

Supplementary article

Design of a Nitrogen-Rich Perylene-Triazine Porous Organic Polymer for Iodine and CO₂ Adsorption

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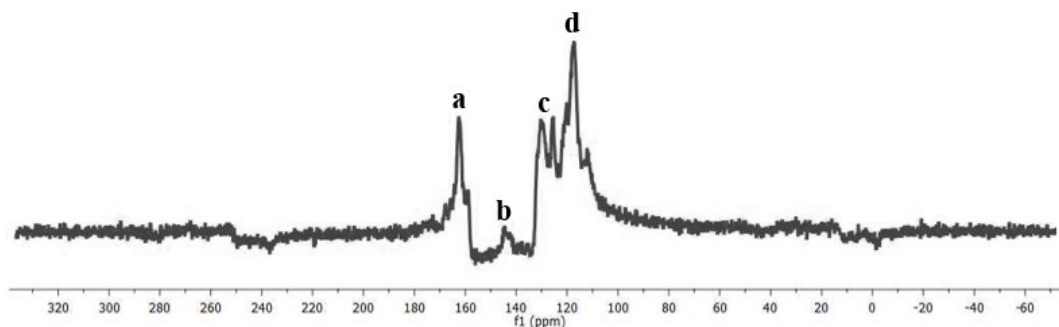


Fig. S1. ^{13}C CP-MAS NMR spectra of PY-TRI

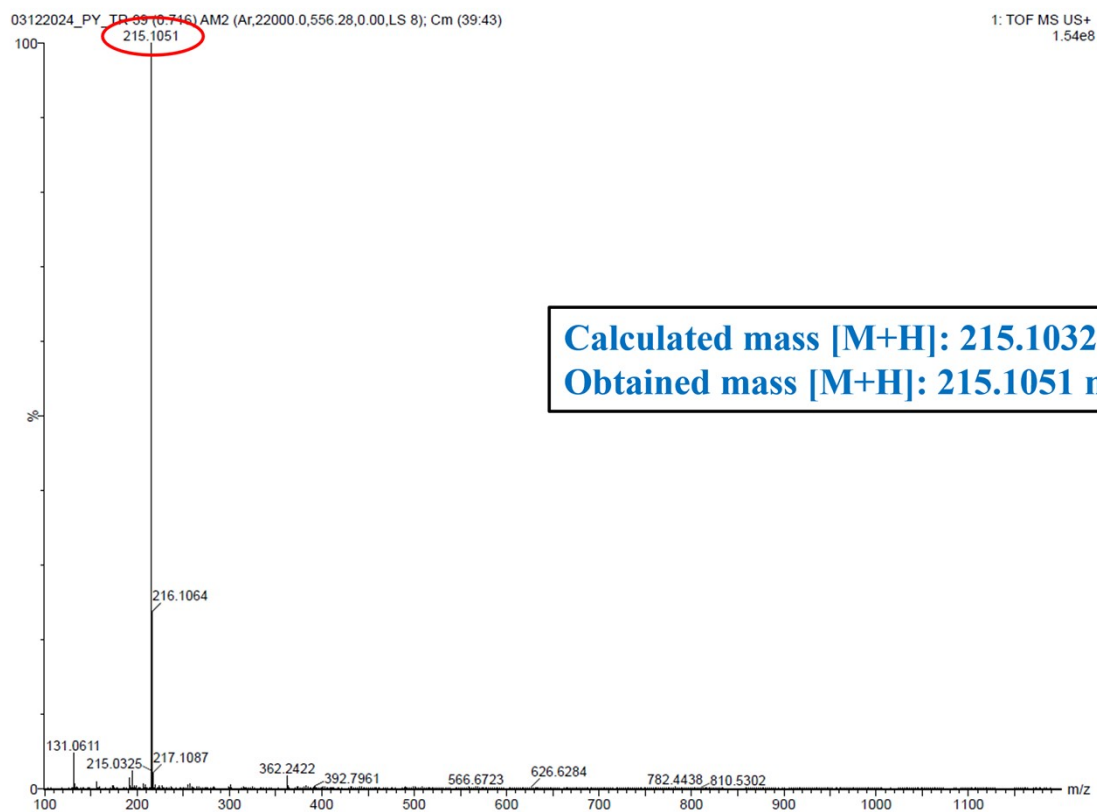


Fig. S2. HRMS spectra of compound PY-TRI

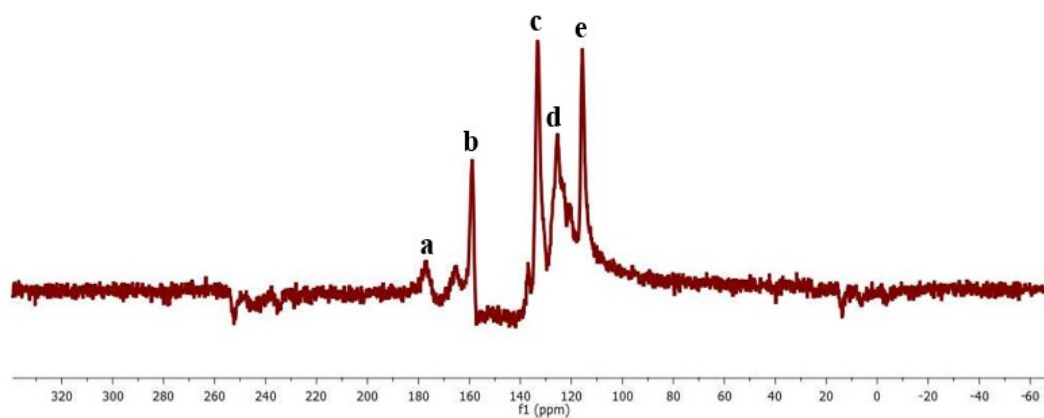


Fig. S3. ^{13}C CP-MAS NMR spectra of PYTP

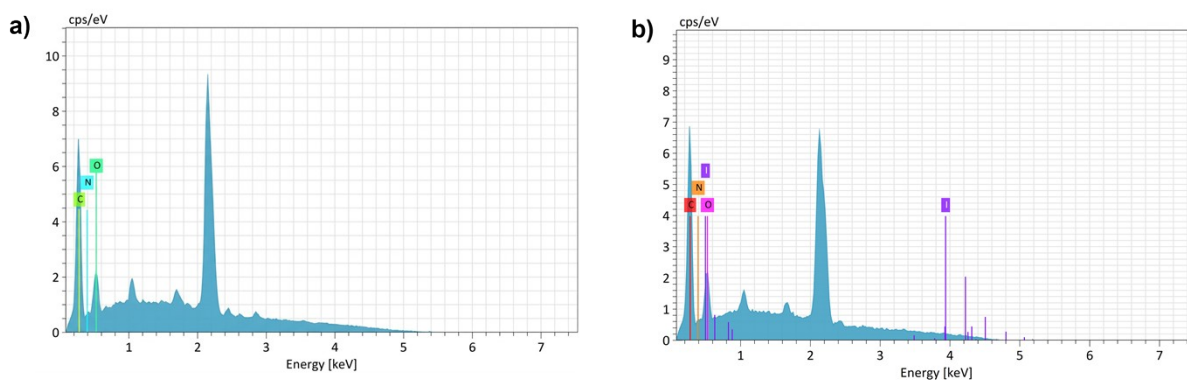


Fig. S4. EDX spectra of Polymer a) PYTP and b) PYTP-I₂

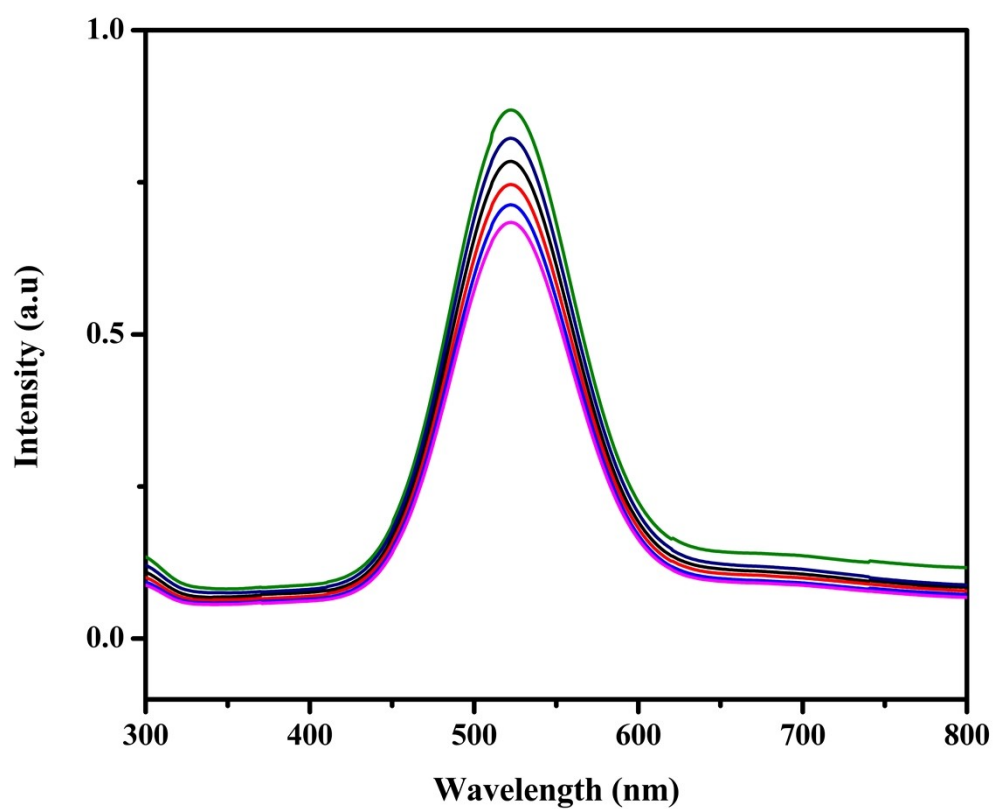


Fig. S5. UV-Vis spectra for PY-TRI

Table S1. Comparison of iodine adsorption performance of PYTP and other reported materials

S.No.	Compounds	$S_{\text{BET}}(\text{m}^2\text{g}^{-1})$	Iodine uptake (wt%)	Reference
1	This work	135	355	-
2	Tfp-Td COF	144.5	300	1
3	PAOP-4	209.9	108	2
4	P-TzTz	137	326	3

5	CTF-CAR	277	286	4
6	TTPT	315.5	177	5
7	NOP-54	1178	202	6
8	HCMP-1	430	159	7
9	Pha-HCOPs	217.31	131	8
10	HCP	584	460	9
11	PT-POP	475	340	10

Table S2. Comparison of CO₂ adsorption performance of PYTP and other reported materials

S.No.	Compounds	Temperature T (K)	CO ₂ uptake cm ³ (cm ⁻³)	Reference
1	This work	273	50	-
2	UTSA-30a	298	44.9	11
3	Cu-F-pymo	273	51.7	12
4	JUC-199	273	35	13
5	NTU-55	273	63	14
6	NTU-67	298	58	15
7	MOF-808-ARG	298	50.3	16
8	CD-MOF-1	273	56	17
9	CD-MOF-2	273	68.2	17
10	Zn-DPNA	273	72	18

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