

## **One-Pot Chloride-Assisted Corrosion Engineering of Nanostructured NiMoCo for Enhanced Alkaline Hydrogen Evolution**

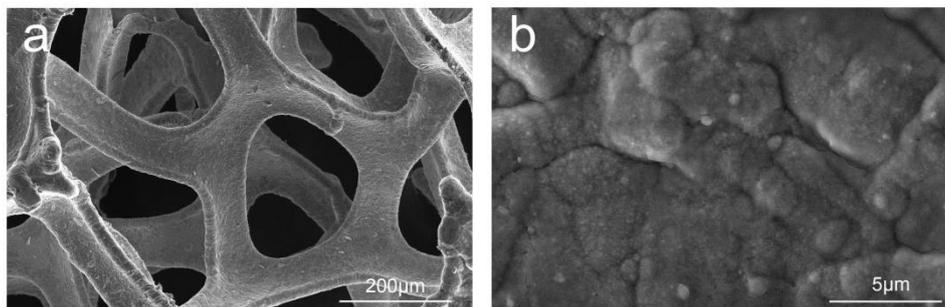
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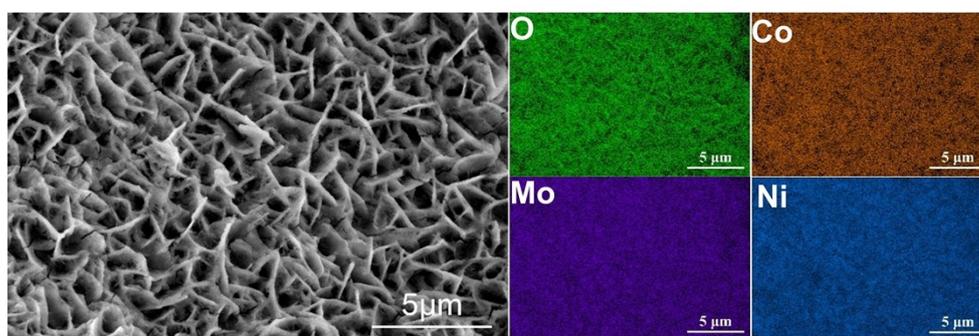
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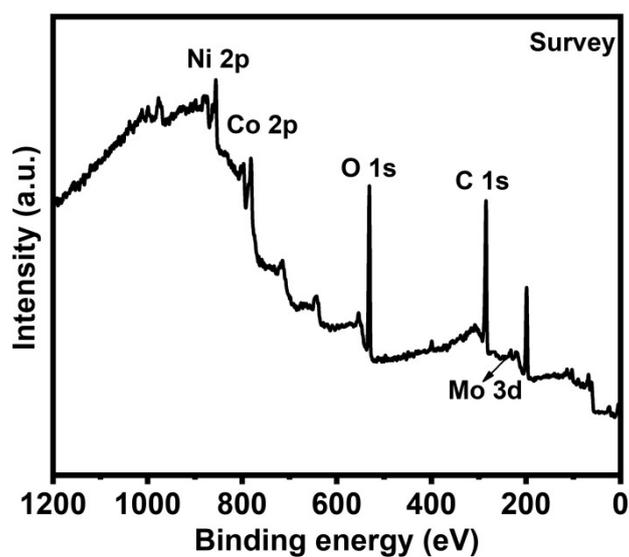
## Supplementary figures



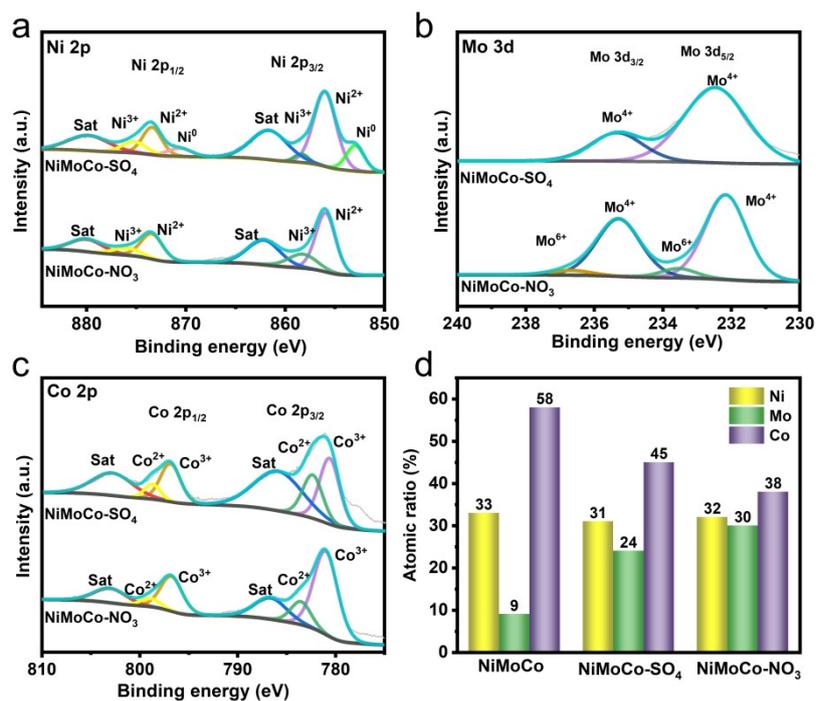
**Figure S1.** SEM images of (a-b) NMF at different scales.



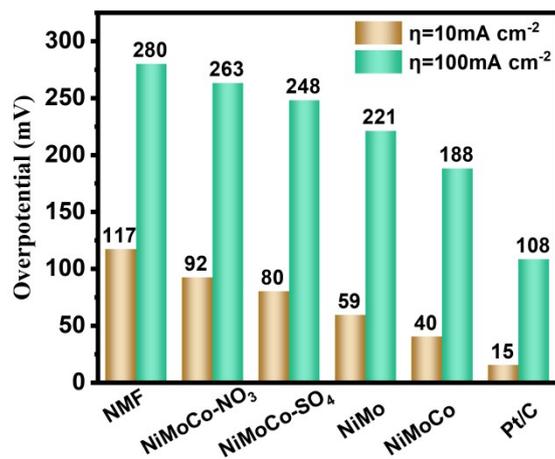
**Figure S2.** SEM image and corresponding EDS elemental mapping images of Ni, Mo, Co, and O acquired from the same region.



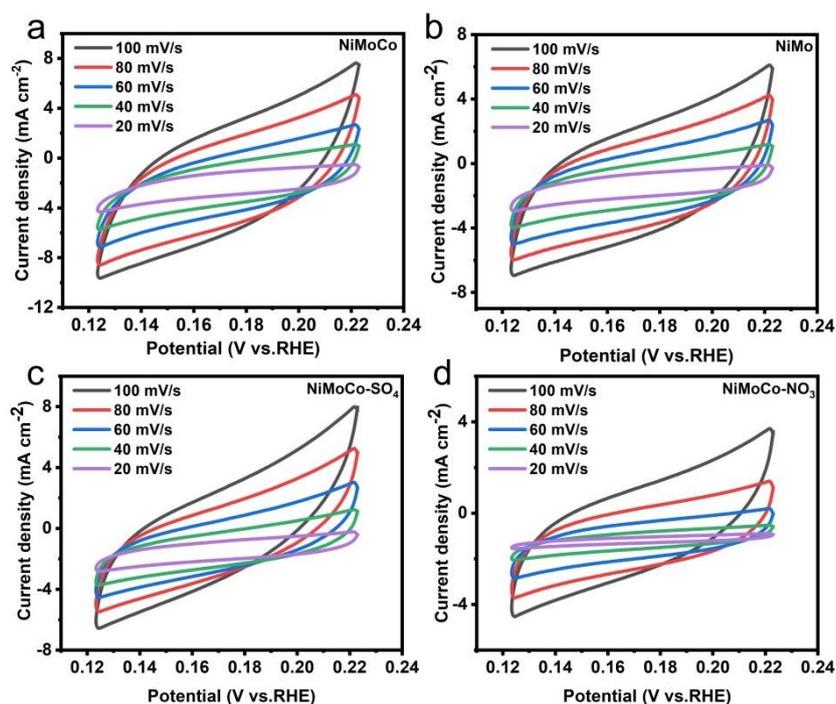
**Figure S3.** Full survey XPS spectra of NiMoCo.



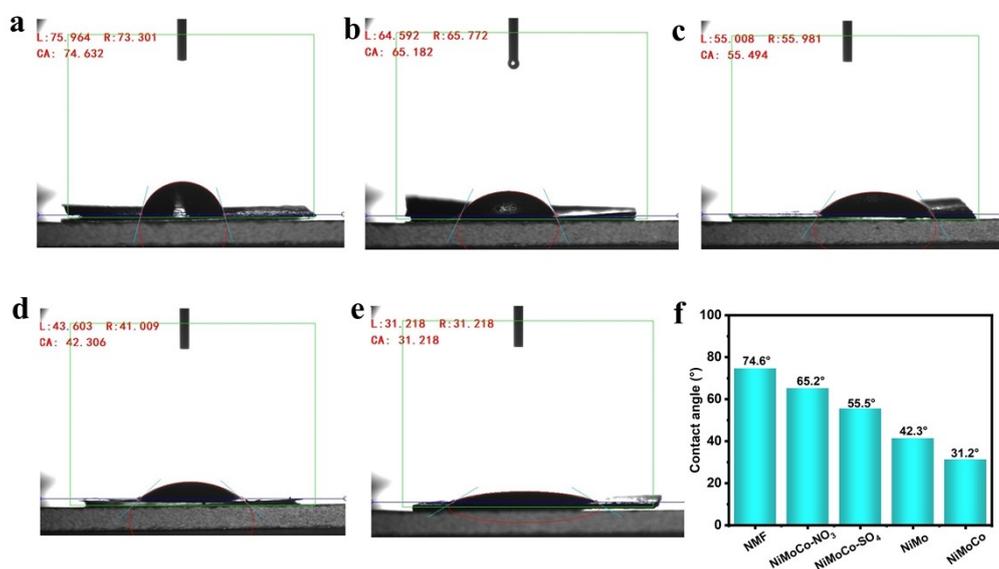
**Figure S4.** High-resolution XPS spectra of (a) Ni 2p, (b) Mo 3d, and (c) Co 2p for NiMoCo-SO<sub>4</sub> and NiMoCo-NO<sub>3</sub>; (d) Surface elemental composition for NiMoCo, NiMoCo-SO<sub>4</sub>, and NiMoCo-NO<sub>3</sub>.



**Figure S5.** HER overpotentials at 10 and 100 mA · cm<sup>-2</sup> for various catalysts measured in 1 M KOH.



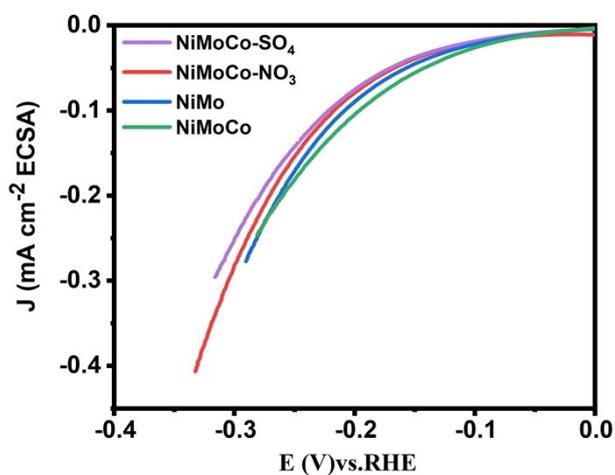
**Figure S6.** CV curves recorded at different scan rates for (a) NiMoCo, (b) chloride-assisted NiMo, (c) NiMoCo-SO<sub>4</sub>, and (d) NiMoCo-NO<sub>3</sub> in 1 M KOH solution.



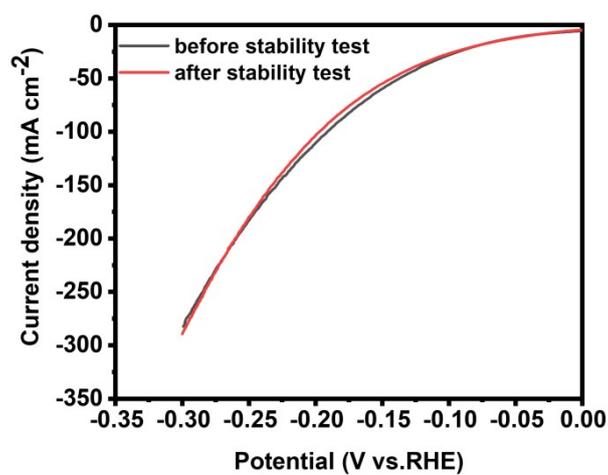
**Figure S7.** (a) Static water contact angle image of pristine NMF without corrosion treatment. (b-e) Static water contact angle images of corrosion-treated electrodes prepared via different anionic environments: (b) NiMoCo-NO<sub>3</sub>, (c) NiMoCo-SO<sub>4</sub>, (d)

chloride-assisted corroded NiMo, and (e) chloride-assisted corroded NiMoCo. (f)

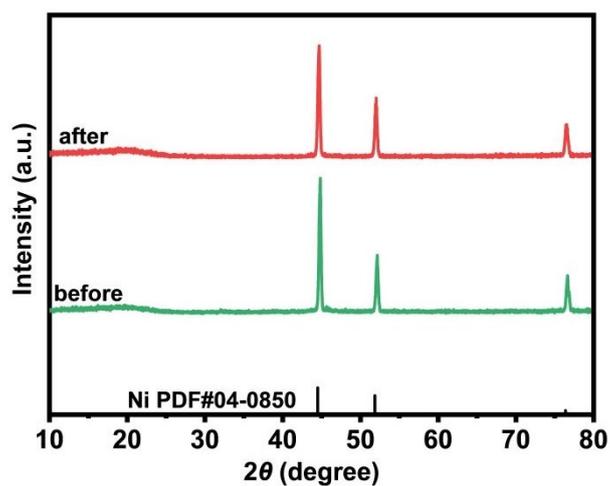
Corresponding contact angle values of the different samples.



**Figure S8.** ECSA-normalized LSV curves for NiMoCo, chloride-assisted NiMo, NiMoCo-SO<sub>4</sub>, NiMoCo-NO<sub>3</sub>, and NMF in 1 M KOH.



**Figure S9.** LSV curves of NiMoCo before and after 48-hour HER stability test.



**Figure S10.** XRD analysis of NiMoCo before and after 48-hour HER stability test.

**Table S1.** The atomic ratio of Ni, Mo, Co and O elements on the surface of NiMoCo obtained from EDS.

Catalyst	Atomic %			
	Ni	Mo	Co	O
NiMoCo	34.43	6.86	10.93	47.78

**Table S2.** Comparison between this work and the recently reported performance of similar catalysts in alkaline electrolyte.

Electrocatalysts	Electrolyte	$\eta_{10}$ (mV)	Ref.
(Co-NiMo)OH <sub>x</sub> /NMF	1 M KOH	40	This work
NiMo/NiCo <sub>2</sub> O <sub>4</sub>	1 M KOH	44	(1)
NiMo@NF-x	1 M KOH	53	(2)
NiMoO/Ni <sub>4</sub> Mo@NC <sub>4</sub>	1 M KOH	61	(3)
Mo/NiMo/NF	1 M KOH	71	(4)
NiMo/Co-NW/CP	1 M KOH	85	(5)
H-Mo <sub>2</sub> C@Co	1 M KOH	103	(6)
NiCo-MOF	1 M KOH	125	(7)
NiMo/Ni(OH) <sub>2</sub> /CC	1 M KOH	132	(8)

**Table S3.** The EIS fit values of catalysts for HER.

Sample	R <sub>s</sub> (Ω)	R <sub>ct</sub> (Ω)
NiMoCo	2.47	5.75
NiMo	2.73	6.58
NiMoCo-SO <sub>4</sub>	2.11	11.19
NiMoCo-NO <sub>3</sub>	2.01	13.23
NMF	2.56	14.70

## References

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