

Supplementary Information

Surface Phosphating Activation of PtMo₆-Ni-BDC Nanosheets for Enhanced Hydrogen Evolution Reaction

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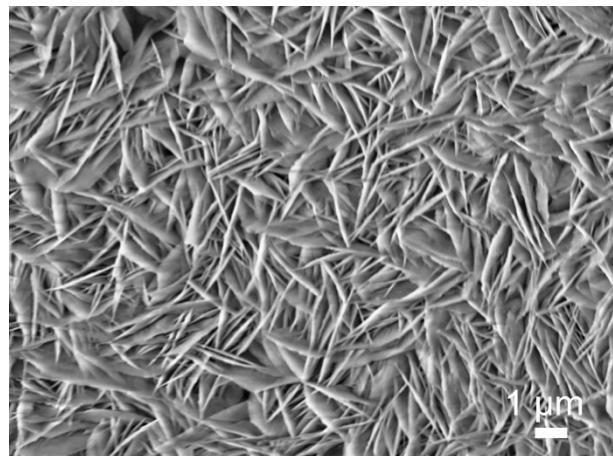


Figure S1. SEM image of Ni-BDC nanosheets grown on NF.

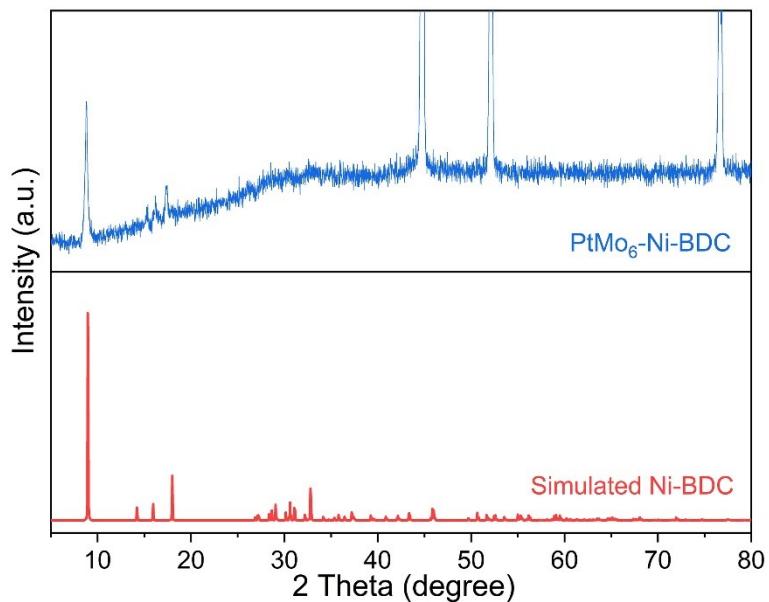


Figure S2. XRD spectra of $\text{PtMo}_6\text{-Ni-BDC}$ on NF and simulated Ni-BDC.

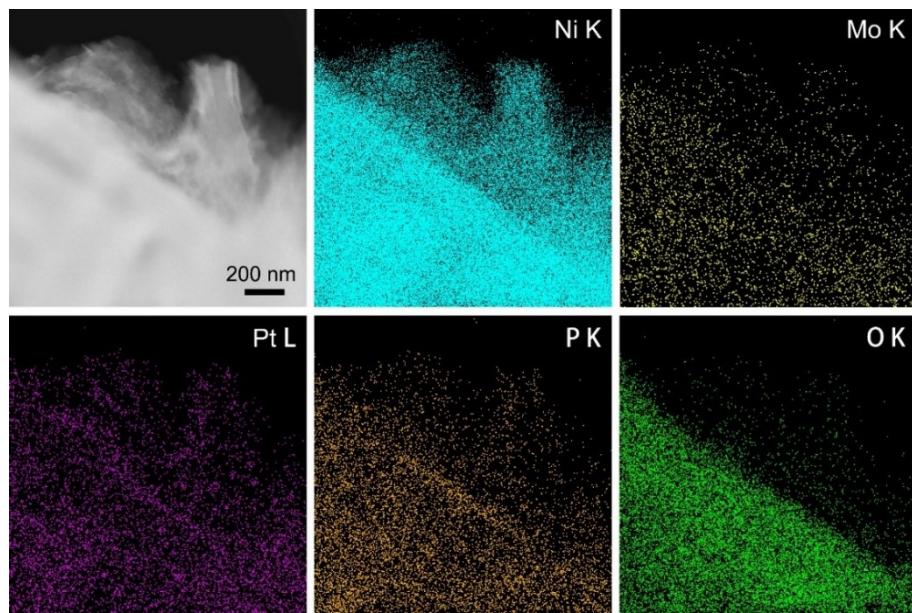


Figure S3. Mapping images of corresponding elements of Ni, Mo, Pt, P, and O of P-PtMo₆-Ni-BDC ultrasonicated from NF.

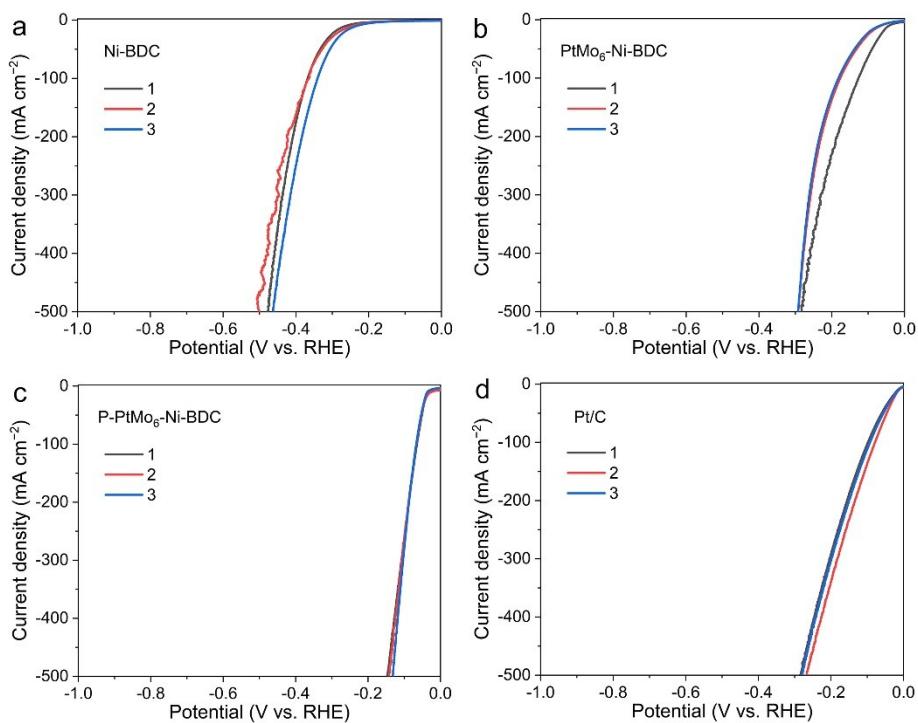


Figure S4. LSV curves of Ni-BDC, PtMo₆-Ni-BDC, P-PtMo₆-Ni-BDC, and Pt/C on NF measured three times for HER in 1.0 M KOH.

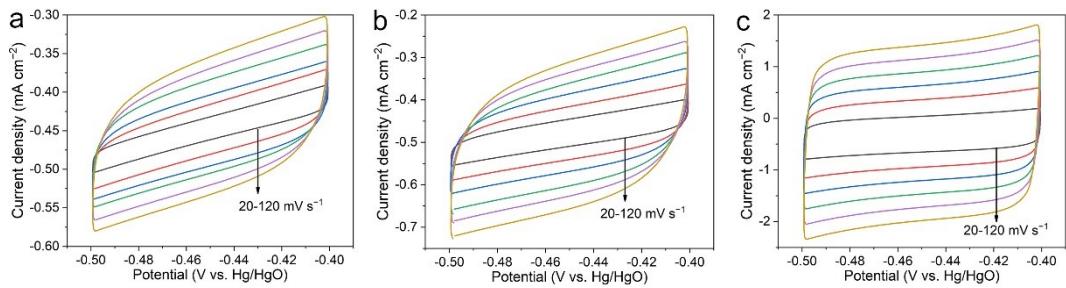


Figure S5. CV curves measured in a non-faradic region of -0.5 to -0.4 V vs. Hg/HgO at different scan rates ($20, 40, 60, 80, 100$, and 120 mV s^{-1}) in 1.0 M KOH. (a) Ni-BDC, (b) PtMo₆-Ni-BDC, and (c) P-PtMo₆-Ni-BDC.

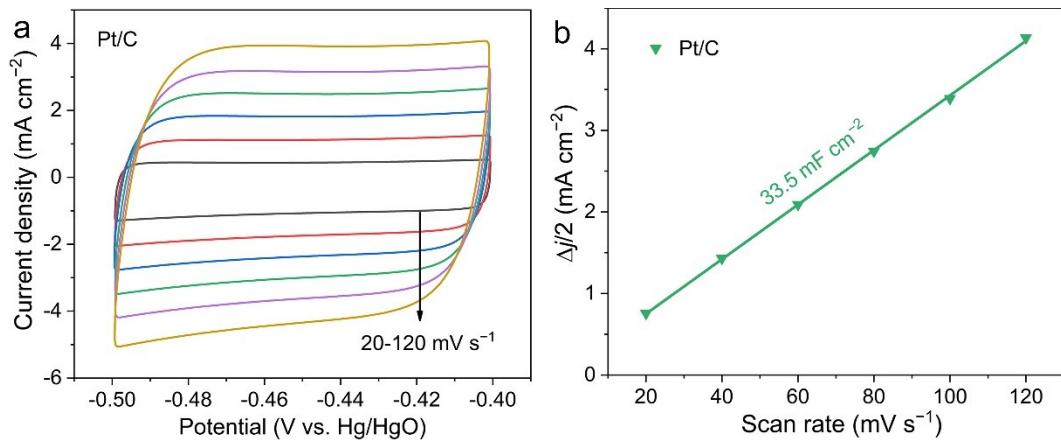


Figure S6. (a) CV curves of Pt/C measured in a non-faradic region of -0.5 to -0.4 V vs. Hg/HgO at different scan rates (20 - 120 mV s^{-1}) in 1.0 M KOH. (b) C_{dl} plot of Pt/C.

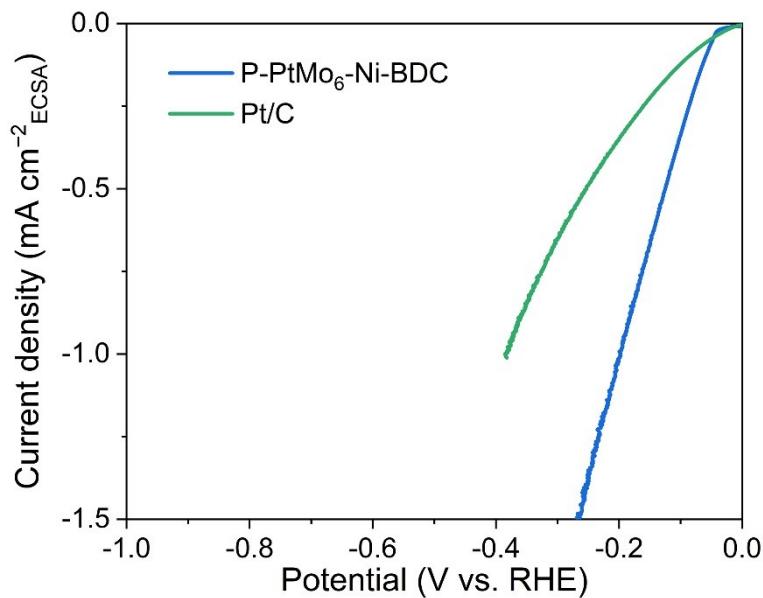


Figure S7. ECSA-normalized LSV curves of P-PtMo₆-Ni-BDC and Pt/C for HER.

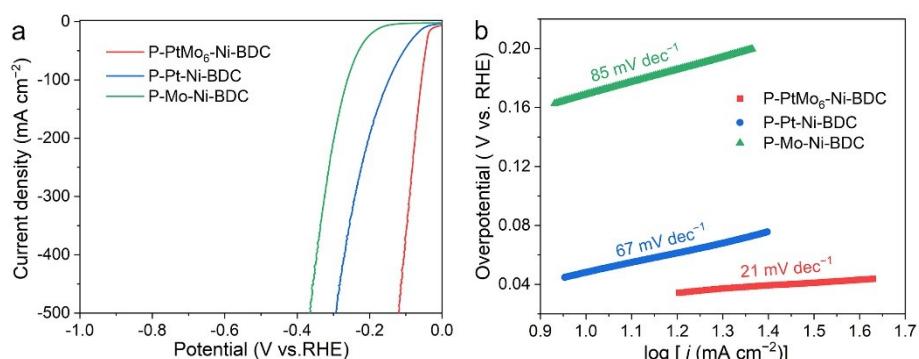


Figure S8. Electrochemical performance of P-PtMo₆-Ni-BDC, P-Pt-Ni-BDC, and P-Mo-Ni-BDC prepared by soaking Ni-BDC/NF precursor in solutions of PtMo₆, K₂Pt(OH)₆, and K₂MoO₄, respectively, for HER measured in 1.0 M KOH. (a) LSV curves and (b) Tafel plots.

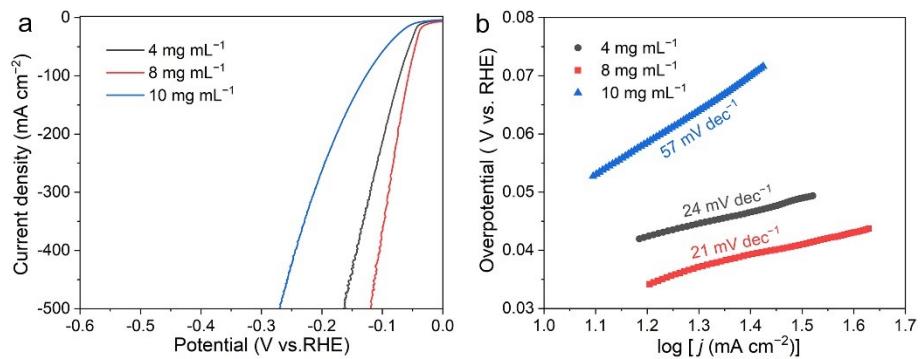


Figure S9. Electrochemical performance of P-PtMo₆-Ni-BDC prepared by soaking Ni-BDC/NF precursor in different concentrations of PtMo₆ solution (4, 8, and, 10 mg mL^{-1}) for HER measured in 1.0 M KOH. (a) LSV curves and (b) Tafel plots.

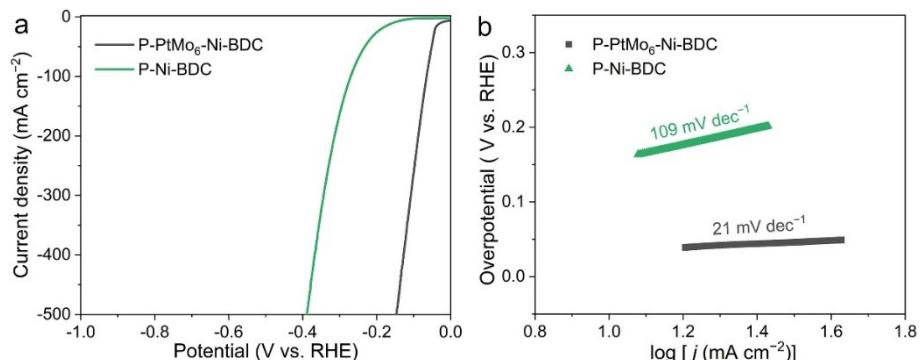


Figure S10. Electrochemical performance of P-PtMo₆-Ni-BDC and P-Ni-BDC for HER in 1.0 M KOH. (a) LSV curves and (b) Tafel plots.

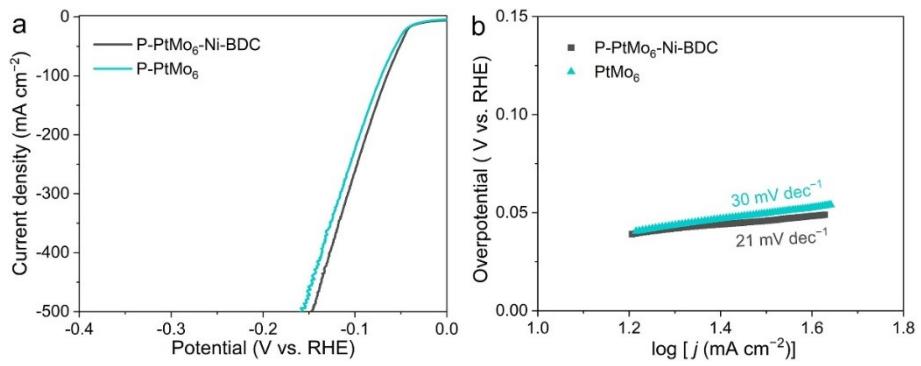


Figure S11. Electrochemical performance of P-PtMo₆ and P-PtMo₆-Ni-BDC for HER in 1.0 M KOH. (a) LSV curves and (b) Tafel plots.

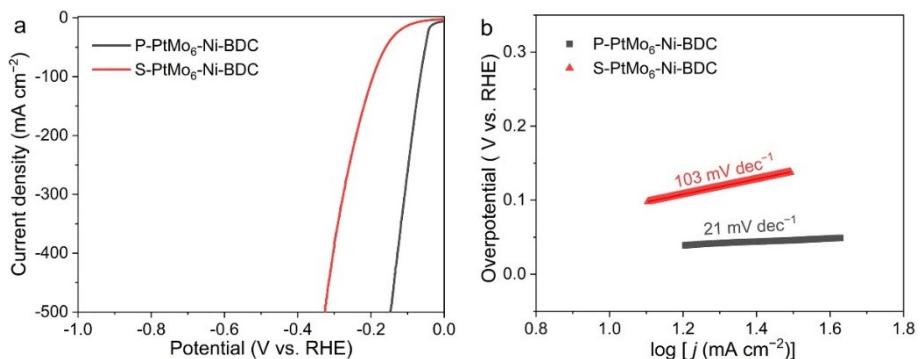


Figure S12. Electrochemical performance of P-PtMo₆-Ni-BDC and S-PtMo₆-Ni-BDC prepared via phosphidation and sulfidation treatments, respectively. (a) LSV curves and (b) Tafel plots.

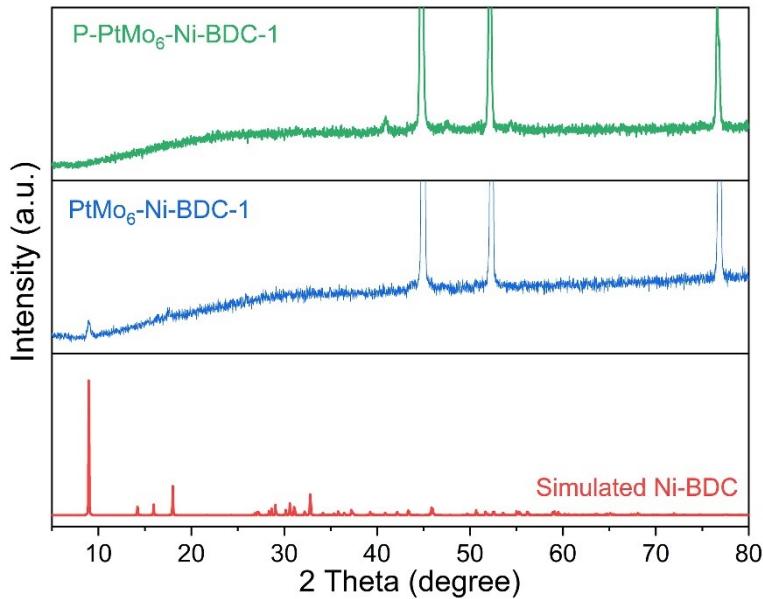


Figure S13. XRD pattern of $\text{PtMo}_6\text{-Ni-BDC-1}$, $\text{P-PtMo}_6\text{-Ni-BDC-1}$ on NF, and simulated Ni-BDC.

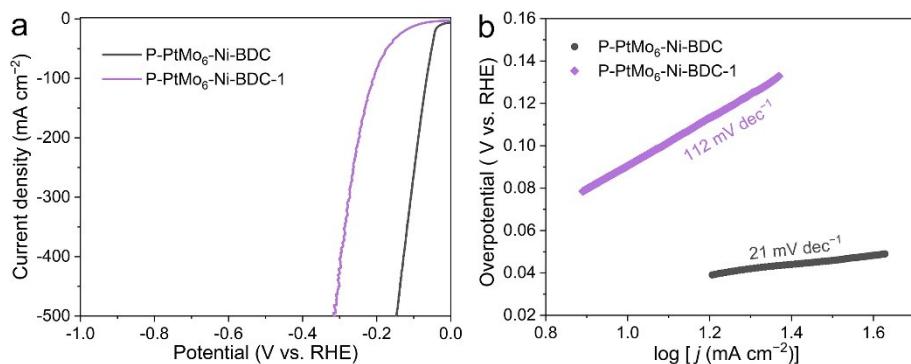


Figure S14. Electrochemical performance of $\text{P-PtMo}_6\text{-Ni-BDC-1}$ and $\text{P-PtMo}_6\text{-Ni-BDC}$ for HER. (a) LSV curves and (b) Tafel plots.

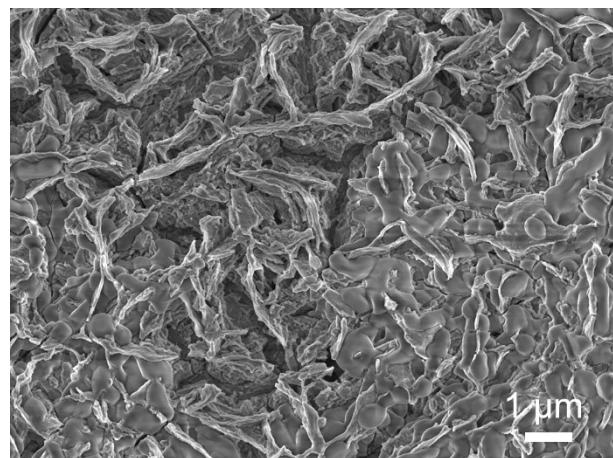


Figure S15. SEM image of P-PtMo₆-Ni-BDC on NF after HER.

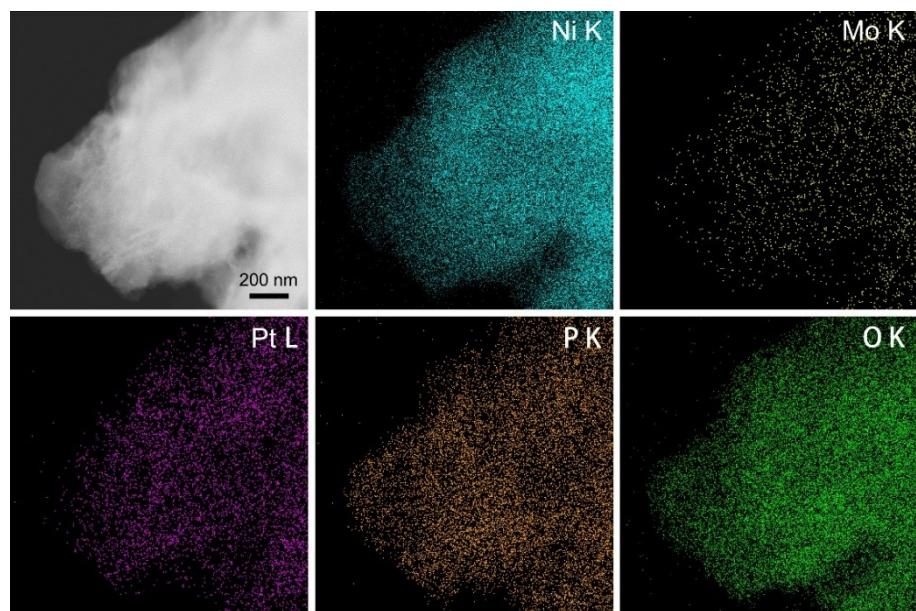


Figure S16. Mapping images of corresponding elements of Ni, Mo, Pt, P, and O of P-PtMo₆-Ni-BDC ultrasonicated after HER.

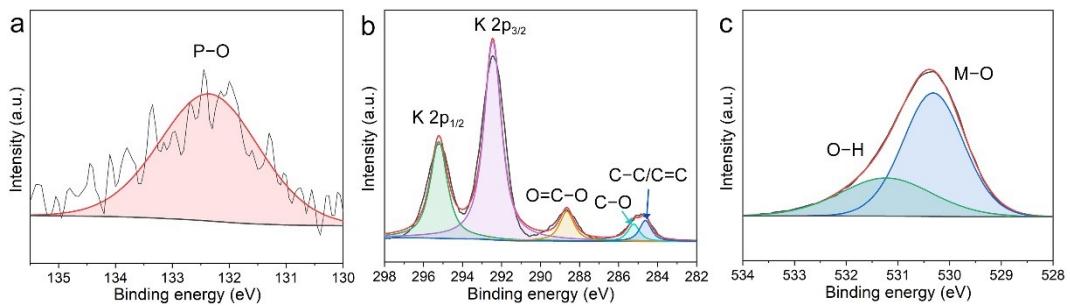


Figure S17. High-resolution XPS spectra of P-PtMo₆-Ni-BDC on NF after HER. (a) P 2p, (b) C 1s, and (c) O 1s.

Table S1. Loading mass of P-PtMo₆-Ni-BDC on NF. (The area of NF is 8 cm²)

	Sample 1	Sample 2	Sample 3
Mass of NF (m)	230 mg	234 mg	231 mg
Mass of P-PtMo ₆ -Ni-BDC on NF (n)	256 mg	263 mg	259 mg
Mass of P-PtMo ₆ -Ni-BDC (n-m)	26 mg	29 mg	28 mg
Average loading mass			3.5 mg cm ⁻²

Table S2. Comparison of HER performance of P-PtMo₆-Ni-BDC and recently reported Mo- and Ni-based electrocatalysts in an alkaline medium.

Electrocatalyst	Overpotential at 10 mA cm ⁻² (mV)	Tafel slope (mV dec ⁻¹)	Substrate	Reference
P-PtMo₆-Ni-BDC	26	21	Ni foam	This work
Co-Mo-0.125–6N	52	54	Ni foam	¹
NiCoCu-Mo _{0.078} /CF	35	50.12	Cu foam	²
Co ₃ O ₄ –Mo ₂ N	100	162.4	Ni foam	³
Pt/Ni-Mo-N-O	40.6@100 mA cm ⁻²	49.3	Ni foam	⁴
FeP/Ni ₂ P/CP	46	50.5	Carbon paper	⁵
MoO ₂ @Ni ₂ P	57	61	Carbon paper	⁶
Ni ₂ P/MoNiP ₂ /MoP-10	20	30.6	Ni foam	⁷
Cu ₃ P/Ni ₂ P	88.1	94	Copper foam	⁸
Ni ₂ P–Ni ₁₂ P ₅	76	68.0	Ni foam	⁹
PS-MoNi	30	37	Ni foam	¹⁰

Table S3. Atomic percentages of corresponding elements of P-PtMo₆-Ni-BDC on NF before and after HER according to EDX results.

	before HER	after HER
Ni (at.%)	19.59	61.46
Mo (at.%)	8.71	1.53
Pt (at.%)	1.30	0.79
P (at.%)	10.54	0.82
C (at.%)	30.91	11.92
O (at.%)	38.95	21.35
K (at.%)	/	2.13

Table S4. Concentrations of corresponding elements of P-PtMo₆-Ni-BDC ultrasonicated from the NF support before and after HER measured by ICP-MS.

	Before HER	After HER
Ni (mg L ⁻¹)	30.08	31.57
Mo (mg L ⁻¹)	13.21	10.07
Pt (mg L ⁻¹)	4.05	5.64
P (mg L ⁻¹)	15.89	5.10

References

- 1 X. Zhang, A. Wu, D. Wang, Y. Jiao, H. Yan, C. Jin, Y. Xie and C. Tian, *Appl. Catal. B-Environ.*, 2023, **328**, 122474.
- 2 G. Qian, Y. Wang, L. Li, M. Lu, C. Chen, D. Min, Z. Wei and P. Tsiakaras, *Adv. Funct. Mater.*, 2024, **34**, 2404055.
- 3 T. Wang, P. Wang, W. Zang, X. Li, D. Chen, Z. Kou, S. Mu and J. Wang, *Adv. Funct. Mater.*, 2022, **32**, 2107382.
- 4 W. Yu, Z. Chen, Y. Fu, W. Xiao, B. Dong, Y. Chai, Z. Wu and L. Wang, *Adv. Funct. Mater.*, 2023, **33**, 2210855.
- 5 M. Gao, P. Gao, T. Lei, C. Ouyang, X. Wu, A. Wu and Y. Du, *J. Mater. Chem. A*, 2022, **10**, 15569-15579.
- 6 Z. Zhang, X. Lin, S. Tang, H. Xie and Q. Huang, *J. Colloid Interf. Sci.*, 2023, **630**, 494-501.
- 7 X. Bu, D. Yin, D. Chen, Q. Quan, Z. Yang, S. Yip, C.-Y. Wong, X. Wang and J. C. Ho, *Small*, 2023, **19**, 2304546.
- 8 H. Liu, J. Gao, X. Xu, Q. Jia, L. Yang, S. Wang and D. Cao, *Chem. Engin. J.*, 2022, **448**,

137706.

9 Z. Wang, S. Wang, L. Ma, Y. Guo, J. Sun, N. Zhang and R. Jiang, *Small*, 2021, **17**, 2006770.

10 J. Song, Y. Q. Jin, L. Zhang, P. Dong, J. Li, F. Xie, H. Zhang, J. Chen, Y. Jin, H. Meng and X. Sun, *Adv. Energy Mater.*, 2021, **11**, 2003511.