

**Nanostructured silver tungstate rods decorated on graphitic carbon nitride: An effective electrocatalyst for electrochemical sensing of carcinogenic organic pollutant metal**

*Elaiyappillai Elanthamilan, Sea-Fue Wang\**

\*Department of Materials and Mineral Resources Engineering, National Taipei University of Technology, No. 1, Sec. 3, Chung-Hsiao East Rd., Taipei 106, Taiwan.

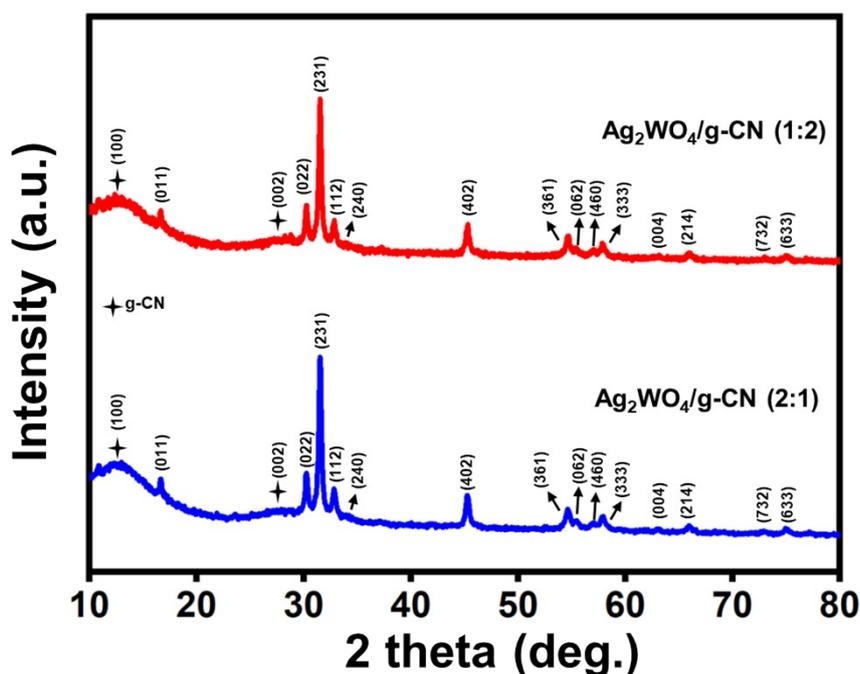
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**Corresponding Author**

Prof. Sea-Fue Wang

Email: [sfwang@ntut.edu.tw](mailto:sfwang@ntut.edu.tw)

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**Fig.S1.** XRD patterns of  $\text{Ag}_2\text{WO}_4/\text{g-CN}$  (2:1 and 1:2) composite

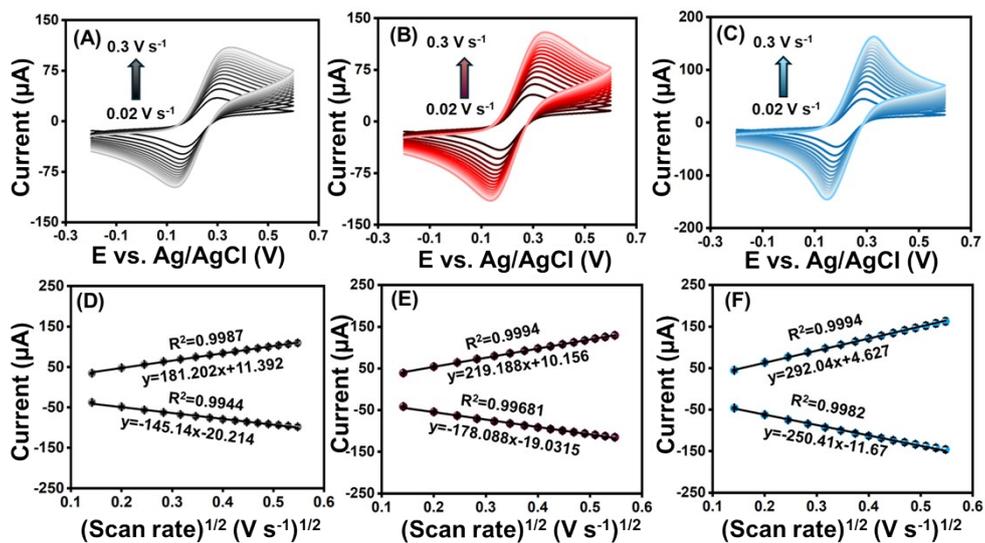


Fig.S2. (A-C) CV curves of bare, g-CN and Ag<sub>2</sub>WO<sub>4</sub> modified SPCE electrode, respectively (D-F) corresponding to the linear plot of bare, g-CN and Ag<sub>2</sub>WO<sub>4</sub> modified SPCE electrode, respectively

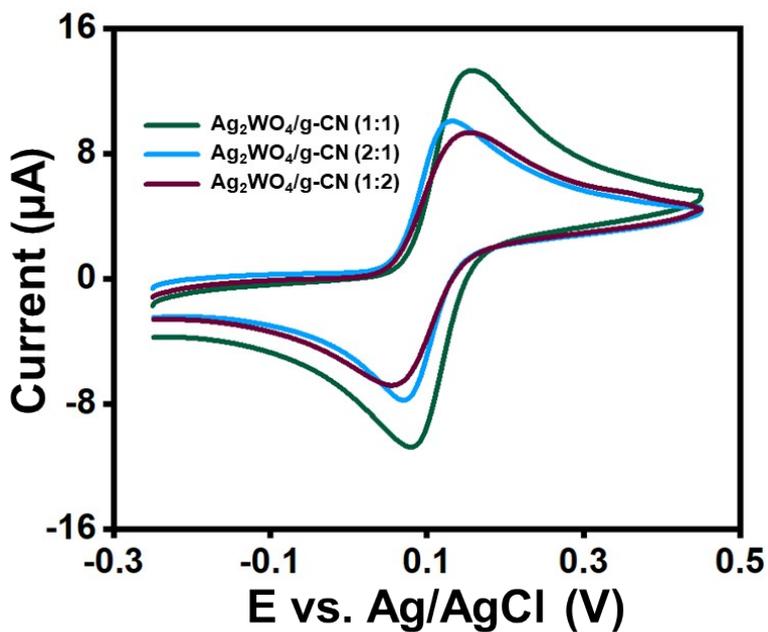
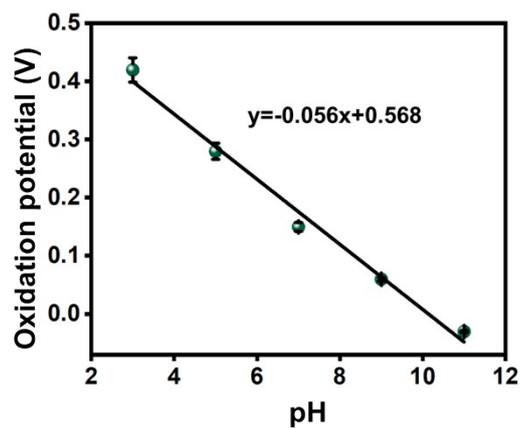
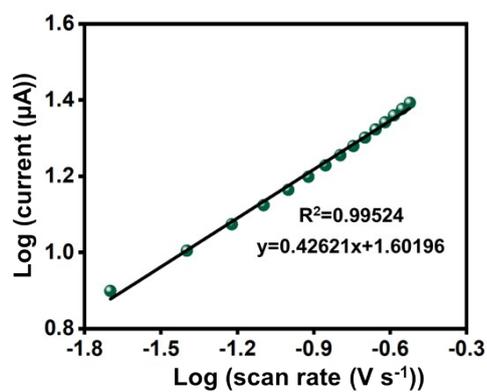


Fig. S3. CV curves of MTL on different ratios of Ag<sub>2</sub>WO<sub>4</sub>/g-CN/SPCE



**Fig. S4. Variation of the anodic peak potential as a function of the pH value of 0.1 M PB solution**



**Fig. S5. Linear relationship between Log (scan rate (V s<sup>-1</sup>)) vs. Log (current (μA))**

**Table S1. Determination of MTL in Real samples with AgW/g-CN (1:1)/SPCE sensors**

<b>Sample</b>	<b>Spiked (<math>\mu\text{M}</math>)</b>	<b>Found (<math>\mu\text{M}</math>)</b>	<b>Recovery (%)</b>
Pond water	5	4.91 $\pm$ 0.06	98.2
	10	9.72 $\pm$ 0.10	97.2
	15	14.83 $\pm$ 0.21	98.9
River water-I	5	4.82 $\pm$ 0.08	96.4
	10	9.85 $\pm$ 0.16	98.5
	15	14.79 $\pm$ 0.18	98.6
River water-II	5	4.89 $\pm$ 0.07	97.8
	10	9.74 $\pm$ 0.17	97.4
	15	14.88 $\pm$ 0.24	99.2
Tap water	5	4.90 $\pm$ 0.06	98.0
	10	9.83 $\pm$ 0.13	98.3
	15	14.82 $\pm$ 0.28	98.8