93.02(10)
C(1)-O(1)	1.257(4)	N(6)-Cu(01)-N(3)	92.99(11)
N(1)-N(2)	1.380 (4)	O(1)-Cu(01)-N(3)	151.99(10)

Table S6. Bond lengths [Å] and angles [°] for **Complex 4**

Bond lengths [Å]		Bond angles [°]	
Zn(01)-O(1A)	2.148(4)	N(2A)-Zn(01)-N(2B)	172.74(17)
Zn(01)-N(2A)	2.071(4)	N(2B)-Zn(01)-O(1B)	74.39(14)
Zn(01)-N(3A)	2.233(4)	N(2B)-Zn(01)-N(3B)	74.92(15)
Zn(01)-O(1B)	2.127(4)	N(2A)-Zn(01)-O(1A)	74.36(15)
Zn(01)-N(2B)	2.078(4)	N(2A)-Zn(01)-N(3A)	75.39(16)
Zn(01)-N(3B)	2.199(4)	O(1A)-Zn(01)-N(3B)	93.80(16)
N(1A)-C(5A)	1.338(7)	O(1B)-Zn(01)-N(3A)	93.91(15)
O(1A)-C(5A)	1.251(6)	N(2B)-Zn(01)-N(3A)	110.38(16)
N(2A)-N(1A)	1.358(6)	N(3B)-Zn(01)-N(3A)	92.27(15)

Table S7. Hydrogen bonds for **Complex 1** [\AA and $^\circ$].

D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	\angle (DHA)
C(1)-H(1A) \cdots Cl(2)#1	0.93	2.96	3.538(11)	121.4
C(2)-H(2A) \cdots Cl(2)#1	0.93	2.74	3.428(14)	131.5
C(11)-H(11A) \cdots Cl(1)#2	0.93	2.84	3.641(8)	144.5
C(13)-H(13A) \cdots Cl(1)#3	0.93	2.80	3.459(13)	128.7
C(15)-H(15A) \cdots O(1W)#4	0.93	2.60	3.42(3)	147.5
C(21A \wedge b)-H(21B \wedge b) \cdots S(1)#5	0.93	2.97	3.618(14)	128.4
O(1W)-H(1W2) \cdots Cl(2)	0.82(2)	2.48(7)	3.18(2)	144(10)

Symmetry transformations used to generate equivalent atoms: #1 $-x+1, y, -z+1/2$ #2 $x+1/2, -y+1/2, z+1/2$

#3 $-x+3/2, y+1/2, -z+1/2$ #4 $-x+2, y, -z+1/2$ #5 $x-1, y, z$

Table S8. Hydrogen bonds for **Complex 2** [\AA and $^\circ$].

D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	\angle (DHA)
C(1)-H(1A) \cdots N(6)#1	0.93	2.57	3.249(5)	130.0
C(10 \wedge a)-H(10A \wedge a) \cdots O(2)#2	0.93	2.52	3.256(8)	136.2
C(11 \wedge a)-H(11A \wedge a) \cdots S(1 \wedge a)#3	0.93	2.75	3.605(7)	152.5
C(13)-H(13A) \cdots N(3)#4	0.93	2.65	3.538(5)	161.1
C(17)-H(17A) \cdots O(1)#5	0.93	2.63	3.361(4)	135.9
C(21 \wedge a)-H(21A \wedge a) \cdots O(2)#6	0.93	2.47	3.353(9)	159.4

Symmetry transformations used to generate equivalent atoms: #1 $-x+1, -y, -z+1$ #2 $x, y+1, z$ #3 $-x+2, y+1/2, -$

$z+3/2$ #4 $-x+2, -y+1, -z+1$ #5 $-x+1, -y+1, -z+1$ #6 $-x+1, y-1/2, -z+3/2$

Table S9. Hydrogen bonds for **Complex 3** auto [\AA and $^\circ$].

D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	\angle (DHA)
C(3)-H(3) \cdots Cl(2)#1	0.93	2.95	3.732(3)	142.4
C(14)-H(14) \cdots Cl(3)#2	0.93	2.92	3.676(4)	138.8
C(19)-H(19) \cdots S(1)#3	0.93	2.97	3.747(4)	142.7
C(8)-H(8) \cdots O(2)#4	0.93	2.49	3.281(4)	142.6
C(5)-H(5) \cdots O(1)#5	0.93	2.35	3.279(5)	177.7
C(23)-H(23) \cdots S(2)	0.93	2.99	3.780(4)	138.1
C(23)-H(23) \cdots N(4)	0.93	2.27	3.161(5)	150.6
C(10)-H(10) \cdots Cl(3)#6	0.93	2.91	3.720(4)	145.8

Symmetry transformations used to generate equivalent atoms: #1 $-x+3/2, y+1/2, -z+1/2$ #2 $-x+1, -y, -z+1$ #3 $-x+3/2, y-1/2, -z+1/2$ #4 $-x+1, -y+1, -z+1$ #5 $-x+1/2, y+1/2, -z+1/2$ #6 $-x+2, -y, -z+1$

Table S10. Hydrogen bonds for **Complex 4** auto [\AA and $^\circ$].

D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	\angle (DHA)
C(11B)-H(11B) \cdots N(1A)#1	0.93	2.57	3.289(7)	134.1
C(1B)-H(1B) \cdots S(1B)#2	0.93	2.79	3.633(6)	151.8
C(2A)-H(2A) \cdots O(1A)#3	0.93	2.51	3.400(8)	160.0

Symmetry transformations used to generate equivalent atoms: #1 $-x+1, -y, -z+1$ #2 $-x, y+1/2, -z+1/2$ #3 $-x+1, y-1/2, -z+1/2$

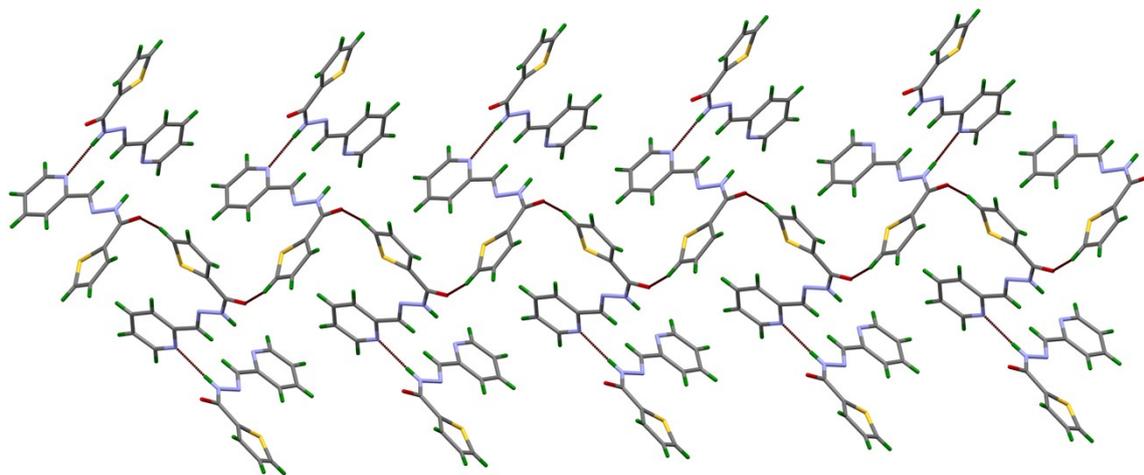


Fig. S13. Showing weak C-H \cdots O intermolecular interactions leading to a wave-like architecture in the **Hptch**.

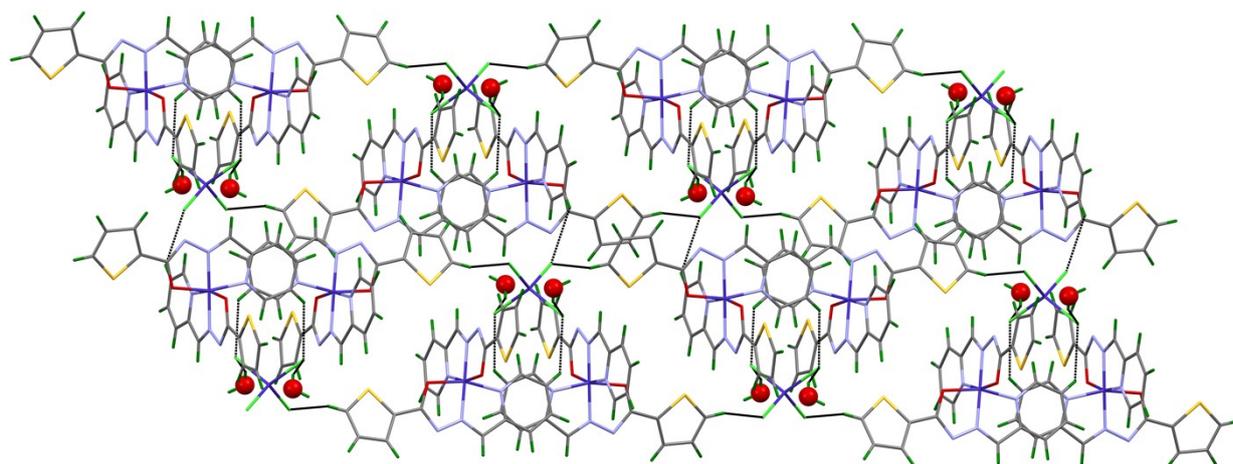


Fig. S14. Showing C-H \cdots Cl interactions in **Complex 1** leading to a paddle wheel-like architecture (Oxygen atoms are presented in spacefill style)

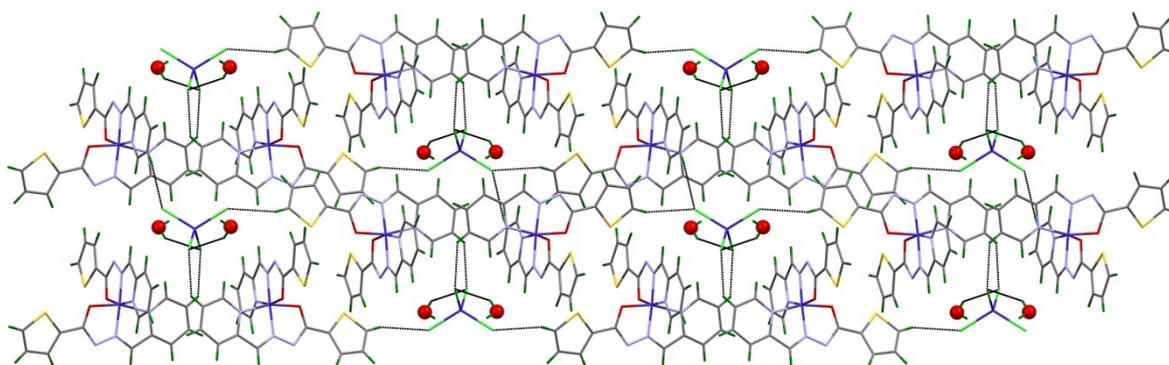


Fig. S15. Showing O-H \cdots Cl and C-H \cdots Cl interactions in Complex **1** leading to a supramolecular architecture.

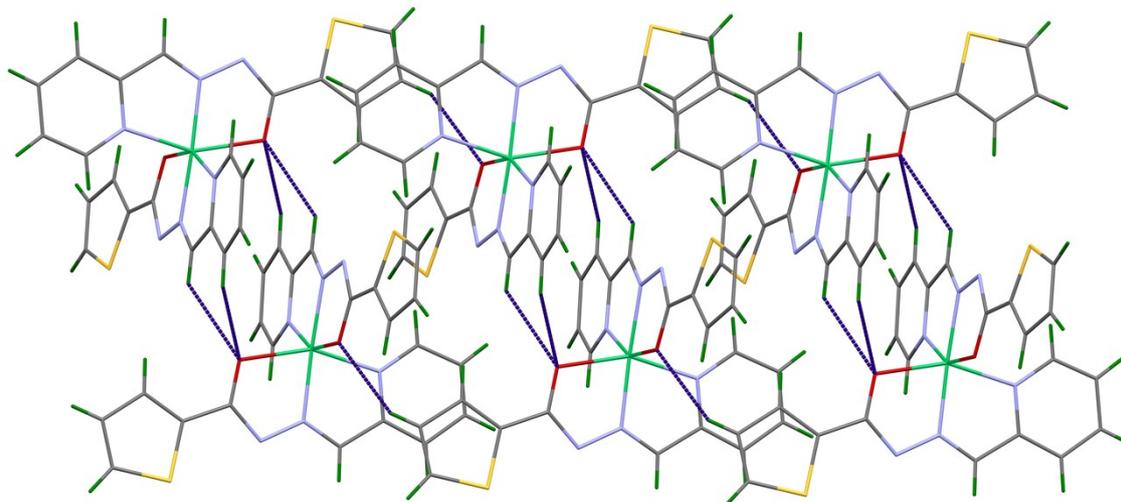


Fig. S16. Showing CH \cdots O interactions in Complex **2**

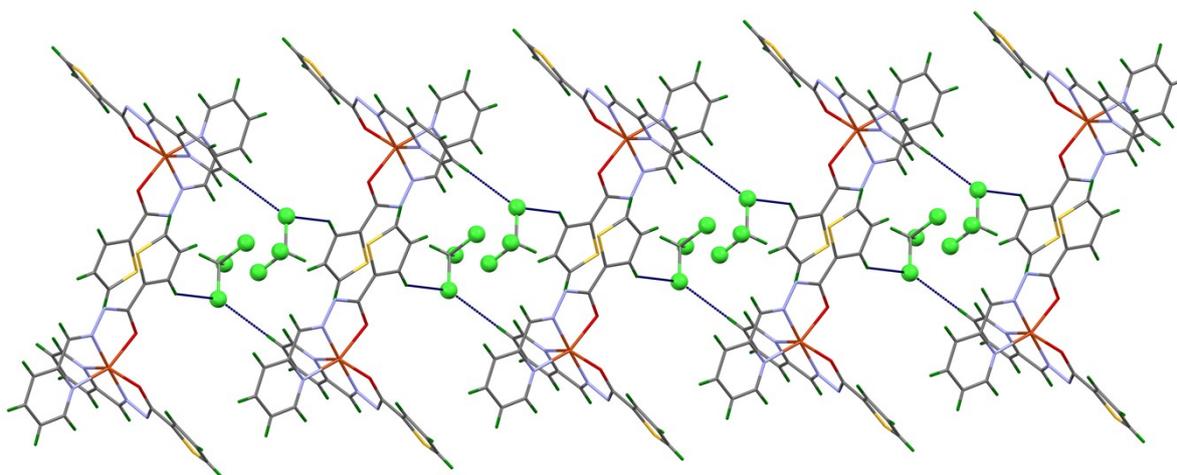


Fig. S17. Showing CH \cdots Cl interactions in Complex **3** leading to a linear chain structure.

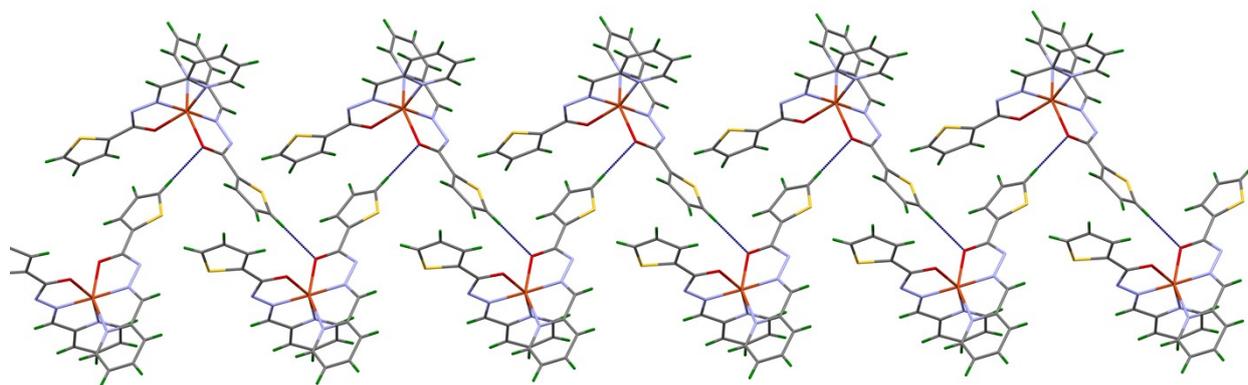


Fig. S18. Showing CH \cdots O interactions in Complex **3** leading to a linear chain structure.

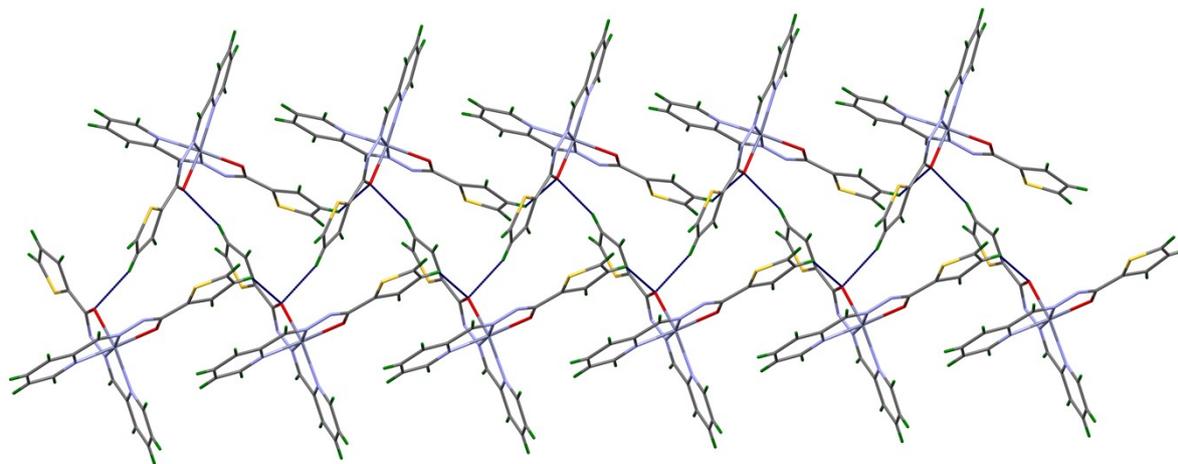


Fig. S19. Showing CH \cdots O interactions in Complex **4** leading to a wave-like architecture.

5. Molecular Docking

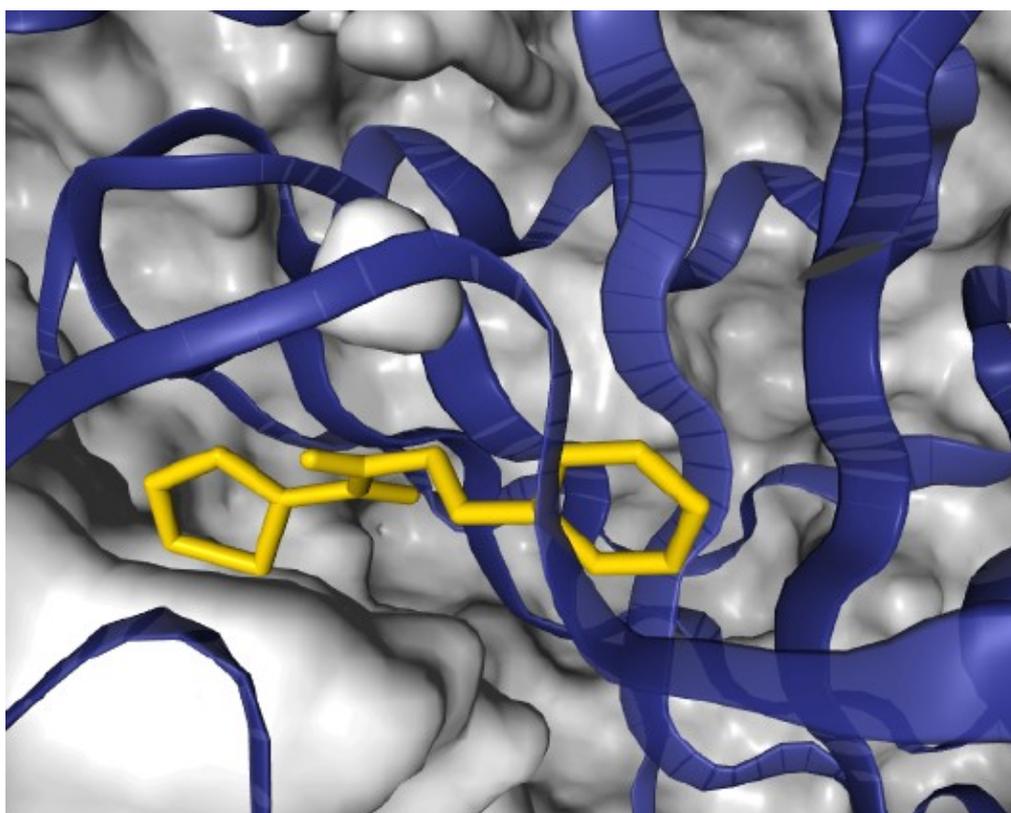


Fig. S20. Protein interaction of ligand **Hptch**.

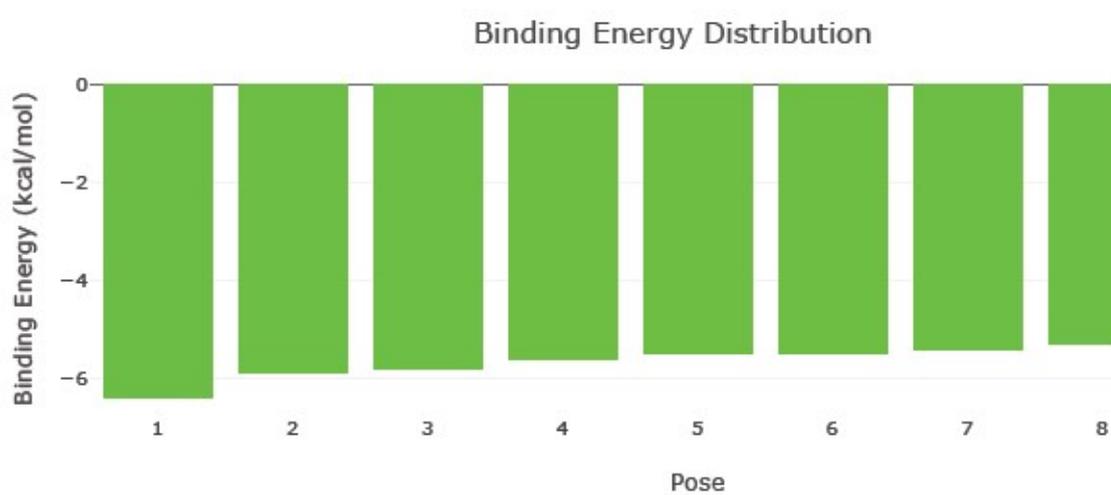


Fig. S21. Binding energy distribution of ligand **Hptch**.

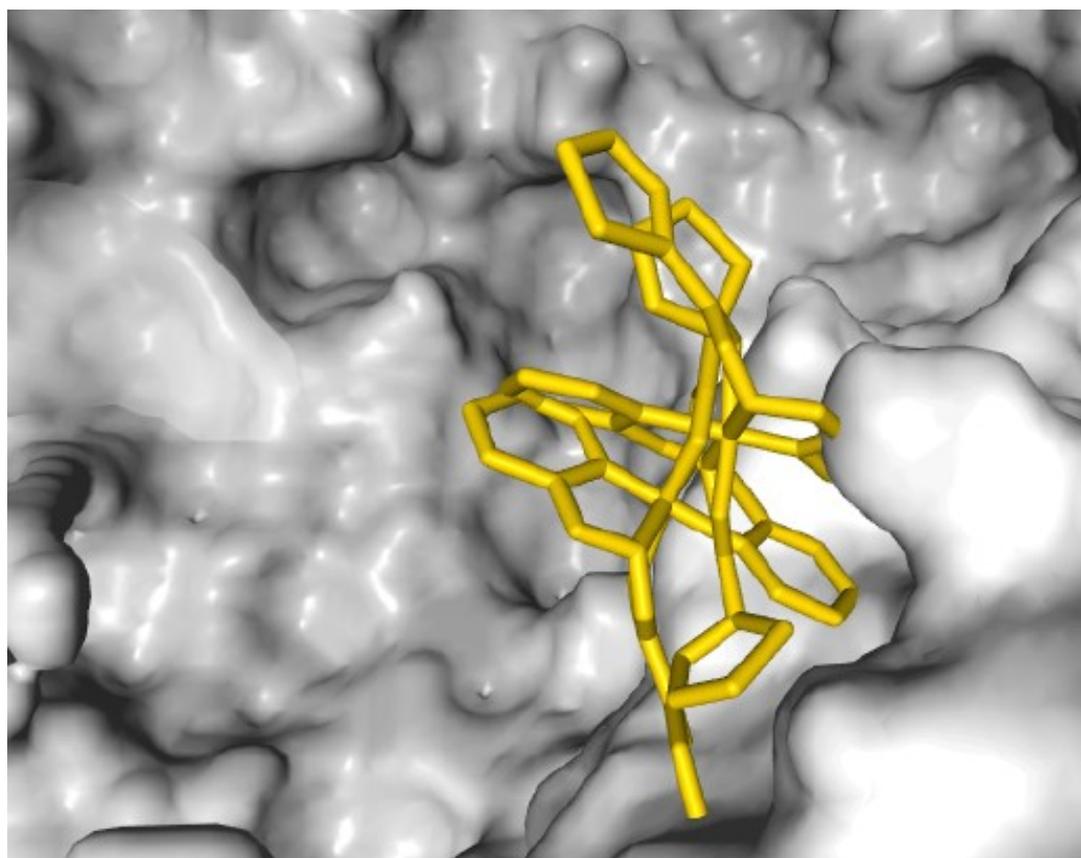


Fig. S22. Protein interaction of complex 1.

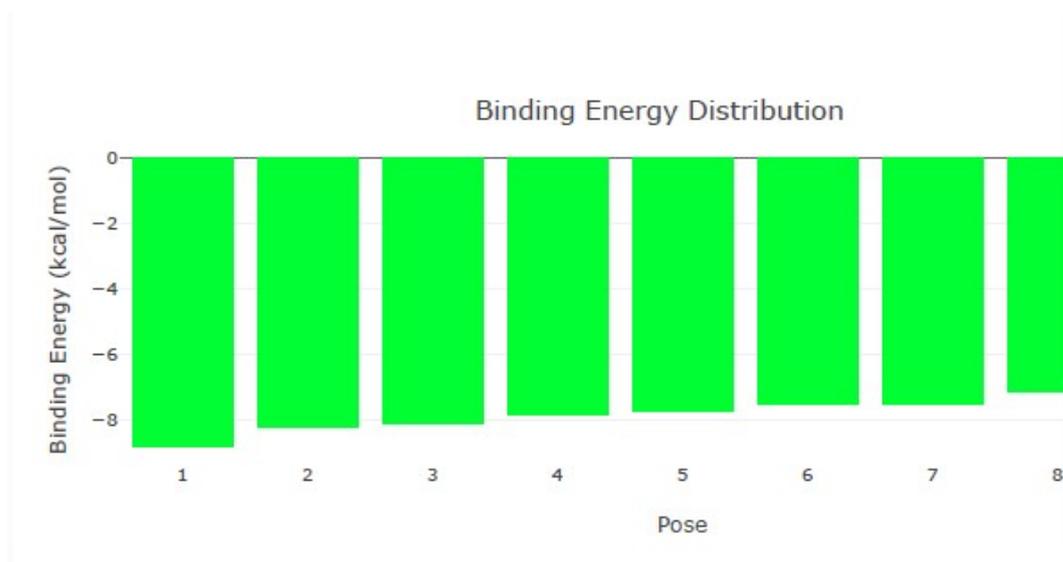


Fig. S23. Binding energy distribution of complex 1.

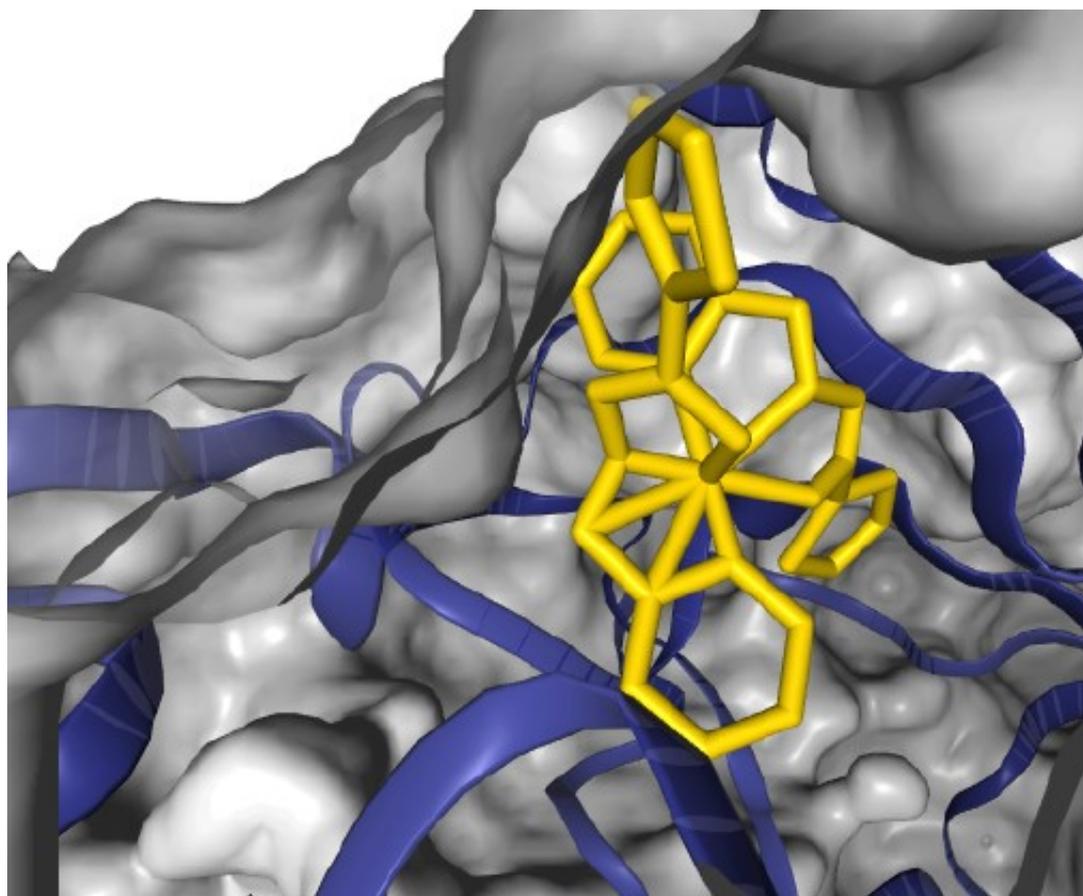


Fig. S24. Protein interaction of complex 2.

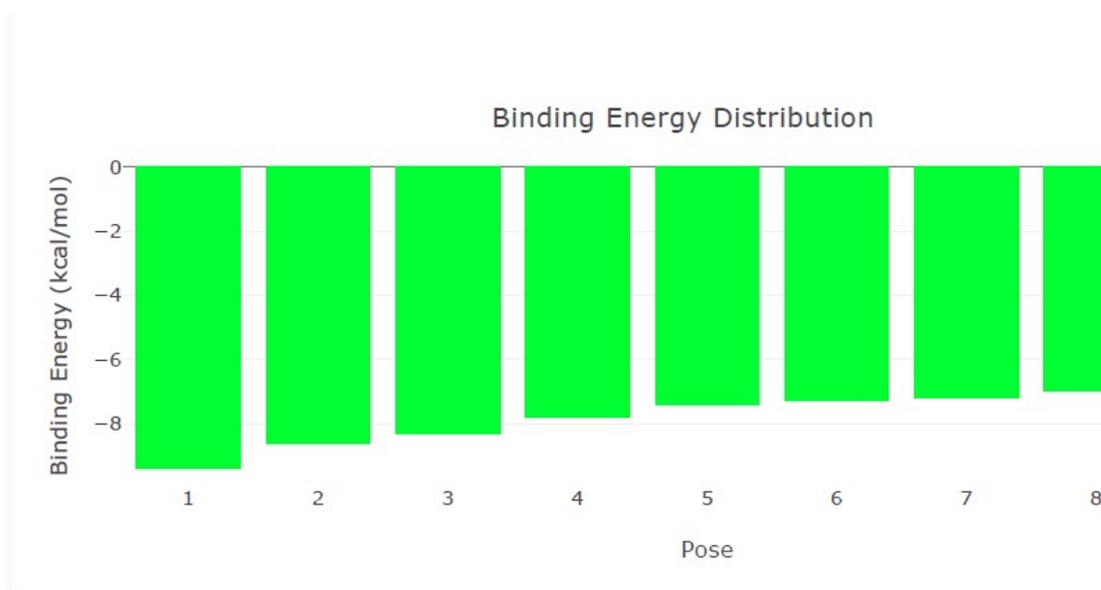


Fig. S25. Binding energy distribution of complex 2.

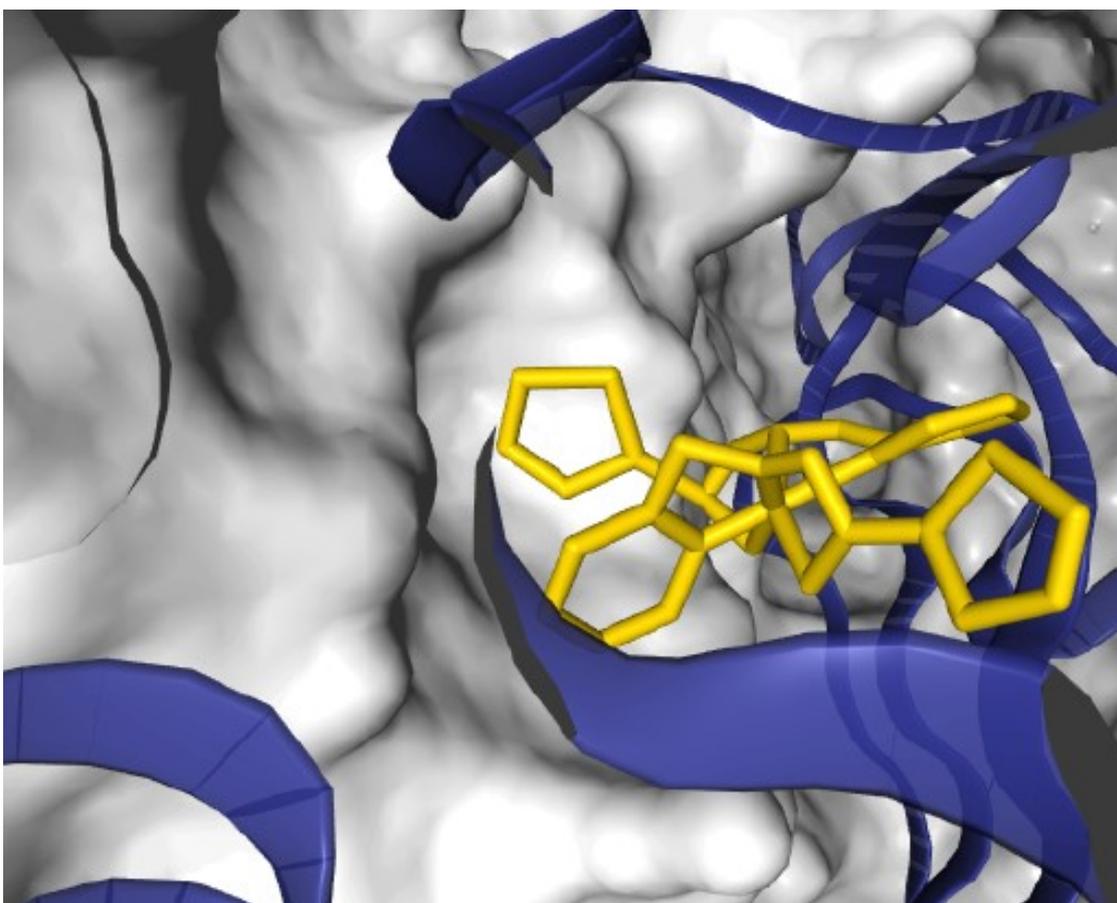


Fig. S26. Protein interaction of complex 3.

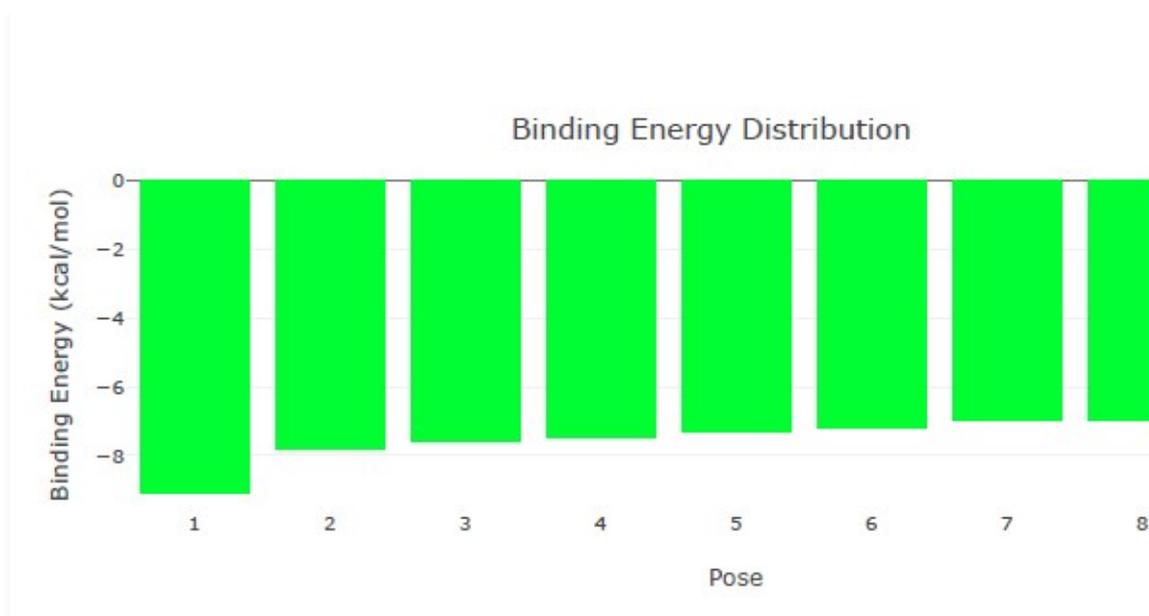


Fig. S27. Binding energy distribution of complex 3.

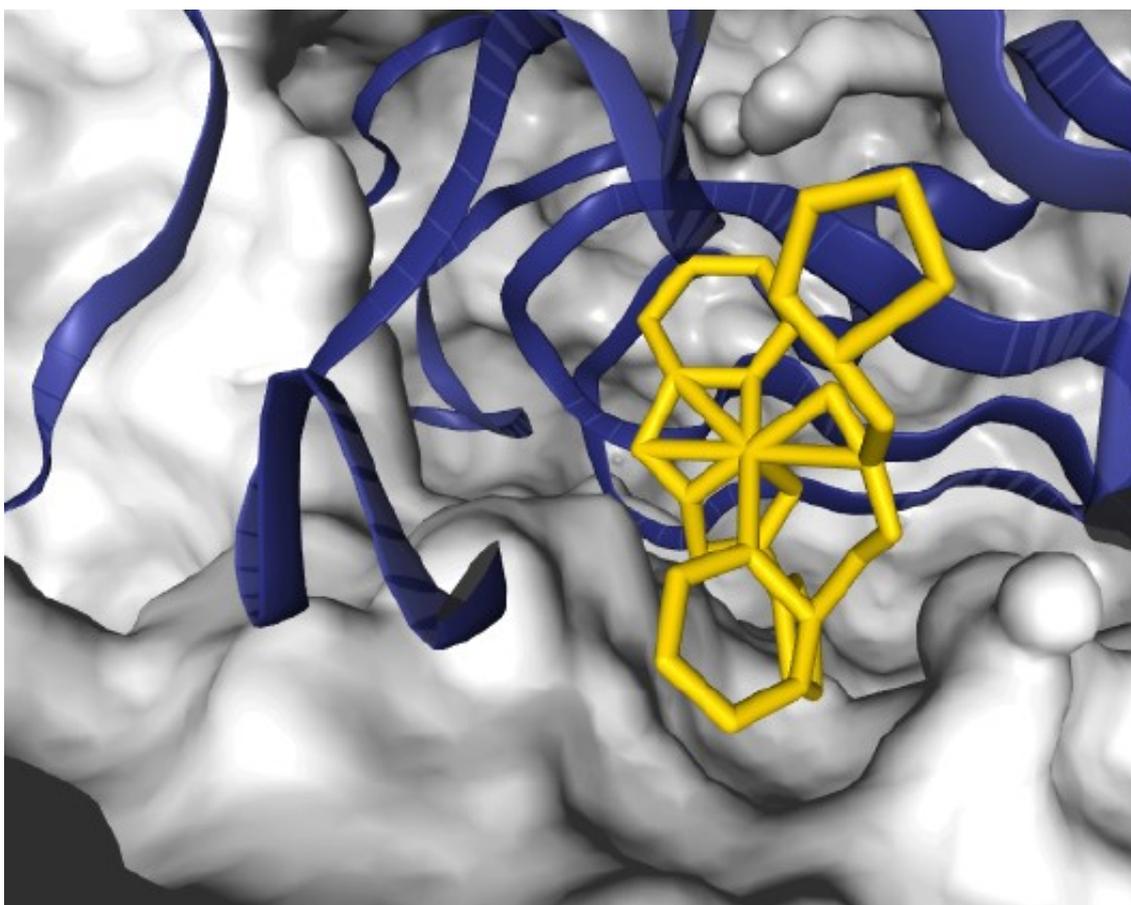


Fig. S28. Protein interaction of complex 4.

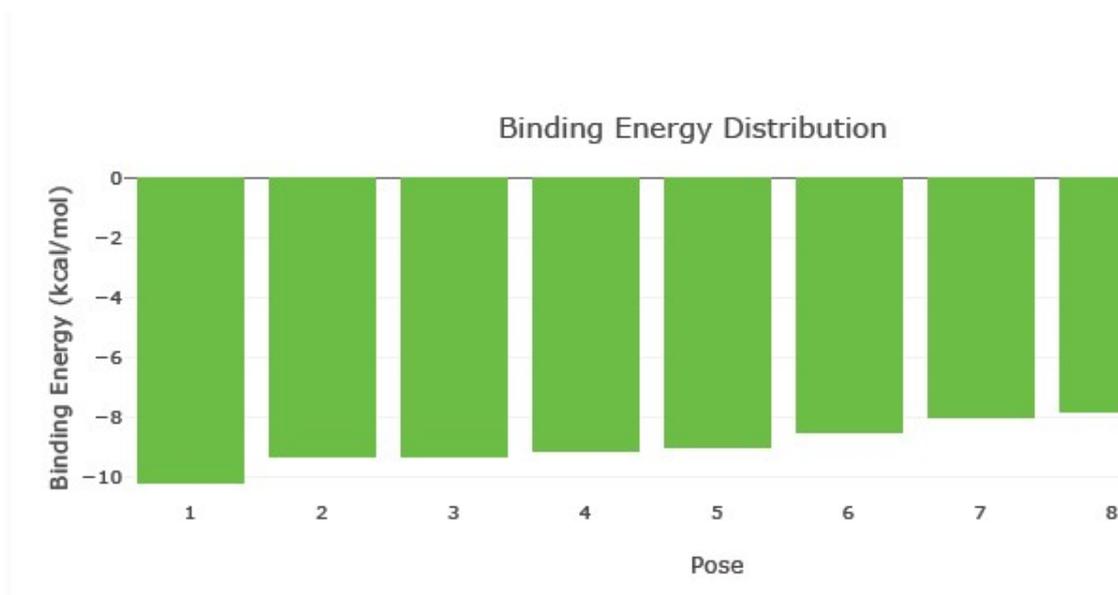
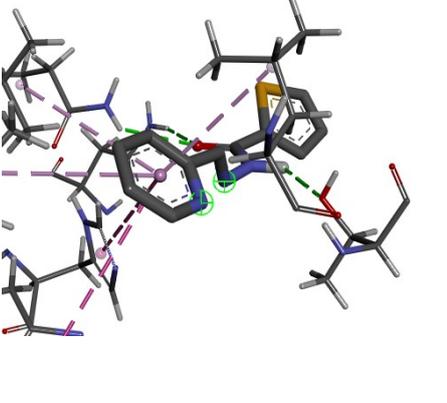
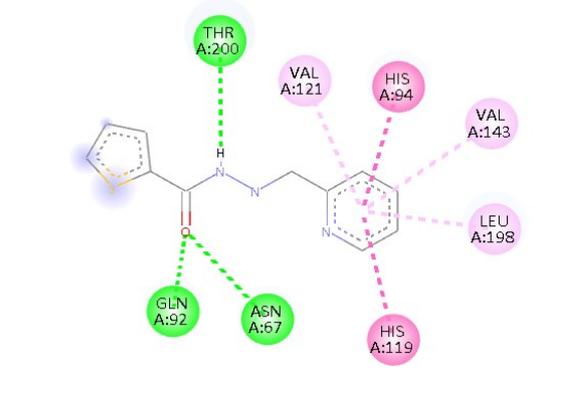
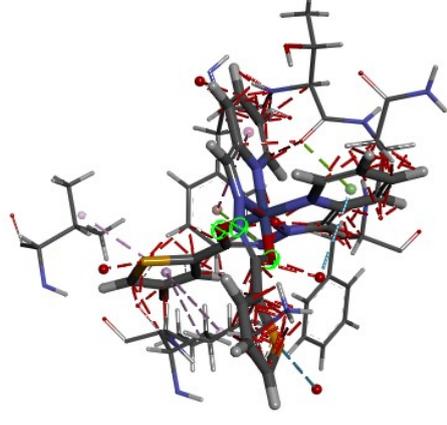
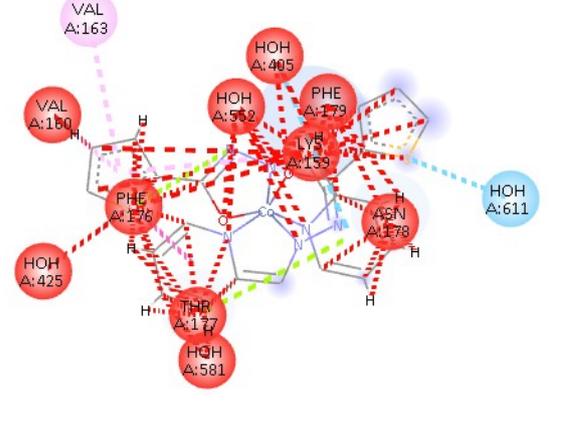


Fig. S29. Binding energy distribution of complex 4.

Table S11: Receptor ligand interaction on a 2D diagrams.

Compound	Receptor-ligand interaction	Receptor-ligand interaction in a 2D-diagram
Hptch		
Complex 1		
Complex 2	