

Supplementary Information

Photostability challenges of catechol-derived LMCT complexes on TiO₂ for visible-light-driven photocatalytic air purification

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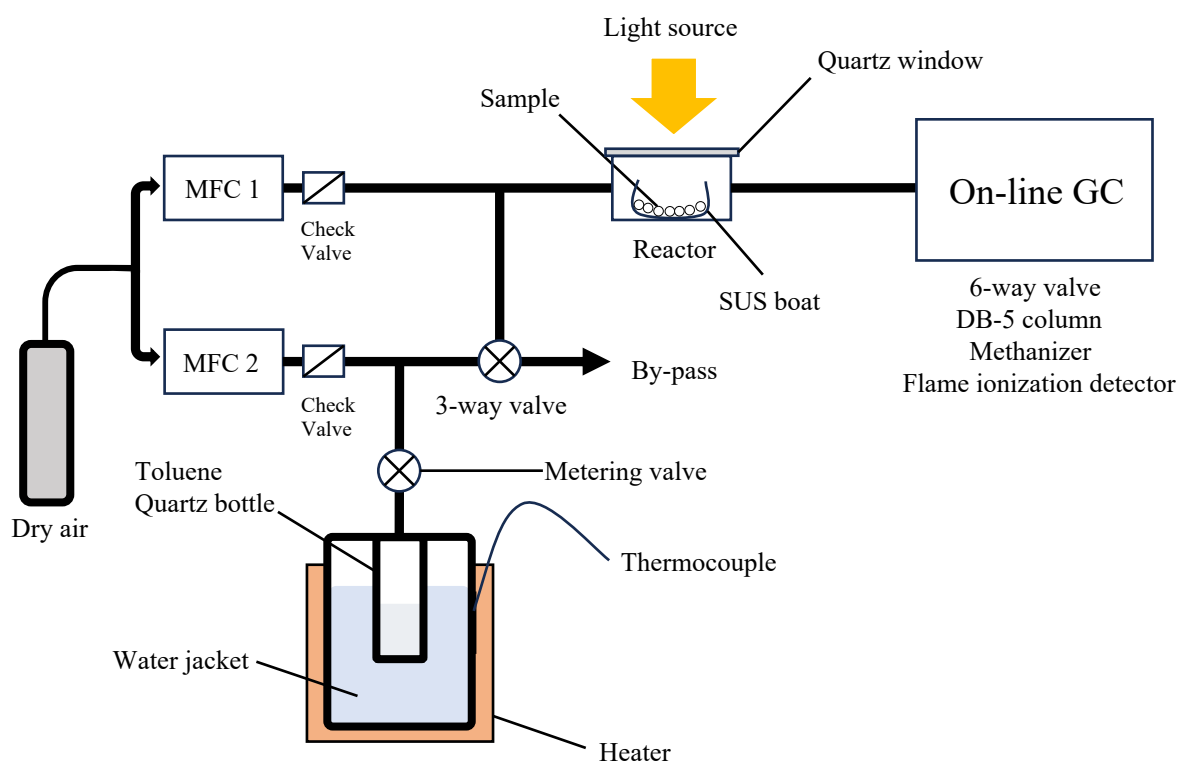


Figure S1. The experimental set-up for photostability and photocatalytic activity measurements.

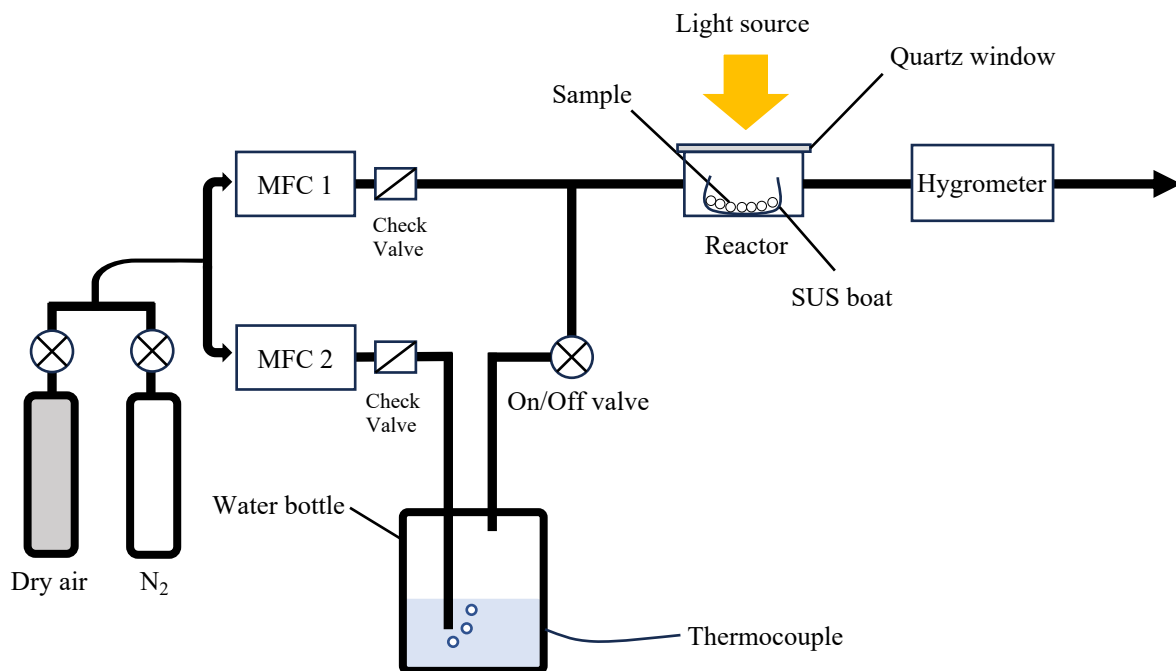


Figure S2. The experimental set-up for photostability tests under various conditions (dry air, humid air, dry N₂, and humid N₂).

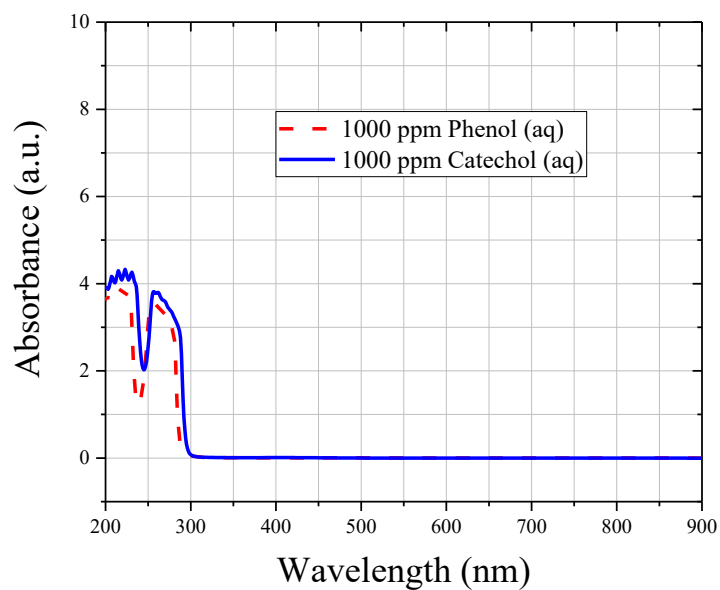


Figure S3. UV-vis absorption spectra of phenol and catechol solutions (1000 ppm).

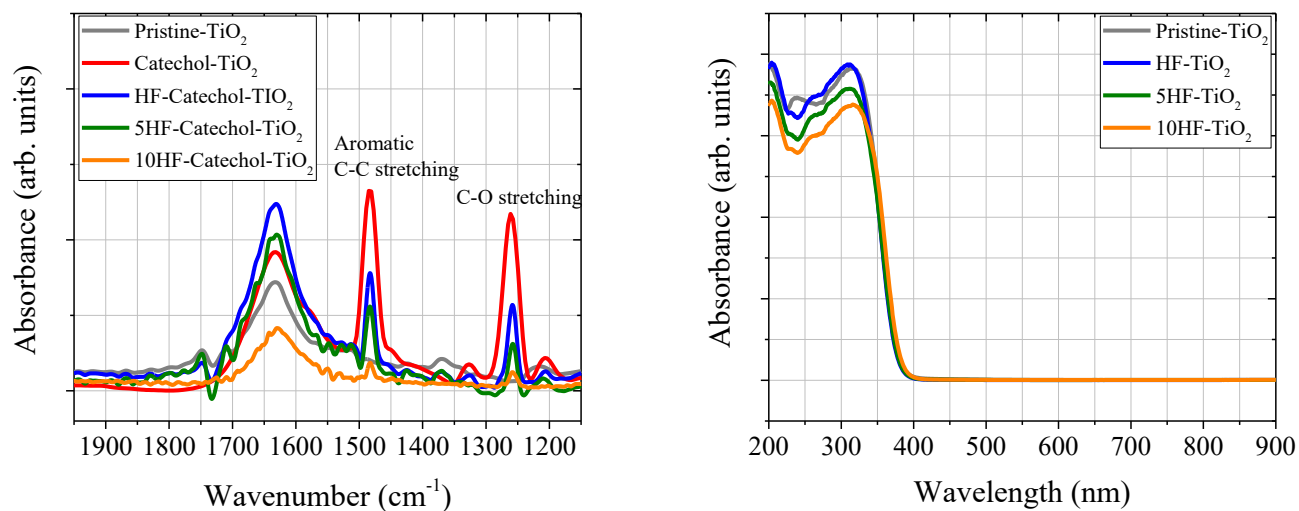


Figure S4. FT-IR spectra of pristine-TiO₂, catechol-TiO₂, HF-Catechol-TiO₂, 5HF-Catechol-TiO₂, and 10HF-Catechol-TiO₂ (Left) and UV-DRS spectra of pristine-TiO₂, HF-TiO₂, 5HF-TiO₂, and 10HF-TiO₂ (Right).

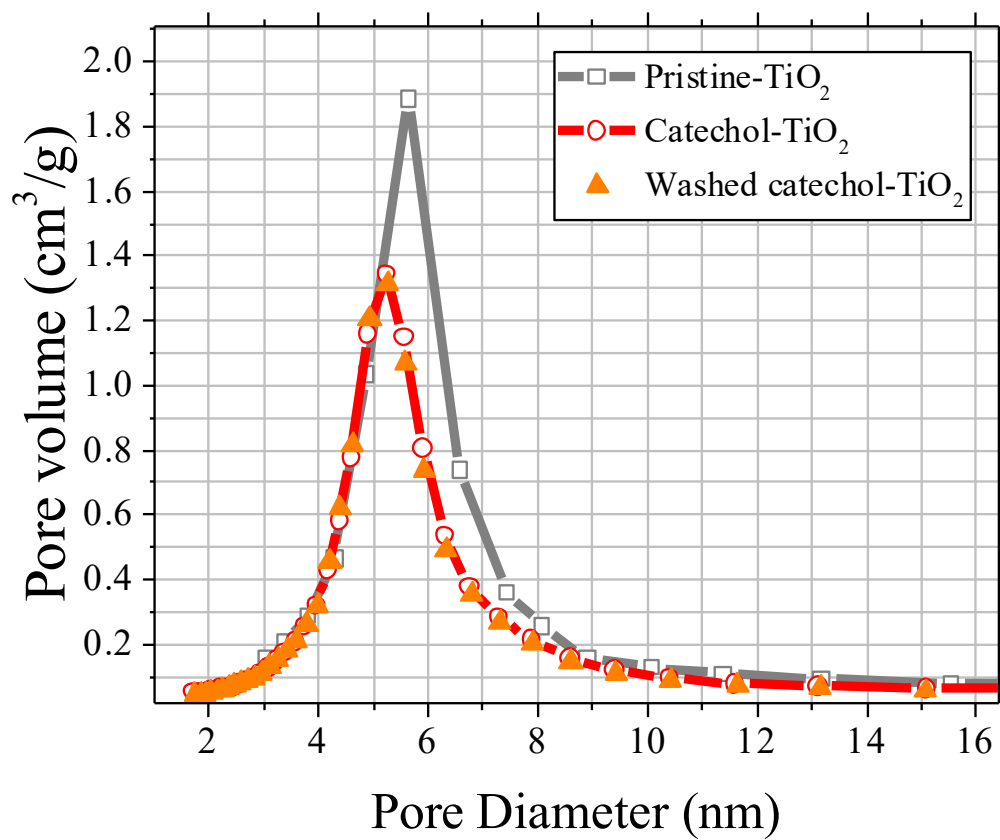


Figure S5. The BJH pore size distributions of pristine-TiO₂, catechol-TiO₂, and washed catechol-TiO₂.

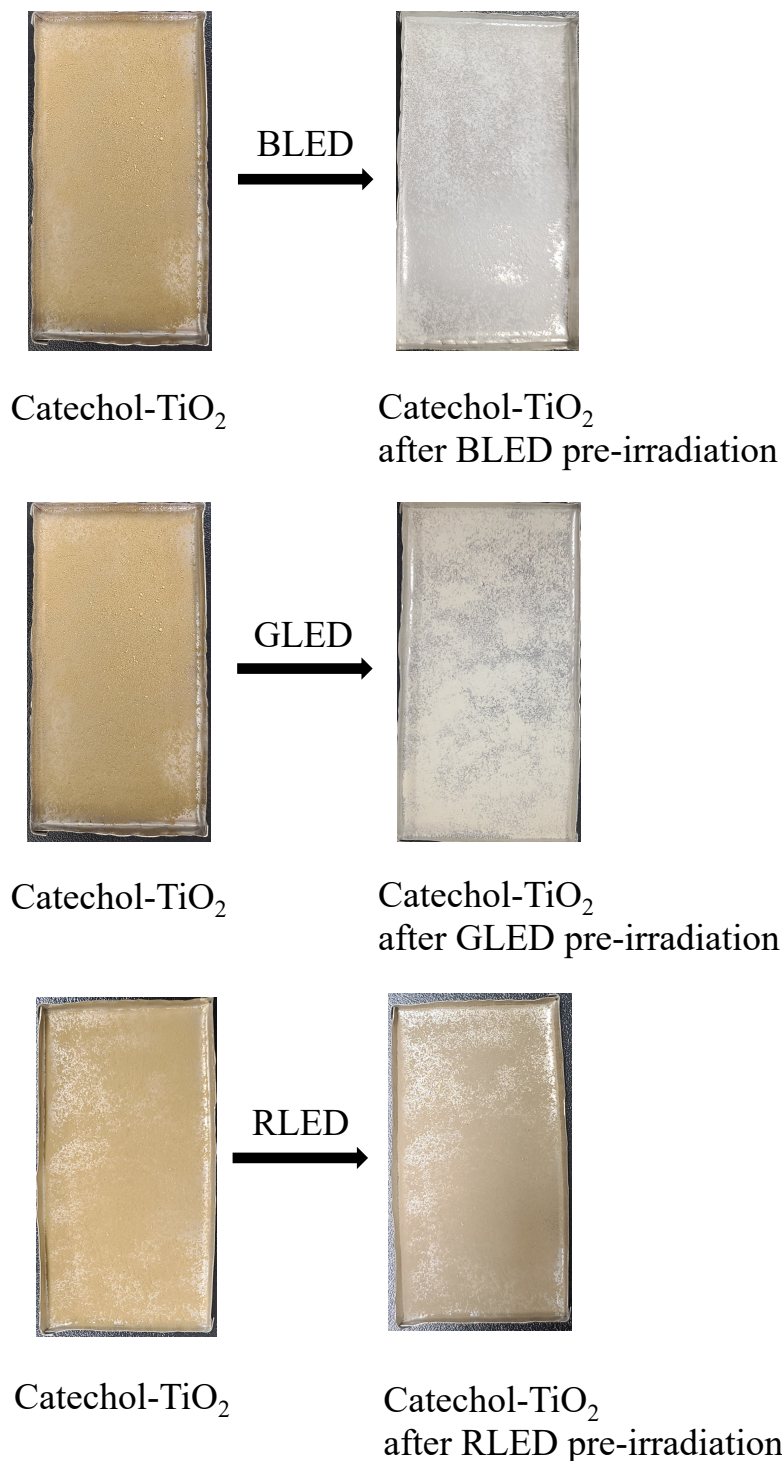


Figure S6. The photos of catechol-TiO₂ taken before and after 6000 minutes of pre-irradiation of blue (top), green (middle), and red (bottom) lights, showing the changes of sample colors after the pre-irradiation.

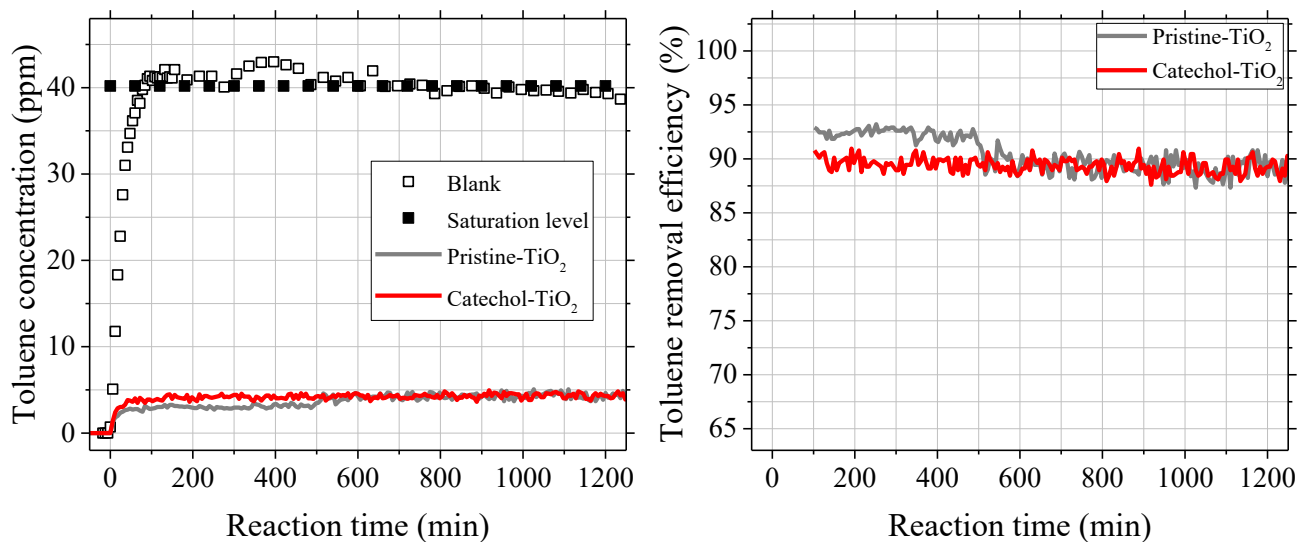


Figure S7. The changes of toluene concentration (Left) and toluene removal efficiency% (Right) in the absence of TiO₂ (blank) and in the presence of pristine-TiO₂ and catechol-TiO₂ under blue light.

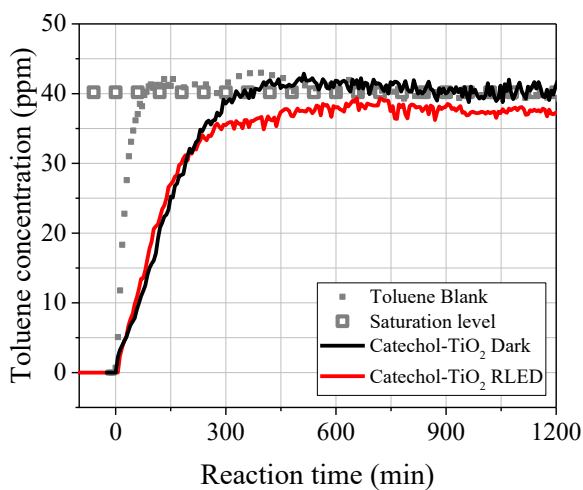
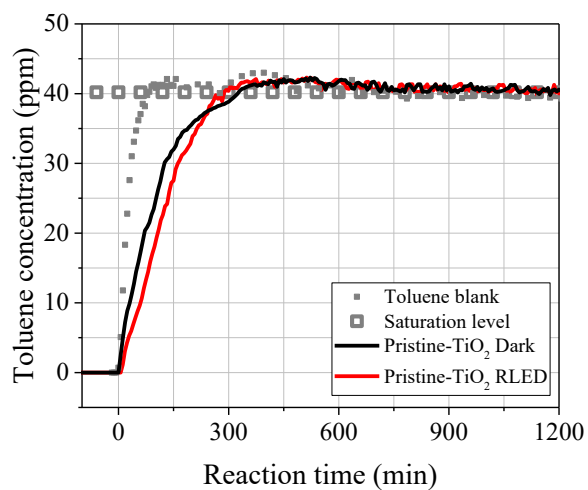


Figure S8. The changes of toluene concentration in the absence of TiO₂ (Toluene blank) and in the presence of pristine-TiO₂ (Left) and catechol-TiO₂ (Right) under dark and red light.