

Supplementary Information

Multiscale Simulation of Eco-Friendly Perovskites under Space Radiation

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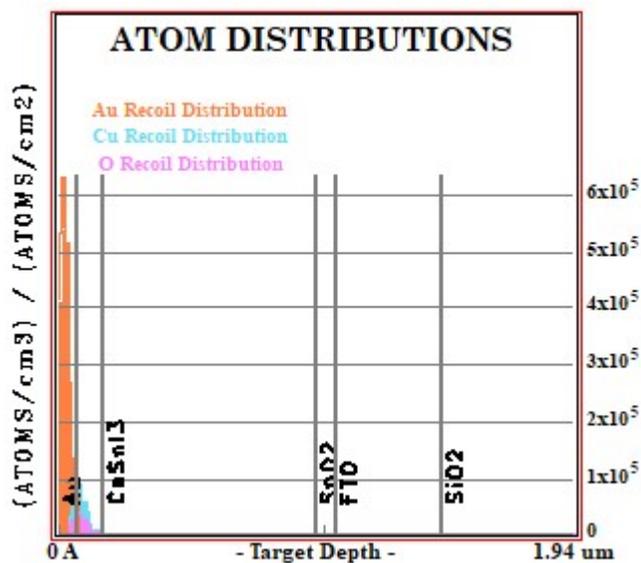


Figure S1: Element-Wise distribution of recoil atoms across the layers of CsSnI₃ device at an incident energy of 0.01 MeV.

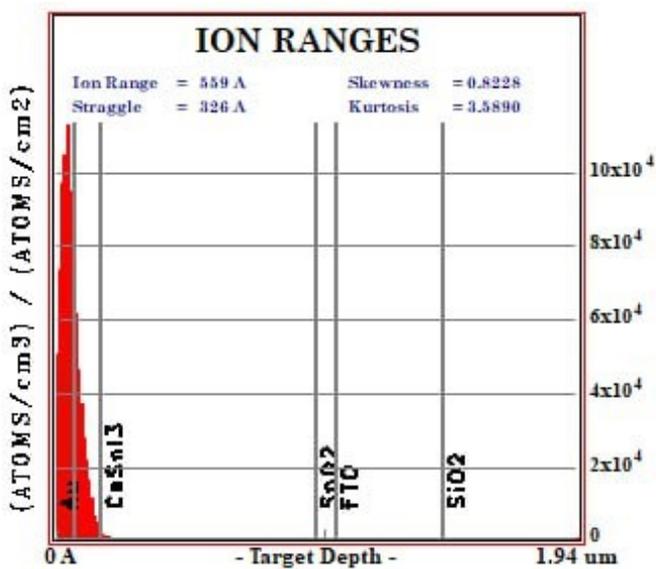


Figure S2: Ion distribution across the layers of CsSnI₃ device at an incident energy of 0.01 MeV.

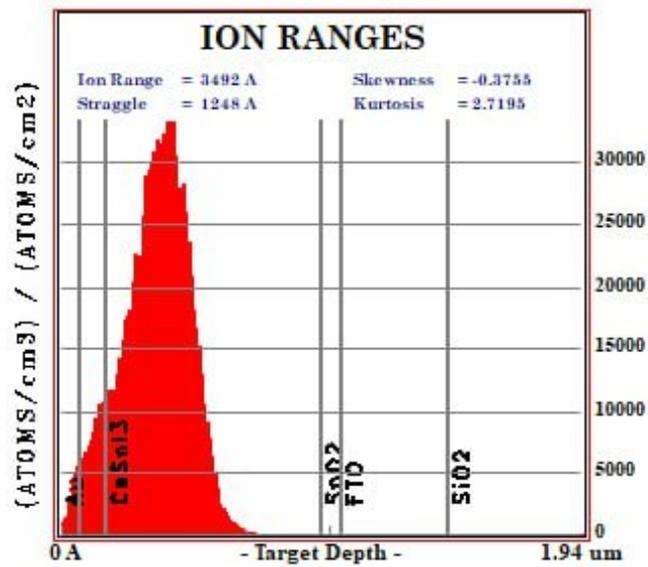


Figure S3: Ion distribution across the layers of CsSnI₃ device at an incident energy of 0.05 MeV.

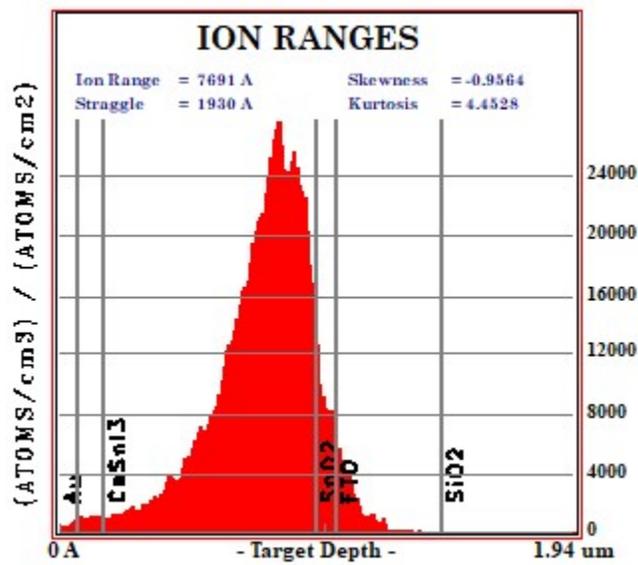


Figure S4: Ion distribution across the layers of CsSnI₃ device at an incident energy of 0.1 MeV.

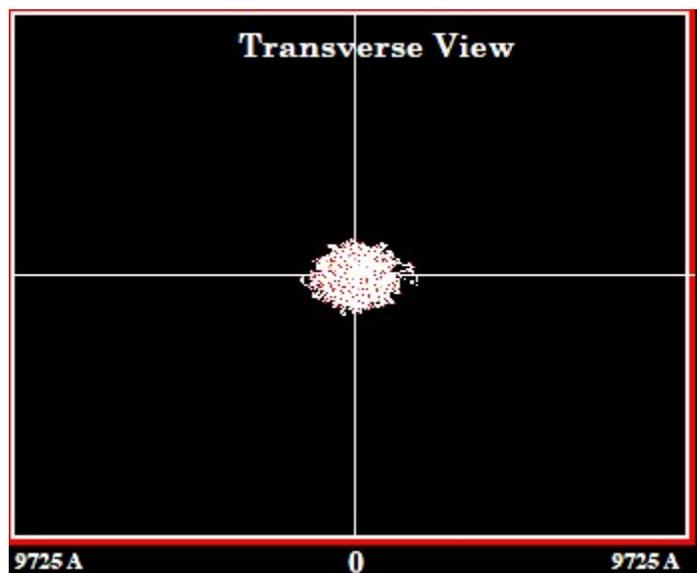


Figure S5: Proton trajectory in xy -plane for CsSnI_3 device at 0.01 MeV.

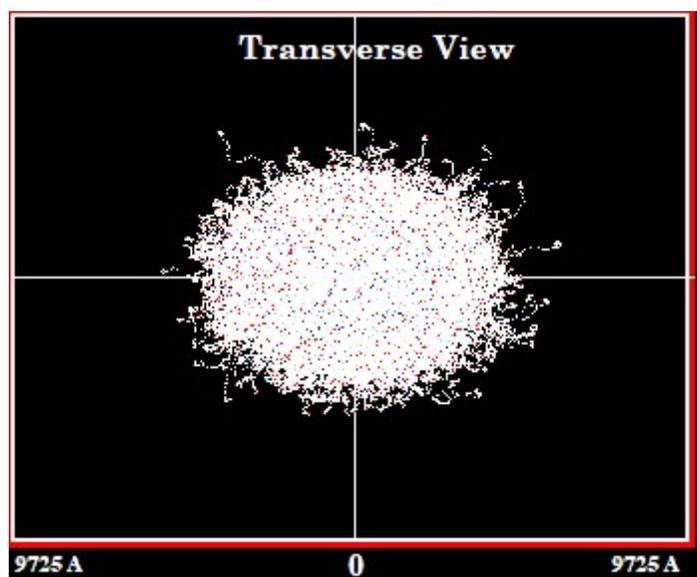


Figure S6: Proton trajectory in xy -plane for CsSnI_3 device at 0.05 MeV.

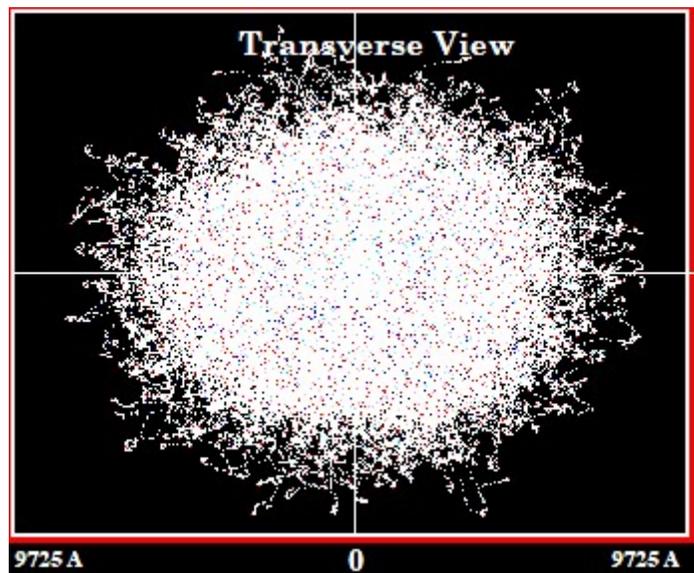


Figure S7: Proton trajectory in *xy*-plane for CsSnI₃ device at 0.1 MeV.

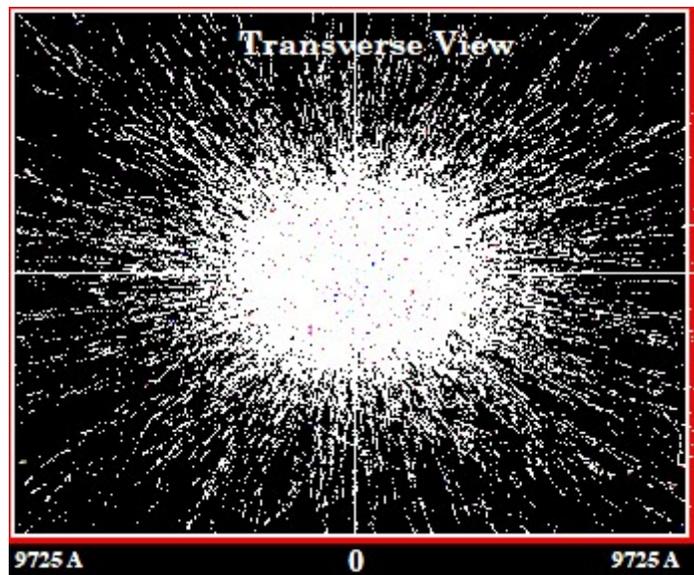


Figure S8: Proton trajectory in *xy*-plane for CsSnI₃ device at 0.3 MeV.



Figure S9: Proton trajectory in xy -plane for CsSnI_3 device at 0.5 MeV.

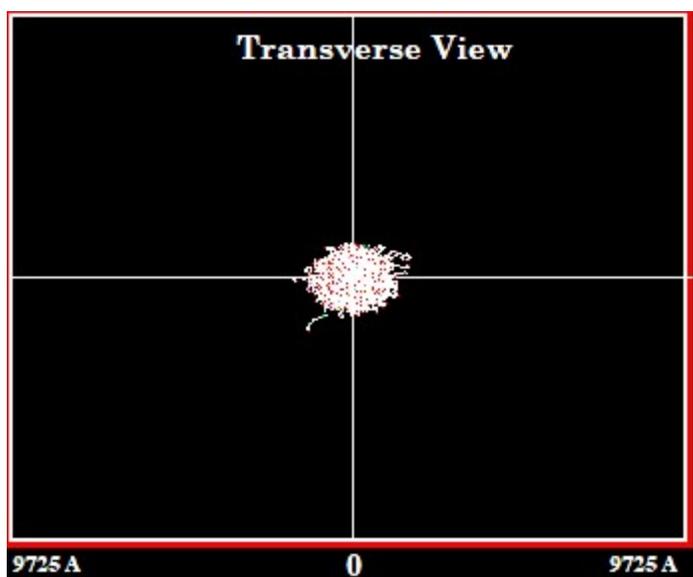


Figure S10: Proton trajectory in xy -plane for MASnI_3 device at 0.01 MeV.

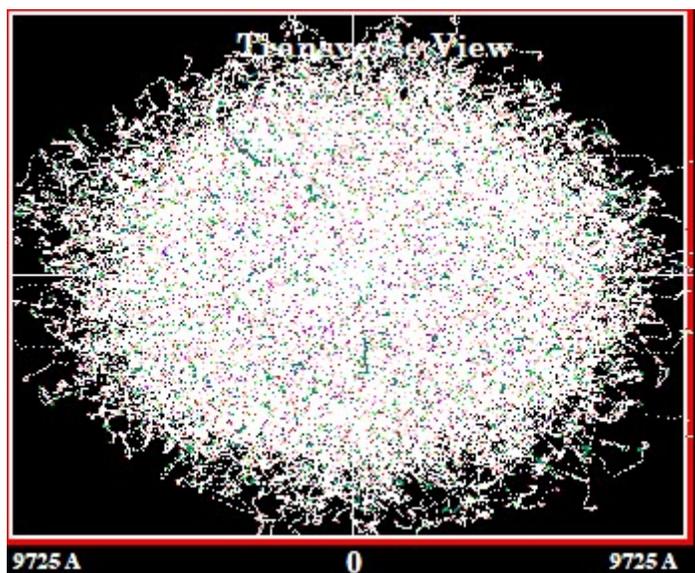


Figure S11: Proton trajectory in xy -plane for MASnI_3 device at 0.05 MeV

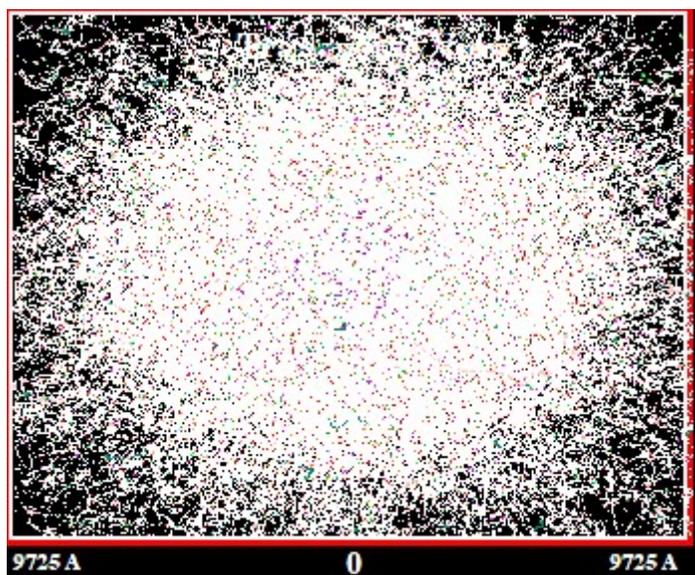


Figure S12: Proton trajectory in xy -plane for MASnI_3 device at 0.1 MeV.

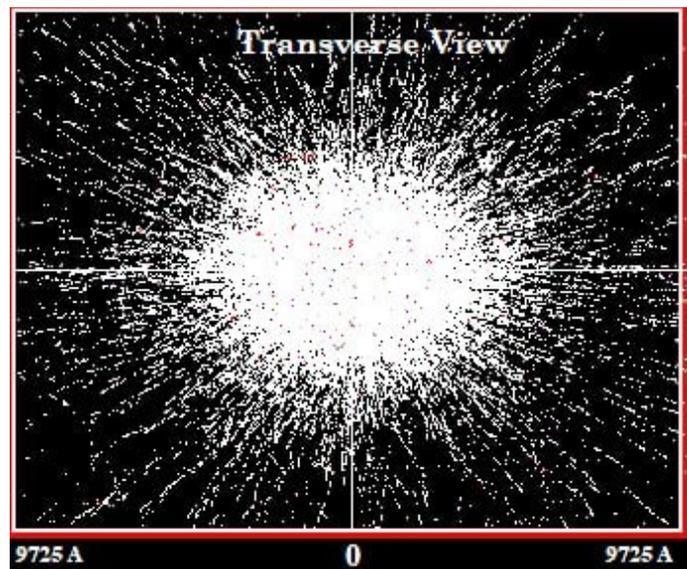


Figure S13: Proton trajectory in xy -plane for MASnI_3 device at 0.3 MeV.

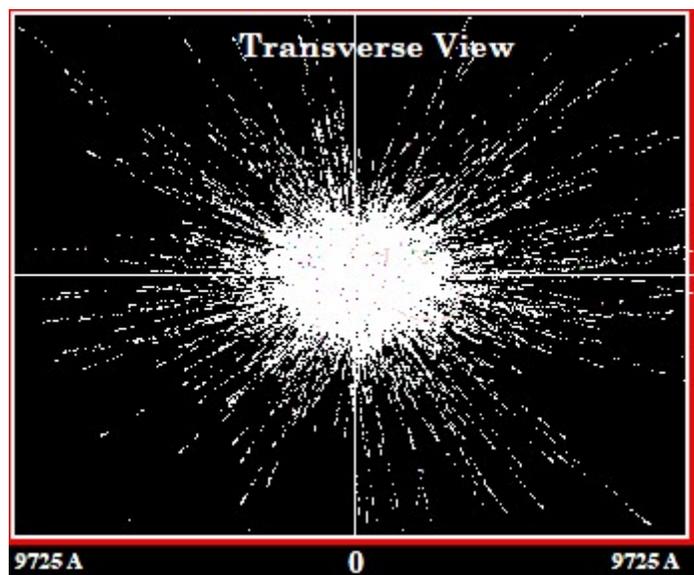


Figure S14: Proton trajectory in xy -plane for MASnI_3 device at 0.5 MeV.

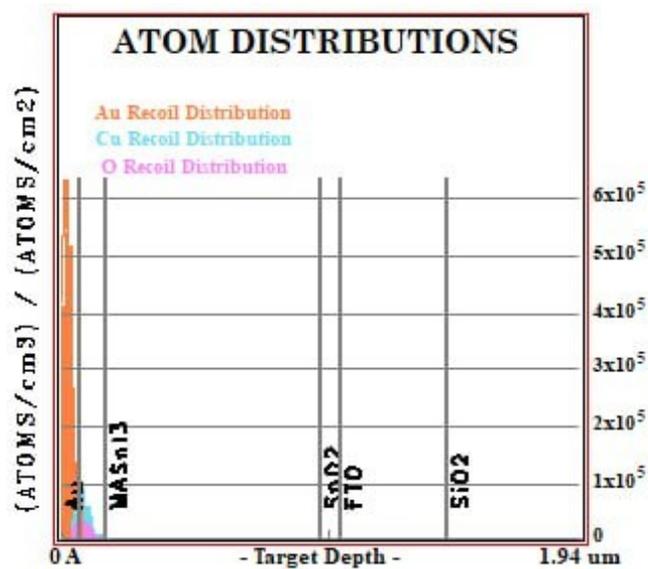


Figure S15: Element-Wise distribution of recoil atoms across the layers of MASnI₃ device at an incident energy of 0.01 MeV.

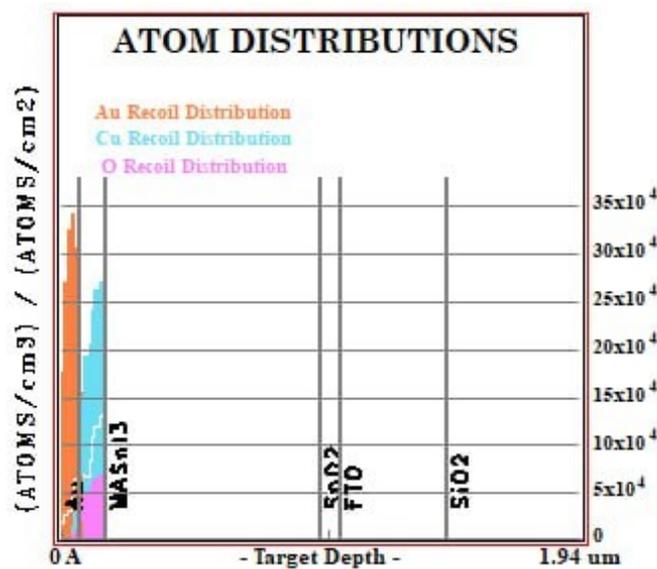


Figure S16: Element-Wise distribution of recoil atoms across the layers of MASnI₃ at an incident energy of 0.05 MeV.

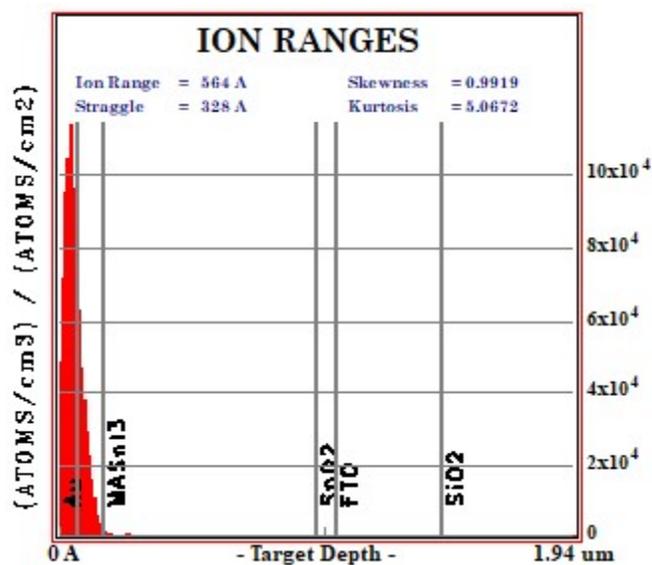


Figure S17: Ion distribution across the layers of MASNi₃ device at an incident energy of 0.01 MeV.

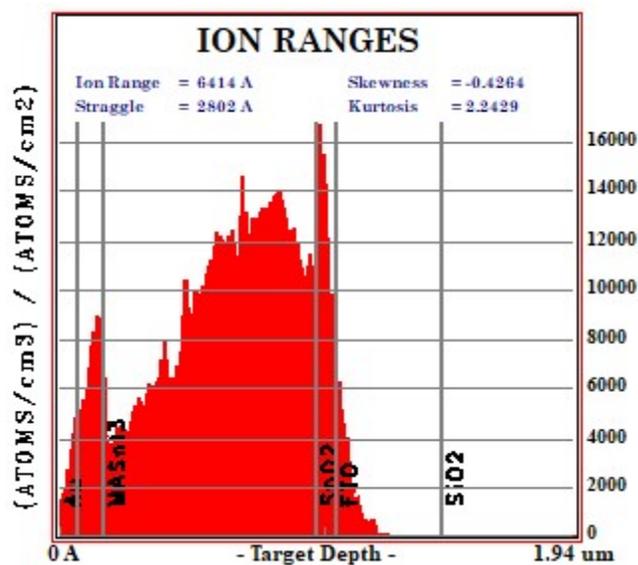


Figure S18: Ion distribution across the layers of MASNi₃ device at an incident energy of 0.05 MeV.

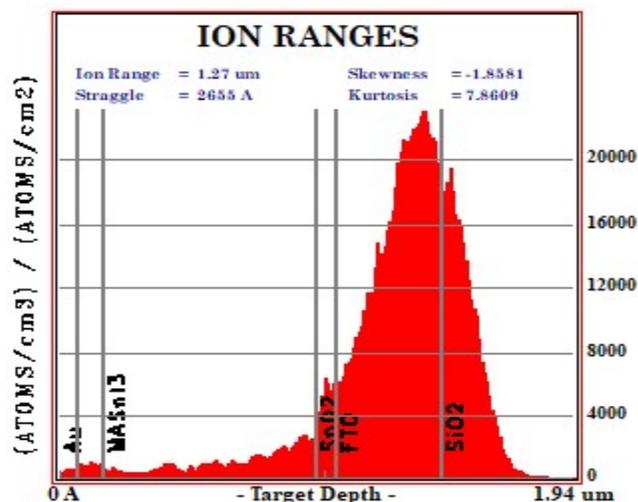


Figure S19: Ion distribution across the layers of MASnI₃ device at an incident energy of 0.1 MeV.

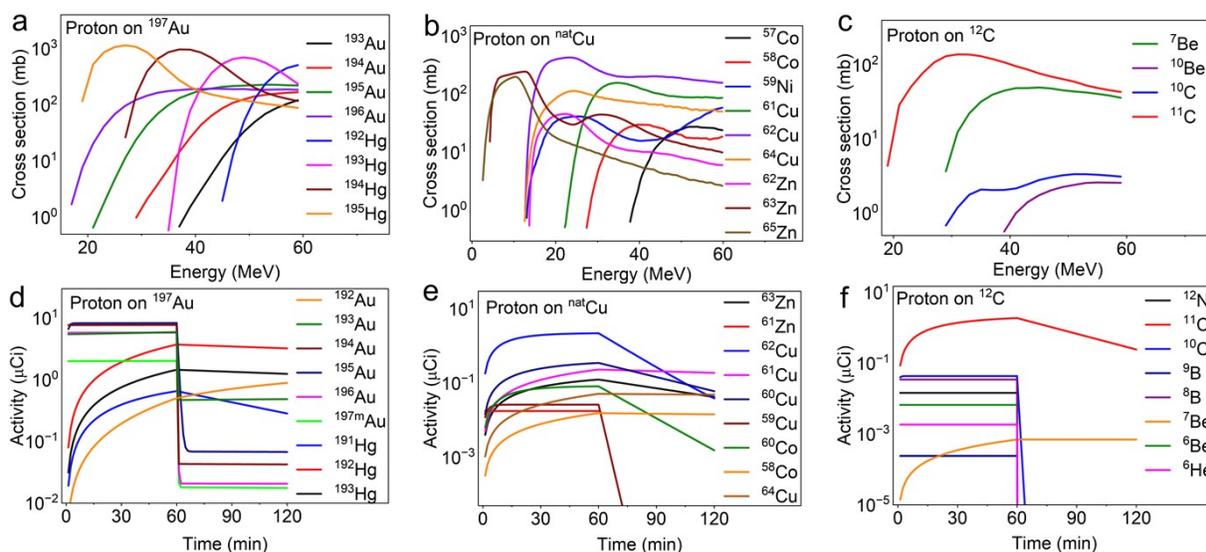


Figure S20: Analysis of proton-matter interactions using the TALYS nuclear reaction code. Simulated proton-induced reaction cross sections for (a) ¹⁹⁷Au, (b) ^{nat}Cu, and (c) ¹²C target isotopes as a function of incident proton energy (0–60 MeV). The corresponding time-dependent radioactivity (up to 25 hours) of (d) ¹⁹⁷Au, (e) ^{nat}Cu, and (f) ¹²C reaction products.

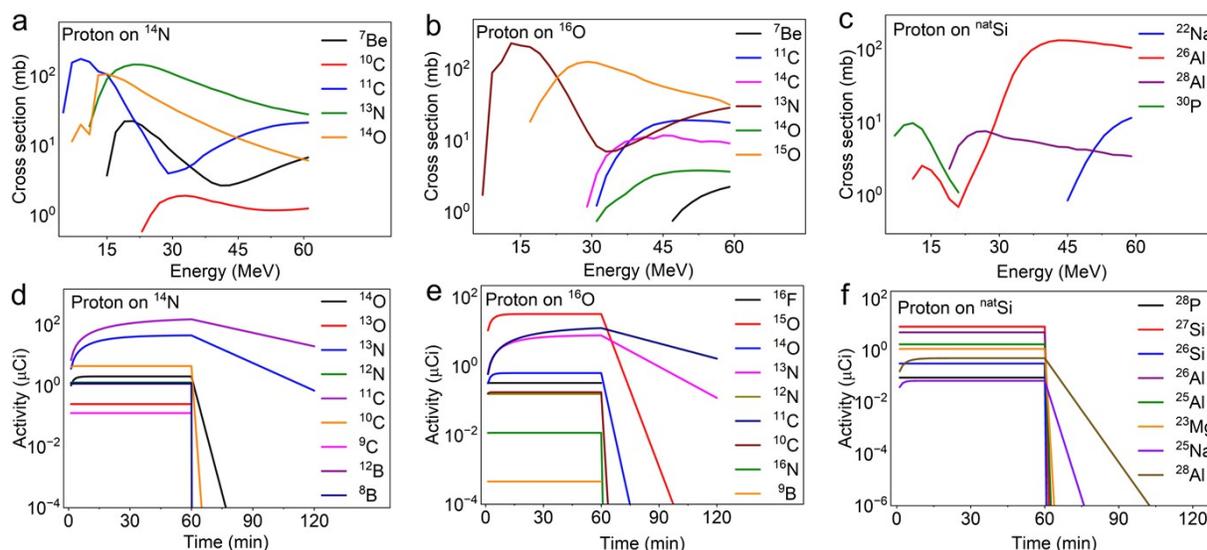


Figure S21: Analysis of proton-matter interactions using the TALYS nuclear reaction code. Simulated proton-induced reaction cross sections for (a) ^{14}N , (b) ^{16}O , and (c) $^{\text{nat}}\text{Si}$ target isotopes as a function of incident proton energy (0-60 MeV). The corresponding time-dependent radioactivity (up to 25 hours) of (d) ^{14}N , (e) ^{16}O , and (f) $^{\text{nat}}\text{Si}$ reaction products.

Note1:

A good overall agreement between the TALYS predicted excitation functions and the experimental cross section data available in the EXFOR database for the selected benchmark reactions. For $^{\text{nat}}\text{Sn}(p,x)^{117}\text{Sb}$, the corresponding EXFOR datasets are provided as the reference set [1-3]. For $^{133}\text{Cs}(p,3n)^{131}\text{Ba}$, the EXFOR datasets are listed in [4-8]. For $^{127}\text{I}(p,3n)^{125}\text{Xe}$, the EXFOR datasets are compiled in [9-15]. For $^{\text{nat}}\text{Cu}(p,x)^{63}\text{Zn}$, the associated EXFOR datasets are given in [16-27]. The TALYS simulations capture the main experimental features, including the reaction onset, the overall energy dependent behavior, and the approximate locations and magnitudes of the cross-section maxima. Minor discrepancies at the highest energies for some channels can be attributed to the expected limitations of global nuclear reaction modeling and to variations among experimental datasets. Overall, this benchmark comparison supports the reliability of the adopted TALYS settings and increases confidence in the calculated cross sections and activation estimates reported for the remaining reactions considered in this Supplementary Information.

Table S1. Interface parameters of SnO₂/Perovskite/Cu₂O structure.

Parameters/Interfaces	SnO ₂ /CsSnI ₃	CsSnI ₃ /Cu ₂ O	SnO ₂ /MASnI ₃	MASnI ₃ /Cu ₂ O
Defect type	Neutral	Neutral	Neutral	Neutral
Capture cross section for electrons (cm ²)	1 × 10 ⁻¹⁹	1 × 10 ⁻¹⁹	1 × 10 ⁻¹⁹	1 × 10 ⁻¹⁹
Capture cross section for holes (cm ²)	1 × 10 ⁻¹⁹	1 × 10 ⁻¹⁹	1 × 10 ⁻¹⁹	1 × 10 ⁻¹⁹
Energetic Distribution	Single	single	single	single
Reference for defect energy level E_t	Above the highest E_v	Above the highest E_v	Above the highest E_v	Above the highest E_v
Energy with respect to reference (eV)	0.600	0.600	0.600	0.600
Total density (1/cm ²)	1 × 10 ¹⁰	1 × 10 ¹⁰	1 × 10 ¹⁰	1 × 10 ¹⁰

Table S2: Comparison of photovoltaic parameters for simulated CsSnI₃ and MASnI₃ PSCs.

Device	V_{oc} (V)	J_{sc} (mA/cm ²)	FF (%)	PCE (%)
CsSnI ₃ PSC	0.758	37.25	67.64	14.05
MASnI ₃ PSC	0.821	40.22	71.71	17.41

Table S3: Summary of proton-induced nuclear reactions on different elements calculated using TALYS.

Target	Natural Abundance (%)	Nuclide	Reaction Channel	Threshold Energy (MeV)	Max. Cross Section (mb)
⁹⁷ Au	100	¹⁹³ Au	(p,2n+t)	21.64	174.89
		¹⁹⁴ Au	(p,n+t)	14.73	181.43
		¹⁹⁵ Au	(p,t)	6.26	229.99
		¹⁹⁶ Au	(p,d)	5.87	193.69
		¹⁹² Hg	(p,6n)	40.47	516.89
		¹⁹³ Hg	(p,5n)	33.31	696.47
		¹⁹⁴ Hg	(p,4n)	24.07	968.09
		¹⁹⁵ Hg	(p,3n)	17.13	1146.23
		¹⁹⁶ Hg	(p,2n)	8.21	982.27
^{nat} Cu	⁶³ Cu: 69.17 ⁶⁵ Cu: 30.83	⁶⁰ Ni			151.27
		⁶¹ Ni			154.95
		⁶¹ Cu			156.90
		⁶² Cu			427.33
		⁶³ Zn	(p,x)	Non	243.97
		⁶⁴ Zn			161.18
		⁶⁵ Zn			199.00
¹³³ Cs	100	¹²⁹ Cs	(p,2n+t)	24.56	157.09
		¹³⁰ Cs	(p,n+t)	17.03	164.72
		¹³¹ Cs	(p,t)	77.34	235.12
		¹³² Cs	(p,d)	6.82	227.49
		¹²⁸ Ba	(p,6n)	44.17	225.34
		¹²⁹ Ba	(p,5n)	36.35	375.88
		¹³⁰ Ba	(p,4n)	26.00	658.46
		¹³¹ Ba	(p,3n)	18.45	928.86
		¹³² Ba	(p,2n)	8.55	963.34
		¹³³ Ba	(p,n)	1.31	360.82
^{nat} Sn	¹¹² Sn: 0.97 ¹¹⁴ Sn: 0.66 ¹¹⁵ Sn: 0.34 ¹¹⁶ Sn: 14.54 ¹¹⁷ Sn: 7.68 ¹¹⁸ Sn: 24.22 ¹¹⁹ Sn: 8.59 ¹²⁰ Sn: 32.58 ¹²² Sn: 4.63 ¹²⁴ Sn: 5.79	¹¹³ Sn			95.31
		¹¹⁴ Sb			65.13
		¹¹⁵ Sb			144.32
		¹¹⁶ Sb			147.82
		¹¹⁷ Sb	(p,x)	Non	297.70
		¹¹⁸ Sb			323.38
		¹¹⁹ Sb			376.99
		¹²⁰ Sb			58.22
		¹²³ Sb			63.66

^{127}I	100	^{123}I	(p,2n+t)	25.05	149.57
		^{124}I	(p,n+t)	17.48	169.66
		^{125}I	(p,t)	7.87	231.76
		^{126}I	(p,d)	6.97	233.58
		^{122}Xe	(p,6n)	45.12	160.02
		^{123}Xe	(p,5n)	37.09	298.11
		^{124}Xe	(p,4n)	26.52	567.37
		^{125}Xe	(p,3n)	18.85	876.77
		^{126}Xe	(p,2n)	8.75	997.81
		^{127}Xe	(p,n)	1.45	381.42
^{12}C	99	^6Li	(p, ^3He +a)	26.18	48.01
		^7Li	(p,2p+a)	26.68	28.12
		^7Be	(p,d+a)	26.06	50.19
		^8Be	(p,p+a)	7.98	103.47
		^9Be	(p,p+ ^3He)	28.48	11.88
		^9B	(p,a)	8.18	261.42
		^{10}B	(p, ^3He)	21.34	52.26
		^{11}B	(p,2p)	17.29	96.62
		^{11}C	(p,d)	17.88	140.35
^{12}N	(p,n)	19.64	36.35		
^{14}N	99.6	^8Be	(p, ^3He + a)	13.02	88.11
		^9B	(p, d + a)	19.10	28.77
		^{10}B	(p, p + a)	12.44	53.64
		^{11}B	(p, p + ^3He)	22.22	27.11
		^{11}C	(p,a)	3.13	177.74
		^{12}C	(p, ^3He)	5.11	188.44
		^{13}C	(p,2p)	8.09	154.29
		^{13}N	(p,d)	8.92	148.04
		^{14}O	(p,n)	6.35	108.06
^{16}O	99.76	^8Be	(p,2a)	15.44	33.36
		^{11}B	(p,2p + a)	24.57	15.78
		^{11}C	(p,d + a)	25.14	14.93
		^{12}C	(p,p + a)	7.61	246.14
		^{13}C	(p, p + ^3He)	24.23	56.45
		^{13}N	(p,a)	5.56	231.83
		^{14}N	(p, ^3He)	16.20	162.62
		^{15}N	(p,2p)	12.89	128.34
		^{15}O	(p,d)	14.28	123.38
		^{16}F	(p,n)	17.21	18.56
		$^{\text{nat}}\text{Si}$	^{28}Si : 92.23 ^{29}Si : 4.68 ^{30}Si : 3.08	^{23}Na	
^{24}Mg					147.76
^{25}Mg					94.35
^{26}Mg					90.61
^{25}Al	(p,x)			Non	54.34
^{26}Al					135.15
^{27}Al					376.63
^{27}Si			126.79		

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