

Fig. S1 ^1H NMR spectrum of the representative polymerization reaction mixture used in all the calculations for the percentage conversion

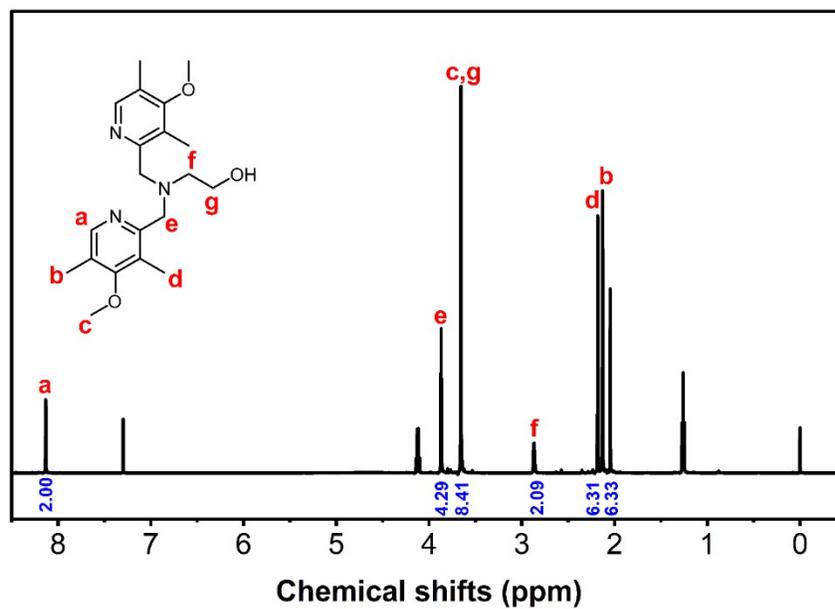


Fig. S2 ¹H NMR analysis of intermediate product 1 performed in CDCl₃ employing TMS as an internal reference

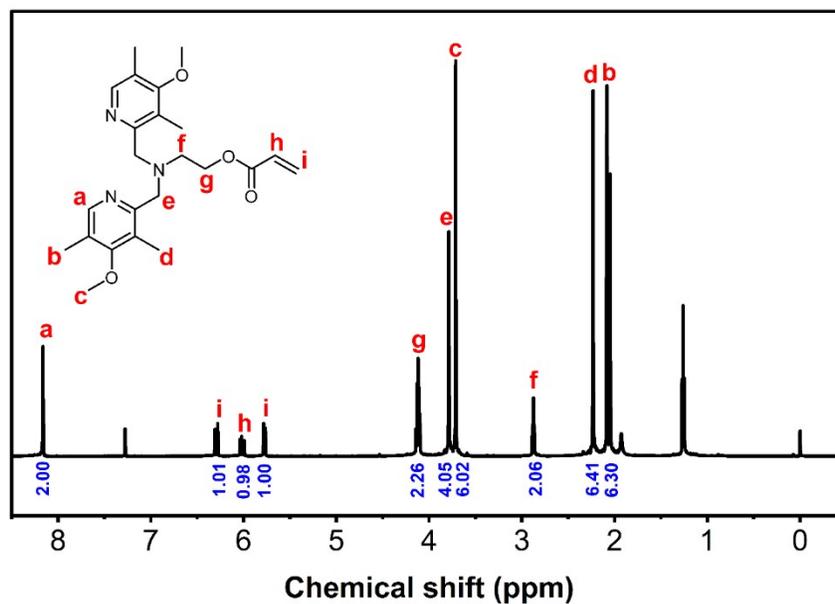


Fig. S3 ¹H NMR analysis of intermediate product **2** performed in CDCl₃ employing TMS as an internal reference

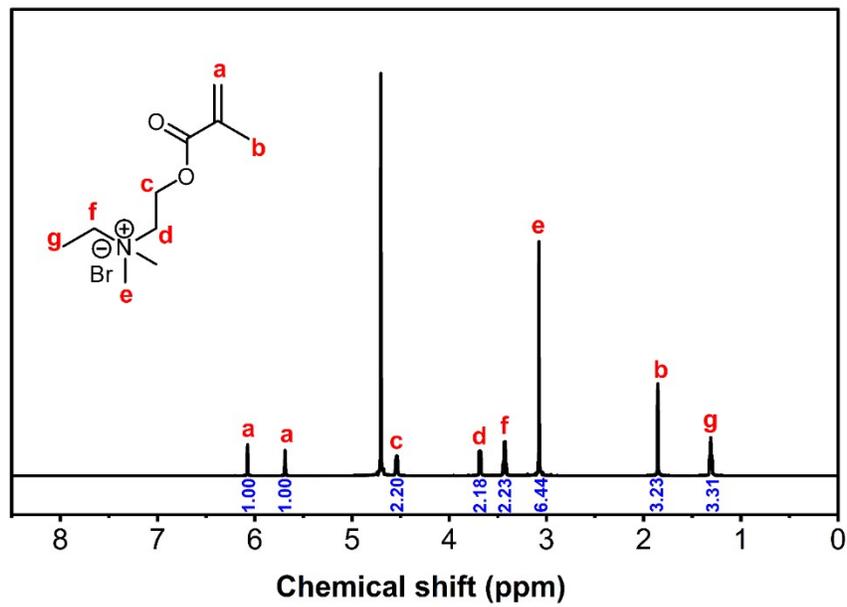


Fig. S4 ¹H NMR analysis of intermediate product **3** performed in D₂O employing TMS as an internal reference

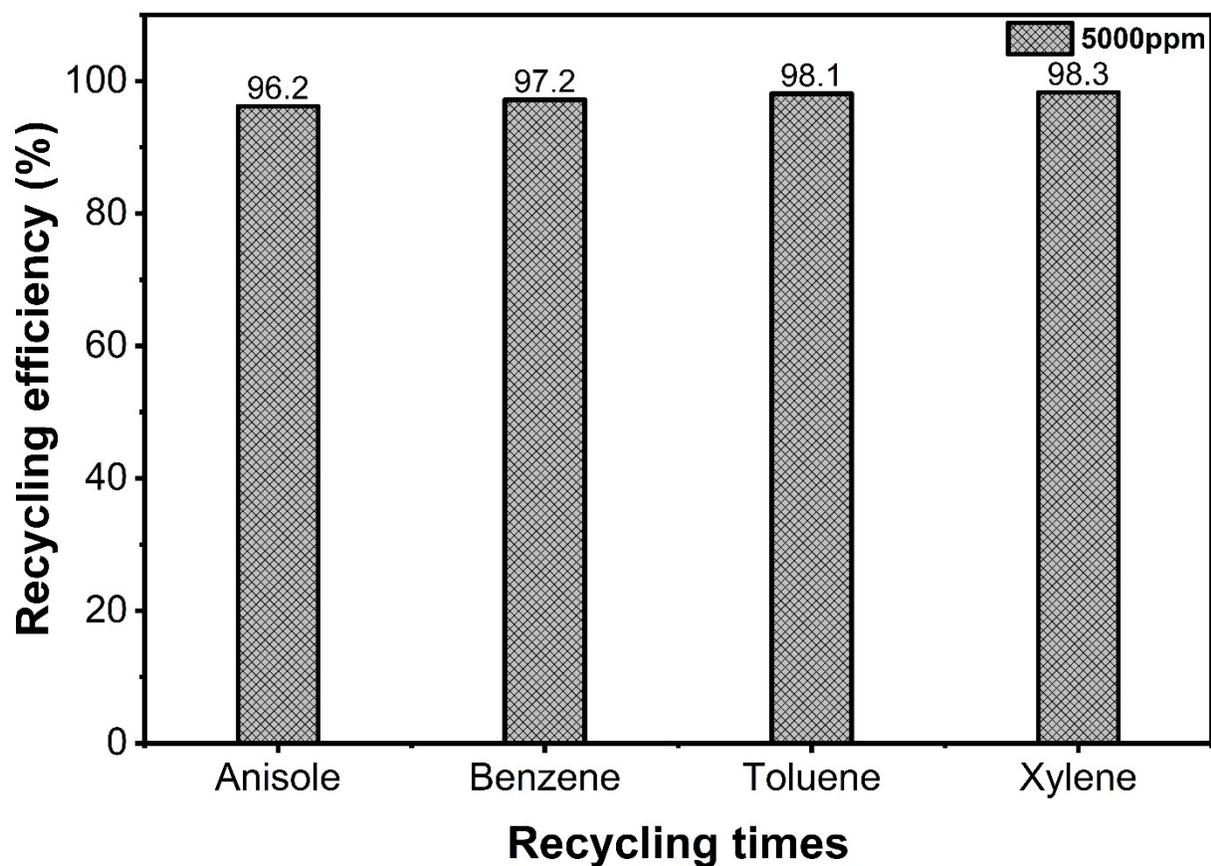


Fig. S5 Recycling efficiency represented as a percentage of residual copper catalyst retained in the PILLL1 phase relative to initial copper catalyst content, determined by ICP-MS, in four different solvents. Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILLL1]_0/[AIBN]_0 = 200/1/1/4/1$, $V_{MMA} = 1$ mL, $V_{anisole} = 2$ mL, temperature = 80 °C

Table S1 Different batches of PEG₃₀₀MMA obtained from the same company and their ratios to attain the TPSC-based ICAR ATRP system

Entry	Company	Anisole (mL)	MMA (mL)	CuBr ₂ (mg)	AIBN (mg)	MA- sLN	ILM	PEG300 MMA	rt ^a	80 °C	rt ^b	No. of Batches
1	Macklin ^a	2.0	1.0	1.0	7.8	1	4.2	5.0	M	M	I	18
2	Macklin ^b	2.0	1.0	1.0	7.8	1	3.2	5.0	M	M	I	4
3	Macklin ^c	2.0	1.0	1.0	0.0	1	2.9	5.0	M	M	I	3

I = Immiscibility and M = Miscibility; rt^a = At room temperature and rt^b = After Cooling cycle to room temperature; ^{a,b,c} PEG₃₀₀MMA batches

Table S2 Sustainability assessment guide of 4 selected organic solvents based on safety, health, and environmental scores¹⁻³

Entry	Solvent	Boiling point (°C)	Safety score	Healthy score	Environmental score	Solvent sustainability ranking	Overall score
1	Anisole	154	4	1	5	Recommended	1
2	Benzene	80	6	10	3	Hazardous	4
3	Xylene	140	4	2	5	Problematic	2
4	Toluene	111	5	6	3	Problematic	3

The color code represents the scoring index: green for 1–3, yellow for 4–6, and red for 7–10

Table S3 Effects of varying Macroligand (PILLL1) for TPSC-based ICAR ATRP system

Entry	x	x (mg)	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n
1	5	279	4	80	16200	13700	1.36
2	4	223	4	88	17800	13400	1.09
3	3	167	4	74	15000	13000	1.20
4	2	112	4	72	14600	14100	1.13
5	1	56	4	68	13800	11600	1.27

Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILLL1]_0/[AIBN]_0 = 200/1/1/x/1$, $V_{MMA} = 1$ mL, $V_{anisole} = 2$ mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times Conv.\% + M_{w,EBiB}$

Table S4 Effects of varying AIBN for TPSC-based ICAR ATRP system

Entry	x	x (mg)	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n
1	1.25	0.980	4	92	18600	15300	1.11
2	1.00	0.780	4	88	17800	13400	1.09
3	0.50	0.390	4	68	13800	12600	1.14
4	0.25	0.195	4	40	8200	7400	1.16
5	0.10	0.078	4	*	*	*	*

Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILLL1]_0/[AIBN]_0 = 200/1/1/4/x$, $V_{MMA} = 1$ mL, $V_{anisole} = 2$ mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times Conv.\% + M_{w,EBiB}$.

*No polymer formed

Table S5 Effects of varying CuBr₂ for TPSC-based ICAR ATRP system

Entry	x	x (mg)	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n
1	1.00	10.4	4	88	17800	13400	1.09
2	0.50	5.2	4	80	16200	10500	1.21
3	0.25	2.6	4	92	18600	16900	1.25
4	0.13	1.3	4	90	18200	15500	1.13

Polymerization conditions: [MMA]₀/[EBiB]₀/[CuBr₂]₀/[PILLL1]₀/[AIBN]₀ = 200/1/1/4/1, V_{MMA} = 1 mL, V_{anisole} = 2 mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times \text{Conv.}\% + M_{w,EBiB}$

Table S6 Effects of varying volume of solvent for TPSC-based ICAR ATRP system

Entry	Anisole (x) (mL)	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n
1	3.0	4	82	16600	13700	1.30
2	2.5	4	75	15200	13500	1.35
3	2.0	4	88	17800	13400	1.09

Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILLL1]_0/[AIBN]_0 = 200/1/1/4/1$, V_{MMA}

$= 1$ mL, $V_{anisole} = x$ mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times Conv.\% + M_{w,EBiB}$

Table S7 Effects of varying Temperature for TPSC via ICAR ATRP system

Entry	Temperature (x) (°C)	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n
1	70	4	65	13200	10700	1.28
2	80	4	88	17800	13400	1.09
3	90	4	87	17600	15900	1.16

Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILL1]_0/[AIBN]_0 = 200/1/1/4/1$, V_{MMA}

$= 1$ mL, $V_{anisole} = 2$ mL, temperature = x °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times Conv.\% + M_{w,EBiB}$

Table S8 Effects of varying alkyl halide initiator (EBiB) for TPSC-based ICAR ATRP system

Entry	x	x (μL)	Time (h)	Conv. (%)	$M_{n,\text{th}}$ (g/mol)	$M_{n,\text{GPC}}$ (g/mol)	M_w/M_n
1	1.5	10.2	4	82	11100	13100	1.18
2	1.0	6.8	4	88	17800	13400	1.09
3	0.5	3.4	4	76	30400	21700	1.43

Polymerization conditions: $[\text{MMA}]_0/[\text{EBiB}]_0/[\text{CuBr}_2]_0/[\text{PILLL1}]_0/[\text{AIBN}]_0 = 200/x/1/4/1$, V_{MMA}

$= 1 \text{ mL}$, $V_{\text{anisole}} = 2 \text{ mL}$, temperature = $80 \text{ }^\circ\text{C}$. $M_{n,\text{th}} = ([\text{M}]_0/[\text{M}]) \times [\text{MMA}] \times \text{Conv.}\% + M_{w,\text{EBiB}}$

Table S9 Recycling performance of the PILL1/CuBr₂ catalyst in TPSC-based ICAR ATRP system

Entry	Recycling times	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n	CR ^a (ppm)
1	1	7	64	13000	18000	1.40	187
2	2	7	52	10600	17400	1.38	114
3	3	7	16	3400	*	*	*

Polymerization conditions: [MMA]₀/[EBiB]₀/[CuBr₂]₀/[PILL1]₀/[AIBN]₀ = 200/1/0.1/1.2/0.2, V_{MMA} = 1 mL, V_{anisole} = 2 mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times \text{Conv.}\%$ + $M_{w,EBiB}$. CR^a refers to the residual catalyst present in the organic polymer solution phase, measured by ICP-MS. *No monomer formed, and the reaction mixture became homogeneous

Table S10 Recycling performance of the PILL1/CuBr₂ catalyst in TPSC-based ICAR ATRP system under varied time and AIBN concentration

Entry	Recycling times	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n	CR ^a (ppm)
1	1	7	64	13000	41300	1.37	34.5
2	2	6	84	17000	84100	1.31	32.9
3	3	5	85	17200	68300	1.20	28.9
4	1*	7	65	13200	45600	1.55	17.5
5	2*	6	83	16800	83500	1.27	16.7
6	3*	5	84	17000	94100	1.24	14.2

Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILL1]_0/[AIBN]_0 = 200/1/0.1/4/x = 0.2, 0.3, 0.4$, and $*[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILL1]_0/[AIBN]_0 = 200/1/0.05/4/x = 0.2, 0.3, 0.4$, $V_{MMA} = 1$ mL, $V_{anisole} = 2$ mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times Conv.\% + M_{w,EBiB}$. CR^a refers to the residual catalyst present in the organic polymer solution phase, measured by ICP-MS

Table S11 Recycling performance of the PILL1/CuBr₂ catalyst in TPSC-based ICAR ATRP system under varied time and AIBN concentration

Entry	Recycling times	Time (h)	Conv. (%)	$M_{n,th}$ (g/mol)	$M_{n,GPC}$ (g/mol)	M_w/M_n	CR ^a (ppm)
1	1 ^a	7	68	13800	45300	1.58	10.3
2	2 ^b	6	80	16200	41000	1.48	9.0
3	3 ^c	5	88	17800	35500	1.48	5.8
4	4 ^c	5	88	17800	30500	1.44	4.4
5	5 ^c	5	84	17000	33000	1.25	3.4

Polymerization conditions: $[MMA]_0/[EBiB]_0/[CuBr_2]_0/[PILL1]_0/[AIBN]_0 = 200/1/0.025/4/x = 0.2^a, 0.3^b, 0.4^c$, $V_{MMA} = 1$ mL, $V_{anisole} = 2$ mL, temperature = 80 °C. $M_{n,th} = ([M]_0/[M]) \times [MMA] \times Conv.\% + M_{w,EBiB}$. CR^a refers to the residual catalyst present in the organic polymer solution phase, measured by ICP-MS