

Supporting Information

Solvothermal Synthesis of Spherical CoMnO_3 Catalysts for Efficient Trace Ethylene Removal under High Humidity

Mingjie Zhao,^a Qiaofei Zhang,^{a,*} Jiayin Liang,^a Liwen Zhang,^a Jianfei Gao,^a

Jurakhonzoda Rauf Jurakhon,^b Chunshan Zhu^{a,*} and Chunzheng Wang^{c,*}

^a College of Chemistry and Chemical Engineering, Henan University of Technology,
Zhengzhou, Henan 450001, China

^b Life Safety and Ecology Department, Mechanical and Technological Faculty, Tajik
Technical University named after academician M.S.Osimi, Dushanbe 734042,
Tajikistan

^c State Key Laboratory of Heavy Oil Processing, College of Chemistry and Chemical
Engineering, China University of Petroleum (East China), Qingdao 266580,
Shandong, China

* Corresponding author E-mails: *qiaofeifay@163.com* (Q. Zhang)

zhuchunshan@haut.edu.cn (C. Zhu)

czwang@upc.edu.cn (C. Wang)

Table S1. The element contents in MnCoO_x catalysts determined by ICP.

Catalyst	Mn content (wt%)	Co content (wt%)	Molar ratio of Mn/Co
Mn ₆ Co ₁ O _x	34.3	7.7	5.1
Mn ₅ Co ₂ O _x	36.2	15.2	2.6
Mn ₄ Co ₃ O _x	29.3	23.5	1.3

Table S2. Textural characteristics of the catalysts.

Catalyst	Specific surface area (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Average pore size (nm)
MnO _x	97.9	0.20	8.3
Mn ₆ Co ₁ O _x	122.7	0.24	8.2
Mn ₅ Co ₂ O _x	142.3	0.25	6.7
Mn ₄ Co ₃ O _x	103.9	0.23	9.1
Co ₃ O ₄	49.3	0.24	19.4

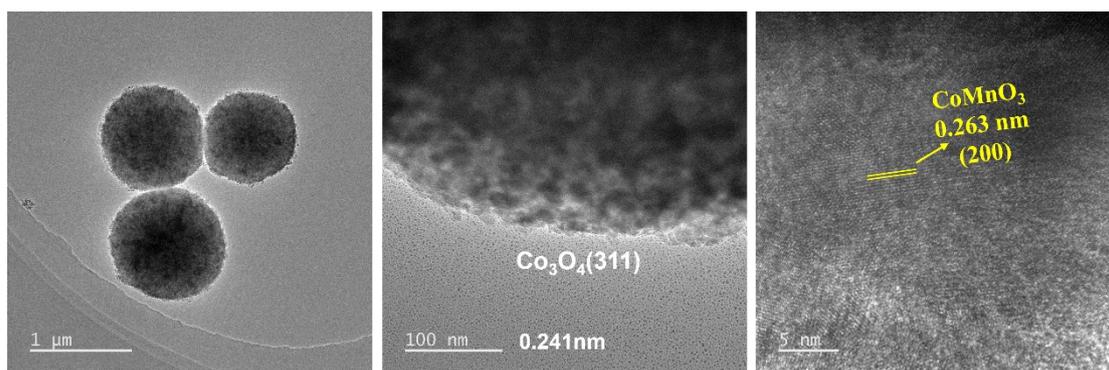


Fig. S1. TEM images of the fresh $\text{Mn}_5\text{Co}_2\text{O}_x$ catalyst.

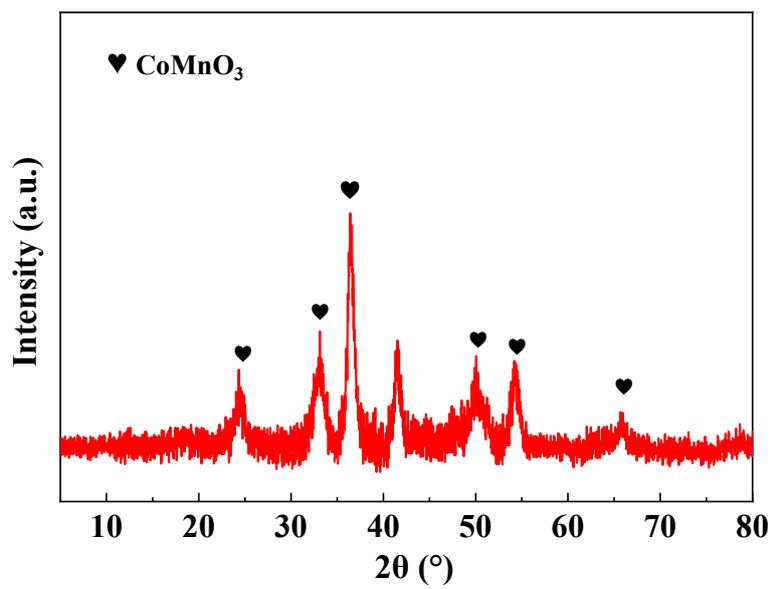


Fig. S2. XRD pattern of the $\text{Mn}_5\text{Co}_2\text{O}_x$ catalyst after stability test.

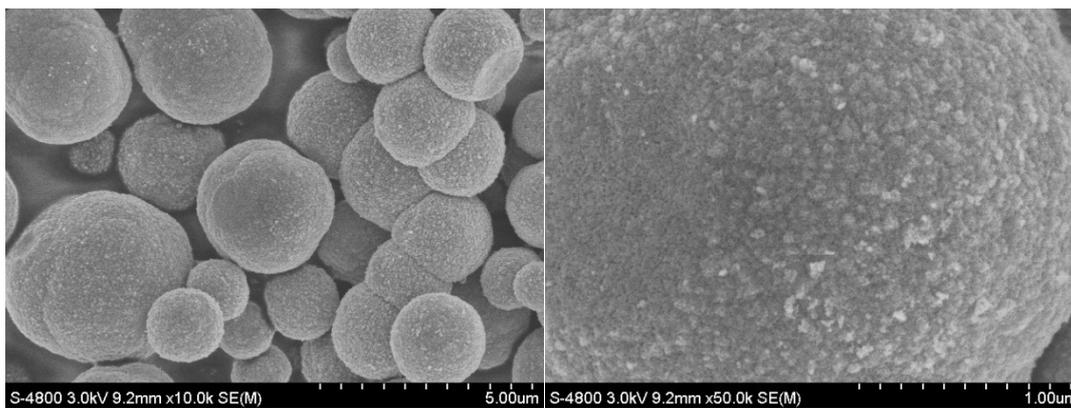


Fig. S3. SEM images of the $\text{Mn}_5\text{Co}_2\text{O}_x$ catalyst after stability test.