

Supplementary Information for:

One-pot microwave synthesis of core-shell C@Bi₂S₃ nanoflake-like structure for humidity-dependent photodetection applications

Marcin Godzierz^{1,*}, Sonia Gorawska^{1,2}, Krystian Mistewicz³, Łukasz Otulakowski¹, Karolina Olszowska¹, Viktoriia Talaniuk¹, Anna Gawron¹, Mirosława Pawlyta⁴, Urszula Szeluga¹

1 – Centre of Polymer and Carbon Materials, Polish Academy of Sciences, Curie-Skłodowskiej 34 Str., 41-819 Zabrze, Poland

2 - Silesian University of Technology, Joint Doctoral School, Akademicka 2A, 44-100 Gliwice, Poland

3 – Silesian University of Technology, Institute of Physics – Centre for Science and Education, Krasińskiego 8, 40-019 Katowice, Poland

4 – Silesian University of Technology, Materials Research Laboratory, Faculty of Mechanical Engineering, Konarskiego 18 A, 44-100 Gliwice, Poland

Corresponding author: mgodzierz@cmpw-pan.pl

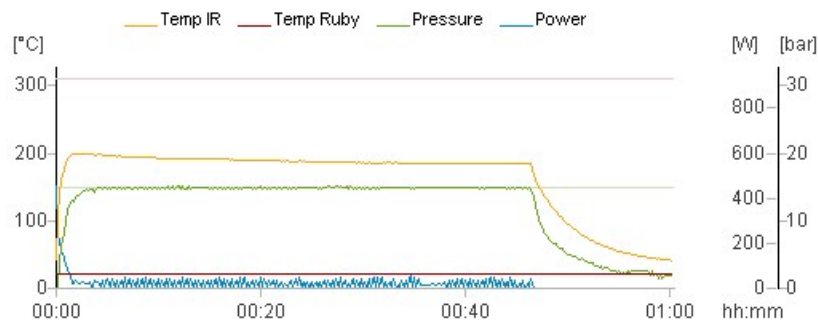


Fig. S1. Changes of pressure and temperature over synthesis of core-shell C@Bi₂S₃ nanoflakes.

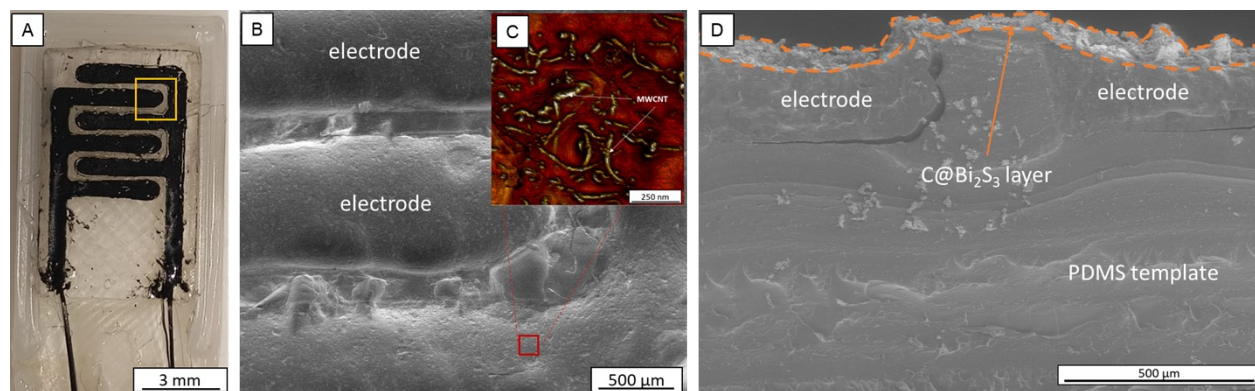


Fig. S2. The digital photograph (A) of fabricated sensor platform, SEM microphotograph (B) from square marked in digital photo, AFM image (C) from red square marked in SEM microphotograph and SEM microphotograph (D) of photodetectors' cross-section, with visible C@Bi₂S₃ layer on the top.

Table S1. Values of bandgap energy reported in the literature for bismuth sulfide.

Material	Method of material preparation	Energy band gap value, eV	Band gap type	Ref.
Bi ₂ S ₃ film	atomic layer deposition	1.03	indirect	1
Bi ₂ S ₃	not applicable	1.32	indirect	2
Bi ₂ S ₃ film	physical vapor deposition	1.32 – 1.36		3
Bi ₂ S ₃ film	reactive evaporation	1.38	direct	4
Bi ₂ S ₃ nanoflowers	hydrothermal method	1.39	direct	5
Bi ₂ S ₃ film	electrochemical synthesis	1.4	direct	6
Bi ₂ S ₃ nanosheets	hydrothermal vulcanization	1.41	direct	7
Bi ₂ S ₃ nanocrystals	organometallic synthesis	1.443		8

Bi ₂ S ₃ nanowires	hydrothermal vulcanization	1.46	direct	7
Bi ₂ S ₃ nanoribbons	hydrothermal vulcanization	1.47	direct	7
Bi ₂ S ₃	not applicable	1.492	indirect	9
Bi ₂ S ₃ film	atomic layer deposition	1.56	direct	1
Bi ₂ S ₃ film	chemical bath deposition	1.56	direct	10
Bi ₂ S ₃ film	successive ionic layer adsorption and reaction	1.61	direct	11
Bi ₂ S ₃ film	pulse-plating method	1.68	direct	12
Bi ₂ S ₃	microwave synthesis	1.34(2)	direct	13
Bi ₂ S ₃ -PVA _{high}	PVA-assisted microwave synthesis	1.41(1)	direct	
Bi ₂ S ₃ -PVA _{low}	PVA-assisted microwave synthesis	1.43(3)	direct	
μ-Bi ₂ S ₃	PVP-assisted microwave synthesis	1.36(2)	direct	14
sμ-Bi ₂ S ₃	PVP-assisted microwave synthesis	1.41(1)	direct	
C@Bi ₂ S ₃ nanoflakes	PEG-assisted microwave synthesis	1.33(3)	direct	this work

Table S2. Values of n exponents of dependence $I \sim U^n$ determined for the ohmic regime ($U < U_{tr}$) and space charge limited current regime ($U > U_{tr}$) under different illumination conditions, where U_{tr} denotes a transition voltage.

	<i>RH</i> , %	<i>n</i> exponent
Dark condition ($I_L = 0$)	75	1.461
	55	1.379
	40	1.385
Red light illumination ($\lambda = 629$ nm, $I_L = 53.4$ μ W/cm ²)	75	1.460
	55	1.403
	40	1.409
Blue light illumination ($\lambda = 465$ nm, $I_L = 69.0$ μ W/cm ²)	75	1.381
	55	1.389
	40	1.369
Green light illumination ($\lambda = 522$ nm, $I_L = 78.9$ μ W/cm ²)	75	1.379
	55	1.381
	40	1.396
White light illumination ($I_L = 178.1$ μ W/cm ²)	75	1.364
	55	1.369
	40	1.382

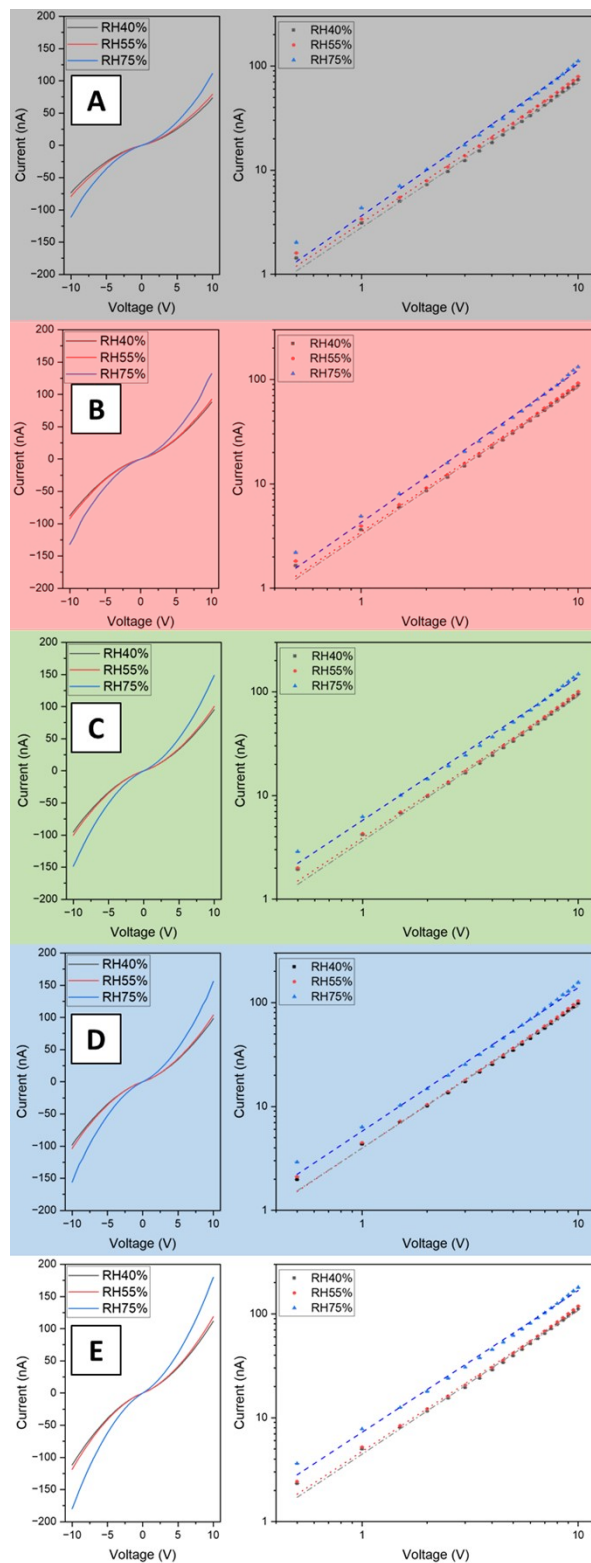


Fig. S3. Humidity-dependent current-voltage characteristics in dark (A), and under red (B, $\lambda = 629$ nm), green (C, $\lambda = 522$ nm), blue (D, $\lambda = 465$ nm) and white light (E) illumination. Right graphs show log-log plot, with two fitted curves up to 3V and above 3V (see Table S1).

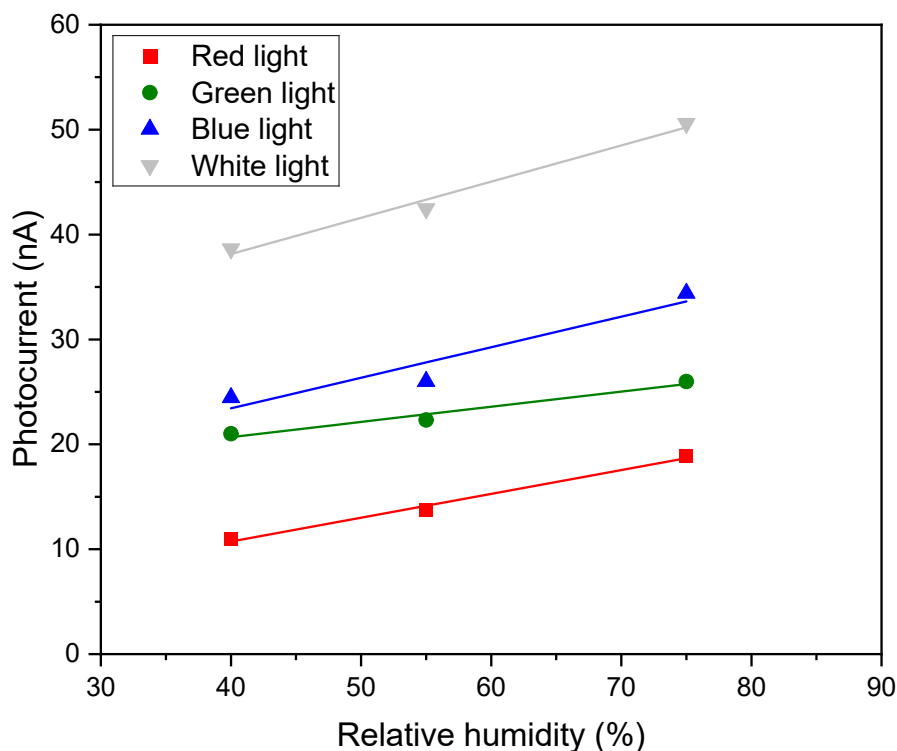


Fig. S4. The relationship between relative humidity and photocurrent induced by red light ($\lambda = 629$ nm, $I_L = 53.4 \mu\text{W}/\text{cm}^2$), green light ($\lambda = 522$ nm, $I_L = 78.9 \mu\text{W}/\text{cm}^2$), blue light ($\lambda = 465$ nm, $I_L = 69.0 \mu\text{W}/\text{cm}^2$), and white light ($I_L = 178.1 \mu\text{W}/\text{cm}^2$).

Table S3. The values of the power exponent determined by best fitting the theoretical dependence of light intensity on photocurrent (Equation (3)) to experimental data presented in Figures 8A-C.

Light color	The power exponent γ		
	$RH = 40\%$	$RH = 55\%$	$RH = 75\%$
Red ($\lambda = 629$ nm)	0.575(7)	0.562(3)	0.484(2)
Green ($\lambda = 522$ nm)	0.661(1)	0.568(4)	0.547(4)
Blue ($\lambda = 465$ nm)	0.689(3)	0.666(6)	0.649(5)
White	0.669(2)	0.648(5)	0.611(6)

References

- 1 N. Mahuli, D. Saha and S. K. Sarkar, *J. Phys. Chem. C*, 2017, **121**, 8136–8144.
- 2 H. Koc, H. Ozisik, E. Deligöz, A. M. Mamedov and E. Ozbay, *J. Mol. Model.*, 2014, **20**, 2180.
- 3 S. ten Haaf, H. Sträter, R. Brüggemann, G. H. Bauer, C. Felser and G. Jakob, *Thin Solid Films*, 2013, **535**, 394–397.
- 4 J. Lukose and B. Pradeep, *Solid State Commun.*, 1991, **78**, 535–538.

- 5 S. Sharma and N. Khare, *Colloid Polym. Sci.*, 2018, **296**, 1479–1489.
- 6 Z. Grubač, J. Katić and M. Metikoš-Huković, *J. Electrochem. Soc.*, 2019, **166**, H433.
- 7 Y.-C. Liang and T.-H. Li, 2022, **11**, 284–297.
- 8 M. Aresti, M. Saba, R. Piras, D. Marongiu, G. Mula, F. Quochi, A. Mura, C. Cannas, M. Mureddu, A. Ardu, G. Ennas, V. Calzia, A. Mattoni, A. Musinu and G. Bongiovanni, *Adv. Funct. Mater.*, 2014, **24**, 3341–3350.
- 9 H. Ben Abdallah and W. Ouerghui, *Opt. Quantum Electron.*, 2021, **54**, 20.
- 10 H. Moreno-García, S. Messina, M. Calixto-Rodríguez and H. Martínez, *Appl. Surf. Sci.*, 2014, **311**, 729–733.
- 11 M. Suresh Kumar, S. P. Madhusudanan, K. Mohanta and S. K. Batabyal, *Mater. Res. Express*, 2020, **7**, 15909.
- 12 F. Ding, Q. Wang, S. Zhou, G. Zhao, Y. Ye and R. Ghomashchi, *R. Soc. Open Sci.*, 2020, **7**, 200479.
- 13 K. Mistewicz, M. Godzierz, A. Gawron, Ł. Otulakowski, A. Hercog, K. Kurtyka, S. Hajra and H. J. Kim, *J. Mater. Chem. C*, 2024, **12**, 15691–15703.
- 14 M. Godzierz, K. Mistewicz, A. Gawron, K. Kurtyka, Ł. Otulakowski and T. Kanti Das, *J. Alloys Compd.*, , DOI:10.1016/j.jallcom.2024.176820.