

Structural Engineering of MXenes Frameworks with Abundant Surface Functionalities for Enhanced Lithium-Sulfur Battery Electrochemistry

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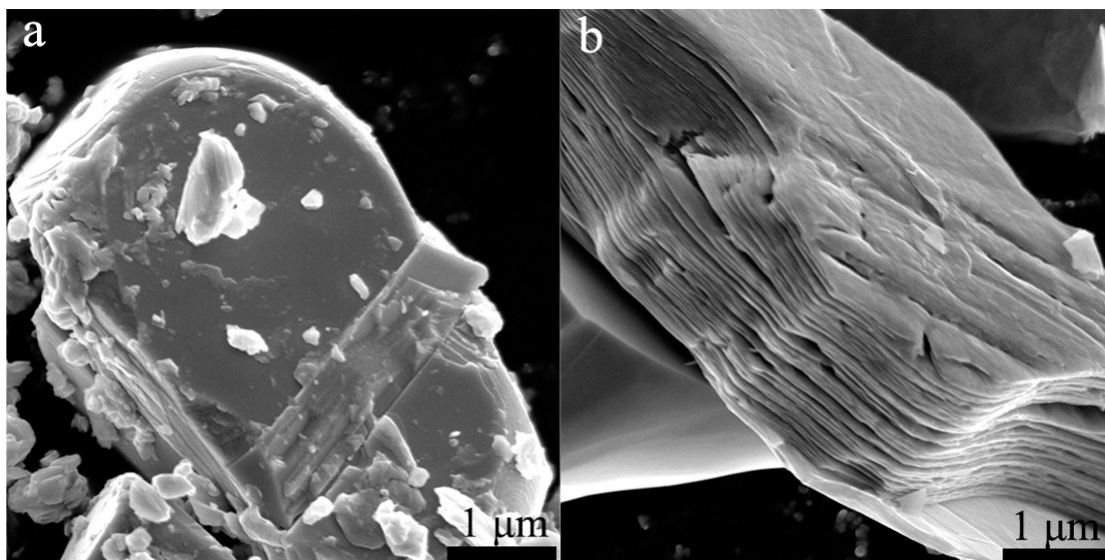


Fig. S1. SEM images of (a) Ti₂AlC, (b) Ti₂CTx.

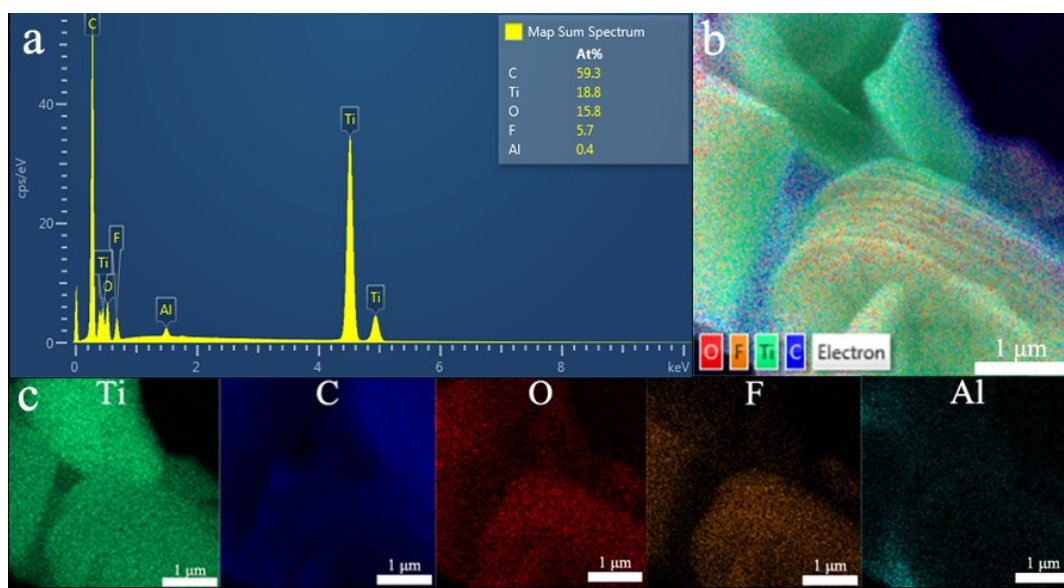


Fig. S2. (a) EDS mapping of Ti_2CT_x . (b-c) EDS elemental mapping of Ti_2CT_x .

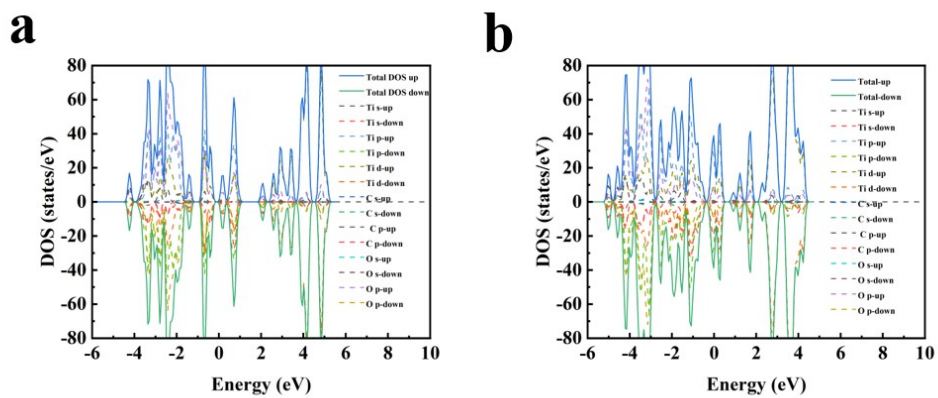


Fig. S3. DOS diagrams for (a) Ti_2CT_x and (b) $\text{Ti}_3\text{C}_2\text{T}_x$.

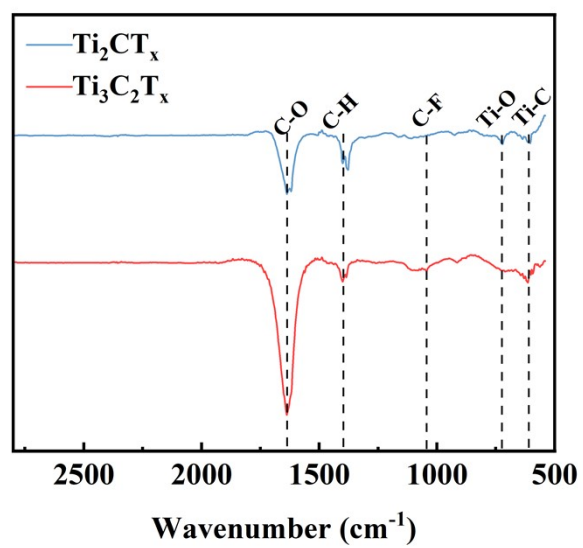


Fig. S4. FTIR spectroscopy of MXenes.

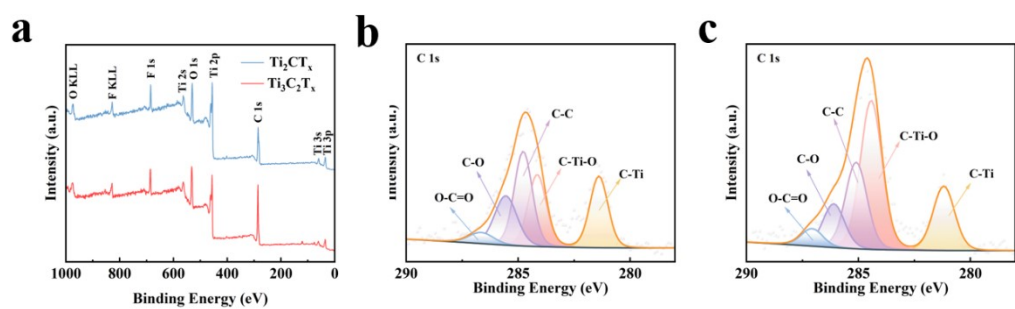


Fig. S5. (a) MXenes full spectra of XPS, C 1s XPS of (b) Ti_2CT_x and (c) $\text{Ti}_3\text{C}_2\text{T}_x$.

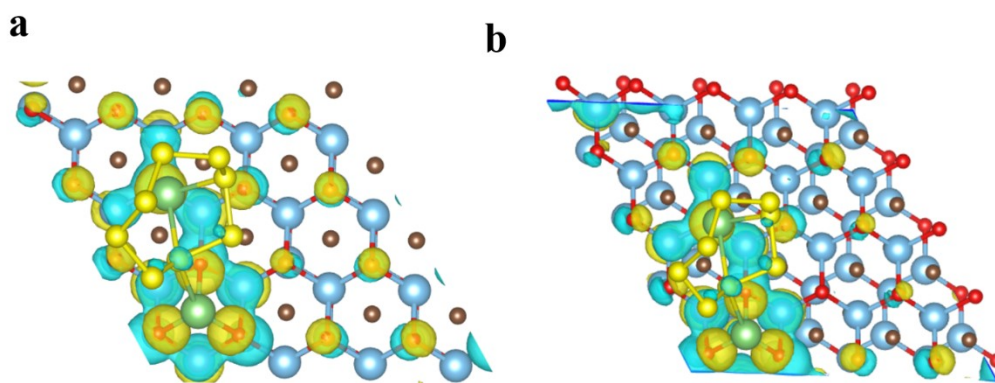


Fig. S6. Top view of the differential charge density for the adsorption of Li_2S_8 on (a) Ti_2CO_2 and (b) $\text{Ti}_3\text{C}_2\text{O}_2$ (Yellow represents electron accumulation; indigo blue represents the electron depletion region)

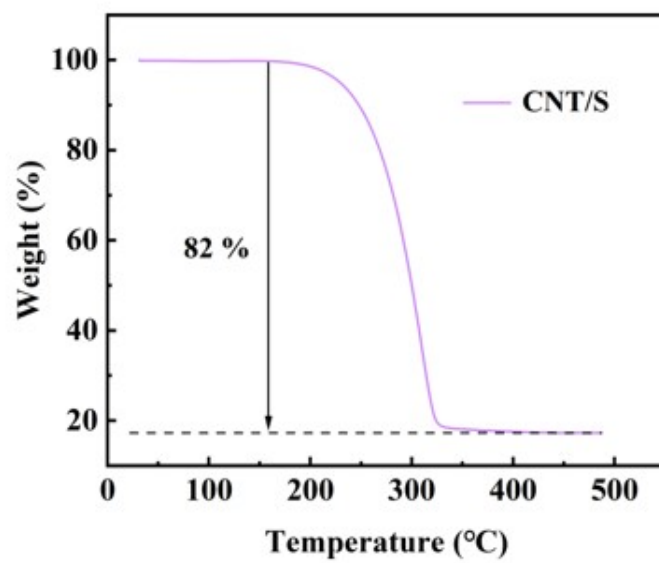


Fig. S7. TGA curves of CNT/S.

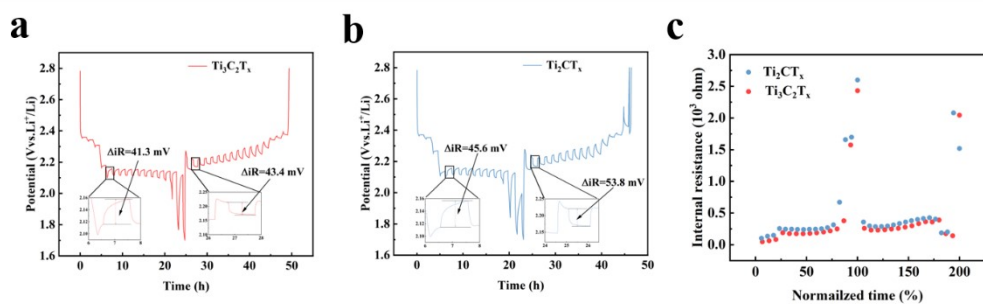


Fig. S8. GITT of (a) $\text{Ti}_3\text{C}_2\text{T}_x$ (b) Ti_2CT_x and corresponding (c) Internal resistance.

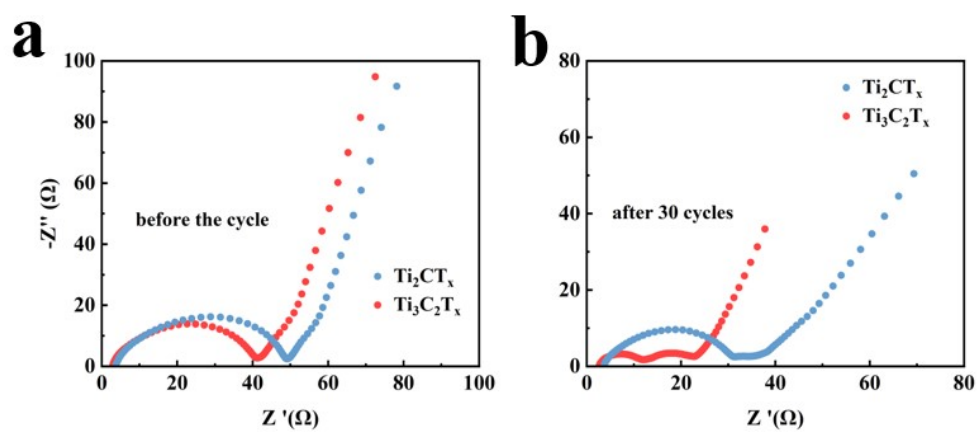


Fig. S9. EIS characterisation of MXenes (a) before the cycle and (b) after 30 cycles.

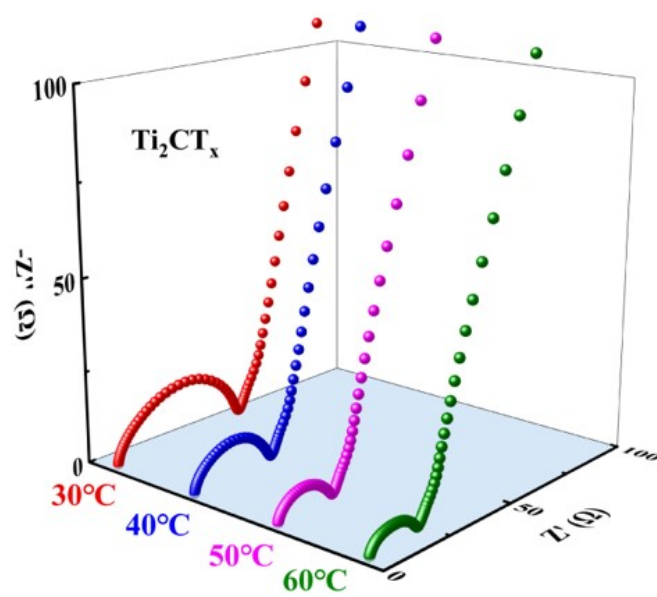


Fig. S10. EIS of Ti_2CT_x at different temperatures.

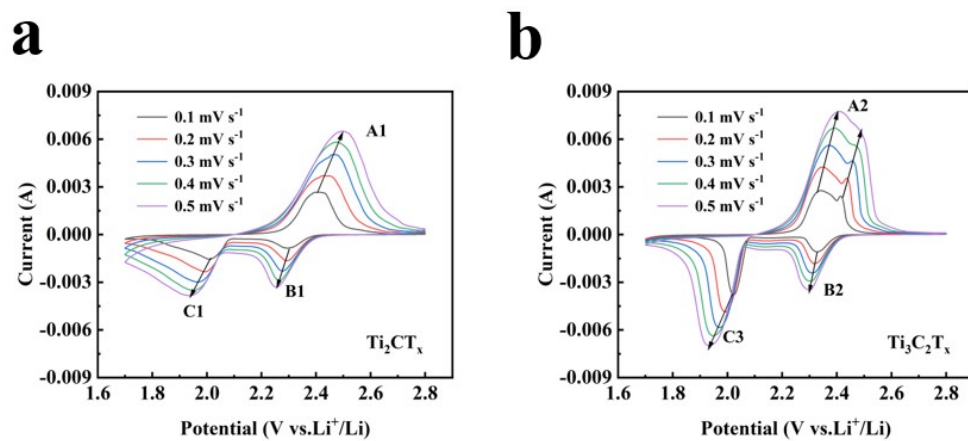


Fig. S11. CV of (a) Ti_2CT_x and (b) $\text{Ti}_3\text{C}_2\text{T}_x$ at different scan rates.

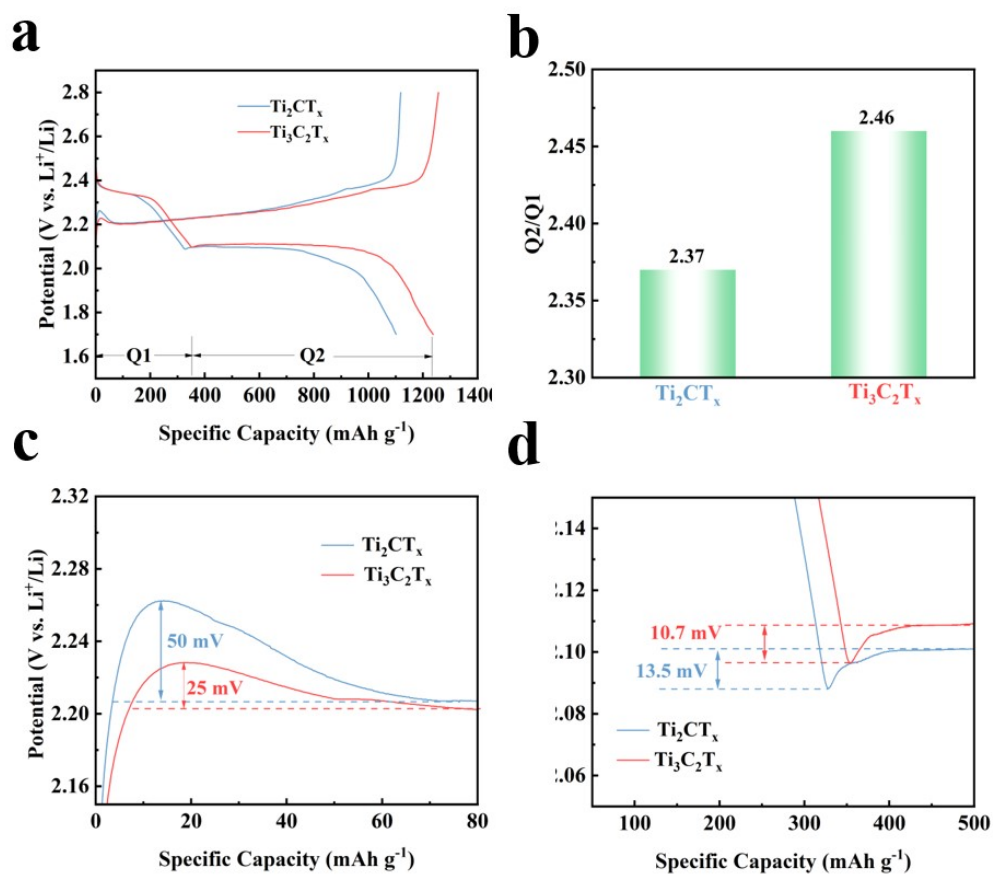


Fig. S12. (a) GCD curves of MXenes. (b) $Q2/Q1$. (c-d) GCD profiles corresponding enlarged parts.

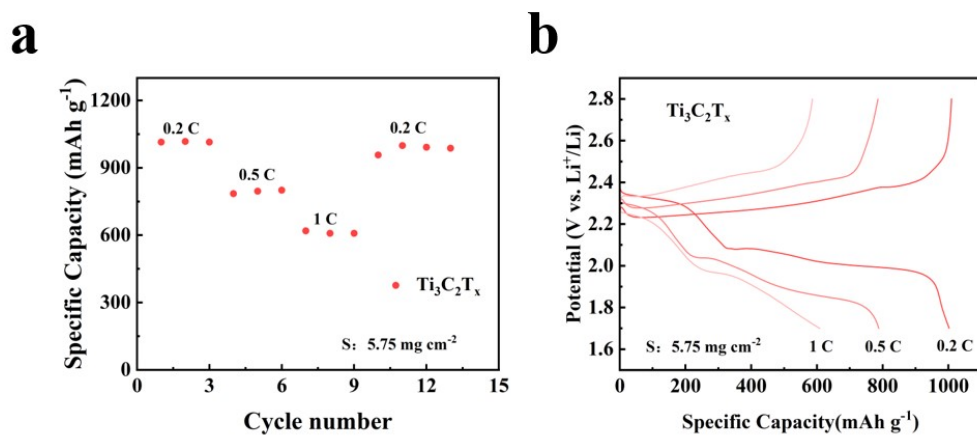


Fig. S13. (a) High-sulfur loading rate performance map of cells with Ti₃C₂T_x. (b) corresponding GCD curves

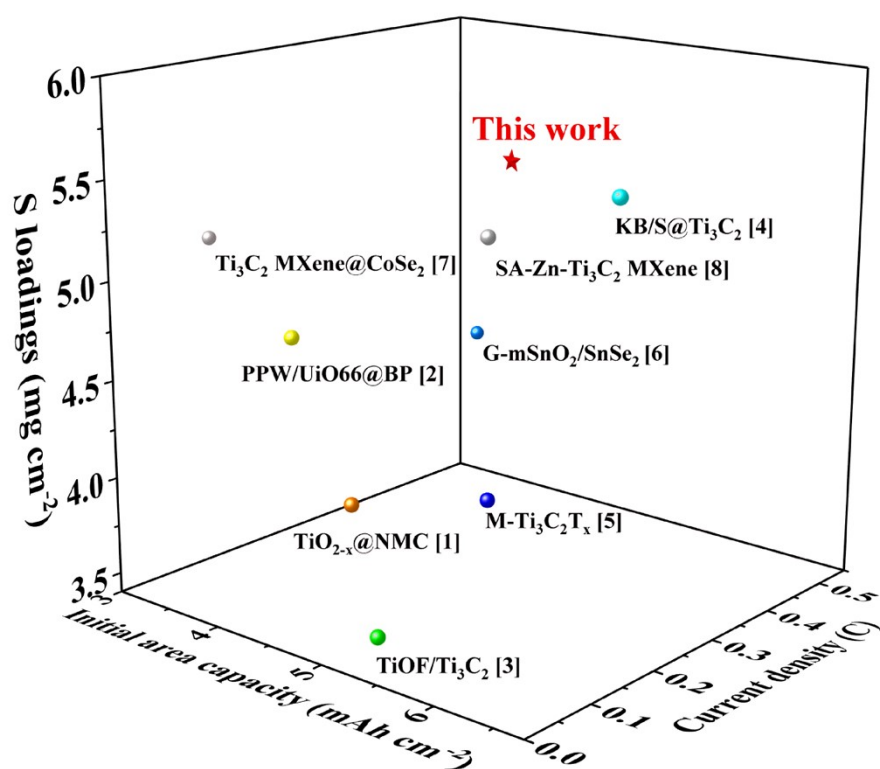


Fig. S14. 3D Performance Comparison Chart

Table S1 O/Ti and F/Ti ratios of MXenes in EDS and XPS.

	EDS:O/Ti	EDS:F/Ti	XPS:O/Ti	XPS:F/Ti
Ti ₃ C ₂ T _x	0.68	0.37	1.32	0.37
Ti ₂ CT _x	0.84	0.30	0.87	0.34

Table S2 Cycling and rate performance comparison of $\text{Ti}_3\text{C}_2\text{T}_x$ with commercial carbon-based or TiO_2 -coated separators.

Materials	Sulfur loading (mg cm^{-2})	Specific capacity (mAh g^{-1})	Cycle life (cycles)	Ref.
$\text{Ti}_3\text{C}_2\text{T}_x$	1.4	0.1 C-1263	200/2 C	This work
		0.2 C-1110		
		0.5 C-970		
		1 C-837		
		2 C-734		
$\text{TiO}_2@\text{p-c-2}$	0.8	0.2 C-1200	100/0.2 C	Ref.9
		0.5 C-951		
		1 C-866		
		2 C-772		
$\text{NCF}@\text{TiO}_2$	1.6	0.05 C-1238	200/0.5 C	Ref.10
		0.1 C-912		
		0.2 C-805		
		0.5 C-720		
		1 C-666		
$\text{NC}@\text{CoNiFe/CNTs}$	2.8	0.1 C-1047	240/ 0.5 C	Ref.11
		0.2 C-909		
		0.5 C-856		
		1 C-793		
		2 C-697		
CNFs/CoS_{2-x}	1.2~1.3	0.1 C-1167	350/1 C	Ref.12
		0.2 C-1074		
		0.5 C-936		
		1 C-839		
		2 C-752		
CNT/HEA-NC	1.5	0.1 C-1060	300/1 C	Ref.13
		0.2 C-922		
		0.5 C-819		
		1 C-744		
		2 C-676		
1T-rich $\text{MoS}_2@\text{PC}$	1.3	0.1 C-1076	500/1 C	Ref.14
		0.2 C-899		
		0.5 C-800		
		1 C-710		
		2 C-619		
		3 C-560		

Table S3 Comparison of electrochemical performance between $\text{Ti}_3\text{C}_2\text{T}_x$ separators and other MXene/S cathode batteries.

Materials	Application	Sulfur loading (mg cm^{-2})	Specific capacity (mAh g^{-1})/Rate (C)	Ref.
$\text{Ti}_3\text{C}_2\text{T}_x$	Separator	1.4	820 / 1 C	This work
A/R- TiO_2 @Ni-N-MXene-S	Cathode	1.5	694 / 1 C	Ref.15
MXene/1T-2H MoS_2 -C-S	Cathode	1.0	797 / 1.5 C	Ref.16
OV- TnQDs@PCN/S	Cathode	2.2	718 / 1 C	Ref.17
S@MXene- CoSe_2	Cathode	1.0	805 / 1 C	Ref.18
S/ Ti_3C_2 @N- CNTs	Cathode	1.2	820 / 1 C	Ref.19
$\text{TiO}_2/\text{H-Ti}_3\text{C}_2\text{T}_x$	Cathode	1.0	800 / 1 C	Ref.20

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