

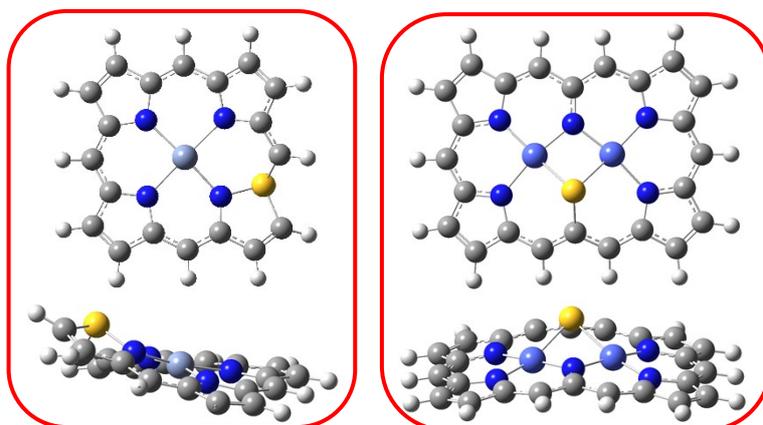
## Supporting Information

### **Theoretical investigation of tandem catalysts utilizing S- modified metalloporphyrin two-dimensional carbon-rich conjugated frameworks for the electrocatalytic reduction synthesis of L- erythrulose**

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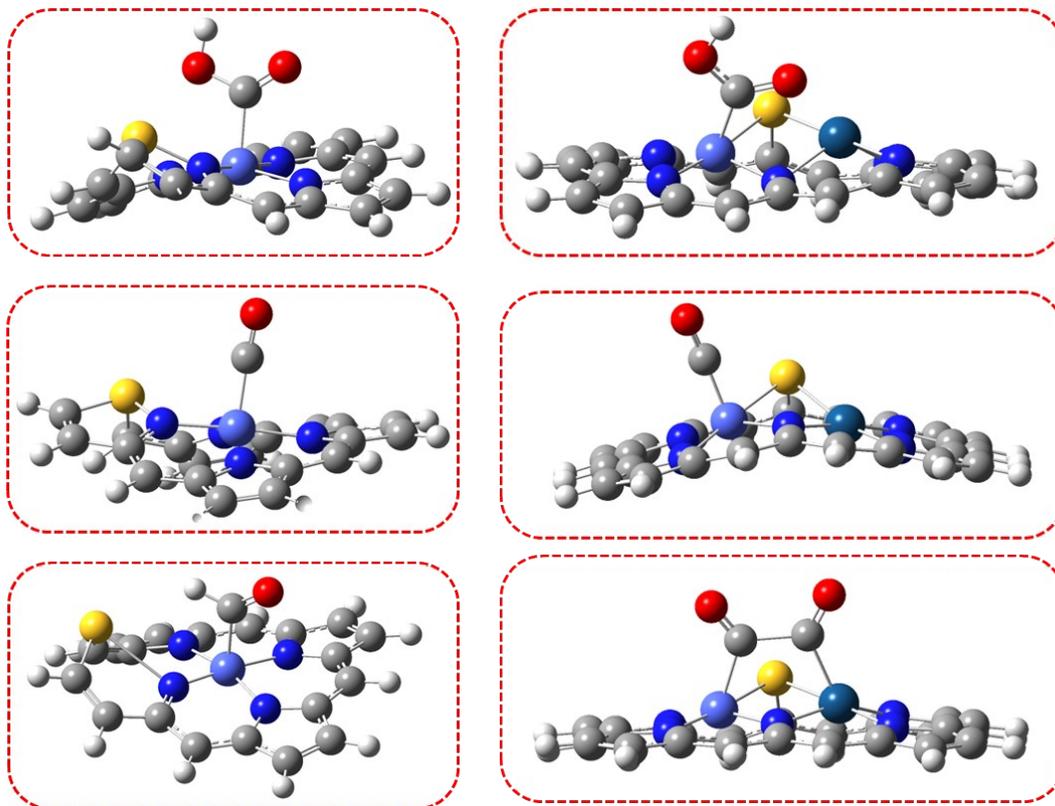
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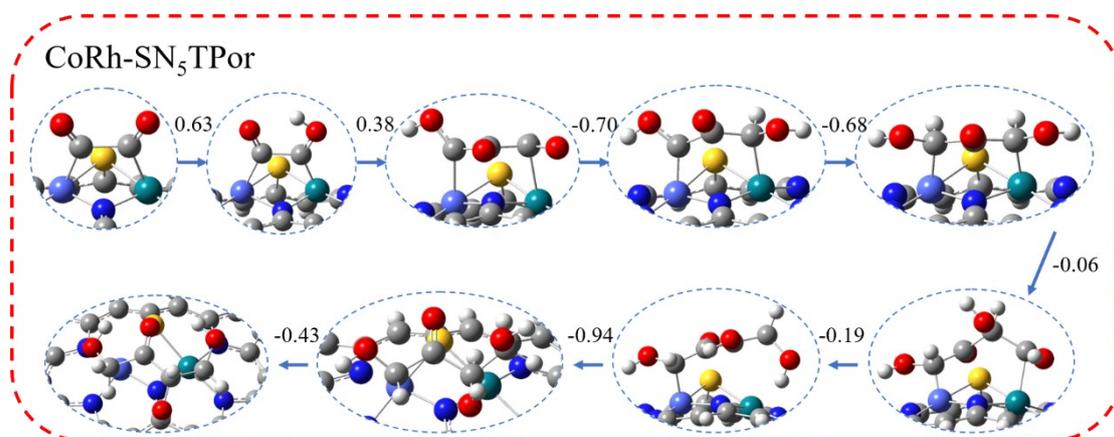


**Fig. S1.** Optimized M-SPor and CoM-SN<sub>5</sub>TPor structures.

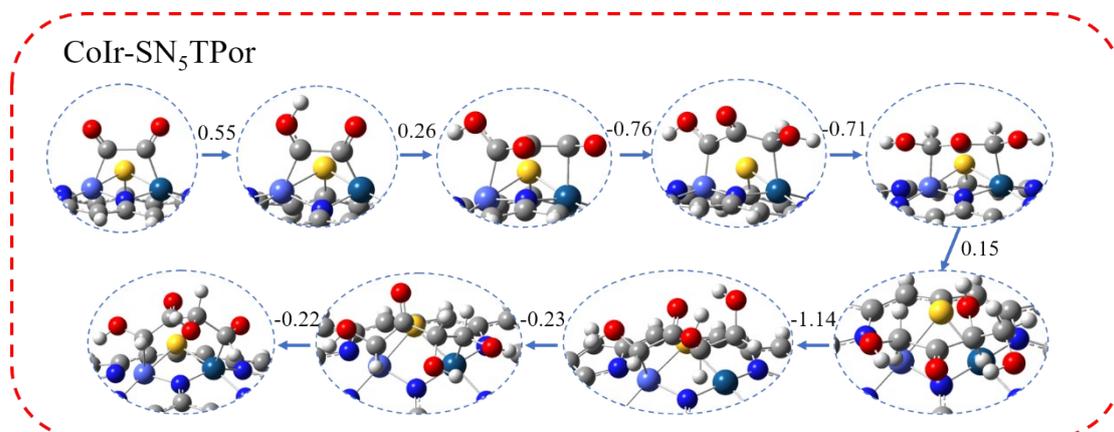
**Fig. S2.** Three key adsorption intermediates over TM-SPor and CoTM-SN<sub>5</sub>TPor catalysts.



**Fig. S3.** Schematic representation of the structural optimisation for L-erythrulose synthesis via C-C coupling on the CoRh-SN<sub>5</sub>TPor catalyst.



**Fig. S4.** Schematic representation of the structural optimisation for L-erythrulose synthesis via C-C coupling on the CoIr-SN<sub>5</sub>TPor catalyst.



**Table S1.** Computed formation energy ( $E_f$ ) and dissolution potential ( $U_{\text{diss}}$ ) of TM-SPor,  $U_{\text{diss}}^0$  and  $n$  are the standard dissolution potential of bulk metal and the number of transferred electrons involved in the dissolution.

<i>M-SPor</i>	<i>n</i>	$U_{\text{diss}}^0$ (V)	$E_f$ (eV)	$U_{\text{diss}}$ (V)
<i>Ti-SPor</i>	2	-1.63	-16.09	6.42
<i>V-SPor</i>	2	-1.18	-10.48	4.06
<i>Cr-SPor</i>	2	-0.91	-19.77	8.97
<i>Mn-SPor</i>	2	-0.19	-10.95	5.29
<i>Fe-SPor</i>	2	-0.45	-18.60	8.85
<i>Co-SPor</i>	2	-0.28	-8.29	3.87
<i>Ni-SPor</i>	2	-0.26	-16.08	7.78
<i>Zr-SPor</i>	4	-1.45	-13.32	1.88
<i>Nb-SPor</i>	3	-1.1	-12.14	2.95
<i>Mo-SPor</i>	3	-0.2	-15.44	4.95
<i>Ru-SPor</i>	2	0.46	-11.90	6.41
<i>Rh-SPor</i>	2	0.6	-8.41	4.81
<i>Hf-SPor</i>	4	-1.55	-12.17	1.49
<i>Ta-SPor</i>	3	-0.6	-12.88	3.69
<i>W-SPor</i>	3	0.1	-17.35	5.88
<i>Re-SPor</i>	3	0.3	-12.18	4.36
<i>Os-SPor</i>	8	0.84	-16.02	2.84
<i>Ir-SPor</i>	3	1.16	-10.29	4.59
<i>Pt-SPor</i>	2	1.18	-10.95	6.66

**Table S2.** The calculated Gibbs free energy changes for CO<sub>2</sub> adsorption ( $\Delta G^*_{CO_2}$ ) on various TM-SPor candidates. (E in eV).

<i>M-SPor</i>	$\Delta G^*_{CO_2}$	<i>M-SPor</i>	$\Delta G^*_{CO_2}$
<i>Ti-SPor</i>	0.05	<i>Ru-SPor</i>	-0.13
<i>V-SPor</i>	-0.50	<i>Rh-SPor</i>	-0.16
<i>Cr-SPor</i>	0.32	<i>Hf-SPor</i>	-1.72
<i>Mn-SPor</i>	-0.15	<i>Ta-SPor</i>	-0.83
<i>Fe-SPor</i>	-0.03	<i>W-SPor</i>	-1.17
<i>Co-SPor</i>	-0.13	<i>Re-SPor</i>	-0.52
<i>Ni-SPor</i>	-0.61	<i>Os-SPor</i>	0.52
<i>Zr-SPor</i>	-0.05	<i>Ir-SPor</i>	-0.16
<i>Nb-SPor</i>	0.96	<i>Pt-SPor</i>	-0.43
<i>Mo-SPor</i>	0.58		

**Table S3.** The calculated Gibbs free energy changes for \*COOH to \*CO on M-SPor candidates(E in eV).

<i>M-SPor</i>	$\Delta G_{*COOH \rightarrow *CO}$	<i>M-SPor</i>	$\Delta G_{*COOH \rightarrow *CO}$
<i>V-SPor</i>	0.40	<i>Rh-SPor</i>	1.04
<i>Mn-SPor</i>	0.28	<i>Hf-SPor</i>	0.92
<i>Fe-SPor</i>	-0.06	<i>Ta-SPor</i>	1.02
<i>Co-SPor</i>	-0.21	<i>W-SPor</i>	0.64
<i>Ni-SPor</i>	0.33	<i>Re-SPor</i>	0.21
<i>Zr-SPor</i>	0.81	<i>Ir-SPor</i>	1.15
<i>Ru-SPor</i>	0.00	<i>Pt-SPor</i>	-0.61

**Table S4.** The calculated Gibbs free energy changes for the first hydrogenation step in CO<sub>2</sub>RR ( $\Delta G^*_{\text{CO}_2 \rightarrow \text{COOH}}$ ) and HER ( $\Delta G^*_{\text{H}}$ ) on various M-SPor candidates. (E in eV).

<i>M-SPor</i>	$\Delta G^*_{\text{CO}_2 \rightarrow \text{COOH}}$	$\Delta G^*_{\text{H}}$	<i>M-SPor</i>	$\Delta G^*_{\text{CO}_2 \rightarrow \text{COOH}}$	$\Delta G^*_{\text{H}}$
<i>V-SPor</i>	0.09	-0.21	<i>Rh-SPor</i>	0.00	-0.19
<i>Mn-SPor</i>	0.35	0.31	<i>Hf-SPor</i>	-0.25	-1.17
<i>Fe-SPor</i>	-0.28	-0.15	<i>Ta-SPor</i>	-0.63	-1.15
<i>Co-SPor</i>	0.34	0.42	<i>W-SPor</i>	0.34	-0.89
<i>Ni-SPor</i>	1.29	0.84	<i>Re-SPor</i>	-0.54	-0.99
<i>Zr-SPor</i>	-0.54	-0.33	<i>Ir-SPor</i>	-0.19	-0.37
<i>Ru-SPor</i>	-0.28	-0.39	<i>Pt-SPor</i>	2.00	0.25

**Table S5.** Computed formation energy ( $E_f$ ) and dissolution potential ( $U_{diss}$ ) of FeTMs@Pc,  $U_{diss}^0$  and  $n$  are the standard dissolution potential of bulk metal and the number of transferred electrons involved in the dissolution, respectively.

$$n = [n(Fe) + n(TM)]/2, U_{diss}^0 = [U_{diss}^0(Fe) + U_{diss}^0(TM)]/2.$$

<i>M-Spor</i>	<i>n</i>	$U_{diss}^0(V)$	$E_f(eV)$	$U_{diss}(V)$
<i>CoTi-SN<sub>5</sub>TPor</i>	2	-0.96	-20.85	9.47
<i>CoV-SN<sub>5</sub>TPor</i>	2	-0.73	-16.16	7.35
<i>CoCr-SN<sub>5</sub>TPor</i>	2	-0.60	-23.58	11.19
<i>CoMn-SN<sub>5</sub>TPor</i>	2	-0.24	-16.17	7.85
<i>CoFe-SN<sub>5</sub>TPor</i>	2	-0.37	-24.22	11.74
<i>CoCo-SN<sub>5</sub>TPor</i>	2	-0.28	-13.61	6.52
<i>CoZr-SN<sub>5</sub>TPor</i>	4	-0.87	-17.73	5.05
<i>CoNb-SN<sub>5</sub>TPor</i>	3	-0.69	-16.53	5.92
<i>CoMo-SN<sub>5</sub>TPor</i>	3	-0.24	-19.32	7.49
<i>CoRu-SN<sub>5</sub>TPor</i>	2	0.09	-16.78	8.48
<i>CoRh-SN<sub>5</sub>TPor</i>	2	0.16	-13.50	6.91
<i>CoPd-SN<sub>5</sub>TPor</i>	2	0.34	-11.95	6.31
<i>CoHf-SN<sub>5</sub>TPor</i>	4	-0.92	-18.21	5.15
<i>CoTa-SN<sub>5</sub>TPor</i>	3	-0.44	-17.35	6.50
<i>CoW-SN<sub>5</sub>TPor</i>	3	-0.09	-21.30	8.43
<i>CoRe-SN<sub>5</sub>TPor</i>	3	0.01	-16.41	6.58
<i>CoOs-SN<sub>5</sub>TPor</i>	8	0.28	-20.50	4.38
<i>CoIr-SN<sub>5</sub>TPor</i>	3	0.44	-15.35	6.58
<i>CoPt-SN<sub>5</sub>TPor</i>	2	0.45	-16.07	8.48

**Table S6.** The calculated Gibbs free energy changes for \*COOH to \*CO on CoTM-SN<sub>5</sub>TPor candidates. (E in eV).

<i>CoTM-SN<sub>5</sub>TPor</i>	$\Delta G^*_{COOH \rightarrow *CO}$	<i>CoTM-SN<sub>5</sub>TPor</i>	$\Delta G^*_{COOH \rightarrow *CO}$
<i>CoV-SN<sub>5</sub>TPor</i>	0.75	<i>CoRh-SN<sub>5</sub>TPor</i>	-0.46
<i>CoCr-SN<sub>5</sub>TPor</i>	1.36	<i>CoPd-SN<sub>5</sub>TPor</i>	0.15
<i>CoMn-SN<sub>5</sub>TPor</i>	-0.82	<i>CoAg-SN<sub>5</sub>TPor</i>	0.40
<i>CoFe-SN<sub>5</sub>TPor</i>	-0.59	<i>CoHf-SN<sub>5</sub>TPor</i>	1.57
<i>CoCo-SN<sub>5</sub>TPor</i>	0.09	<i>CoTa-SN<sub>5</sub>TPor</i>	0.80
<i>CoNi-SN<sub>5</sub>TPor</i>	0.21	<i>CoW-SN<sub>5</sub>TPor</i>	0.62
<i>CoCu-SN<sub>5</sub>TPor</i>	0.13	<i>CoRe-SN<sub>5</sub>TPor</i>	0.26
<i>CoY-SN<sub>5</sub>TPor</i>	1.20	<i>CoOs-SN<sub>5</sub>TPor</i>	-0.11
<i>CoZr-SN<sub>5</sub>TPor</i>	0.88	<i>CoIr-SN<sub>5</sub>TPor</i>	-0.58
<i>CoNb-SN<sub>5</sub>TPor</i>	0.41	<i>CoPt-SN<sub>5</sub>TPor</i>	0.14
<i>CoMo-SN<sub>5</sub>TPor</i>	1.49	<i>CoAu-SN<sub>5</sub>TPor</i>	0.47
<i>CoRu-SN<sub>5</sub>TPor</i>	0.09		

**Table S7.** The calculated Gibbs free energy changes for the first hydrogenation step in CO<sub>2</sub>RR ( $\Delta G^*_{\text{COOH}}$ ) and HER ( $\Delta G^*_{\text{H}}$ ) on various CoTM-SN<sub>5</sub>TPor candidates. (E in eV).

<i>CoTM-SN<sub>5</sub>TPor</i>	$\Delta G^*_{\text{H}}$	$\Delta G^*_{\text{CO}_2 \rightarrow \text{COOH}}$	<i>CoTM-SN<sub>5</sub>TPor</i>	$\Delta G^*_{\text{CO}_2 \rightarrow \text{COO}}$	$\Delta G^*_{\text{H}}$
				<i>H</i>	
<i>CoTi-SN<sub>5</sub>TPor</i>	0.46	0.47	<i>CoRu-SN<sub>5</sub>TPor</i>	0.49	-0.02
<i>CoV-SN<sub>5</sub>TPor</i>	0.49	0.33	<i>CoRh-SN<sub>5</sub>TPor</i>	0.40	-0.08
<i>CoCr-SN<sub>5</sub>TPor</i>	-0.39	0.51	<i>CoPd-SN<sub>5</sub>TPor</i>	0.40	0.05
<i>CoMn-SN<sub>5</sub>TPor</i>	1.61	0.65	<i>CoHf-SN<sub>5</sub>TPor</i>	0.79	0.48
<i>CoFe-SN<sub>5</sub>TPor</i>	0.17	-0.30	<i>CoTa-SN<sub>5</sub>TPor</i>	-0.72	0.84
<i>CoCo-SN<sub>5</sub>TPor</i>	0.21	0.25	<i>CoW-SN<sub>5</sub>TPor</i>	0.40	0.36
<i>CoZr-SN<sub>5</sub>TPor</i>	0.29	0.49	<i>CoRe-SN<sub>5</sub>TPor</i>	0.73	0.55
<i>CoNb-SN<sub>5</sub>TPor</i>	0.64	0.83	<i>CoOs-SN<sub>5</sub>TPor</i>	0.48	0.11
<i>CoMo-SN<sub>5</sub>TPor</i>	0.30	0.28	<i>CoIr-SN<sub>5</sub>TPor</i>	0.39	0.01

**Table S8.** Summary of valence-electron number  $Sv_1$ ,  $Sv_2$  and  $Sv_3$  of transition metal dopant and nitrogen atom, the electronegativity  $\chi_1$ ,  $\chi_2$ ,  $\chi_3$  and  $\chi_4$  of transition metal element and nitrogen atoms and descriptor  $\psi$  for CoTM-SN<sub>5</sub>TPor.

<i>CoTM-SN<sub>5</sub>TPor</i>	$Sv_1$	$Sv_2$	$Sv_3$	$Sv_4$	$\chi_1$	$\chi_2$	$\chi_3$	$\chi_4$	$\psi$
<i>CoTi-SN<sub>5</sub>TPor</i>	4	9	6	5	1.54	1.88	2.58	3.04	15.05
<i>CoV-SN<sub>5</sub>TPor</i>	5	9	6	5	1.63	1.88	2.58	3.04	16.59
<i>CoCr-SN<sub>5</sub>TPor</i>	6	9	6	5	1.66	1.88	2.58	3.04	18.10
<i>CoMn-SN<sub>5</sub>TPor</i>	7	9	6	5	1.55	1.88	2.58	3.04	19.88
<i>CoFe-SN<sub>5</sub>TPor</i>	8	9	6	5	1.83	1.88	2.58	3.04	20.39
<i>CoCo-SN<sub>5</sub>TPor</i>	9	9	6	5	1.88	1.88	2.58	3.04	21.48
<i>CoCu-SN<sub>5</sub>TPor</i>	11	9	6	5	1.9	1.88	2.58	3.04	23.69
<i>CoZr-SN<sub>5</sub>TPor</i>	4	9	6	5	1.33	1.88	2.58	3.04	15.62
<i>CoNb-SN<sub>5</sub>TPor</i>	5	9	6	5	1.6	1.88	2.58	3.04	16.67
<i>CoMo-SN<sub>5</sub>TPor</i>	6	9	6	5	2.16	1.88	2.58	3.04	16.94
<i>CoRu-SN<sub>5</sub>TPor</i>	8	9	6	5	2.2	1.88	2.58	3.04	19.47
<i>CoRh-SN<sub>5</sub>TPor</i>	9	9	6	5	2.28	1.88	2.58	3.04	20.47
<i>CoPd-SN<sub>5</sub>TPor</i>	10	9	6	5	2.2	1.88	2.58	3.04	21.77
<i>CoAg-SN<sub>5</sub>TPor</i>	11	9	6	5	1.93	1.88	2.58	3.04	23.60
<i>CoHf-SN<sub>5</sub>TPor</i>	4	9	6	5	1.3	1.88	2.58	3.04	15.71
<i>CoTa-SN<sub>5</sub>TPor</i>	5	9	6	5	1.5	1.88	2.58	3.04	16.94
<i>CoW-SN<sub>5</sub>TPor</i>	6	9	6	5	2.36	1.88	2.58	3.04	16.57
<i>CoRe-SN<sub>5</sub>TPor</i>	7	9	6	5	1.9	1.88	2.58	3.04	18.90
<i>CoOs-SN<sub>5</sub>TPor</i>	8	9	6	5	2.2	1.88	2.58	3.04	19.47
<i>CoIr-SN<sub>5</sub>TPor</i>	9	9	6	5	2.2	1.88	2.58	3.04	20.66
<i>CoPt-SN<sub>5</sub>TPor</i>	10	9	6	5	2.28	1.88	2.58	3.04	21.58
<i>CoAu-SN<sub>5</sub>TPor</i>	11	9	6	5	2.54	1.88	2.58	3.04	22.03

**TableS9.** The adsorption energy of \*CO,\*COOH and \*HCOO on CoTM-SN<sub>5</sub>TPor.

<i>CoTM-SN<sub>5</sub>TPor</i>	<i>E*CO</i>	<i>E*COOH</i>	<i>E*HCOO</i>
<i>CoTi-SN<sub>5</sub>TPor</i>	-0.43	-2.54	-4.12
<i>CoV-SN<sub>5</sub>TPor</i>	-0.58	-2.16	-3.54
<i>CoCr-SN<sub>5</sub>TPor</i>	-0.75	-2.94	-4.49
<i>CoMn-SN<sub>5</sub>TPor</i>	-0.64	-0.64	-1.57
<i>CoFe-SN<sub>5</sub>TPor</i>	-1.25	-1.49	-2.32
<i>CoCo-SN<sub>5</sub>TPor</i>	-0.96	-1.02	-1.84
<i>CoCu-SN<sub>5</sub>TPor</i>	-0.85	-1.81	-1.79
<i>CoZr-SN<sub>5</sub>TPor</i>	-1.26	-2.98	-4.60
<i>CoNb-SN<sub>5</sub>TPor</i>	-0.89	-2.13	-4.02
<i>CoMo-SN<sub>5</sub>TPor</i>	-0.09	-2.41	-3.93
<i>CoRu-SN<sub>5</sub>TPor</i>	-0.85	-1.77	-2.83
<i>CoRh-SN<sub>5</sub>TPor</i>	-0.84	-1.21	-1.93
<i>CoPd-SN<sub>5</sub>TPor</i>	-0.92	-1.90	-1.78
<i>CoAg-SN<sub>5</sub>TPor</i>	-0.82	-2.05	-1.85
<i>CoHf-SN<sub>5</sub>TPor</i>	-0.31	-2.72	-4.87
<i>CoTa-SN<sub>5</sub>TPor</i>	-0.40	-2.03	-4.11
<i>CoW-SN<sub>5</sub>TPor</i>	-0.75	-2.21	-3.87
<i>CoRe-SN<sub>5</sub>TPor</i>	-0.55	-1.64	-3.20
<i>CoOs-SN<sub>5</sub>TPor</i>	-0.76	-1.48	-2.77
<i>CoIr-SN<sub>5</sub>TPor</i>	-0.83	-1.08	-1.91
<i>CoPt-SN<sub>5</sub>TPor</i>	-1.35	-2.32	-2.21
<i>CoAu-SN<sub>5</sub>TPor</i>	-0.83	-2.13	-2.01