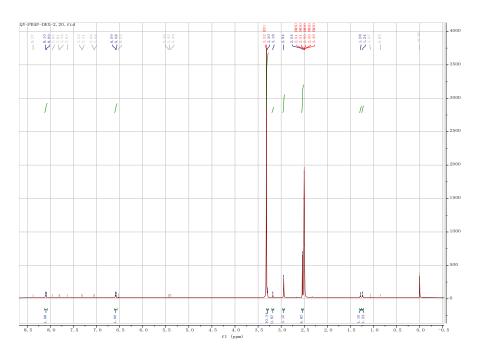


Figure S1. NMR hydrogen spectra of Ace-DEX. The successful reaction of DEX with 2-ethoxypropene was confirmed by <sup>1</sup>H NMR spectroscopy. In the spectrum, the signal at  $\delta = 4.9$  ppm is assigned to the anomeric proton on the DEX pyranose ring. The multiplet in the range of  $\delta = 3.42$ -3.92 ppm corresponds to the protons on the main chain and the -CH<sub>2</sub>- groups of the side chains. Characteristically, signals appearing at  $\delta = 1.0$ -1.5 ppm are attributed to the -CH<sub>3</sub> and -CH<sub>2</sub>- protons from the attached ethoxy functional group. Furthermore, the peak at  $\delta = 4.7$  ppm belongs to the residual signal of the deuterated solvent D<sub>2</sub>O, while the peaks at  $\delta = 2.5$  ppm and  $\delta = 3.32$  ppm are characteristic of the solvent d6-DMSO.



**Figure S2. NMR hydrogen spectra of PBAP-DEX.** The <sup>1</sup>H NMR spectrum was assigned as follows: the signal at  $\delta = 1.0$ –1.5 ppm corresponds to the -CH<sub>3</sub> protons of PBAP; the peak at  $\delta = 2.50$  ppm is characteristic of the solvent DMSO; the resonance at  $\delta = 2.94$  ppm is attributed to the -CH<sub>2</sub>- protons on the main chain; the signal at  $\delta = 3.32$  ppm arises from water (H<sub>2</sub>O); the peak at  $\delta = 6.59$  ppm is assigned to the -OH proton on the main chain; and the signal at  $\delta = 8.10$  ppm belongs to the aromatic protons of the PBAP benzene ring.

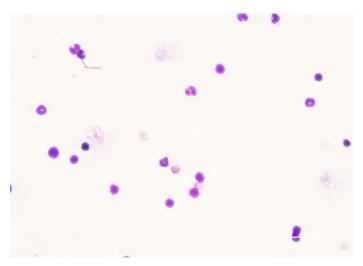


Figure S3. The morphology of neutrophils was observed by Giemsa staining. Scale bar: 50 um.