

Fe single-atom catalysts embedded in reduced graphene oxide for electrochemical detection of uric acid and hydrogen peroxide

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S1. Materials and instruments

Indium tin oxide glass (SPE) was purchased from Rui Zhi Han Xing Technology Co., Ltd (Beijing, China). Zinc nitrate hexahydrate ($Zn(NO_3)_2 \cdot 6H_2O$) was purchased from Chuandong Chemical Industry Group (Chongqing, China). Dimethylimidazole (2-mIM), iron nitrate ($Fe(NO_3)_3 \cdot 9H_2O$), graphene oxide (GO), uric acid (UA), ascorbic acid (AA), dopamine hydrochloride (DA), glucose (Glu), urea (Ure), creatinine (Cre), etc. were obtained from Aladdin Biochemical Technology Co., Ltd. (Shanghai, China). Human Serum AB was purchased from NovoBiotechnology Co., Ltd. (Beijing, China).

The morphology was characterized by field-emission scanning electron microscope (SEM, Zeiss Gemini 360) and transmission electron microscopy (TEM, JEOL JEM-F200). The elemental compositions were clarified by X-ray photoelectron spectroscopy (XPS, Thermo Scientific-K-ALPHA). The atomic sites were determined by aberration-corrected high-angle annular dark-field scanning transmission electron microscopy (AC-HAADF-STEM, JEOL JEM-ARM200F). All electrochemical measurements were performed on a portable electrochemical analyzer (EmStat4R).

S2. Supplementary Figures

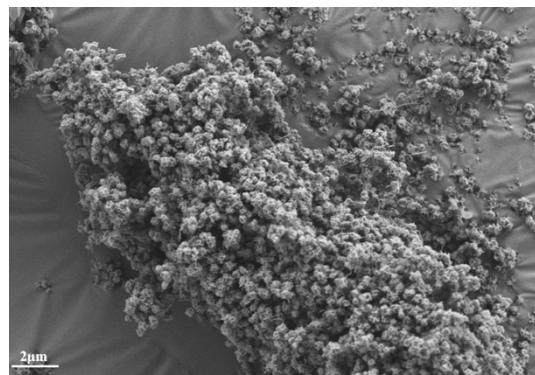


Fig. S1. SEM image of FeSACs.

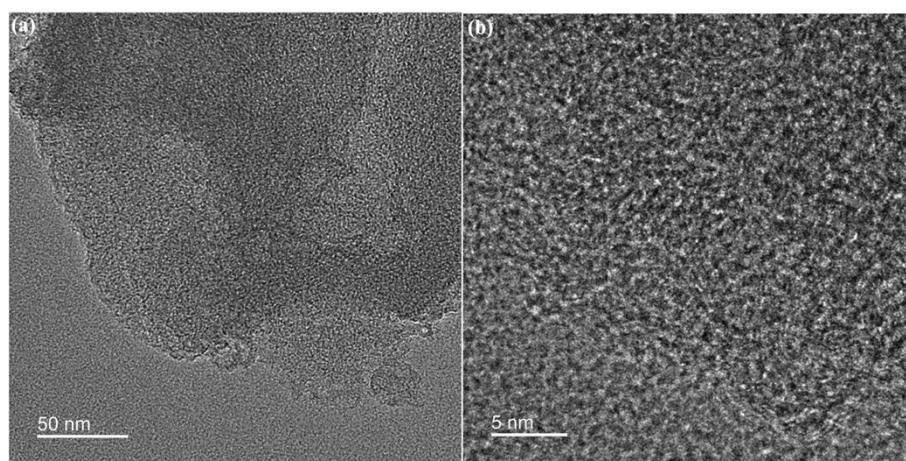


Fig. S2. High-resolution TEM images of FeSACs.

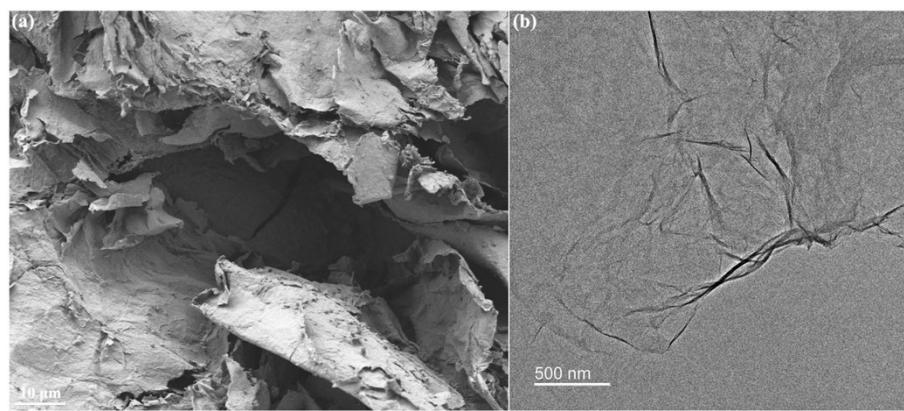


Fig. S3. SEM and TEM images of RGO.

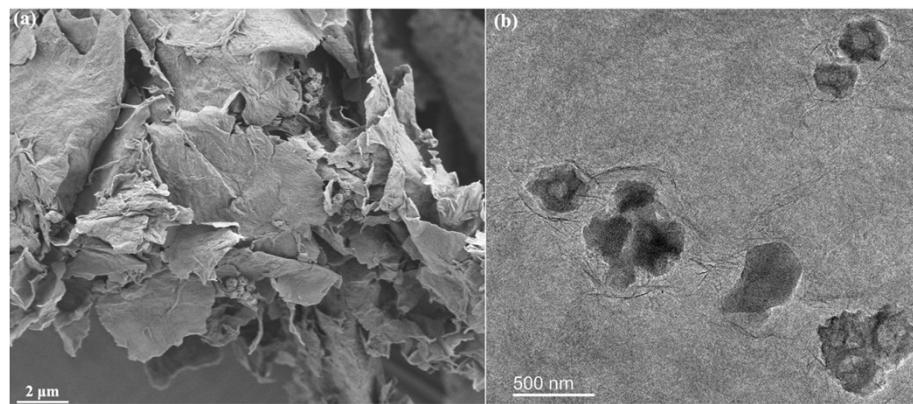


Fig. S4. SEM and TEM images of FeSACs@RGO.

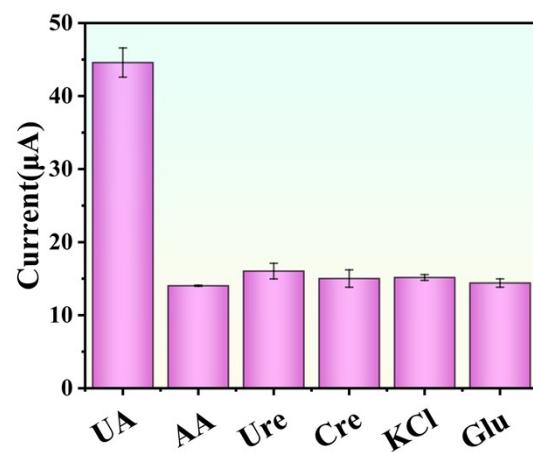


Fig. S5. Current of FeSACs@RGO/SPE toward 200 μ M UA.

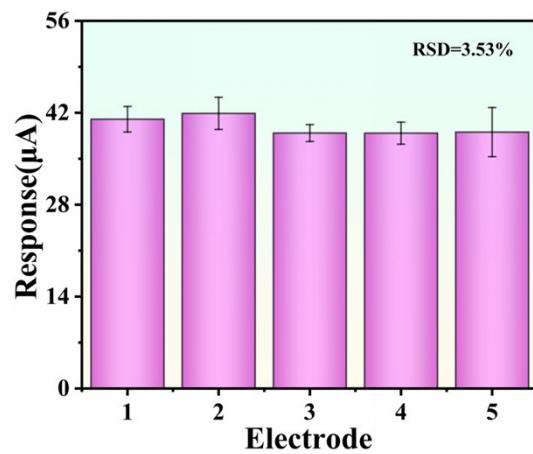


Fig. S6. Responses of five independent FeSACs@RGO/SPE toward 200 μ M UA .

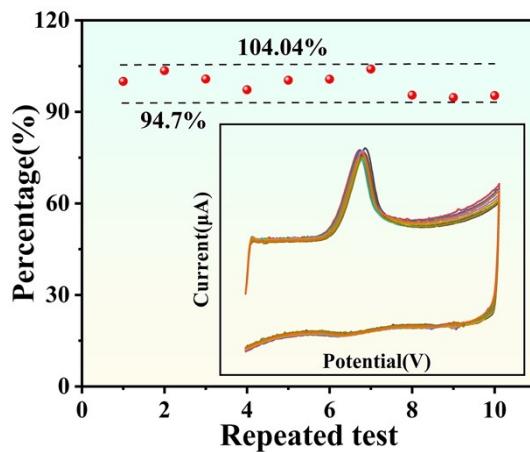


Fig. S7. Responses of FeSACs@RGO/SPE toward 200 μ M UA during 10 repeated test.

The Fe-N coordination structure and the catalytic performance of FeSACs distinctly follows the principle of “structure-activity relationship”. Specifically, the coordination structure could modulate the electronic structure of the Fe active sites (such as the d-band center position), which not only directly influences its electron transport capacity but also significantly regulates the adsorption strength for H_2O_2 . The catalytic mechanism is illustrated in Fig. S8. The H_2O_2 molecule is initially adsorbed onto the single-metal atom in SAC. Then, the adsorbed H_2O_2 readily undergoes homolytic cleavage, producing two hydroxyl groups. One hydroxyl group desorbs, yielding a free hydroxyl radical. The remaining adsorbed hydroxyl reacts with a proton under acidic conditions, forming a H_2O molecule attached to the metal center. Finally, desorption of the H_2O molecule regenerates the catalyst to its original state.

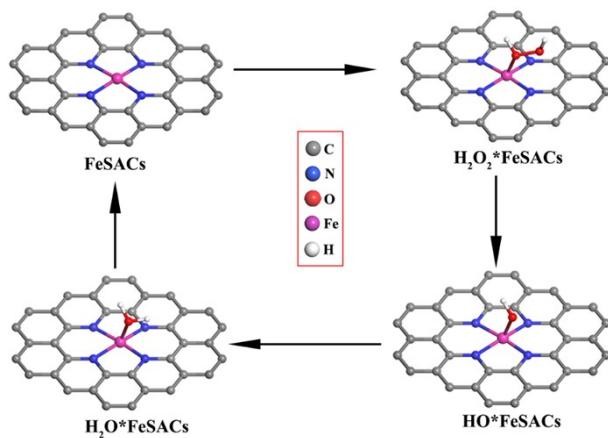


Fig. S8. Schematic illustration of the catalytic mechanism of H_2O_2 on FeSACs.

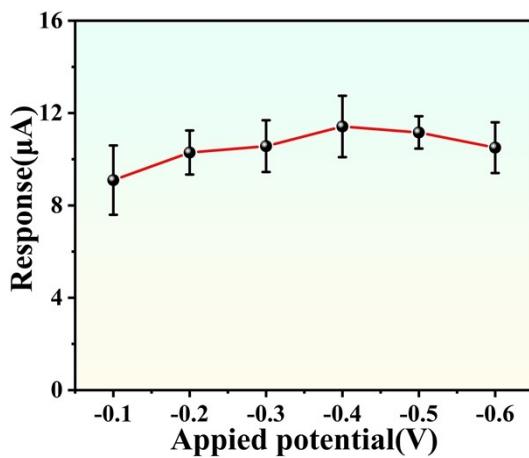


Fig. S9. The influence of applied potential on H_2O_2 response.

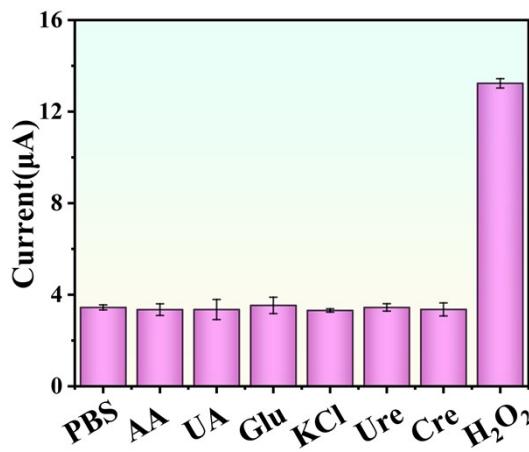


Fig. S10. Current responses of FeSACs@RGO/SPE toward different substances.

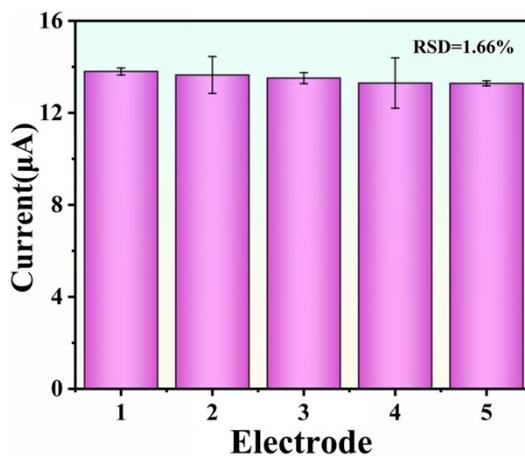


Fig. S11. Responses of five independent FeSACs@RGO/SPE toward 200 μM H_2O_2 .

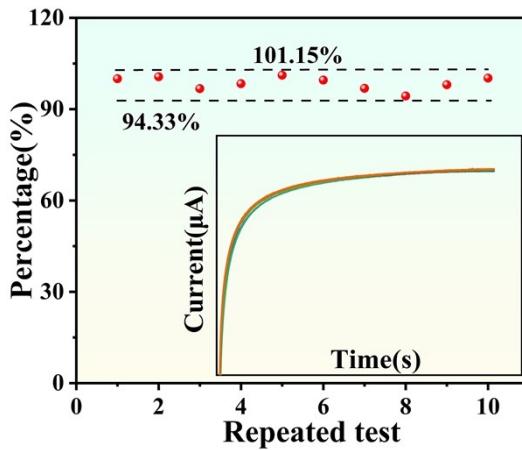


Fig. S12. Responses of FeSACs@RGO/SPE toward 200 μM H_2O_2 during 10 repeated test.

Table S1. Comparison of some recent electrochemical sensors for UA detection

Electrode material	Linear range (μM)	LOD (μM)	Ref.
FeNS/MWCNT	5-500	3.26	[1]
CoNiFe ₂ O ₄	4-5280	6.38	[2]
PyTS@Ti ₃ C ₂ T _x	5-100	0.48	[3]
Ce-BTC@MoS ₂	5-2500	5.00	[4]
MWCNT-COOH	0-1600	3.58	[5]
ENi-hMoS ₂	100-9000	7.30	[6]
PTA/CeO ₂ @Pt	10-138	1.04	[7]
Co ₃ O ₄ nanosheets	5-200	5.00	[8]
FeSACs@RGO	5-100	3.06	This work

Table S2. Comparison of some recent electrochemical sensors for H_2O_2 detection

Electrode material	Linear range (μM)	LOD (μM)	Ref.
CuO-CeO ₂ /MXene	5-100	1.67	[9]
FeCu-NZs	0.1-3800	0.06	[10]
Co@MOF-808	10-450	1.3	[11]
NiCo ₂ O ₄ -Ti ₃ C ₂ T _x	20-100	6	[12]
Pt1-rGO@PILs	2.3-250	1.5	[13]
FeSACs@RGO	5-4000	5.11	This work

Table S3. Results for UA detection in serum samples

Sample	Added (μM)	Found (μM)	Recovery (%)	RSD (%)
1	20	20.96	112.85	5.57
2	50	45.16	90.33	4.44
3	100	90.30	90.30	6.45

Table S4. Results for H₂O₂ detection in serum samples

Sample	Added (μM)	Found (μM)	Recovery (%)	RSD (%)
1	50	48.61	87.96	4.19
2	100	89.37	89.37	7.78
3	200	197.15	98.58	3.63

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