

Supporting Information

Porphyrin-based covalent organic polymers connected by imine-bonds: An efficient and recyclable heterogeneous photocatalyst for the oxidation of thioanisole

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Synthesis of 5,10,15,20-tetrakis(4-hydroxyphenyl) porphyrin (H₂THPP).

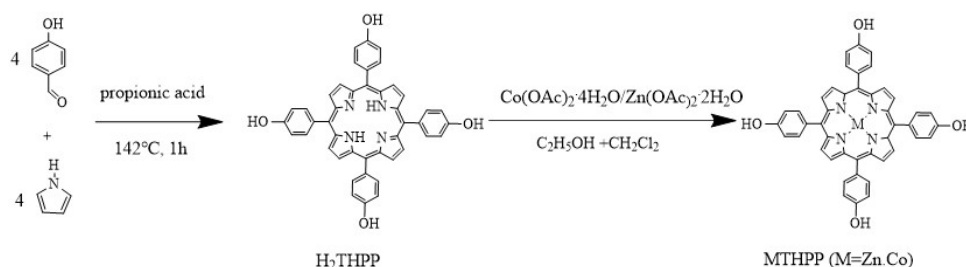
A mixture of 4-hydroxybenzaldehyde (3.0 g, 0.025 mmol) was placed in propionic acid (80 mL), then freshly distilled pyrrole (1.65 mL, 0.025 mol) in 10mL of propionic acid was dropwise added to the mixture with stirring. After the mixture was continuously stirred for 1 h under reflux, the mixture was allowed to distill out of propionic acid. Added 50 ml of dichloromethane, cooled overnight and a purple crude product are collected by filtration. The crude product was purified by SiO₂ column chromatography (CH₂Cl₂ / EtOH =10:1) to give purple porphyrin H₂THPP. Yield: 20 %; MS, (m/z) calcd for [M + H]⁺: 679.20; UV-vis (C₂H₅OH): λ_{max} / nm: 421 (Soret band), 514, 551, 692, 647 (Q band).

Synthesis of ZnTHPP.

H₂THPP (0.1356 g, 0.2 mmol), zinc acetate (1.317 g, 6mmol) were dissolved in a mixture of ethanol (20 ml) and dichloromethane (20 ml) and stirred for 48 h at room temperature. The aim of a 30-fold amount of zinc acetate to the porphyrin is to ensure full metallization of porphyrin under mild conditions. The crude product was purified by SiO₂ column chromatography (CH₂Cl₂ / EtOH =25:2) to give porphyrin ZnTHPP. Yield: 38 %; MS, (m/z) calcd for [M + H]⁺: 743.2; UV-vis (C₂H₅OH): λ_{max} / nm: 428 (Soret band), 539, 587 (Q band).

Synthesis of CoTHPP.

H₂THPP (0.1356 g, 0.2 mmol), cobalt acetate (1.494 g, 6 mmol) were dissolved in a mixture of ethanol (20 ml) and dichloromethane (20 ml) and refluxed for 48 h by heating. The crude product was purified by SiO₂ column chromatography (CH₂Cl₂ / EtOH =25:2) to give porphyrin CoTHPP. Yield: 27 %; MS, (m/z) calcd for [M + H]⁺: 736.3; UV-vis (CH₂Cl₂): λ_{max} / nm: 418 (Soret band), 533 (Q band).



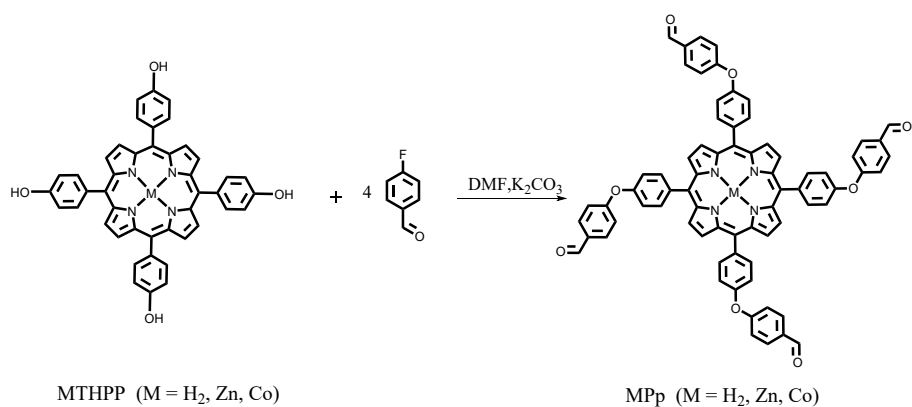
SCHEME S1 Synthesis of MTHPP

Synthesis of 5,10,15,20-[4-(4-Formyl) phenoxy] phenyl porphyrin (H₂Pp)

H₂THPP (0.136 g, 0.2 mmol) and 4-fluorobenzaldehyde (1.46 g, 2 mmol) were dissolved in N,N dimethylformamide (150 ml) and heated with a constant stirring under a nitrogen atmosphere. After that, K₂CO₃ (0.691 g, 5 mmol) was added to the solution when the reaction reached reflux temperature. TLC was checked at the conclusion of the reaction. The crude product was purified by SiO₂ column chromatography (CH₂Cl₂/EtOH = 10:1) to give purple porphyrin H₂Pp.

Synthesis of ZnPp and CoPp

ZnPp and CoPp was synthesized in the same route as the preparation of H₂Pp, only changed ZnTHPP and CoTHPP as the start material.



SCHEME S2 Synthesis of MPp

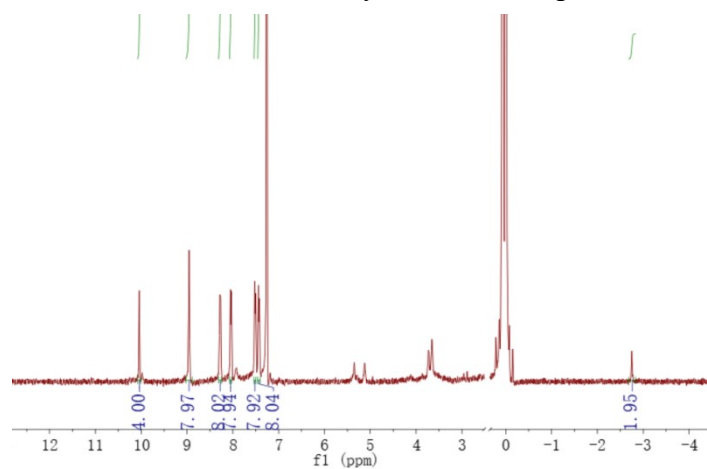


FIGURE S1 ¹H NMR spectra of H₂Pp taken at 400 MHz in CDCl₃

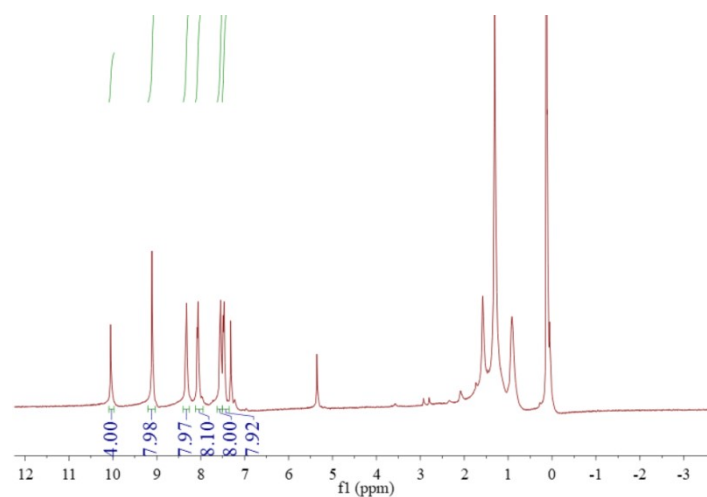


FIGURE S2 ¹H NMR spectra of ZnPp taken at 400 MHz in CDCl₃

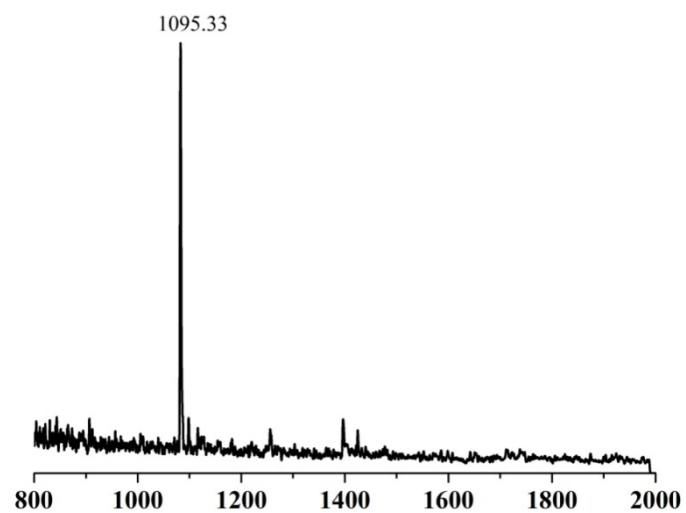


FIGURE S3 MS of H₂Pp

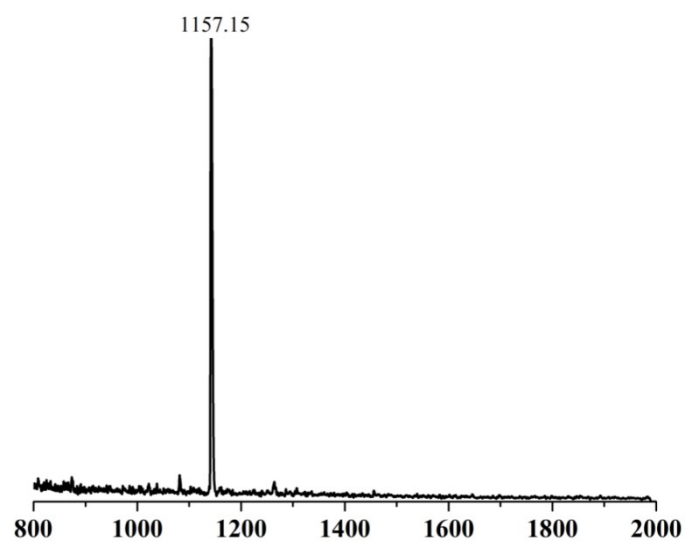


FIGURE S4 MS of ZnPp

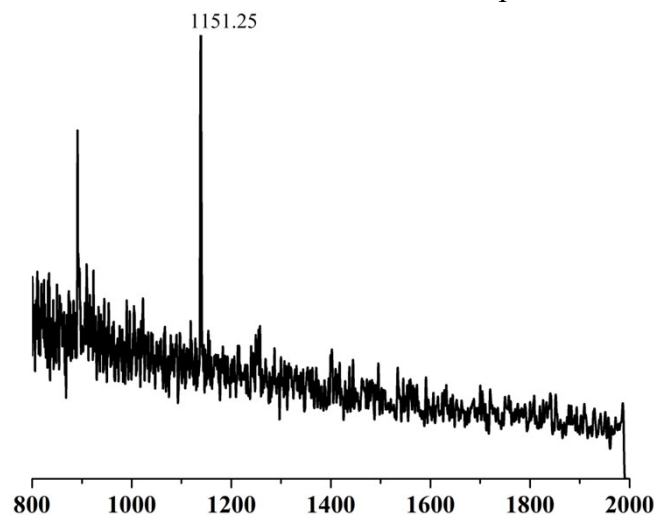
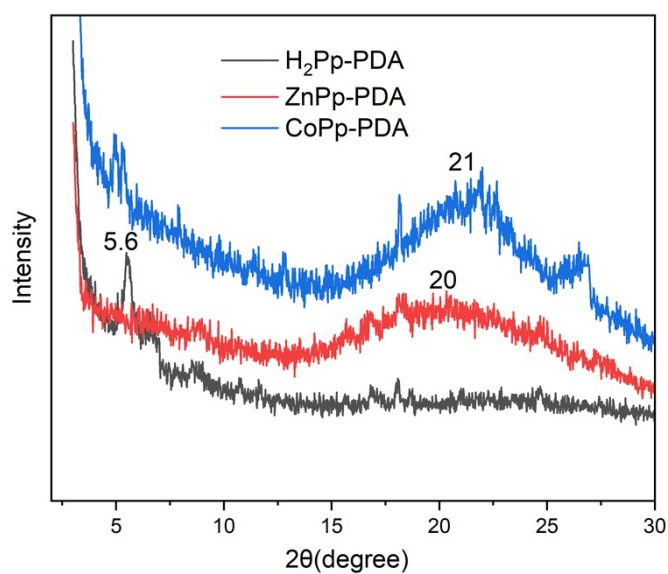
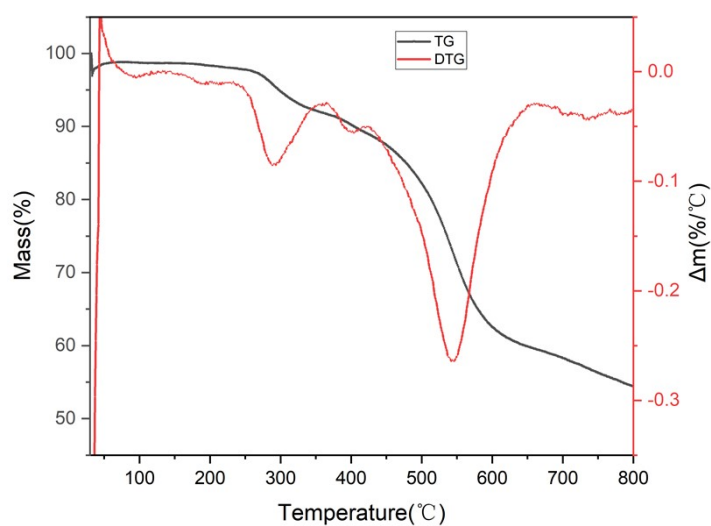


FIGURE S5 MS of CoPp

Table S1 UV-vis spectroscopy data of H₂Pp、ZnPp and CoPp

porphyrin	Soret band (nm)	Q band (nm)
H ₂ Pp	420	515 , 551 , 590 , 647
CoPp	412	529
ZnPp	422	549 , 587

**FIGURE S6** PXRD patterns of H₂Pp-PDA (black curve), ZnPp-PDA (red curve) and CoPp-PDA (blue curve)**FIGURE S7** DTG curves of H₂Pp-PDA

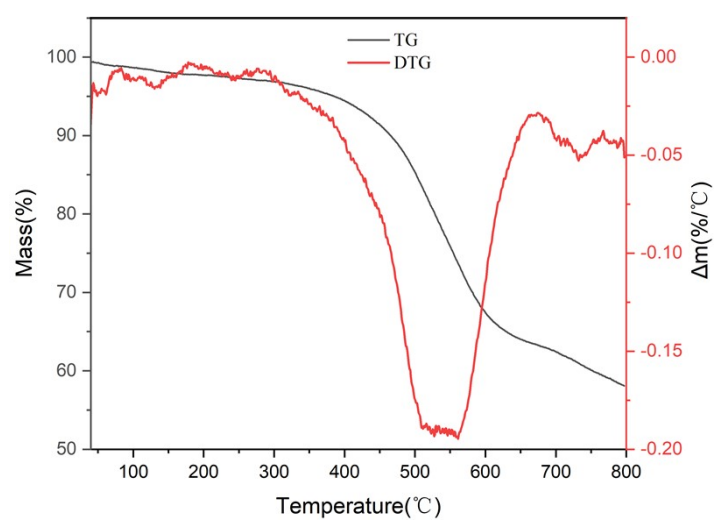


FIGURE S8 DTG curves of ZnPp-PDA

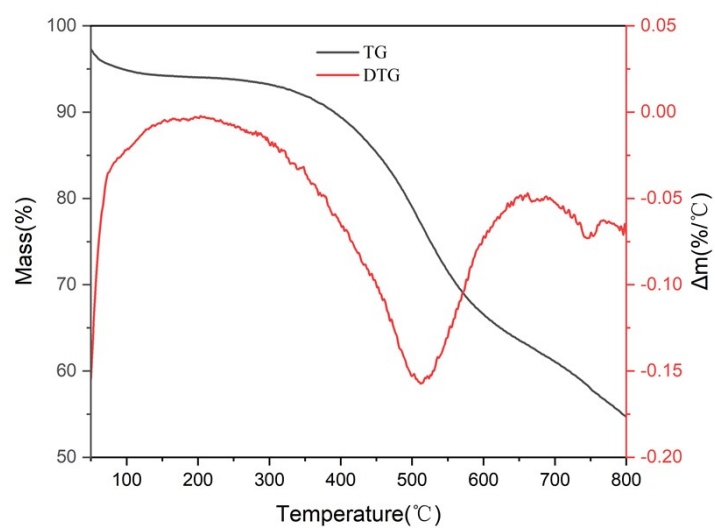


FIGURE S9 DTG curves of CoPp-PDA