

Supplementary Information

γ -Graphyne modulated g-C₃N₄ nanostructures for boosted photocatalytic ammonia production

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Table 1 Corresponding absorbance of ammonia nitrogen standard solution.

Chemicals and materials

Most chemicals were purchased from Sinopharm Chemical Reagent Co. Ltd, including melamine (CP), benzene/hexahydrobenzene (CP), benzene ($\geq 99.5\%$, AR), absolute ethanol (AR), HNO_3 (68 wt.%, AR), anhydrous sodium sulfate (AR), methanol ($\geq 99.5\%$, AR), ammonium chloride ($\geq 99.5\%$, CP), potassium sodium tartrate ($\geq 99.5\%$, AR) and Nessler's reagent. Fluorine-doped tin oxide (FTO) conducting glasses were obtained from Foshan City Mei jing Glass Co. Ltd. Nafion (5 wt.%) and CaC_2 ($\geq 80\%$) were gained from Sigma-Aldrich (USA). N_2 (99.5 %) and Ar (99.5 %) were supplied from Shanghai Haoqi Gases Co., Ltd. They were used as received, without any further treatment.

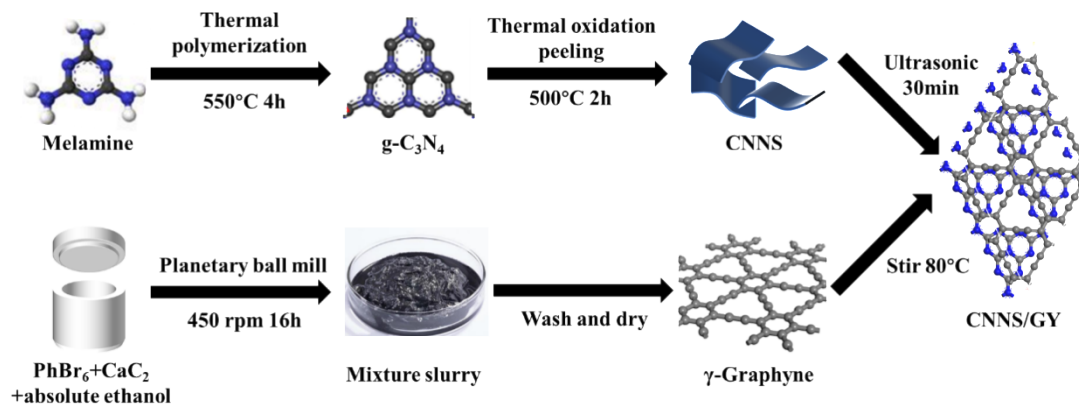


Fig. S1 Schematic diagram of preparation process of CNNS/GY

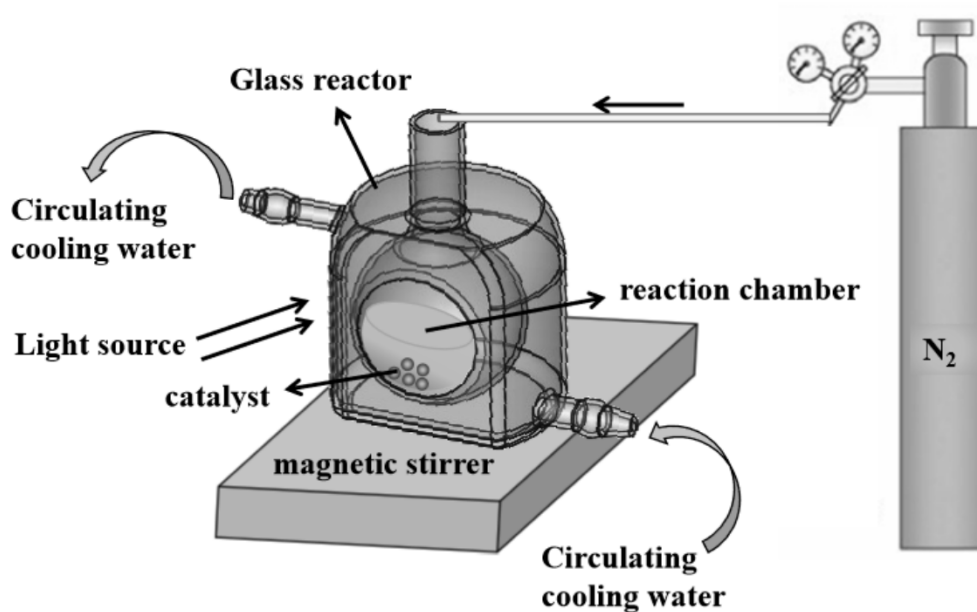


Fig. S2 The schematic picture of the used reactor system.

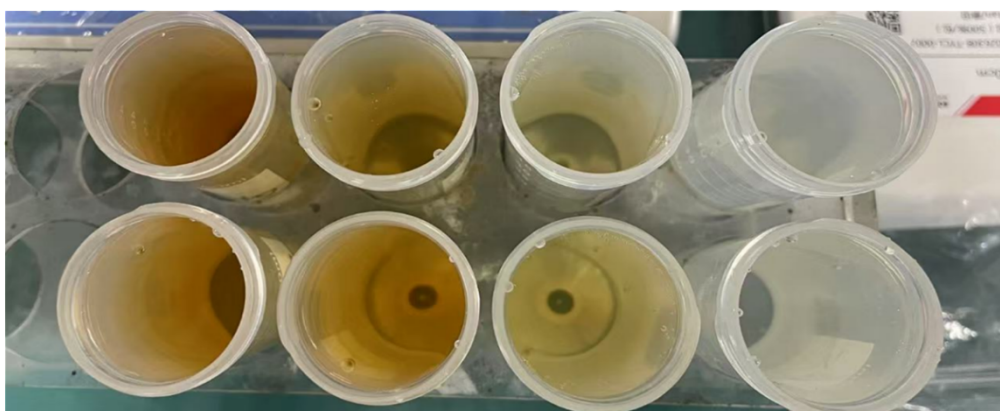


Fig. S3 Colorimetric photos of the solution at different ammonia nitrogen concentrations.

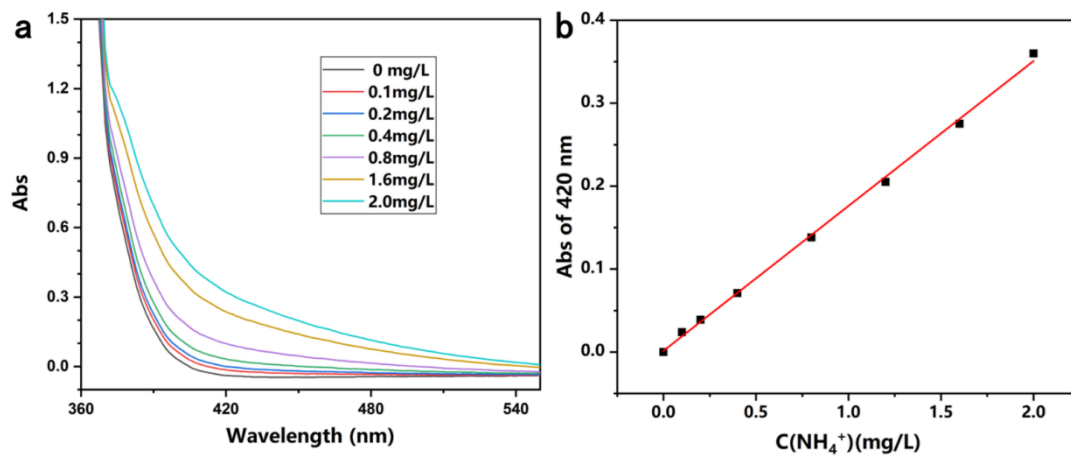


Fig. S4 Determination of ammonia nitrogen standard curve. (a) Absorption spectra of NH_4^+ standard solutions, (b) Linear plot of the absorbance values at 420 nm versus NH_4^+ concentration.

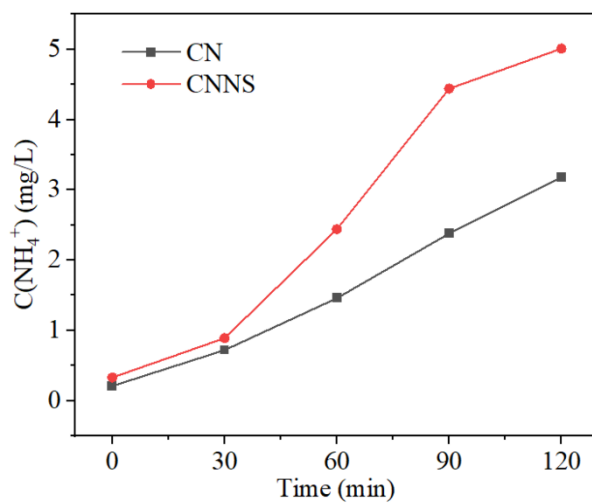


Fig. S5 Time-dependent NH_4^+ concentrations generated by CN and CNNS.

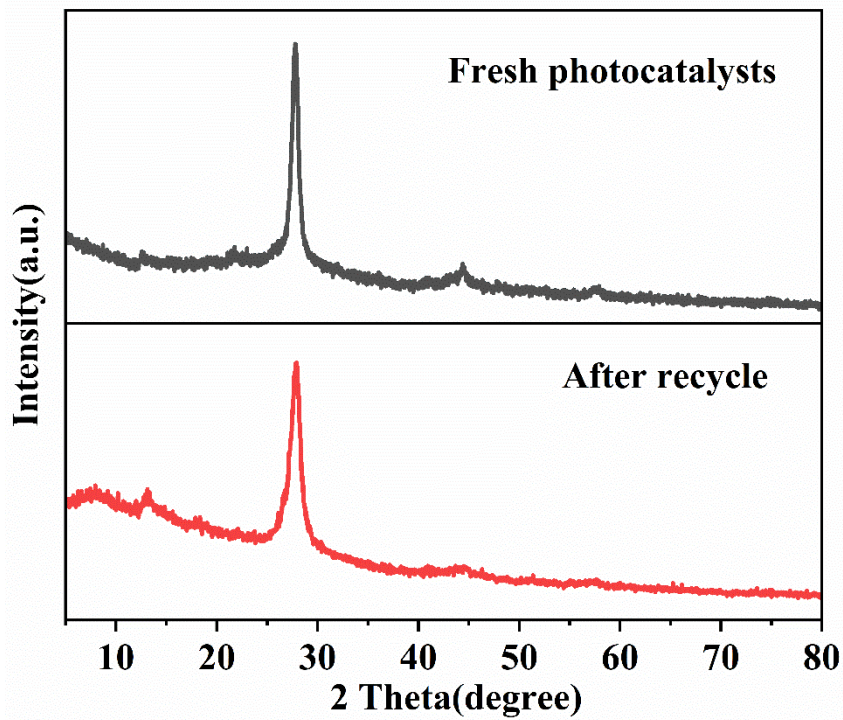


Fig. S6 XRD patterns of CNNS/GY composites before and after three recycles.

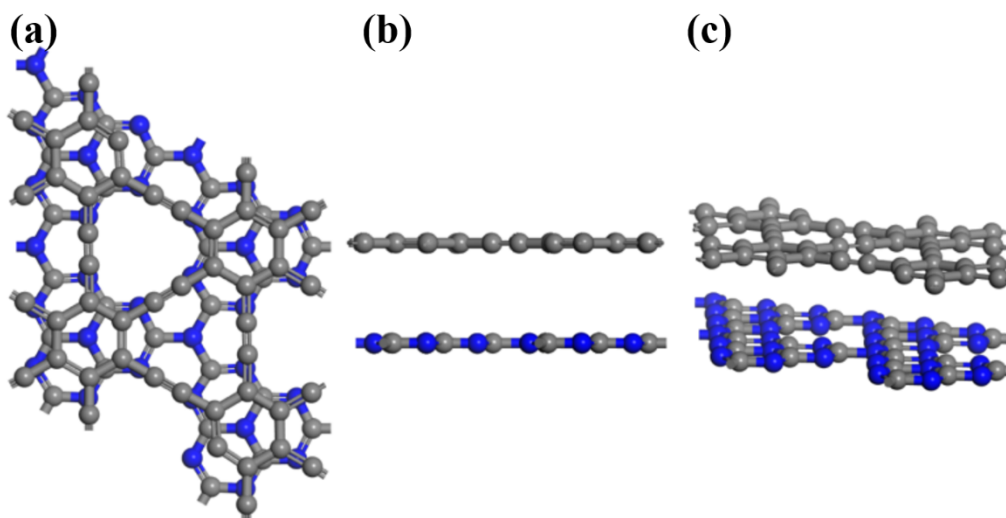


Fig. S7 Theoretical models of the CNNS/GY electronic structure: (a) top view; (b) side view; (c) oblique view.

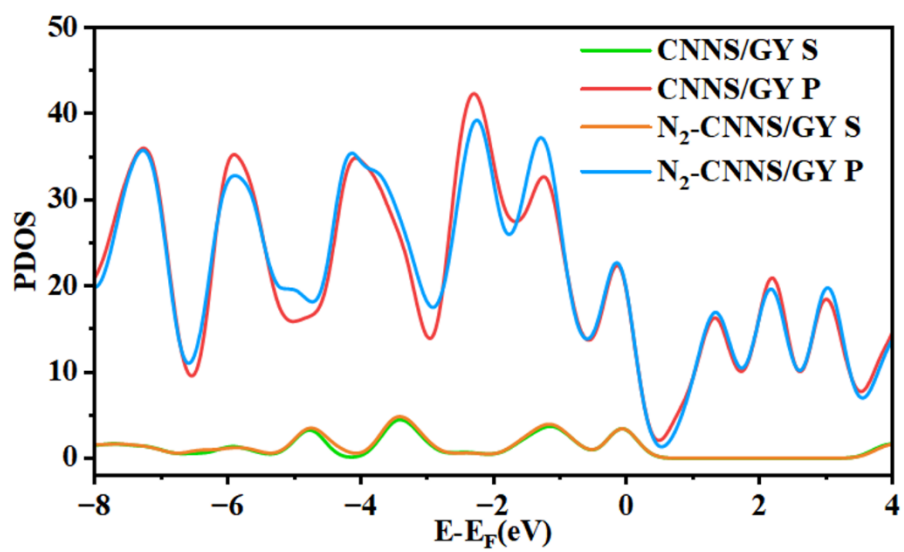


Fig. S8 Partial density of states (PDOS) of the CNNS/GY heterostructure before and after N₂ adsorption.

Table 1 Corresponding absorbance of ammonia nitrogen standard solution

Concentration (mg/L)	Actual Absorbance (Abs)
0	0
0.1	0.024
0.2	0.039
0.4	0.071
0.8	0.138
1.2	0.205
1.6	0.275
2.0	0.360