

## Interfacial Charge Engineering in TiSSe/Borophene van der Waals Heterostructures for Enhanced Quantum Capacitance and Electro-Mechanical Stability

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### Supporting Information

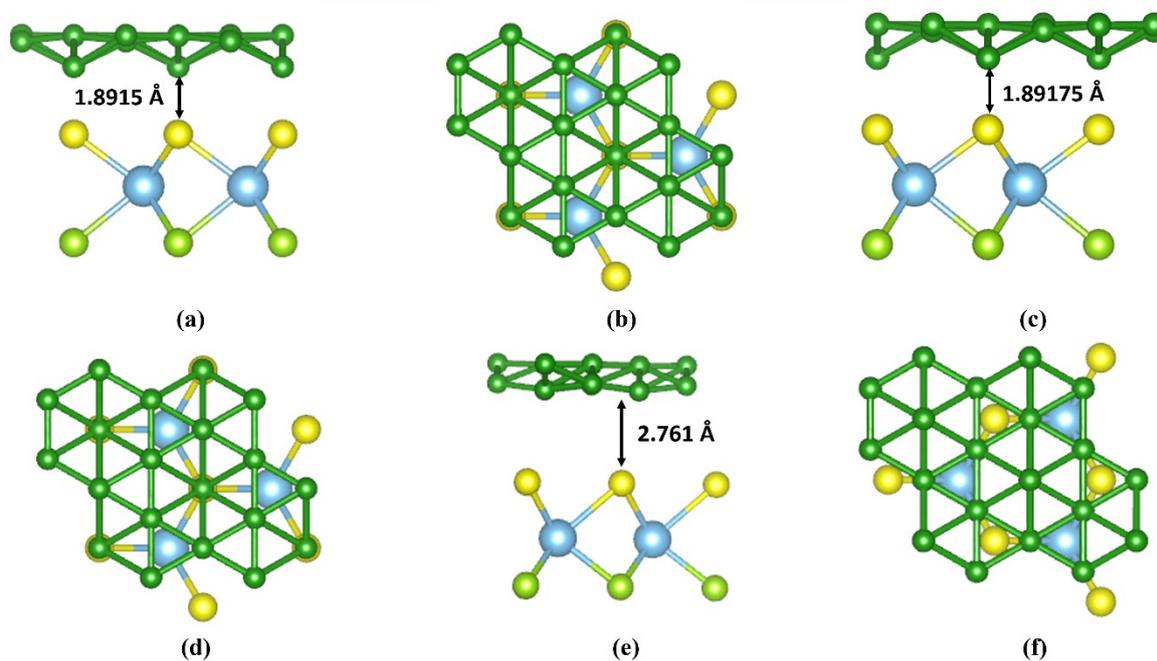
#### Stacking configurations

Stacking configurations as shown in **Figs. S1 and S2** were considered during stacking screening to identify structurally stable vdW heterostructures. However, lower binding energy configurations were excluded due to severe borophene distortion and non-vdW-like interlayer bonding, as observed from structural relaxation.

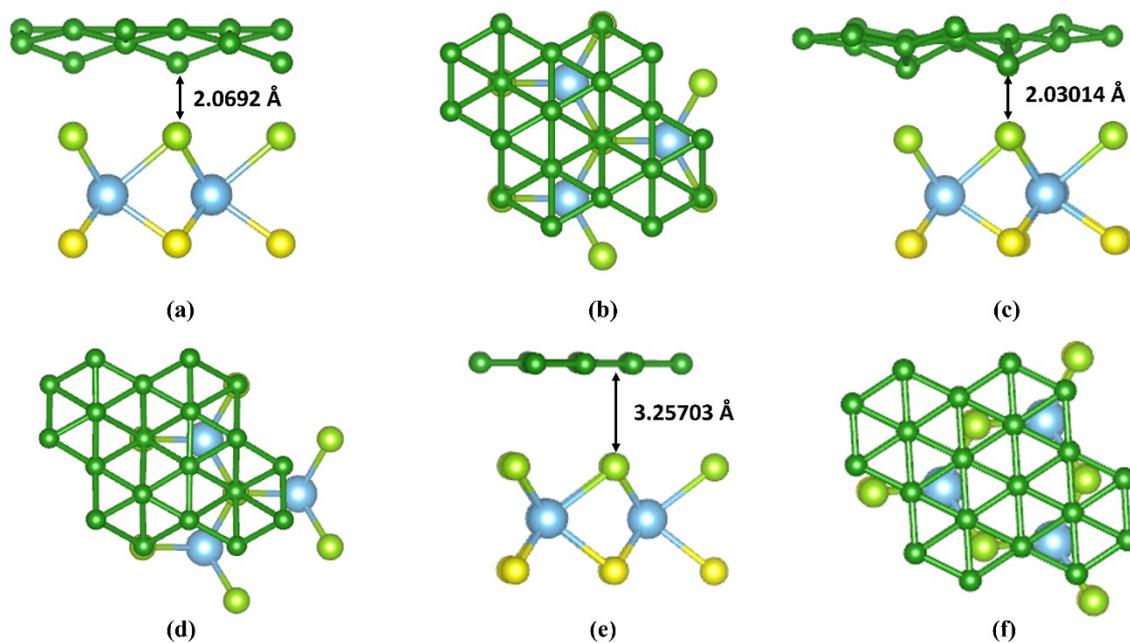
**Table S1.** Summary of investigated stacking configurations for TiSSe/H-borophene heterostructure for S-B and Se-B interfaces.

Interface	Initial B position	Binding energy	Interlayer distance (Å)	Structural outcome	Considered for analysis
S-B	Hollow / bridge	Reference	2.76100	Stable, minor borophene distortion	Yes
S-B	On top of S	Lower	1.89175	Significant borophene distortion.	No
S-B	On top of Ti	Lower	1.8915	Relaxes to B-on-S, distorted	No
Se-B	Hollow / bridge	Reference	3.25703	Stable, no borophene distortion	Yes
Se-B	On top of Se	Lower	2.03014	Significant borophene	No

				distortion	
Se-B	On top of Ti	Lower	2.06929	Relaxes to B-on-Se, distorted	No

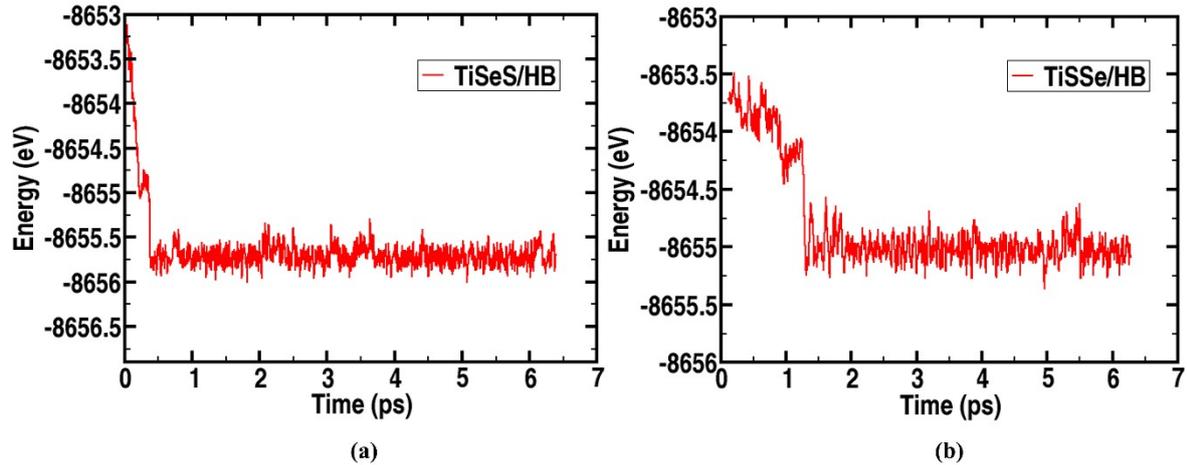


**Fig. S1.** Side and top views of the TiSSe/H-borophene heterostructure showing various stacking configurations for the S-B interface, including B atoms positioned (a, b) above Ti atoms, (c, d) S atoms, and (e, f) hollow/bridge sites.



**Fig. S2.** Side and top views of the TiSSe/H-borophene heterostructure showing various stacking configurations for the Se–B interface, including B atoms positioned **(a, b)** above Ti atoms, **(c, d)** S atoms, and **(e, f)** hollow/bridge sites. These configurations were considered during stacking screening to identify structurally stable vdW heterostructures.

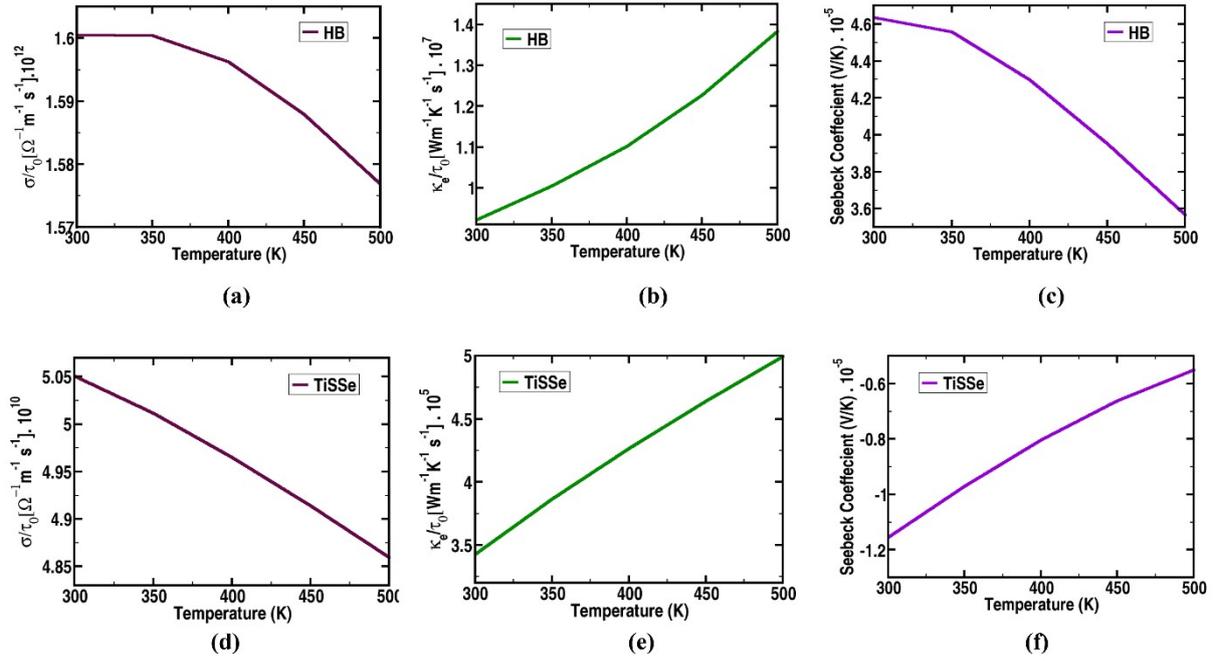
## 2. AIMD calculations



**Fig. S3.** Variation of energy as a function of simulation time during AIMD run at 300 K of **(a)** S-B interface and **(b)** Se-B interface of TiSSe/H-borophene heterostructure.

## 3. Transport properties

To clarify the origin of carrier polarity and strengthen the thermoelectric discussion, transport properties of pristine-Janus monolayer and H-borophene were evaluated within the same BoltzTraP framework.



**Fig. S4.** Thermoelectric properties of constituent monolayers: **(a, d)** Electrical conductivity, **(b, e)** Thermal conductivity, and **(c, f)** Seebeck coefficient of pristine H-borophene, and Janus TiSSe monolayer, respectively.

In **Fig. S4 (a-c)**, it is shown that H-borophene exhibits extremely high electrical conductivity ( $\sim 1.60 \times 10^{12}$  S/m at 300 K), confirming its metallic character and strong carrier mobility. The Seebeck coefficient is small and positive ( $\sim +4.63 \times 10^{-5}$  V/K at 300 K), indicating hole-dominated transport consistent with metallic behavior. The electronic thermal conductivity increases with temperature and reaches a maximum value of  $\sim 1.38 \times 10^7$  W/m·K at 500 K, reflecting efficient electronic heat transport associated with high carrier concentration.

In contrast, **Fig. S4 (d-f)** shows the Janus TiSSe monolayer exhibiting lower electrical conductivity ( $\sim 5.05 \times 10^{10}$  S/m at 300 K), suggesting reduced carrier density relative to H-borophene. Its Seebeck coefficient is small and negative ( $\sim -1.15 \times 10^{-5}$  V/K at 300 K), indicating electron-dominated transport. The electronic thermal conductivity remains comparatively moderate ( $\sim 4.99 \times 10^5$  W/m·K at 500 K), consistent with its lower carrier mobility and weaker metallic response.

Overall, both monolayers exhibit metallic transport characteristics with small Seebeck coefficients. However, H-borophene demonstrates higher electrical and electronic thermal conductivities.