### **Supporting Information**

### Rhodium Catalysts Supported on Phosphorus-Rich Covalent Organic Polymers for Sustainable Hydrosilylation of Alkynes

Xin Liu, Xinxin Han, Jiaxi Cai, Qiaoya Wang, Aizhao Pan and Mingyou Hu\*

School of Chemistry, Engineering Research Center of Energy Storage Materials and Devices, Ministry of Education, Xi'an Key Laboratory of Sustainable Energy Material Chemistry, Xi'an Jiaotong University, Xi'an, Shaanxi 710049, People's Republic of China.

E-mail: mingyouhu@xjtu.edu.cn

### **Table of Contents**

1. General Information	3
2. Synthesis and Characterisation of Polymers and Metallized Polymeric Catalysts	4
3. Detailed Optimization of Reaction Conditions	6
4. General Procedures for the Syntheses of Vinylsilane Compounds	7
5. Kinetic Isotope Effect Experiments	8
6. Reuse Experiment of COP-PPh <sub>3</sub> -RhCl	11
7. XPS Characterisation of COP-PPh3-RhCl after 5 Cycles.	12
8. Characterisation Data for Products	13
9. References	
10. NMR Spectra	29

#### **1. General Information**

#### **Reagents and Materials.**

Unless otherwise mentioned, solvents and reagents were purchased from commercial sources and used as received. The solvents were purchased from Aladdin in glass bottle and used directly. Alkynes **1a**-**1t**, and hydrosilanes **2a-2d** are commercially available and no further purification was needed.

#### **Characterisation Methods.**

<sup>1</sup>H, <sup>13</sup>C, and <sup>19</sup>F NMR spectra were recorded on a Bruker 400 MHz (100 MHz, 376 MHz and for <sup>13</sup>C and <sup>19</sup>F, respectively) spectrometer. All <sup>1</sup>H NMR spectra were reported in parts per million (ppm) relative to the signals for CDCl<sub>3</sub> (7.26 ppm) and DMSO- $d_6$  (2.50 ppm). All <sup>13</sup>C NMR spectra were reported in ppm relative to residual CDCl<sub>3</sub> (77.16 ppm) and DMSO- $d_6$  (39.52 ppm). For the isolated compounds, <sup>19</sup>F NMR chemical shifts were determined relative to CFCl<sub>3</sub> at  $\delta$  0 ppm. <sup>13</sup>C solid NMR spectrum for **COP-PPh<sub>3</sub>-RhCl** were recorded on an Agilent DD2-500 MHz NMR spectrometer. Field-Emission Scanning Electron Microscopy (FE-SEM) images were conducted on a Zeiss Sigma300 field-emission scanning electron microscopy. Fourier transform infrared (FTIR) spectroscopy was performed using a VERTEX70 spectrophotometer in the range of 4000 to 400 cm<sup>-1</sup>. X-ray photoelectron spectroscopy (XPS) measurements were conducted on a ThermoFisher Scientific ESCALAB Xi+ spectrometer. The Rh loading of **COP-PPh<sub>3</sub>-RhCl** before and after the reaction were detected by Inductively Coupled Plasma Mass Spectrometer (ICP-MS, NexION 350D). The thermogravimetric analysis (TGA) spectra were recorded on METTLER TOLEDO TGA/DSC3+ thermogravimeter from 30 °C to 800 °C at a rate of 10 °C/min under an air atmosphere.

**Abbreviations Used:** DBE: 1,2-dibromoethane, MeCN: acetonitrile, DCE: dichloroethane, DCM: dichloromethane, MeOH: methanol, THF: tetrahydrofuran, DMF: *N*,*N*-dimethylformamide, DMSO: dimethyl sulfoxide, EA: ethylacetate, PPh<sub>3</sub>: triphenylphosphine.

#### 2. Synthesis and Characterisation of Polymers and Metallized Polymeric Catalysts

#### 2.1 Synthesis of COP-PPh<sub>3</sub>.

**COP-PPh<sub>3</sub>** was synthesized by a facile modified solvent knitting strategy based on the Friedel-Crafts alkylation reaction using DBE as both the solvent and the external crosslinker. Under N<sub>2</sub> atmosphere, triphenylphosphine (5 mmol, 1.31 g) was dispersed in 20 mL DBE and mixed for 30 min. Then anhydrous AlCl<sub>3</sub> (15 mmol, 2.0 g) was added to the reaction medium and the mixture was allowed to stir vigorously at 60 °C for 24 h, and 80 °C for 24 h. The resulting precipitate was quenched by the slow addition of 40 mL H<sub>2</sub>O, filtered and washed several times with ethanol and deionized water respectively. Finally, the crude product was extracted with ethanol in a Soxhlet extractor for 48 h, the obtained brown solid powder was vacuum dried at 70 °C for another 48 h.

#### 2.2 Synthesis of COP-PPh<sub>3</sub>-RhCl.

A 100 mL round bottom flask equipped with a teflon-coated magnetic stirring bar was charged with **COP-PPh<sub>3</sub>** (1 g) and rhodium chloride (RhCl<sub>3</sub>•3H<sub>2</sub>O, 100 mg), then EtOH (20 mL) was injected into the bottom flask by a syringe under nitrogen atmosphere, the reaction mixture was stirred at 70 °C for 6 h. After the completion of the reaction, the resulting precipitate was filtered and washed several times with deionized water, ethanol and DCM respectively. Finally, the obtained brown solid powder was vacuum dried at 70 °C for another 24 h.

#### 2.3 Characterisation of COP-PPh<sub>3</sub>-RhCl.



Figure S1. The TGA curve of COP-PPh<sub>3</sub>-RhCl.



Figure S2. The full-scan XPS spectra of COP-PPh<sub>3</sub>-RhCl.



Figure S3. High-resolution C 1s XPS spectra of COP-PPh<sub>3</sub>-RhCl.



Figure S4. High-resolution Cl 2p XPS spectra of COP-PPh<sub>3</sub>-RhCl.



Figure S5. SEM image of COP-PPh<sub>3</sub>-RhCl and the corresponding EDS elemental mapping of P and Rh elements.

#### 3. Detailed Optimization of Reaction Conditions

+ $+$ $+$ $+$ $+$ $+$ $+$ $+$ $+$ $+$	COP-PPh <sub>3</sub> -RhCl ► solvent, 70 °C, N <sub>2</sub> , 12 h	β-3a	β <sup>Si</sup> α-3a
Entry	Solvent	Yield ( <b>3a</b> , %)	$E/\alpha$
1	THF	39	90:10
2	1,4-dioxane	46	89:11
3	MeCN	38	92:8
4	Toluene	60	92:8
5	DMF	35	90:10
6	DMSO	20	91:11

Table S1. Screen of solvents for reaction

Reaction conditions: **1a** (0.5 mmol), **2a** (0.6 mmol), COP-PPh<sub>3</sub>-RhCl (20 mg, 1.77 mol%), solvent (3 mL), 70 °C, N<sub>2</sub>, 12 h. Yields were determined by <sup>1</sup>H NMR with CH<sub>2</sub>Br<sub>2</sub> as an internal standard.

Table S2. Screen of reaction temperature for reaction

1a $+$ $2a$ $+$ $2a$	COP-PPh <sub>3</sub> -RhCI toluene, <i>Temp</i> , N <sub>2</sub> , 12 h	β-3a + (	βi α-3a
Entry	<i>Temp</i> (°C)	Yield ( <b>3a</b> , %)	$E/\alpha$
1	50	56	91:9
2	70	60	92:8
3	90	77	92:8
4	110	75	90:10
5	130	73	88:12

Reaction conditions: **1a** (0.5 mmol), **2a** (0.6 mmol), COP-PPh<sub>3</sub>-RhCl (20 mg, 1.77 mol%), toluene (3 mL), N<sub>2</sub>, 12 h. Yields were determined by <sup>1</sup>H NMR with CH<sub>2</sub>Br<sub>2</sub> as an internal standard.

Table S3. Screen of reaction time for reaction



Entry	<i>Time</i> (h)	Yield ( <b>3a</b> , %)	E/α
1	4	47	92:8
2	8	75	91:9
3	12	77	92:8
4	16	73	93:7
5	20	76	91:9

Reaction conditions: **1a** (0.5 mmol), **2a** (0.6 mmol), COP-PPh<sub>3</sub>-RhCl (20 mg, 1.77 mol%), toluene (3 mL), 90 °C, N<sub>2</sub>. Yields were determined by <sup>1</sup>H NMR with CH<sub>2</sub>Br<sub>2</sub> as an internal standard.

#### 4. General Procedures for the Syntheses of Vinylsilane Compounds

A sealed tube was charged with alkynes 1 (0.5 mmol), silanes (0.6 mmol), COP-PPh<sub>3</sub>-RhCl (20 mg, 1.77 mol%) and toluene (3 mL) under N<sub>2</sub> atmosphere. The mixture was stirred at 90 °C for 12 h. When the reaction was completed, the mixture was filtered, then the filtrate was concentrated in vacuo. The crude product was purified by column chromatography on silica gel using a mixture of hexanes and ethyl acetate as eluent to give the title compounds **3**.

#### 5. Kinetic Isotope Effect Experiments

#### 5.1 The Procedures for the Syntheses of Phenylacetylene- $d_1$ (1a- $d_1$ )



The compound **1a**- $d_1$  was prepared following a modified literature procedure.<sup>1</sup> To a stirred solution of phenylacetylene (1.02 g, 10 mmol, 1.0 equiv) in THF (20 mL) at -78 °C was added n-BuLi (6 mL, 2.5 M in hexanes, 15 mmol, 1.5 equiv) dropwise. The reaction mixture was quenched with D<sub>2</sub>O (2.0 mL) after 30 minutes and further stirred at room temperature for 5 minutes. The resulting solution was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated under reduced pressure to yield phenylacetylene- $d_1$  (788 mg, 77% yield) as a colorless oil, which was used directly without further purification. The deuterium incorporation was determined to be 99% by <sup>1</sup>H NMR. The NMR spectrum was identical with the starting material with the acetylene proton being absent. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.44 – 7.31 (m, 2H), 7.24 – 7.14 (m, 3H).

#### 5.2 The Procedures for the Syntheses of Dimethyl(phenyl)silane-d1 (2a-d1)



The compound **2a**-*d*<sub>1</sub> was prepared following a modified literature procedure.<sup>2</sup> To a stirring suspension of LiAlD<sub>4</sub> (210 mg, 5 mmol) in dry Et<sub>2</sub>O (12 mL) was added PhMe<sub>2</sub>SiCl (2.55 g, 15 mmol) dropwise at ambient temperature under N<sub>2</sub>. The reaction mixture was refluxed at 40 °C for 12 h. The reaction was cooled to room temperature. Then, the reaction was quenched by adding aqueous solution of sodium hydroxide (15 mL, 10 wt%) into the crude reaction mixture, which was subsequently extracted by diethyl ether for three times. The resulting solution was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated under reduced pressure to yield PhMe<sub>2</sub>SiD as a colorless oil in 74% yield (1.5 g, 99% D), which was used directly without further purification. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.51 – 7.40 (m, 2H), 7.32 – 7.23 (m, 3H), 0.25 (s, 6H).

#### **5.3 Deuterium Labelling Experiments**



(1) A sealed tube was charged with **1a** (1.5 mmol, 153 mg), **2a** (1.8 mmol, 245 mg), **COP-PPh<sub>3</sub>-RhCl** (30 mg, 0.87 mol%), 1,3,5-trimethoxybenzene (0.825 mmol, 138.8 mg) and toluene (9 mL)

under N<sub>2</sub> atmosphere. The reaction mixture was stirred at 90 °C under precise time control, with samples taken every 5 minutes to monitor the yield of product  $\beta(E)$ -3a by <sup>1</sup>H NMR with 1,3,5-trimethoxybenzene as an internal standard.

(2) A sealed tube was charged with  $1a-d_1$  (1.5 mmol, 155 mg), silane 2a (1.8 mmol, 245 mg), COP-PPh<sub>3</sub>-RhCl (30 mg, 0.87 mol%), 1,3,5-trimethoxybenzene (0.51 mmol, 85.7 mg) and toluene (9 mL) under N<sub>2</sub> atmosphere. The reaction mixture was stirred at 90 °C under precise time control, with samples taken every 5 minutes to monitor the yield of product  $\beta(E)$ -3a- $d_1$  by <sup>1</sup>H NMR with 1,3,5trimethoxybenzene as an internal standard.

(3) A sealed tube was charged with **1a** (1.5 mmol, 153 mg), silane **2a**- $d_1$  (1.8 mmol, 247 mg), **COP-PPh<sub>3</sub>-RhCl** (30 mg, 0.87 mol%), 1,3,5-trimethoxybenzene (0.329 mmol, 55.4 mg) and toluene (9 mL) under N<sub>2</sub> atmosphere. The reaction mixture was stirred at 90 °C under precise time control, with samples taken every 5 minutes to monitor the yield of product  $\beta(E)$ -3a- $d_2$  by <sup>1</sup>H NMR with 1,3,5-trimethoxybenzene as an internal standard.



Figure S6. The kinetic curve of yield versus reaction time.



Figure S7. The crude <sup>1</sup>H NMR spectrum of compound  $\beta(E)$ -3a- $d_1$ .



Figure S8. The crude <sup>1</sup>H NMR spectrum of compound  $\beta(E)$ -3a- $d_2$ .

#### 6. Reuse Experiment of COP-PPh<sub>3</sub>-RhCl



To a 10 mL Pyrex reactor equipped with a magnetic stir bar, added 1-ethynyl-4-methoxybenzene (1.0 mmol, 1.0 equiv), dimethylphenylsilane (1.2 mmol, 1.2 equiv) and **COP-PPh<sub>3</sub>-RhCl** (40 mg, 1.77 mol%), with 6 mL of toluene as solvent. The mixture was stirred at 90 °C for 12 h. When the reaction was completed, the catalyst is recovered through suction filtration, followed by washing and drying before initiating the next catalytic cycle. The resulting filtrate is concentrated under vacuum to yield the crude product. The Yield of **3h** was determined by <sup>1</sup>H NMR with  $CH_2Br_2$  as an internal standard. The catalytic cycle was repeated five times, and the recovered catalyst was subjected to Characterisation analysis.





Figure S9. The full-scan XPS spectra of COP-PPh<sub>3</sub>-RhCl after 5 cycles.



Figure S10. High-resolution Cl 2p XPS spectra of COP-PPh<sub>3</sub>-RhCl after 5 cycles.

#### 8. Characterisation Data for Products

We determined the configuration of the vinylsilane products based on the differences in the chemical shifts and coupling constants of the alkene protons. The proton coupling constant is largest in  $\beta(E)$ configured aryl alkenes, followed by  $\beta(Z)$  isomers, while the  $\alpha$ -configured aryl alkenes exhibit the smallest values. For vinyl silicon compounds, the coupling constant of hydrogen protons in  $\beta(E)$ configured aryl alkenes is approximately 19 Hz, while that in  $\beta(Z)$ -configured aryl alkenes is about 15
Hz, and in  $\alpha$ -configured aryl alkenes, it is around 2.8 Hz.

#### (*E*)-Dimethyl(phenyl)(styryl)silane ( $\beta$ (*E*)-3a) and dimethyl(phenyl)(1-phenylvinyl)silane ( $\alpha$ -3a)





The mixture of compound  $\beta(E)$ -**3a** and compound  $\alpha$ -**3a** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3a** could not be separated as pure forms from major isomer  $\beta(E)$ -**3a**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  6.01 (d, J = 2.8 Hz, 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> with that at  $\delta$  6.01 (d, J = 2.9 Hz, 1H) and  $\delta$  5.69 (d, J = 2.8 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>3</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.63 – 7.52 (overlapped, 2.76H), 7.50 – 7.43 (overlapped, 1.84H), 7.42 – 7.36 (overlapped, 2.76H), 7.36 – 7.30 (overlapped, 1.84H), 6.97 (d, J = 19.2 Hz, 0.92H), 6.61 (d, J = 19.2 Hz, 0.92H), 0.46 (overlapped, 5.52H).  $\delta$  the  $\alpha$ isomer 7.63 – 7.52 (overlapped, 0.24H), 7.50 – 7.43 (overlapped, 0.16H), 7.42 – 7.36 (overlapped, 0.24H), 7.36 – 7.30 (overlapped, 0.16H), 6.01 (d, J = 2.8 Hz, 0.08H), 5.69 (d, J = 2.8 Hz, 0.08H), 0.46 (overlapped, 0.48H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 145.4, 138.7, 138.3, 134.1, 129.2, 128.7, 128.3, 128.0, 127.2, 126.6, -2.4. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>4</sup>

(*E*)-(2-([1,1'-Biphenyl]-4-yl)vinyl)dimethyl(phenyl)silane ( $\beta(E)$ -3b) and (1-([1,1'-biphenyl]-4-yl)vinyl)dimethyl(phenyl)silane ( $\alpha$ -3b)



The mixture of compound  $\beta(E)$ -3b and compound  $\alpha$ -3b was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -3b could not be separated as pure forms from major isomer  $\beta(E)$ -3b. Because compound  $\alpha$ -3b has not been reported in the literature, the comparison with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of previously reported  $\alpha$ -**3a** bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  6.02 (d, J = 2.8, 0.13H) and  $\delta$  5.66 (d, J = 2.8 Hz, 0.13H) in CDCl<sub>3</sub> with that at  $\delta$  6.01 (d, J= 2.8 Hz, 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> reported previously. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.58 – 7.52 (overlapped, 3.48H), 7.53 – 7.51 (overlapped, 1.74H), 7.48 – 7.45 (overlapped, 1.74H), 7.39 – 7.35 (overlapped, 1.74H), 7.34 – 7.31 (overlapped, 2.61H), 7.30 - 7.26 (overlapped, 0.87H), 6.94 (d, J = 19.2 Hz, 0.87H), 6.59 (d, J = 18.8 Hz, 0.87H), 0.41 (s, 5.22H).  $\delta$  the  $\alpha$  isomer 7.58 – 7.52 (overlapped, 0.52H), 7.53 – 7.51 (overlapped, 0.26H), 7.48 -7.45 (overlapped, 0.26H), 7.39 - 7.35 (overlapped, 0.26H), 7.34 - 7.31 (overlapped, 0.39H), 7.30 -7.26 (overlapped, 0.13H), 6.02 (d, J = 2.8 Hz, 0.13H), 5.66 (d, J = 2.8 Hz, 0.13H), 0.41 (s, 0.78H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 144.9, 141.0, 140.8, 138.7, 137.3, 134.1, 129.2, 128.9, 128.0, 127.5, 127.4, 127.4, 127.1, 127.0, -2.4. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>5</sup> HRMS (ESI) *m/z*: calcd for C<sub>22</sub>H<sub>23</sub>Si [M+H]<sup>+</sup>: 315.1564, found 315.1570.

## (*E*)-Dimethyl(4-methylstyryl)(phenyl)silane ( $\beta(E)$ -3c) and dimethyl(phenyl)(1-(*p*-tolyl)vinyl)silane ( $\alpha$ -3c)





The mixture of compound  $\beta(E)$ -**3c** and compound  $\alpha$ -**3c** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3c** could not be separated as pure forms from major isomer  $\beta(E)$ -**3c**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  6.00 (d, J = 2.8 Hz, 0.08H) and  $\delta$  5.65 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> with that at  $\delta$  6.04 (d, J = 2.9 Hz, 1H) and  $\delta$  5.69 (d, J = 2.9 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>6</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.60 – 7.56 (overlapped, 1.84H), 7.38 – 7.34 (overlapped, 4.60H), 7.14 (overlapped, 1.84H), 6.93 (d, J = 19.2 Hz, 0.92H), 6.53 (d, J = 19.2 Hz, 0.92H), 2.34 (s, 2.76H), 0.44 (s, 5.52H).  $\delta$  the  $\alpha$  isomer 7.60 – 7.56 (overlapped, 0.16H), 7.38 – 7.34 (overlapped, 0.40H), 7.14 (overlapped, 0.16H), 6.00 (d, J = 2.8 Hz, 0.08H), 5.65 (d, J = 2.8 Hz, 0.08H), 2.34 (s, 0.24H), 0.44 (s, 0.48H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 145.3, 134.1, 133.1, 129.4, 129.1, 127.9, 127.8, 126.6, 126.2, 125.8, 21.4, -2.3. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>5</sup>







The mixture of compound  $\beta(E)$ -3d and compound  $\alpha$ -3d was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -3d could not be separated as pure forms from major isomer  $\beta(E)$ -3d. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  6.04 (d, J = 2.9 Hz, 0.11H) and  $\delta$  5.66 (d, J = 2.9 Hz, 0.11H) in CDCl<sub>3</sub> with that at  $\delta$  6.11 – 6.00 (d, J = 2.8 Hz, 1H) and  $\delta$  5.72 – 5.62 (d, J = 2.9 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>7</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.60 – 7.53 (overlapped, 1.78H), 7.42 – 7.39 (overlapped, 1.78H), 7.38 – 7.32 (overlapped, 4.45H), 6.95 (d, J =19.1 Hz, 0.89H), 6.55 (d, J = 19.1 Hz, 0.89H), 1.32 (s, 8.01H), 0.44 (s, 5.34H).  $\delta$  the  $\alpha$  isomer 7.60 – 7.53 (overlapped, 0.22H), 7.42 – 7.39 (overlapped, 0.22H), 7.38 – 7.32 (overlapped, 0.55H), 6.04 (d, J = 2.9 Hz, 0.11H), 5.66 (d, J = 2.9 Hz, 0.11H), 1.32 (s, 0.99H), 0.44 (s, 0.66H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 151.5, 145.3, 138.9, 134.1, 133.1, 129.1, 127.9, 126.4, 126.1, 125.6, 34.7, 31.4, -2.3. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>8</sup>

# (*E*)-Dimethyl(2-(naphthalen-2-yl)vinyl)(phenyl)silane ( $\beta(E)$ -3e) and dimethyl(1-(naphthalen-2-yl)vinyl)(phenyl)silane ( $\alpha$ -3e)





The mixture of compound  $\beta(E)$ -**3e** and compound  $\alpha$ -**3e** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3e** could not be separated as pure forms from major isomer  $\beta(E)$ -**3e**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  6.13 (d, J = 2.8 Hz, 0.12H) and  $\delta$  5.80 (d, J = 2.8 Hz, 0.12H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3a** at  $\delta$  6.01 (d, J = 2.8 Hz, 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.84 – 7.80 (overlapped, 2.64H), 7.73 – 7.70 (overlapped, 0.88H), 7.66 – 7.57 (overlapped, 2.64H), 7.48 – 7.46 (overlapped, 1.76H), 7.43 – 7.40 (overlapped, 2.64H), 7.15 (d, J = 18.8 Hz, 0.88H), 6.75 (d, J = 19.2 Hz, 0.88H), 0.51 (s, 5.28H).  $\delta$  the  $\alpha$  isomer 7.84 – 7.80 (overlapped, 0.12H), 7.66 – 7.57 (overlapped, 0.36H), 7.48 – 7.46 (overlapped, 0.12H), 7.66 – 7.57 (overlapped, 0.36H), 7.48 – 7.46 (overlapped, 0.32H), 7.43 – 7.40 (overlapped, 0.32H), 6.13 (d, J = 2.8 Hz, 0.12H), 5.80 (d, J = 2.8 Hz, 0.12H), 0.49 (s, 0.72H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 145.4, 138.7, 135.7, 134.1, 133.5, 133.1, 129.2, 128.3, 128.3, 128.0, 127.8, 127.7, 127.0, 126.4, 126.2, 123.5, -2.3. HRMS (ESI) m/z: calcd for C<sub>20</sub>H<sub>21</sub>Si [M+H]<sup>+</sup>: 289.1407, found 289.1410.

#### (*E*)-(2-Methoxystyryl)dimethyl(phenyl)silane methoxyphenyl)vinyl)dimethyl(phenyl)silane (α-3f)





and

(1-(2-

 $(\beta(E)-3f)$ 

The mixture of compound  $\beta(E)$ -**3f** and compound  $\alpha$ -**3f** was isolated by silica gel column chromatography (Hexane/EA = 20:1) as a colorless oil. Compounds  $\alpha$ -**3f** could not be separated as pure forms from major isomer  $\beta(E)$ -**3f**. Because compound  $\alpha$ -**3f** has not been reported in the literature, the comparison with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3h** bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.91 (d, J = 2.8 Hz, 0.11H) and 5.67 (d, J = 2.8 Hz, 0.11H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3h** at  $\delta$  5.89 (d, J = 2.8 Hz, 0.15H) and  $\delta$  5.53 (d, J = 2.8 Hz, 0.15H) in CDCl<sub>3</sub>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.60 – 7.55 (overlapped, 2.67H), 7.42 – 7.35 (overlapped, 2.67H), 7.42 – 7.35 (overlapped, 0.89H), 7.26 – 7.22 (overlapped, 0.89H), 6.95 – 6.91 (overlapped, 0.89H), 6.87 (d, J = 8.0 Hz, 0.89H), 6.55 (d, J = 19.2 Hz, 0.89H), 3.84 (s, 2.67H), 0.44 (s, 5.34H).  $\delta$  the  $\alpha$  isomer 7.60 – 7.55 (overlapped, 0.33H), 7.42 – 7.35 (overlapped, 0.11H), 6.95 – 6.91 (overlapped, 0.33H), 7.42 – 7.35 (overlapped, 0.33H), 7.26 – 7.22 (overlapped, 0.11H), 6.95 – 6.91 (overlapped, 0.11H), 5.91 (d, J = 2.8 Hz, 0.11H), 5.67 (d, J = 2.8 Hz, 0.11H), 3.54 (s, 0.33H), 0.35 (s, 0.66H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 156.8, 139.7, 139.2, 134.1, 129.3, 129.1, 127.9, 127.5, 127.4, 126.5, 120.7, 111.1, 55.6, -2.3. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>9</sup> HRMS (ESI) m/z: calcd for C<sub>17</sub>H<sub>21</sub>OSi [M+H]<sup>+</sup>: 269.1356, found 269.1353.



OCH

1g

2a

осн₃

β(E)-**3**g



*α*-3g

The mixture of compound  $\beta(E)$ -3g and compound  $\alpha$ -3g was isolated by silica gel column chromatography (Hexane/EA = 20:1) as a colorless oil. Compounds  $\alpha$ -3g could not be separated as pure forms from major isomer  $\beta(E)$ -3g. Because compound  $\alpha$ -3g has not been reported in the literature, the comparison with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -3f bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.97 (d, J = 2.8 Hz, 0.11H) and  $\delta$  5.65 (d, J = 2.8 Hz, 0.11H) in CDCl<sub>3</sub> with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3f** at  $\delta$  5.91 (d, J = 2.8 Hz, 0.11H) and  $\delta$  5.67 (d, J = 2.8 Hz, 0.11H) in CDCl<sub>3</sub>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.56 – 7.54 (overlapped, 1.78H), 7.35 – 7.33 (overlapped, 2.67H), 7.21 (t, J = 8.0 Hz, 0.89H), 7.02 – 7.00 (overlapped, 0.89H), 6.97 - 6.96 (m, 0.89H), 6.89 (d, J = 18.8 Hz, 0.89H), 6.79 (dd, J = 8.0, 2.4 Hz, 0.89H), 6.56 (d, J = 18.8 Hz, 0.89H), 6.79 (dd, J = 8.0, 2.4 Hz, 0.89H), 6.56 (d, J = 18.8 Hz, 0.89H), 6.79 (dd, J = 1819.2 Hz, 0.89H), 3.78 (s, 2.67H), 0.42 (s, 5.34H). δ the α isomer 7.56 – 7.54 (overlapped, 0.22H), 7.35 -7.33 (overlapped, 0.33H), 7.14 - 7.08 (m, 0.22H), 7.02 - 7.00 (overlapped, 0.11H), 6.71 - 6.68 (m, 0.11H), 5.97 (d, J = 2.8 Hz, 0.11H), 5.65 (d, J = 2.8 Hz, 0.11H), 3.64 (s, 0.33H), 0.40 (s, 0.66H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 160.0, 145.3, 139.8, 138.7, 134.1, 129.6, 129.2, 128.0, 127.6, 119.4, 114.3, 111.5, 55.4, -2.4. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>9</sup> HRMS (ESI) *m/z*: calcd for C<sub>17</sub>H<sub>21</sub>OSi [M+H]<sup>+</sup>: 269.1356, found 269.1361.







The mixture of compound  $\beta(E)$ -**3h** and compound  $\alpha$ -**3h** was isolated by silica gel column chromatography (Hexane/EA = 20:1) as a colorless oil. Compounds  $\alpha$ -**3h** could not be separated as pure forms from major isomer  $\beta(E)$ -**3h**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.89 (d, J = 2.8 Hz, 0.15H) and  $\delta$  5.53 (d, J = 2.8 Hz, 0.15H) in CDCl<sub>3</sub> with that at  $\delta$  5.99 (d, J = 2.4 Hz, 1H) and  $\delta$  5.63 (d, J = 2.4 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>7</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.50 – 7.46 (overlapped, 1.70H), 7.31 – 7.29 (overlapped, 1.70H), 7.28 – 7.26 (m, 2.55H), 6.80 (d, J = 19.2 Hz, 0.85H), 6.80– 6.77 (overlapped, 1.70H), 6.33 (d, J = 19.2 Hz, 0.85H), 3.71 (s, 2.55H), 0.34 (s, 5.10H).  $\delta$  the  $\alpha$  isomer 7.50 – 7.46 (overlapped, 0.30H), 7.31 – 7.29 (overlapped, 0.30H), 7.01 – 6.98 (m, 0.45H), 6.80– 6.77 (overlapped, 0.30H), 5.89 (d, J = 2.8 Hz, 0.15H), 5.53 (d, J = 2.8 Hz, 0.15H), 3.67 (s, 0.45H), 0.33 (s, 0.90H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 159.5, 144.9, 139.9, 136.3, 134.1, 133.1, 129.4, 127.8, 127.5, 114.0, 55.4, -1.0. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>9</sup>

## (*E*)-(3,4-Dimethoxystyryl)dimethyl(phenyl)silane ( $\beta(E)$ -3i) and (1-(3,4-di methoxyphenyl)vinyl)dimethyl(phenyl)silane ( $\alpha$ -3i)





The mixture of compound  $\beta(E)$ -**3i** and compound  $\alpha$ -**3i** was isolated by silica gel column chromatography (Hexane/EA = 20:1) as a colorless oil. Compounds  $\alpha$ -**3i** could not be separated as pure forms from major isomer  $\beta(E)$ -**3i**. Because compound  $\alpha$ -**3i** has not been reported in the literature, the comparison with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3h** bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  6.05 (d, J = 2.8 Hz, 0.14H) and  $\delta$  5.70 (d, J = 2.8 Hz, 0.14H) in CDCl<sub>3</sub> with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3h** at 5.89 (d, J = 2.8 Hz, 0.15H) and  $\delta$  5.53 (d, J = 2.8 Hz, 0.15H) in CDCl<sub>3</sub> reported previously. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.65 – 7.60 (overlapped, 1.72H), 7.42 – 7.39 (overlapped, 2.58H), 7.07 (overlapped, 0.86H), 6.47 (d, J = 19.2 Hz, 0.86H), 3.93 (s, 2.58H), 3.90 (s, 2.58H), 0.49 (s, 5.16H).  $\delta$  the  $\alpha$  isomer 7.65 – 7.60 (overlapped, 0.42H), 7.07 (overlapped, 0.14H), 7.02 (overlapped, 0.14H), 6.86 – 6.83 (overlapped, 0.14H), 5.70 (d, J = 2.8 Hz, 0.14H), 3.86 (s, 0.42H), 3.69 (s, 0.42H), 0.47 (s, 0.84H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 149.3, 149.0, 144.9, 138.7, 134.0, 131.3, 129.1, 127.8, 124.4, 120.0, 110.8, 108.4, 55.9, 55.8, -2.4. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>10</sup> HRMS (ESI) *m/z*: calcd for C<sub>18</sub>H<sub>23</sub>O<sub>2</sub>Si [M+H]<sup>+</sup>: 299.1462, found 299.1460.







The mixture of compound  $\beta$ -**3j** and compound  $\alpha$ -**3j** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\beta(Z)$ -**3j** and  $\alpha$ -**3j** could not be separated as pure forms from major isomer  $\beta(E)$ -**3j**. As for  $\beta(Z)$ -**3j**, the generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  5.80 (d, J = 15.2 Hz, 0.10H) in CDCl<sub>3</sub> with that at  $\delta$  5.99 (d, J = 15.0 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>11</sup> As for  $\alpha$ -**3j**, the generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.76 (d, J = 2.8 Hz, 0.06H) and  $\delta$  5.48 (d, J = 2.8 Hz, 0.06H) in CDCl<sub>3</sub> with that at  $\delta$  5.98 – 5.93 (d, J = 2.73 Hz, 1H), 5.70 - 5.65 (d, J = 2.76 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>12</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, three isomers)  $\delta$  the  $\beta(E)$  isomer 7.39 – 7.31 (overlapped, 1.68H), 7.23 – 7.13 (overlapped, 4.20H), 6.88 – 6.79 (overlapped, 1.68H), 6.70 (d, J = 19.2 Hz, 0.84H), 6.56 (d, J = 18.8 Hz, 0.84H), 0.24 (s, 5.04H).  $\delta$  the  $\beta(Z)$  isomer 7.39 – 7.31 (overlapped, 0.20H), 7.23 – 7.13 (overlapped, 0.60H), 6.88 – 6.79 (overlapped, 0.20H), 5.80 (d, J = 15.2 Hz, 0.10H), 0.14 (s, 0.60H).  $\delta$  the  $\alpha$  isomer 7.39 – 7.31 (overlapped, 0.20H), 7.23 –

7.13 (overlapped, 0.30H), 6.88 – 6.79 (overlapped, 0.12H), 5.76 (d, J = 2.8 Hz, 0.06H), 5.48 (d, J = 2.8 Hz, 0.06H), 0.21 (s, 0.36H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 162.8 (d, J = 246 Hz), 144.1, 138.6, 134.0, 133.1, 129.2, 128.2 (d, J = 8.1 Hz), 128.0, 127.0 (d, J = 2.2 Hz), 115.6 (d, J = 21.5 Hz), -2.4. <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>, three isomers)  $\delta$  the  $\beta(E)$  isomer -113.7 (s, 0.84F);  $\delta$  the  $\beta(Z)$  isomer -114.3 (s, 0.10F);  $\delta$  the  $\alpha$  isomer -116.8 (s, 0.06F). The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>9</sup> HRMS (ESI) *m/z*: calcd for C<sub>16</sub>H<sub>18</sub>FSi [M+H]<sup>+</sup>: 257.1156, found 257.1161.





*α*-3k

β(E)-**3k** 

The mixture of compound  $\beta(E)$ -**3k** and compound  $\alpha$ -**3k** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3k** could not be separated as pure forms from major isomer  $\beta(E)$ -**3k**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.93 (d, J = 2.8 Hz, 0.11H) and  $\delta$  5.66 (d, J = 2.8 Hz, 0.11H) in CDCl<sub>3</sub> with that at  $\delta$  5.99 (d, J = 2.7 Hz, 1H) and  $\delta$  5.71 (d, J = 2.7 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>3</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.56 – 7.47 (overlapped, 1.78H), 7.36 – 7.30 (overlapped, 4.45H), 7.28 – 7.24 (overlapped, 1.78H), 6.85 (d, J = 19.2 Hz, 0.89H), 6.53 (d, J = 19.2 Hz, 0.89H), 0.41 (s, 5.34H).  $\delta$  the  $\alpha$  isomer 7.56 – 7.47 (overlapped, 0.22H), 7.36 – 7.30 (overlapped, 0.55H), 7.28 – 7.24 (overlapped, 0.22H), 5.93 (d, J = 2.8 Hz, 0.11H), 5.66 (d, J =2.8 Hz, 0.11H), 0.41 (s, 0.66H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 144.0, 138.4, 136.8, 134.0, 133.1, 129.3, 128.8, 128.2, 128.0, 127.8, -2.5. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>9</sup>



S-21

The mixture of compound  $\beta(E)$ -**31** and compound  $\alpha$ -**31** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**31** could not be separated as pure forms from major isomer  $\beta(E)$ -**31**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  5.87 (d, J = 2.8 Hz, 0.13H) and  $\delta$  5.59 (d, J = 2.8 Hz, 0.13H) in CDCl<sub>3</sub> with that at  $\delta$  5.99 (d, J = 2.7 Hz, 1H) and  $\delta$  5.71 (d, J = 2.7 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>7</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the *E* isomer 7.48 – 7.43 (overlapped, 1.74H), 7.33 (d, J = 8.4 Hz, 1.74H), 7.29 – 7.24 (overlapped, 2.61H), 7.18 (d, J = 8.4 Hz, 1.74H), 6.76 (d, J = 19.2 Hz, 0.87H), 6.48 (d, J = 19.2 Hz, 0.87H), 0.34 (s, 5.22H).  $\delta$  the *Z* isomer 7.48 – 7.43 (overlapped, 0.26H), 7.29 – 7.24 (overlapped, 0.65H), 6.86 (d, J = 8.4 Hz, 0.26H), 5.87 (d, J = 2.8 Hz, 0.13H), 5.59 (d, J = 2.8 Hz, 0.13H), 0.31 (s, 0.78H).<sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the *E* isomer 144.0, 138.3, 137.2, 134.0, 131.7, 129.3, 128.4, 128.1, 128.0, 122.1, -2.5. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>4</sup>

#### (*E*)-4-(2-(Dimethyl(phenyl)silyl)vinyl)benzonitrile ( $\beta$ (*E*)-3m)



52.7 mg, 40%, β(E)/α > 99/1

The product was purified by silica gel column chromatography (Hexane/EA = 100:1) to afford  $\beta(E)$ -**3m** (52.7 mg, 40%) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.63-7.59 (m, 2H), 7.59-7.54 (m, 2H), 7.53-7.48 (m, 2H), 7.41-7.38 (m, 3H), 6.92 (d, *J* = 19.2 Hz, 1H), 6.76 (d, *J* = 18.8 Hz, 1H), 0.47 (s, 6H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  143.2, 142.3, 137.7, 134.0, 132. 7, 132.5, 129.4, 128.1, 127.1, 119.1, 111.3, -2.7. The data are consistent with those previously reported.<sup>4</sup>

(*E*)-Dimethyl(phenyl)(4-(trifluoromethyl)styryl)silane ( $\beta(E)$ -3n) and dimethyl(phenyl)(1-(4-(trifluoromethyl)phenyl)vinyl)silane ( $\alpha$ -3n)



113.4 mg, 74%,  $\beta(E)/\alpha$  = 88/12

The mixture of compound  $\beta(E)$ -**3n** and compound  $\alpha$ -**3n** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3n** could not be separated as pure forms from major isomer  $\beta(E)$ -**3n**. The generation thereof was additionally confirmed by comparison of its

distinct alkenyl proton signals at  $\delta$  5.94 (d, J = 2.4 Hz, 0.12H) and  $\delta$  5.71 (d, J = 2.8 Hz, 0.12H) in CDCl<sub>3</sub> with that at  $\delta$  6.02 (d, J = 2.7 Hz, 1H) and  $\delta$  5.78 (d, J = 2.7 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>3</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.55 – 7.44 (overlapped, 5.28H), 7.35 – 7.30 (overlapped, 2.64H), 6.91 (d, J = 19.2 Hz, 0.88H), 6.65 (d, J = 18.8 Hz, 0.88H), 0.42 (s, 5.28H);  $\delta$  the  $\alpha$  isomer 7.55 – 7.44 (overlapped, 0.72H), 7.35 – 7.30 (overlapped, 0.36H), 5.94 (d, J = 2.4 Hz, 0.12H), 5.71 (d, J = 2.8 Hz, 0.12H), 0.38 (s, 0.72H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 143.8, 141.6, 138.1, 134.1, 131.0, 130.1, 129.8, 129.4, 128.1, 126.8, 125.7, 125.7, 125.6, 125.6, -2.5. <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer -62.3 (s, 2.64F);  $\delta$  the  $\alpha$  isomer -62.2 (s, 0.36F). The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>13</sup>







The mixture of compound  $\beta(E)$ -30 and compound  $\alpha$ -30 was isolated by silica gel column chromatography (Hexane/EA = 10:1) as a colorless oil. Compounds  $\alpha$ -30 could not be separated as pure forms from major isomer  $\beta(E)$ -30. Because compound  $\alpha$ -30 has not been reported in the literature, the comparison with the alkenvl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -3a bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$ 5.97 (d, J = 2.4 Hz, 0.13H) and  $\delta$  5.76 (d, J = 2.4 Hz, 0.13H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -3a at  $\delta$  6.01 (d, J = 2.8, Hz 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> reported previously. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 8.11 (d, J = 8.8 Hz, 1.74H), 7.52 - 7.45 (overlapped, 3.48H), 7.36 - 7.28 (overlapped, 2.61H), 6.91 (d, J = 8.8 Hz, 1.74H), 7.52 - 7.45 (overlapped, 3.48H), 7.36 - 7.28 (overlapped, 2.61H), 6.91 (d, J = 8.8 Hz, 1.74H), 7.52 - 7.45 (overlapped, 3.48H), 7.36 - 7.28 (overlapped, 2.61H), 6.91 (d, J = 8.8 Hz, 1.74H), 7.52 - 7.45 (overlapped, 3.48H), 7.36 - 7.28 (overlapped, 2.61H), 6.91 (d, J = 8.8 Hz, 1.74H), 7.52 - 7.45 (overlapped, 3.48H), 7.36 - 7.28 (overlapped, 2.61H), 6.91 (d, J = 8.8 Hz, 1.74H), 7.52 - 7.45 (overlapped, 3.48H), 7.36 - 7.28 (overlappeJ = 19.2 Hz, 0.87H), 6.75 (d, J = 19.2 Hz, 0.87H), 0.41 (s, 5.22H).  $\delta$  the  $\alpha$  isomer 8.01 (d, J = 8.8 Hz, 0.26H, 7.52 - 7.45 (overlapped, 0.52H), 7.36 - 7.28 (overlapped, 0.39H), 5.97 (d, J = 2.4 Hz, 0.13H), 5.76 (d, J = 2.4 Hz, 0.13H), 0.37 (s, 0.78H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 147.3, 144.2, 142.8, 137.6, 134.0, 134.0, 129.5, 128.1, 127.2, 124.1, -2.7. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>4</sup> HRMS (ESI) *m/z*: calcd for C<sub>16</sub>H<sub>18</sub>NO<sub>2</sub>Si [M+H]<sup>+</sup>: 284.1101, found 284.1105.

Methyl (*E*)-4-(2-(dimethyl(phenyl)silyl)vinyl)benzoate ( $\beta(E)$ -3p) and methyl 4-(1-(dimethyl(phenyl)silyl)vinyl)benzoate ( $\alpha$ -3p)





The mixture of compound  $\beta(E)$ -**3p** and compound  $\alpha$ -**3p** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -3p could not be separated as pure forms from major isomer  $\beta(E)$ -3p. Because compound  $\alpha$ -3p has not been reported in the literature, the comparison with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -3a bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$ 6.01 (d, J = 2.8 Hz, 0.11H) and  $\delta$  5.74 (d, J = 2.8 Hz, 0.11H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -3a at  $\delta$  6.01 (d, J = 2.8 Hz, 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> reported previously. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 8.00 (d, J = 8.4 Hz, 1.78H), 7.58 - 7.52 (overlapped, 1.78H), 7.49 (d, J = 8.4 Hz, 1.78H), 7.39 - 7.35(overlapped, 2.67H), 6.96 (d, J = 19.2 Hz, 0.89H), 6.73 (d, J = 19.2 Hz, 0.89H), 3.91 (s, 2.67H), 0.46 (s, 5.34H).  $\delta$  the  $\alpha$  isomer 7.96 (d, J = 8.4 Hz, 0.22H), 7.58 – 7.52 (overlapped, 0.22H), 7.39 – 7.35 (overlapped, 0.33H), 7.15 (d, J = 8.4 Hz, 0.22H), 6.01 (d, J = 2.8 Hz, 0.11H), 5.74 (d, J = 2.8 Hz, 0.11H), 3.88 (s, 0.33H), 0.42 (s, 0.66H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 170.0, 144.2, 142.5, 138.1, 134.0, 131.0, 130.0, 129.6, 129.3, 128.0, 126.5, 52.2, -2.5. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>14</sup> HRMS (ESI) *m/z*: calcd for C<sub>18</sub>H<sub>21</sub>O<sub>2</sub>Si [M+H]<sup>+</sup>: 297.1305, found 297.1309.



2a

1a

18.0 mg, 15%,  $\beta(E)/\beta(Z)/\alpha = 86/7/7$ 

*α*-3q

*β*-3q

The mixture of compound  $\beta$ -**3q** and compound  $\alpha$ -**3q** was isolated by silica gel column chromatography (Hexane/EA = 20:1) as a colorless oil. Compounds  $\beta(Z)$ -**3q** and  $\alpha$ -**3q** could not be separated as pure forms from major isomer  $\beta(E)$ -**3q**. As for  $\beta(Z)$ -**3q**, the generation thereof was additionally confirmed

by comparison of its distinct alkenyl proton signal at  $\delta$  6.10 (d, J = 15.2 Hz, 0.07H) in CDCl<sub>3</sub> with that at  $\delta$  6.13 (d, J = 15.2 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>15</sup> As for  $\alpha$ -3q, because compound  $\alpha$ -3q has not been reported in the literature, the comparison with the alkenyl proton signal in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -**3a** bearing the closely related structure was made as an additional effort to identify the structure of the isomer. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.93 (d, J = 2.4 Hz, 0.07H) and  $\delta$  5.71 (d, J = 2.4Hz, 0.07H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\alpha$ -3a at  $\delta$ 6.01 (d, J = 2.8 Hz, 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> reported previously. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, three isomers)  $\delta$  the  $\beta(E)$  isomer 8.54 (s, 0.86H), 8.40 – 8.38 (overlapped, 0.86H), 7.70 - 7.67 (overlapped, 0.86H), 7.50 - 7.47 (overlapped, 1.72H), 7.32 - 7.24 (overlapped, 2.58H), 7.20 - 7.16 (overlapped, 0.86H), 6.83 (d, J = 19.2 Hz, 0.86H), 6.61 (d, J = 18.8 Hz, 0.86H), 0.38 (s, 5.58H).  $\delta$  the  $\beta(Z)$  isomer 8.40 – 8.38 (overlapped, 0.14H), 7.70 – 7.67 (overlapped, 0.07H), 7.50 – 7.47 (overlapped, 0.14H), 7.32 – 7.24 (overlapped, 0.21H), 7.20 – 7.16 (overlapped, 0.07H), 5.93 (d, J = 2.4 Hz, 0.07H), 5.71 (d, J = 2.4 Hz, 0.07H), 0.33 (s, 0.42H).  $\delta$  the  $\alpha$  isomer 8.40 – 8.38 (overlapped, 0.14H), 7.70 – 7.67 (overlapped, 0.07H), 7.50 – 7.47 (overlapped, 0.21H), 7.32 – 7.24 (overlapped, 0.21H), 7.20 - 7.16 (overlapped, 0.07H), 6.10 (d, J = 15.2 Hz, 0.07H), 0.33 (s, 0.42H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ the *E* isomer 149.1, 148.7, 141.6, 138.0, 134.0, 133.1, 133.0, 130.8, 129.4, 128.1, 123.6, -2.6. The data the  $\beta(E)$  isomer are consistent with those previously reported.<sup>16</sup> HRMS (ESI) m/z: calcd for C<sub>15</sub>H<sub>18</sub>NSi [M+H<sup>+</sup>]: 240.1203, found 240.1203.

# (*E*)-(2-(4,4-Dimethylthiochroman-6-yl)vinyl)dimethyl(phenyl)silane ( $\beta(E)$ -3r) and (1-(4,4-dimethylthiochroman-6-yl)vinyl)dimethyl(phenyl)silane ( $\alpha$ -3r)





The mixture of compound  $\beta(E)$ -**3r** and compound  $\alpha$ -**3r** was isolated by silica gel column chromatography (Hexane/EA = 100:1) as a colorless oil. Compounds  $\alpha$ -**3r** could not be separated as pure forms from major isomer  $\beta(E)$ -**3r**. Because compound  $\beta(E)$ -**3r** and  $\alpha$ -**3r** have not been reported in the literature, the comparison with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\beta(E)$ -**3a** and  $\alpha$ -**3a** bearing the closely related structure was made as an additional effort to identify the structure of the isomers. As for  $\beta(E)$ -**3r**, the generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  6.78 (d, J = 18.8 Hz, 0.82H), 6.39 (d, J = 19.2 Hz, 0.82H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of  $\beta(E)$ -**3a** at  $\delta$  6.97 (d, J = 19.2 Hz, 0.92H) and 6.61 (d, J = 19.2 Hz, 0.92H) in CDCl<sub>3</sub> reported previously. As for *a*-**3r**, the generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  5.92 (d, J = 2.0 Hz, 0.18H) and  $\delta$  5.55 (d, J = 2.4 Hz, 0.18H) in CDCl<sub>3</sub> with the alkenyl proton signals in the <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of *a*-**3a** at  $\delta$  6.01 (d, J = 2.8 Hz, 0.08H) and  $\delta$  5.69 (d, J = 2.8 Hz, 0.08H) in CDCl<sub>3</sub> reported previously. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.52 – 7.42 (overlapped, 1.64H), 7.36 – 7.19 (overlapped, 3.28H), 7.14 – 7.07 (overlapped, 0.82H), 7.00 – 6.91 (overlapped, 0.82H), 6.78 (d, J = 18.8 Hz, 0.82H), 6.39 (d, J = 19.2 Hz, 0.82H), 2.94 – 2.91 (m, 1.64H), 1.87 – 1.84 (m, 1.64H), 1.24 (s, 4.92H), 0.34 (s, 4.92H).  $\delta$  the *a* isomer 7.52 – 7.42 (overlapped, 0.82H), 6.78 (d, J = 18.8 Hz, 0.82H), 6.18 Hz, 0.18H), 2.89 – 2.85 (m, 0.36H), 1.81 – 1.78 (m, 0.36H), 1.24 (s, 1.08H), 0.33 (s, 1.08H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 145.3, 142.0, 138.9, 134.3, 134.1, 132.2, 129.1, 127.9, 126.8, 125.2, 125.1, 123.9, 37.7, 33.1, 30.2, 23.3, -2.3. HRMS (ESI) *m/z*: calcd for C<sub>21</sub>H<sub>27</sub>SSi [M+H]<sup>+</sup>: 339.1597, found 339.1601.

#### (*E*)-(1,2-Diphenylvinyl)dimethyl(phenyl)silane (3s)



(E)-3s, 108.5 mg, 69%

The product was purified by silica gel column chromatography (Hexane/EA = 100:1) to afford *E*-**3s** (108.5 mg, 69%) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.49 – 7.42 (m, 2H), 7.28 – 7.23 (m, 3H), 7.15 – 7.06 (m, 3H), 7.00 – 6.94 (m, 3H), 6.87 – 6.78 (m, 4H), 6.74 (s, 1H), 0.30 (s, 6H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  145.1, 142.4, 139.3, 137.8, 137.4, 134.4, 129.7, 129.2, 128.7, 128.0, 127.9, 127.8, 127.3, 125.8, -2.9. The data are consistent with those previously reported.<sup>14</sup>

#### Hex-1-en-1-yldimethyl(phenyl)silane ( $\beta$ -3t) and Hex-1-en-2-yldimethyl(phenyl)silane ( $\alpha$ -3t)



55.7 mg, 51%,  $\beta(E)/\beta(Z)/\alpha = 87/4/9$ 

The mixture of compound  $\beta$ -3t and compound  $\alpha$ -3t was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\beta(Z)$ -3t and  $\alpha$ -3t could not be separated as pure forms from

major isomer  $\beta(E)$ -**3t**. As for  $\beta(Z)$ -**3t**, the generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signal at  $\delta$  6.47 (dt, J = 14.4, 7.6 Hz, 0.04H) in CDCl<sub>3</sub> with that at 6.43 (dt, J = 14.5, 7.4 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>11</sup> As for  $\alpha$ -**3t**, the generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.74 – 5.69 (m, 0.09H) in CDCl<sub>3</sub> with that at  $\delta$  5.75 – 5.72 (m, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>1</sup> H NMR (400 MHz, CDCl<sub>3</sub>, three isomers)  $\delta$  the  $\beta(E)$  isomer 7.57 – 7.52 (overlapped, 1.74H), 7.40 – 7.37 (overlapped, 2.61H), 6.16 (dt, J = 18.8, 6.0 Hz, 0.87H), 5.79 (d, J = 18.4 Hz, 0.87H), 2.23 – 2.14 (overlapped, 1.74H), 1.45 – 1.31 (overlapped, 3.48H), 0.98 – 0.82 (overlapped, 2.61H), 0.35 (s, 5.22H).  $\delta$  the  $\alpha$ isomer 7.57 – 7.52 (overlapped, 0.18H), 7.40 – 7.37 (overlapped, 0.27H), 5.74 – 5.69 (m, 0.09H), 5.46 – 5.44 (overlapped, 0.09H), 2.01 – 2.91 (overlapped, 0.18H), 1.45 – 1.31 (overlapped, 0.36H), 0.98 – 0.82 (overlapped, 0.27H), 0.41 (s, 0.54H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 149.6, 139.5, 134.0, 128.9, 127.8, 127.3, 36.7, 30.9, 22.4, 14.2, -2.3. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>9</sup>

#### (*E*)-Diphenyl(styryl)silane ( $\beta(E)$ 3u) and diphenyl(1-phenylvinyl)silane ( $\alpha$ -3u)



54.4 mg, 38%, β(E)/α = 92/8

The mixture of compound  $\beta(E)$ -**3u** and compound  $\alpha$ -**3u** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3u** could not be separated as pure forms from major isomer  $\beta(E)$ -**3u**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  6.20 (dd, J = 2.5, 1.0 Hz, 1H) and  $\delta$  5.60 (d, J = 2.5 Hz, 1H) in CDCl<sub>3</sub> with that at  $\delta$  6.28 (dd, J = 2.5, 1.0 Hz, 1H) and  $\delta$  5.68 (d, J = 2.5 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>7</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.56 – 7.51 (m, 4H), 7.40 – 7.34 (m, 2H), 7.33 – 7.22 (m, 9H), 7.00 (d, J = 19.2 Hz, 1H), 6.63 (dd, J = 19.2 Hz, J = 3.2 Hz, 1H), 5.17 (d, J = 3.2 Hz, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 149.3, 138.0, 135.7, 133.7, 130.0, 128.7, 128.7, 128.2, 126.9, 121.7. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>17</sup>

#### (*E*)-Triethyl(styryl)silane ( $\beta$ (*E*)-3v) and triethyl(1-phenylvinyl)silane ( $\alpha$ -3v)



54.6 mg, 50%, *β*(*E*)/*α* = 92/8

The mixture of compound  $\beta(E)$ -**3v** and compound  $\alpha$ -**3v** was isolated by silica gel column chromatography (Hexane) as a colorless oil. Compounds  $\alpha$ -**3v** could not be separated as pure forms from major isomer  $\beta(E)$ -**3v**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  5.79 (d, J = 3.2 Hz, 0.08H) and  $\delta$  5.49 (d, J = 3.2 Hz, 0.08H) in CDCl<sub>3</sub> with that at  $\delta$  5.86 (dd, J = 2.5, 1.0 Hz, 1H) and  $\delta$  5.56 (d, J = 2.5 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>18</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.37 – 7.34 (overlapped, 1.84H), 7.25 – 7.21 (overlapped, 1.84H), 7.18 – 7.12 (overlapped, 0.92H), 6.82 (d, J = 19.2 Hz, 0.92H), 6.34 (d, J = 19.2 Hz, 0.92H), 0.93 – 0.89 (overlapped, 8.28H), 0.61 – 0.55 (overlapped, 5.52H);  $\delta$  the  $\alpha$  isomer 7.37 – 7.34 (overlapped, 0.16H), 7.25 – 7.21 (overlapped, 0.16H), 7.18 – 7.12 (overlapped, 0.08H), 5.79 (d, J = 3.2 Hz, 0.08H), 5.49 (d, J = 3.2 Hz, 0.08H), 0.93 – 0.89 (overlapped, 0.16H), 7.18 – 7.12 (overlapped, 0.08H), 5.79 (d, J = 3.2 Hz, 0.08H), 5.49 (d, J = 3.2 Hz, 0.08H), 0.93 – 0.89 (overlapped, 0.72H), 0.61 – 0.55 (overlapped, 0.48H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 145.0, 138.6, 128.6, 128.0, 126.5, 126.0, 7.6, 3.7. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>19</sup>

#### (*E*)-Triethoxy(styryl)silane ( $\beta$ (*E*)-3w) and triethoxy(1-phenylvinyl)silane ( $\alpha$ -3w)



38.6 mg, 29%,  $\beta(E)/\alpha$  = 94/6

The mixture of compound  $\beta(E)$ -**3w** and compound  $\alpha$ -**3w** was isolated by silica gel column chromatography (Hexane/EA = 20:1) as a colorless oil. Compounds  $\alpha$ -**3w** could not be separated as pure forms from major isomer  $\beta(E)$ -**3w**. The generation thereof was additionally confirmed by comparison of its distinct alkenyl proton signals at  $\delta$  6.06 (d, J = 3.2 Hz, 0.06H) and  $\delta$  5.89 (d, J = 2.8 Hz, 0.06H) in CDCl<sub>3</sub> with that at  $\delta$  6.15 (d, J = 3.0 Hz, 1H) and  $\delta$  5.97 (d, J = 3.0 Hz, 1H) in CDCl<sub>3</sub> reported in the literature.<sup>20</sup> <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, two isomers)  $\delta$  the  $\beta(E)$  isomer 7.41 – 7.38 (overlapped, 1.88H), 7.28 – 7.20 (overlapped, 2.82H), 7.14 (d, J = 19.2 Hz, 0.94H), 6.10 (d, J = 19.2 Hz, 0.94H), 3.81 (q, J = 6.8 Hz, 5.64H), 1.19 (t, J = 7.2 Hz, 8.46H).  $\delta$  the  $\alpha$  isomer 7.41 – 7.38 (overlapped, 0.12H), 7.28 – 7.20 (overlapped, 0.18H), 6.06 (d, J = 3.2 Hz, 0.06H), 5.89 (d, J = 2.8 Hz, 0.06H), 3.76 (q, J = 6.8 Hz, 0.36H), 1.14 (t, J = 6.4 Hz, 0.54H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  the  $\beta(E)$  isomer 149.3, 137.7, 128.9, 128.7, 126.9, 117.8, 58.7, 18.4. The data of the  $\beta(E)$  isomer are consistent with those previously reported.<sup>21</sup>

#### 9. References

- [1] T. Zhou, P. Gao, R. Lalancette, R. Szostak and M. Szostak, Nat. Chem., 2024, 16, 2025–2035.
- [2] Y. Wang, Q. Wang, L. Wu, K. Jia, M. Wang and Y Qiu, Nat. Commun., 2024, 15, 2780.
- [3] L. Song, Q. Feng, Y. Wang, S. Ding, Y.-D. Wu, X. Zhang, L. W. Chung and J. Sun, J. Am. Chem. Soc., 2019, 141, 17441–17451.
- [4] C. Chen, Q. Mo, Y. Wang and L. Zhang, Inorg. Chem., 2023, 62, 16882–16889.
- [5] L. Zhang and M. Oestreich, Org. Lett., 2018, 20, 8061-8063.
- [6] Y. Gao, S. Yazdani, A. Kendrick IV, G. P. Junor, T. Kang, Do. B. Grotjahn, G. Bertrand, R. Jazzar and K. M. Engle, *Angew. Chem. Int. Ed.*, 2021, 60, 19871–19878.
- [7] M. Skrodzki, V. Patroniak and P. Pawluc, Org. Lett., 2021, 23, 663–667.
- [8] J. R. McAtee, S. B. Krause and D. A. Watson, Adv. Synth. Catal., 2015, 357, 2317–2321.
- [9] Y. Duan, G. Ji, S. Zhang, X. Chen and Y. Yang, Catal. Sci. Technol., 2018, 8, 1039–1050.
- [10] B. K. Allam, S. Azeez and J. Kandasamy, Appl. Organomet. Chem., 2019, 33, e5192.
- [11] H. Liang, Y.-X. Ji, R.-H. Wang, Z.-H. Zhang and B. Zhang, Org. Lett., 2019, 21, 2750–2754.
- [12] M. Skrodzki, V. O. Garrido, A. G. Csáky, P. Pawluc, J. Catal., 2022, 411, 116–121.
- [13] Y. Nagashima, D. Yukimori, C. Wang and M. Uchiyama, Angew. Chem. Int. Ed., 2018, 57, 8053–8057.
- [14] C. Xu, B. Huang, T. Yan and M. Cai, Green Chem., 2018, 20, 391-397.
- [15] H. Xu, S. Yamaguchi, T. Mitsudome and T. Mizugaki, Synlett, 2024, 35, 1296–1300.
- [16] M. Zhong, X. Pannecoucke, P. Jubault and T. Poisson, Chem. Eur. J., 2021, 27, 11818–11822.
- [17] Z. Mo, J. Xiao, Y. Gao and L. Deng, J. Am. Chem. Soc., 2014, 136, 17414-17417.
- [18] R. Cano, M. Yus and D. J. Ramoń, ACS Catal., 2012, 2, 1070–1078.
- [19] H. Andoh, K. Nakamura, Y. Nakazawa, T. Ikeda-Fukazawa, S. Okabayashi and T. Tsuchimoto, *Adv. Synth. Catal.*, 2023, 365, 4583–4594.
- [20] D. Wang, Y. Lai, P. Wang, X. Leng, J. Xiao and L. Deng, J. Am. Chem. Soc., 2021, 143, 12847–12856.
- [21] L. Yang, M. Yi, X. Wu, Y. Lu and Z. Zhang, Chem. Eur. J., 2024, 30, e202402406.

### 10. NMR Spectra





















































