

Electronic supplementary information (ESI)

Structural and magnetic properties of lanthanide - tetraphenol complexes based cyclen: slow magnetic relaxation of the Nd^{III} compound

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Fig. S1. IR spectrum of complex 1	2
Fig. S2. IR spectrum of complex 2	3
Fig. S3. IR spectrum of complex 3	4
Fig. S4. IR spectrum of complex 4	5
Fig. S5. IR spectrum of complex 5	6
Fig. S6. IR spectrum of H₄L^{Me,Cl}	7
Fig. S7. ESI-MS of complex 1 in CH ₃ CN	8
Fig. S8. ESI-MS of complex 2 in CH ₃ CN	9
Fig. S9. ESI-MS of complex 4 in CH ₃ CN	10
Fig. S10. ESI-MS of complex 5 in CH ₃ CN	11
Fig. S11. UV spectrum of complex 1 in CH ₃ CN	12
Fig. S12. UV spectrum of complex 2 in CH ₃ CN	14
Fig. S13. UV spectrum of complex 5 in CH ₃ CN.....	15
Fig. S14. UV spectrum of H₄L^{Cl,Me} in CH ₃ CN.....	16
Table S1. Crystallographic data and details of measurements for compounds 1, 2, 4 and 5	17
Table S2. SHAPE analyses for compounds 1, 2, 4 and 5	18
Table S3. Possible hydrogen bonds for compound 1	18
Fig. S15. a) Perspective view and b) Crystal packing of 2	19
Fig. S16. a) Perspective view and b) Crystal packing of 4	20
Fig. S17. a) Perspective view and b) Crystal Packing of 5	21
Table S4. Possible hydrogen bonds for compound 2	22
Table S5. Possible hydrogen bonds for compound 4	22
Table S6. Possible hydrogen bonds for compound 5	22
Fig. S18. Field dependence of the magnetization at 2 K in the 0-50 kOe field range for 1 (purple), 2 (black), 3 (green), 4 (red) and 5 (blue).	23
Fig. S19. Field dependence of the in-phase component of the magnetic susceptibility at 2 K in the 0-4000 Oe field range for 1	23
Table S7. Best fitted parameters (χ_T , χ_S , τ and α) with the extended Debye model for compound 1 at 2 K in the field range 0-4000 Oe.....	24
Fig. S20. Temperature dependence of the in-phase component of the magnetic susceptibility under an applied field of 3200 Oe in the 2-7 K temperature range for 1	25
Table S8. Best fitted parameters (χ_T , χ_S , τ and α) with the extended Debye model for compound 1 at 3200 Oe in the temperature range 2-5.1 K.....	25
Fig. S21. Normalized Argand plot for 1 in the 0-5.1 K temperature range.....	26
Table S9. <i>SINGLE_ANISO</i> computed energy barrier and angles between the g_{zz} axes of KDs	26
Table S10. <i>SINGLE_ANISO</i> computed Cystal-Field parameters B_k^q for the complex 1	26
Table S11. Decomposition of the RASSI wave functions corresponding to the lowest atomic multiplet $J=9/2$ in wave functions with definite projection of the total moment on the quantization axis for the states of the complex 1	27
Fig. S22. ESI-MS of H₄L^{Me,Cl} in MeOH	28

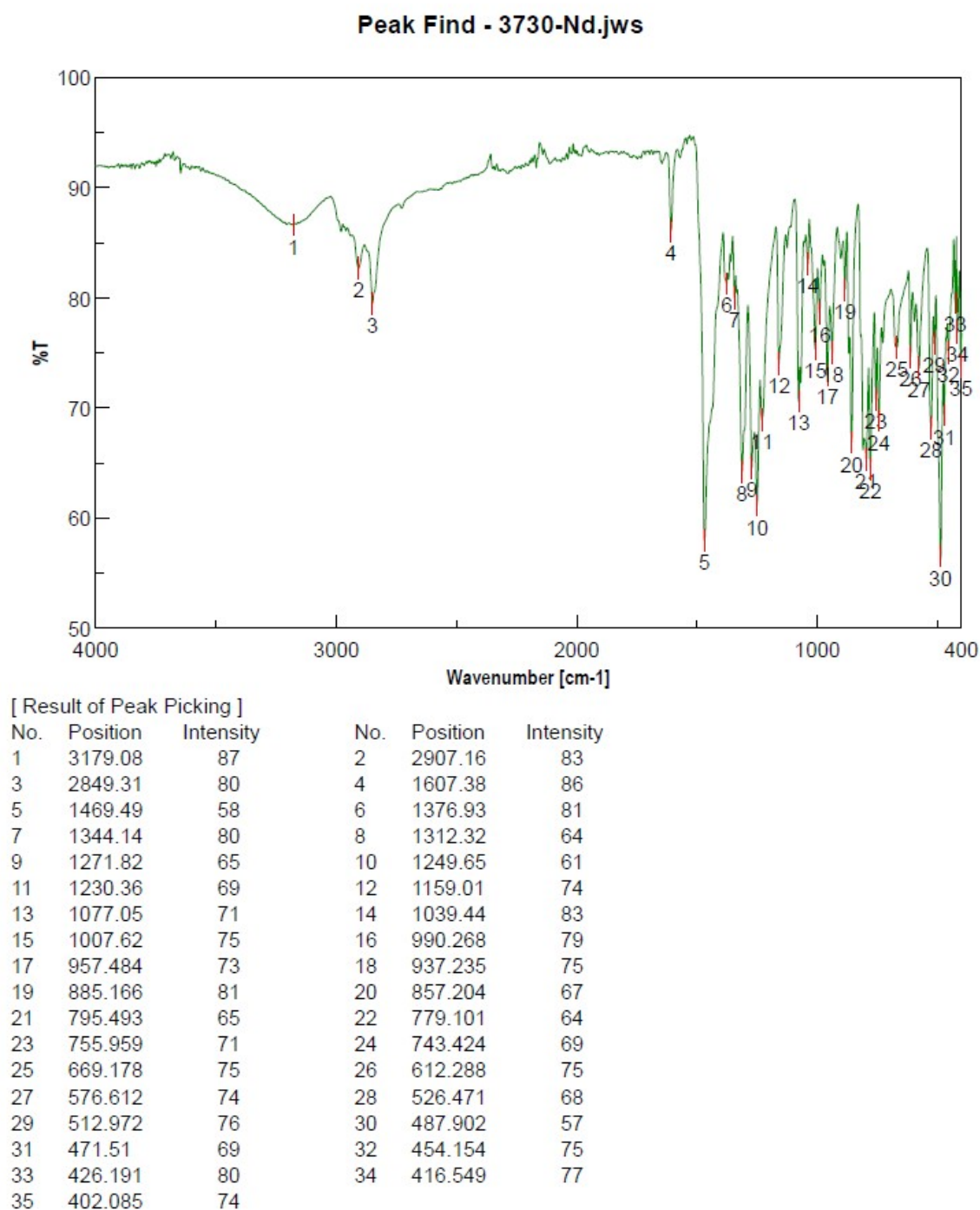


Fig. S1. IR spectrum of complex **1**

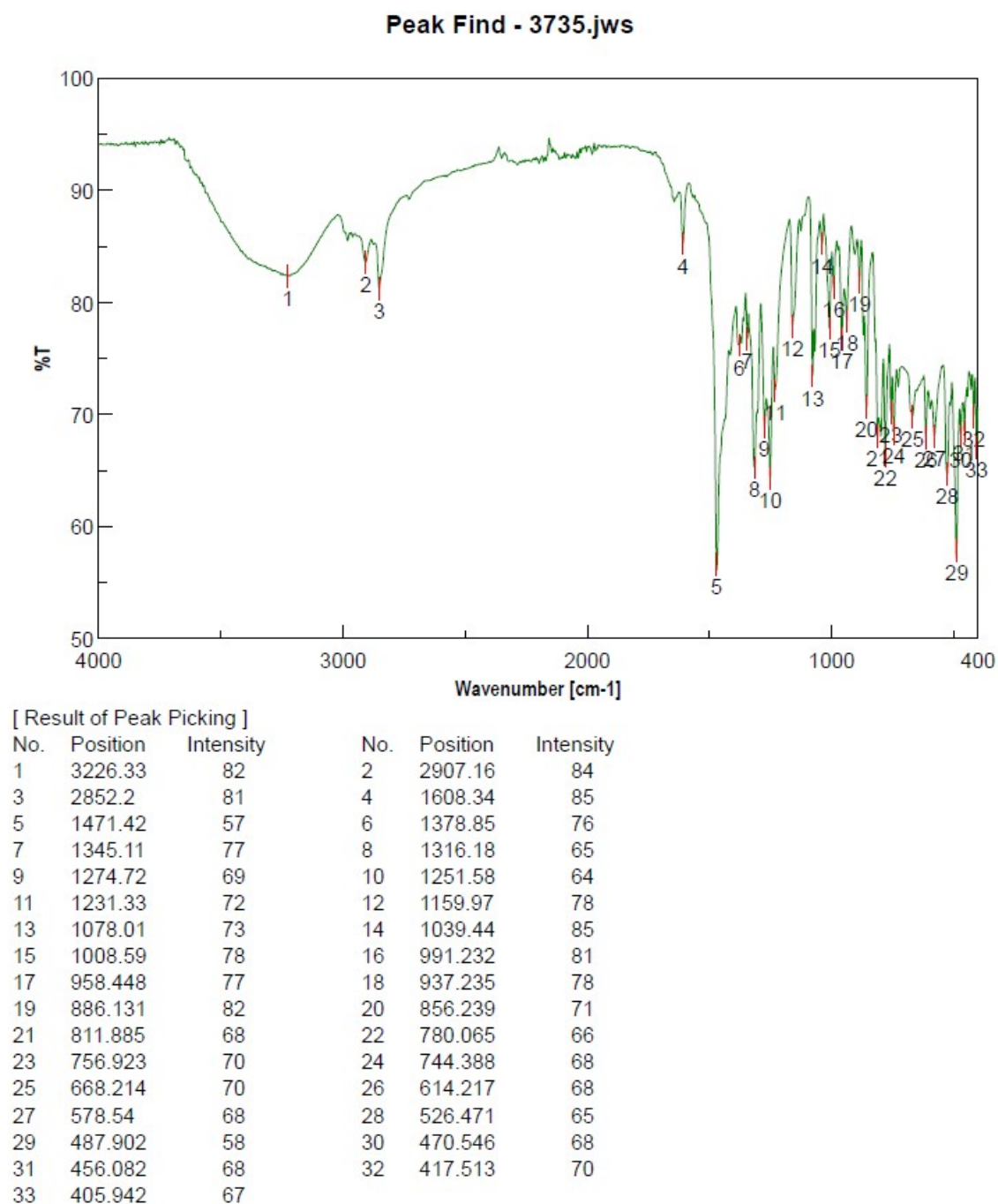


Fig. S2. IR spectrum of complex **2**

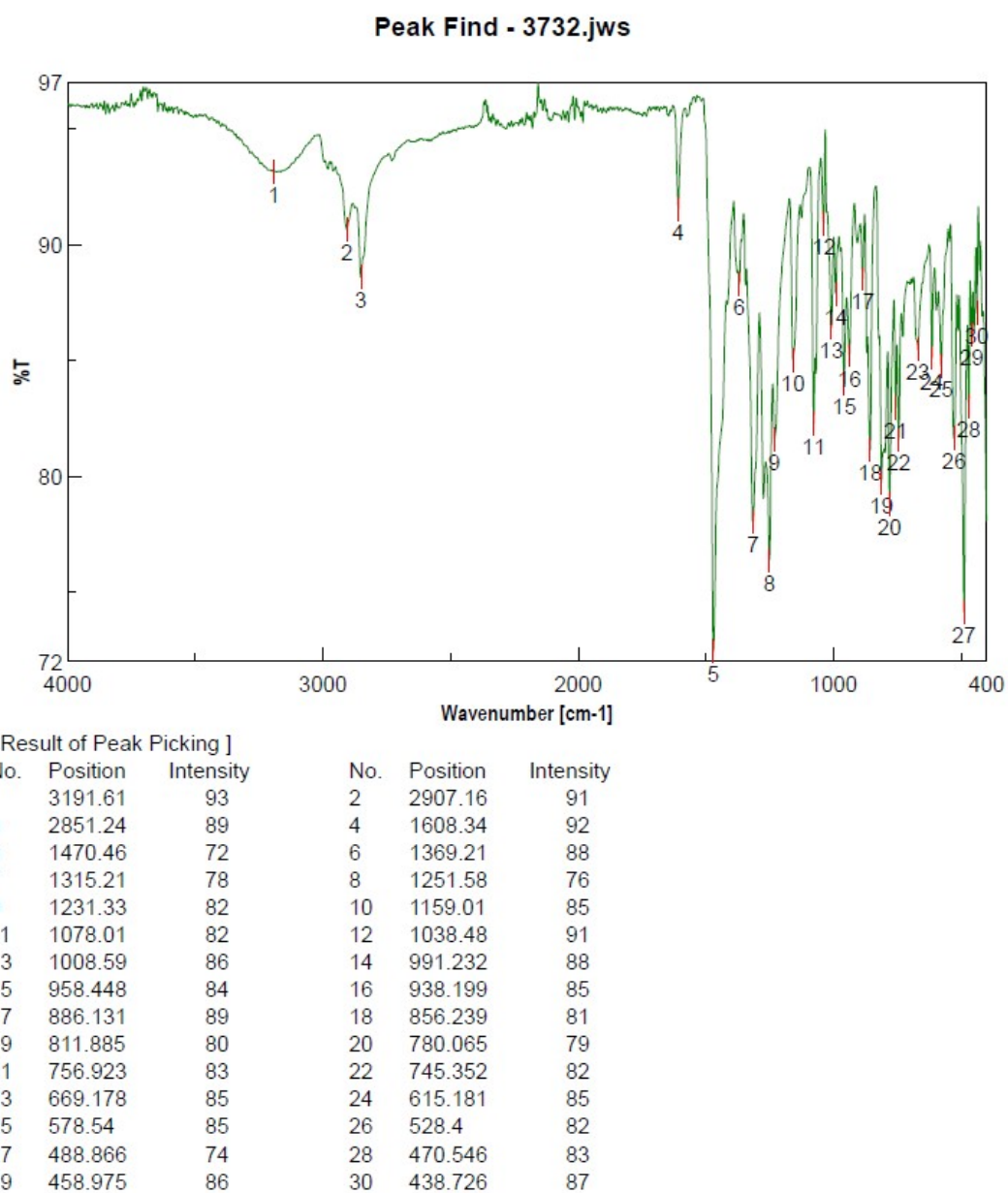


Fig. S3. IR spectrum of complex **3**

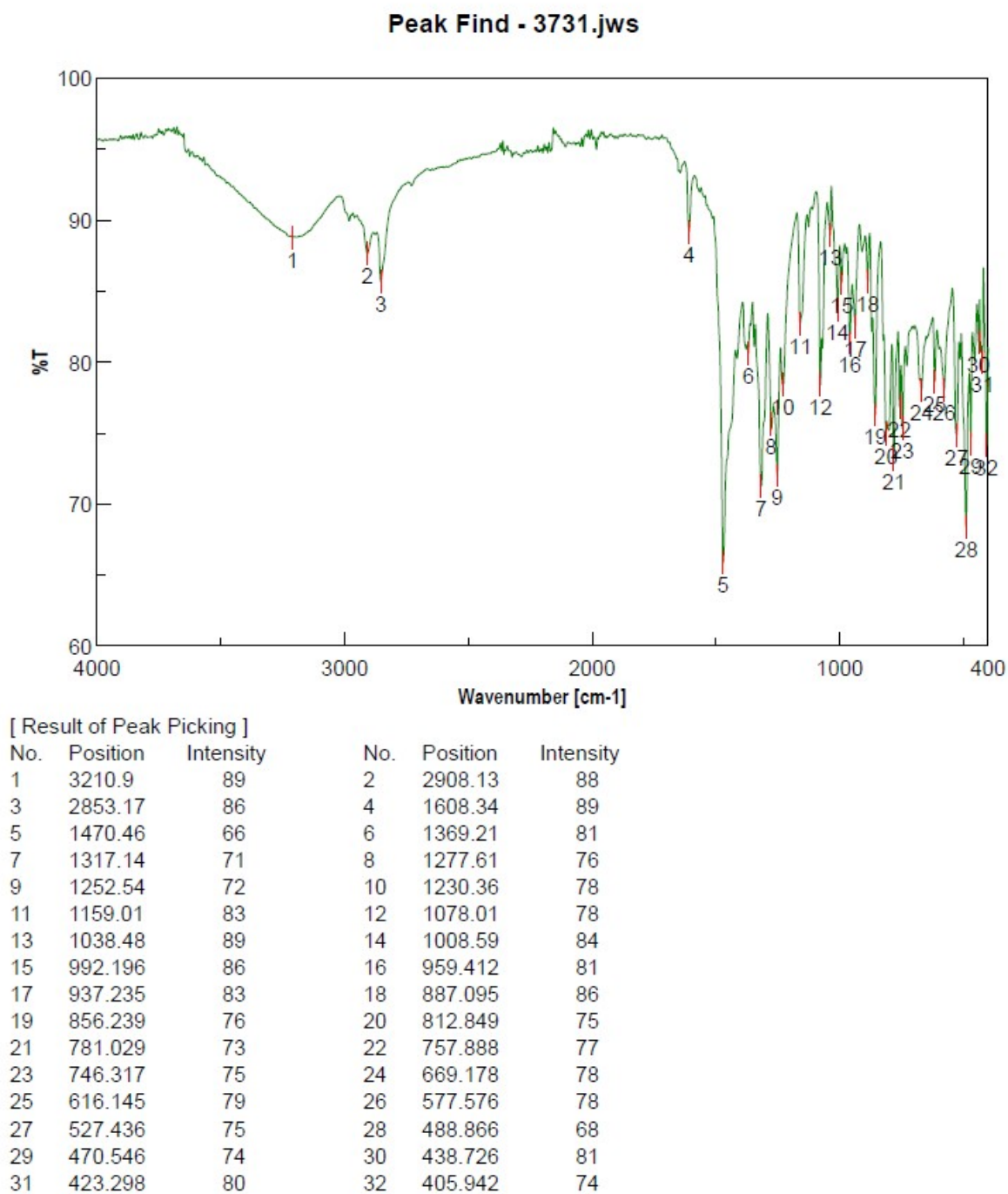
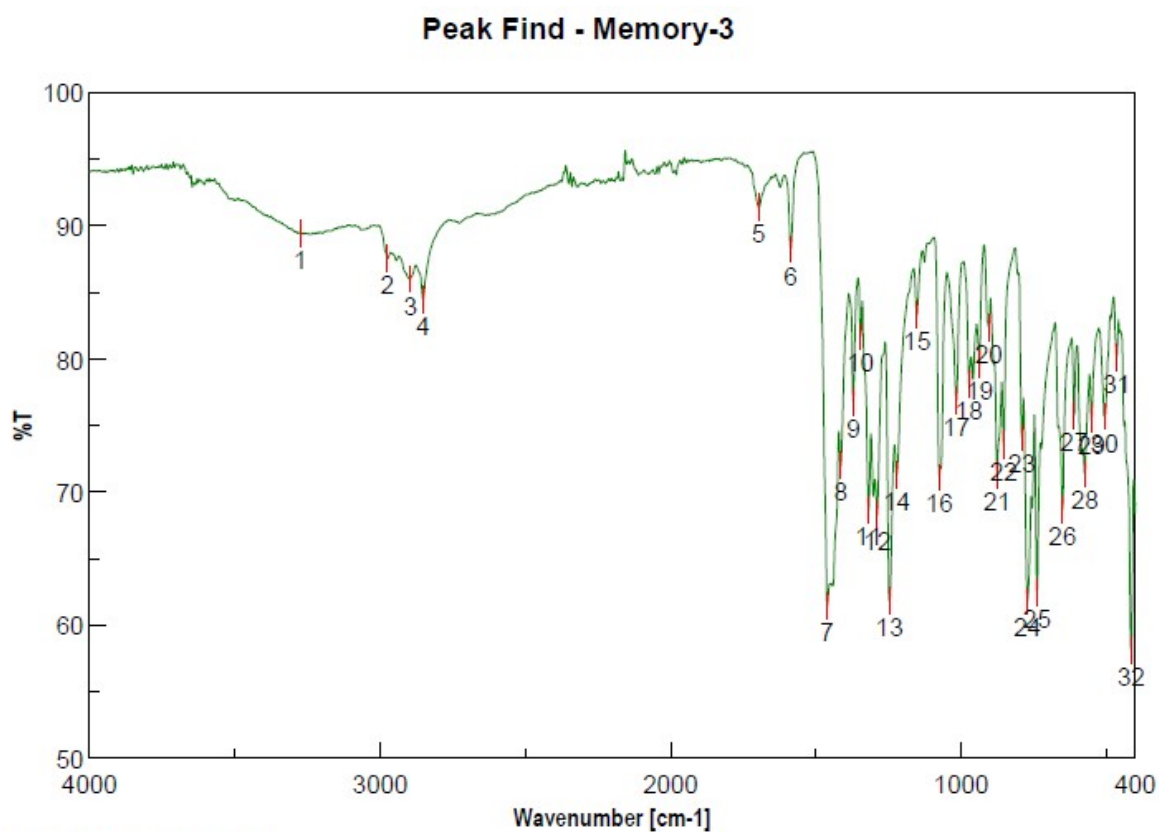


Fig. S4. IR spectrum of complex **4**

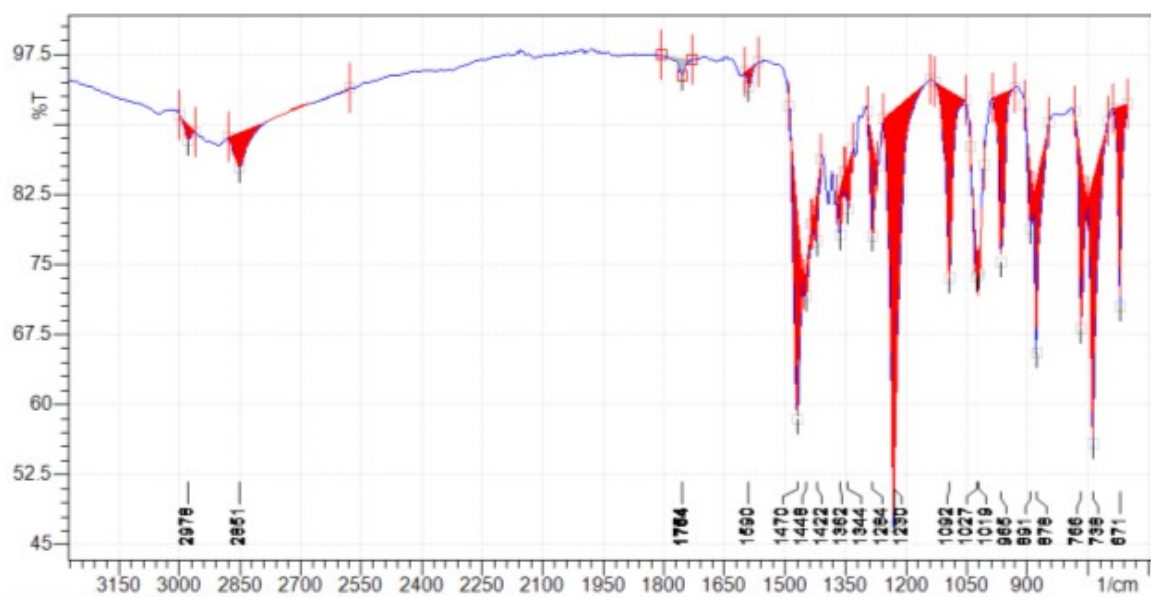


[Result of Peak Picking]

No.	Position	Intensity	No.	Position	Intensity
1	3274.54	89	2	2975.62	88
3	2896.56	86	4	2852.2	84
5	1697.05	91	6	1586.16	88
7	1460.81	61	8	1414.53	72
9	1371.14	77	10	1346.07	82
11	1318.11	69	12	1289.18	68
13	1246.75	62	14	1221.68	71
15	1152.26	83	16	1075.12	71
17	1016.3	77	18	972.912	78
19	937.235	80	20	904.451	82
21	876.488	71	22	853.347	74
23	790.671	74	24	771.387	62
25	738.603	62	26	650.858	69
27	611.324	76	28	574.683	71
29	551.542	75	30	506.223	76
31	463.796	80	32	414.62	58

Fig. S5. IR spectrum of complex **5**

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Department: chemistry



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Fig. S6. IR spectrum of $\text{H}_4\text{L}^{\text{Me,Cl}}$

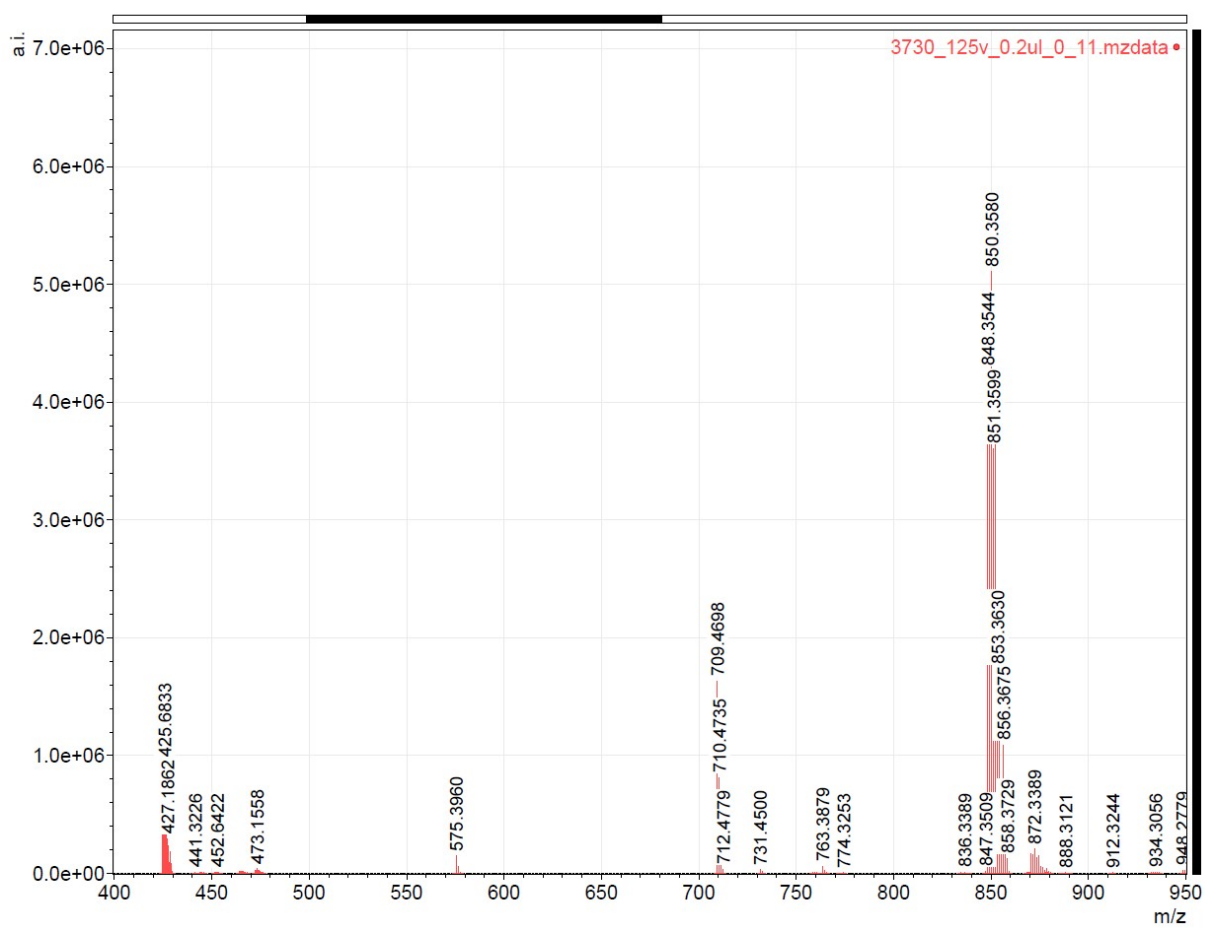


Fig. S7. ESI-MS of complex **1** in CH₃CN

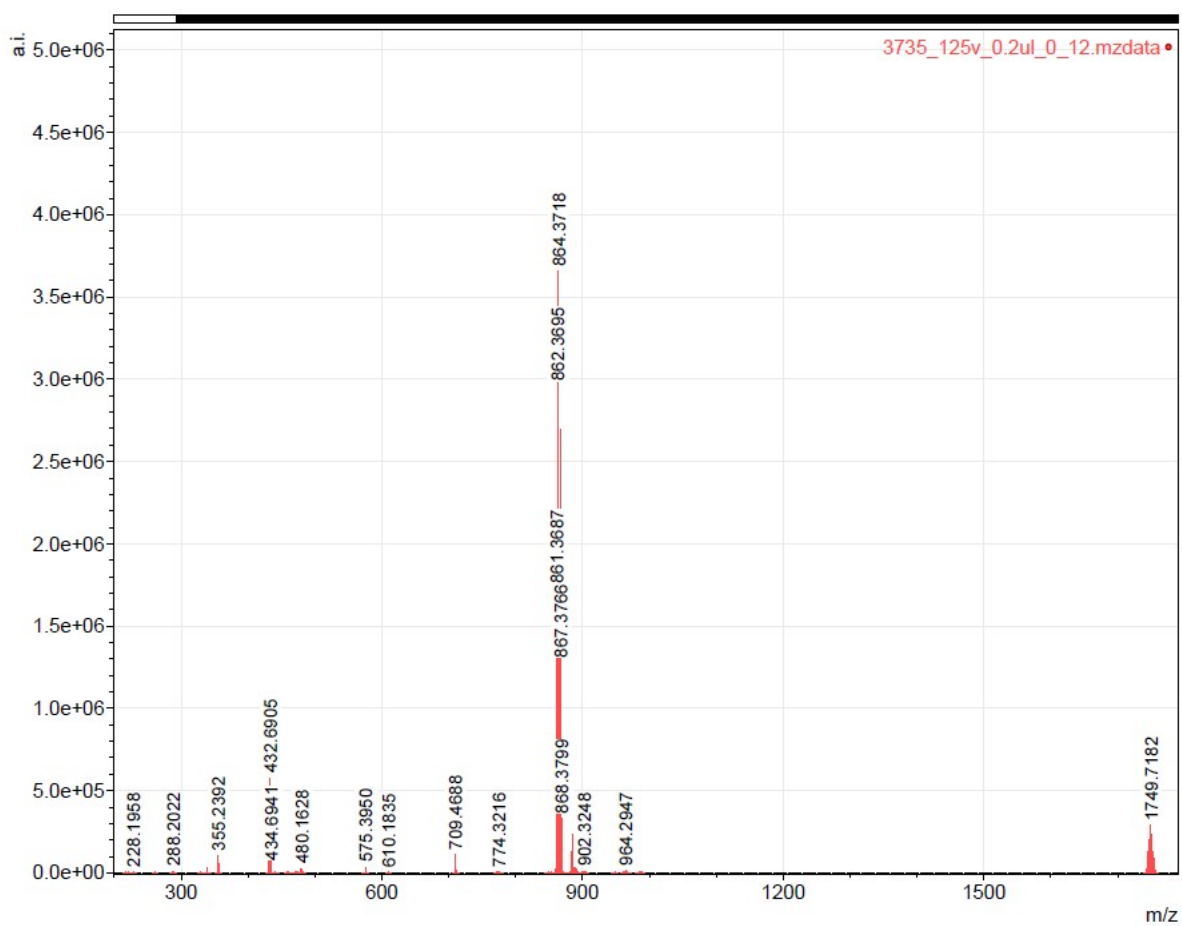


Fig. S8. ESI-MS of complex **2** in CH₃CN

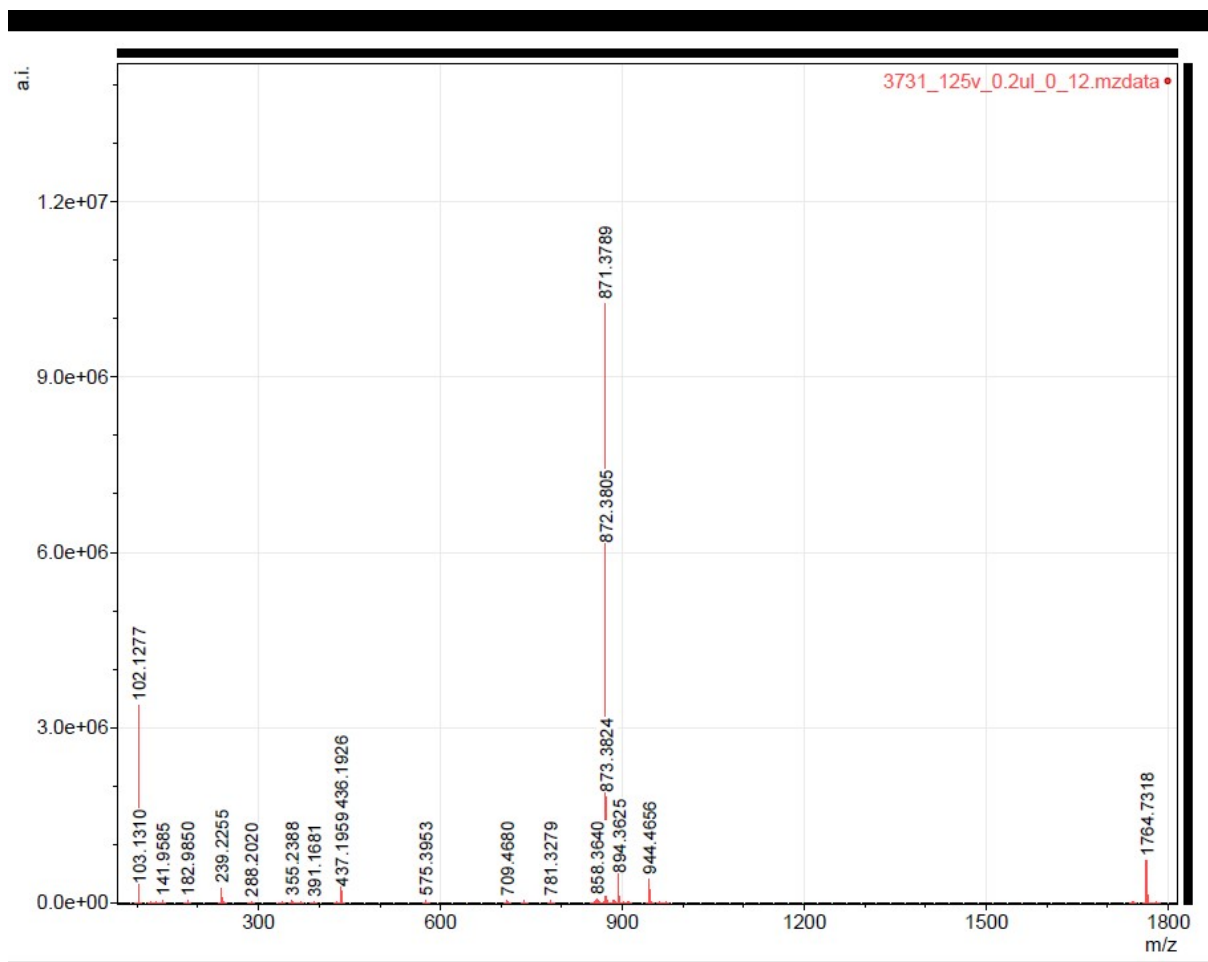


Fig. S9. ESI-MS of complex **4** in CH₃CN

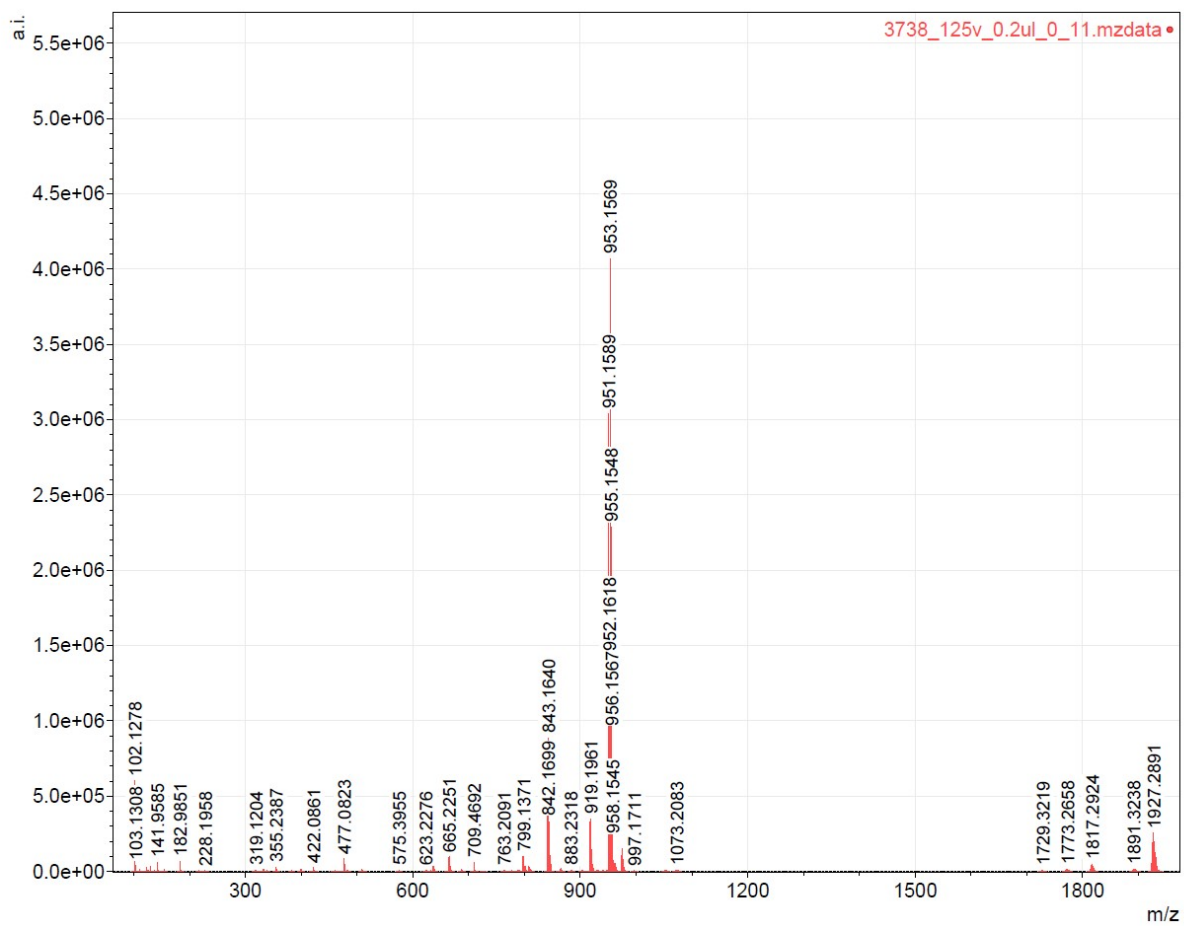


Fig. S10. ESI-MS of complex **5** in CH₃CN

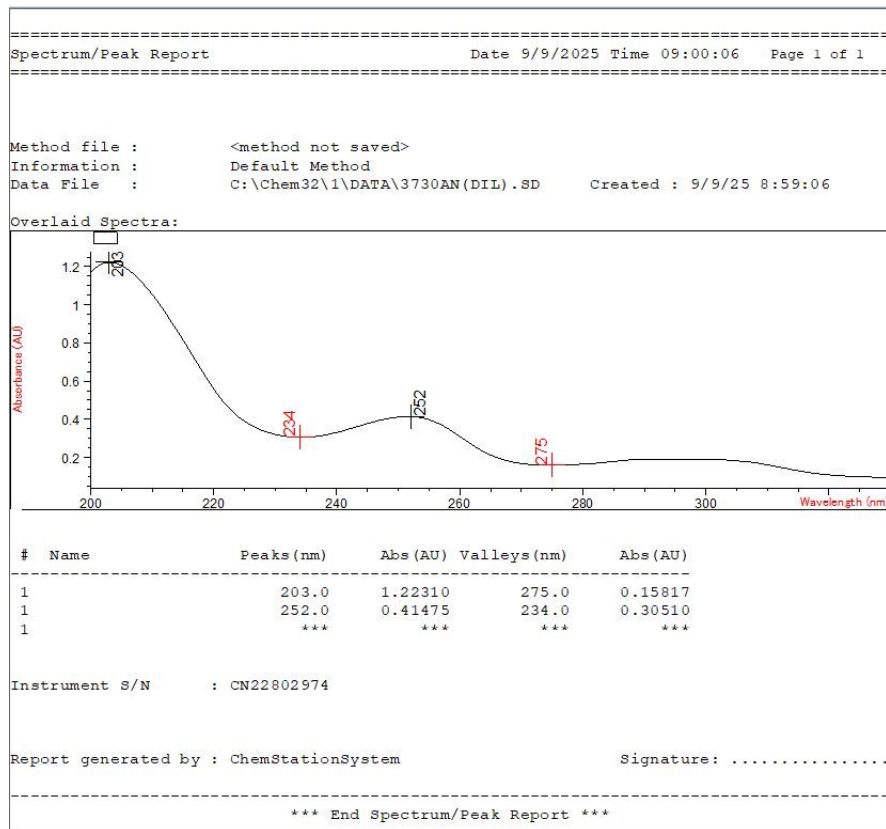
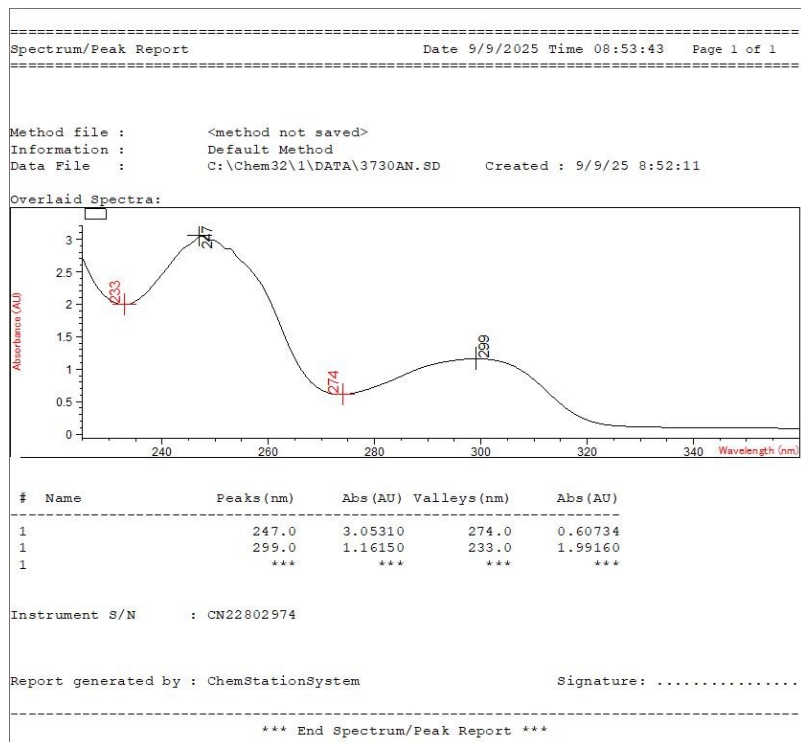


Fig. S11. UV spectrum of complex **1** in CH₃CN (top: 1.35×10^{-4} M; bottom: 1.35×10^{-5} M).

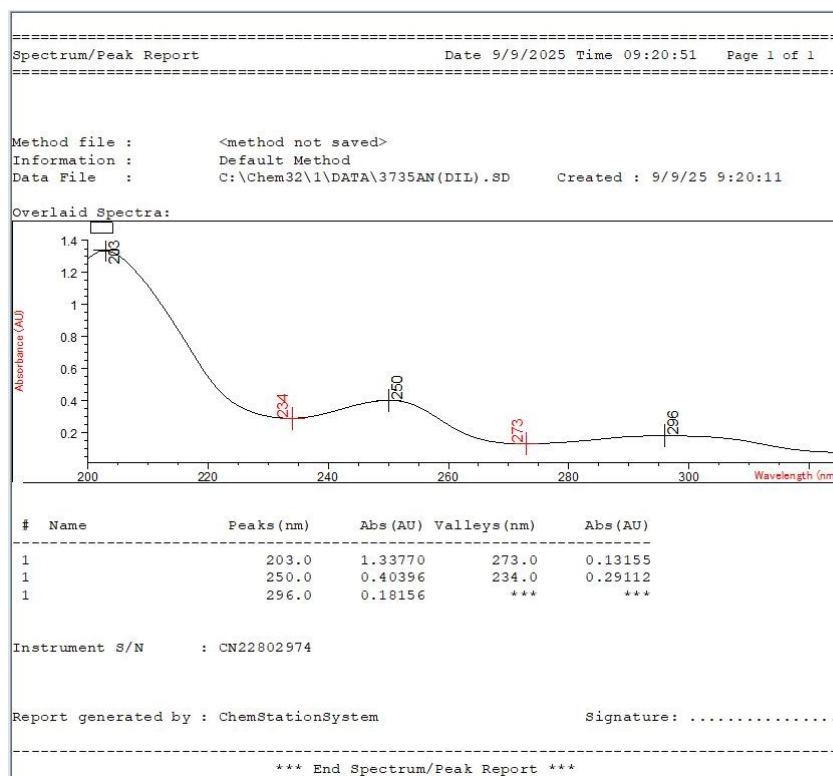
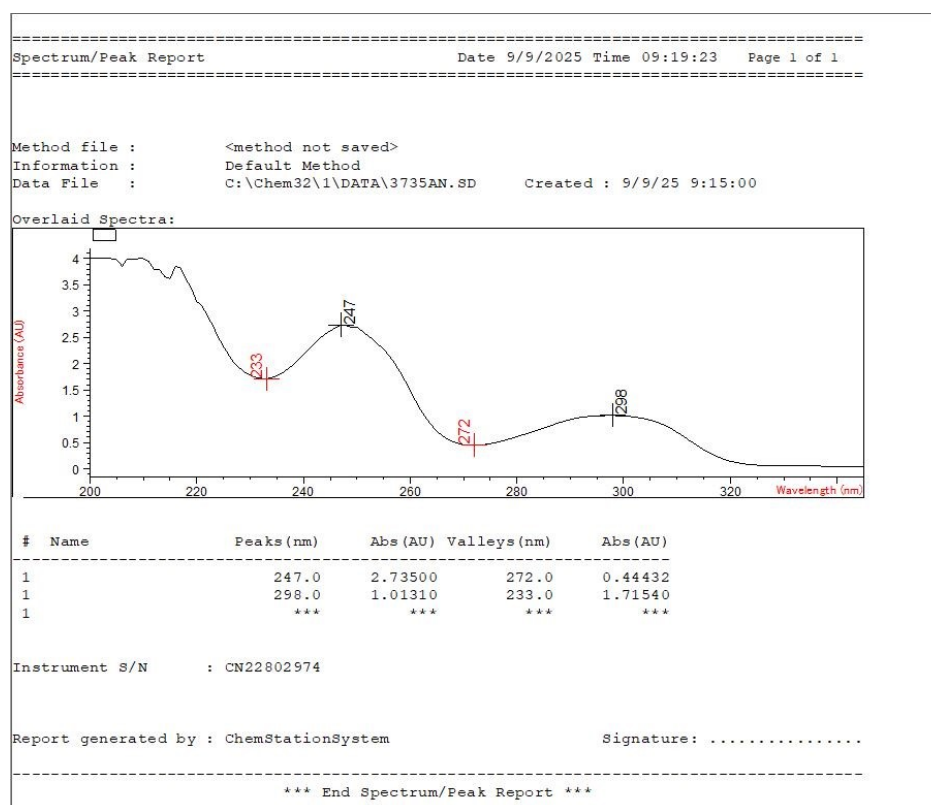
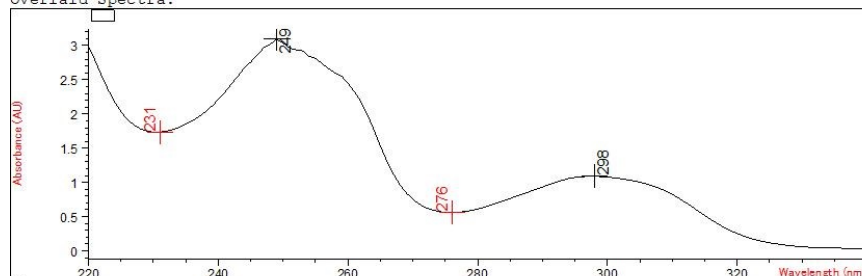


Fig. S12. UV spectrum of complex **2** in CH₃CN (top: 1.35×10^{-4} M; bottom: 1.35×10^{-5} M)

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Overlaid Spectra:



#	Name	Peaks (nm)	Abs (AU)	Valleys (nm)	Abs (AU)
1		249.0	3.09180	276.0	0.55907
1		298.0	1.08920	231.0	1.73450
1		***	***	***	***

Instrument S/N : CN22802974

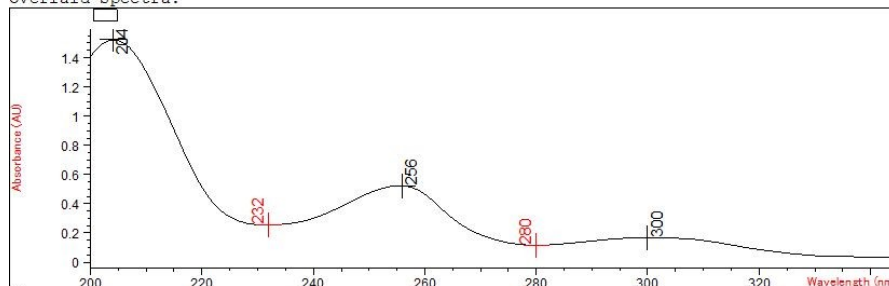
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Signature:

*** End Spectrum/Peak Report ***

Method file : <method not saved>
 Information : Default Method
 Data File : C:\Chem32\1\DATA\3738AN(DIL).SD Created : 9/9/25 9:09:52

Overlaid Spectra:



#	Name	Peaks (nm)	Abs (AU)	Valleys (nm)	Abs (AU)
1		204.0	1.52250	280.0	0.11598
1		256.0	0.52314	232.0	0.25359
1		300.0	0.16645	***	***

Instrument S/N : CN22802974

Report generated by : ChemStationSystem

Signature:

*** End Spectrum/Peak Report ***

Fig. S13. UV spectrum of complex **5** in CH₃CN (top: 1.35×10^{-4} M; bottom: 1.35×10^{-5} M)



Fig. S14. UV spectrum of $\text{H}_4\text{L}^{\text{Cl,Me}}$ in CH_3CN (top: 1.35×10^{-4} M; bottom: 1.35×10^{-5} M).

Table S1. Crystallographic data and details of measurements for compounds **1**, **2**, **4** and **5**
 $R1 = \Sigma |F_o| - |F_c| / \Sigma |F_o|$; $wR2 = [\Sigma w(F_o^2 - F_c^2)^2 / \Sigma w(F_o^2)^2]^{1/2}$

Compound	[Nd(HL ^{Me,Me})]·MeOH (1)	[Gd(HL ^{Me,Me})]·MeOH (2)	[Ho(HL ^{Me,Me})]·MeOH (4)	[Ho(HL ^{Me,Cl})]·2MeOH (5)
CCDC No.	2489782	2489783	2489784	2489785
Formula	C ₄₄ H ₅₇ N ₄ NdO ₄ ·CH ₄ O	C ₄₄ H ₅₇ GdN ₄ O ₄ ·CH ₄ O	C ₄₄ H ₅₇ HoN ₄ O ₄ ·CH ₄ O	C ₄₂ H ₅₃ Cl ₄ HoN ₄ O ₆
Fw (g mol ⁻¹)	882.21	895.22	902.90	1016.61
<i>a</i> (Å)	14.8343(6)	14.8229(4)	14.9515(7)	13.1871(6)
<i>b</i> (Å)	12.4668(5)	12.4491(4)	12.4578(6)	18.6904(8)
<i>c</i> (Å)	22.4721(9)	22.4326(8)	22.3763(11)	17.0530(6)
α (°)	90	90	90	90
β (°)	90	90	90	102.217(2)
γ (°)	90	90	90	90
<i>V</i> (Å ³)	4155.9(3)	4139.5(2)	4167.9(3)	4107.9(3)
<i>Z</i>	4	4	4	4
Crystal size (mm)	0.09 × 0.07 × 0.06	0.15 × 0.1 × 0.08	0.17 × 0.13 × 0.09	0.17 × 0.14 × 0.08
Crystal habit	Block, light blue	Block, colourless	Block, green	Block, green
Crystal system	Orthorhombic	Orthorhombic	Orthorhombic	Monoclinic
Space group	<i>Pca</i> 2 ₁	<i>Pca</i> 2 ₁	<i>Pca</i> 2 ₁	<i>P</i> 2 ₁ / <i>n</i>
<i>d</i> _{calc} (Mg m ⁻³)	1.410	1.436	1.439	1.598
μ (mm ⁻¹)	1.30	1.65	1.95	2.24
Radiation type	Mo K α	Mo K α	Mo K α	Mo K α
Wavelength (Å)	0.71073	0.71073	0.71073	0.71073
<i>T</i> (K)	100(2)	100(2)	100(2)	100(2)
θ range (°)	2.3–33.1	2.4–28.3	2.3–27.9	2.4–27.9
<i>F</i> (000)	1836	1852	1864	2000
<i>T</i> _{min} , <i>T</i> _{max}	0.622, 0.747	0.477, 0.746	0.542, 0.746	0.616, 0.746
<i>R</i> _{int}	0.098	0.103	0.101	0.080
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	203178, 15842, 14048	112309, 112309, 8798	235200, 10760, 9522	81536, 9945, 8019
independent reflections	15842	112309	10760	9945
No. of parameters, restraints	514, 1	507, 1	507, 1	533, 0
$\Delta\rho_{max}$, $\Delta\rho_{min}$ (e Å ⁻³)	1.31, -1.14	1.46, -3.67	0.85, -1.05	0.74, -1.14
GooF	1.06	1.05	1.04	1.04
Absolute structure parameter	0.030(9)	0.011(7)	0.028(5)	—
R1, wR2 (all)	R1 = 0.0347	R1 = 0.0528	R1 = 0.0365	R1 = 0.0506

data)	wR2 = 0.0614	wR2 = 0.0734	wR2 = 0.0614	wR2 = 0.0590
R1, wR2	R1 = 0.0265	R1 = 0.0409	R1 = 0.0274	R1 = 0.0327
(>2 σ)	wR2 = 0.0575	wR2 = 0.0698	wR2 = 0.0572	wR2 = 0.0550

Table S2. Bond lengths and angles around metal centers in compounds **1-5**.

	1 , M=Nd	2 , M=Gd	3 , M=Tb	4 , M=Ho	5 , M=Ho
M(III)-N1 (Å)	2.730(5)	2.666(5)	2.660(4)	2.660(4)	2.597(3)
M(III)-N2 (Å)	2.682(5)	2.663(5)	2.641(4)	2.646(3)	2.631(3)
M(III)-N3 (Å)	2.703(6)	2.686(5)	2.651(4)	2.617(4)	2.578(3)
M(III)-N4 (Å)	2.746(5)	2.688(5)	2.668(4)	2.637(4)	2.608(3)
M(III)-O1 (Å)	2.199(5)	2.198(4)	2.146(3)	2.138(3)	2.201(2)
M(III)-O2 (Å)	2.286(4)	2.240(4)	2.226(3)	2.203(3)	2.201(3)
M(III)-O3 (Å)	2.231(4)	2.162(4)	2.170(3)	2.170(3)	2.129(2)
N1-M(III)-N2 (°)	67.33(15)	69.42(16)	69.1(1)	68.12(10)	70.71(9)
N2-M(III)-N3 (°)	69.15(18)	67.80(15)	69.6(1)	70.07(11)	68.92(9)
N3-M(III)-N4 (°)	66.51(16)	67.28(15)	68.6(1)	69.22(11)	69.41(9)
N4-M(III)-N1 (°)	66.03(17)	67.85(15)	67.3(1)	68.29(11)	70.67(9)
O1-M(III)-O2 (°)	96.85(16)	93.98(15)	93.6(1)	94.04(12)	89.88(9)
O2-M(III)-O3 (°)	96.33(16)	95.17(15)	94.8(1)	91.52(11)	91.88(10)
O3-M(III)-O1 (°)	93.01(16)	91.53(16)	86.9(1)	90.33(11)	89.18(9)

Table S3. Coordination geometries of metal centers in compounds **1-5**.

Compound	{N1>N4} plane* RMSD (Å)	{N1>N4} - M(III) (Å) ^a	O1>O3 - M(III) (Å) ^a	{N1>N4} - M(III) - {O1>O3} (Å) ^b	{N1>N4} - M(III) - {O1>O3} (°) ^b	O1>O3 to plane twist angle (°)	{O1>O3} to {N1>N4} plane fold angle (°)
1	0.013	1.6953(11)	1.1624(11)	2.8536(15)	173.76(7)	116(2)	4.29(6)
2	0.012	1.63533(3)	1.18820(2)	2.81630(6)	174.53(5)	124.88(7)	4.47667(4)
3	0.010	1.603(2)	1.2201(18)	2.820(3)	174.62(11)	107(2)	4.49(11)
4	0.010	1.5834(17)	1.2092(17)	2.790(2)	175.41(11)	111(3)	4.27(10)
5	0.014	1.5252(16)	1.2497(16)	2.772(2)	175.06(9)	128(2)	5.37(9)

* MPLN – Olex2⁷, {N1>N4} refers to plane spanned by coordinated N atoms, {O1>O4} to phenolate O-atoms

^a plane centroid to metal

^b plane centroid to metal to plane centroid

Table S4. SHAPE analyses for compounds **1**, **2**, **4** and **5**.

		Shape calculations						
Compounds		HP-7 (D _{7h})	HPY-7 (C _{6v})	PBPY-7 (D _{5h})	COC-7 (C _{3v})	CTPR-7 (C _{2v})	JPBPY-7 (D _{5h})	JETPY-7 (C _{3v})
(1)·CH ₃ OH	Nd1	35.537	19.284	6.224	1.438	1.850	10.594	21.314
(2)·CH ₃ OH	Gd1	35.665	19.162	6.165	1.269	1.630	10.364	21.070
(4)·CH ₃ OH	Ho1	35.382	19.211	6.243	1.243	1.308	10.281	20.781
(5)·2CH ₃ OH	Ho1	35.709	19.090	6.038	0.904	1.647	10.059	20.534

*HP-7 corresponds to an Heptagon, HPY-7 to an Hexagonal pyramid, PBPY-7 to a Pentagonal bipyramid, COC-7 to a Capped octahedron, CTPR-7 to a Capped trigonal prism, JPBPY-7 to a Johnson pentagonal bipyramid J13 and JETPY-7 to a Johnson elongated triangular pyramid J7.

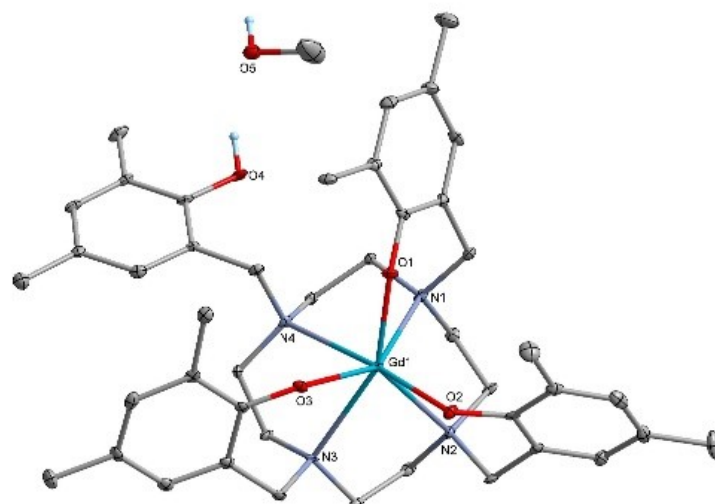
Table S5. Possible hydrogen bonds for compound **1**.

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O4—H4O···O5	0.72(4)	1.99(5)	2.659(3)	154(5)
O5—H5A···O2 ⁱ	0.81(5)	2.01(5)	2.807(3)	171(5)
C10—H10B···O3 ⁱⁱ	0.97	2.56	3.453 (3)	153
C38—H38B···O5 ⁱⁱⁱ	0.97	2.59	3.487 (4)	154
C43—H43B···O4	0.97	2.36	3.047 (3)	128
C45—H45A···O4	0.96	2.59	3.059 (5)	110

Symmetry codes: (i) *x*, *y*+1, *z*; (ii) *x*-1/2, -*y*, *z*; (iii) *x*-1/2, -*y*+1, *z*.

* D = Donor, A = Acceptor.

a)



b)

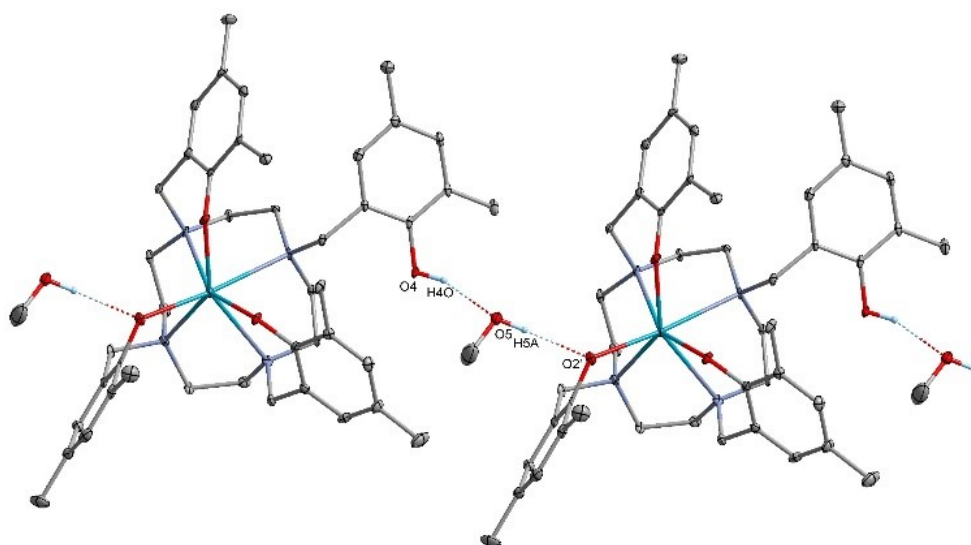
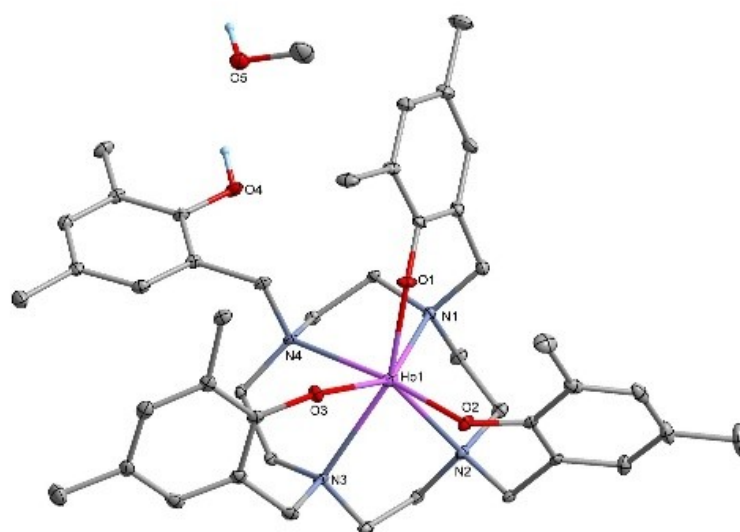


Fig. S15. a) Perspective view and b) Crystal packing of **2**.

a)



b)

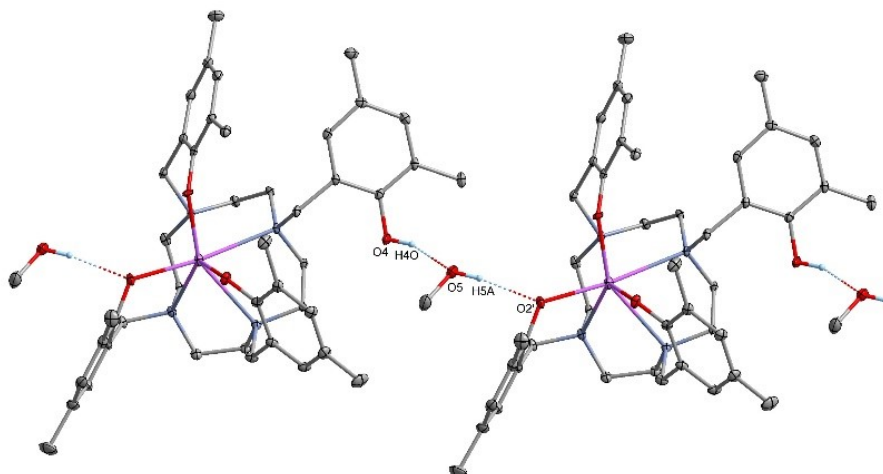
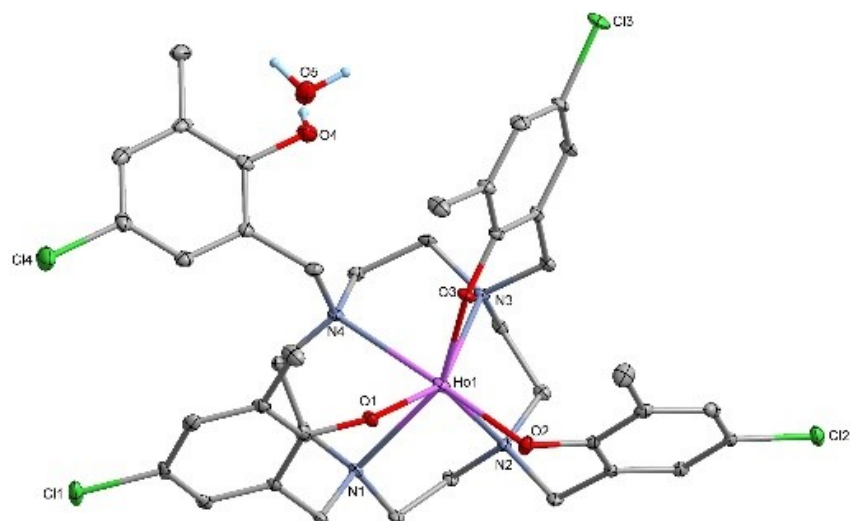


Fig. S16. a) Perspective view and b) Crystal packing of **4**.

a)



b).

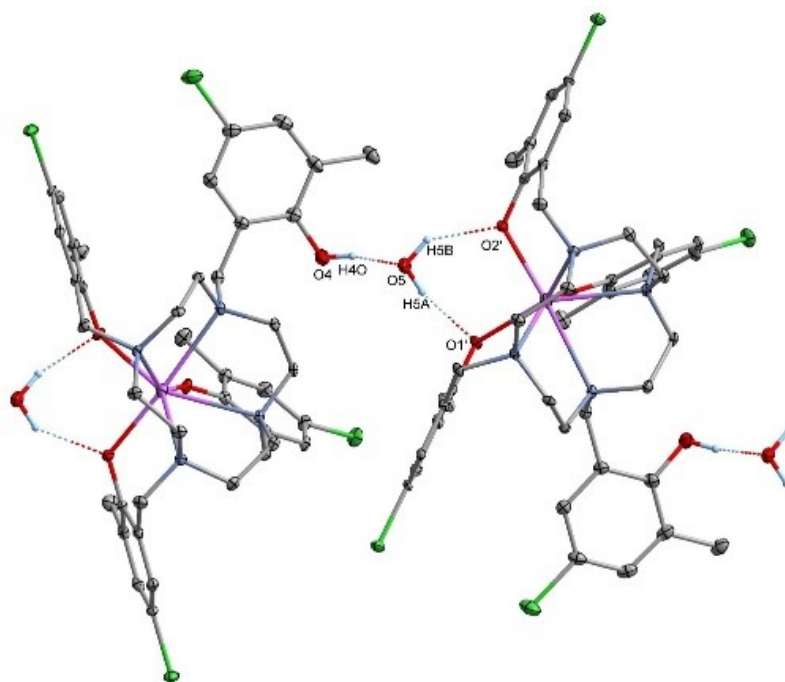


Fig. S17. a) Perspective view and b) Crystal packing of **5**

Table S6. Possible hydrogen bonds for compound **2**.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O4—H4O $\cdots$ O5	0.84	1.85	2.661(6)	160.8
O5—H5A $\cdots$ O2 ⁱ	0.84	1.98	2.821(6)	174.0
C12—H12A $\cdots$ O3 ⁱⁱ	0.99	2.58	3.501(7)	154.6
C11—H11B $\cdots$ O5 ⁱⁱⁱ	0.99	2.55	3.455(8)	152.6
C43—H43B $\cdots$ O4	0.99	2.33	3.048(7)	128.3

Symmetry codes: (i) $x, -1+y, z$; (ii) $1/2+x, 2-y, z$; (iii) $1/2+x, 1-y, z$.

Table S7. Possible hydrogen bonds for compound **4**.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O4—H4O $\cdots$ O5	0.84	1.87	2.659(5)	156
O5—H5A $\cdots$ O2 ⁱⁱ	0.84	1.99	2.820(5)	171
C43—H43A $\cdots$ O4	0.99	2.36	3.085(5)	129
C11—H11A $\cdots$ O5 ⁱ	0.99	2.57	3.441(6)	147
C34—H34B $\cdots$ O3	0.99	2.59	3.332(5)	131

Symmetry codes: (i) $x, y-1, z$; (ii) $x+1/2, -y+1, z$.

Table S8. Possible hydrogen bonds for compound **5**.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O4—H4O $\cdots$ O5	0.84	1.71	2.506(7)	157
O5—H5A $\cdots$ O1 ⁱ	0.87	1.91	2.768(6)	150
O5—H5B $\cdots$ O2 ⁱⁱ	0.87	2.05	2.733(6)	135
C20—H20B $\cdots$ C11 ⁱ	0.99	2.86	3.793(4)	157
C30—H30A $\cdots$ C12 ⁱⁱⁱ	0.99	2.99	3.842(4)	145
C30—H30B $\cdots$ O4	0.99	2.49	3.147(5)	123
C40—H40B $\cdots$ C11 ^{iv}	0.99	2.96	3.681(4)	131
C7—H7 $\cdots$ C14 ^{iv}	0.95	2.94	3.841(4)	158
C31—H31A $\cdots$ O1	0.99	2.66	3.374(5)	130
C31—H31B $\cdots$ O3	0.99	2.62	3.122(5)	112
C35—H35 $\cdots$ O7	0.95	2.52	3.389(7)	152

Symmetry codes: (i) $1/2+x, 1/2-y, 1/2+z$; (ii) $x+1/2, -y+1/2, z+1/2$; (iii) $-x+3/2, y-1/2, -z+1/2$; (iv) $-x+1, -y, -z$.

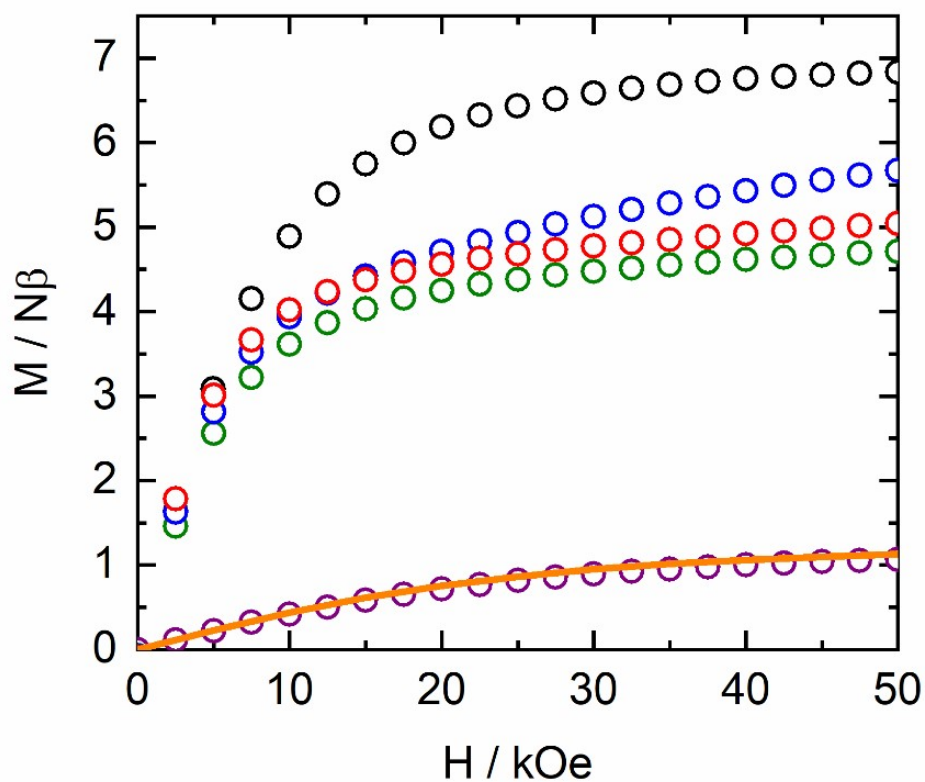


Fig. S18. Field dependence of the magnetization at 2 K in the 0-50 kOe field range for **1** (purple), **2** (black), **3** (green), **4** (red) and **5** (blue). Full orange line corresponds to calculated data.

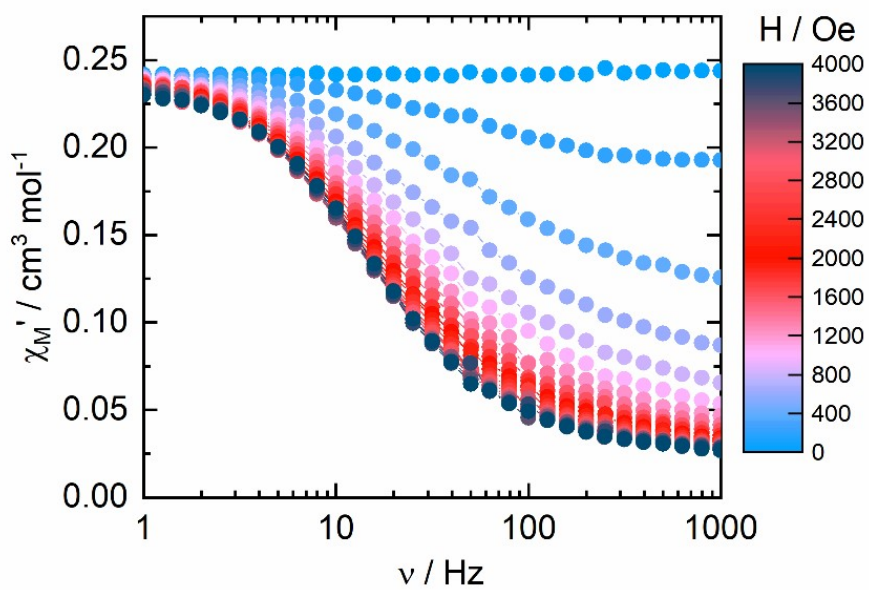


Fig. S19. Field dependence of the in-phase component of the magnetic susceptibility at 2 K in the 0-4000 Oe field range for **1**.

Extended Debye model (Eq. S1).

$$\chi_M' = \chi_S + (\chi_T - \chi_S) \frac{1 + (\omega\tau)^{1-\alpha} \sin\left(\alpha \frac{\pi}{2}\right)}{1 + 2(\omega\tau)^{1-\alpha} \sin\left(\alpha \frac{\pi}{2}\right) + (\omega\tau)^{2-2\alpha}}$$

$$\chi_M'' = (\chi_T - \chi_S) \frac{(\omega\tau)^{1-\alpha} \cos\left(\alpha \frac{\pi}{2}\right)}{1 + 2(\omega\tau)^{1-\alpha} \sin\left(\alpha \frac{\pi}{2}\right) + (\omega\tau)^{2-2\alpha}}$$

With χ_T the isothermal susceptibility, χ_S the adiabatic susceptibility, τ the relaxation time and α an empiric parameter which describe the distribution of the relaxation time. For SMM with only one relaxation time, α is close to zero. The extended Debye model was applied to fit simultaneously the experimental variations of χ_M' and χ_M'' with the frequency ν of the oscillating field ($\omega = 2\pi\nu$). Typically, only the temperatures for which a maximum on the χ'' vs. ν curves, have been considered. The best fitted parameters τ , α , χ_T , χ_S are listed in Tables S6 and S7 with the coefficient of determination R^2 .

Table S7. Best fitted parameters (χ_T , χ_S , τ and α) with the extended Debye model for compound **1** at 2 K in the field range 0-4000 Oe.

H / Oe	χ_S / cm ³ mol ⁻¹	χ_T / cm ³ mol ⁻¹	τ / s	α	R^2
200	0.18994	0.24233	0.00302	0.25836	0.99994
400	0.11702	0.24619	0.0035	0.32863	0.99991
600	0.07726	0.24999	0.00451	0.33621	0.99968
800	0.05789	0.25158	0.00555	0.32487	0.99935
1000	0.04708	0.2522	0.00629	0.31275	0.99919
1200	0.04101	0.25051	0.00695	0.29346	0.99899
1400	0.03719	0.25048	0.00754	0.28326	0.99925
1600	0.03464	0.25007	0.00817	0.2707	0.99929
1800	0.03284	0.24914	0.00873	0.25988	0.99936
2000	0.03161	0.24813	0.00926	0.25139	0.99925
2200	0.03082	0.24808	0.00958	0.24732	0.99918
2400	0.02937	0.24594	0.00986	0.22955	0.99928
2600	0.02935	0.24585	0.01023	0.22911	0.99916
2800	0.02832	0.24395	0.01024	0.22144	0.99936
3000	0.02761	0.24242	0.01024	0.21213	0.99944
3200	0.02855	0.24659	0.01012	0.21801	0.99925
3400	0.02803	0.24475	0.01001	0.20695	0.99953
3600	0.02704	0.2425	0.00969	0.19424	0.99962
3800	0.02757	0.24106	0.00942	0.19016	0.99914
4000	0.02681	0.23857	0.00901	0.17753	0.9996

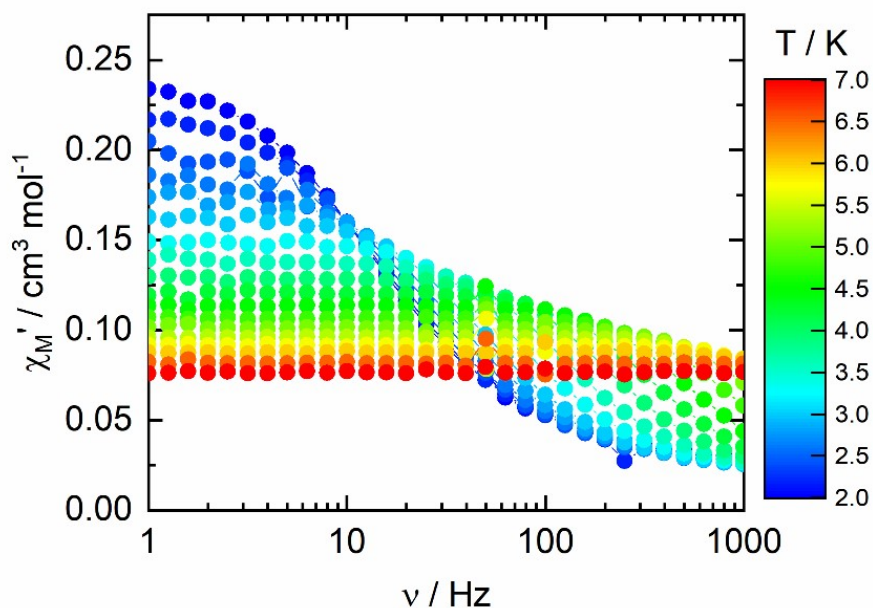


Fig. S20. Temperature dependence of the in-phase component of the magnetic susceptibility under an applied field of 3200 Oe in the 2-7 K temperature range for **1**.

Table S8. Best fitted parameters (χ_T , χ_S , τ and α) with the extended Debye model for compound **1** at 3200 Oe in the temperature range 2-5.1 K.

T / K	$\chi_S / \text{cm}^3 \text{mol}^{-1}$	$\chi_T / \text{cm}^3 \text{mol}^{-1}$	τ / s	α	R^2
2	0.02793	0.24652	0.01084	0.21976	0.99942
2.2	0.02698	0.22594	0.00891	0.19813	0.99923
2.4	0.02593	0.20262	0.00669	0.17597	0.99761
2.6	0.02569	0.18965	0.00528	0.15619	0.99827
2.8	0.02376	0.1782	0.00405	0.15263	0.99905
3	0.02367	0.16505	0.00297	0.11943	0.99941
3.3	0.0235	0.15027	0.00185	0.08438	0.99937
3.6	0.023	0.13938	0.00118	0.07371	0.99973
3.9	0.02147	0.12906	7.20313E-4	0.06354	0.99919
4.2	0.02243	0.1208	4.59435E-4	0.03693	0.99971
4.5	0.01997	0.11403	2.89834E-4	0.0275	0.99878
4.8	0.02303	0.10694	1.91928E-4	0.01937	0.99933
5.1	0.02394	0.10181	1.29749E-4	0.01017	0.99886

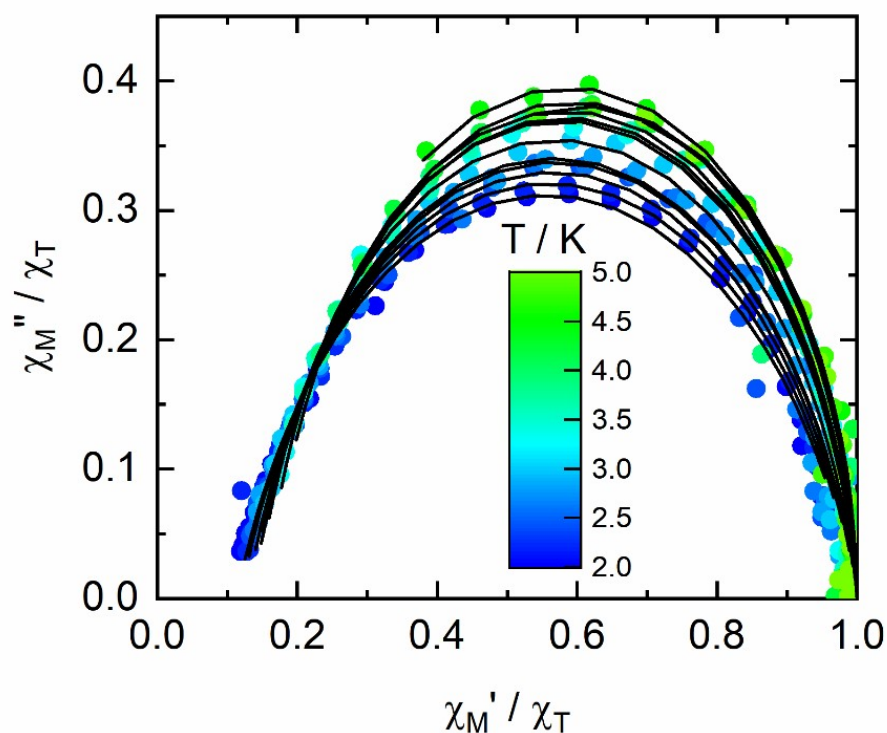


Fig. S21. Normalized Argand plot for **1** in the 0-5.1 K temperature range.

Table S9. SINGLE_ANISO computed energy barrier and angles between the g_{zz} axes of KDs.

Energy (cm ⁻¹)	g_{xx}	g_{yy}	g_{zz}	θ
0.0	1.09	1.61	3.50	-
91.0	2.74	2.63	0.09	52.8
174.4	0.31	1.56	3.30	42.2
268.6	0.41	0.57	5.68	105.9
503.5	1.84	2.42	3.62	75.9

Table S10. SINGLE_ANISO computed Crystal-Field parameters B_k^q for the complex **1**.

k	q	B_k^q
2	-2	-1.0899
2	-1	0.0465
2	0	-0.5075
2	1	-1.6063
2	2	0.4716
4	-4	-0.0564
4	-3	0.2230
4	-2	0.0307
4	-1	0.0613

4	0	0.0103
4	1	-0.0468
4	2	0.0456
4	3	0.1004
4	4	-0.0462

Table S11. Decomposition of the RASSI wave functions corresponding to the lowest atomic multiplet $J=9/2$ in wave functions with definite projection of the total moment on the quantization axis for the states of the complex **1**.

m_J	ab initio state 1		ab initio state 2	
	Real	Imaginary	Real	Imaginary
-9/2	-0.118526	0.451748	0.105851	-0.403439
-7/2	0.160579	-0.093461	0.167131	0.160105
-5/2	-0.143755	-0.411600	0.061136	0.283856
-3/2	0.100695	-0.087969	-0.070898	-0.059466
-1/2	0.199503	-0.164648	-0.382028	-0.078312
1/2	0.021203	-0.389395	0.209888	-0.151187
3/2	0.039526	0.083668	-0.110644	0.075074
5/2	0.259048	0.131172	0.361642	0.243505
7/2	-0.112449	-0.202291	-0.131153	0.131603
9/2	-0.417094	0.000000	-0.467038	0.000000
	ab initio state 3		ab initio state 4	
-9/2	0.065050	0.123192	-0.143071	-0.270948
-7/2	0.068861	-0.037134	-0.400359	-0.322437
-5/2	-0.046484	0.191334	0.074343	-0.366326
-3/2	-0.262170	0.392669	-0.124652	0.225656
-1/2	0.131079	-0.022509	-0.263676	-0.247491
1/2	-0.341974	-0.117603	-0.041302	-0.126422
3/2	-0.141340	0.215596	0.224815	-0.415187
5/2	-0.289225	0.236792	-0.147489	0.130447
7/2	0.472071	0.203474	-0.000683	0.078233
9/2	-0.306402	0.000000	-0.139312	0.000000

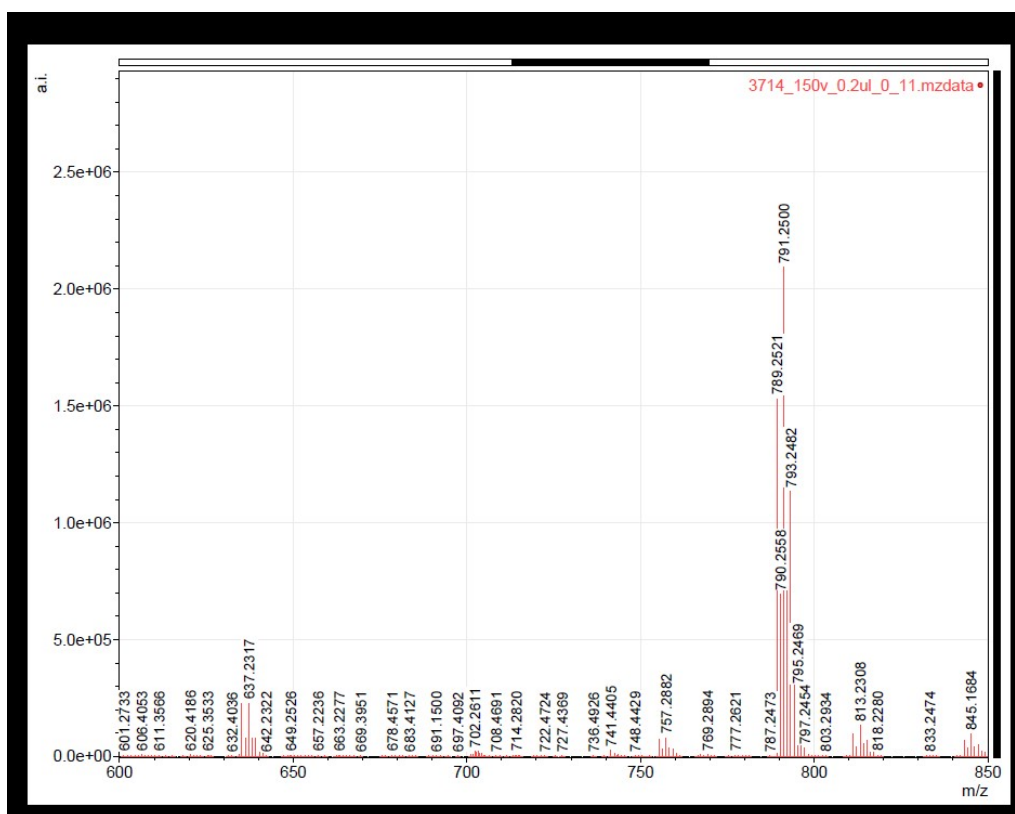


Fig. S22. ESI-MS of $H_4L^{Me,Cl}$ in MeOH

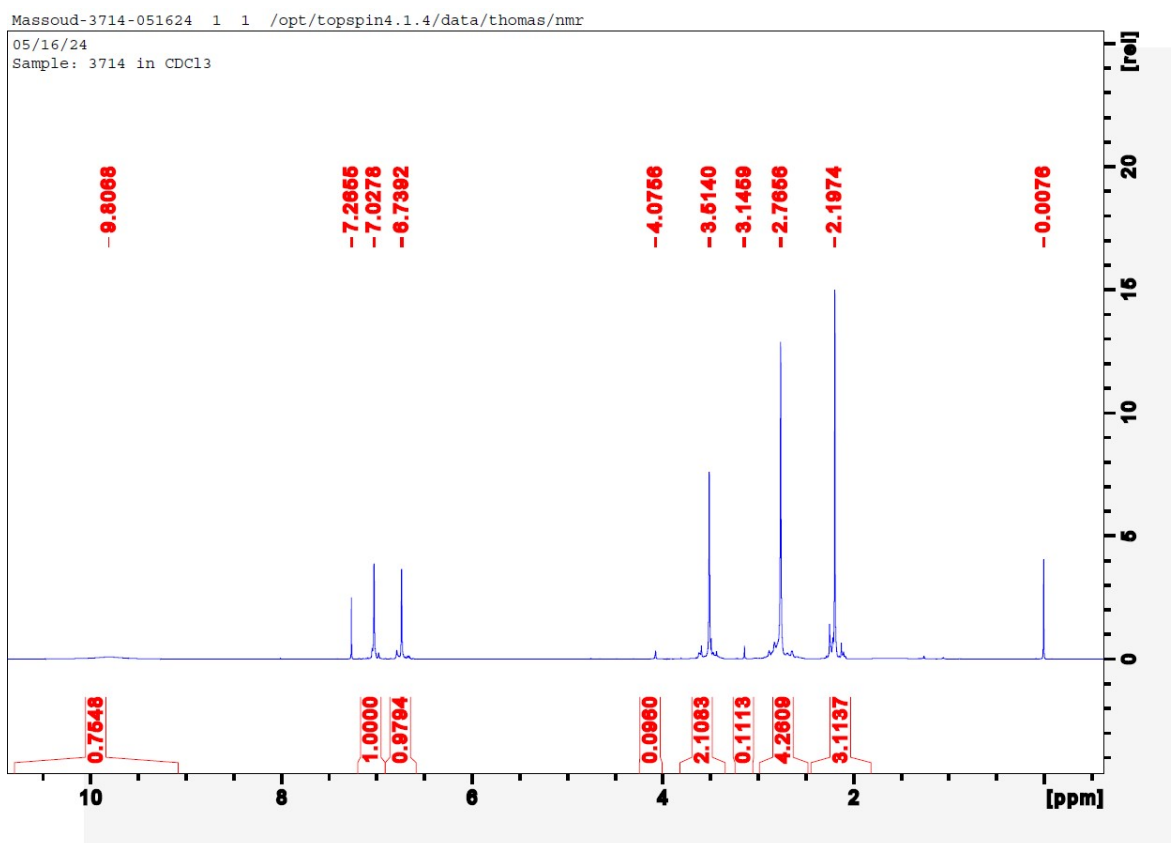


Fig. S23. ^1H NMR of $\text{H}_4\text{L}^{\text{Me,Cl}}$ in CDCl_3