

A Coal-Derived Carbon Fiber Scaffold with Superior Mechanical Robustness for Ultra-Stable Lithium Metal Anodes

Xian Lu ^a, Qingtao Ma^{a*}, Xiaoquan Zhu^{b*}, Mengjiao Xu^a, Nannan Guo^a, Lili Ai^a, Changyu Leng^a,
Luxiang Wang^{a*}

^aState Key Laboratory of Chemistry and Utilization of Carbon-Based Energy Resources, College of Chemistry, Xinjiang University, Urumqi 830017, Xinjiang, PR China

^bCollege of Textiles and Clothing, Xinjiang University, Urumqi 830017, Xinjiang, PR China

*Corresponding authors

E-mail: qingtaoma@xju.edu.cn; zhuxiaoquan@xju.edu.cn; wangluxiangxju@163.com

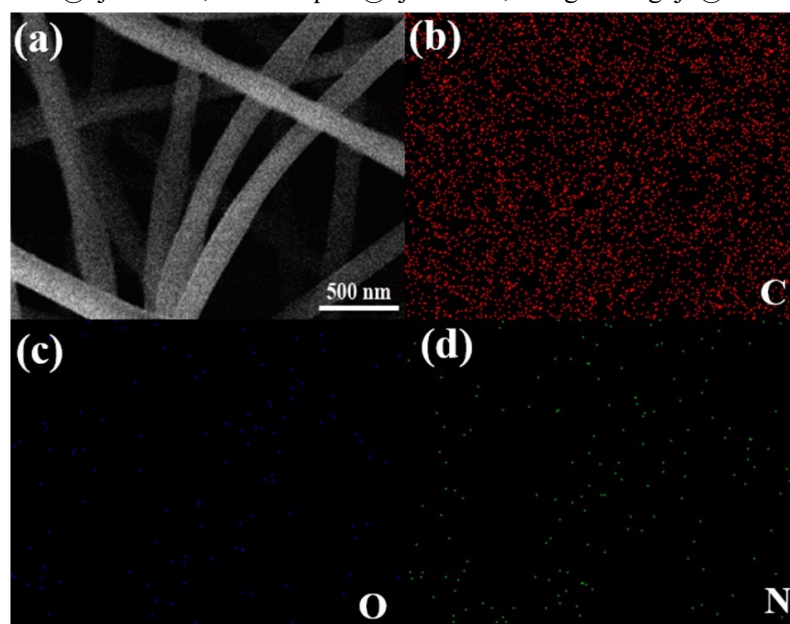


Fig. S1. (a) SEM plots and (b-d) corresponding elemental mappings of CBCNFs.

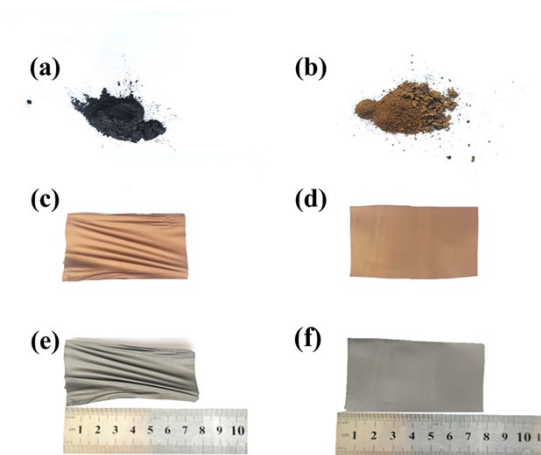


Fig. S2. Digital photograph of (a) raw coal, (b) oxidized coal, (c) CNFs and (d) CBCNFs after pre-oxidation and (e) CNFs and (f) CBCNFs after carbonation.

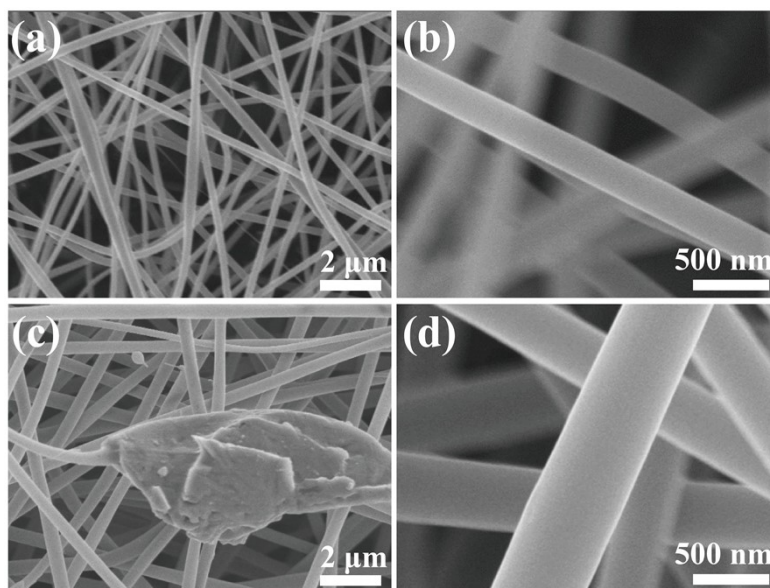


Fig. S3. Scanning electron microscope images of CBCNFs-1 (a-b) and CBCNFs-2 (c-d)

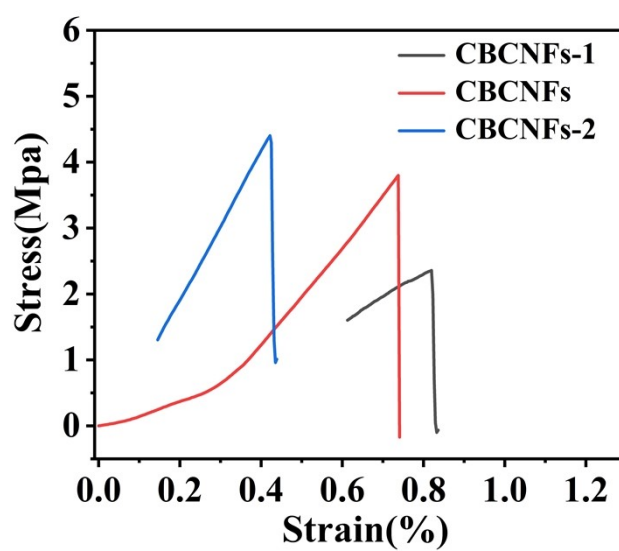


Fig. S4. Tensile strength of CBCNFs-1、CBCNFs and CBCNFs-2.

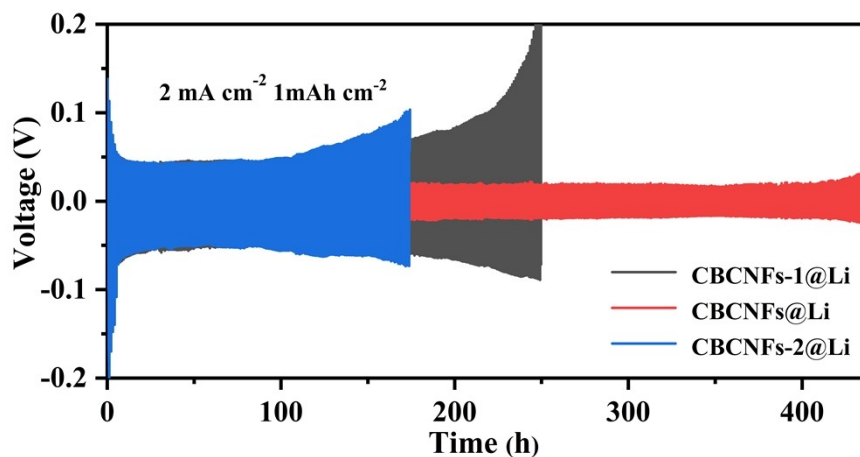


Fig. S5. The Voltage-time curves of CBCNFs-1@Li, CBCNFs@Li, and CBCNFs-2@Li symmetric batteries at a current density of 2 mA cm^{-2} and an area capacity of 1 mAh cm^{-2} .

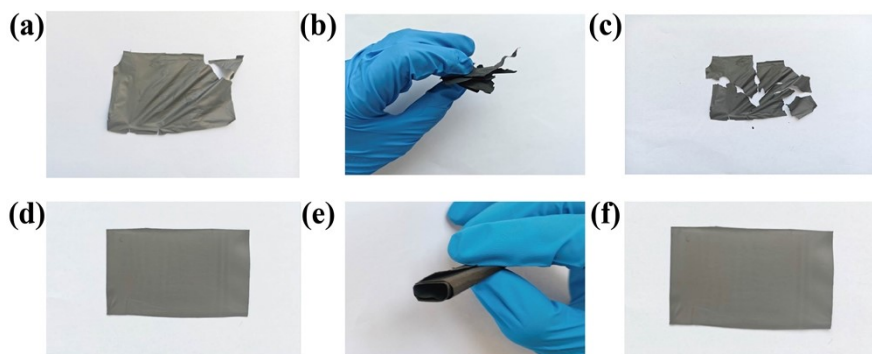


Fig. S6. Flexibility tests of (a-c) CNFs, (d-f) CBCNFs.

Table S1. Specific surface area and pore structure parameters of the samples.

Samples	^a S _{BET} m ² g ⁻¹	^b V _{total} cm ³ g ⁻¹	^c V _{micro} cm ³ g ⁻¹	V _{micro} /V _{total} %	^e D _{ap} nm
CNFs	436.27	0.189	0.124	65.61	1.70
CBCNFs	551.25	0.187	0.154	82.35	1.36

Table S2. summary of elemental content analysis of XPS spectra of prepared samples.

Sample	C 1s (at%)	N 1s (at%)	O 1s (at%)
CNFs	85.13	10.33	4.54
CBCNFs	86.21	5.72	7.75

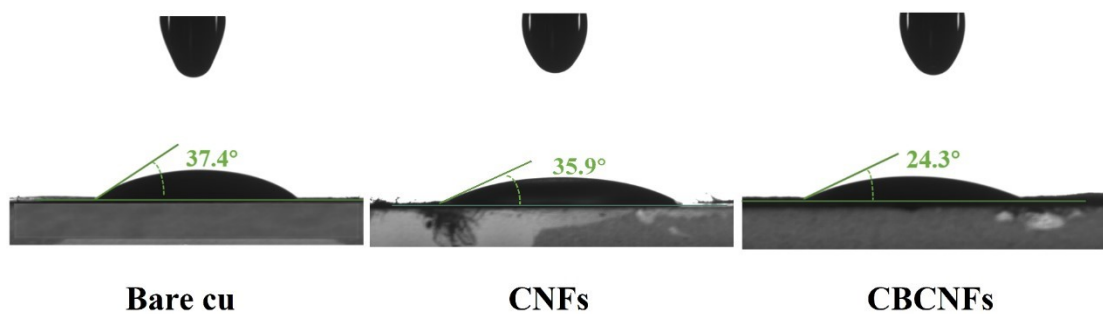


Fig. S7. Contact angle of ether-based electrolyte droplets on (a) bare copper, (b) CNFs, and (c) CBCNFs.

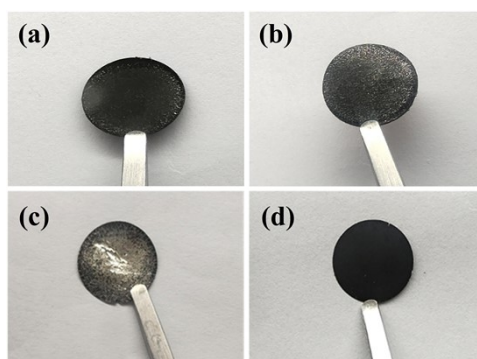


Fig. S8. Digital photograph of CBCNFs at different lithium plating capacities: (a) 5 mAh cm⁻², (b) 8 mAh cm⁻², (c) 10 mAh cm⁻², (d) after being charged to 1 V.

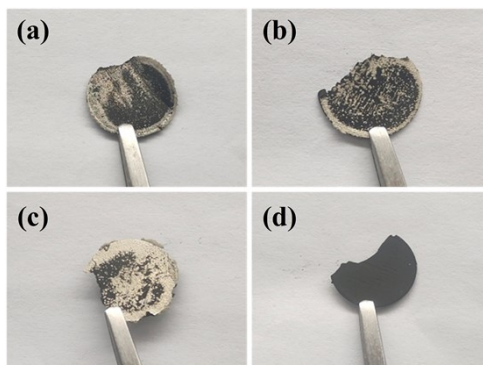


Fig. S9. Digital photograph of CNFs at different lithium plating capacities: (a) 5 mAh cm⁻², (b) 8 mAh cm⁻², (c) 10 mAh cm⁻², (d) after being charged to 1 V.

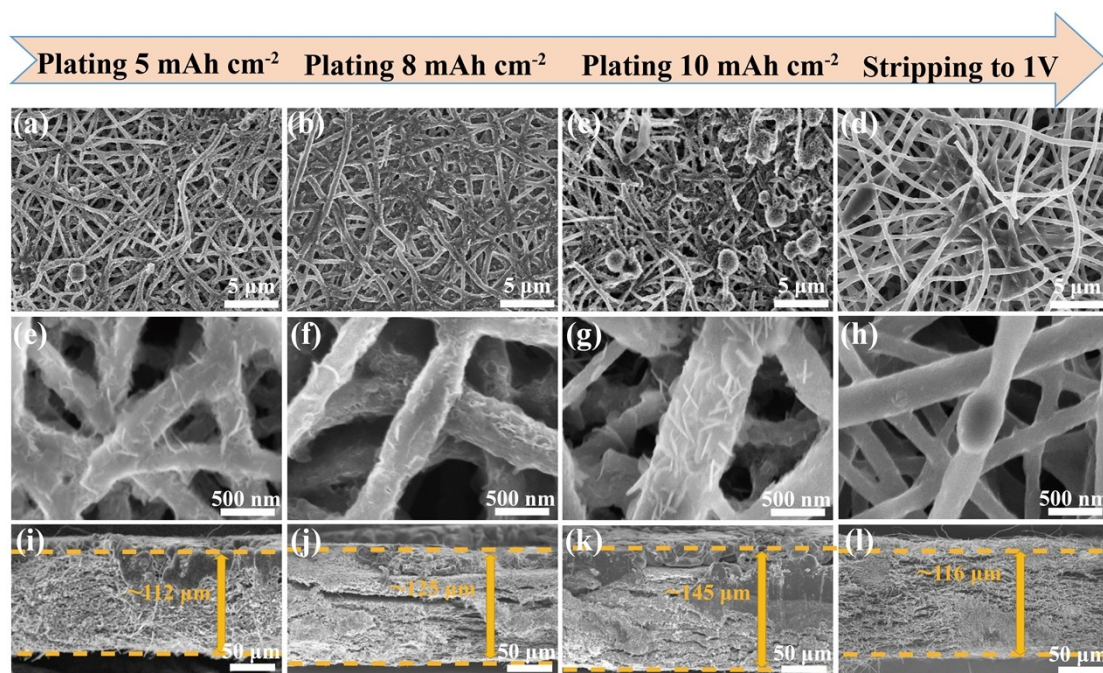


Fig. S10. The SEM images of CNFs after different lithium depositions are (a, e) 5 mAh cm^{-2} , (b, f) 8 mAh cm^{-2} , (c, g) 10 mAh cm^{-2} and (d, h) corresponding cross-sectional images after charging to 1 V (i-l).

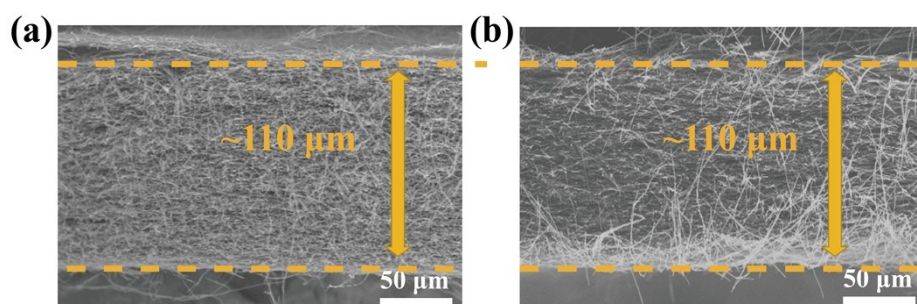


Fig. S11. Cross-sectional SEM images of (a) CBCNFs and (b) CNFs before electroplating.

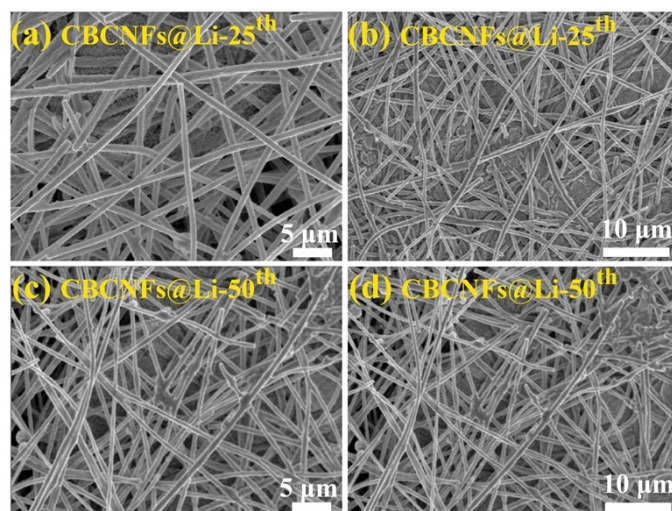


Fig. S12. SEM images of CBCNFs following 25 cycles (a, b) and 50 cycles (c, d) under a cycling condition of 1 mA cm^{-2} and 1 mAh cm^{-2} .

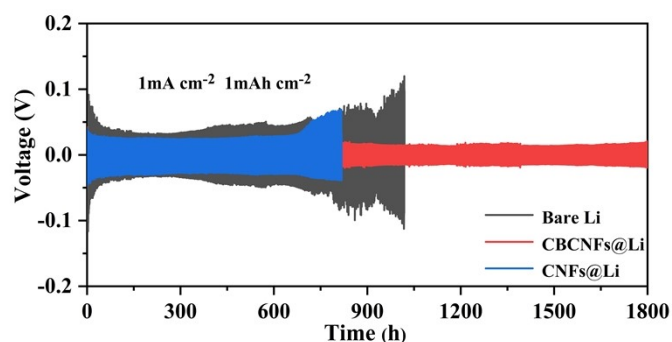


Fig. S13. The deposition/stripping curves of bare lithium, CBCNFs@Li, and CNFs@Li symmetric batteries at a current density of 1 mA cm^{-2} and an area capacity of 1 mAh cm^{-2} .

Table S3 Fitted values of resistances (R_{ct}) using the model in Fig. 6d and Fig. 6e

Material	Fresh R_{ct} (ohm)	25 th R_{ct} (ohm)	50 th R_{ct} (ohm)	100 th R_{ct} (ohm)
Bare Li	115.20	19.52	9.72	6.06
CBCNFs@Li	79.52	13.91	6.93	3.33

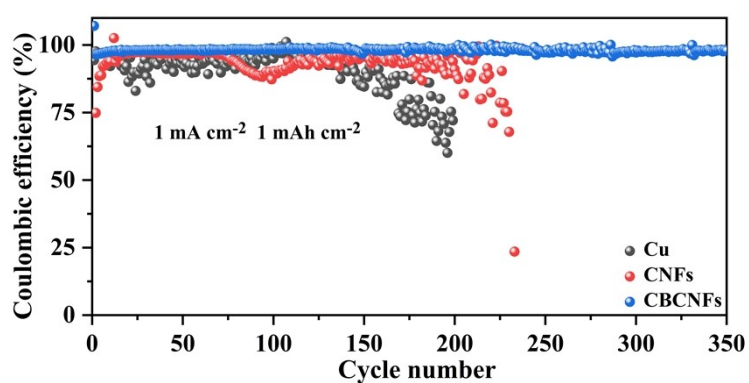


Fig. S14. CE of Cu||Li, CNFs||Li and CBCNFs||Li batteries at current of 1 mA cm^{-2} and capacity of 1 mAh cm^{-2} .

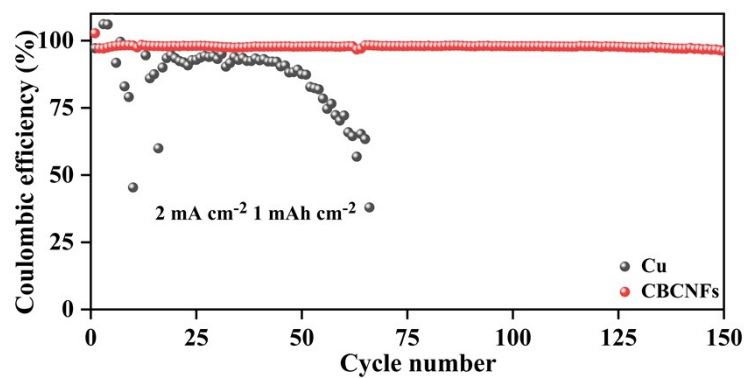


Fig. S15. CE of Cu||Li and CBCNFs||Li batteries at current of 2 mA cm^{-2} and capacity of 1 mAh cm^{-2} .

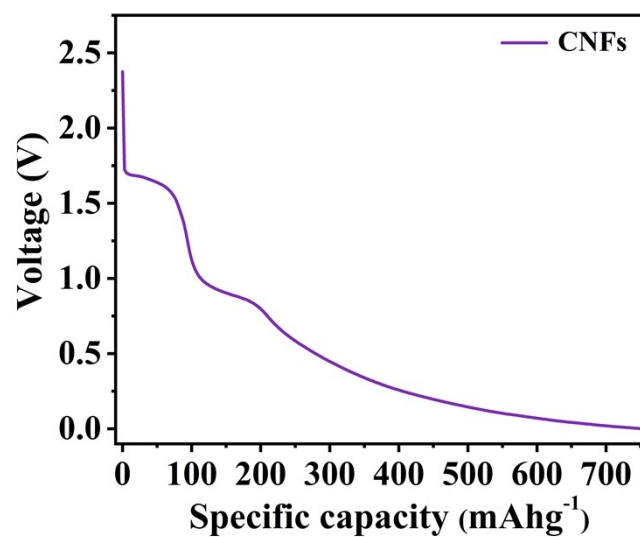


Fig. S16. voltage-specific capacity curve of CNFs||Li cell.

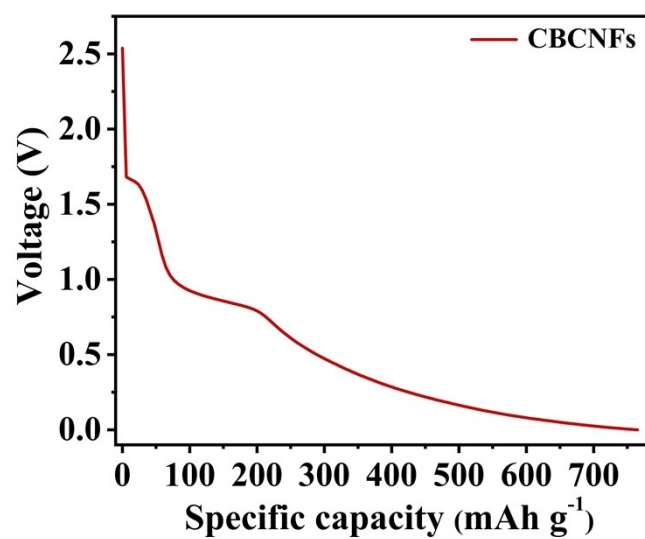


Fig. S17. voltage-specific capacity curve of CBCNFs||Li cell.

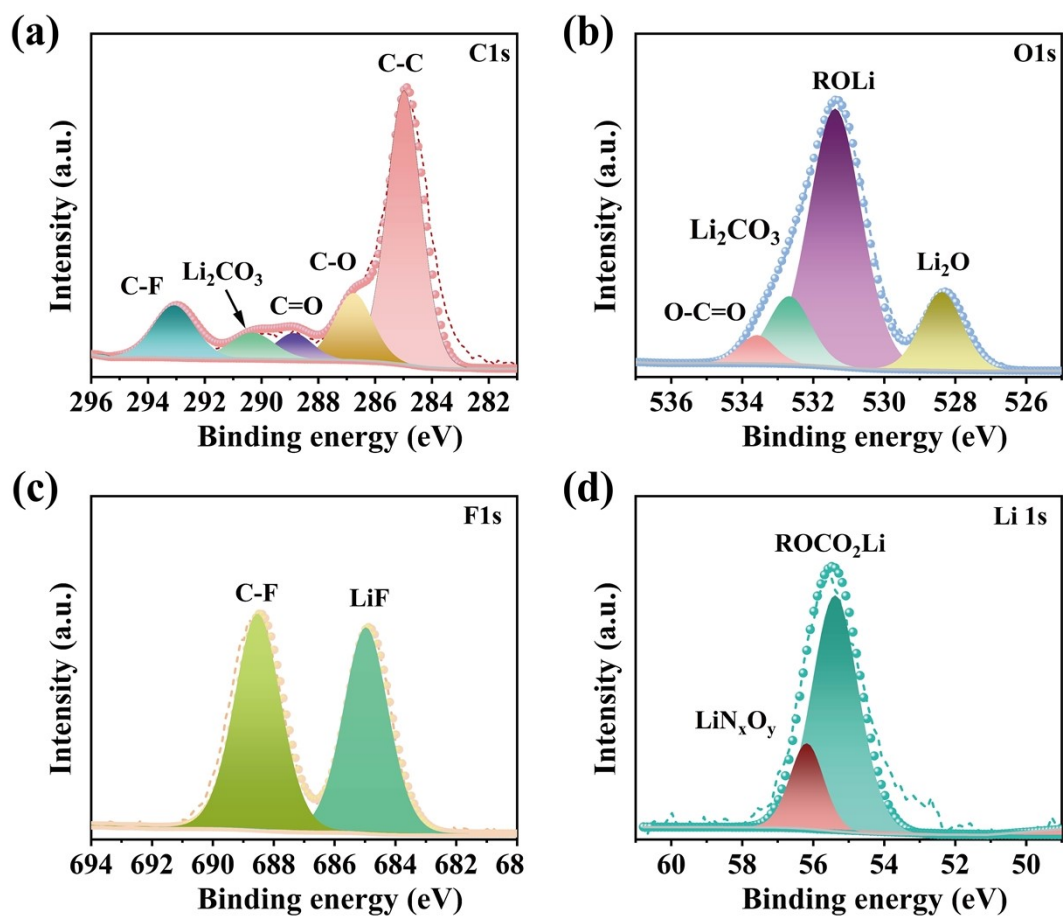


Fig. S18. (a) XPS spectra of the SEI composition on the lithium metal surface after 10 cycles of CBCNFs: (a) C1s spectrum; (b) O1s spectrum; (c) F1s spectrum; (d) Li1s spectrum