

Supplementary Information (SI) for

**Enhanced Carrier Extraction and Photostability in Perovskite Solar
Cells via Band-Engineered Sb₂O_x/TiO₂ Bilayer Heterojunction**

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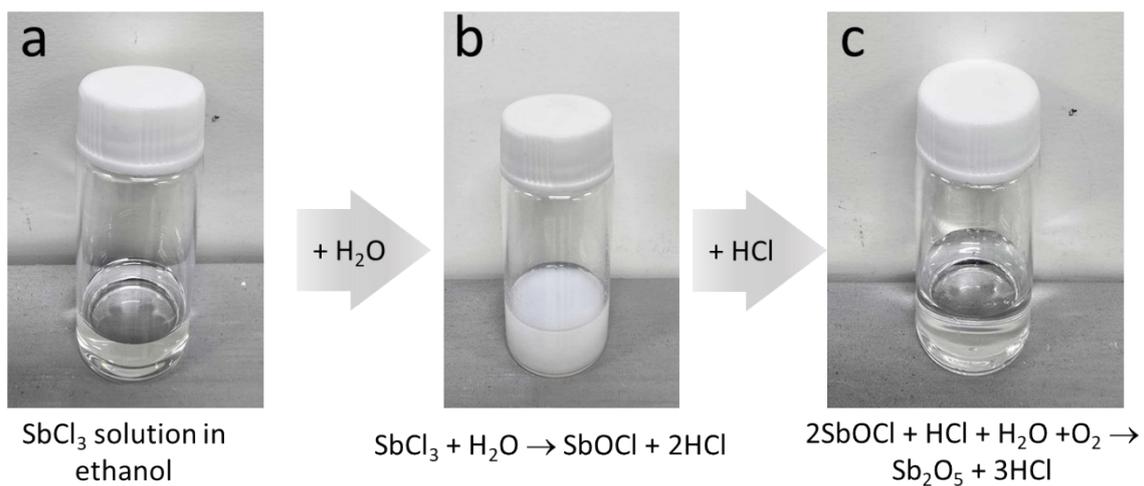


Fig. S1. Photographs illustrating the stepwise preparation of the antimony oxide precursor solution: (a) dissolution of SbCl₃ powder in ethanol to form a clear solution, (b) formation of antimony oxide micelles upon gradual addition of deionized water, and (c) generation of a transparent and homogeneous hydrolysate after the introduction of concentrated HCl.

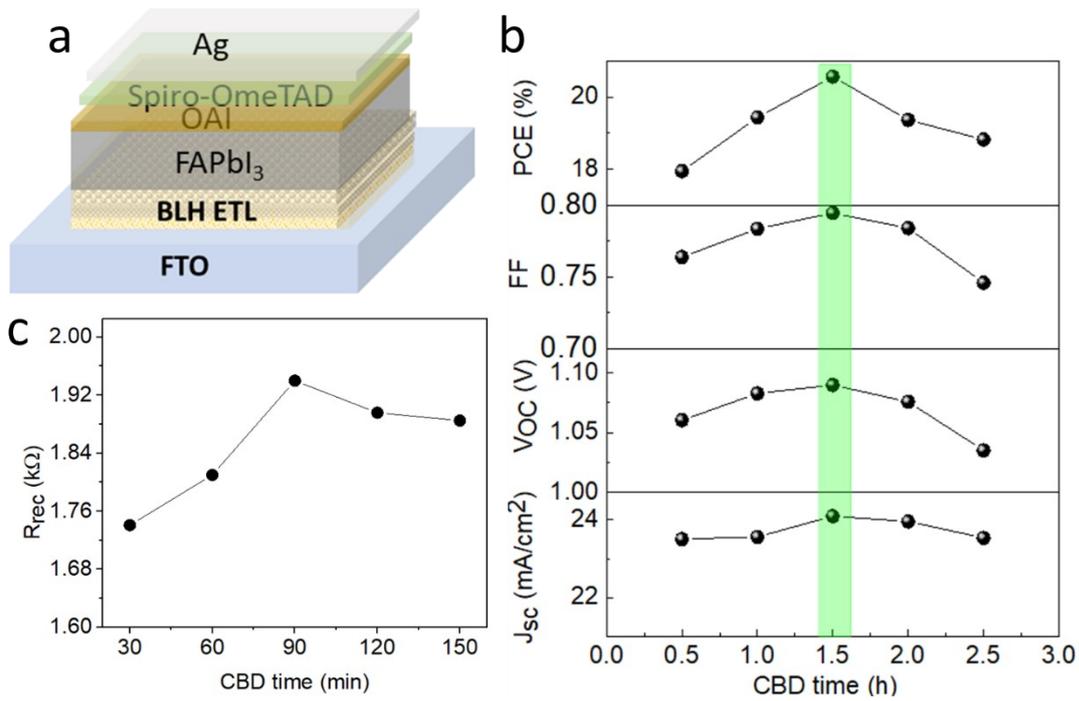


Fig. S2. (a) Schematic illustration of the planar PSC architecture. (b, c) Photovoltaic parameters (J_{sc} , V_{oc} , FF, and PCE) and R_{rec} of BLH-based PSCs as a function of Sb_2O_x deposition time. The R_{rec} values were extracted from Nyquist plots measured under dark conditions at an applied bias of 0.8 V.

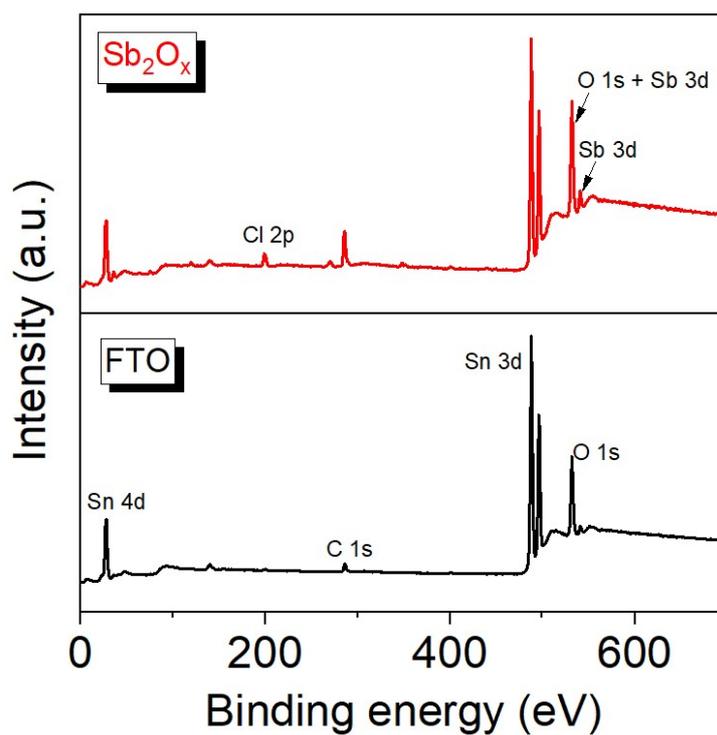


Fig. S3. Survey XPS spectra of bare FTO and Sb_2O_x -coated FTO substrate.

Table S1. Peak fitting parameters obtained from the XPS spectra of bare FTO and Sb_2O_x -coated FTO substrate.

		O 1s	Sb	
		O-Sn	-	-
FTO	Binding energy (ev)	531.10		
	Peak area	306830.31		
		O-Sn + Sb 3d _{5/2} + O-Sb	Sb 3d _{3/2}	Sb 3d _{5/2}
Sb_2O_x -coated FTO	Binding energy (ev)	530.98	540.35	-
	Peak area	519527	101461.83	152192.74

Stoichiometric calculation of Sb_2O_x

The total O 1s peak area for the Sb_2O_x -coated FTO substrate was 519527, while the corresponding O–Sn signal from the bare FTO was 306830.31. Subtracting the latter yielded an excess O 1s area of 212696.70, which accounts for both the oxygen bonded to antimony (O–Sb) and the overlapping Sb 3d_{5/2} contribution.^{S1} Using the known ratio that the area of Sb 3d_{5/2} is 1.5 times that of Sb 3d_{3/2},^{S2} and given the Sb 3d_{3/2} area as 101461.83, the estimated area for Sb 3d_{5/2} was calculated to be 152192.74. Subtracting this from the excess O 1s signal gave a net O–Sb contribution of 60503.95. After normalizing using the relative sensitivity factors (RSF = 2.93 for O 1s and 16.39 for Sb 3d_{5/2}),^{S3} the corrected signal intensities were 20649.80 for O–Sb and 9285.70 for Sb. This results in an atomic ratio of O to Sb of approximately 2.22.

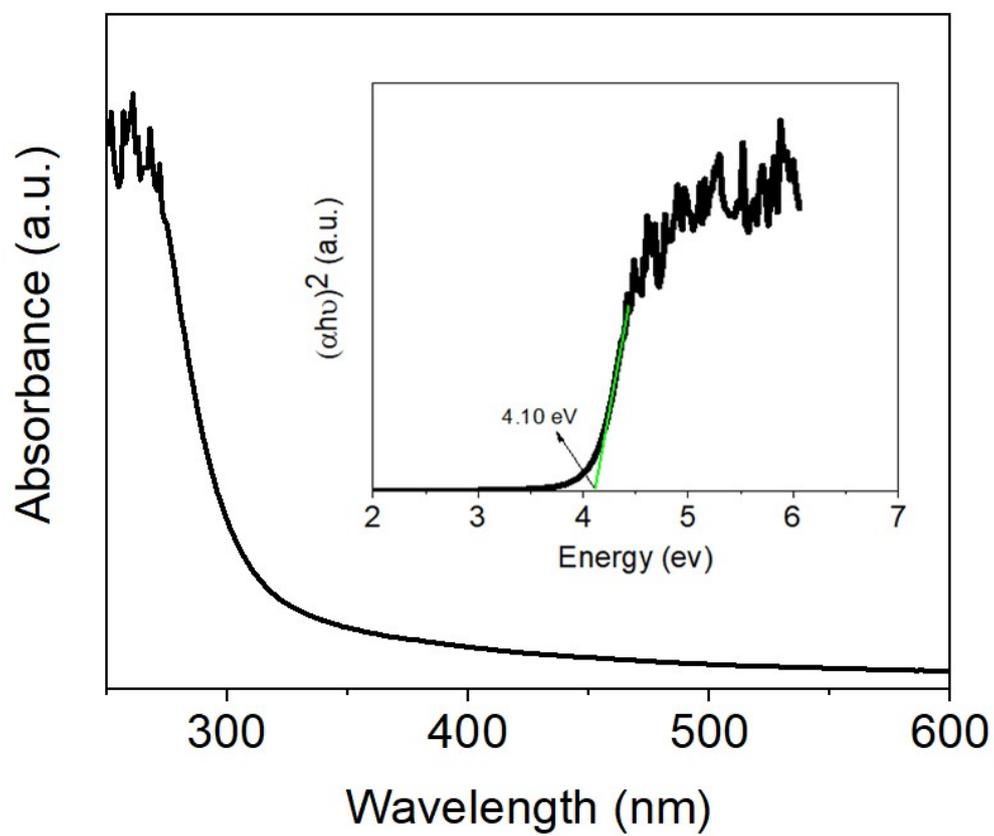


Fig. S4. UV-vis absorption spectrum of the Sb_2O_x layer deposited on a silica glass substrate. The inset displays the corresponding Tauc plot used to estimate the optical bandgap.

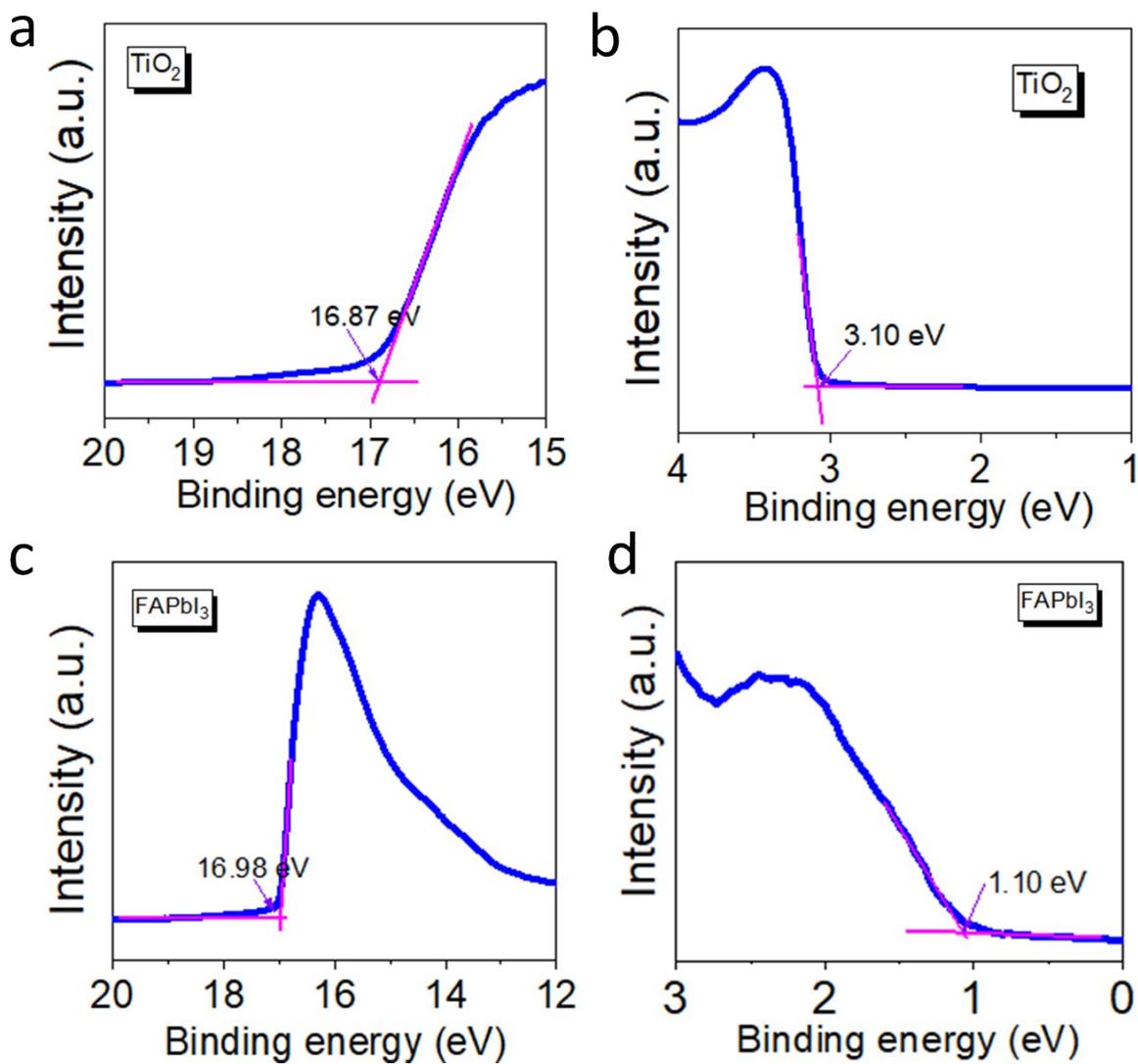


Fig. S5. Ultraviolet photoelectron spectroscopy (UPS) spectra of (a, b) the TiO₂ layer and (c, d) the FAPbI₃ layer, showing the secondary electron cutoff and valence band region, respectively.

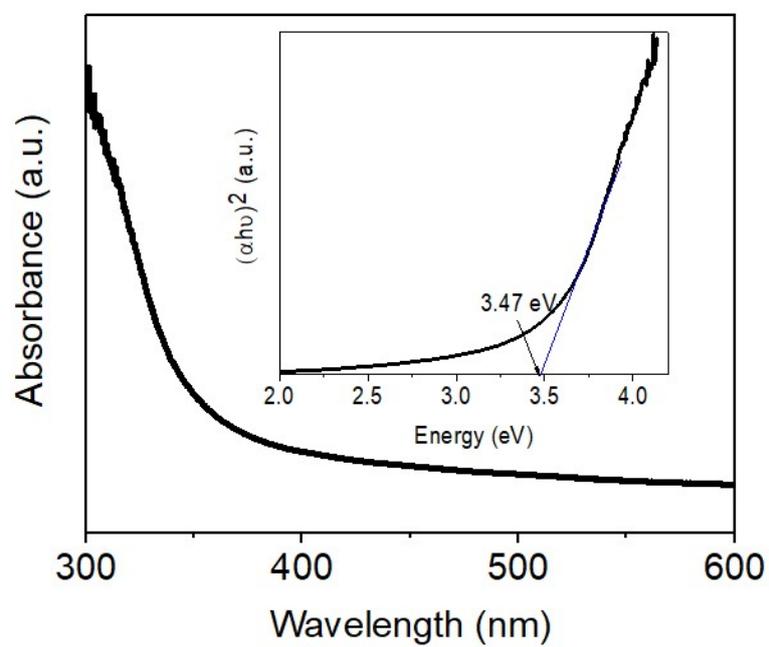


Fig. S6. UV-vis absorption spectrum of the TiO₂ layer. The inset displays the corresponding Tauc plot.

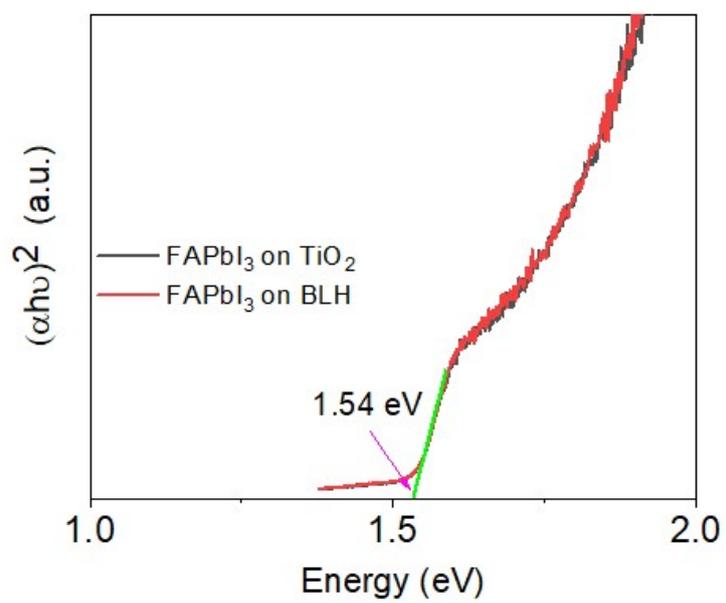


Fig. S7. Tauc plot derived from the UV-vis absorption spectrum of the FAPbI₃ layer, used to estimate its optical bandgap.

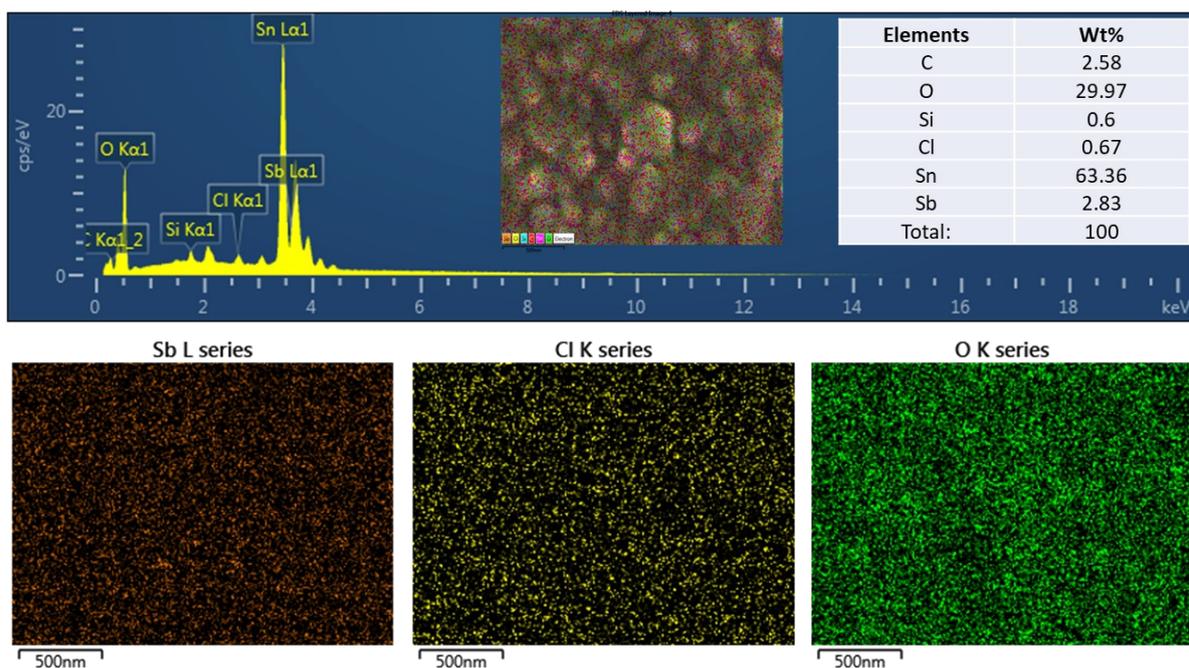


Fig. S8. EDX spectrum and elemental mapping of Sb, O, and Cl for the Sb_2O_x -coated FTO substrate. The inset displays the corresponding FE-SEM image along with the elemental weight percentages.

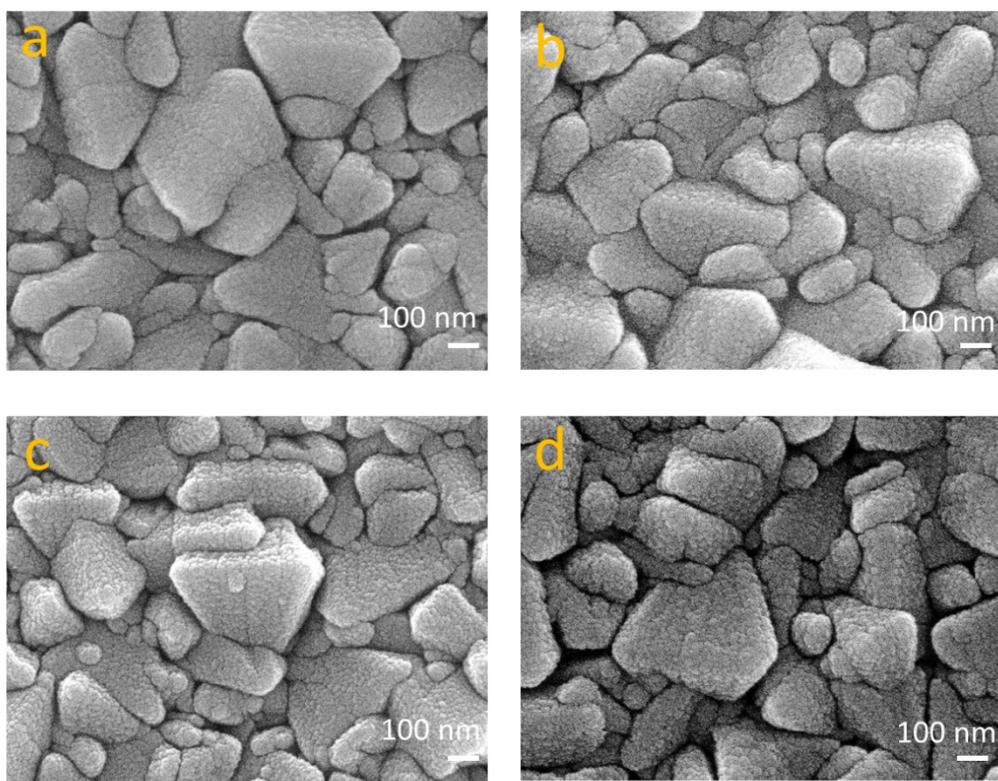


Fig. S9. Top-view FE-SEM images of Sb_2O_x layers deposited on FTO substrates with varying deposition times of 0.5, 1, 2, and 2.5 h.

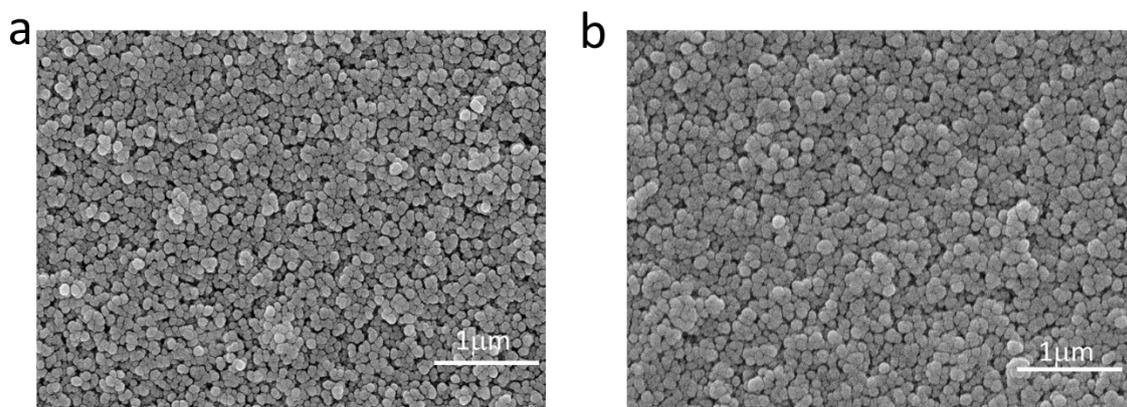


Fig. S10. Top-view FE-SEM images of mesoporous TiO_2 (m- TiO_2) layers deposited on compact TiO_2 (c- TiO_2) and Sb_2O_x substrates.

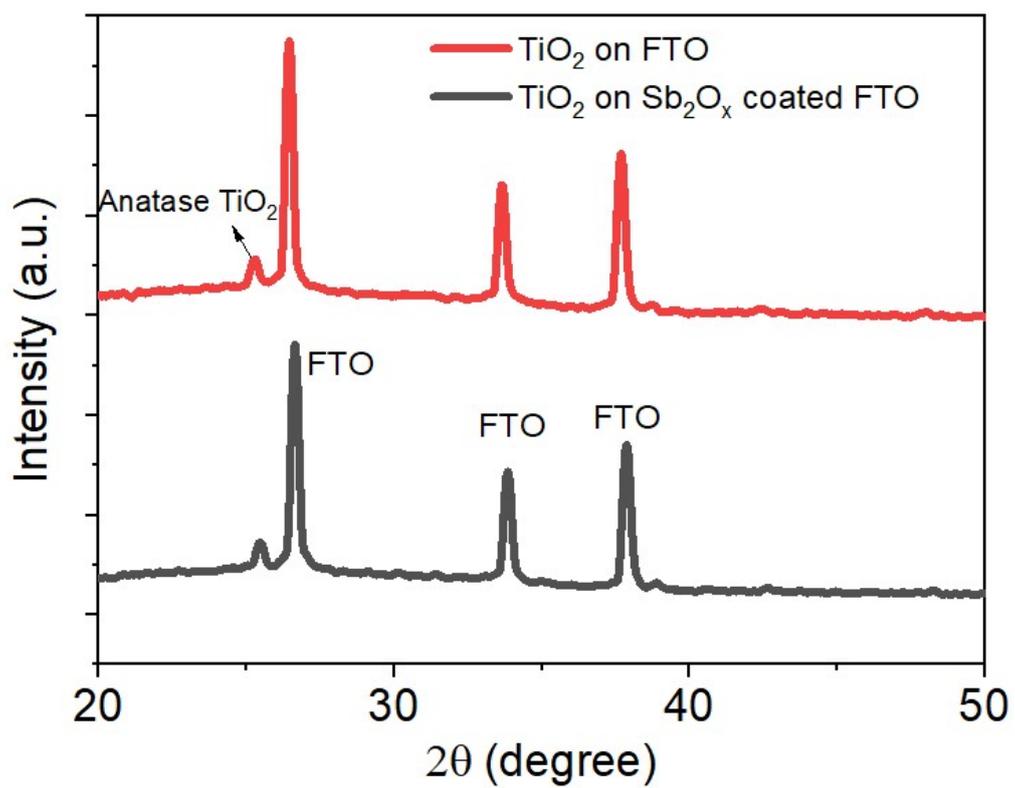


Fig. S11. XRD patterns of TiO₂ layers deposited on bare FTO and Sb₂O_x-coated FTO substrates.

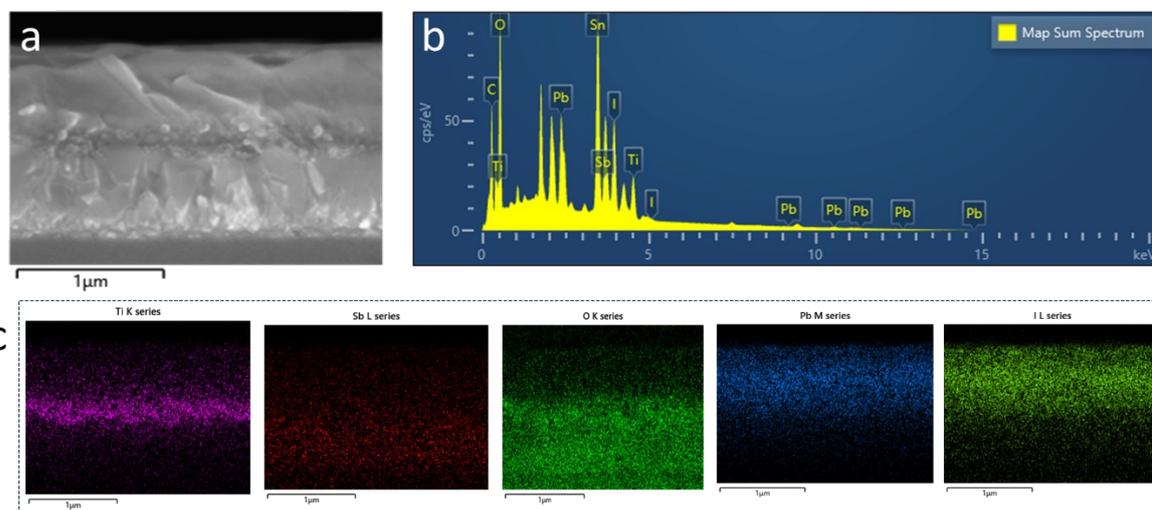


Fig. S12. (a) Cross-sectional FE-SEM image of the $\text{Sb}_2\text{O}_x/\text{TiO}_2/\text{perovskite}$ interface. (b) Corresponding EDS spectra. (c) Elemental mapping showing the spatial distribution of Ti, Sb, O, Pb, and I.

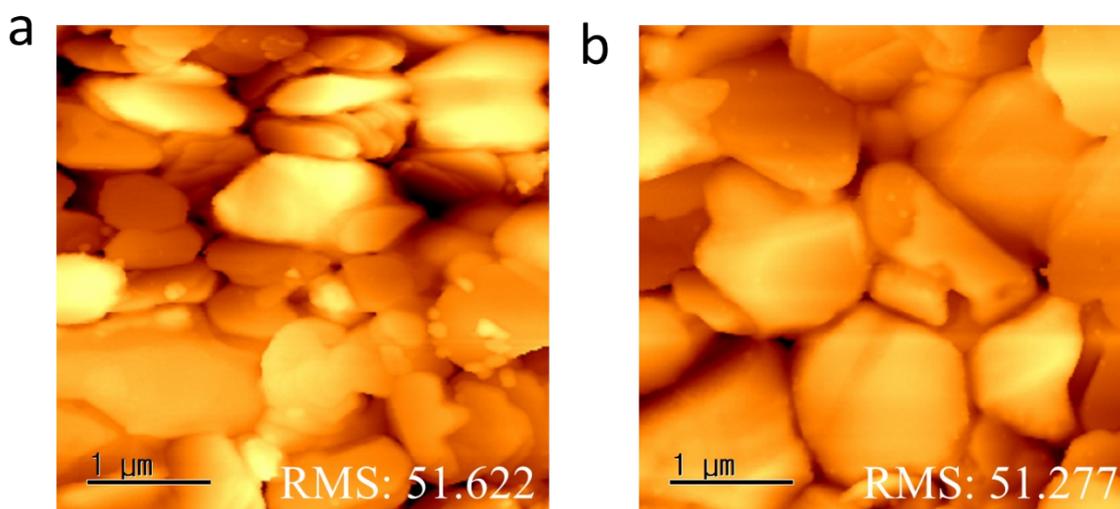


Fig. S13. AFM images of FAPbI₃ layers formed on (a) TiO₂- and (b) BLH-coated FTO substrate.

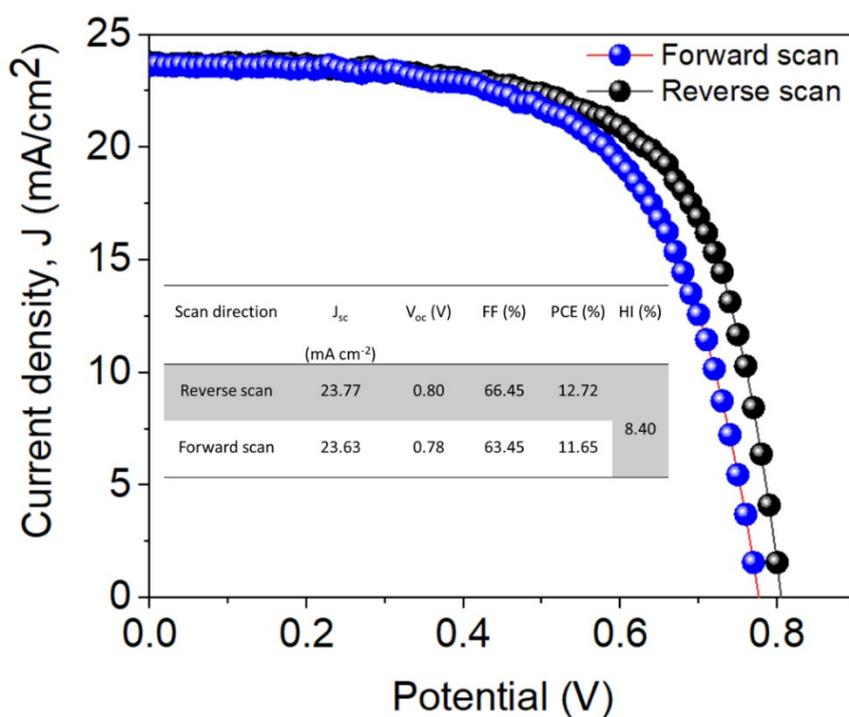


Fig. S14. J-V characteristics of the Sb₂O_x-based PSC measured under both reverse and forward

scan directions. The inset displays the corresponding photovoltaic parameters for each scan.

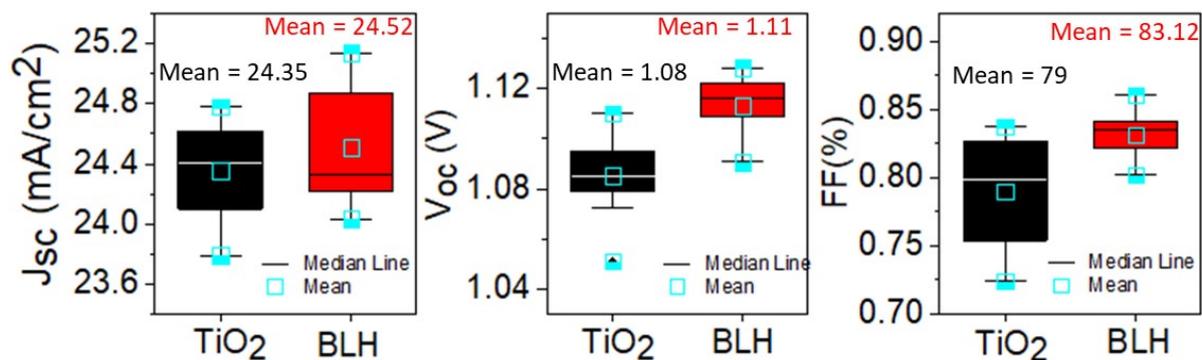


Fig. S15. Statistical distribution of the PV parameters (V_{oc} , J_{sc} , FF) of TiO₂- and BLH-ETLs-based PSCs.

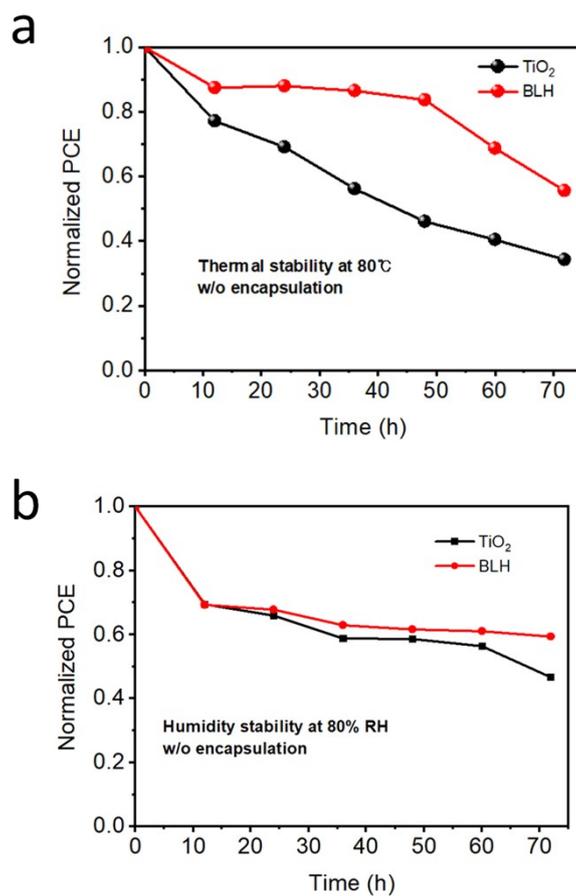


Fig. S16. Stability evaluation of PSCs: (a) thermal stability at 80 °C in ambient air and (b)

humidity stability at 80% RH in ambient air without encapsulation.

Table S2. Comparison of the photovoltaic performance of various reported bilayer ETL-based PSCs with $\text{Sb}_2\text{O}_x/\text{TiO}_2$ ETL-based PSCs, including details on PCE, perovskite materials, and device structures.

ETLs	Perovskite	Device structure	PCE (%)	Ref.
c-TiO ₂ /Fe ₂ O ₃ /SnO ₂	CsPbBr ₃		10.23	S4
Bilayer SnO ₂	MAPbI ₃		17.61	S5
TiO ₂ /SnO ₂	Cs/FA/MA perovskite		17.64	S6
TiO ₂ /SnO ₂	MA/FA perovskite		23.45	S7
TiO ₂ /SnO ₂	Cs/MA/FA perovskite		15.39	S8
TiO ₂ /SnO ₂ -quantum dots	MA/FA perovskite		10.41	S9
TiO ₂ /TiO ₂	MAPbI ₃	n-i-p	14.143	S10
ZnO/SnO ₂	MAPbI ₃		20.43	S11
ZnO/TiO _x	MA/FA perovskite		19.34	S12
$\text{Sb}_2\text{O}_x/\text{TiO}_2$	FAPbI ₃		23.72%	This work

Table S3. Photovoltaic parameters of PSCs incorporating SnO₂/TiO₂ BLH ETLs.

Cell number	J_{sc} (mA/cm²)	V_{oc} (V)	FF	PCE (%)
1	22.70	1.04	0.75	17.81
2	23.62	1.10	0.80	20.70
3	23.21	1.06	0.77	19.05
4	23.83	1.06	0.78	19.76
5	24.31	1.08	0.76	19.84
6	23.47	1.02	0.74	17.72
7	23.04	1.02	0.74	17.30
8	24.02	1.06	0.80	20.31
9	24.13	1.06	0.78	20.09
10	23.40	1.04	0.78	19.03
11	23.82	1.07	0.81	20.77
12	23.96	1.09	0.80	20.87
13	23.27	1.05	0.77	18.67
14	23.10	1.02	0.72	16.83
15	24.09	1.08	0.80	20.79
16	24.18	1.05	0.76	19.17
17	23.40	1.00	0.72	16.91
18	22.94	0.98	0.72	16.13
19	23.95	1.08	0.82	21.04
20	23.52	1.02	0.74	17.69
Average	23.60±0.45	1.05±0.03	0.77±0.03	19.02 ±1.51

Table S4. TRPL parameters obtained by fitting.

Sample	τ_1 (ns)	A_1	τ_2 (ns)	A_2	τ_{avg} (ns)
Sb ₂ O _x	230.15	0.18	1384.40	0.41	1305.88
TiO ₂	114.52	0.19	607.64	0.28	551.70
BLH	86.35	0.08	425.0	0.21	400.0

The decay in PL intensity from the TRPL measurements was analyzed using a bi-exponential decay model [S13]:

$$I = I_0 + A_1 \exp\left(-\frac{x}{\tau_1}\right) + A_2 \exp\left(-\frac{x}{\tau_2}\right)$$
, where τ_1 represents the fast decay component associated with interfacial charge transfer at the ETL/perovskite interface and τ_2 denotes the slower component, attributed to radiative recombination of free carriers within the bulk FAPbI₃ layer. The average lifetime (τ_{avg}) was calculated using the equation $\tau_{\text{avg}} = (A_1\tau_1^2 + A_2\tau_2^2)/(A_1\tau_1 + A_2\tau_2)$.^{S13}

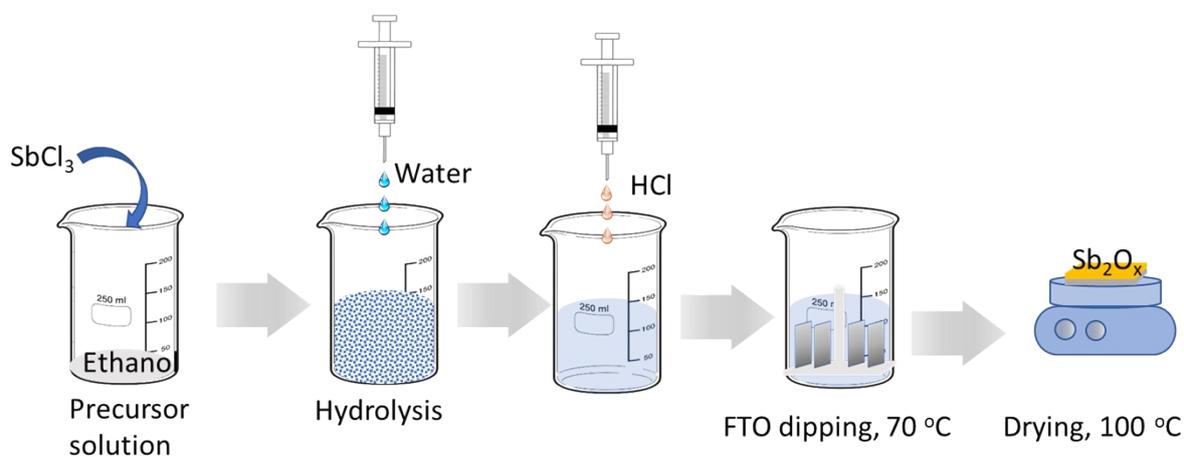


Fig. S17. Stepwise schematic diagram of the chemical bath deposition (CBD) process for forming antimony oxide (Sb_2O_x) layers on FTO substrates.

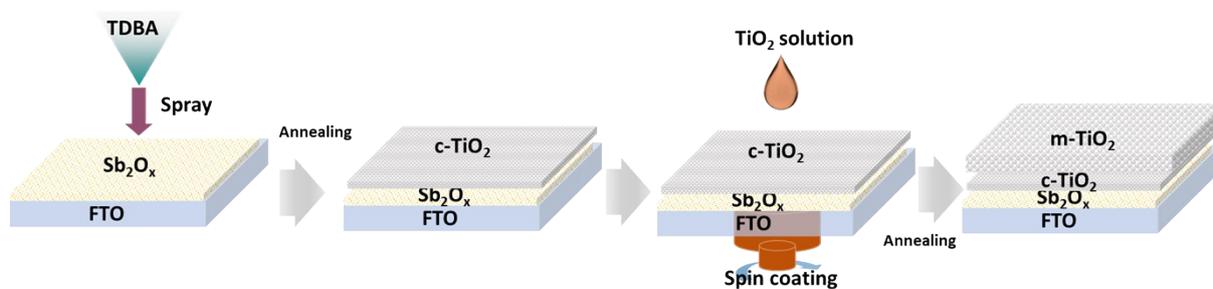


Fig. S18. Schematic illustration of the deposition of compact and mesoporous TiO_2 layers onto Sb_2O_x layers.

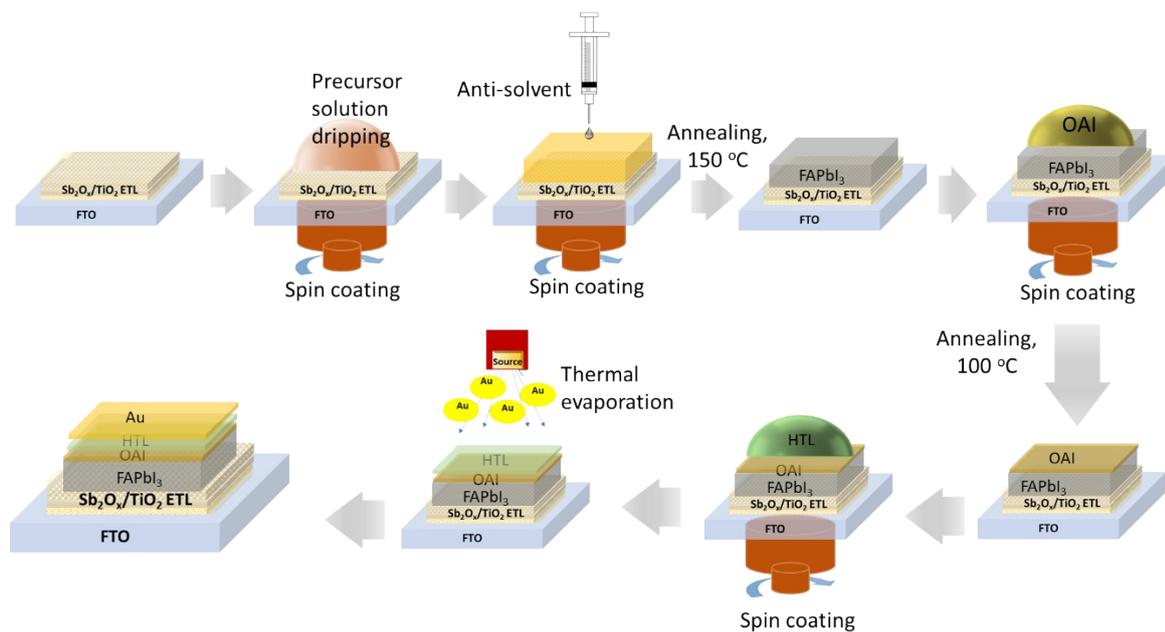


Fig. S19. Schematic diagram of the fabrication of perovskite solar cells using Sb_2O_x/TiO_2 heterostructure electron transport layers (ETLs).

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