Supplementary Information (SI) for Organic & Biomolecular Chemistry. This journal is © The Royal Society of Chemistry 2025

Visible-light-promoted hydroxysulfenylation of alkenes for the synthesis of β -hydroxysulfides

Rekha Bai,^a Yi-Xuan Jian,^a Shih-Jie Chen,^a Indrajit Karmakar^a and Chin-Fa Lee*^{a,b,c}

- ^a Department of Chemistry, National Chung Hsing University, Taichung, Taiwan 402, Republic of China. Email: cfalee@dragon.nchu.edu.tw
- ^b *i*-Center for Advanced Science and Technology (iCAST), National Chung Hsing University, Taichung, Taiwan 402, Republic of China.
- ^c Innovation and Development Center of Sustainable Agriculture (IDCSA), National Chung Hsing University, Taichung, Taiwan 402, Republic of China.

Entry	Contents	Pg. No
1	Light on-off experiment	S2
2	¹ H-NMR titration experiment	S3-S5
3	UV-vs spectroscopic studies	S5-S6
4	Stern-Volmer Studies	S6-S7
5	EPR experiments	S7-S9
6	References	S9
7	Copies of ¹ H, ¹³ C, ¹⁹ F-NMR and Mass	S10-S164

Light on-off experiment^{S1}

For the on-off experiments, five oven-dried 10 mL Schlenk tubes were charged with styrene 1a (0.5 mmol), thiophenol 2a (1.0 equiv.), DABCO (1.3 equiv.), and EtOH (2 mL). An oxygen-filled balloon was attached to each tube, after which the reaction mixtures were stirred and irradiated under 410 nm Blue-light emitting diode at room temperature. After 1h, the LED was turn off, and one tube was removed from the irradiation setup for analysis. The remaining four tubes were stirred in the absence of light for an additional 1 h. We than, one tube was removed for analysis, and light turn back to on to irradiated remaining three reaction mixtures. After an additional 1h of irradiation, the LED was turn off, and one tube removed for analysis. The remaining two tubes were stirred in the absence of light for an additional 1h. Then, one tube was removed for analysis and the blue LED was turned back on to irradiate the last one reaction mixture. After an additional 1h irradiation, the last one was analyzed. Yield was determined by ¹H NMR spectroscopy using 1,3,5 trimethoxy benzene as internal standard.

Entry	Time (h)	Experiments	3a yield (%)
1	0	-	0
2	1	on	49
3	2	off	50
4	3	on	76
5	4	off	77
6	5	on	90

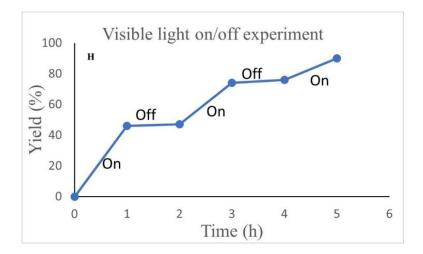
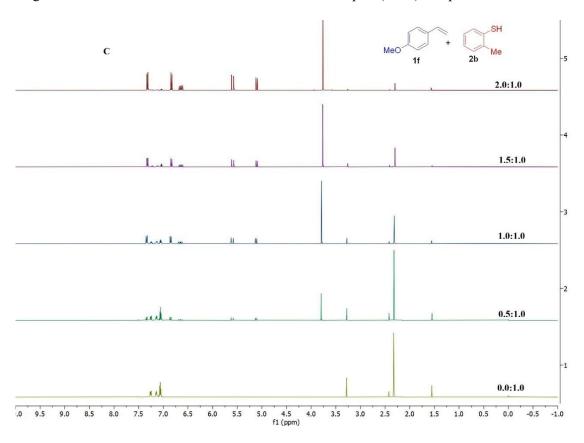
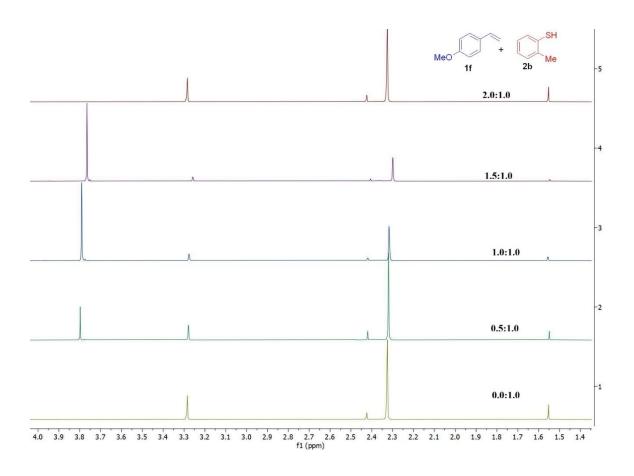


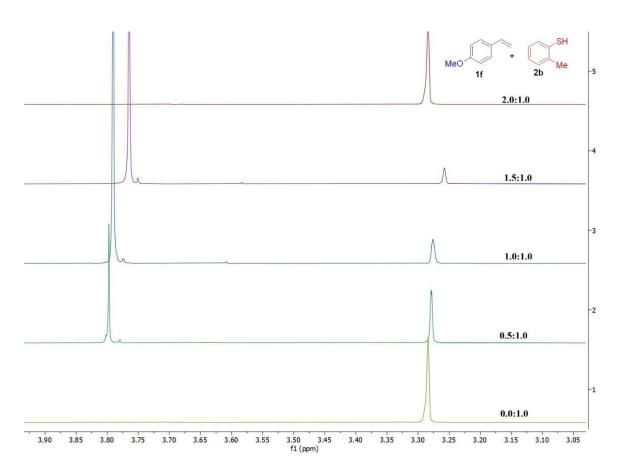
Fig S1: Light on-off experiments

¹H-NMR titration experiment to detect EDA complex formation^{S1}

¹H NMR experiments were carried out in CDCl₃ solution containing 4-methoxy styrene **1f** and 2-methyl thiol **2b** in varying ratio. As the concentration of 4-methoxy styrene **2f** increased, the methyl proton of **2b** shifted little up-field, while the methoxy protons of **1f** shifted little downfield. These reciprocal changes confirmed the formation of an electron donor-acceptor (EDA) complex between **1f** and **2b**







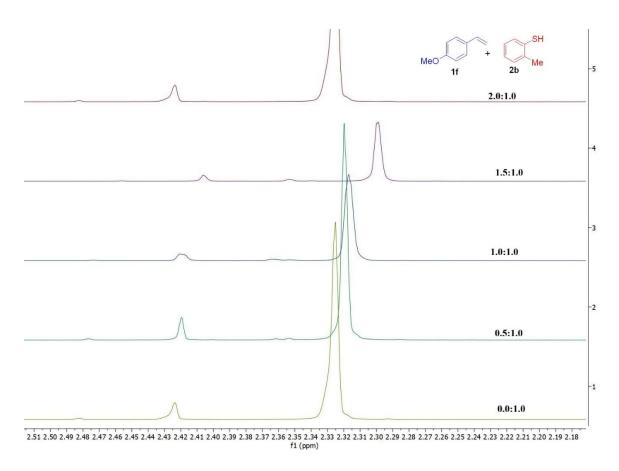


Fig S2: Evidence for the formation of Electron-donor-acceptor complex through ¹H NMR

Spectroscopic studies^{S1}

The UV-Vis absorption spectra of styrene **1a** (25 mM), 2-methoxythiophenol **2d** (50 mM), and DABCO (50 mM) in ethanol were recorded using 1 cm quartz cuvettes on a U-3900 spectrophotometer (Fig. S3-S4). The spectra were obtained for each component individually as well as for their 1:1 binary mixture. In the mixed samples, two new absorption bands appear at wavelength above 300 nm, which were absent in the spectra of the individual components. The appearance of these additional bands suggests the formation of an EDA (electron donor-acceptor) complex between thiolate species and the styrene **1a**, thereby supporting the involvement of an EDA complex in this blue-LED driven reaction pathway.

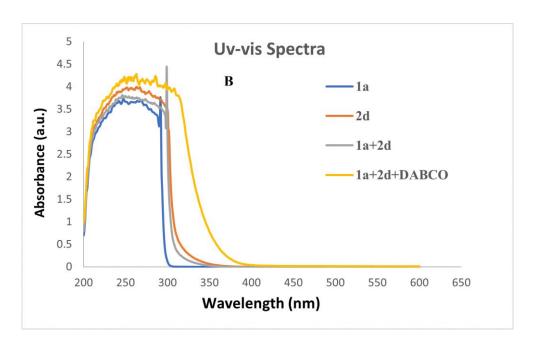


Fig. S3: UV-vis spectroscopic measurements on various 1a, 2d, and DABCO and their combination in EtOH

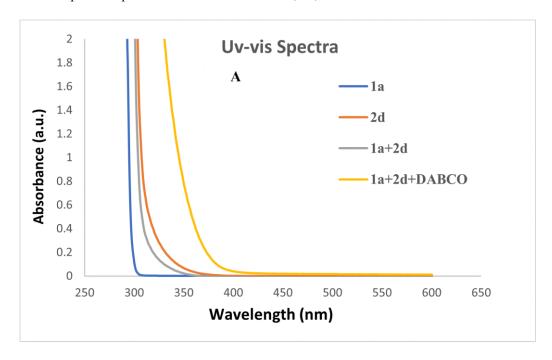


Fig. S4: UV-vis spectroscopic measurements on various 1a, 2d and DABCO their combination in EtOH

Fluorescence studies (Stern-Volmer Studies)

The fluorescence emission intensities were recorded on a F-7000 FL Spectrophotometer. The excitation wavelength was set at 240 nm, and the emission wavelength was measured over the range of 250-470 nm. Samples were prepared by mixing thiophenol 2a (3.0×10^{-5} mol/L) with varying concentration of styrene 1a and DABCO in EtOH (total volume = 2.0 mL) in a quartz fluorescence cuvette. For each quenching experiment, 1.0 mL of styrene 1a or DABCO stock solution at different concentration was added to 1.0 mL of the thiophenol solution, yielding a total volume of 2.0 mL. Then the emission

intensity was collected and the data were presented in Figure S5 and S6.

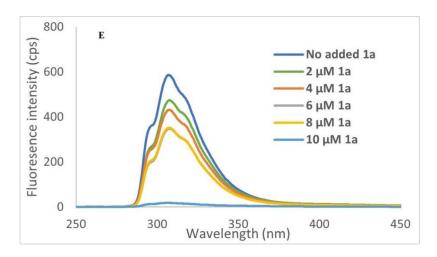


Fig S5: Fluorescence emission in the presence of Styrene (1a).

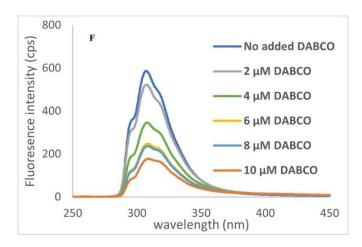


Fig S6: Fluorescence emission in the presence of DABCO

EPR experiments^{S1}

Without Blue-LED irradiation: A mixture of styrene 1a (0.5 mmol), thiophenol 2a (1.0 mmol), DABCO (1.3 equiv.), EtOH (3 mL), and DMPO (12 μ L) was prepared and directly analyzed by EPR.

Standard reaction Conditions:

A mixture of styrene 1a (0.5 mmol), thiophenol 2a (1.0 mmol), DABCO (1.3 equiv.), EtOH (3 mL), and DMPO (12 μ L) was stirred under Blue LED irradiation under oxygen atmosphere at room temperature. Aliquots were taken and analyzed by EPR after 1, 5, 10, 15, and 30 minutes of irradiation.

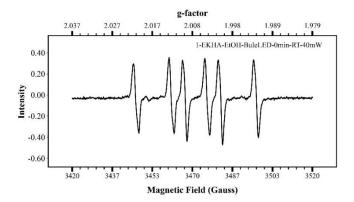


Fig S7: EPR Without Blue-LED irradiation

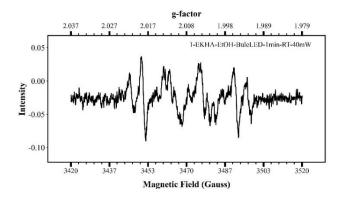


Fig S8: EPR under Blue-LED irradiation after 1 min.

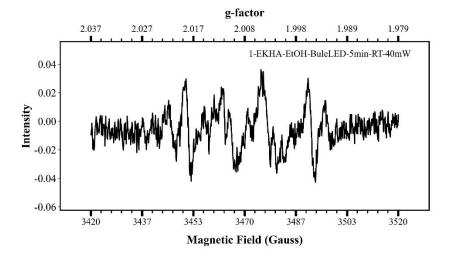


Fig S9: EPR under Blue-LED irradiation after 5 min.

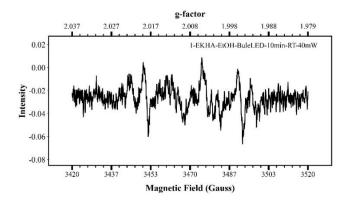


Fig S9: EPR under Blue-LED irradiation after 10 min.

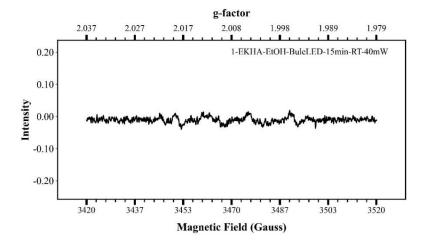


Fig S10: EPR under Blue-LED irradiation after 15 min.

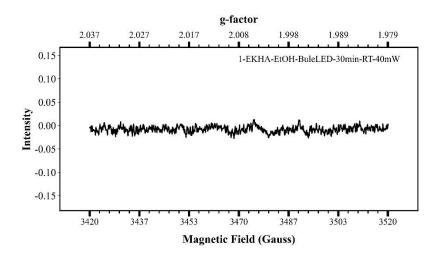
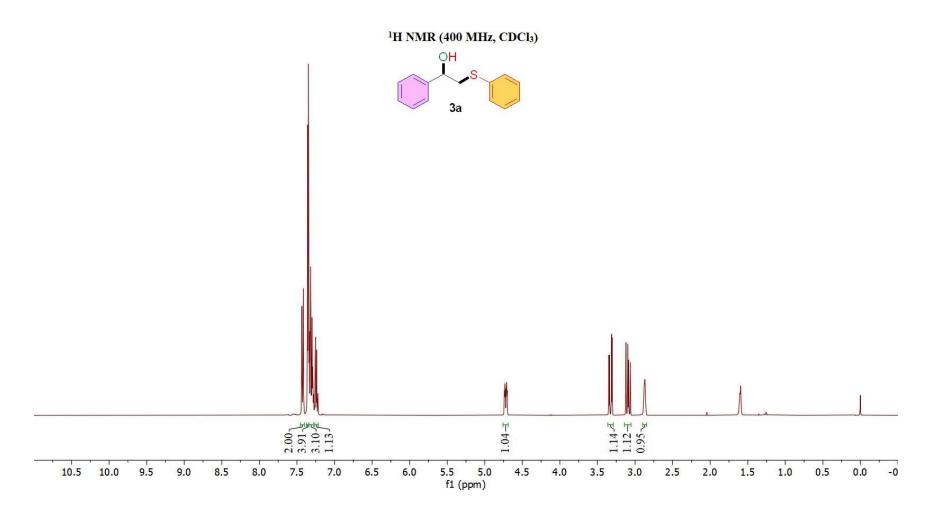


Fig S11: EPR under Blue-LED irradiation after 30 min.

References:

S1 (a) H. Ruan, L.-G. Meng, L. Zhu and L. Wang, *Adv. Synth. Catal.*, 2019, **361**, 3217; (b) Z. Chen, F. Xue, T. Liu, B. Wang, Y. Zhang, W. Jin, Y. Xia, and C. Liu, *Green Chem.*, 2022, **24**, 3250; (c) J. Shi, X. W. Gao, Q. X. Tong, and J. J. Zhong, *J. Org. Chem.*, 2021, **86**, 12922; (d) J. E. Hong, Y. Jung, D. Min, M. Jang, S. Kim, J. Park, and Y. Park, *J. Org. Chem.*, 2022, **87**, 7378; (e) A. Das, and K. J. Thomas, ACS omega, 2023, **8**, 18275. (f) Z. Xing, M. Yang, H. Sun, Z. Wang, P. Chen, L. Liu, X. Wang, X. Xie and X. She, *Green Chem.*, 2018, **20**, 5117.

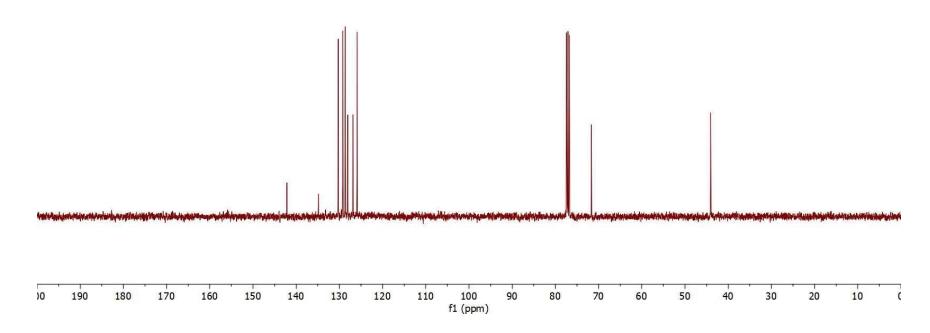
Copies of ¹H, ¹³C, ¹⁹F-NMR and Mass





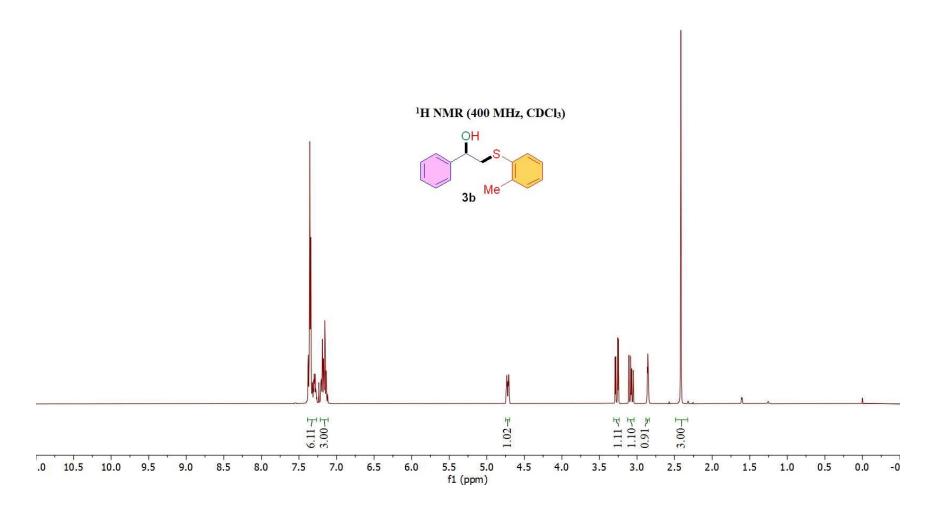




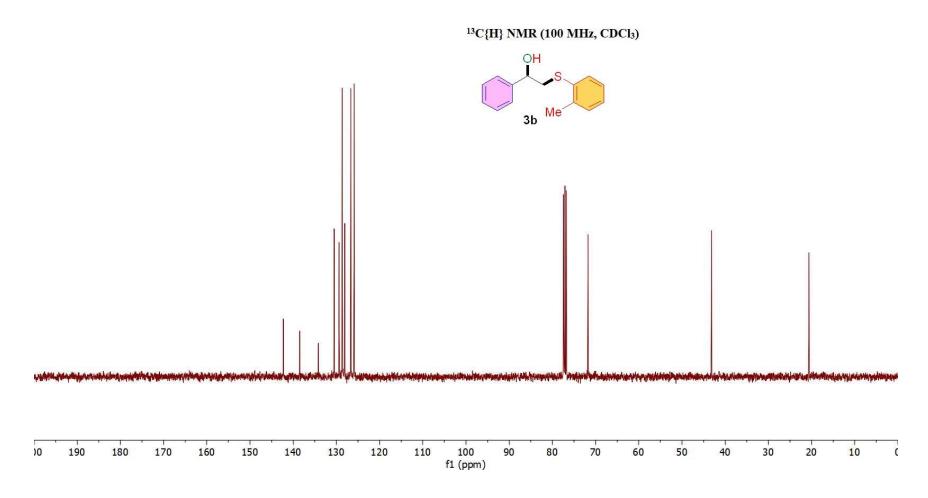


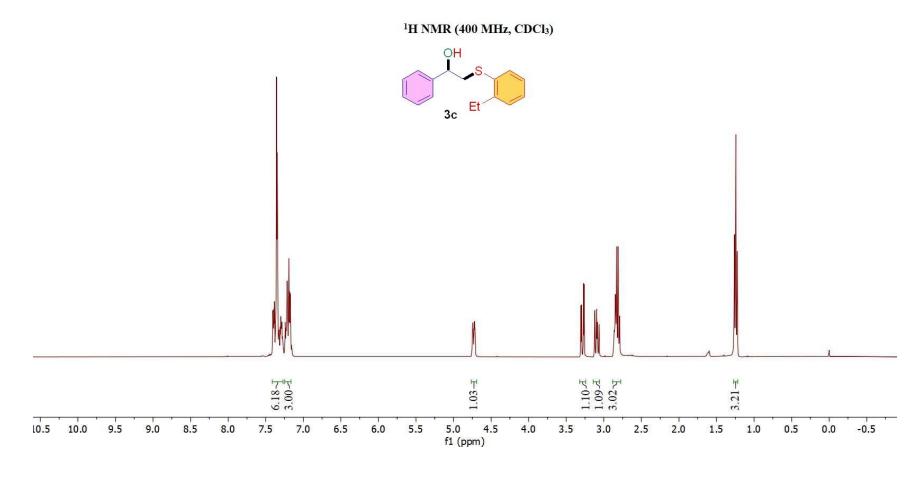




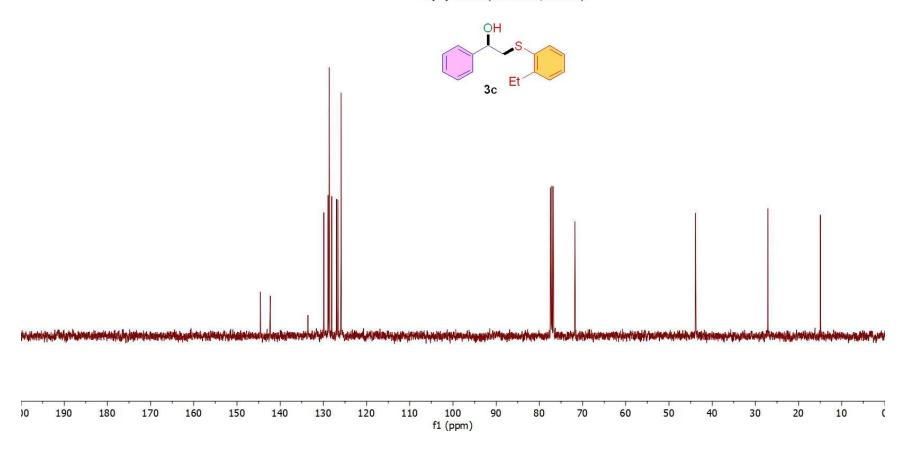






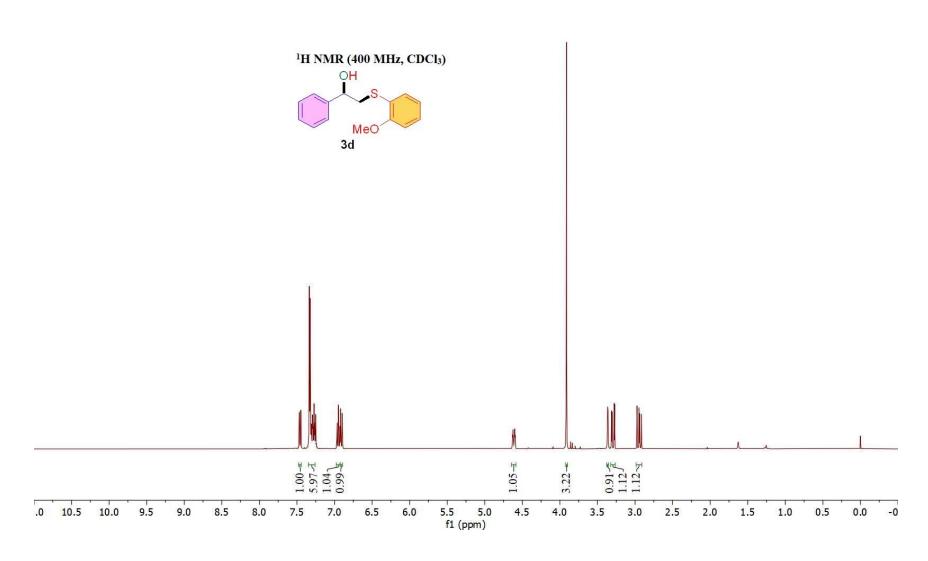


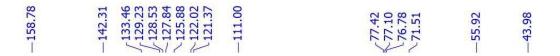


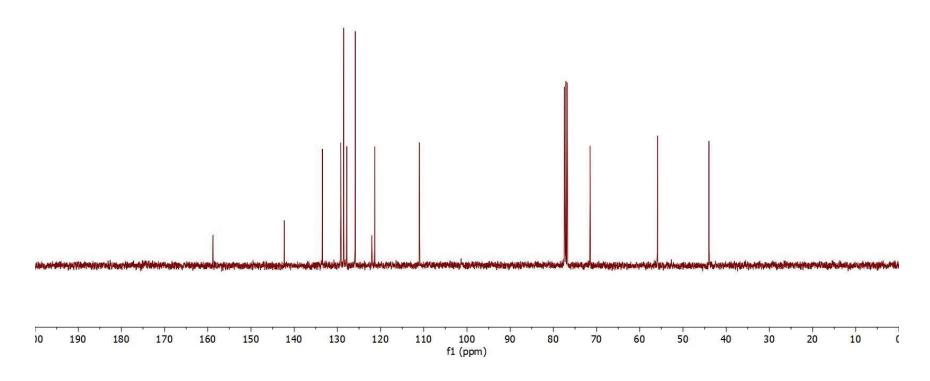




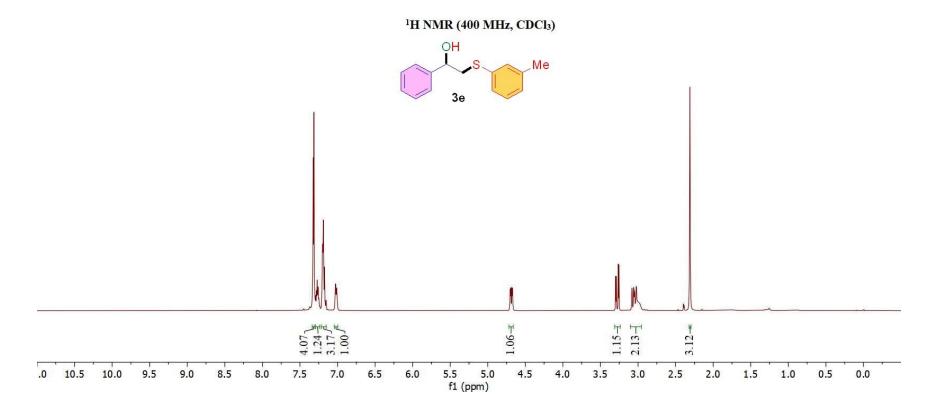


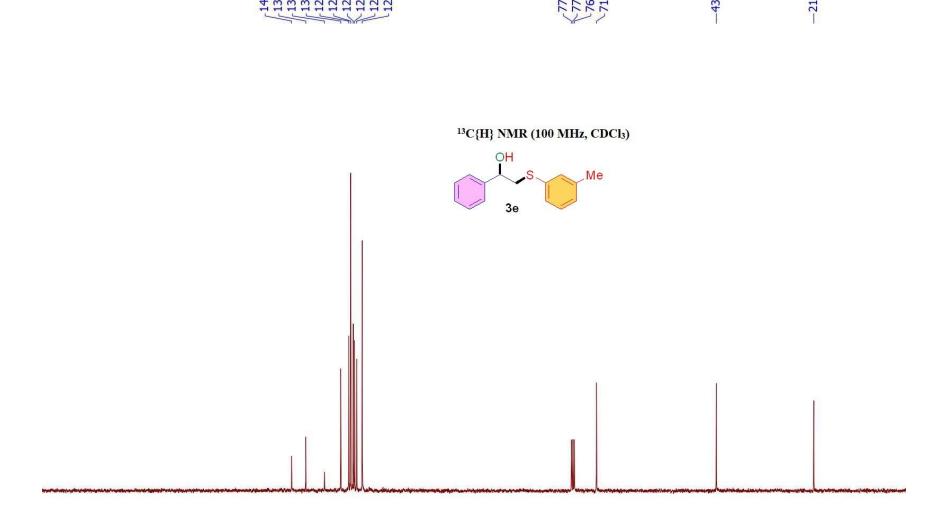










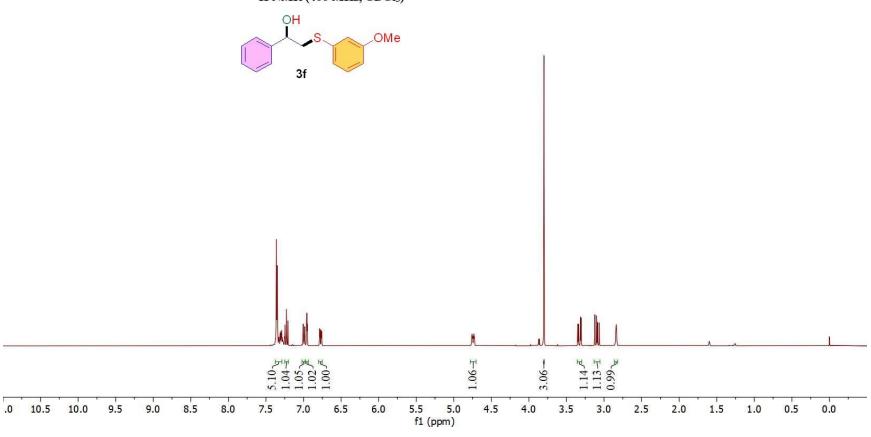


f1 (ppm)



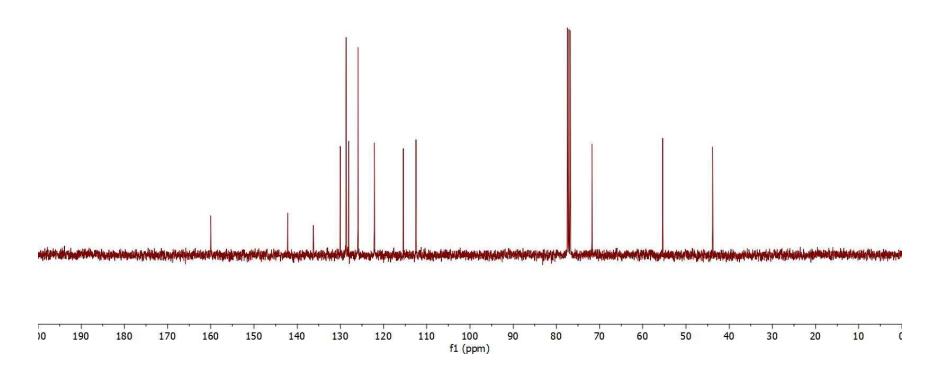


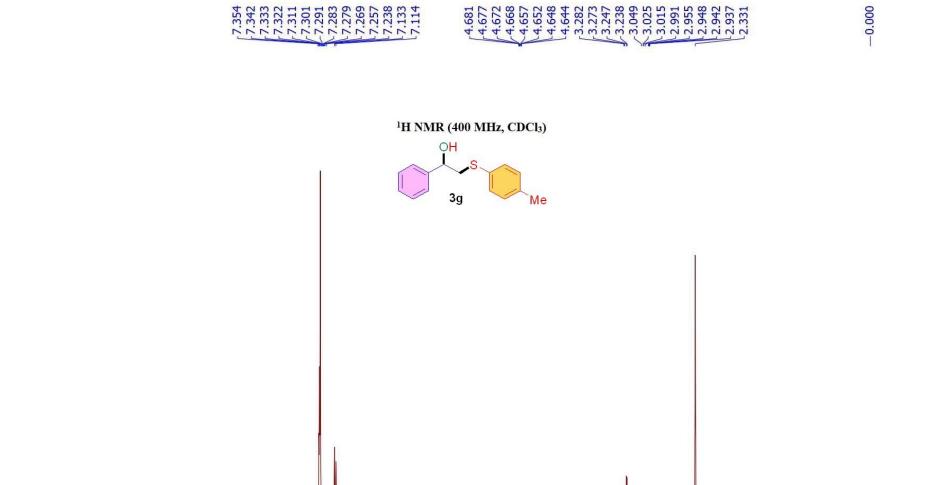






13C{H} NMR (100 MHz, CDCl₃)





5.5 5.0 f1 (ppm) D-66.0

4.5

4.0

3.5

3.0

3.04-₹

2.0

1.5

1.0

0.5

0.0

-0

2.5

7.00년 1.92곡

7.0

6.5

6.0

7.5

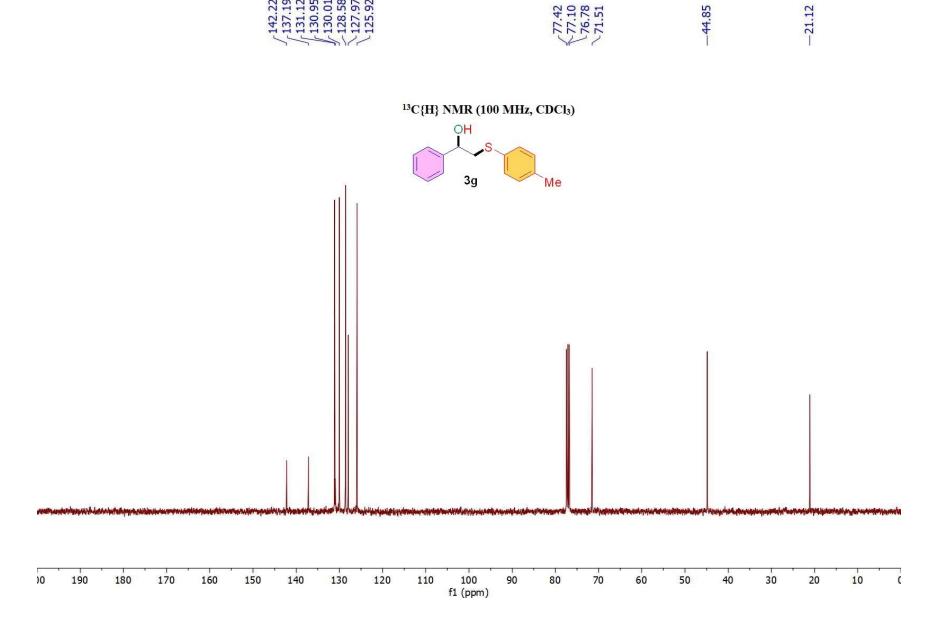
8.0

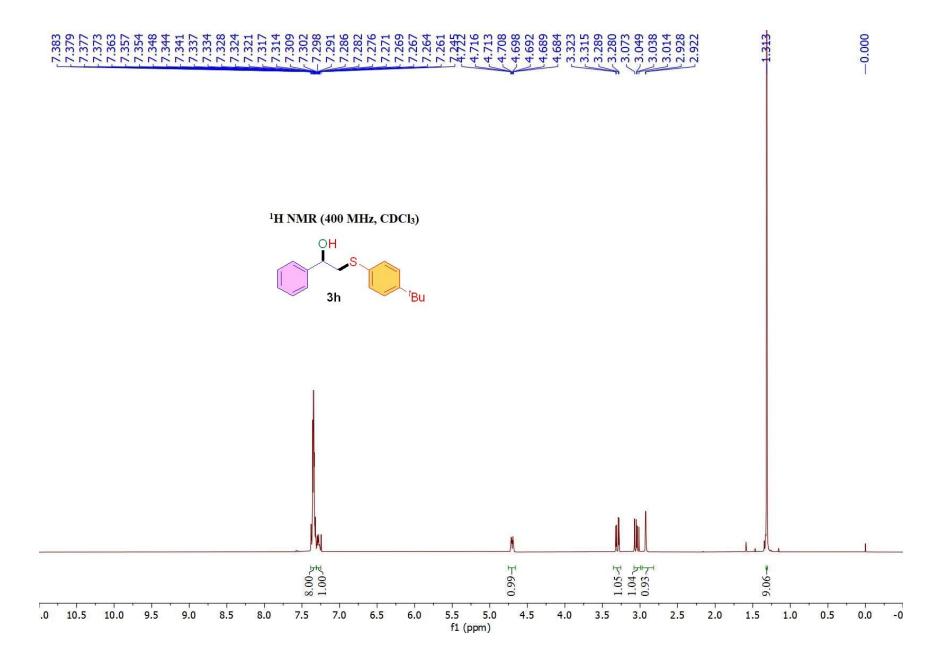
8.5

.0 10.5 10.0

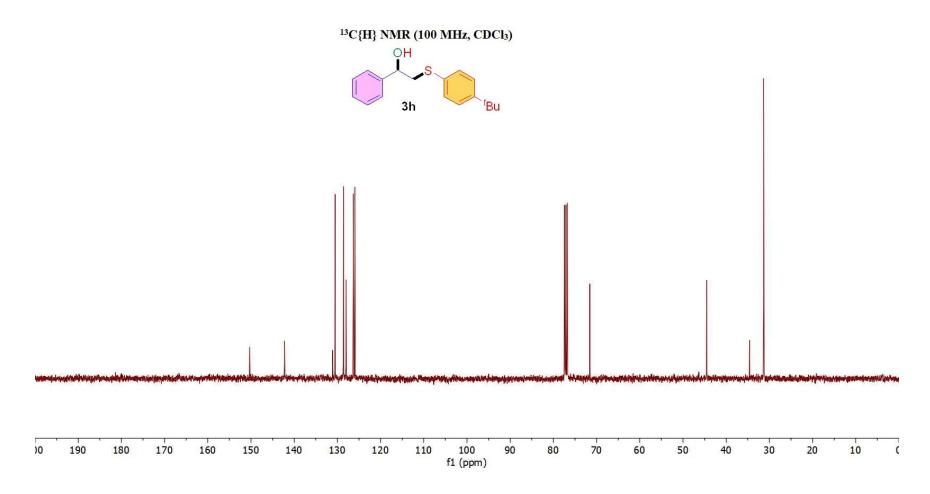
9.5

9.0







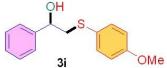


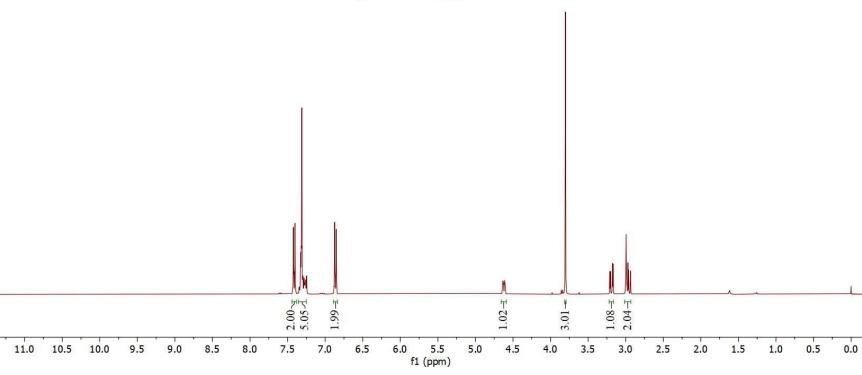




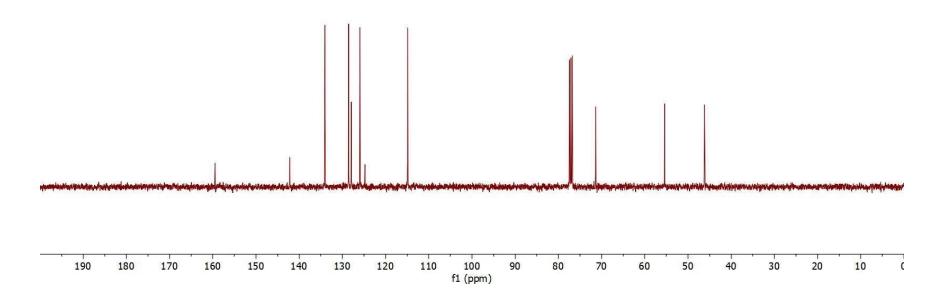






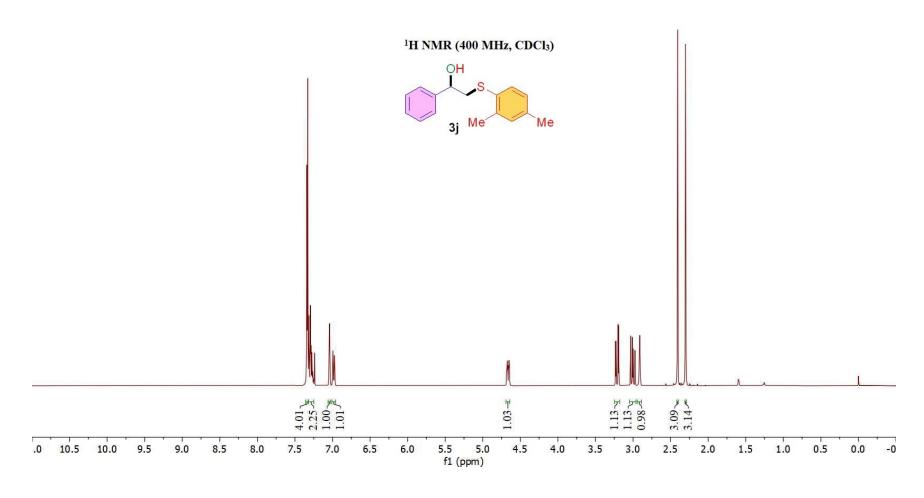








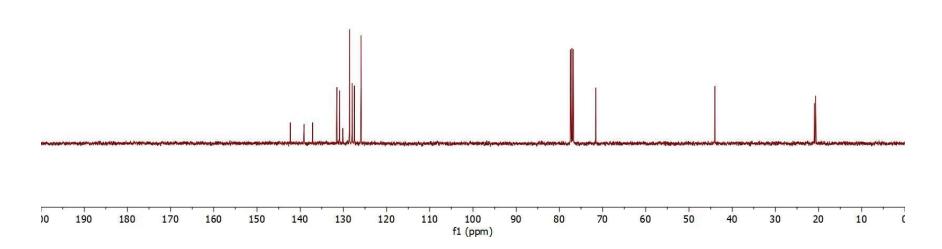






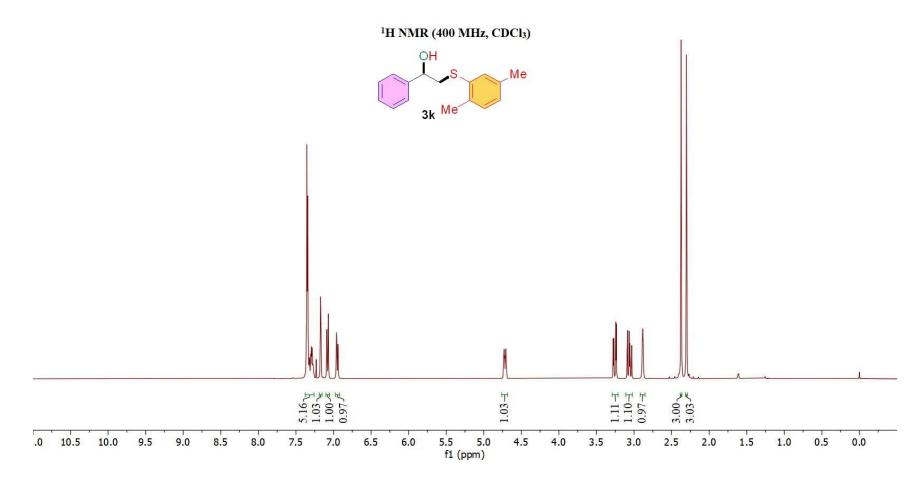


¹³C{H} NMR (100 MHz, CDCl₃)

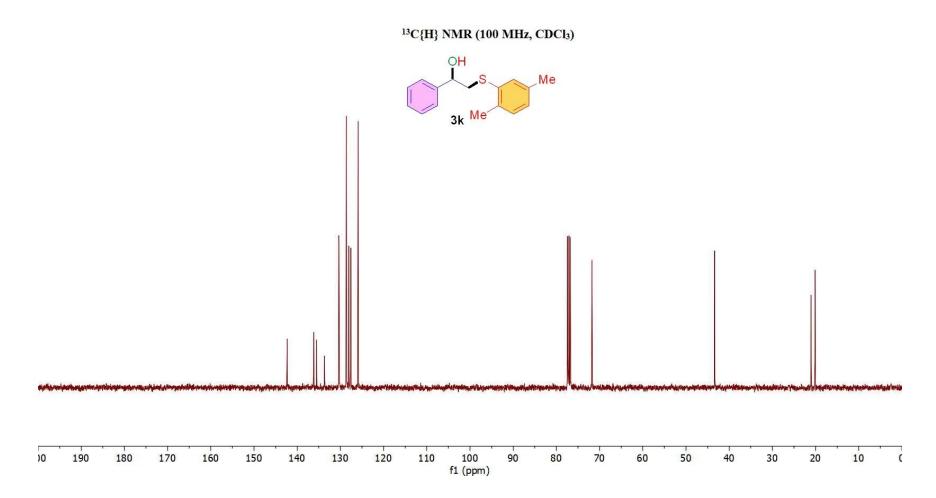


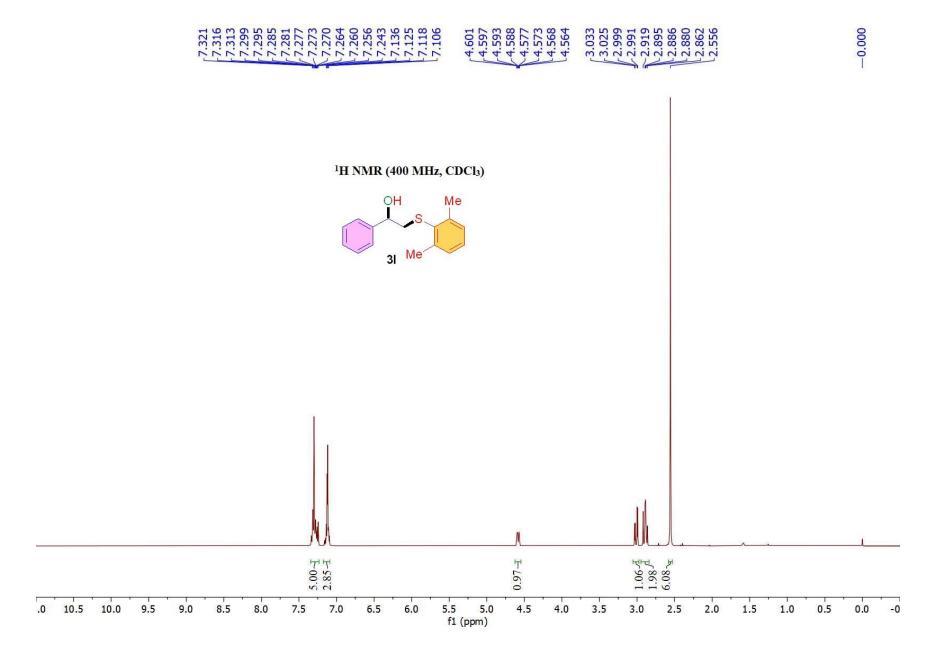




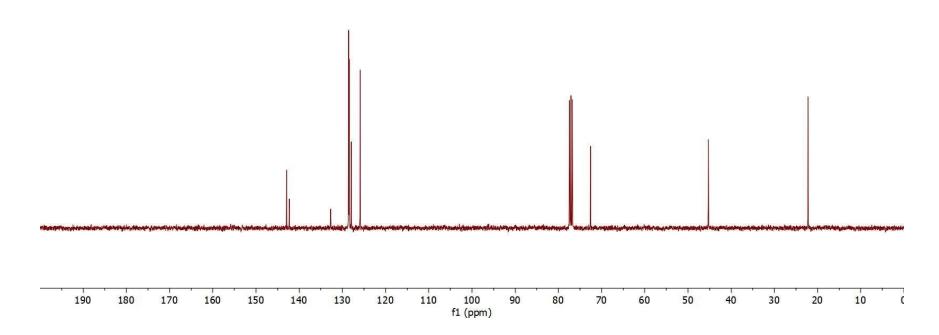


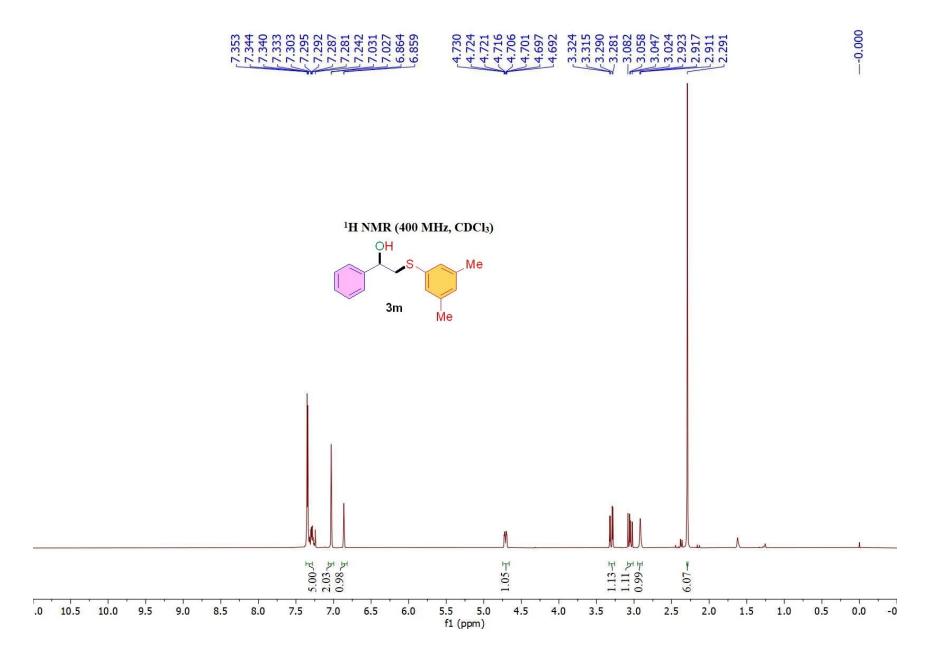




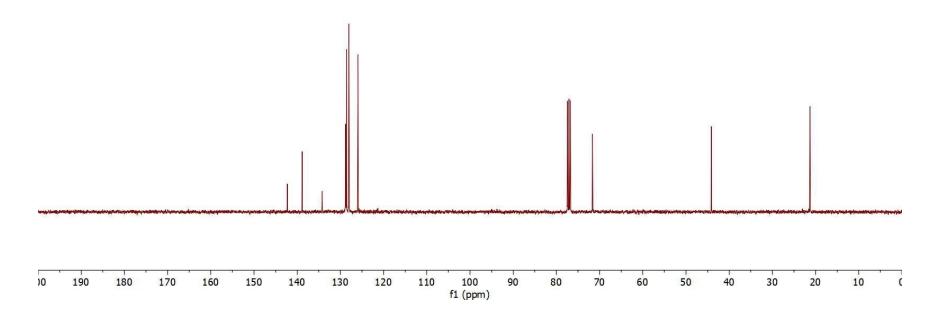








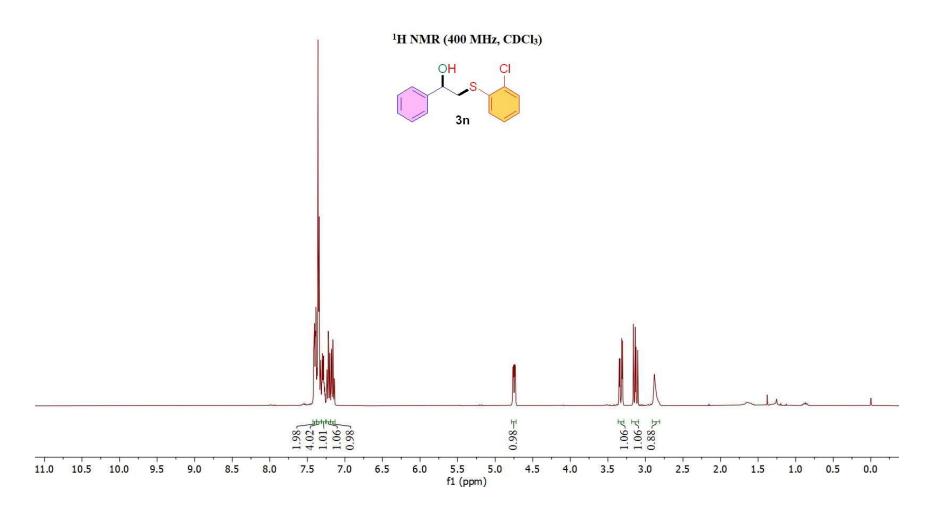




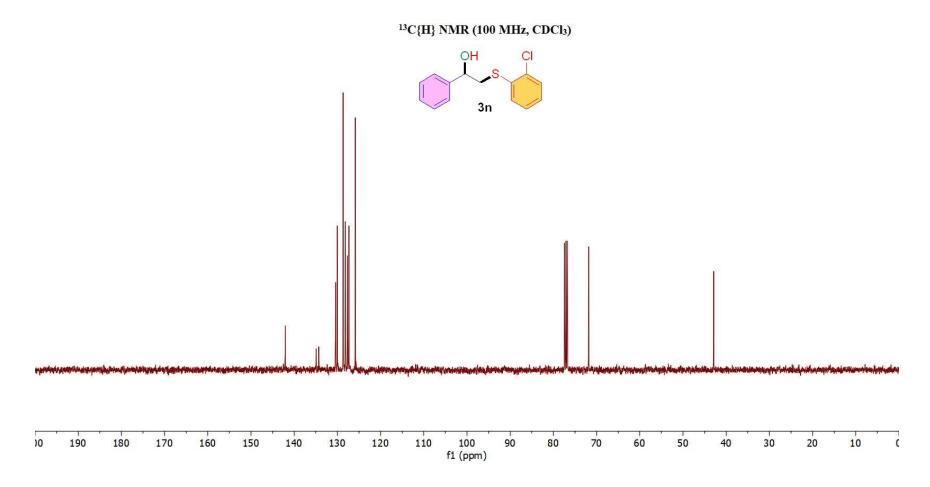




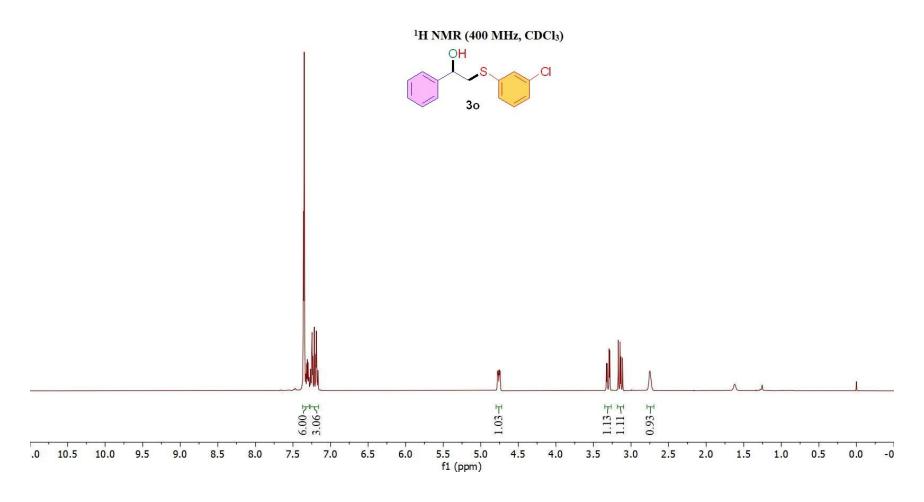








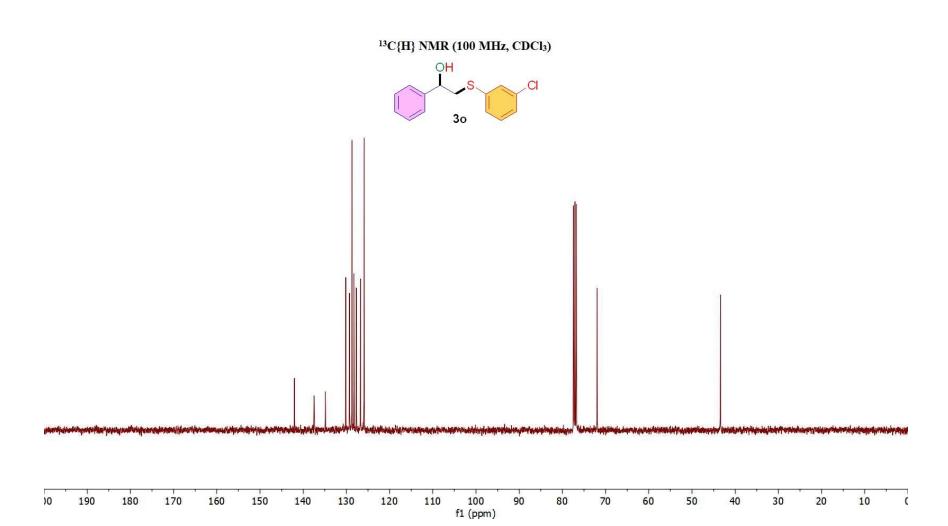




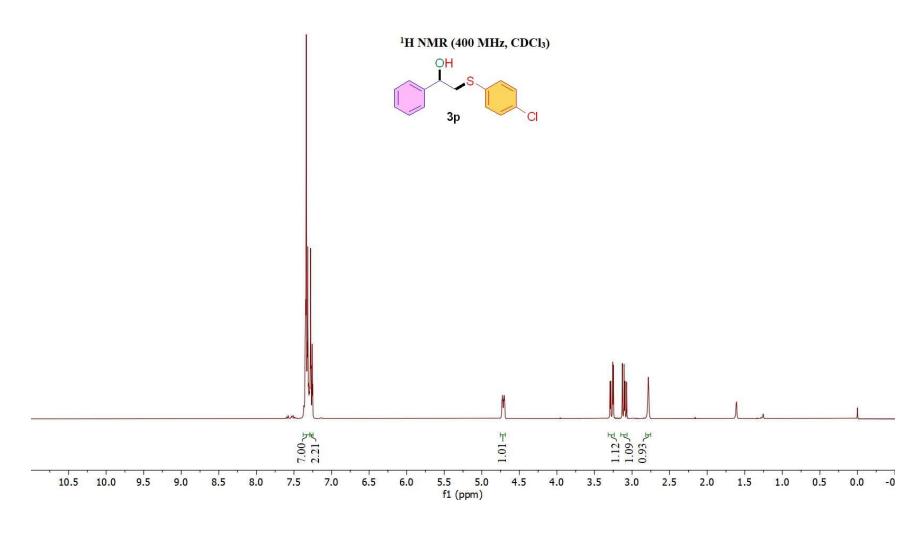








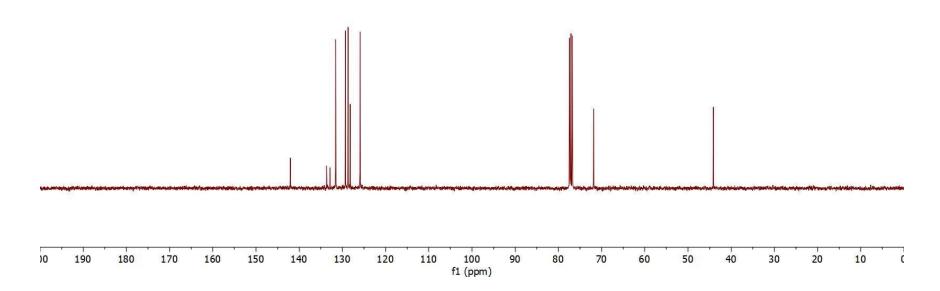




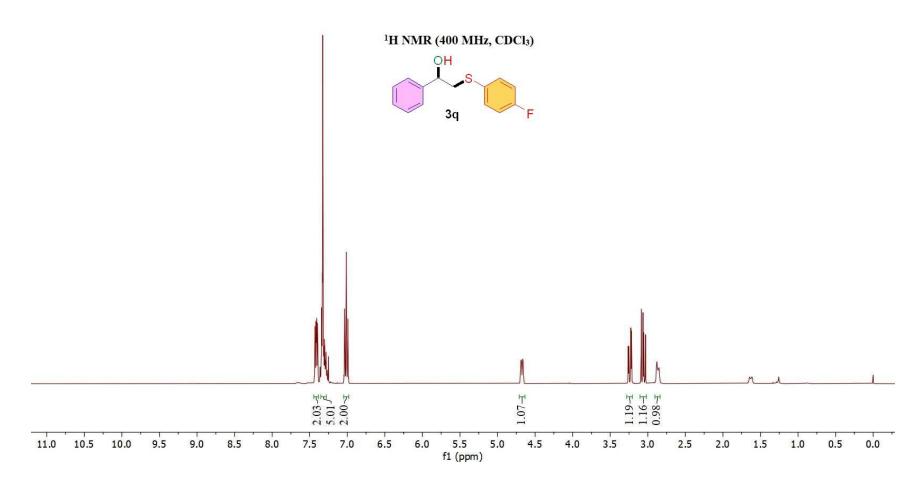


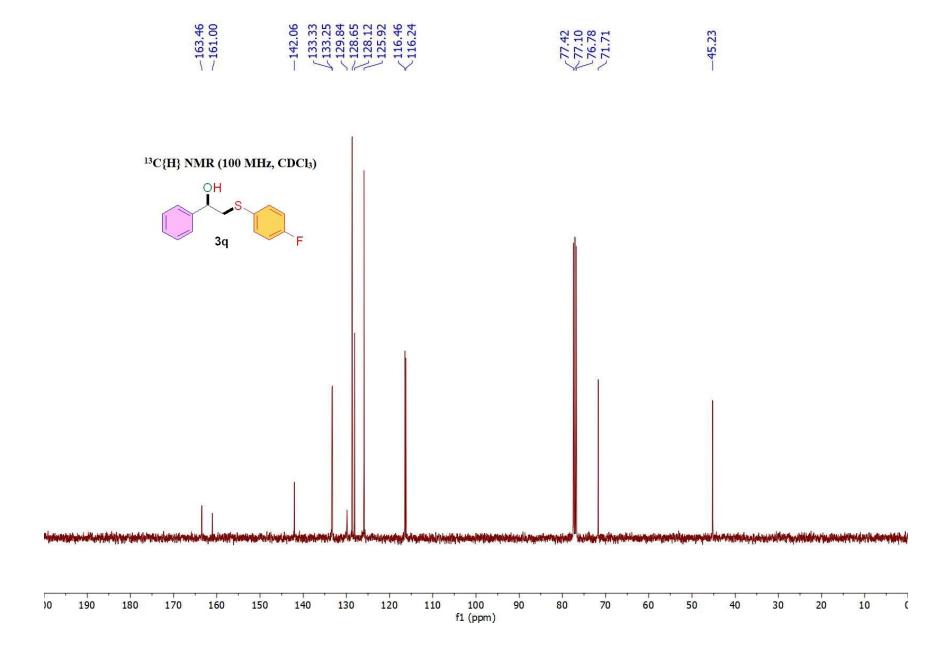


-44.14

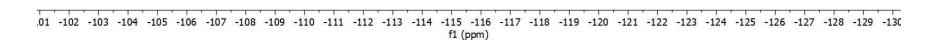


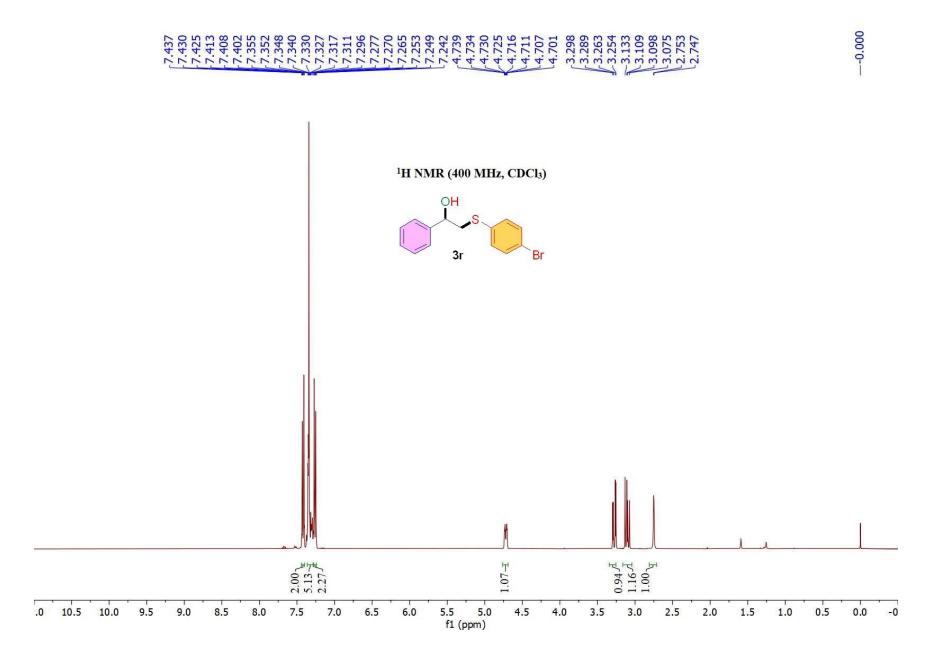






¹⁹F NMR (376 MHz, CDCl₃)

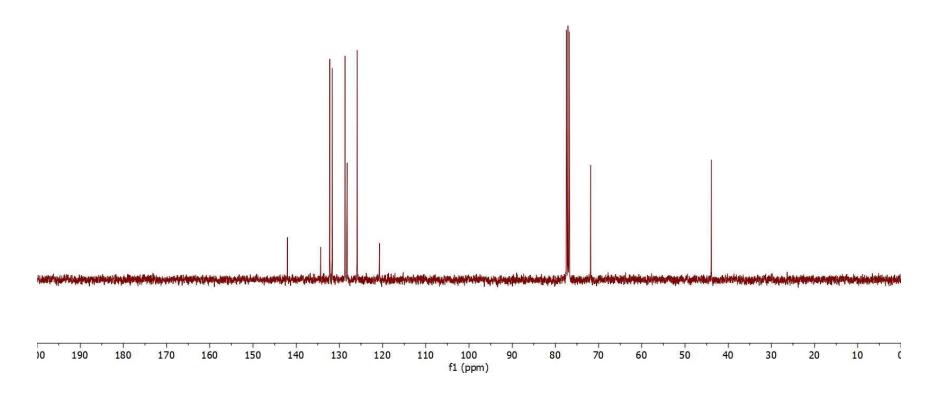








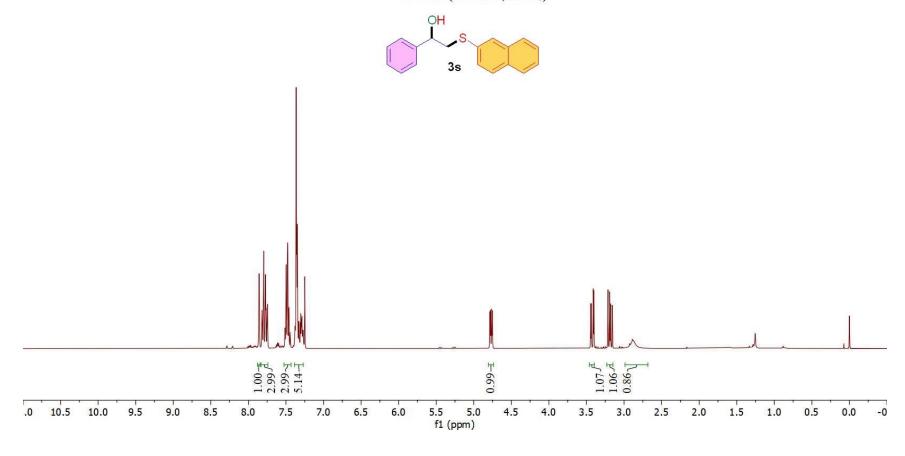
-43.92



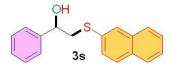


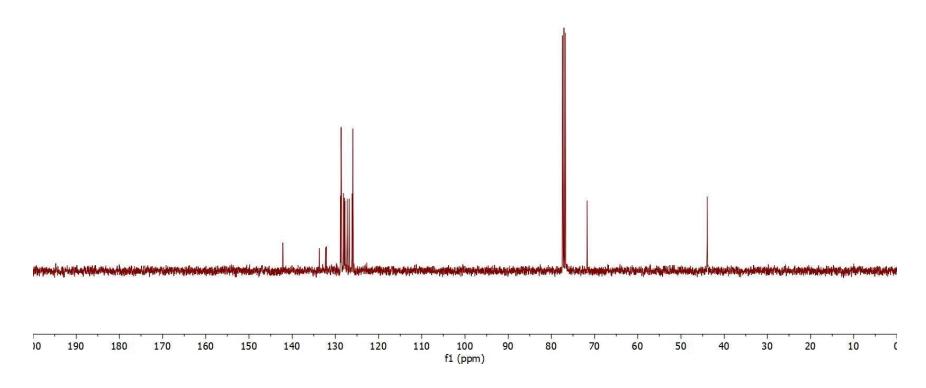


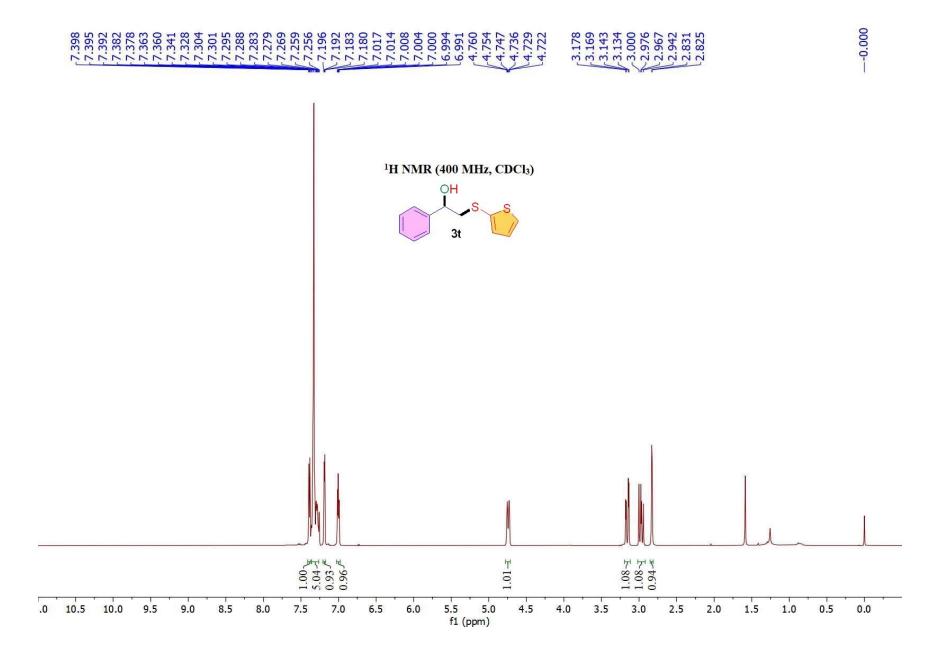


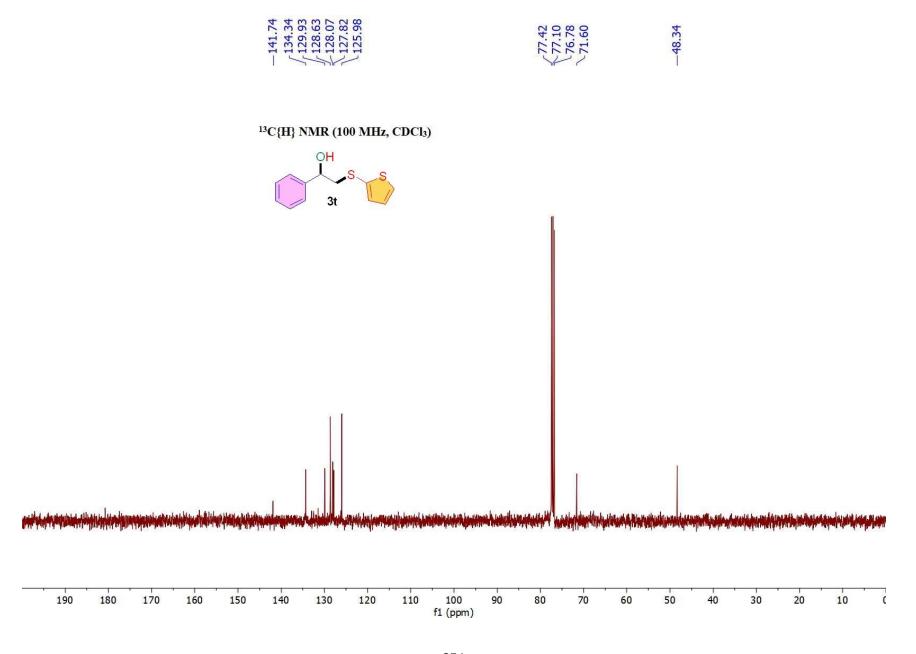


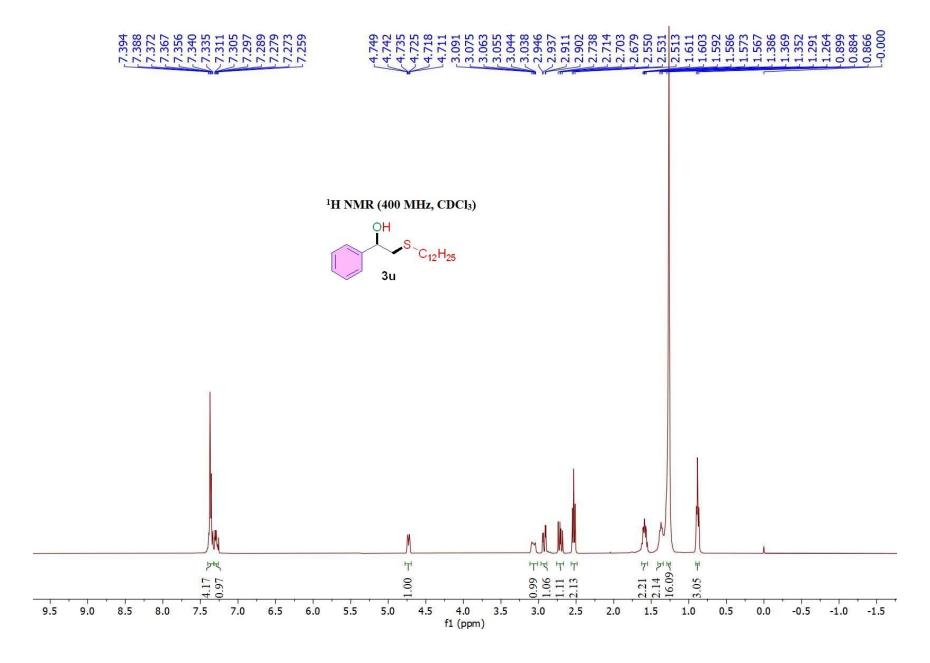






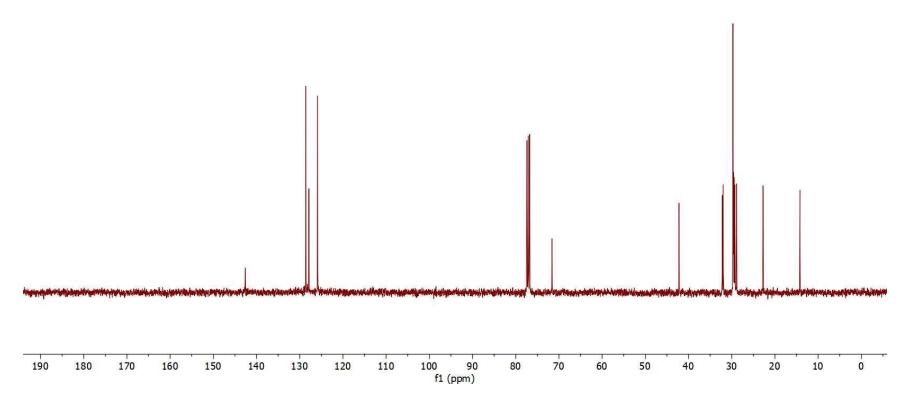


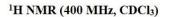


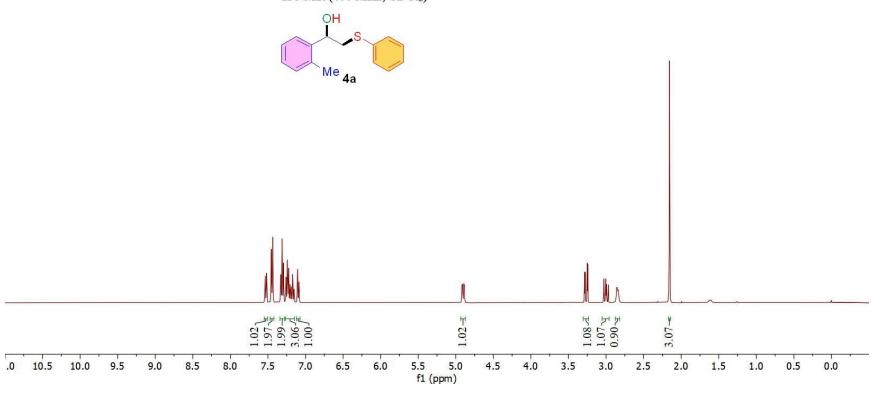




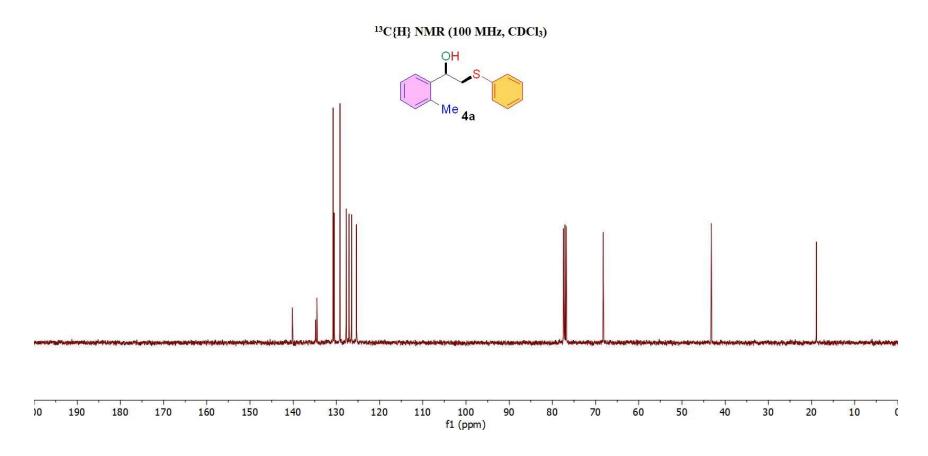


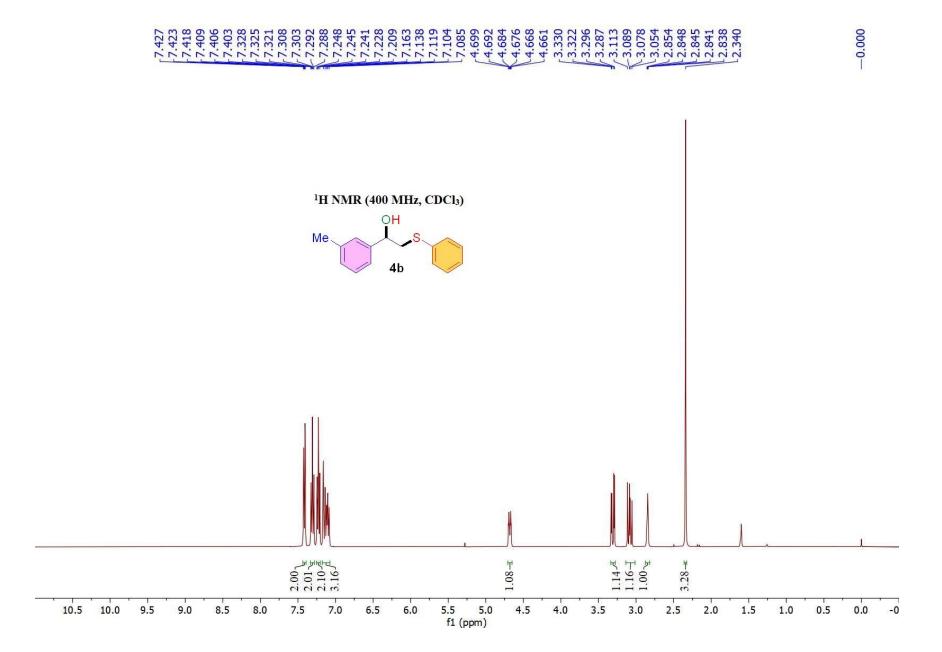




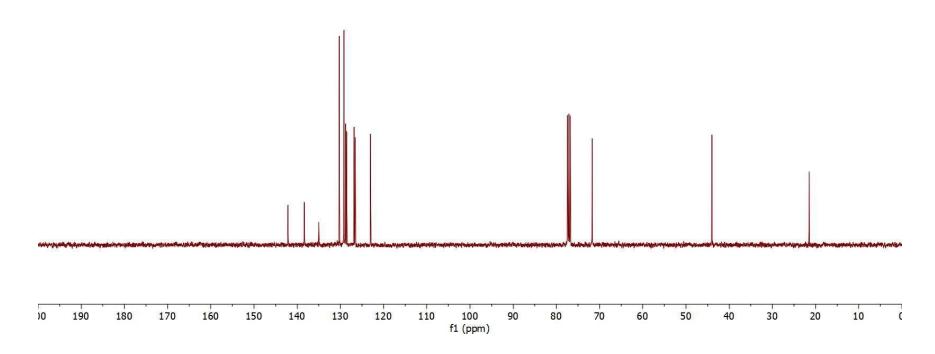




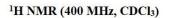


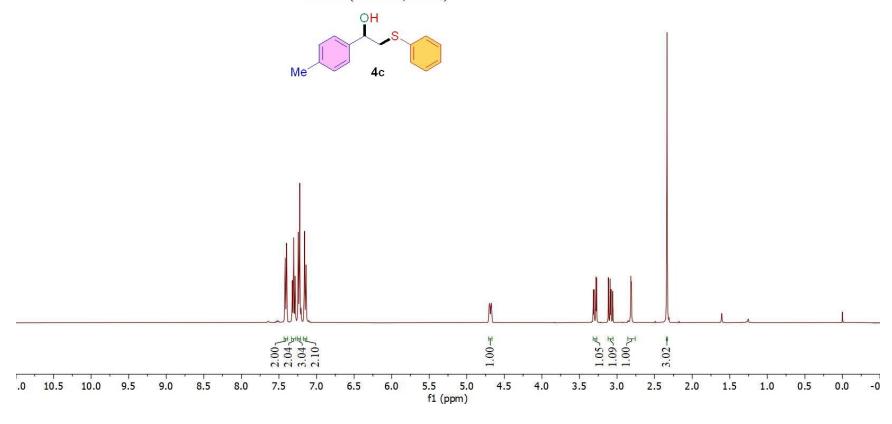




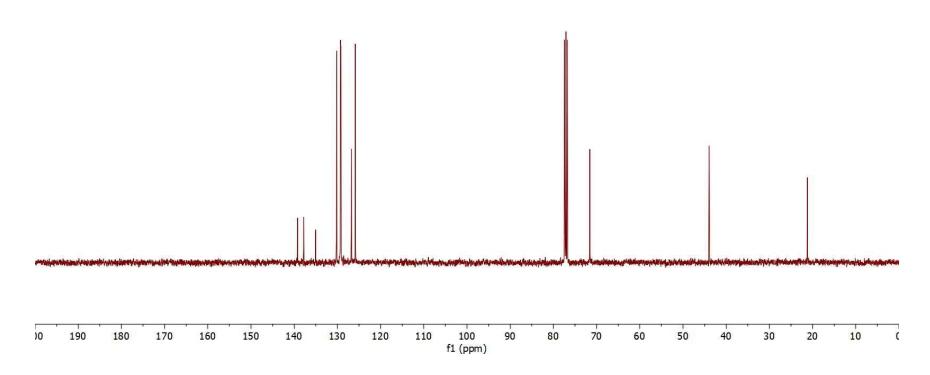


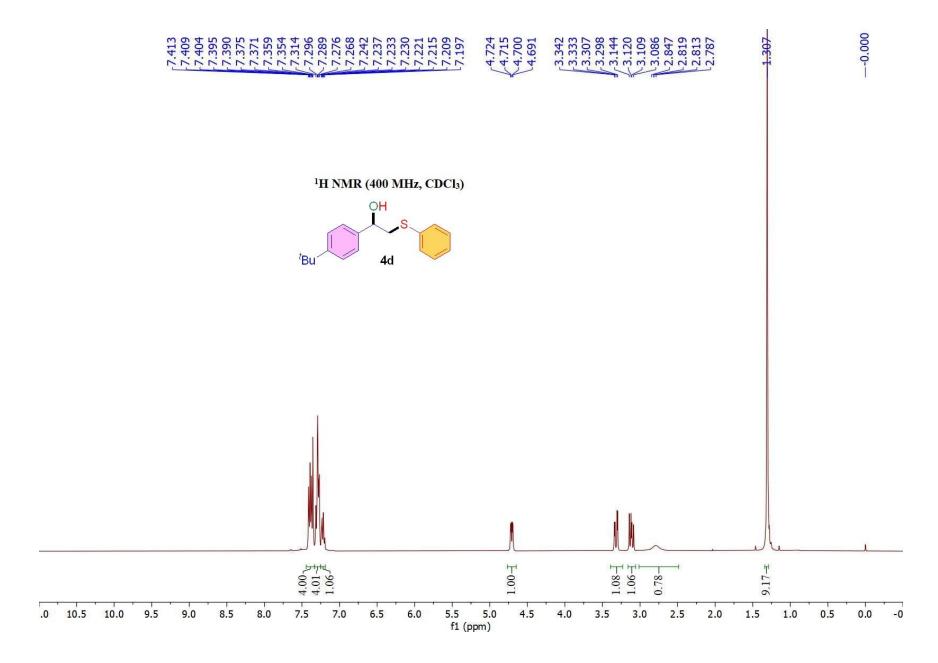




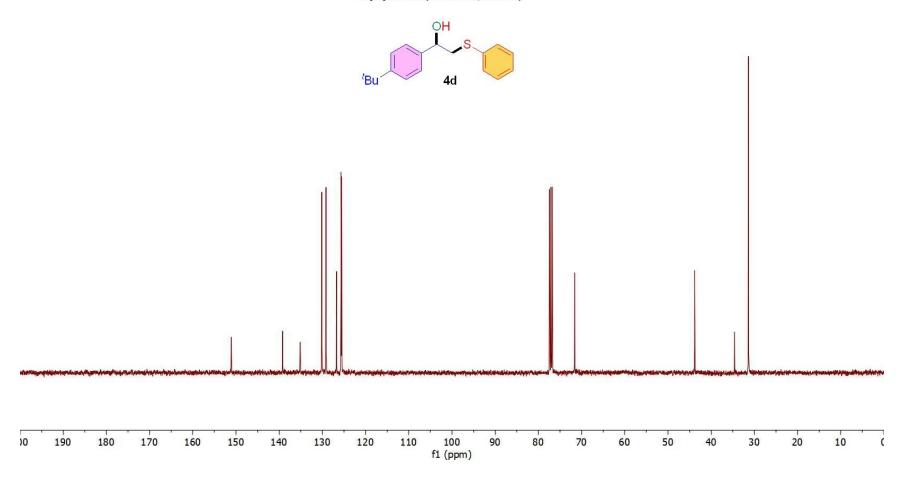




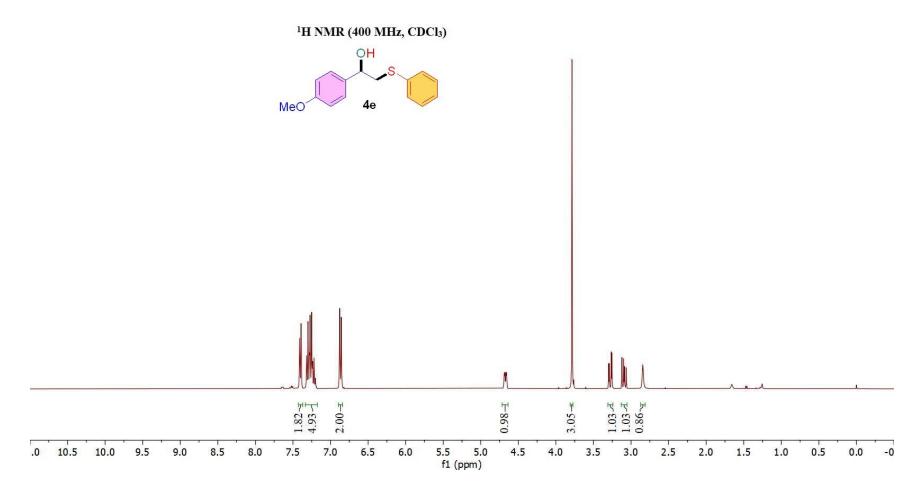




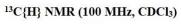


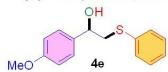


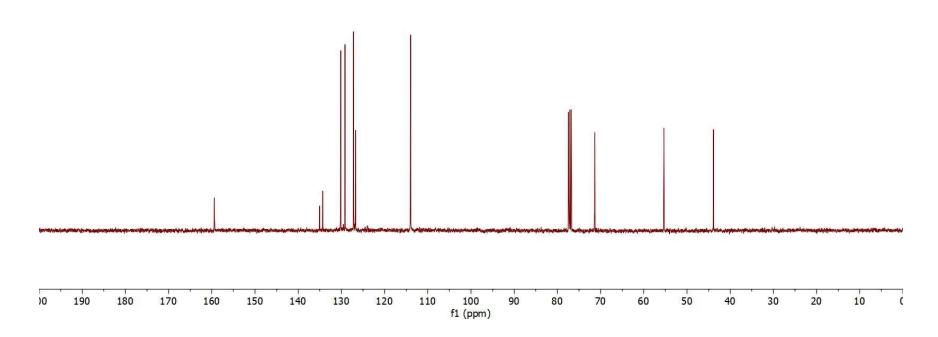




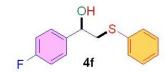


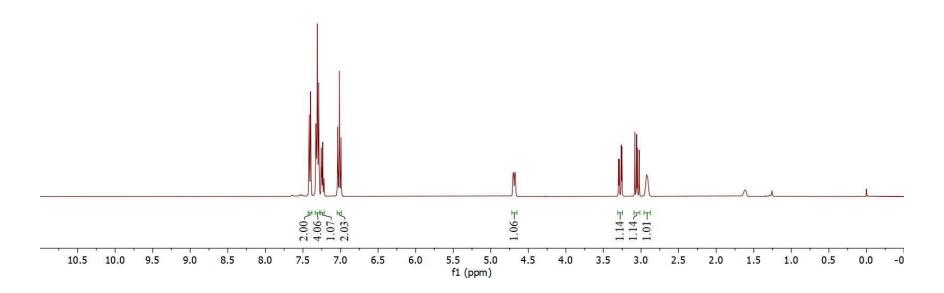


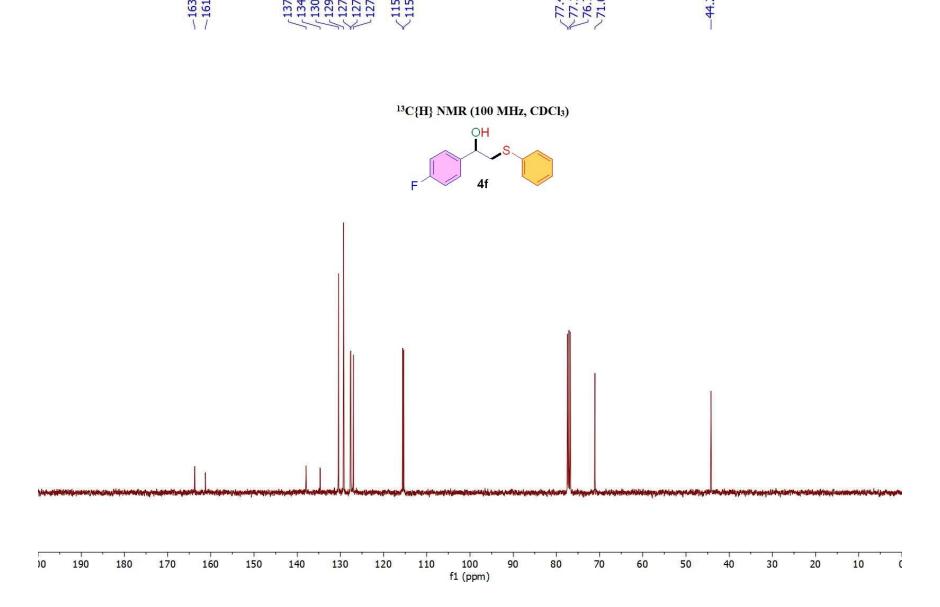






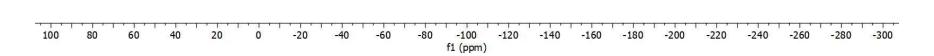


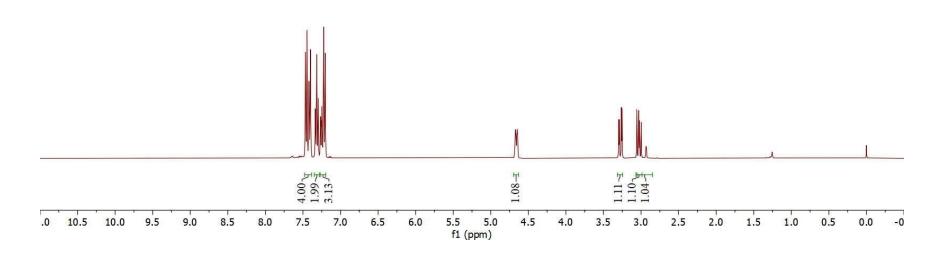


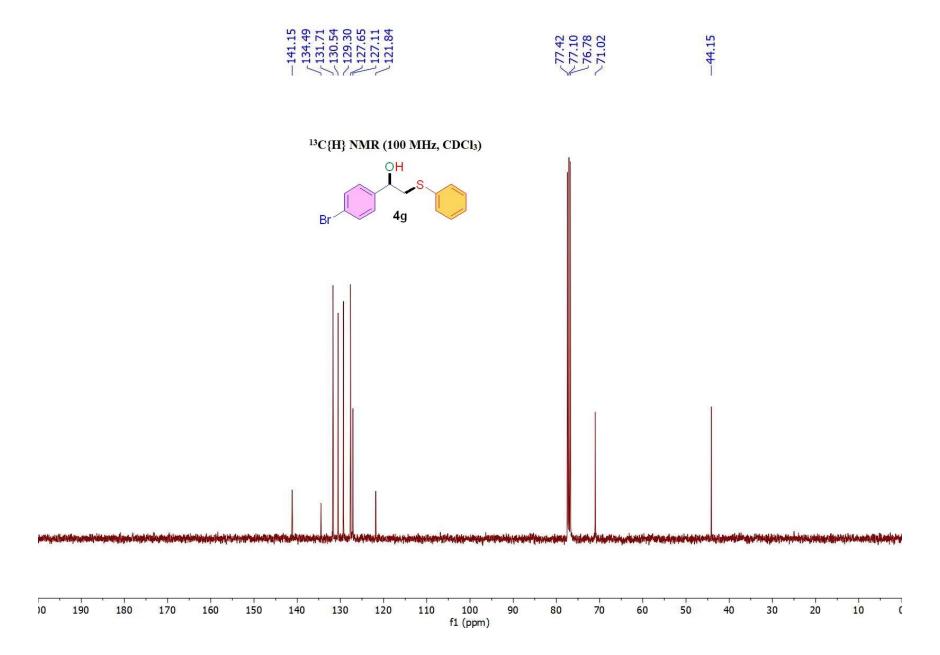




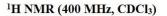
¹⁹F NMR (376 MHz, CDCl₃)

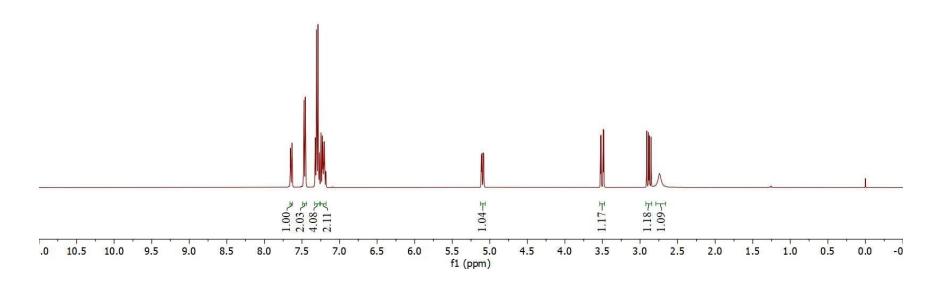


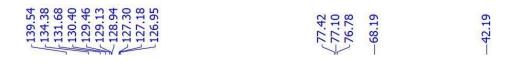




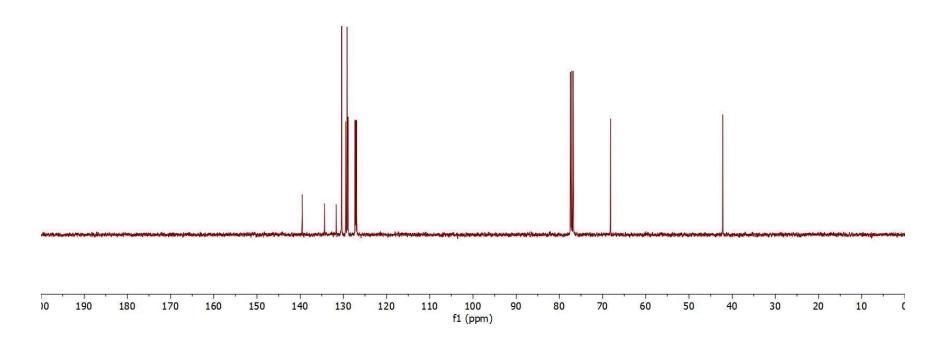


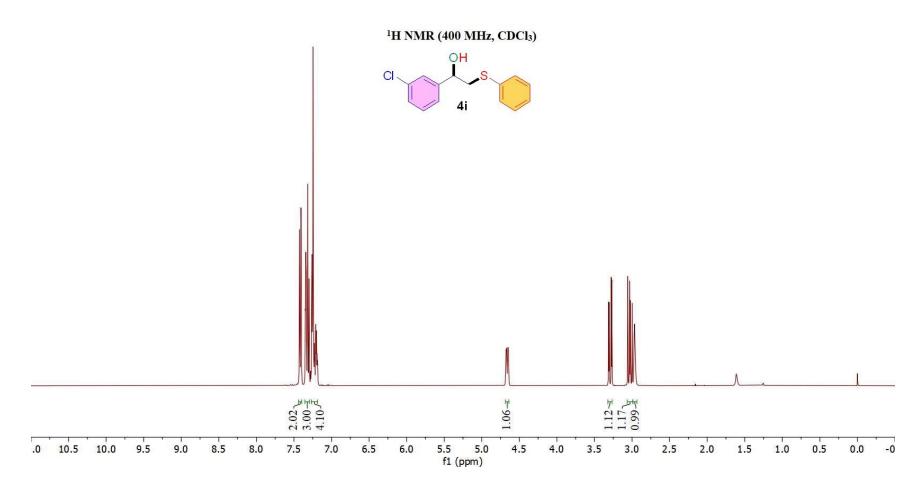








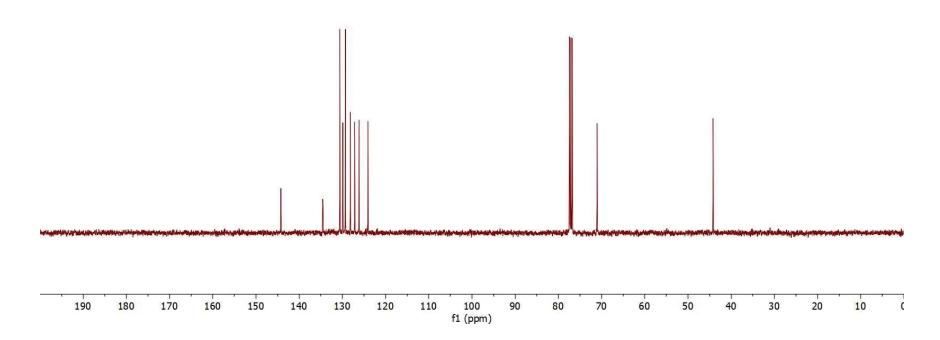






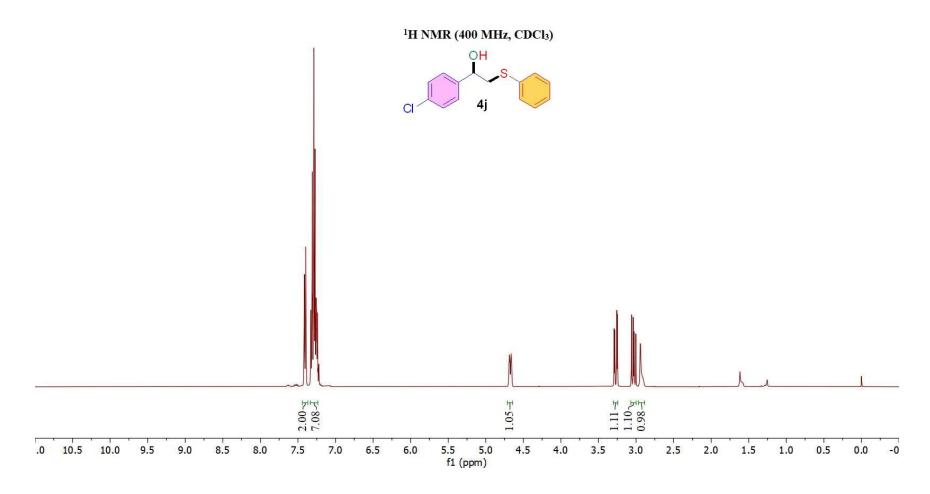


-44.20







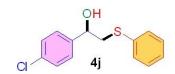


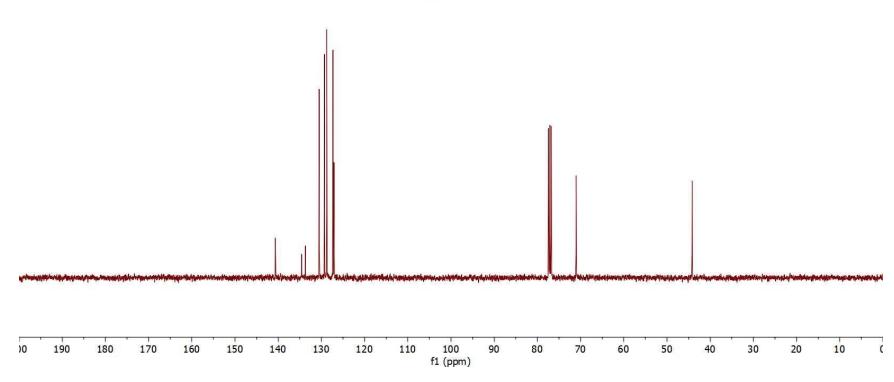


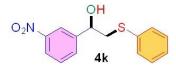


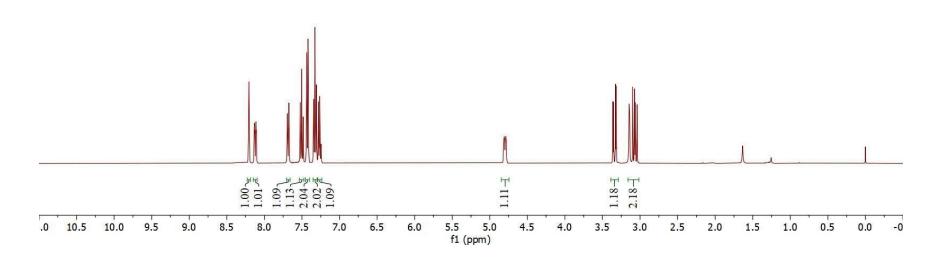








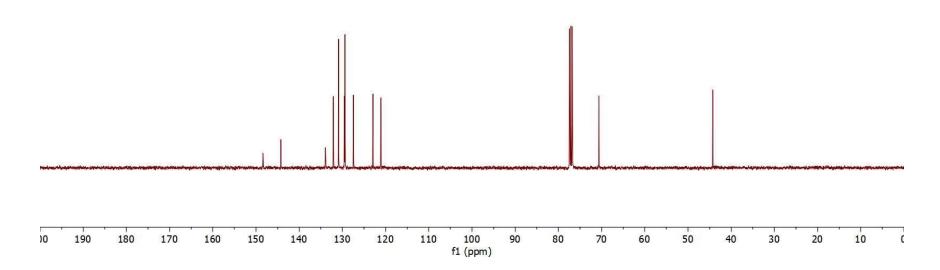








-44.26

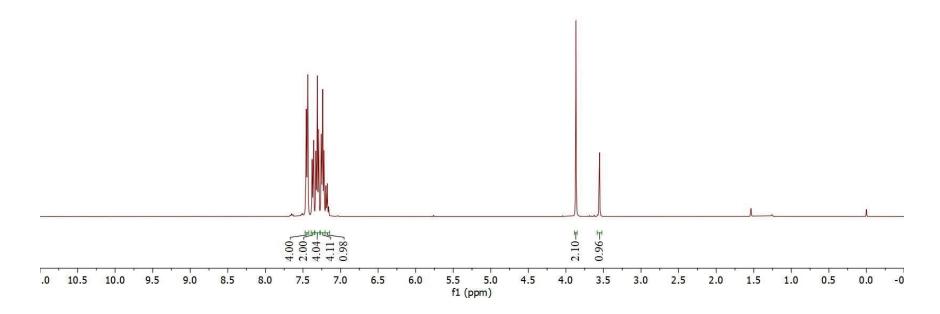










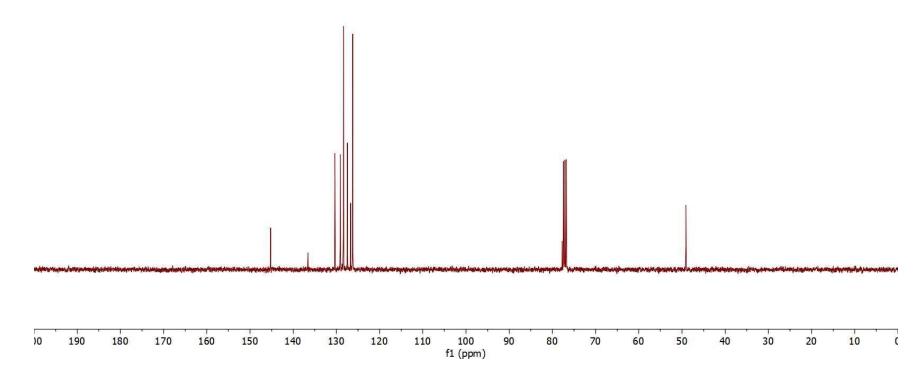






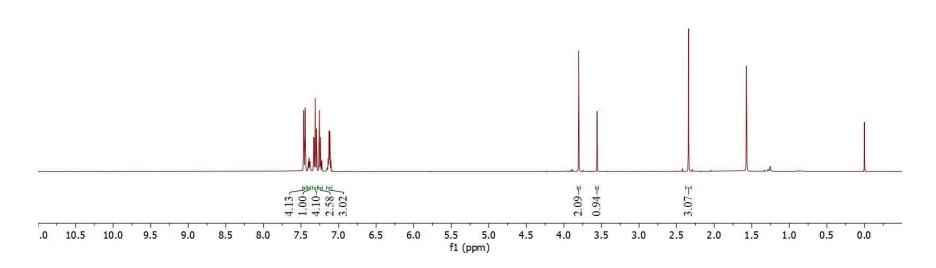
-49.09

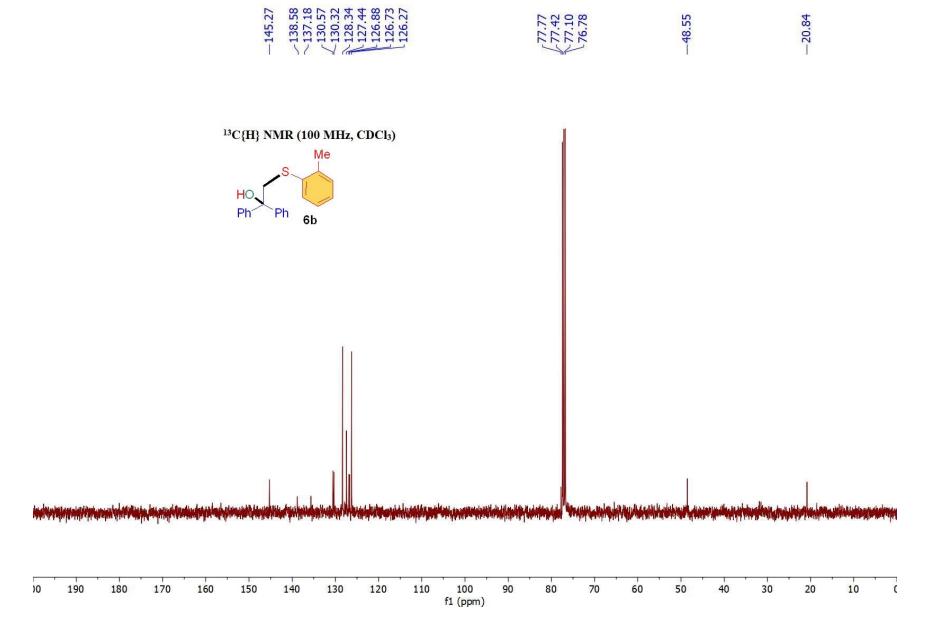




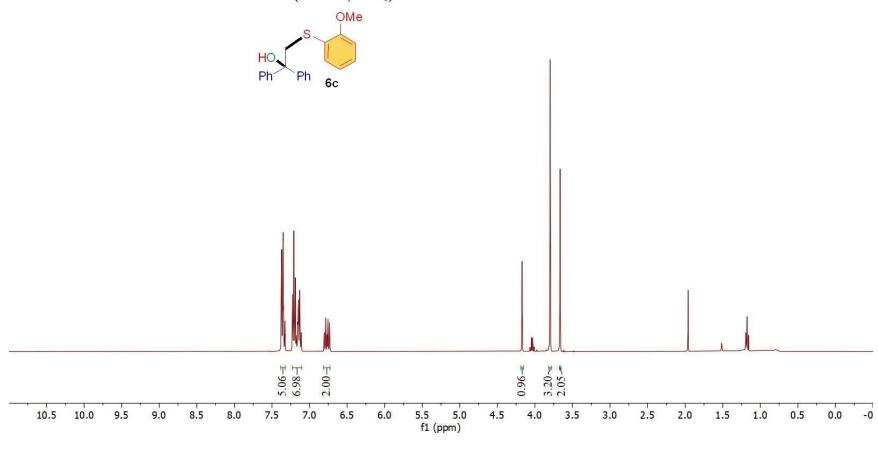




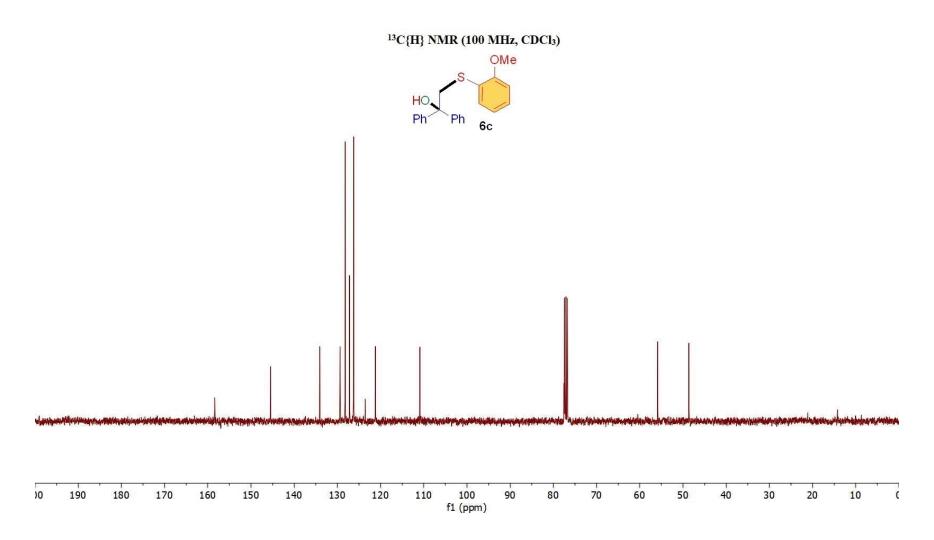


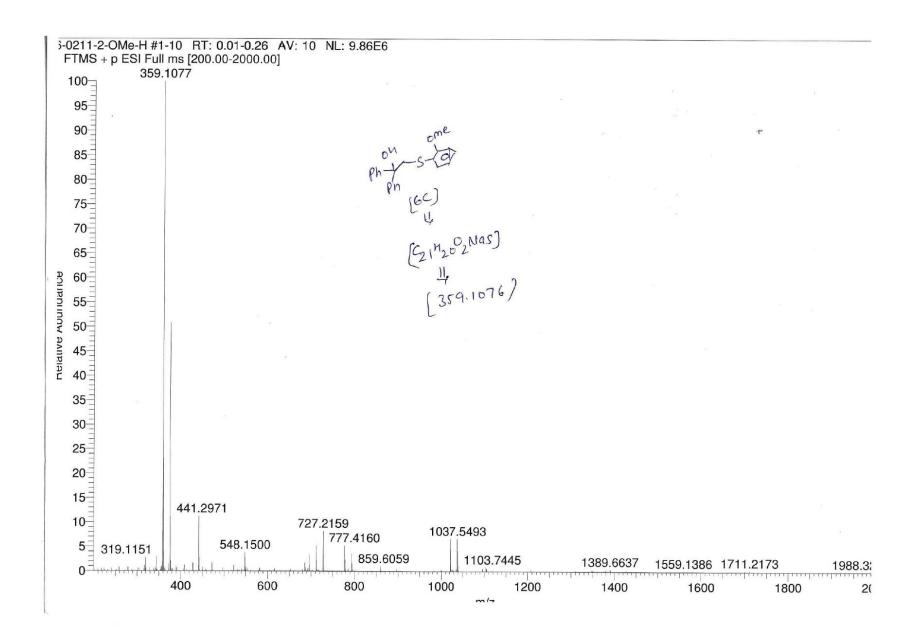






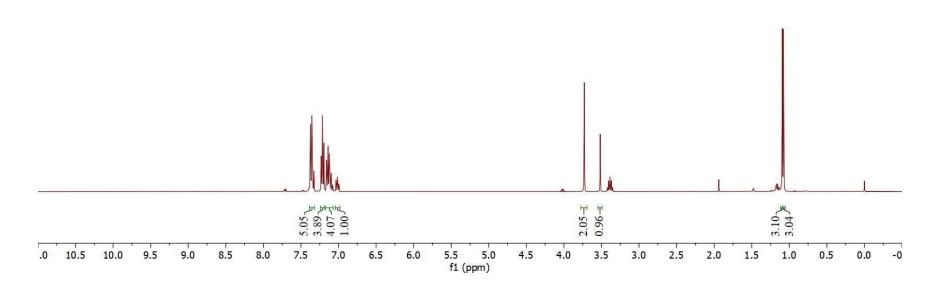


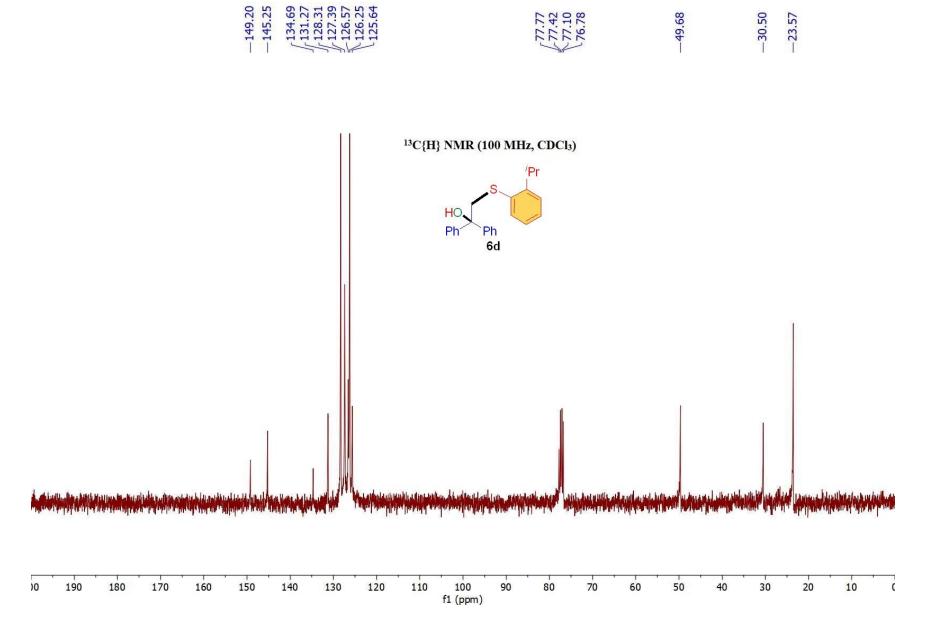


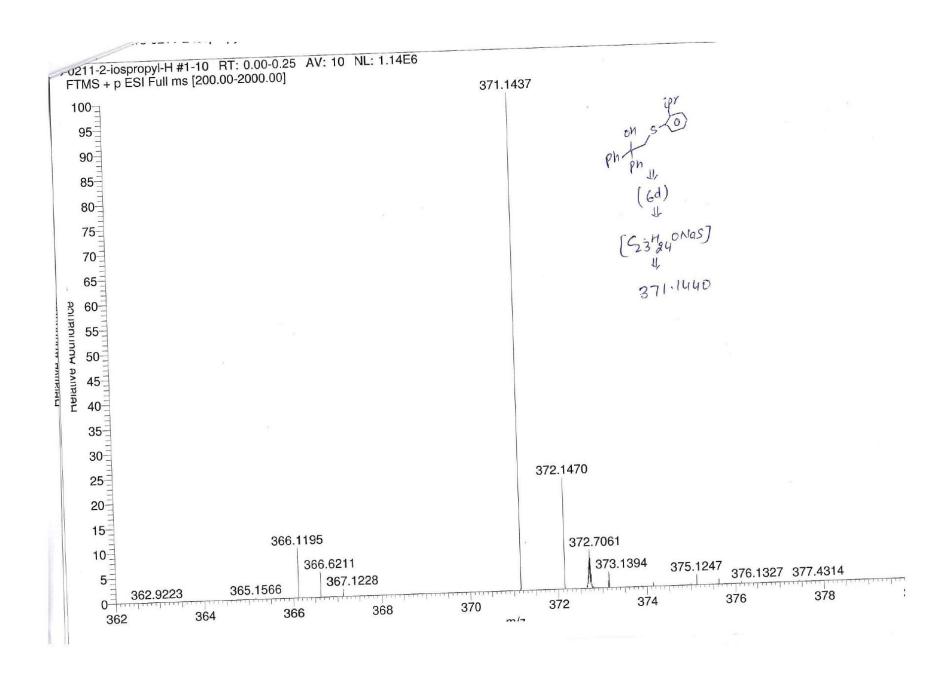






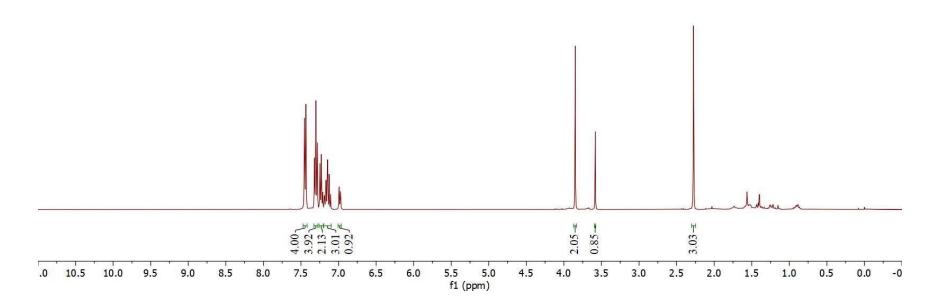




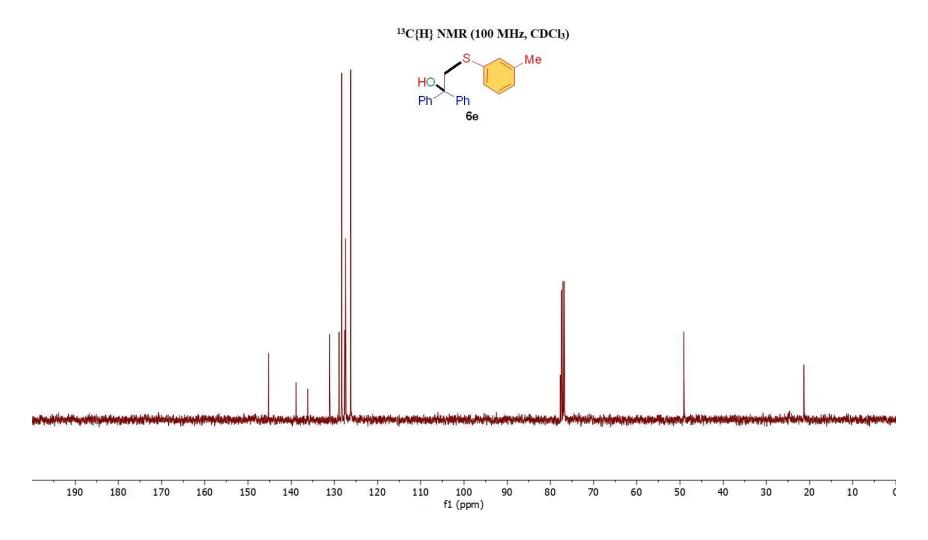








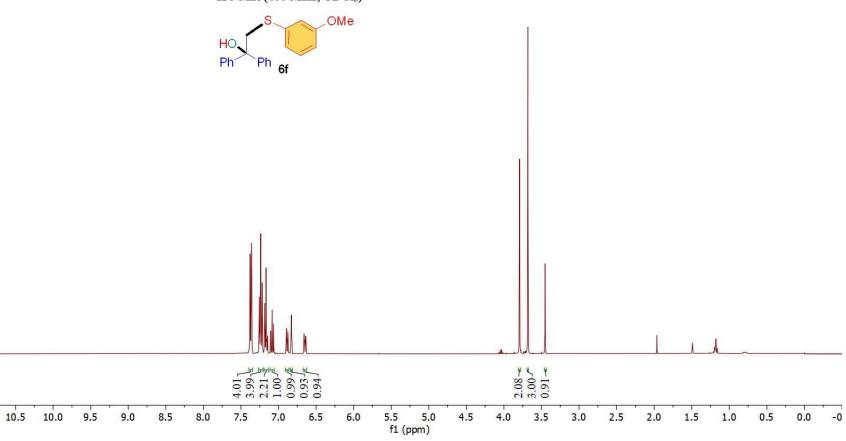




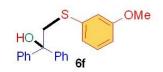


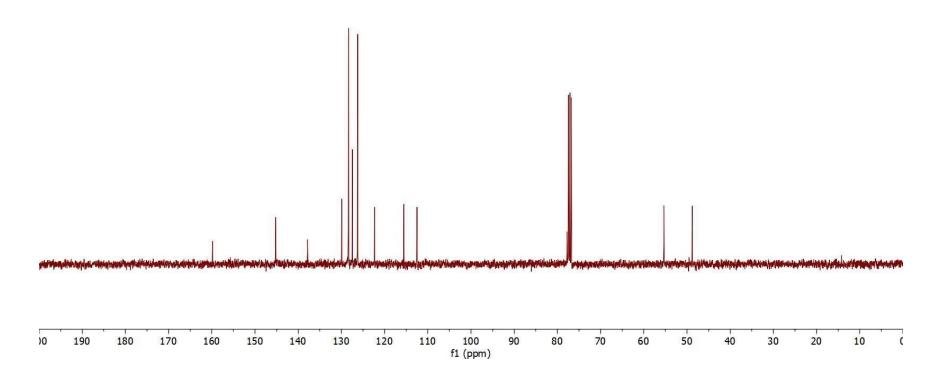


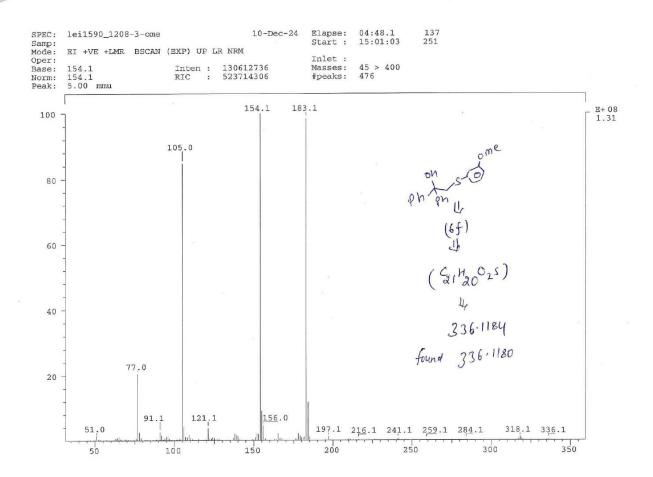




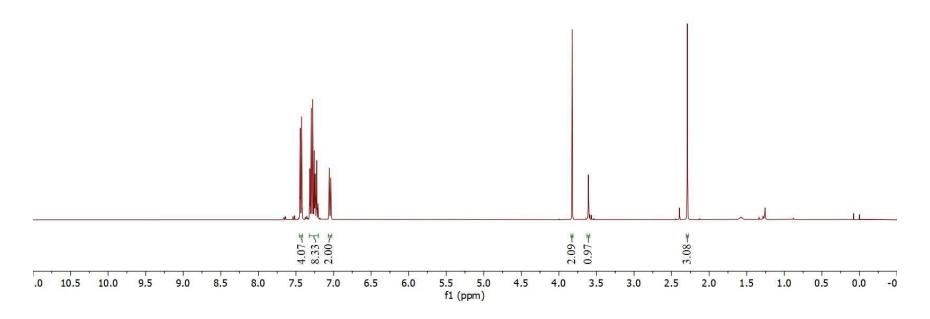






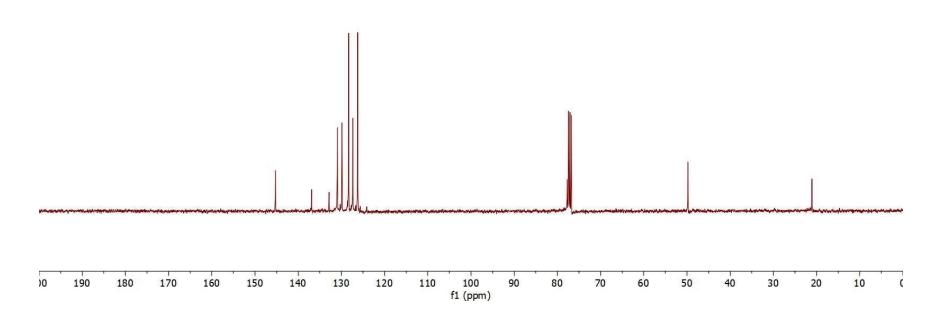






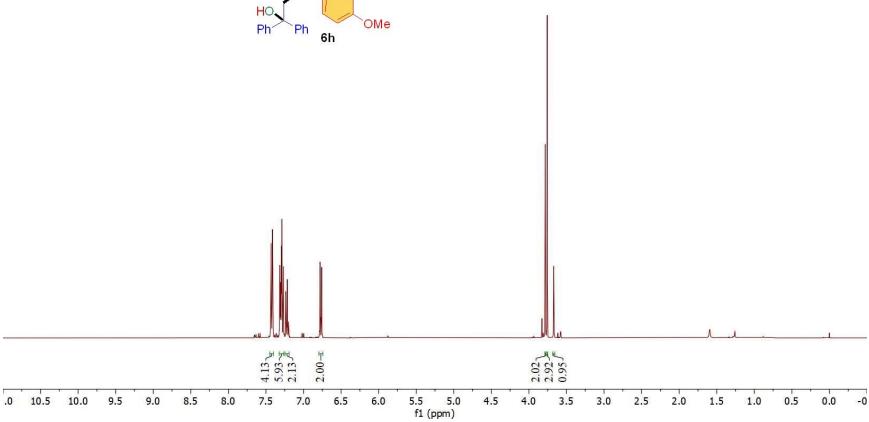


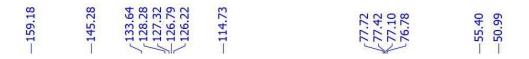


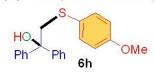


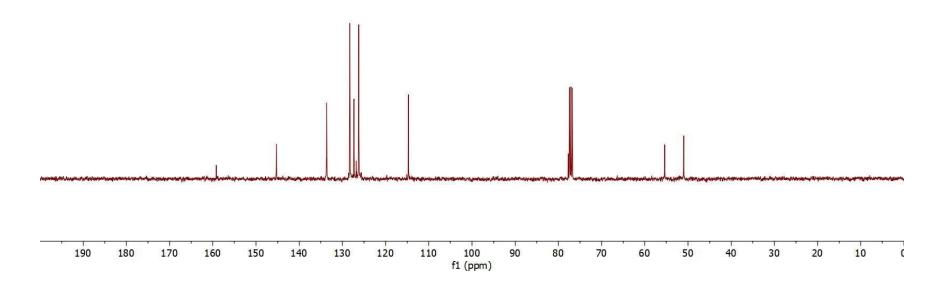






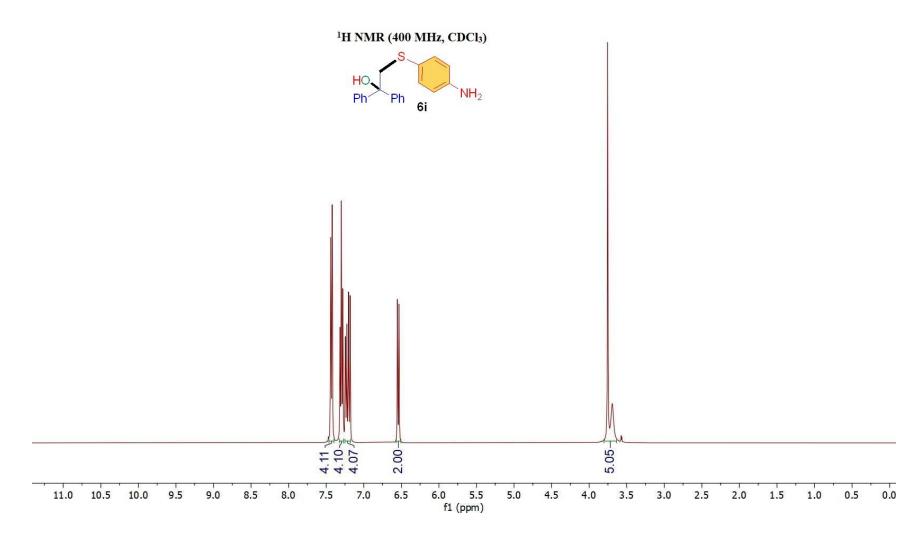




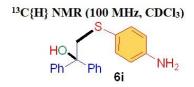


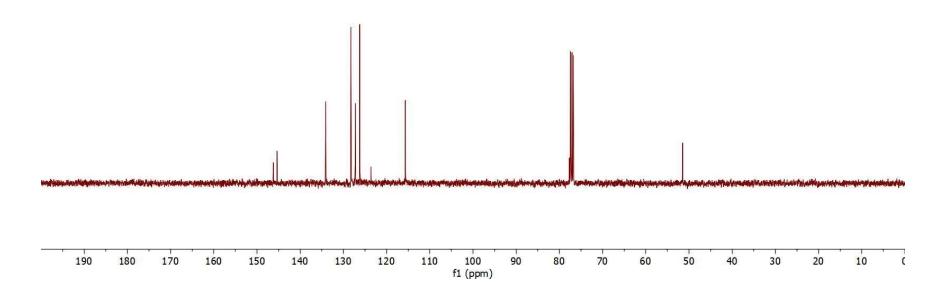


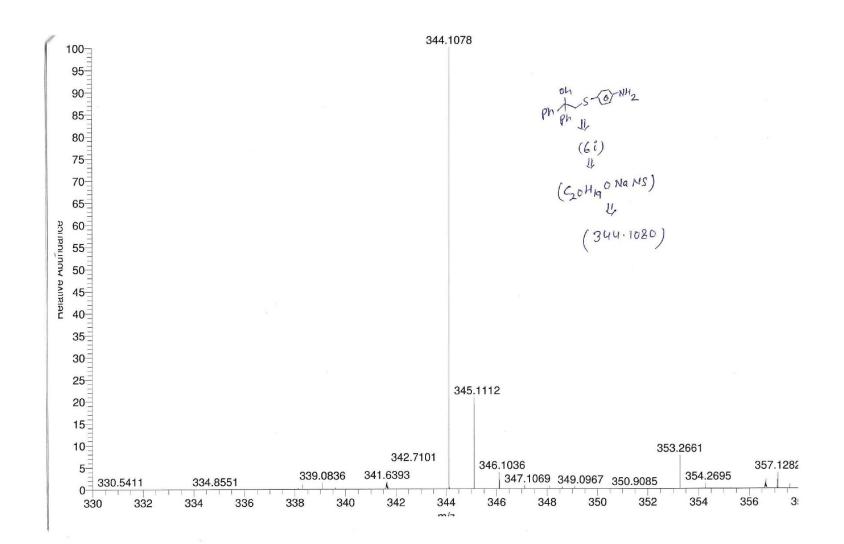




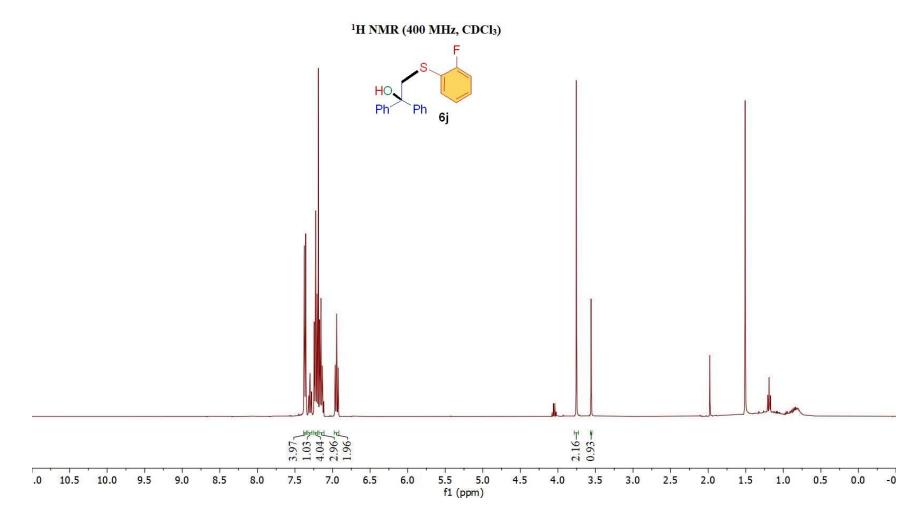




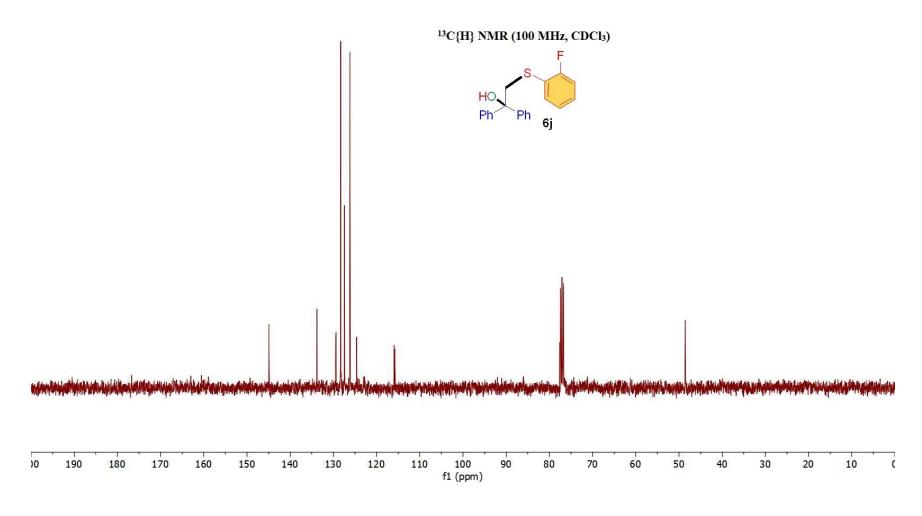


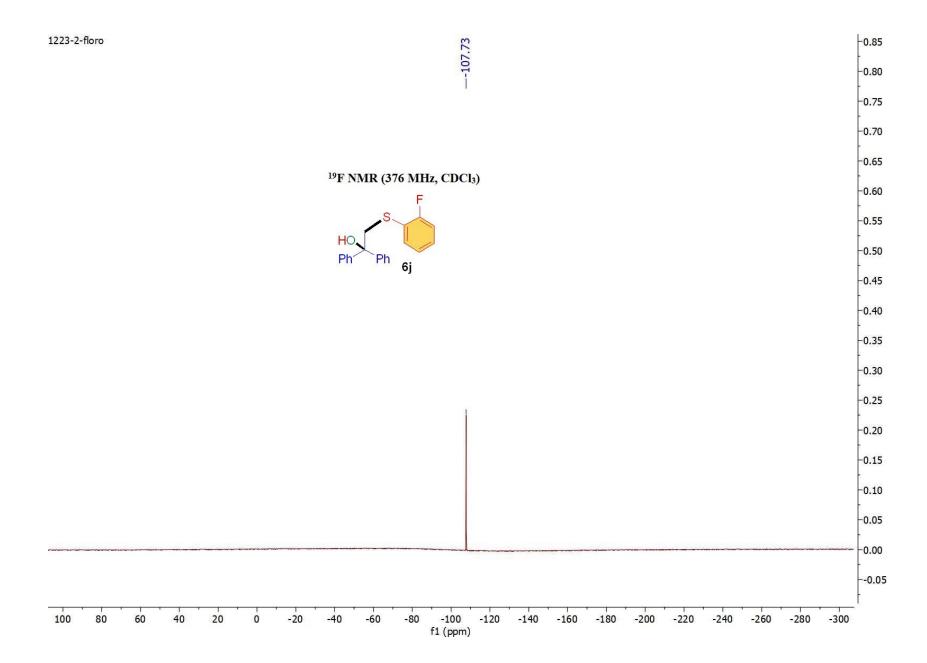


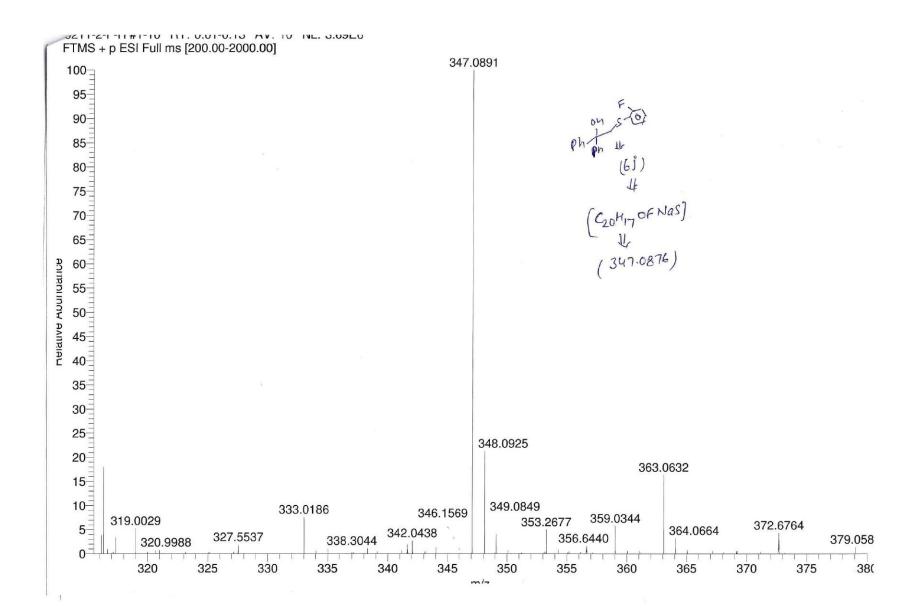




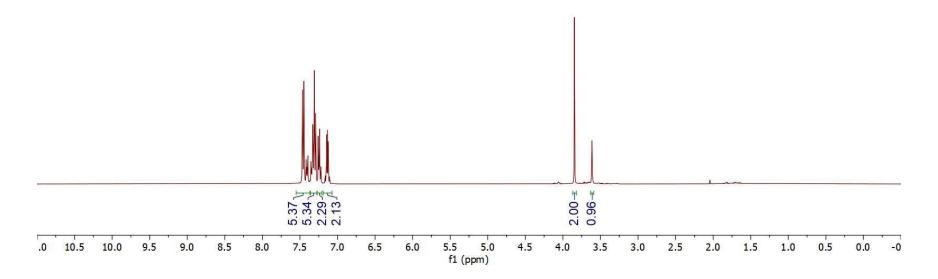










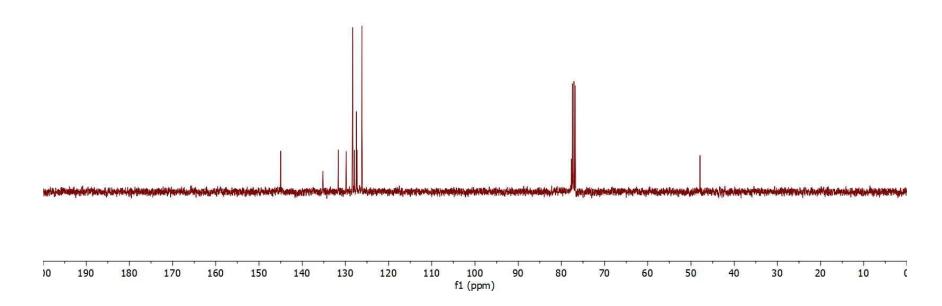


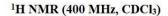


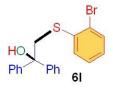


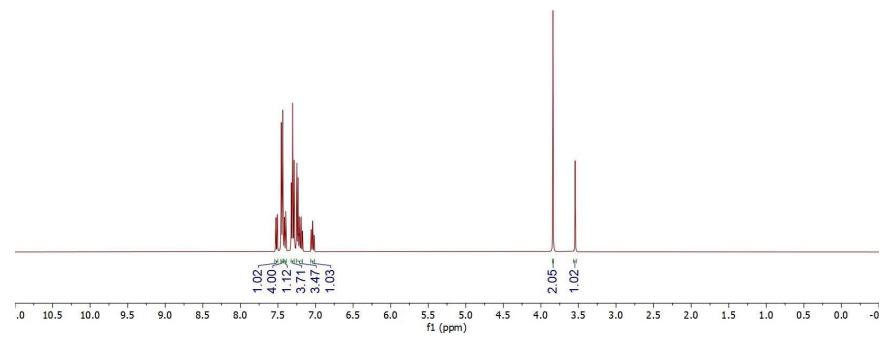
-47.92



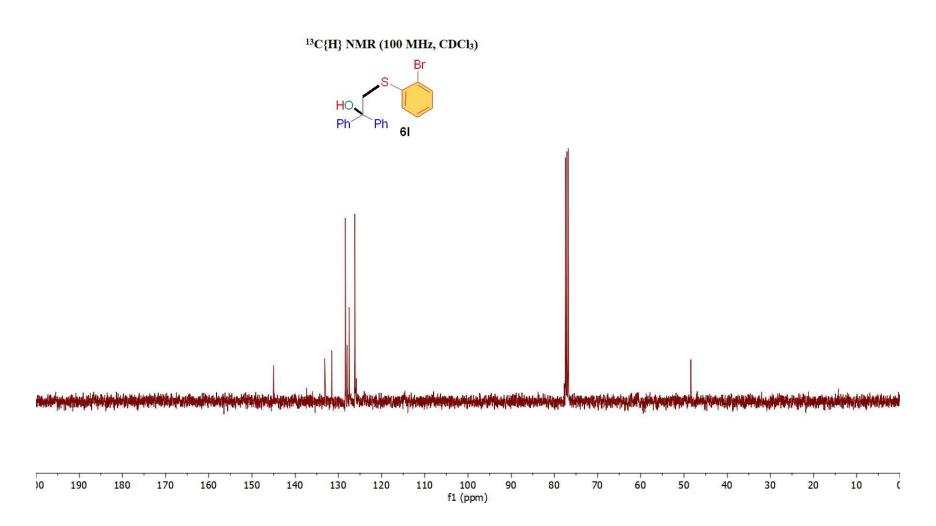


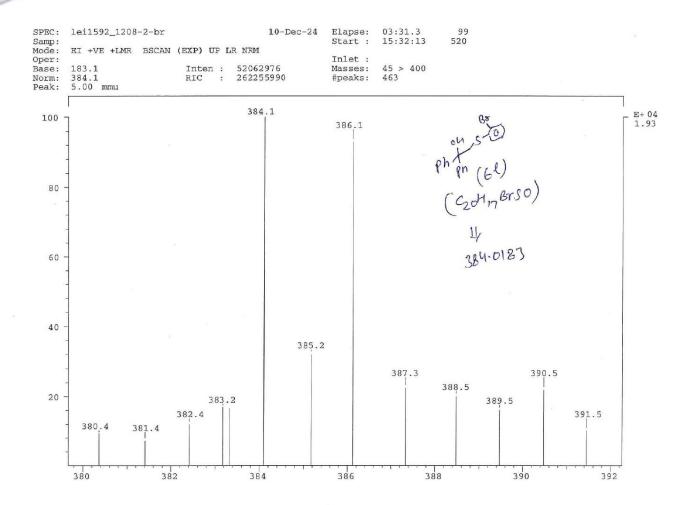


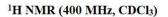


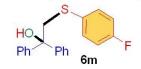


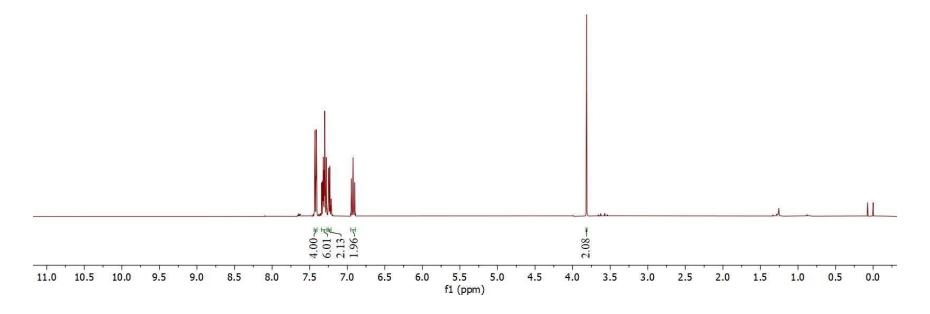






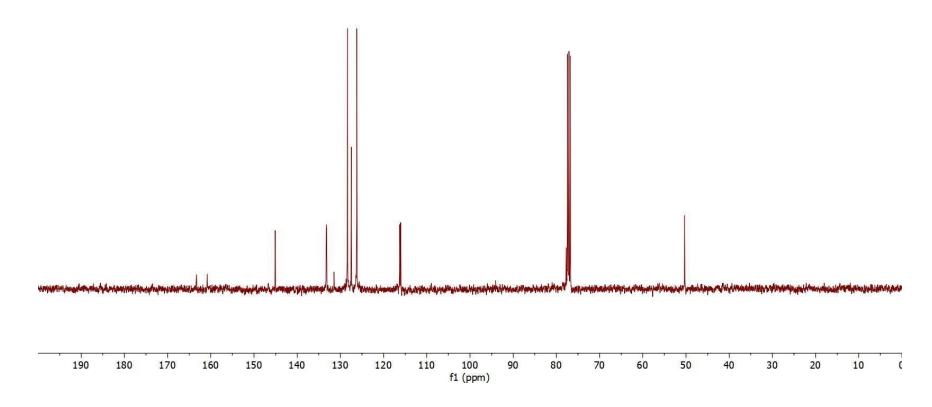


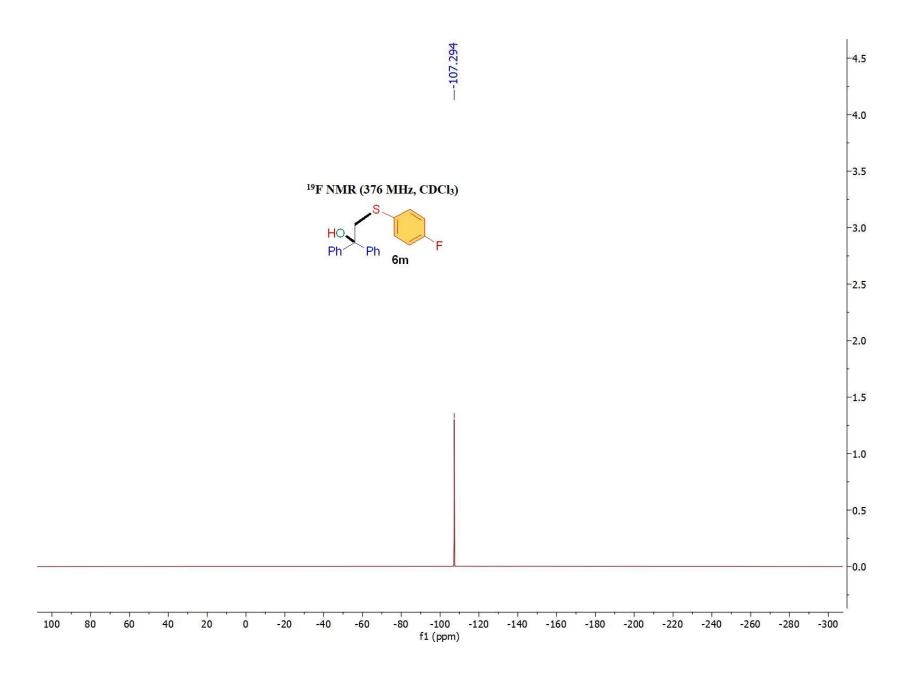




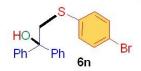


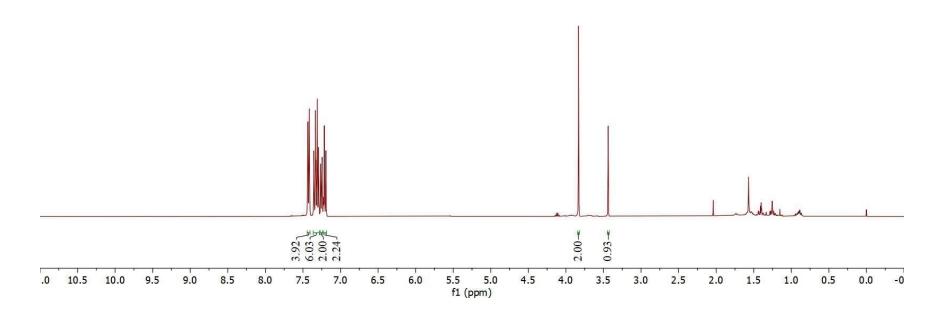


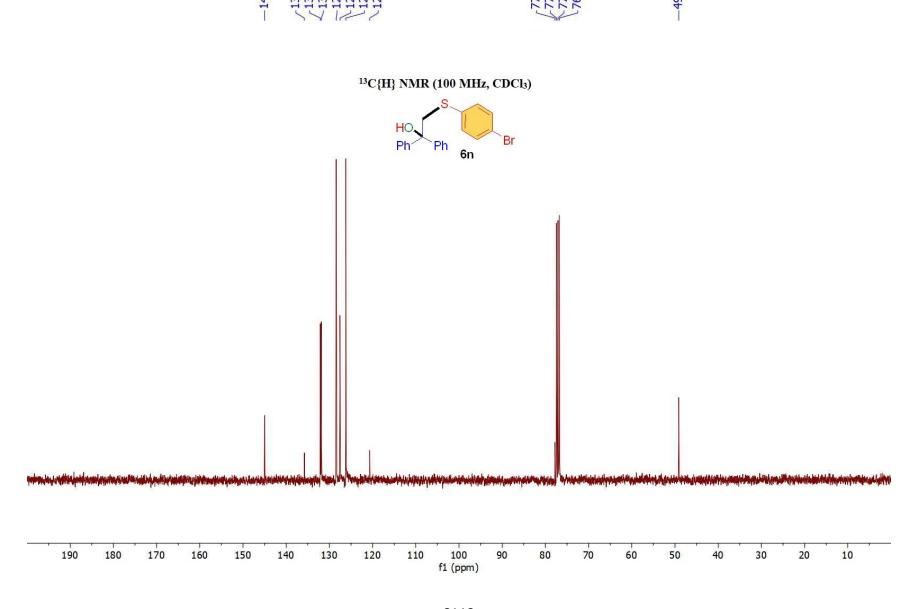


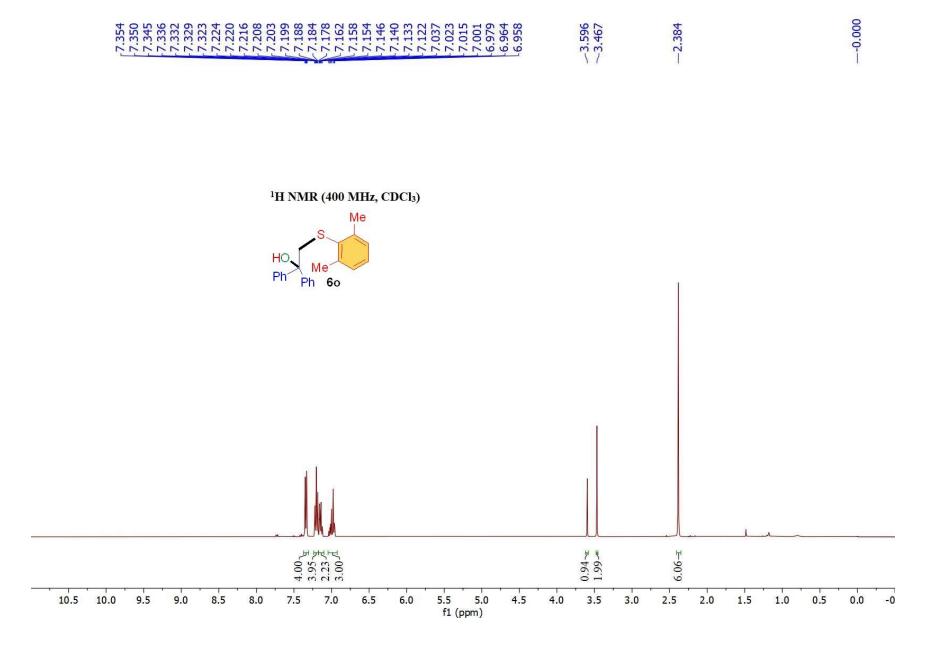


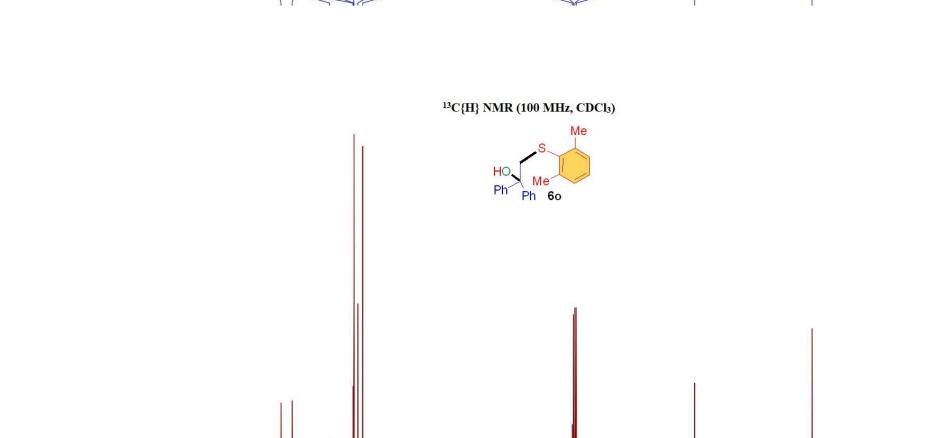




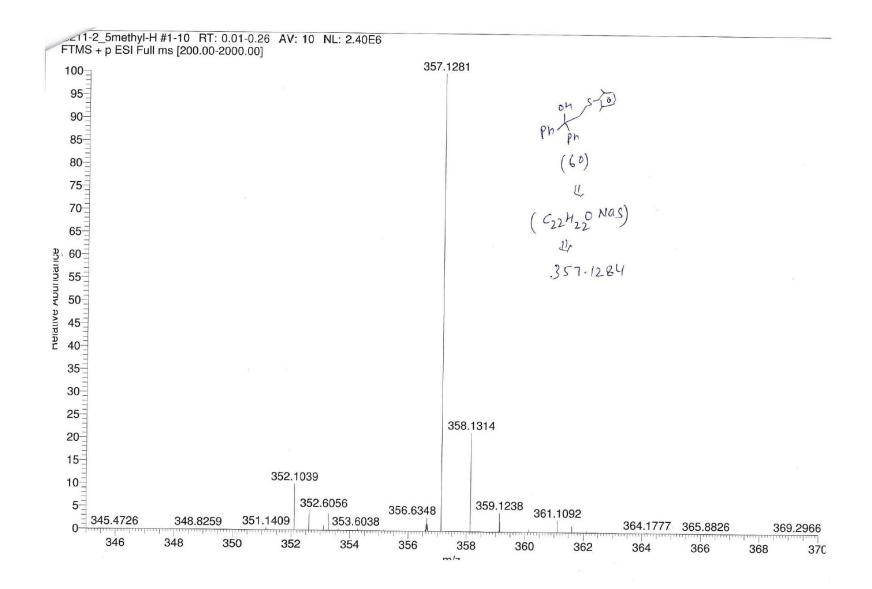






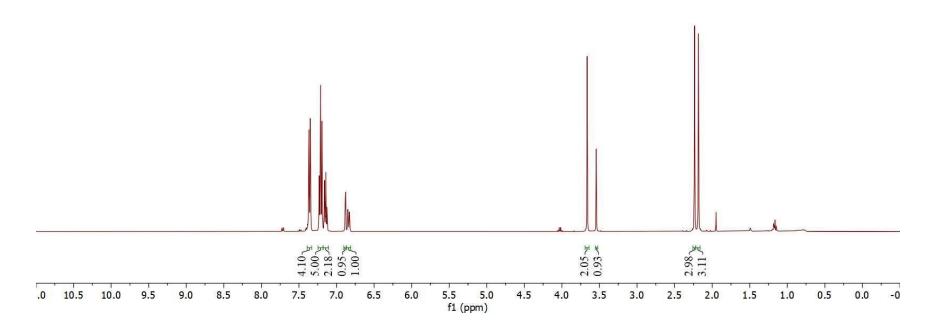


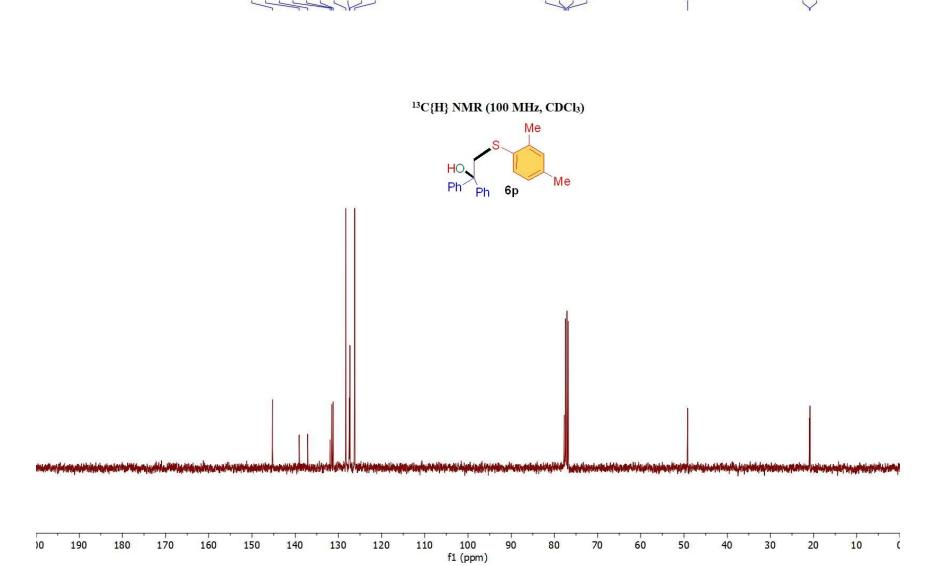
f1 (ppm)

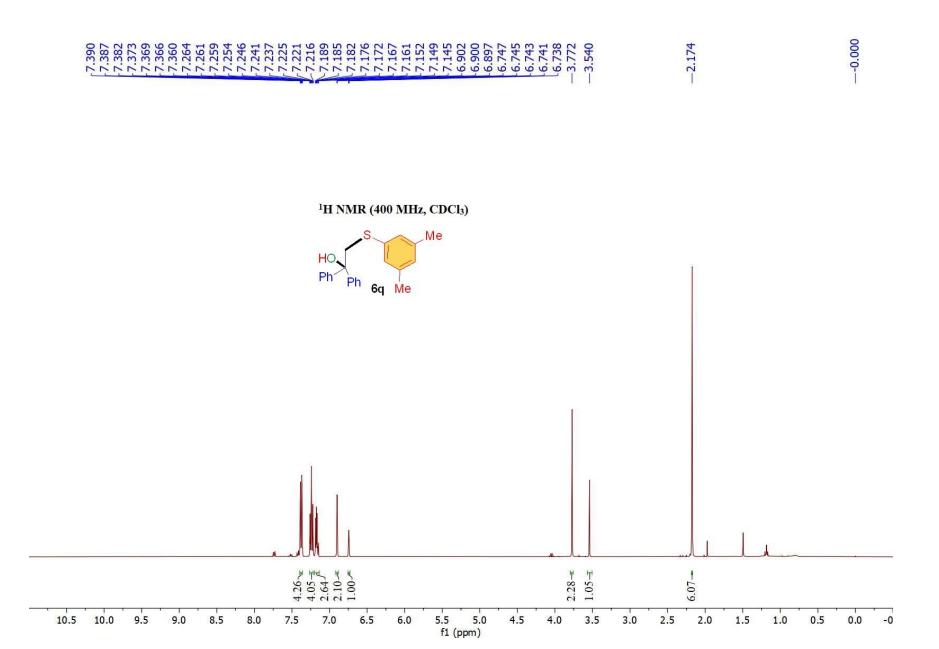




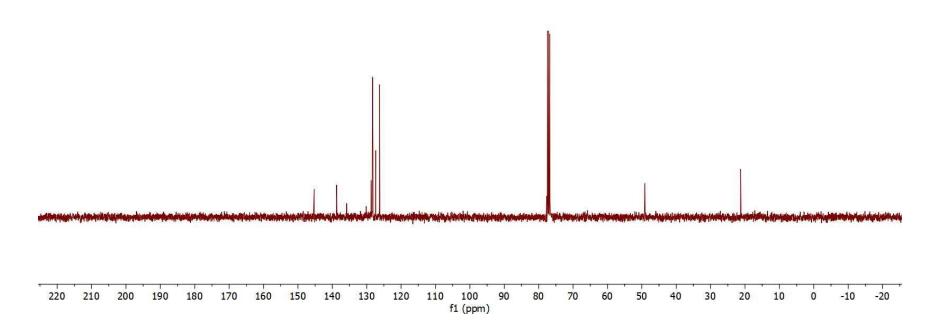






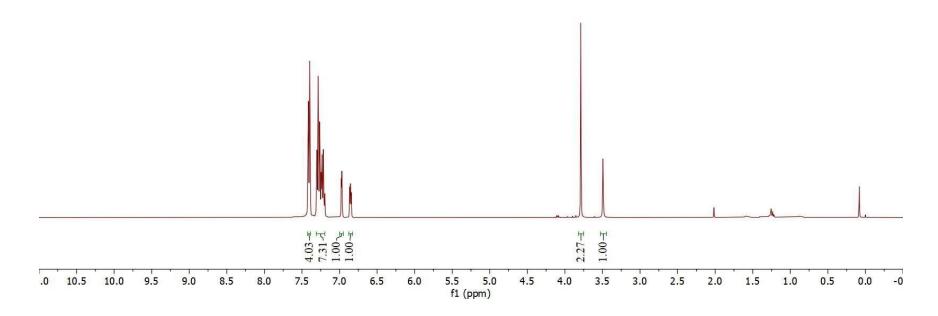




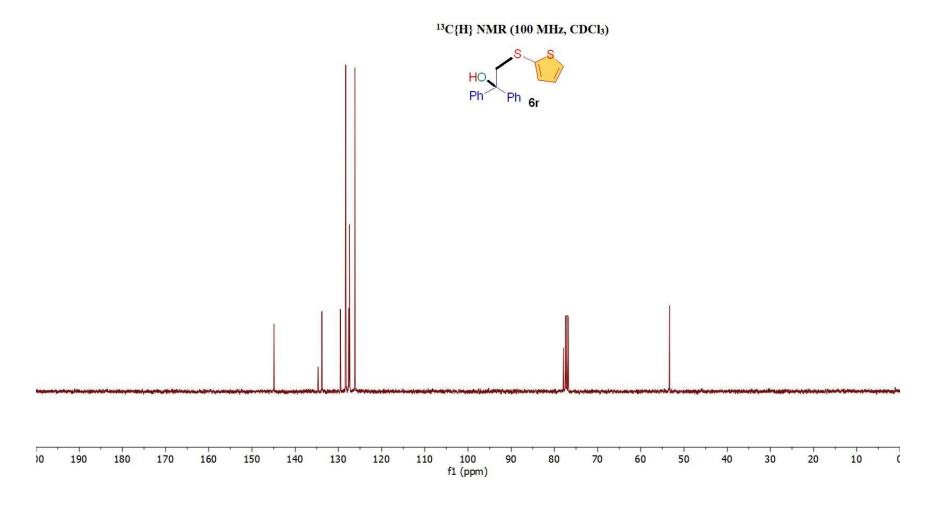


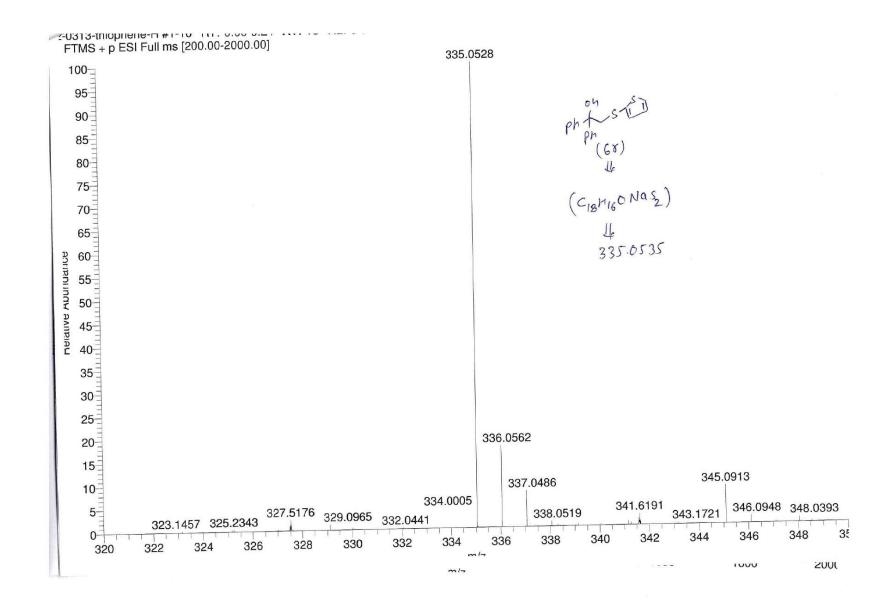




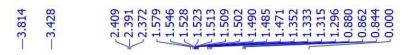






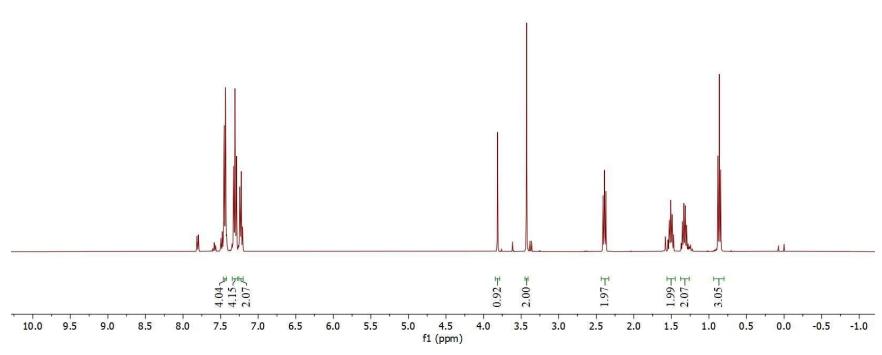




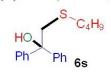


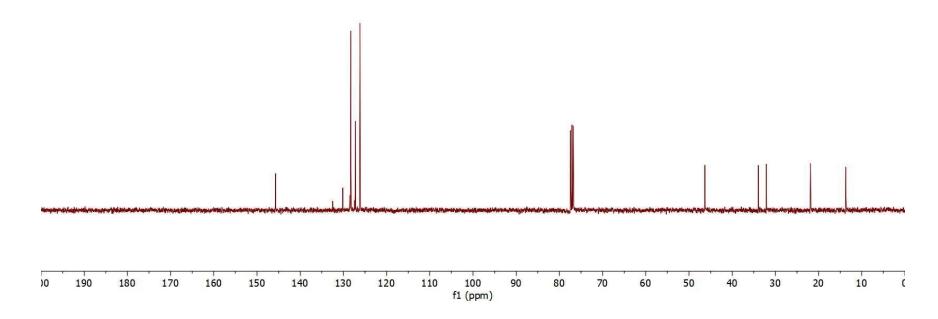




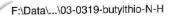


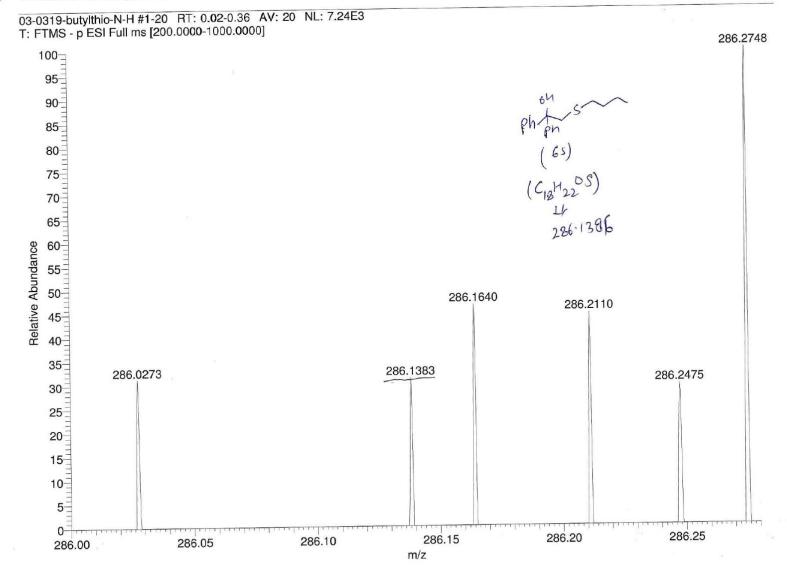






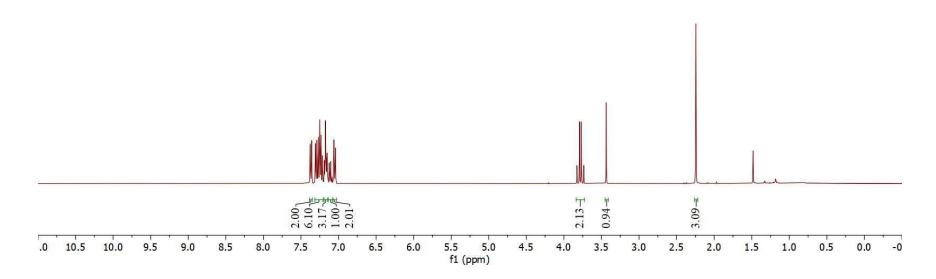


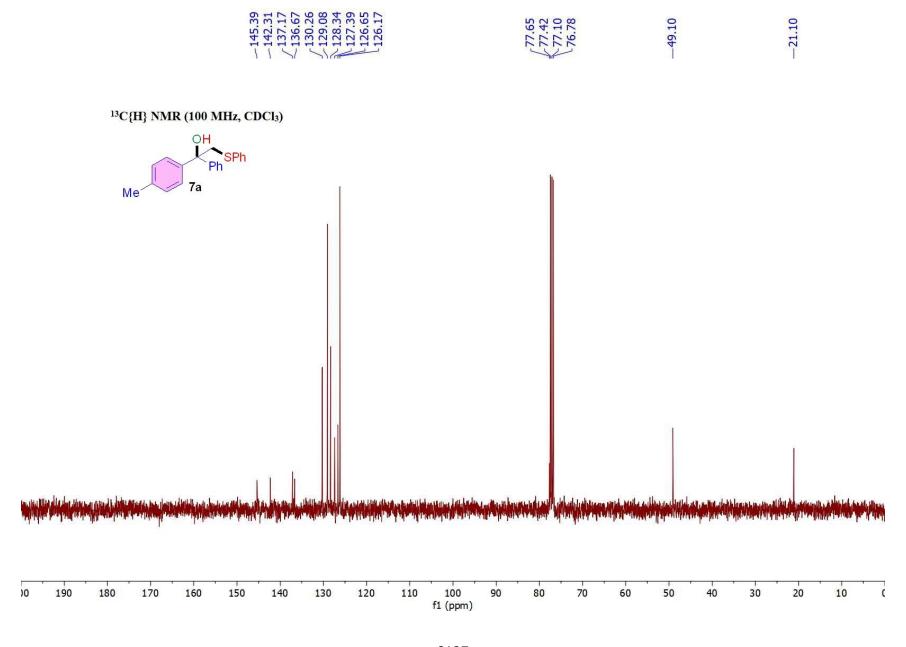


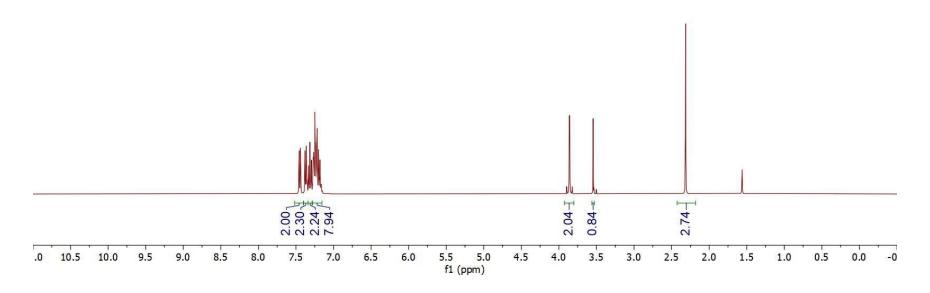


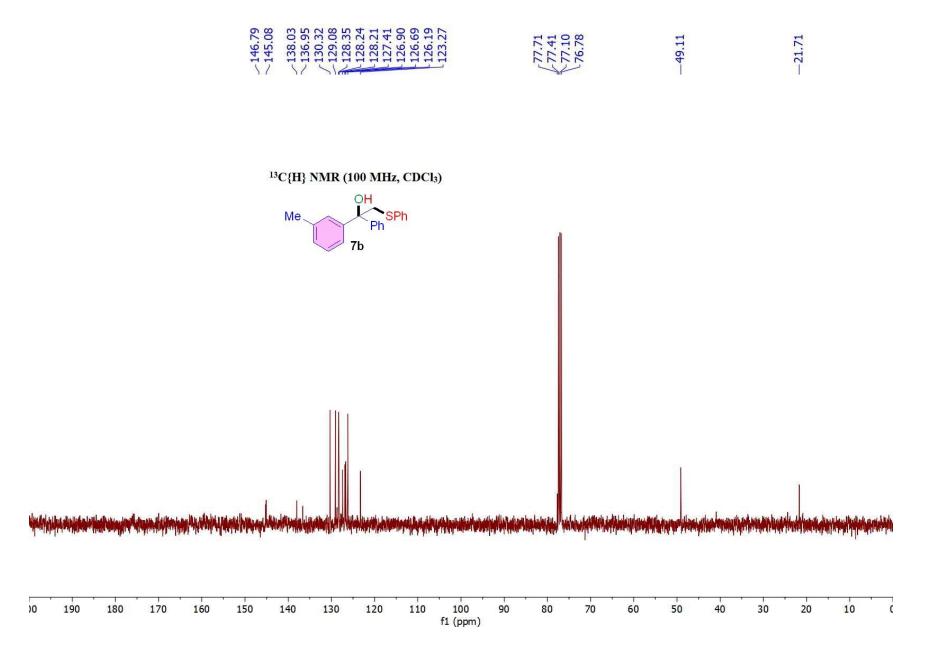


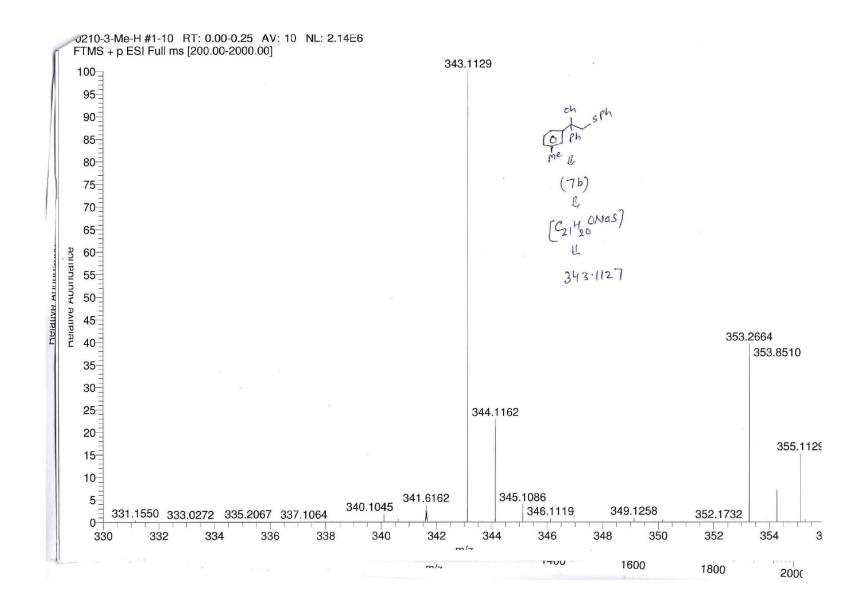


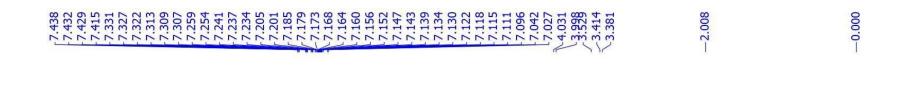


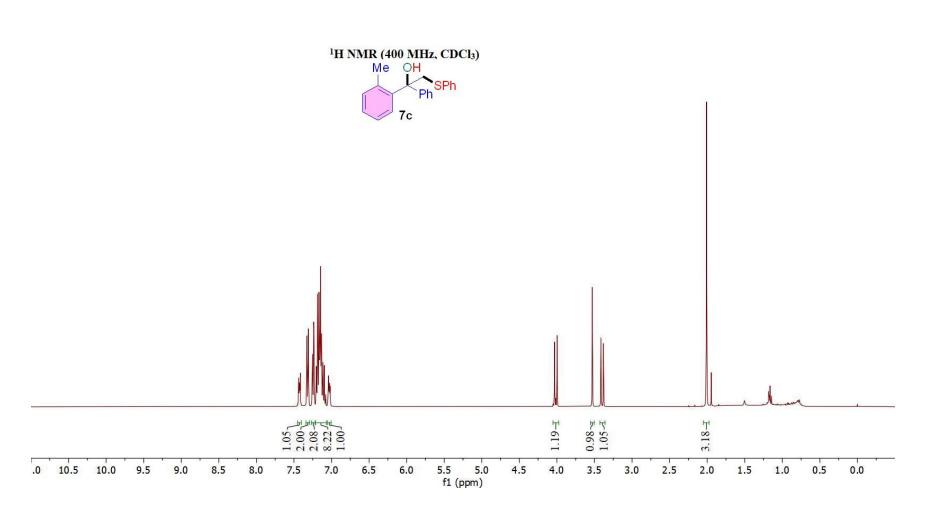






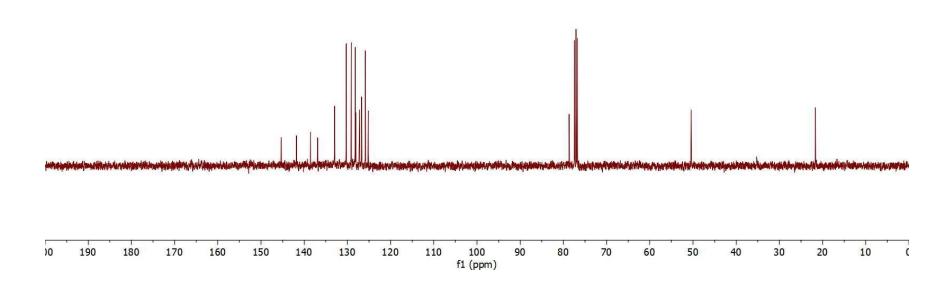


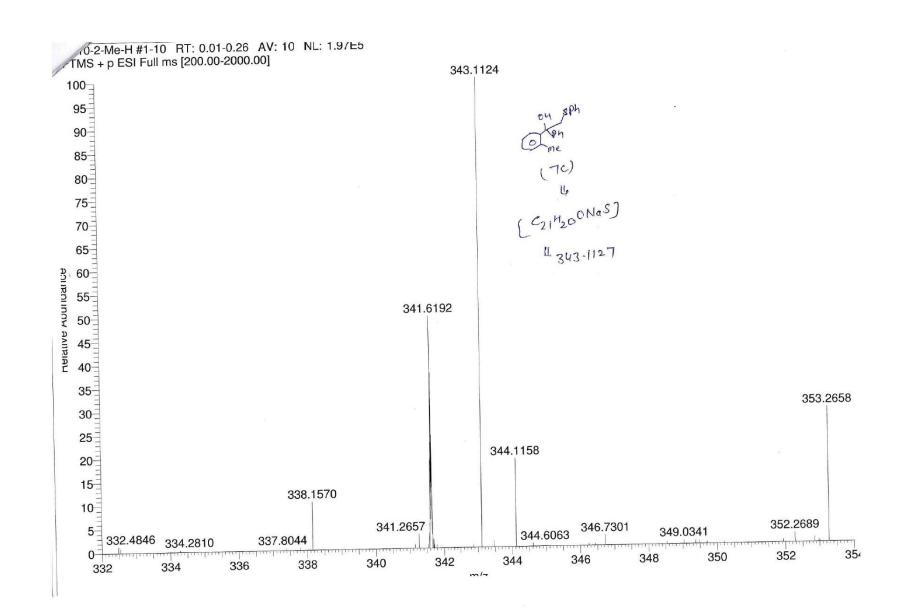


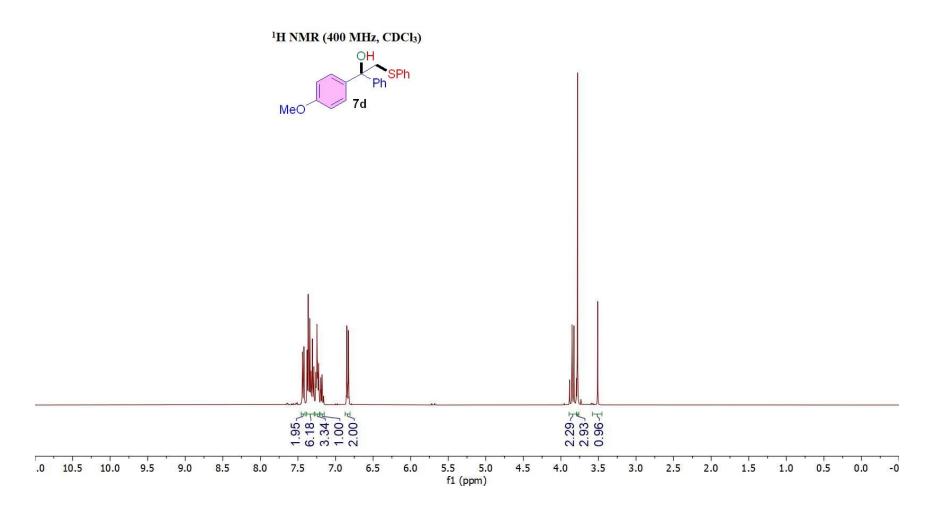




$^{13}\mathrm{C}\{\mathrm{H}\}$ NMR (100 MHz, CDCl₃)

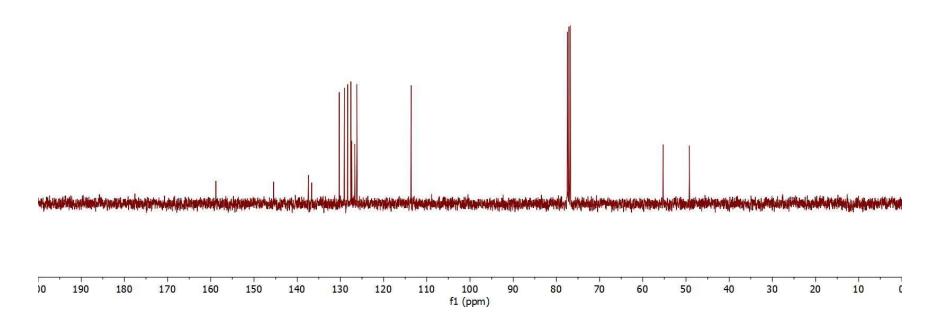




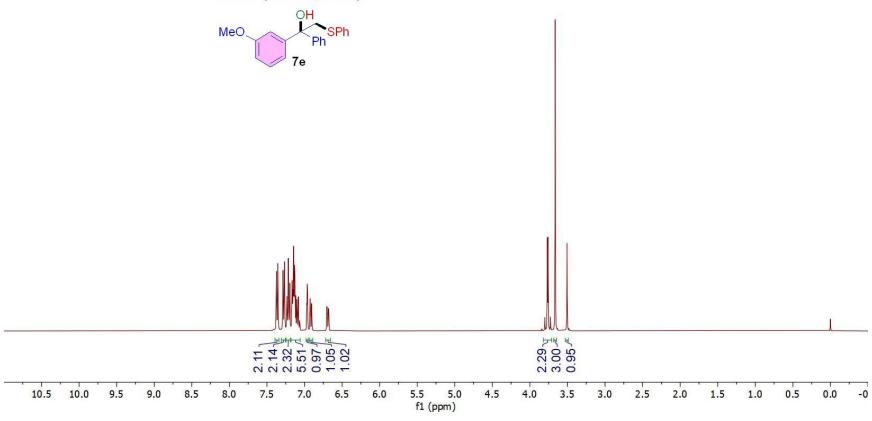




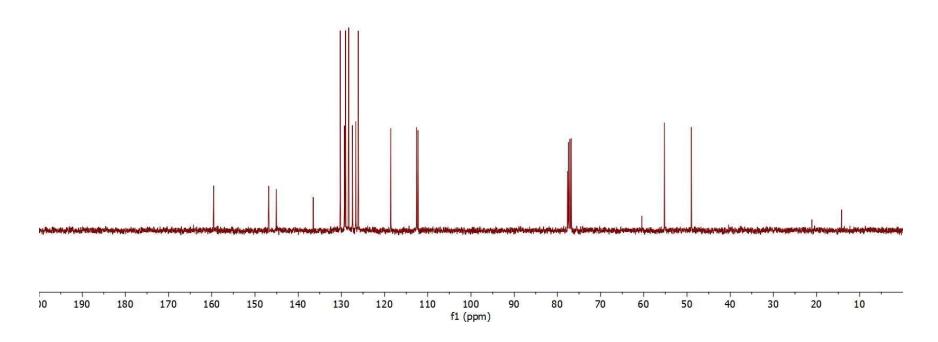
13C{H} NMR (100 MHz, CDCl₃)

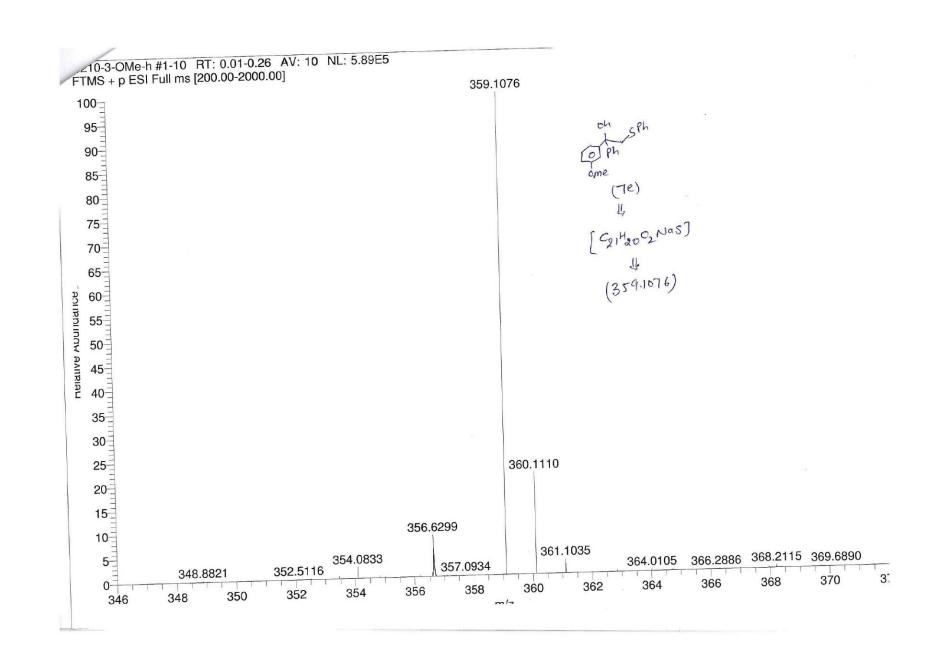


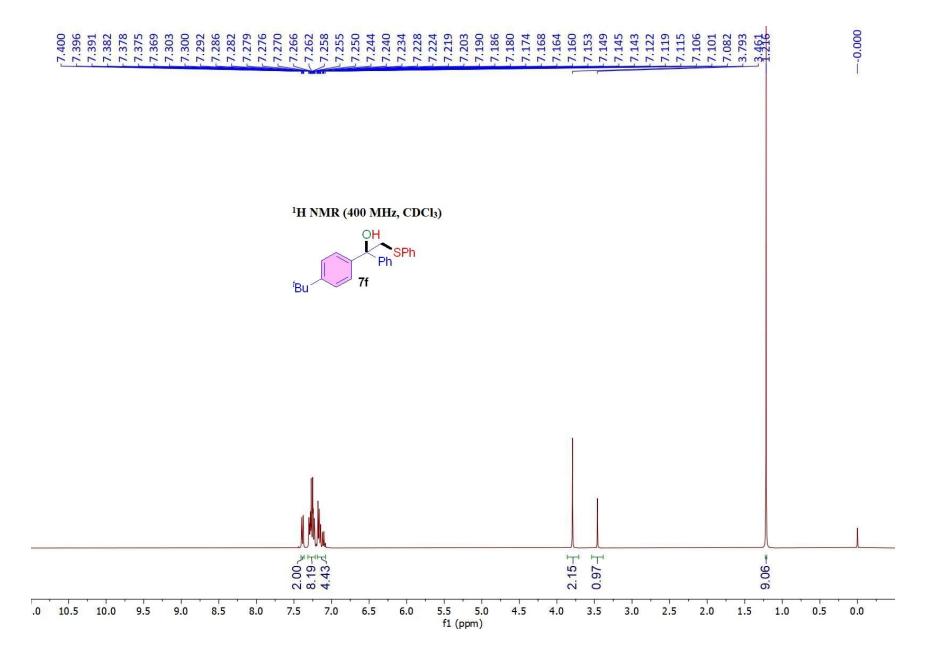




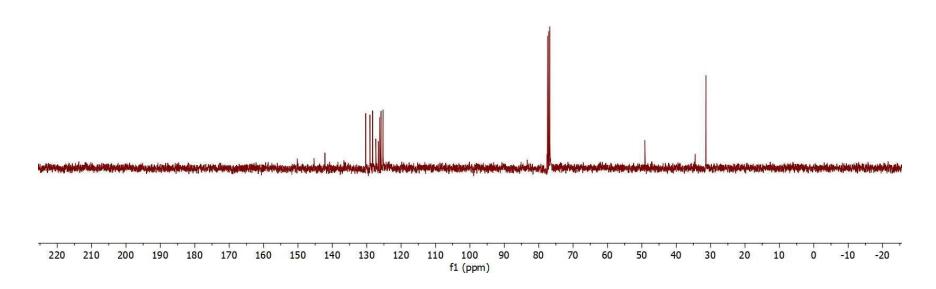


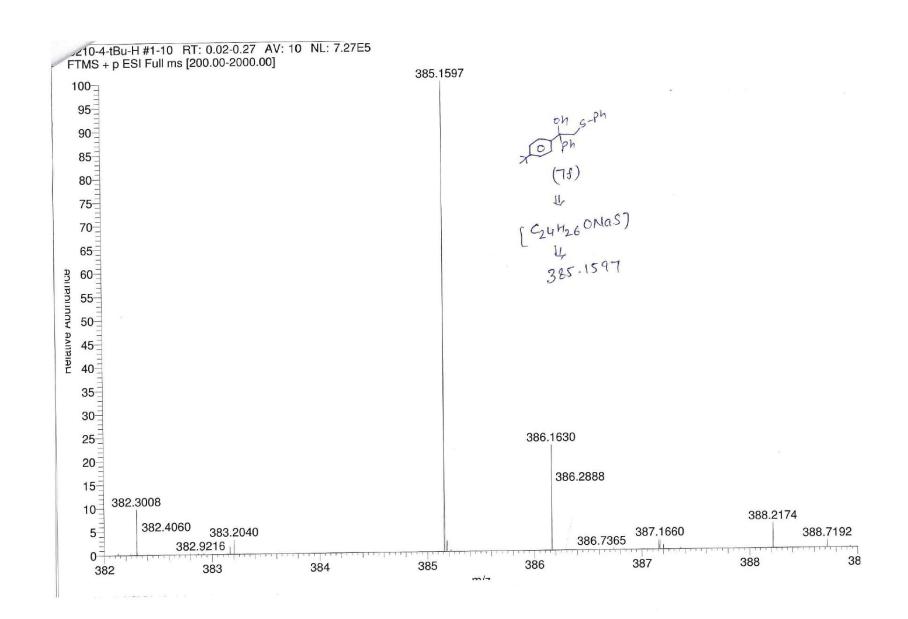




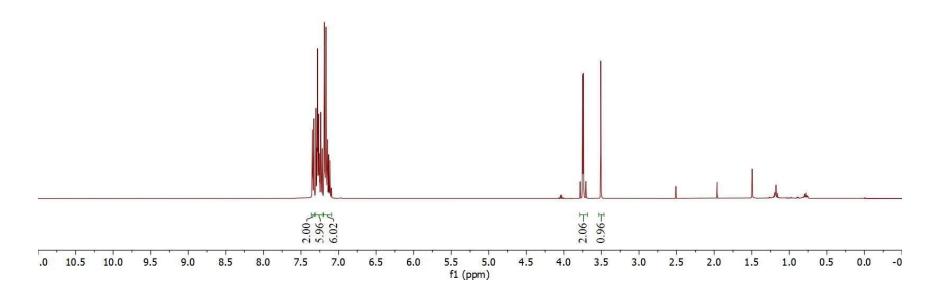








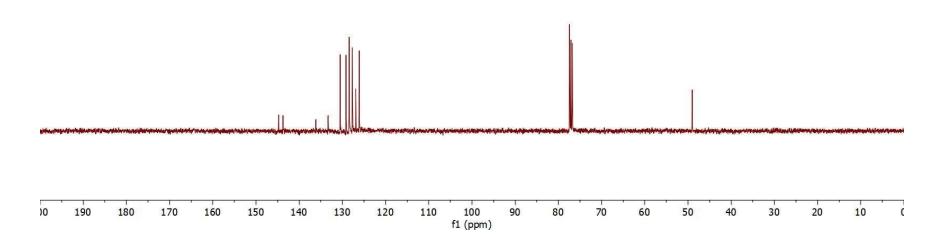


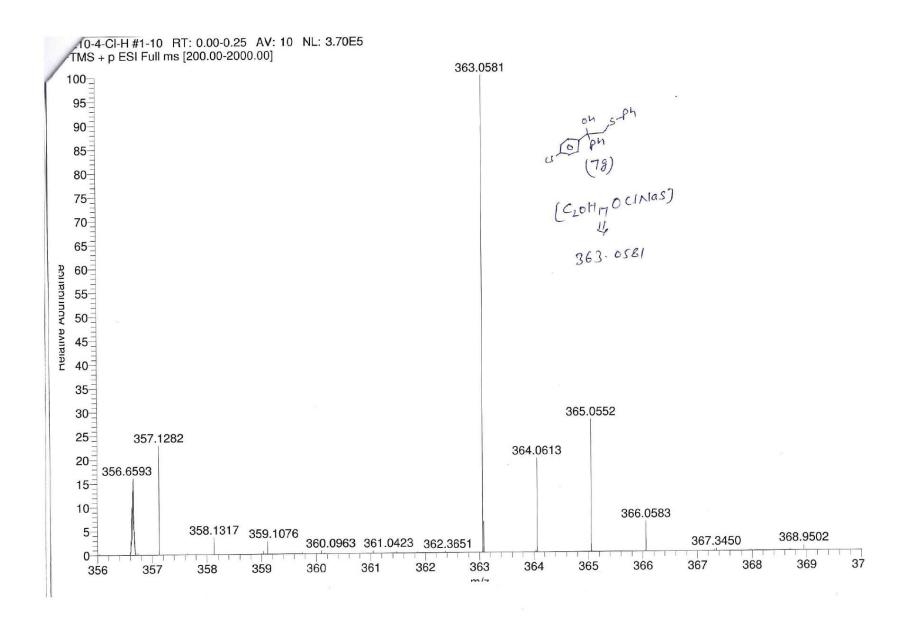


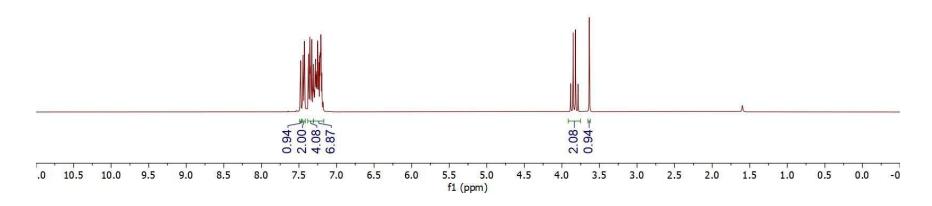




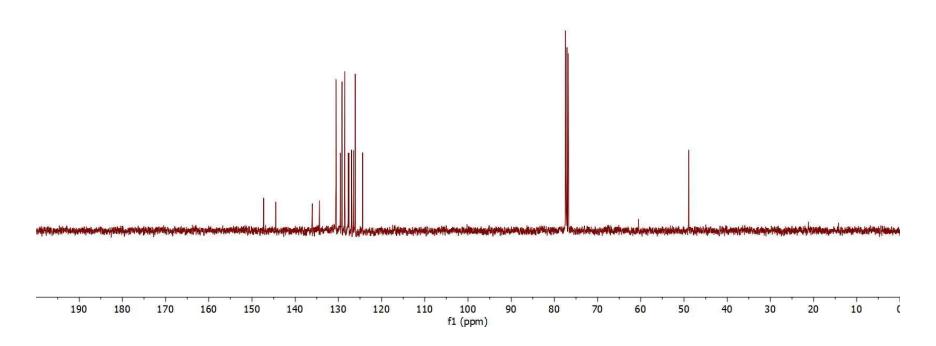
-49.02

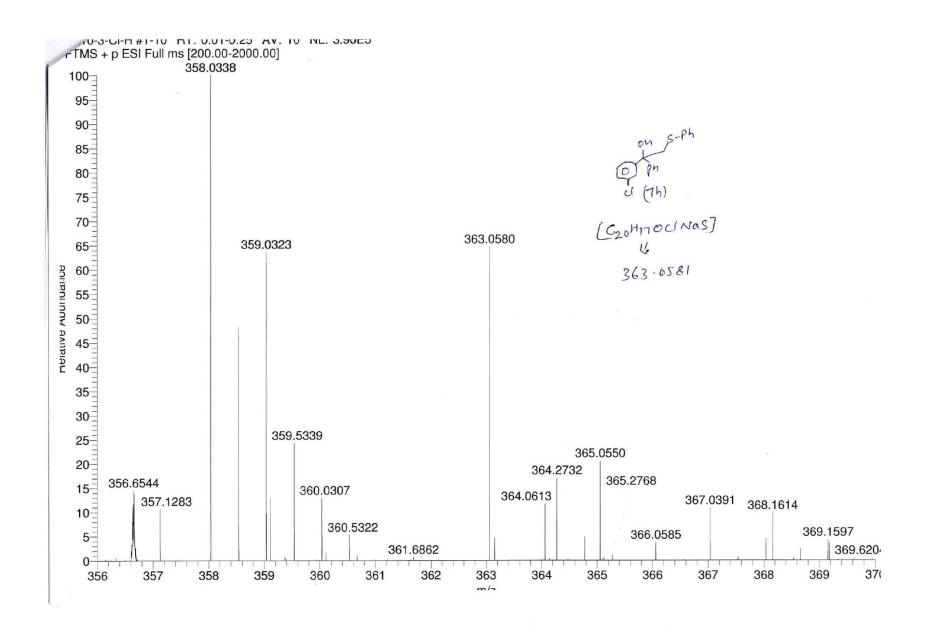






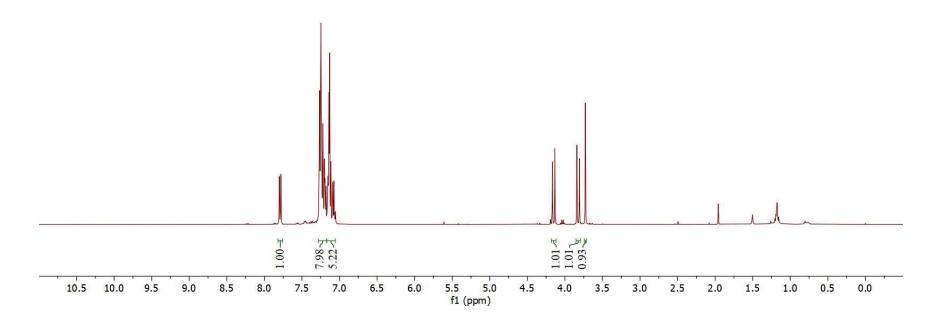








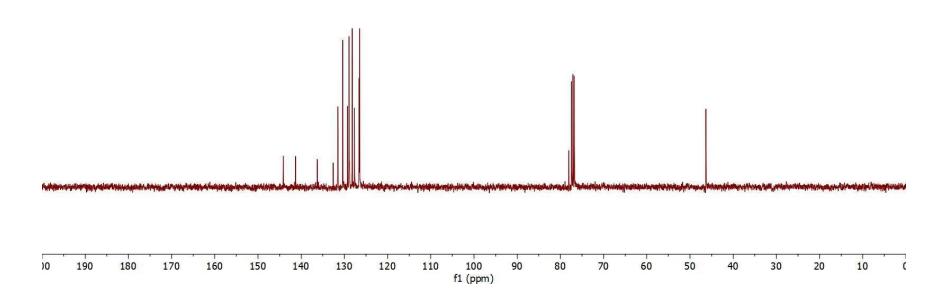


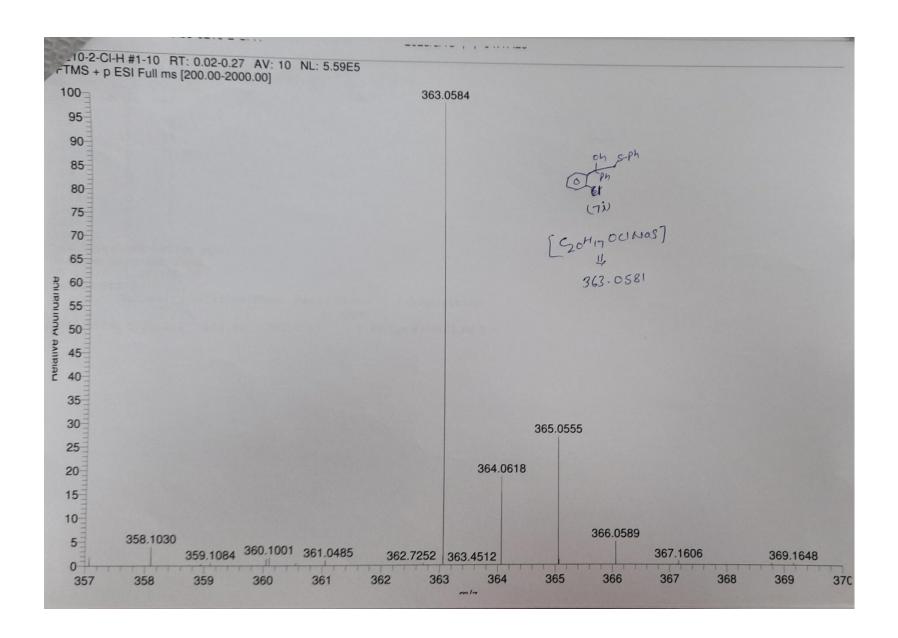


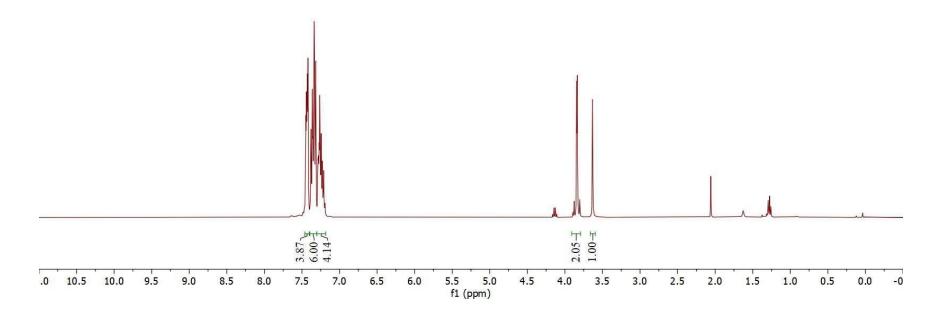




-46.32



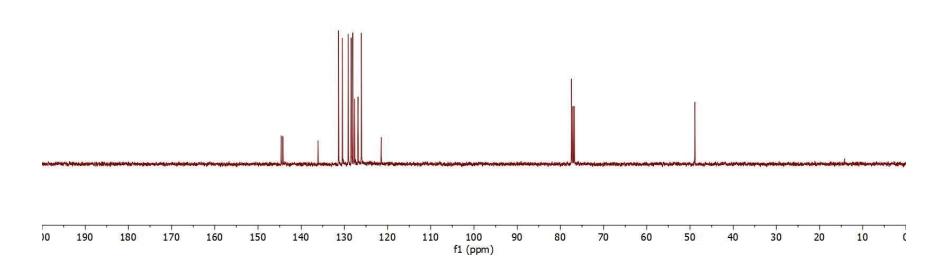


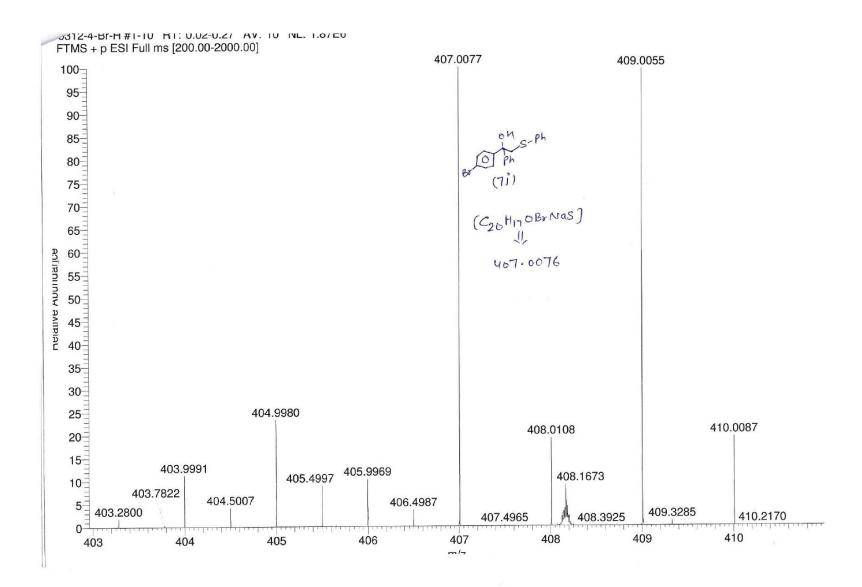


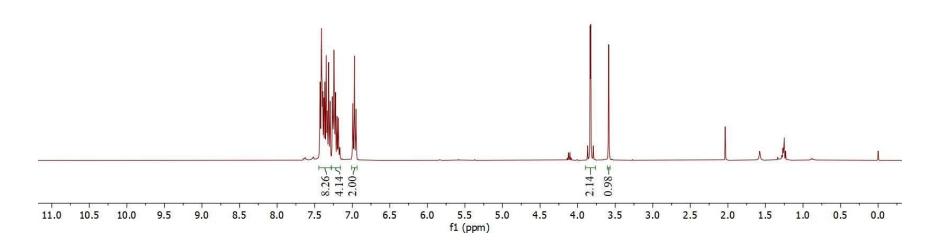






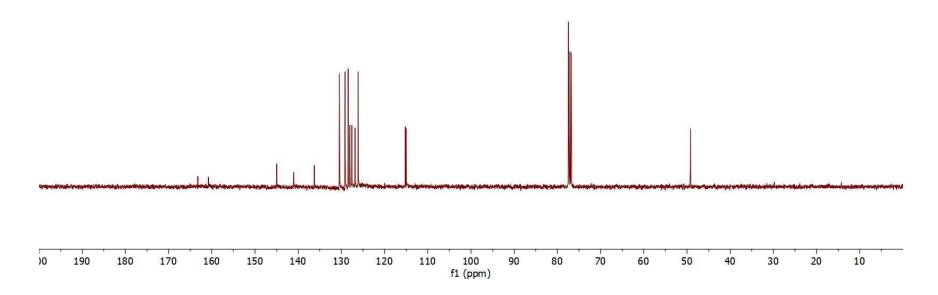


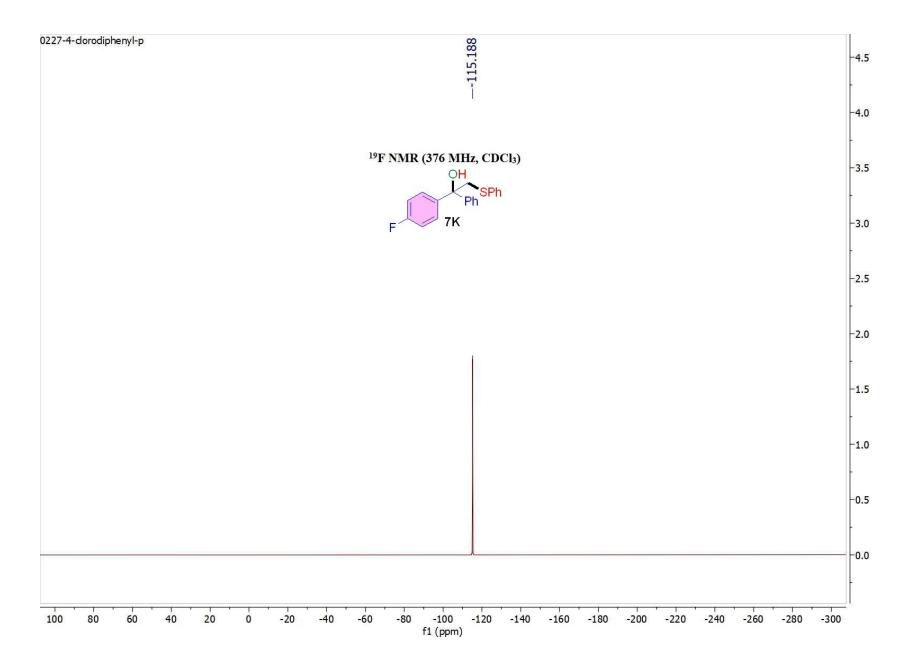


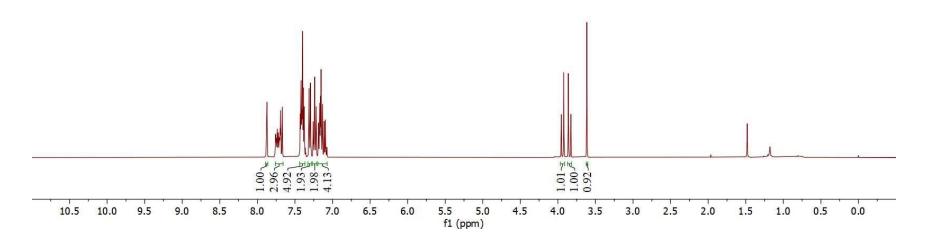


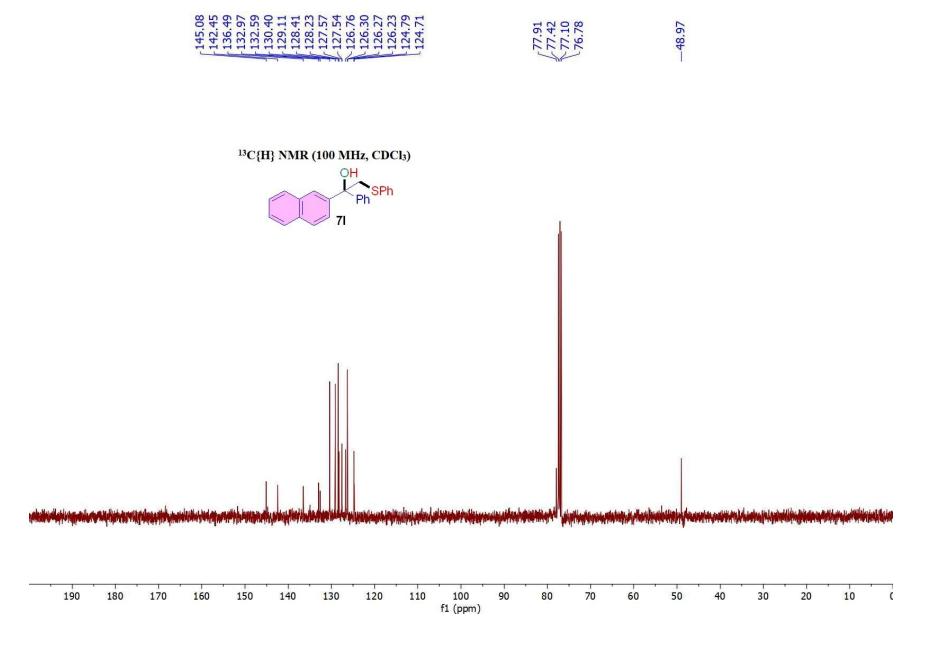


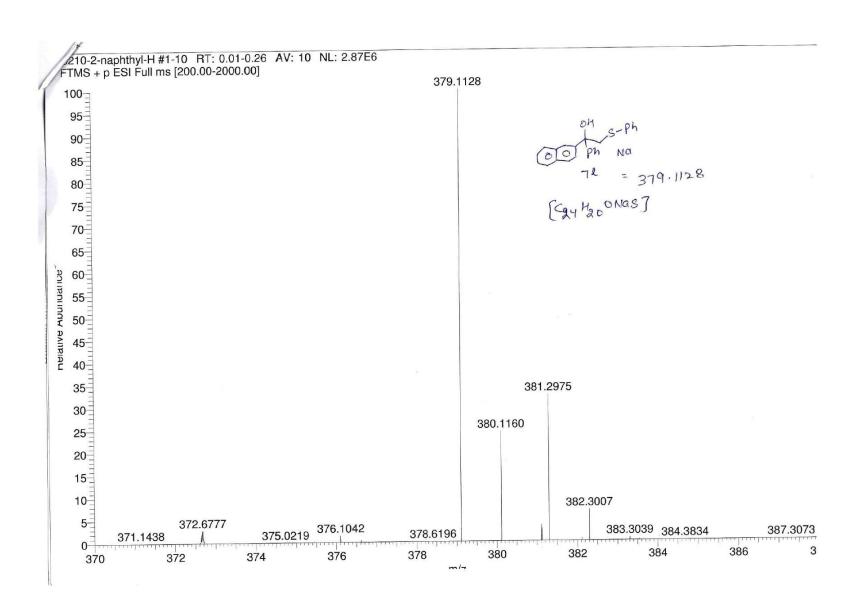
13C{H} NMR (100 MHz, CDCl₃)

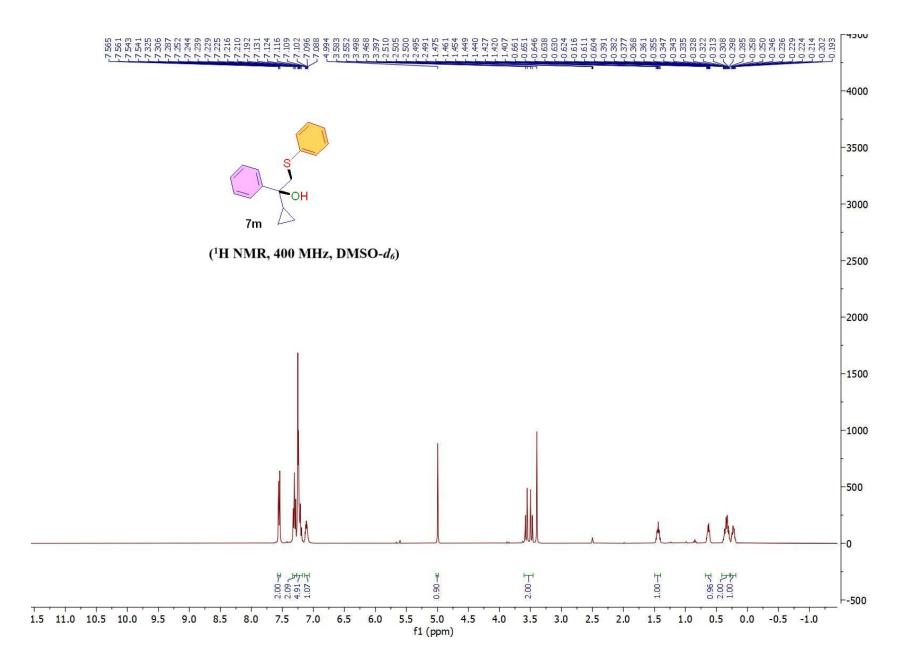


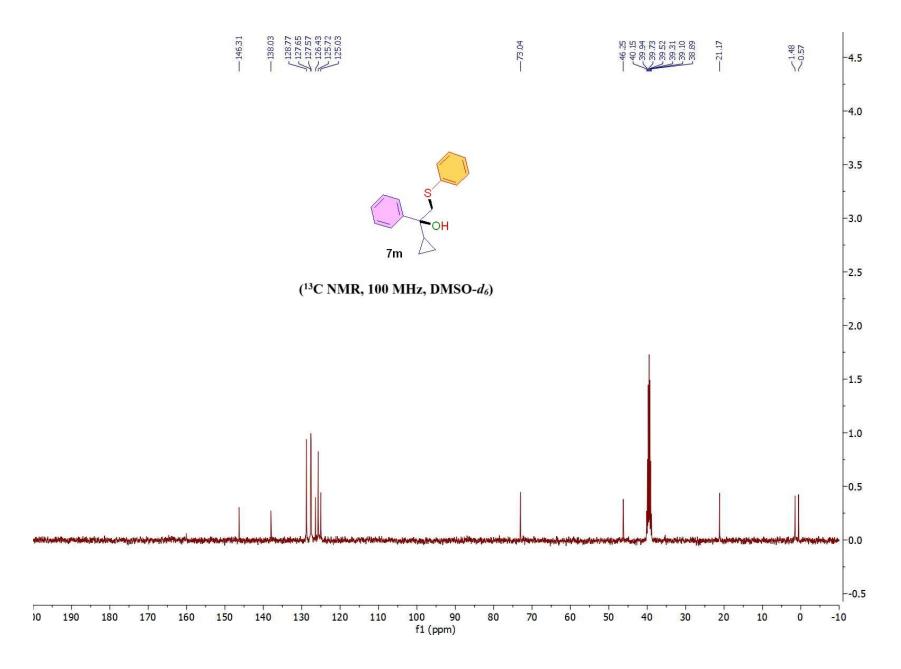


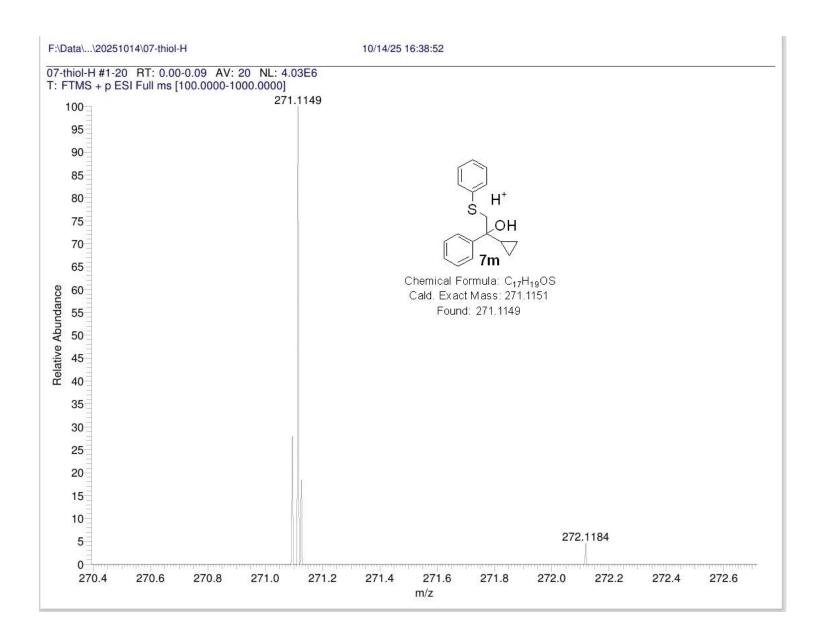












File :D:\MassHunter\GCMS\1\data\830\SJC24040201.D

Operator :

Acquired : 02 Apr 2024 14:31 using AcqMethod 20190906-1.M

Instrument: 830-GCMS Sample Name: 1

Misc Info : Vial Number: 13



Exact Mass 266.

