

# Supplementary Information

## Hierarchical mesoporous $\text{CuMn}_2\text{O}_4/\text{ZnMn}_2\text{O}_4$ hollow microspheres as cathode for aqueous zinc ion batteries

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### 1. Materials characterization

The surface morphology of the synthesized materials was characterized using a field-emission scanning electron microscope (Zeiss Gemini 500). High-resolution transmission electron microscopy (HRTEM) analysis was performed on an FEI-Themis Z instrument operated at 300 kV to resolve crystallographic features. X-ray diffraction (XRD) patterns were recorded with a Bruker D8 ADVANCE diffractometer using  $\text{Cu K}\alpha$  radiation ( $\lambda=1.5406 \text{ \AA}$ ) in the  $2\theta$  range of  $10^\circ$ – $80^\circ$ . Raman spectroscopy measurements were conducted on a LabRAM HR800 spectrometer equipped with a 632.8 nm He-Ne laser, collecting spectra from 100 to  $1000 \text{ cm}^{-1}$ . Surface chemical states and elemental compositions were analyzed via X-ray photoelectron spectroscopy (XPS) using an ESCALAB 250XI spectrometer with monochromatic  $\text{Al K}\alpha$  excitation source.

### 2. Electrochemical measurement

CR2032 coin-type cells were assembled for electrochemical evaluation. The cathode slurry was prepared by uniformly mixing active material, conductive carbon (acetylene black), and poly(vinylidene fluoride) (PVDF) binder in an N-methyl-2-pyrrolidone (NMP) solvent at a mass ratio of 7:2:1. After mechanical grinding for 30 min followed by magnetic stirring for 12 h, the homogeneous slurry was coated onto carbon cloth substrates and dried at  $80^\circ \text{C}$  for 12 h. The resulting electrodes exhibited an area loading

of approximately  $2 \text{ mg cm}^{-2}$ . Cell assembly involved zinc foil anodes, glass fiber separators, and an aqueous electrolyte comprising  $2 \text{ M ZnSO}_4$  supplemented with  $0.2 \text{ M MnSO}_4$  additive.

Galvanostatic charge-discharge (GCD) profiles and rate capability tests were collected on a Neware BTS-4008 battery tester within a voltage window of  $0.8\text{--}1.8 \text{ V}$ . Cyclic voltammetry (CV) measurements were carried out on a CHI660E electrochemical workstation (Shanghai Chenhua Instrument Co., Ltd.) at scan rates ranging from  $0.2$  to  $1.0 \text{ mV s}^{-1}$ . Electrochemical impedance spectroscopy (EIS) data were acquired over the frequency range of  $10^{-2}\text{--}10^5 \text{ Hz}$  with a perturbation amplitude of  $5 \text{ mV}$ .