

Supporting Information

Table Captions:

Table. S1 Comparison of CO production rate, CO selectivity for R-0.25H₂ catalysts and literature reported catalysts.

Table. S1 Comparison of CO production rate, CO selectivity for R-0.25H₂ catalysts and literature reported catalysts.

catalyst	H ₂ : CO ₂ ratio	WHSV (mL/g/h)	Temperature (°C)	Rate (mmol _{CO} ·g _{cat} ⁻¹ ·h ⁻¹)	CO selectivity (%)	Ref.
R-0.25H ₂	4:1	60000	350	50.2	100	This work
			400	77.3	100	
			450	104.2	100	
			500	133.7	100	
			550	144.0	100	
			600	163.4	100	
CuZnAl	3:1	60000	400	72.5	100	[1]
			500	104.5	100	
			350	28.9	100	
Cu/CeO ₂	4:1	30000	400	45.9	100	[2]
			450	58.3	100	
			500	68.3	100	
			550	75.0	100	
			600	81.7	100	
Ni@Au/SiO ₂	3:1	60000	350	30.5	95	[3]
			400	55.0	95	
			450	80.0	94	
			500	102.6	96	
			550	109.9	95	
			600	110.5	95	
Mo ₂ C@N-C	3:1	24000	400	73.6	98	[4]
			450	89.1	99	
			500	102.9	100	
			550	111.4	100	
			600	126.4	100	
Ni-Cu/γ-Al ₂ O ₃	4:1	60000	350	25.7	75	[5]
			400	64.2	80	
Ni/CeO ₂	4:1	14000	450	43.8	10	[6]
			500	82.5	20	
CuIn/CeO ₂	4:1	3000	500	45.2	100	[7]
			550	54.9	100	
			350	8.9	52	
RuNi/CeZr	4:1	24000	400	12.1	63	[8]
			450	15.1	64	
			500	28.3	60	
			550	52.3	61	
			600	82.4	63	
Ni ₂ P-SiO ₂	4:1	12000	450	7.7	85	[9]

catalyst	H ₂ : CO ₂ ratio	WHSV (mL/g/h)	Temperature (°C)	Rate (mmol _{CO} ·g _{cat} ⁻¹ ·h ⁻¹)	CO selectivity (%)	Ref.
Ni ₂ P-SiO ₂	4:1	12000	500	14.7	84	[9]
			550	18.9	80	
			600	22.0	82	

Figure Captions:

Fig. S1 CO₂ conversion of R-0.25H₂ and R-0.2H₂ at different temperatures.

Fig. S2. Relationship between CO₂ conversion and Cu⁺ ratio (a)/oxygen vacancy content (b).

Fig. S3 (a) CO₂ conversion and (b) H₂ conversion of R-0.25H₂, R-0.5H₂ and R-H₂ catalysts at different particle sizes and WHSV (at 320 °C under P=1 bar).

Fig. S4 Changes of reaction rates of (a) R-0H₂ catalysts as a function of partial pressure of reactants at different temperatures (b) Apparent activation energy (E_a) of R-0H₂.

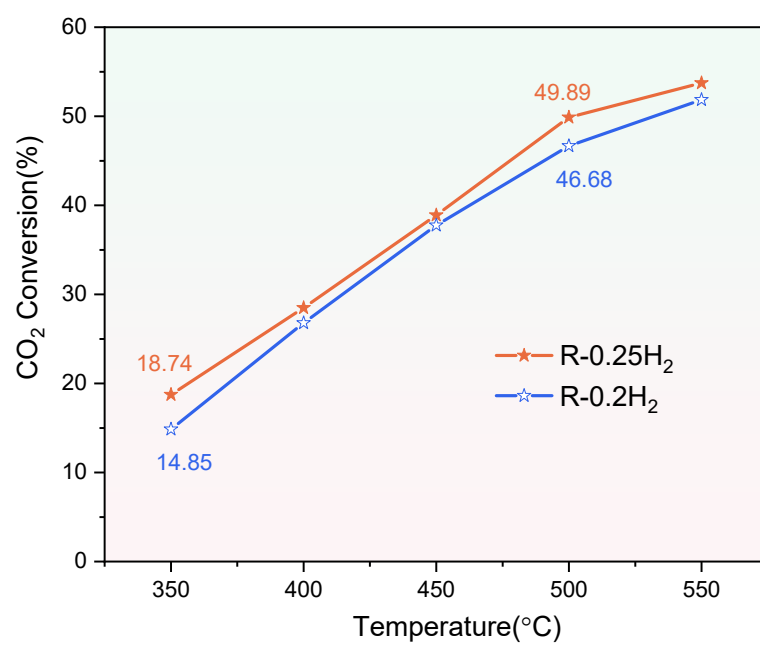


Fig. S1 CO₂ conversion of R-0.25H₂ and R-0.2H₂ at different temperatures.

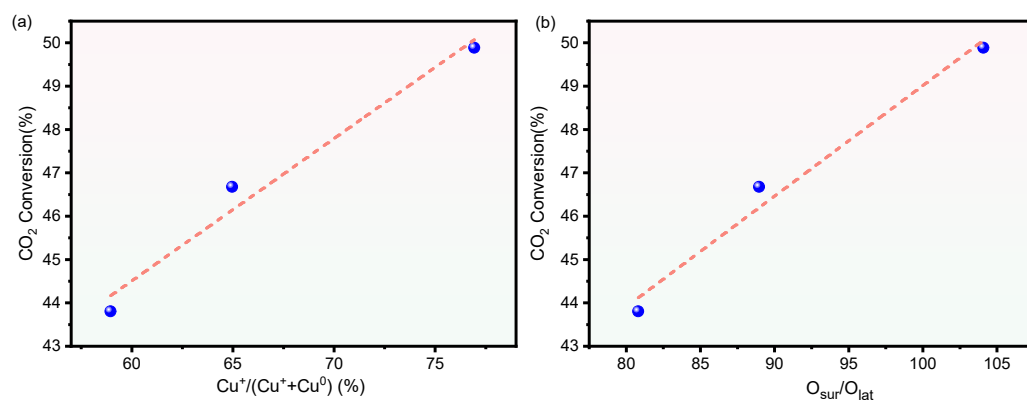


Fig. S2. Relationship between CO₂ conversion and Cu⁺ ratio (a)/oxygen vacancy content (b).

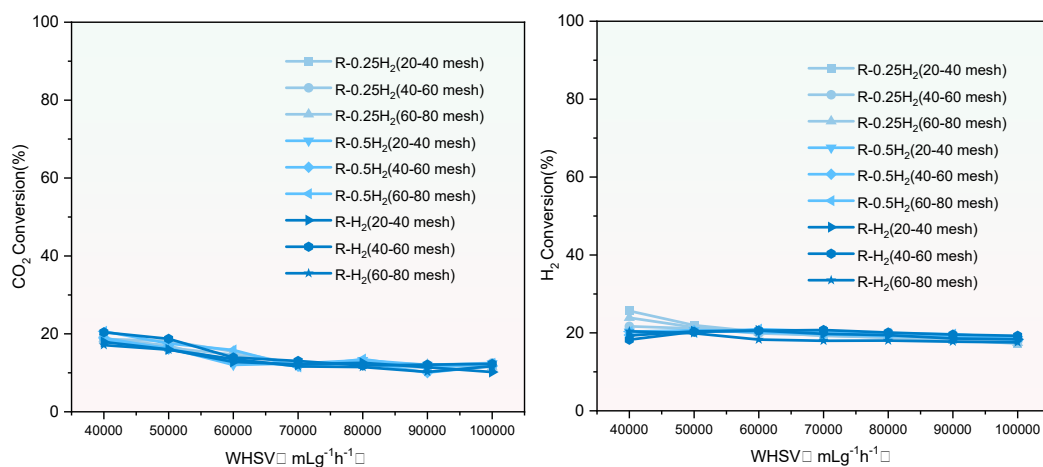


Fig. S3 (a) CO₂ conversion and (b) H₂ conversion of R-0.25H₂, R-0.5H₂ and R-H₂ catalysts at different particle sizes and WHSV (at 320 °C under P=1 bar).

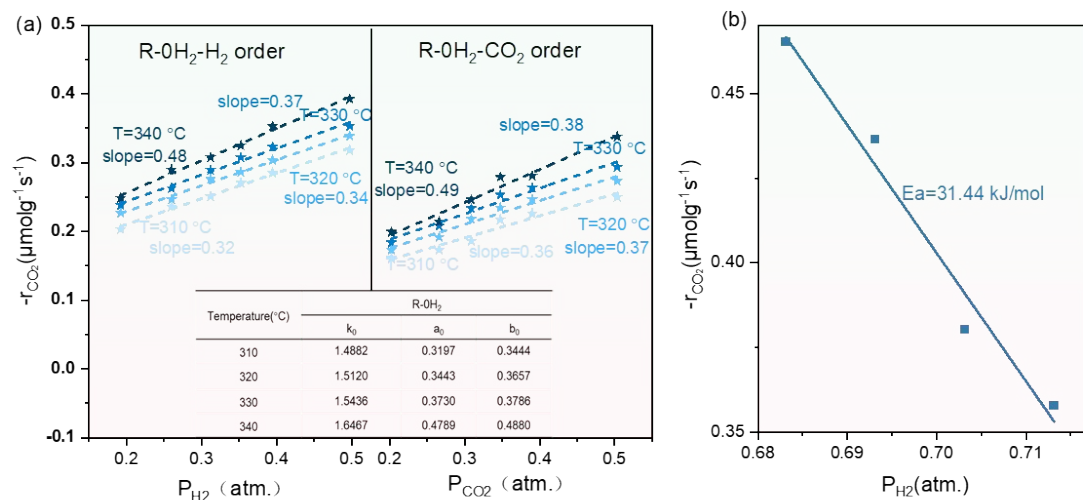


Fig. S4 Changes of reaction rates of R-0H₂ (a) catalysts as a function of partial pressure of reactants at different temperatures (b) Apparent activation energy (Ea) of R-0H₂.

References

- [1] Kang H, Zhu L, Li S, et al. Generation of oxide surface patches promoting H-spillover in Ru/(TiO_x)MnO catalysts enables CO₂ reduction to CO[J]. *Nature Catalysis*, 2023, 6(11): 1062-1072.
- [2] Li M, My Pham T H, Ko Y, et al. Support-dependent Cu-In bimetallic catalysts for tailoring the activity of reverse water gas shift reaction[J]. *ACS Sustainable Chemistry & Engineering*, 2022, 10(4): 1524-1535.
- [3] Zhang X, Han S, Zhu B, et al. Reversible loss of core-shell structure for Ni-Au bimetallic nanoparticles during CO₂ hydrogenation[J]. *Nature Catalysis*, 2020, 3(4): 411-417.
- [4] Ma Y, Guo Z, Jiang Q, et al. Molybdenum carbide clusters for thermal conversion of CO₂ to CO via reverse water-gas shift reaction[J]. *Journal of Energy Chemistry*, 2020, 50: 37-43.
- [5] Deng L, Ai X, Xie F, et al. Efficient Ni-based catalysts for low-temperature reverse water-gas shift (RWGS) reaction[J]. *Chemistry-An Asian Journal*, 2021, 16(8): 949-958.
- [6] Kuhaudomlap S, Srifa A, Koo-Amornpattana W, et al. Insight and comprehensive study of Ni-based catalysts supported on various metal oxides for CO₂ methanation[J]. *Scientific Reports*, 2024, 14(1): 23149..
- [7] Zhen Z, Tang W, Chu W W, et al. Microemulsion solvating-out co-precipitation strategy for fabricating highly active Cu-ZnO/Al₂O₃ dual site catalysts for reverse water gas shift[J]. *Catalysis Science & Technology*, 2020, 10(8): 2343-2352.
- [8] Le Saché E, Pastor-Perez L, Haycock B J, et al. Switchable catalysts for chemical CO₂ recycling: A step forward in the methanation and reverse water-Gas shift reactions[J]. *ACS sustainable chemistry & engineering*, 2020, 8(11): 4614-4622.
- [9] Zhang Q, Pastor-Pérez L, Villora-Picó J J, et al. Ni-Phosphide catalysts as versatile systems for gas-phase CO₂ conversion: Impact of the support and evidences of structure-sensitivity[J]. *Fuel*, 2022, 323: 124301.