

Supporting Information

Strong direct-bandgap photoluminescence of suspended few-layer MoS₂ via interlayer decoupling

Jiahao Wu¹, Jinyan Huang², Juncong She^{1,}, and Shasha Li^{2,*}*

¹State Key Laboratory of Optoelectronic Materials and Technologies, Guangdong
Province Key Laboratory of Display Material and Technology, School of Electronics
and Information Technology, Sun Yat-sen University, Guangzhou 510275, People's
Republic of China

²School of Integrated Circuits, Sun Yat-sen University, Shenzhen 518107, People's
Republic of China

*Email: lishsh89@mail.sysu.edu.cn; shejc@mail.sysu.edu.cn

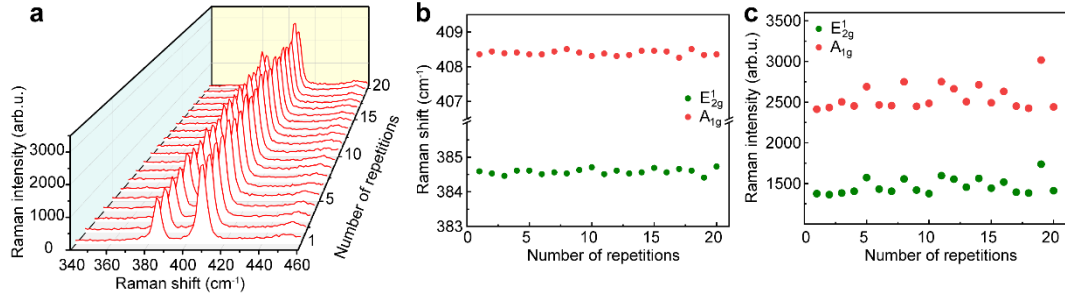


Figure S1. Raman measurements of MoS₂. (a) Raman spectra of MoS₂ obtained with a laser power of 6 μ W. The exposure time of each measurement is 10 s. (b) Raman shifts of the E_{12g} and A_{1g} peaks. (c) Raman intensities of the E_{12g} and A_{1g} peaks. It can be seen that the Raman measurements maintain good repeatability over multiple measurements. The results suggest that the structural quality of the tested MoS₂ samples can be well maintained after the optical measurements with such a low power.

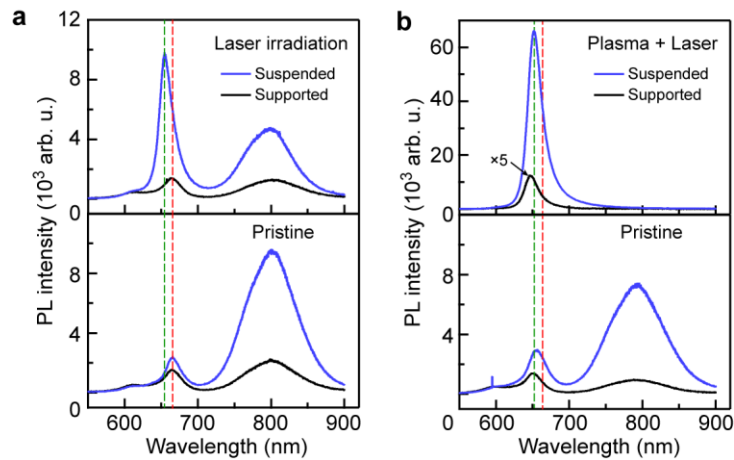


Figure S2. Enhanced radiative emission of A excitons. (a) PL spectra from both suspended and supported regions of MoS₂ before (bottom) and after (top) laser irradiation. (b) PL spectra from MoS₂ before (bottom) and after (top) the combined plasma and laser treatment. The green and red dashed line indicate the positions of A excitons (~654 nm) and trions (~663 nm) extracted from Gaussian fitting. The as-

prepared MoS₂ flake for laser treatment was highly n-doped, which dominated the emission spectrum and made the A exciton peak nearly indistinguishable. Following laser irradiation, a pronounced enhancement of the A exciton emission was observed, indicating a significant reduction in electron doping. A similar enhancement, accompanied by the complete suppression of indirect bandgap emission, was also achieved with the combined plasma and laser treatment.

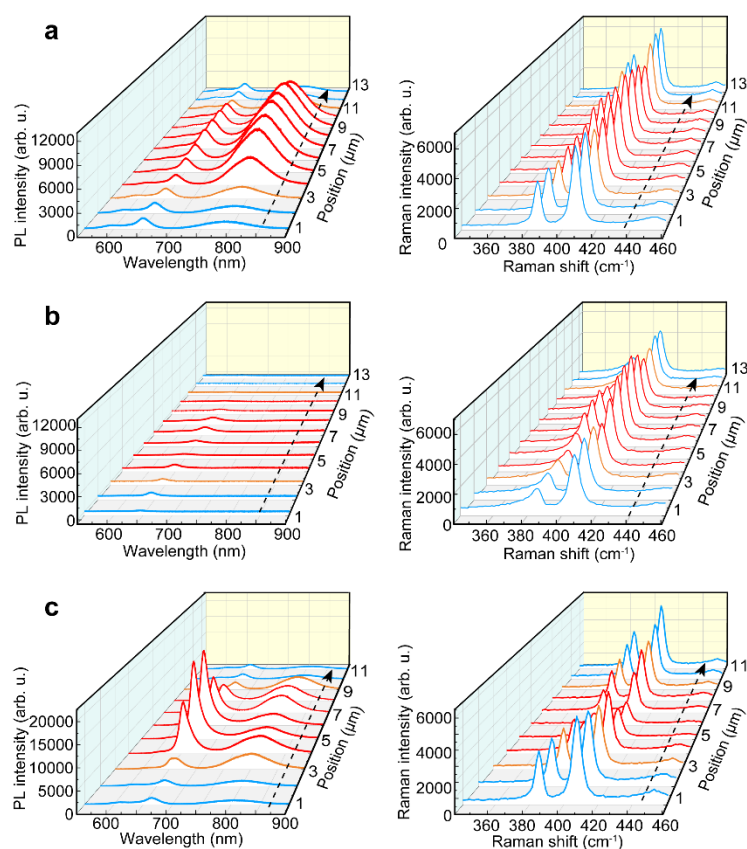


Figure S3. Optical characterizations. (a–c) PL and Raman spectra of the pristine bilayer MoS₂ (a), plasma-treated bilayer MoS₂ (b), and laser-treated bilayer MoS₂ (c). The PL and Raman spectra are shown in the left and right panels, respectively.

Table S1 Average values and standard deviations of Raman peak frequencies and their difference for different types of MoS₂ samples.

| MoS ₂ sample | ω_E (cm ⁻¹) | ω_A (cm ⁻¹) | $\Delta\omega$ (cm ⁻¹) |
|-------------------------|--------------------------------|--------------------------------|------------------------------------|
| Pristine 1L | 385.94 \pm 0.14 | 405.91 \pm 0.05 | 19.97 \pm 0.28 |
| Pristine sus 1L | 385.99 \pm 0.24 | 405.98 \pm 0.20 | 19.99 \pm 0.18 |
| Pristine 2L | 385.63 \pm 0.11 | 407.89 \pm 0.04 | 22.26 \pm 0.13 |
| Pristine sus 2L | 385.28 \pm 0.03 | 407.67 \pm 0.05 | 22.39 \pm 0.04 |
| Plasma 2L | 385.61 \pm 0.13 | 407.65 \pm 0.05 | 22.04 \pm 0.15 |
| Plasma sus 2L | 384.98 \pm 0.16 | 407.37 \pm 0.10 | 22.39 \pm 0.12 |
| Laser 2L | 385.64 \pm 0.07 | 407.94 \pm 0.07 | 22.3 \pm 0.10 |
| Laser sus 2L | 385.40 \pm 0.19 | 407.88 \pm 0.10 | 22.48 \pm 0.20 |
| (Plasma + laser) 2L | 385.94 \pm 0.14 | 408.06 \pm 0.05 | 22.12 \pm 0.16 |
| (Plasma + laser) sus 2L | 386.20 \pm 0.24 | 406.66 \pm 0.20 | 20.46 \pm 0.34 |

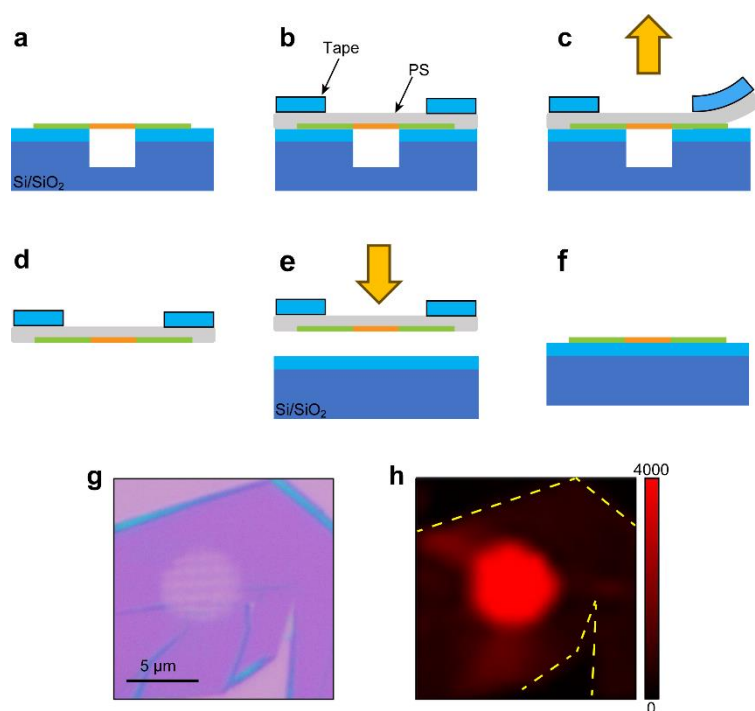


Figure S4 Transfer process. (a) The treated MoS₂ sample on the Si/SiO₂ substrate with micro-holes. (b) The MoS₂ sample was spin-coated with the polystyrene (PS) solution, forming a PS film after drying in air. (c) The PS film was exfoliated by tape with the MoS₂ sample attached on the PS film. (d–f) The PS film was transferred to a bare

Si/SiO₂ substrate and removed by toluene solution, leaving the MoS₂ sample on the substrate. (g,h) Optical microscopy (g) and PL mapping (h) images of the treated MoS₂ transferred from the hole region to a bare Si/SiO₂ substrate.

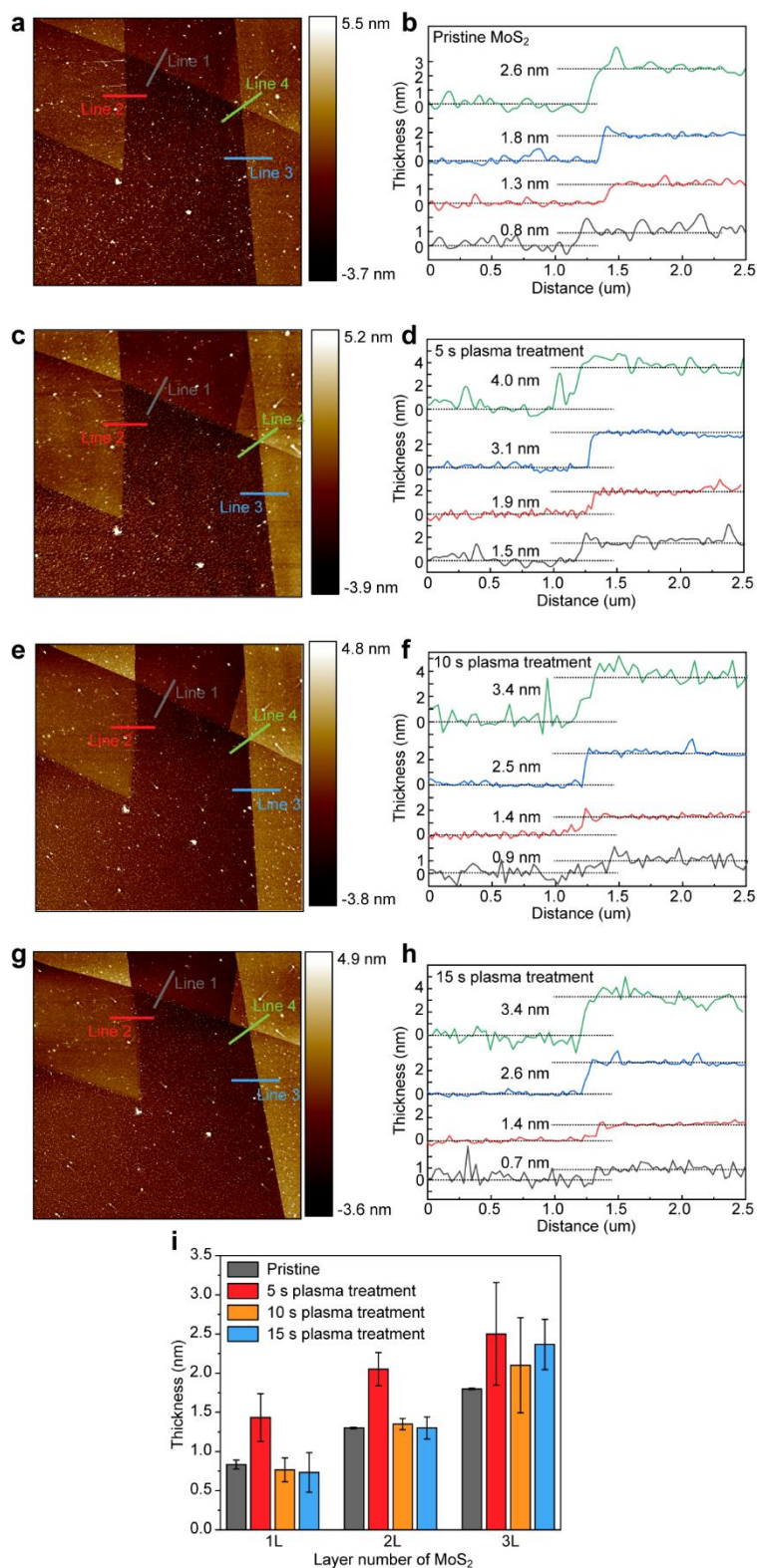


Figure S5 AFM characterization of plasma-treated MoS₂. (a,b) AFM image (a) of the pristine MoS₂ and height profiles (b) along the lines in (a). (c–h) AFM images and corresponding height profiles obtained from the MoS₂ samples after plasma treatments for 5 s (c,d), 10 s (e,f), and 15 s (g,h). (i) Histogram summarizing the thickness variations of the MoS₂ samples after plasma treatments.

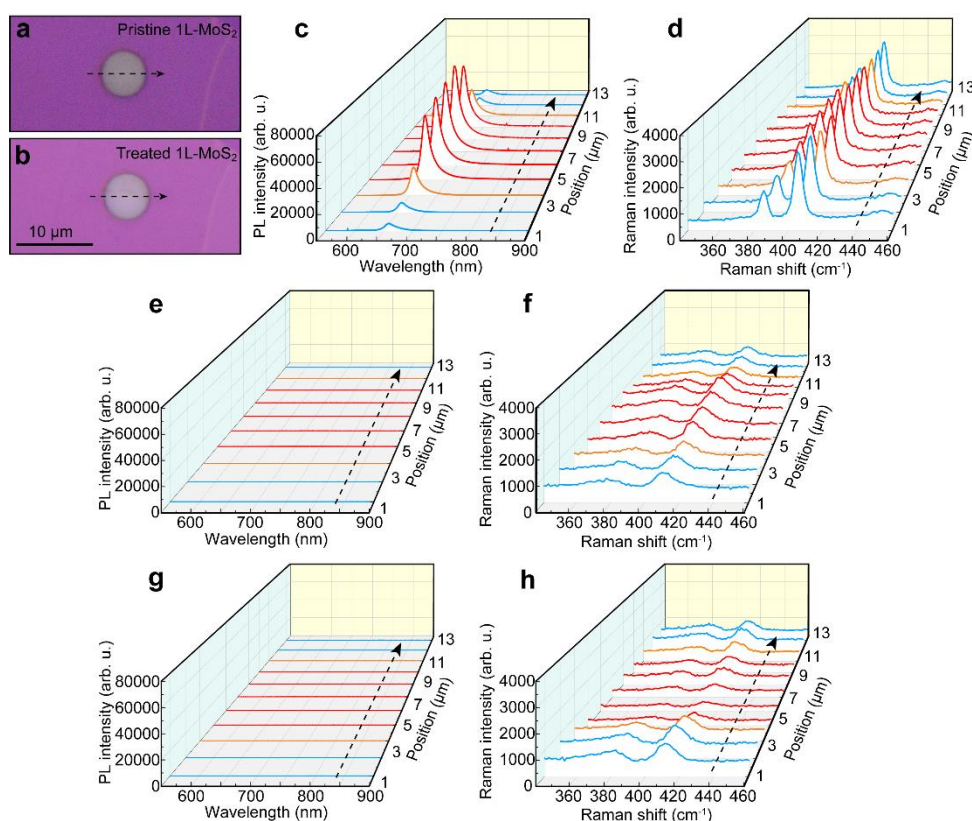


Figure S6 Optical characterizations on the monolayer MoS₂. (a, b) Optical microscopy images of monolayer MoS₂ before (a) and after (b) the plasma and laser treatments. (c, d) PL (c) and Raman (d) spectra of the pristine monolayer MoS₂ sample. The PL emission of suspended monolayer MoS₂ is significantly stronger than that of the supported one, which is consistent with other studies.^{1,2} The Raman spectra of pristine

monolayer MoS₂ show no obvious difference in the suspended and supported region. Here, Raman peaks corresponding to the A'₁ and E'₁ modes of monolayer MoS₂ due to D_{3h} point group are provided.³ (e, f) PL and Raman spectra of the plasma-treated monolayer MoS₂ sample. (g, h) PL and Raman spectra of the monolayer MoS₂ after plasma treatment followed by laser irradiation.

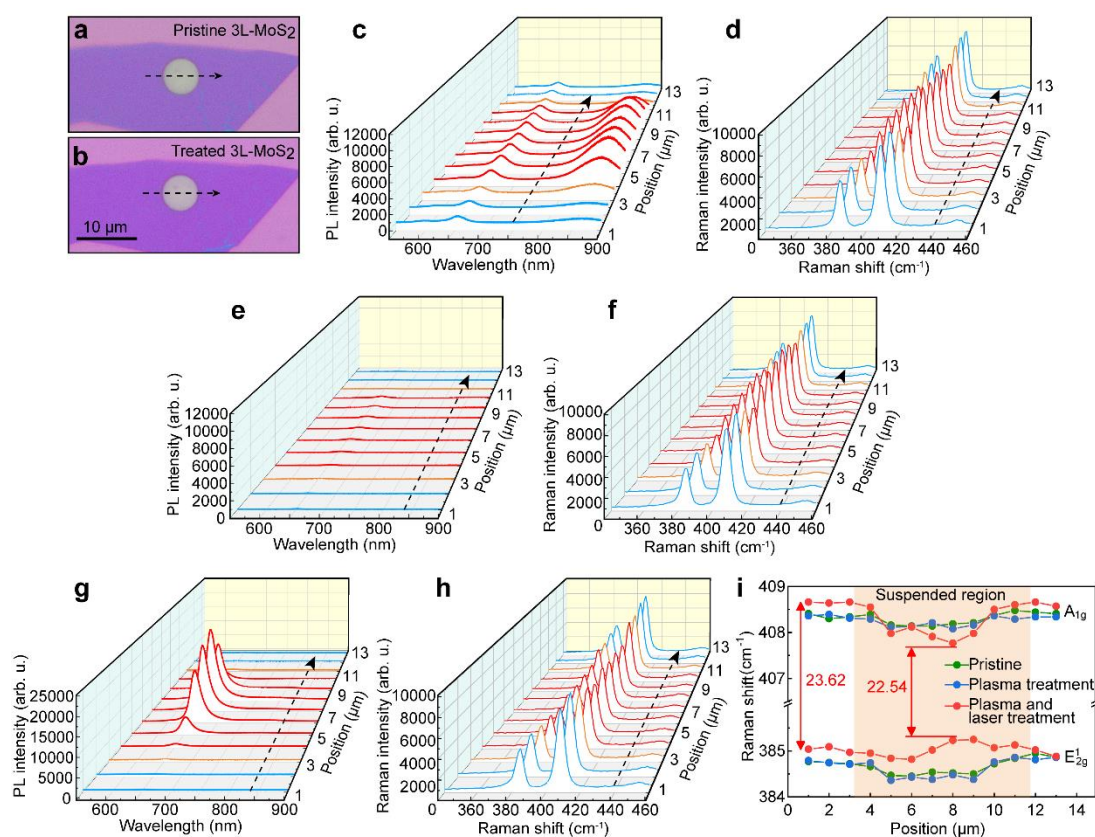


Figure S7. Optical characterizations on the trilayer MoS₂. (a, b) Optical microscopy images of trilayer MoS₂ before (a) and after (b) the plasma and laser treatments. (c, d) PL (c) and Raman (d) spectra of the pristine trilayer MoS₂ sample. (e, f) PL and Raman spectra of the plasma-treated trilayer MoS₂ sample. (g, h) PL and Raman spectra of the trilayer MoS₂ after plasma treatment followed by laser irradiation. (i) Raman shifts of

the E'_1 and A'_1 modes obtained from different types of MoS_2 samples. The wavenumber difference of the two dominant Raman modes decreased after the plasma and laser treatments, suggesting the emergence of interlayer decoupling.

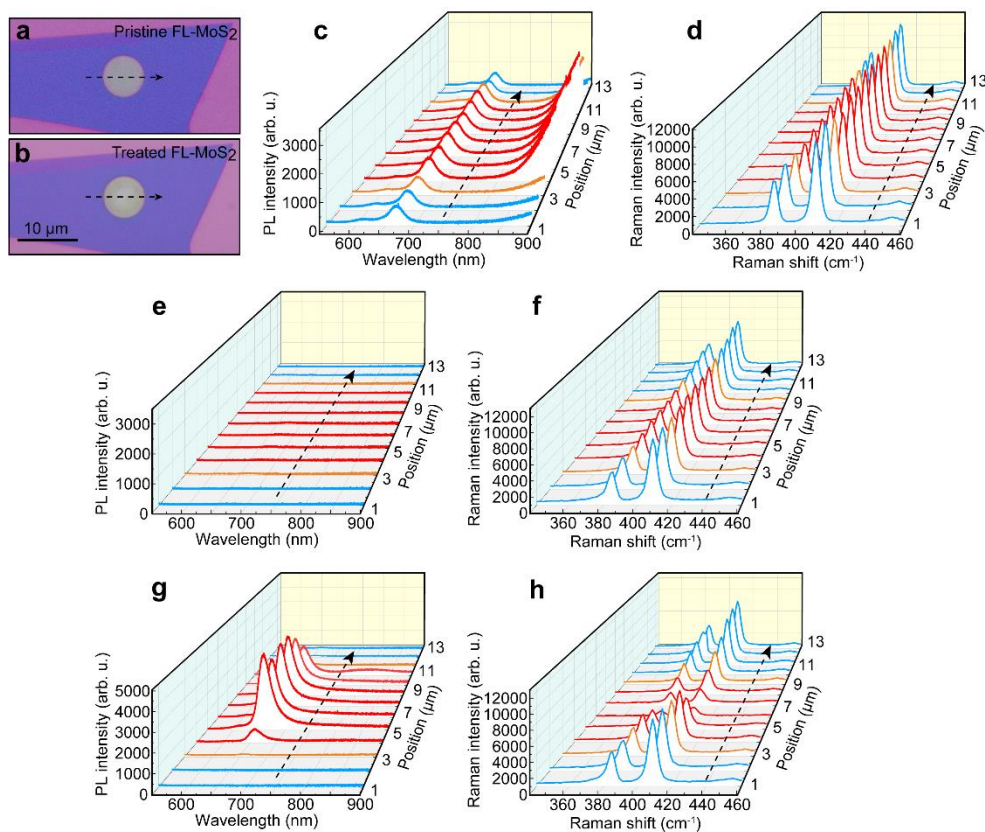


Figure S8. Optical characterizations on the few-layer MoS_2 . (a, b) Optical microscopy images of few-layer MoS_2 before (a) and after (b) the plasma and laser treatments. (c, d) PL (c) and Raman (d) spectra of the pristine few-layer MoS_2 sample. (e, f) PL and Raman spectra of the plasma-treated few-layer MoS_2 sample. (g, h) PL and Raman spectra of the few-layer MoS_2 after plasma treatment followed by laser irradiation.

Table S2 Comparative summary of key parameters in plasma treatment and laser irradiation for PL modulation of MoS₂

| Method | Layer number | Processing parameters | Mechanism for PL modulation | Enhancement factor | Ref. |
|---|--------------|---|--|--------------------|----------|
| O ₂ plasma | 1L | RF power: 5 W; Pressure: 37.5 mTorr | Healing of sulfur vacancies | ~10 | 4 |
| O ₂ plasma | 1L | RF power: 100 W; Pressure: 250 mTorr | Oxidation | quenching | 5 |
| O ₂ plasma | Few-layer | RF power: 20 W; Pressure: 200 mTorr; Treated time: 3 min | Layer decoupling due to ion intercalation | ~16 | 6 |
| O ₂ plasma | Few-layer | RF power: ~15 W; Pressure: 400 mTorr; Treated time: 3 min; DC bias: 15–20 V | Layer decoupling due to ion intercalation | ~20 (4L) | 7 |
| O ₂ plasma | Few-layer | RF power: 180 W; Treated time: 10 s | Shielding from a surface oxide layer | ~3 (3L) | 8 |
| O ₂ plasma | Few-layer | RF power: 5~30 W; Pressure: 285 mTorr; Treated time: 2 min | Layer decoupling due to ion intercalation | ~20 (2L) | 9 |
| O ₂ plasma | Few-layer | Plasma power: 40 W; Pressure: 1 Pa; treated time: 2–60 s | Oxidation and etching | quenching | 10 |
| O ₂ /Ar plasma | multilayer | Oxygen: Ar = 3:1; Pressure: ~5 mbar; Treated time: 14 s | Layer decoupling due to ion intercalation | ~1.2 (4L) | 11 |
| Laser irradiation | 1L | Power intensity: 4.5×10^5 W/cm ² ; Wavelength: 514.5 nm, Treated time: 10 s to 25 min | Healing of sulfur vacancies | ~5.5 | 12 |
| Laser irradiation | 1L | Power: 50 mW; Wavelength: 532 nm; Treated time: 10 s to 15 min; Under pure oxygen atmosphere | Healing of sulfur vacancies | 3.8 | 13 |
| Laser irradiation | 1L | Power intensity: 5.8×10^5 W/cm ² ; Wavelength: 532 nm; Treated time: 10 s to 25 min | Healing of sulfur vacancies | ~2 | 14 |
| Laser irradiation | Multi-layer | Power: 10–17mW; Wavelength: 532 nm; Treated time: 10 s to 15 min; Under pure oxygen atmosphere | Etching | / | 15 |
| Laser irradiation | Few-layer | Power: 10–40 mW; Wavelength: 532 nm; Treated time: 60 s | Etching and shielding from a surface oxide layer | ~3 (4L) | 16 |
| O ₂ plasma and laser irradiation | Few-layer | RF power: 5 W; Pressure: 37.5 mTorr; DC bias: 70-120 V; plasma treated time: 5 s; Laser power: ~460 μ W; Treated time: 1 s | Layer decoupling | ~30 | Our work |

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