

Supporting Information

Facilitated gallium incorporation in blue-emitting $\text{In}_{1-x}\text{Ga}_x\text{P}$ alloy quantum dots via monomeric gallium precursors

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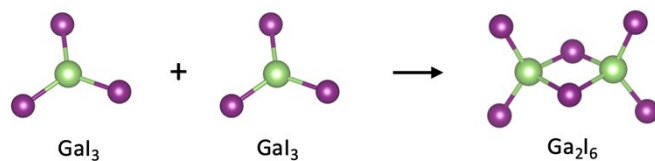
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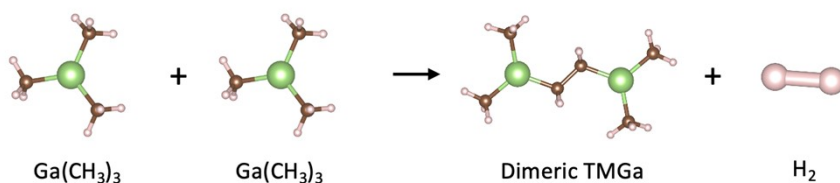
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(a) $\Delta E = -0.839$ eV (preferred)



(b) $\Delta E = 0.934$ eV (unpreferred)



(c) $\Delta E = -3.769$ eV (preferred)

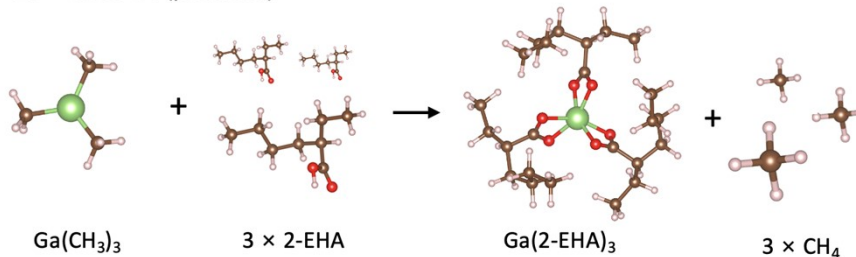


Fig. S1 Energy change (ΔE) for the reaction of the (a) dimerization of gallium iodide, (b) dimerization of TMGa, and (c) formation of gallium tri-2-ethylhexanoate from TMGa, calculated using first-principles density functional theory (DFT).

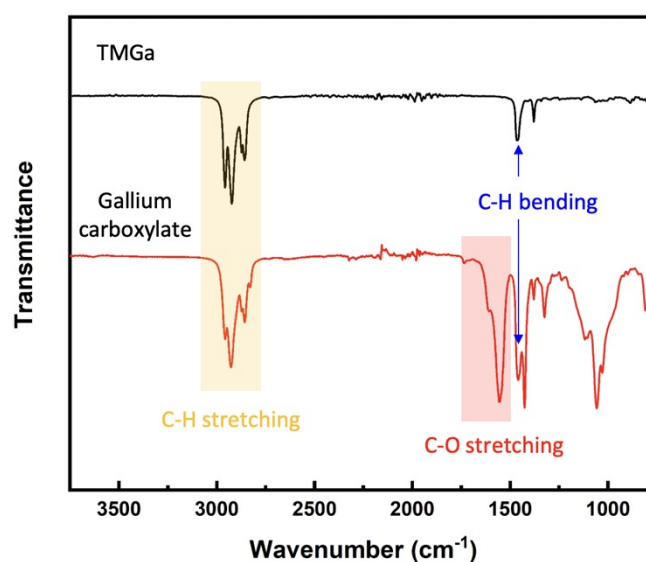


Fig. S2 FTIR spectra of gallium 2-ethylhexanoates derived from TMGa. Four equivalents of 2-EHA were reacted with TMGa.

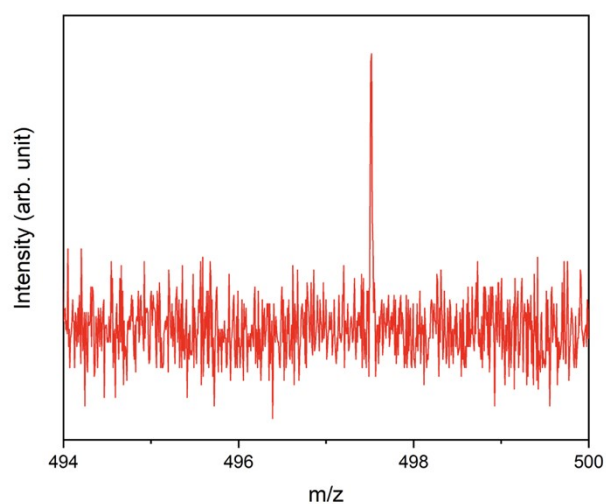


Fig. S3 LDI-TOF spectra of gallium 2-ethylhexanoates synthesized by the reaction of TMGa with 2-EHA.

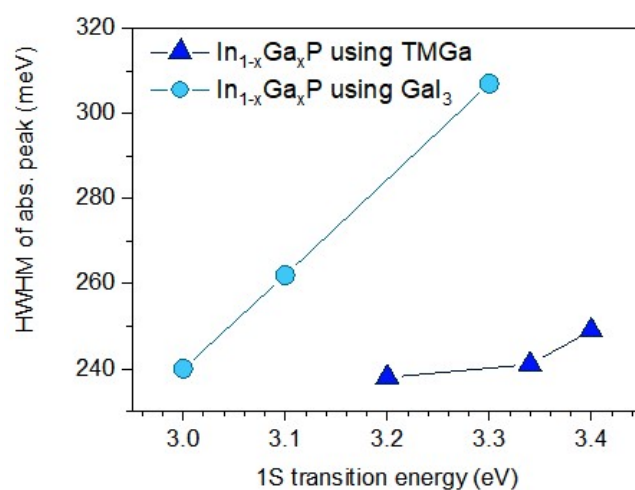


Fig. S4 HWHM values of the absorption peak of $\text{In}_{1-x}\text{Ga}_x\text{P}$ QDs according to their 1S transition peak energies. The HWHM values are shown for QDs derived from TMGa (sky blue) and gallium iodide (blue) at different Ga-to-In feed ratios.

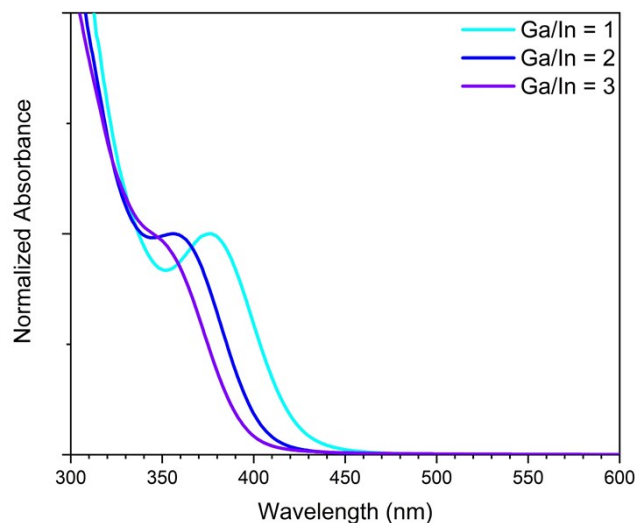


Fig. S5 UV-vis absorption spectra of $\text{In}_{1-x}\text{Ga}_x\text{P}$ cores synthesized from $\text{Ga}(\text{acac})_3$ with different Ga-to-In feed ratios.

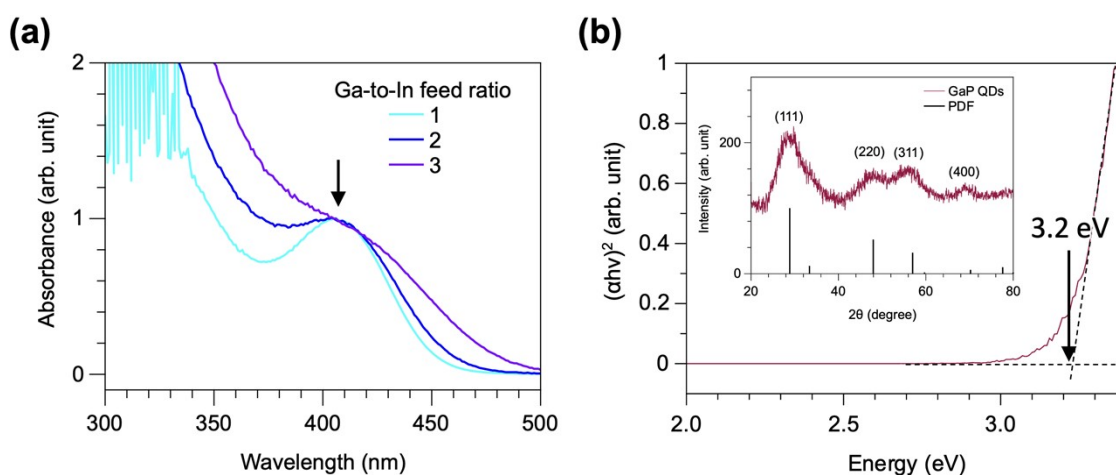


Fig. S6 Characterization of $\text{In}_{1-x}\text{Ga}_x\text{P}$ and GaP QDs using di-substituted gallium carboxylate derived from TMGa. (a) Absorption spectra of $\text{In}_{1-x}\text{Ga}_x\text{P}$ QDs using $\text{Ga}(\text{CH}_3)(\text{LA})_2$ with different Ga-to-In feed ratios. The black arrow indicates the 1S transition peak of the QDs (b) Tauc plot of GaP QDs using $\text{Ga}(\text{CH}_3)(\text{LA})_2$. The black arrow indicates the optical band gap of GaP QDs. The inset shows the XRD pattern for the GaP QDs (red) and standard peaks for GaP (JCPDS #32-0397, vertical lines).

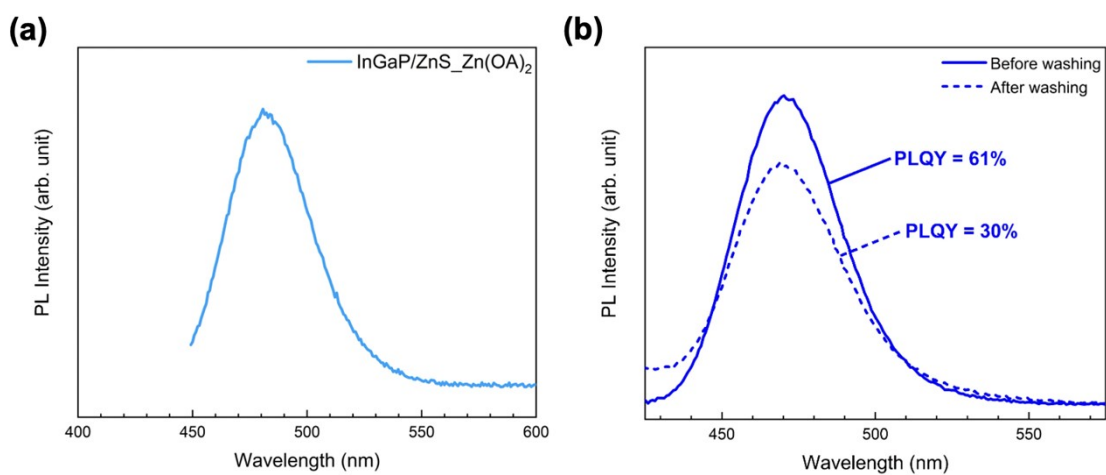


Fig. S7 PL properties of In_{1-x}Ga_xP/ZnS QDs according to Zn precursors. (a) PL spectrum of In_{1-x}Ga_xP/ZnS QDs synthesized using Zn(OA)₂ as the only Zn precursor for ZnS shell growth. (b) Change in the PL spectra and PL QYs of In_{1-x}Ga_xP/ZnS QDs synthesized by ZnCl₂-OAm depending on the purification process.

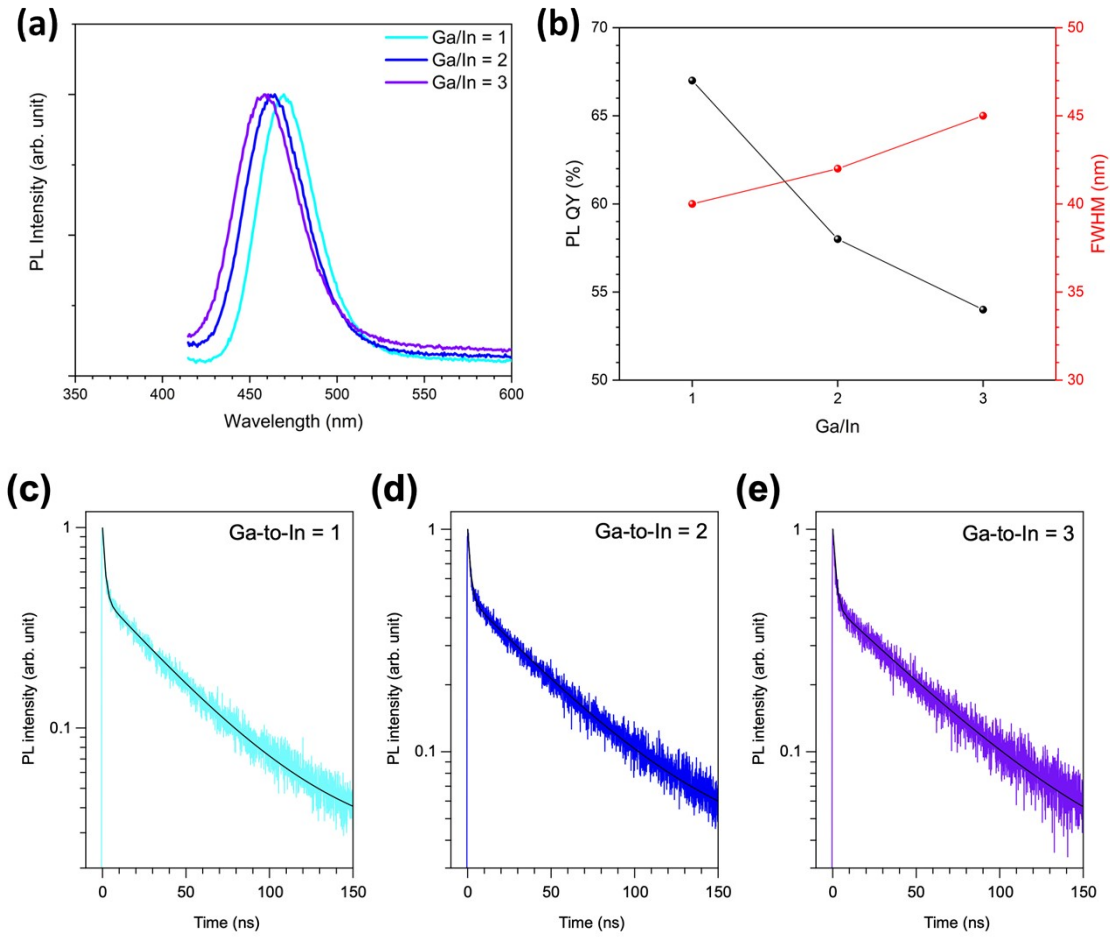


Fig. S8 PL properties of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs as function of Ga-to-In feed ratio. (a) Steady-state PL spectra of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs with varying Ga-to-In ratios (1, 2, and 3). (b) PL QY and FWHM of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs for Ga-to-In ratios of 1, 2, and 3. (c–e) Time-resolved PL decay curves of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs with Ga-to-In ratios of (c) 1, (d) 2, and (e) 3. Black solid lines represent the biexponential fits using the function $y = A_1 e^{-t/\tau_1} + A_2 e^{-t/\tau_2}$.

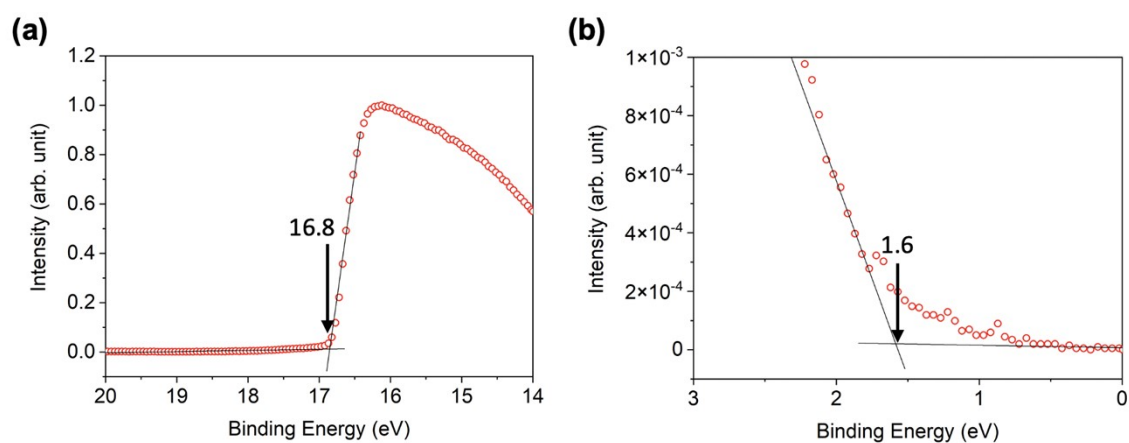


Fig. S9 UPS spectra for (a) secondary electron cutoff and (b) valence band edge regions of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs.

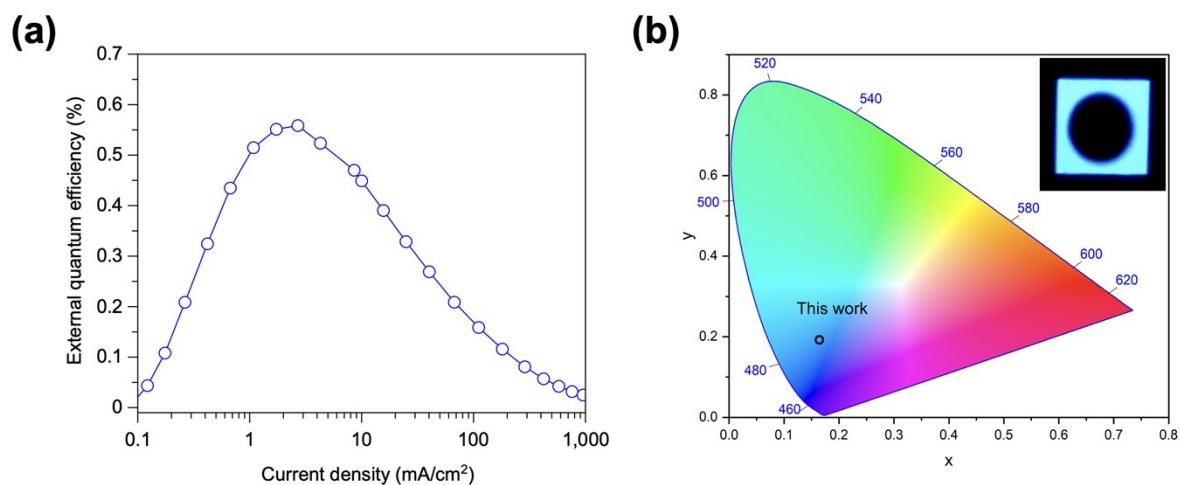


Fig. S10 (a) External quantum efficiency (EQE)–current density characteristics of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QD-based QLEDs. (b) Chromaticity diagram of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QD-based QLEDs on CIE 1931 coordinates. Inset shows a photograph of QLEDs in operation at 5 V.

Table S1 Summary of the elemental analysis for gallium 2-ethylhexanoate derived from gallium iodide (i.e., $\text{Ga}_2\text{I}_2(2\text{-EHA})_4$, synthesized with a gallium iodide-to-2-ethylhexanoic acid ratio of 1:4)

Elements	C	H	O
Expected mass percentage (%)	39.2	6.7	13.2
Measured mass percentage (%)	36.0	6.6	20.1

Table S2 Molar ratio (%) of Zn, In, Ga, and P in $\text{In}_{1-x}\text{Ga}_x\text{P}$ QDs characterized by ICP-OES results

	using $\text{Ga}(\text{LA})_3$				using $\text{Ga}(\text{CH}_3)(\text{LA})_2$			
	Zn	In	Ga	P	Zn	In	Ga	P
$\text{In}_{1-x}\text{Ga}_x\text{P}$ (2:1:1)	26.6	29.4	19.3	24.8	33.2	20.9	22.0	23.9
$\text{In}_{1-x}\text{Ga}_x\text{P}$ (2:0.67:1.33)	28.4	18.0	27.4	26.2	26.1	10.9	37.8	25.2
$\text{In}_{1-x}\text{Ga}_x\text{P}$ (2:0.5:1.5)	30.2	14.1	31.1	24.6	28.7	8.1	40.4	22.8

Table S3 PL decay fitting results of $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs with different Ga-to-In feed ratios

Ga-to-In ratio	τ_1 (ns)	A_1	τ_2 (ns)	A_2
1	1.5	0.56	45.9	0.44
2	1.7	0.52	53.2	0.48
3	1.6	0.45	58.7	0.55

Table S4 Literature survey for the photoluminescence characteristics of $\text{In}_{1-x}\text{Ga}_x\text{P}$ QDs (emission at 430–500 nm)

Synthesis method	Structure	PL peak (nm)	PLQY (%)	FWHM (nm)	Reference
Cation exchange in organic solvents	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnSeS}/\text{ZnS}$	465-475	80-82	47	1
	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnSeS}/\text{ZnS}$	457	84	52	2
Cation exchange in molten salts	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$	497	30	50	3
	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$	475	20	49	4
	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$	465-475	42	56	5
Colloidal synthesis	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$	500	3	77	6
	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$	486	65	46	7
	$\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$	470	67	40	This work

Table S5 Summary on the EL properties of QLEDs fabricated with $\text{In}_{1-x}\text{Ga}_x\text{P}/\text{ZnS}$ QDs

EL peak wavelength (nm)	477
Turn-on voltage (V)	2.6
Maximum EQE (%)	0.56

Reference

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