

Electronic supplementary information (ESI)

Mechanism of the generation of ultra-stable radicals in fast photochromic naphthalenediimide-based coordination polymers

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1. Figure

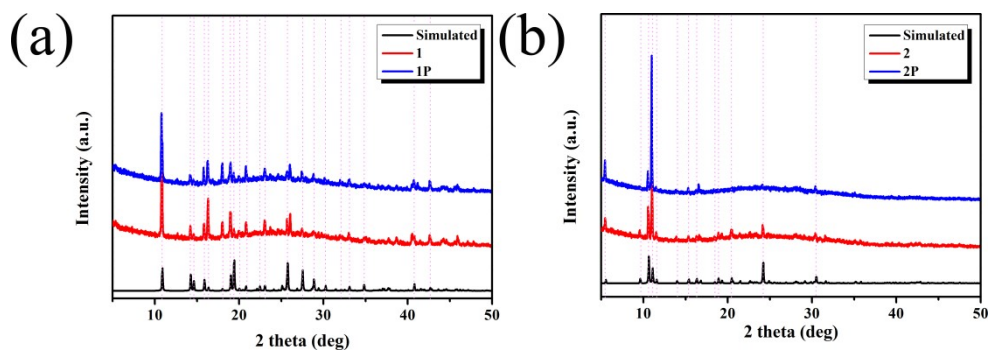


Fig. S1. (a) PXRD patterns of **1** and **1P** at room temperature. (b) PXRD patterns of **2** and **2P** at room temperature.

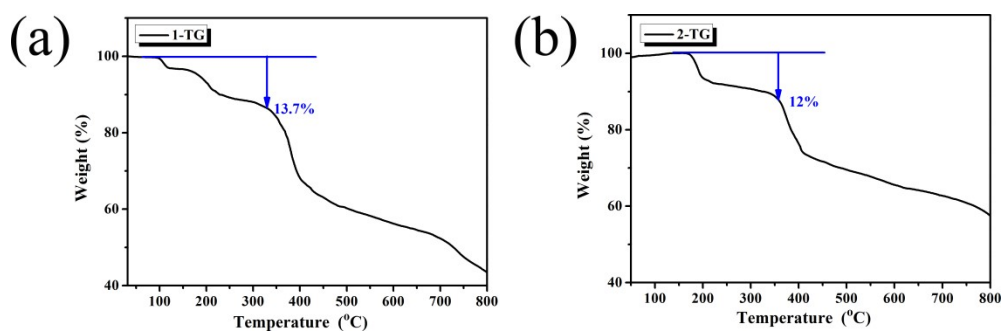


Fig. S2. (a) TGA curve of **1**. (b) TGA curve of **2**.

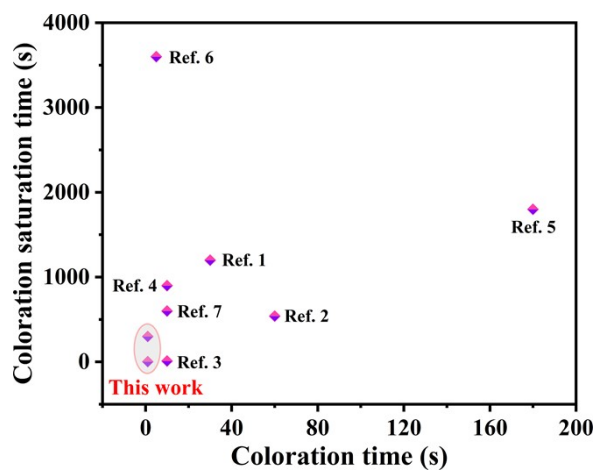


Fig. S3. Summary of photochromic CPs in the literature compared to this work.

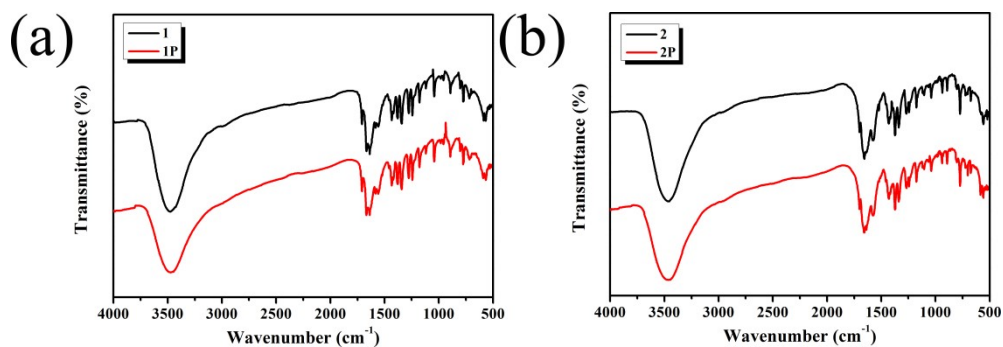


Fig. S4. (a) FT-IR spectra of **1** and **1P**. (b) FT-IR spectra of **2** and **2P**.

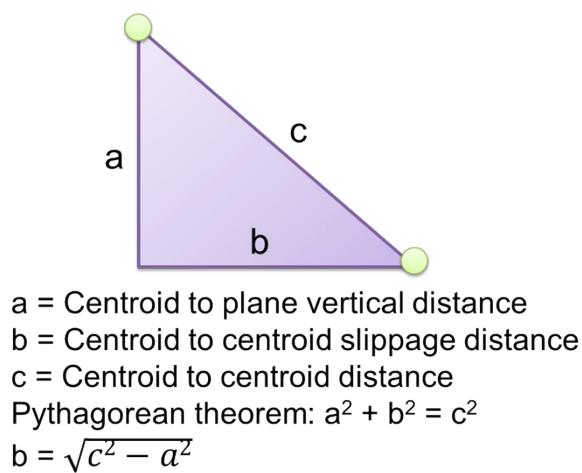


Fig. S5. The calculation principle of DORS

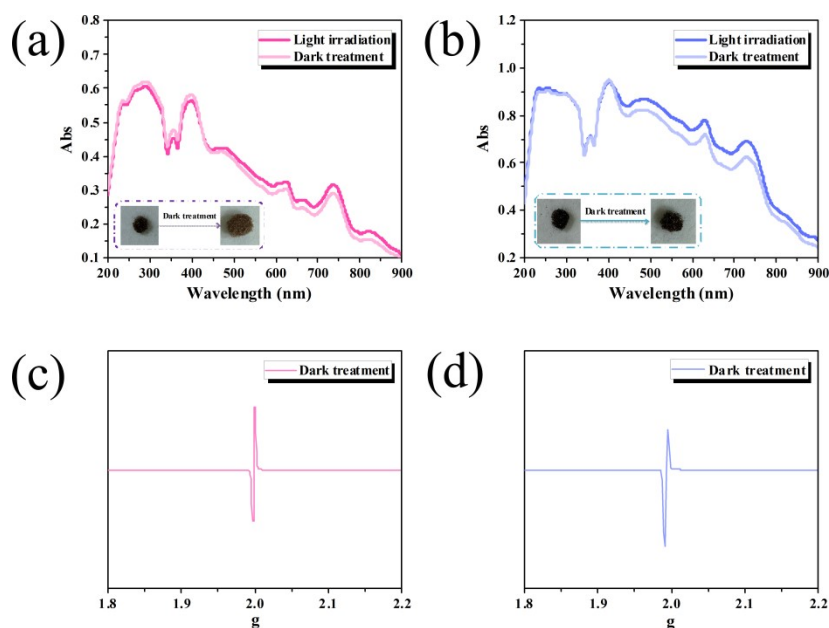


Fig. S6. (a) The UV-vis absorption spectra of **1** after dark treatment (insert: electronic picture of **1** after dark treatment). (b) The UV-vis absorption spectra of **2** after dark treatment (insert: electronic picture of **2** after dark treatment). (c) EPR spectra of **1** after dark treatment. (d) EPR spectra of **2** after dark treatment.

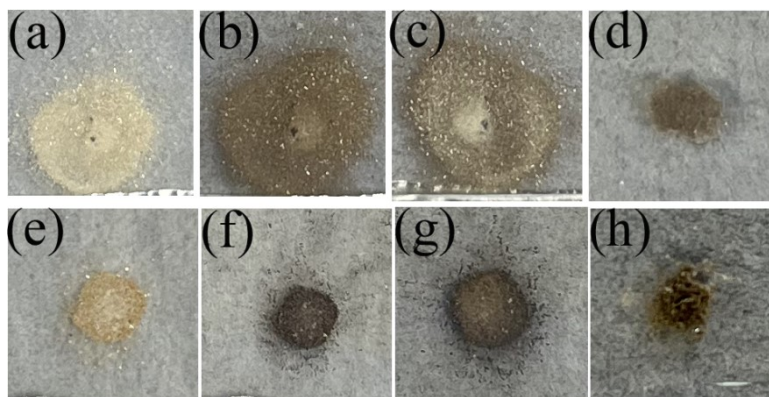


Fig. S7. (a) Primary sample of **1**. (b) Electronic picture of **1** after UV light irradiation. (c) Back side of **1**. (d) Electronic picture of **1** after grind. (e) Primary sample of **2**. (f) Electronic picture of **2** after UV light irradiation. (g) Back side of **2**. (h) Electronic picture of **2** after grind.

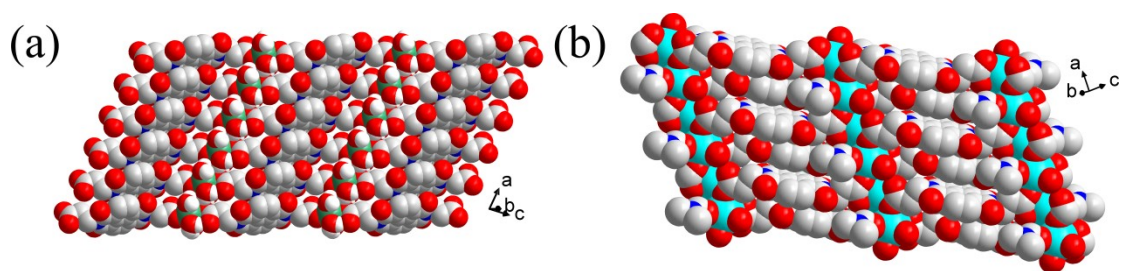


Fig. S8. (a) the space-filling view of **1**. (b) the space-filling view of **2**.

2. Table

Table S1. Crystallographic data and refinement parameters of **1** and **1P**.

Compound	1	1P
CCDC Code	2394240	2394241
Empirical formula	$\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_{12}\text{Mg}$	$\text{C}_{20}\text{H}_{20}\text{N}_2\text{O}_{12}\text{Mg}$
Formula weight	504.69	504.69
Temperature (K)	293(2)	293(2)
Crystal system	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	5.0535(12)	5.057(3)
<i>b</i> (Å)	6.3853(16)	6.384(3)
<i>c</i> (Å)	16.505(4)	16.511(8)
α (°)	81.549(6)	81.545(12)
β (°)	82.117(6)	82.082(12)
γ (°)	78.223(7)	78.212(8)
<i>V</i> (Å ³)	512.6(2)	512.9(4)
<i>Z</i>	1	1
<i>D_c</i> (g cm ⁻³)	1.635	1.634
μ (mm ⁻¹)	0.163	0.163
<i>F</i> (000)	262.0	262.0

Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
ϑ range ($^{\circ}$)	6.568 to 54.992	6.568 to 56.728
Reflections collected	7271	8151
Unique reflections	2358	2546
R_{int}	0.0498	0.0450
Data/restraints/parameters	2358/0/162	2546/0/162
Goodness-of-fit on F^2	1.028	1.010
R_1/wR_2 , [$I \geq 2\sigma(I)$] ^{a,b}	0.0538/0.1049	0.0472/0.1062
R_1/wR_2 , (all data)	0.1051/0.1206	0.0852/0.1218
$\Delta\rho_{\text{max}}/\Delta\rho_{\text{min}}$ (e $\text{\AA}^{-3}$)	0.24/-0.28	0.29/-0.29

$$^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Table S2. Crystallographic data and refinement parameters of **2** and **2P**.

Compound	2	2P
CCDC Code	2394242	2394243
Empirical formula	C ₂₃ H ₂₁ N ₃ O ₁₀ Sr	C ₂₃ H ₂₁ N ₃ O ₁₀ Sr
Formula weight	587.05	587.05
Temperature (K)	293(2)	293(2)
Crystal system	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1
<i>a</i> ($\text{\AA}$)	8.0182(18)	8.019(3)
<i>b</i> ($\text{\AA}$)	9.2611(18)	9.258(3)
<i>c</i> ($\text{\AA}$)	16.060(4)	16.090(6)
α ($^{\circ}$)	84.046(6)	84.054(10)
β ($^{\circ}$)	85.142(5)	85.034(10)
γ ($^{\circ}$)	84.613(4)	84.591(9)
<i>V</i> ($\text{\AA}^3$)	1177.5(5)	1179.3(7)
<i>Z</i>	2	2

D_c (g cm ⁻³)	1.656	1.653
μ (mm ⁻¹)	2.354	2.351
F (000)	596.0	596.0
Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
ϑ range (°)	4.438 to 56.658	4.44 to 56.694
Reflections collected	18958	17541
Unique reflections	5864	5865
R_{int}	0.0425	0.0421
Data/restraints/parameters	5864/1/339	5865/1/339
Goodness-of-fit on F^2	1.056	1.031
R_1/wR_2 , [$I \geq 2\sigma(I)$] ^{a,b}	0.0307/0.0719	0.0343/0.0750
R_1/wR_2 , (all data)	0.0384/0.0747	0.0496/0.0797
$\Delta\rho_{\text{max}}/\Delta\rho_{\text{min}}$ (e Å ⁻³)	0.48/-0.48	0.40/-0.46

$$^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Table S3. Selected bond lengths (Å) and angles (°) of **1**.

Compound 1			
Mg1-O2	2.0993 (16)	Mg1-O3 ¹	2.0925 (17)
Mg1-O2 ¹	2.0993 (16)	Mg1-O4 ¹	2.0626 (16)
Mg1-O3	2.0925 (17)	Mg1-O4	2.0627 (16)
O2 ¹ -Mg1-O2	180.0	O4 ¹ -Mg1-O2 ¹	90.55 (6)
O3 ¹ -Mg1-O2 ¹	87.52 (7)	O4 ¹ -Mg1-O2	89.45 (7)
O3-Mg1-O2 ¹	92.48 (7)	O4-Mg1-O3	91.15 (7)
O3 ¹ -Mg1-O2	92.48 (7)	O4 ¹ -Mg1-O3 ¹	91.15 (7)
O3-Mg1-O2	87.52 (7)	O4-Mg1-O3 ¹	88.85 (7)
O3 ¹ -Mg1-O3	180.0	O4 ¹ -Mg1-O3	88.85 (7)
O4-Mg1-O2 ¹	89.45 (7)	O4 ¹ -Mg1-O4	180.0

O4-Mg1-O2 90.55 (6)

symmetry codes: ¹1-x, 2-y, 1-z; ²-x, 1-y, -z.

Table S4. Selected bond lengths (Å) and angles (°) of **1P**.

Compound 1P			
Mg1-O2	2.1034 (15)	Mg1-O3 ¹	2.0940 (15)
Mg1-O2 ¹	2.1035 (15)	Mg1-O4 ¹	2.0624 (15)
Mg1-O3	2.0940 (15)	Mg1-O4	2.0624 (15)
O2 ¹ -Mg1-O2	180.00 (5)	O4 ¹ -Mg1-O2 ¹	90.56 (6)
O3 ¹ -Mg1-O2 ¹	87.52 (7)	O4 ¹ -Mg1-O2	89.44 (6)
O3-Mg1-O2 ¹	92.59 (7)	O4-Mg1-O3	91.15 (7)
O3 ¹ -Mg1-O2	92.59 (7)	O4 ¹ -Mg1-O3 ¹	91.15 (7)
O3-Mg1-O2	87.41 (7)	O4-Mg1-O3 ¹	88.85 (7)
O3 ¹ -Mg1-O3	180.0	O4 ¹ -Mg1-O3	88.85 (7)
O4-Mg1-O2 ¹	89.44 (6)	O4 ¹ -Mg1-O4	180.0
O4-Mg1-O2	90.56 (6)		
symmetry codes: ¹ 1-x, 2-y, 1-z; ² -x, 1-y, -z.			

Table S5. Selected bond lengths (Å) and angles (°) of **2**.

Compound 2			
Sr1-O1	2.6039 (16)	Sr1-O4 ¹	2.8056 (16)
Sr1-O2	2.4846 (19)	Sr1-O9 ²	2.6533 (15)
Sr1-O3 ¹	2.6095 (14)	Sr1-O10 ³	2.4958 (14)
Sr1-O3	2.5123 (15)	Sr1-O10 ²	2.6017 (14)
O1-Sr1-O3 ¹	161.74 (5)	O3 ¹ -Sr1-O9 ²	93.04 (5)

O1-Sr1-O4 ¹	150.26 (5)	O3-Sr1-O9 ²	79.34 (5)
O1-Sr1-O9 ²	76.61 (6)	O3-Sr1-O10 ²	128.54 (4)
O2-Sr1-O1	94.86 (7)	O9 ² -Sr1-O4 ¹	110.71 (5)
O2-Sr1-O3	77.22 (6)	O10 ³ -Sr1-O1	79.54 (5)
O2-Sr1-O3 ¹	88.25 (7)	O10 ² -Sr1-O1	74.28 (5)
O2-Sr1-O4 ¹	88.52 (7)	O10 ³ -Sr1-O3	153.24 (5)
O2-Sr1-O9 ²	154.86 (6)	O10 ³ -Sr1-O3 ¹	118.68 (5)
O2-Sr1-O10 ²	151.19 (6)	O10 ² -Sr1-O3 ¹	110.51 (5)
O2-Sr1-O10 ³	78.59 (6)	O10 ³ -Sr1-O4 ¹	72.18 (4)
O3-Sr1-O1	91.25 (5)	O10 ² -Sr1-O4 ¹	88.69 (4)
O3-Sr1-O3 ¹	71.87 (5)	O10 ³ -Sr1-O9 ²	121.85 (4)
O3-Sr1-O4 ¹	118.24 (4)	O10 ² -Sr1-O9 ²	49.45 (4)
O3 ¹ -Sr1-O4 ¹	47.60 (4)	O10 ³ -Sr1-O10 ²	73.27 (5)

symmetry codes: ¹2-x, 1-y, -z; ²-x+1, -y, 1-z; ³x, 1+y, 1-z.

Table S6. Selected bond lengths (Å) and angles (°) of **2P**.

Compound 2P			
Sr1-O1	2.6029 (19)	Sr1-O4 ¹	2.6542 (17)
Sr1-O2	2.481 (2)	Sr1-O9 ²	2.8042 (19)
Sr1-O3	2.4989 (16)	Sr1-O10 ³	2.5151 (17)
Sr1-O3 ¹	2.6027 (16)	Sr1-O10 ²	2.6103 (17)
O1-Sr1-O4 ¹	76.57 (7)	O3-Sr1-O4 ¹	121.91 (5)
O1-Sr1-O9 ²	150.21 (6)	O3 ¹ -Sr1-O9 ²	88.61 (5)
O1-Sr1-O10 ²	161.74 (6)	O3-Sr1-O9 ²	72.10 (5)
O2-Sr1-O1	95.11 (8)	O3-Sr1-O10 ³	153.27 (6)
O2-Sr1-O3 ¹	151.28 (6)	O3 ¹ -Sr1-O10 ²	110.39 (6)
O2-Sr1-O3	78.58 (6)	O3-Sr1-O10 ²	118.65 (5)

O2-Sr1-O4 ¹	154.92 (7)	O4 ¹ -Sr1-O9 ²	110.77 (6)
O2-Sr1-O9 ²	88.31 (7)	O10 ³ -Sr1-O1	91.12 (6)
O2-Sr1-O10 ³	77.36 (6)	O10 ³ -Sr1-O3 ¹	128.42 (5)
O2-Sr1-O10 ²	88.10 (7)	O10 ² -Sr1-O4 ¹	93.01 (5)
O3-Sr1-O1	79.58 (6)	O10 ³ -Sr1-O4 ¹	79.18 (5)
O3 ¹ -Sr1-O1	74.32 (6)	O10 ³ -Sr1-O9 ²	118.42 (5)
O3-Sr1-O3 ¹	73.30 (6)	O10 ² -Sr1-O9 ²	47.64 (5)
O3 ¹ -Sr1-O4 ¹	49.49 (5)	O10 ³ -Sr1-O10 ²	72.01 (6)

symmetry codes: ¹1-x, 2-y, -z; ²-x, 1-y, 1-z; ³x, 1+y, z-1.

Table S7. Hydrogen bonds of **1** (Å and °).

Compound 1				
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O3-H3A...O1 ¹	0.85	1.93	2.760 (3)	163
O3-H3B...O2 ²	0.85	2.24	3.073 (3)	166
O4-H4A...O1	0.85	1.87	2.653 (2)	153
O4-H4B...O2 ³	0.85	2.24	3.047 (2)	159

Symmetry codes: ¹1-x, 1-y, 1-z; ²-x, 2-y, 1-z; ³1+x, y, z.

Table S8. Hydrogen bonds of **1P** (Å and °).

Compound 1P				
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O3-H3A...O1 ¹	0.85	1.93	2.756 (2)	164
O3-H3B...O2 ²	0.85	2.25	3.075 (2)	165
O4-H4A...O1	0.85	1.87	2.651 (2)	153
O4-H4B...O2 ³	0.85	2.23	3.043 (2)	160

Symmetry codes: ¹1-x, 1-y, 1-z; ²-x, 2-y, 1-z; ³1+x, y, z.

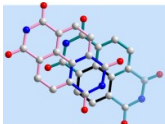
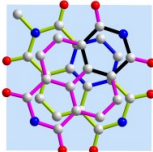
Table S9. Hydrogen bonds of **2** (Å and °).

Compound 2				
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O1-H1A...O4 ¹	0.85	2.02	2.838 (2)	161
O1-H1B...O7 ²	0.85	2.07	2.908 (2)	170
Symmetry codes: ¹ 1-x, y, z; ² 1-x, 1-y, 1-z.				

Table S10. Hydrogen bonds of **2P** (Å and °).

Compound 2P				
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O1-H1A...O9 ¹	0.85	2.02	2.837 (3)	161
O1-H1B...O6 ²	0.85	2.06	2.907 (3)	171
Symmetry codes: ¹ 1+x, 1+y, z-1; ² 1-x, 1-y, -z.				

Table S11. Summary of photochromic parameters in the literature compared to this work.

Compound	Pack	Light source	Ring-slippage (Å)	d _{π-π} (Å)	Ref.
[Zn(NDI-ATZ)(DMF) ₂]		Sunlight	0.854	3.778	8
[Ca(BIPNDI) _{0.5} (DMF) ₂]		Xe lamp >420 nm	0.877	3.778	9

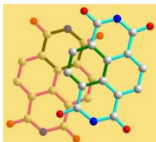
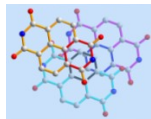
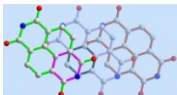
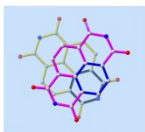
RC		UV	2.010	3.887	10
2		Xe lamp >420 nm	1.172	3.607	This work
1		Xe lamp >420 nm	1.483	3.683	This work
MC		Xe lamp >420 nm	0.759	3.587	11

Table S12. The change of single-crystal parameters before and after UV light irradiation.

Parameters	1	1P	2	2P
a (Å)	5.054	5.057	8.018	8.019
b (Å)	6.385	6.384	9.261	9.258
c (Å)	16.506	16.511	16.060	16.090
α (°)	81.549	81.545	84.046	84.054
β (°)	82.117	82.082	85.142	85.034
γ (°)	78.223	78.212	84.613	84.591
V (Å ³)	512.6	512.9	1177.5	1179.3
π - π interactions (Å)	3.683	3.685	3.607 3.727	3.615 3.730

References

1. C. Fu, G. S. Zhang and H. Y. Wang, et al. Different photochromic properties induced by lone pair- π interactions with varying strengths in two stereocontrolled self-assembly isomeric coordination polymers, *CrystEngComm*, 2018, **20**, 6821–6827.
2. J. J. Liu, Z. J. Wang and S. B. Xia, et al. Photochromic and photocontrolled luminescence properties of two metalorganic frameworks constructed from a

naphthalene diimide derivative, *Dyes Pigm.*, 2020, **172**, 107856.

3. P. F. Hao, B. H. Gao and G. P. Li, et al. Ultrafast visible-light photochromic properties of naphthalenediimide-based coordination polymers for the visual detection/filtration of blue light, *Inorg. Chem. Front.*, 2022, **9**, 2852–2861.

4. P. F. Hao, H. H. Zhu and Y. Pang, et al. The impact of positional isomerism on electronic and photochromic properties of 1-D zinc-based naphthalene diimide coordination polymers, *CrystEngComm*, 2020, **22**, 3371–3377.

5. H. H. Zhu, P. F. Hao and Q. Shen, et al. The modulation effect of electron-rich solvents on the supramolecular networks and photochromic properties of naphthalene diimide molecules, *CrystEngComm*, 2021, **23**, 3356–3363.

6. G. P. Li, H. H. Zhu and P. F. Hao, et al. The modulation effect of carboxylic acid ligands on the electron transfer photochromism of NDI derived coordination polymers, *Dalton Trans.*, 2021, **50**, 15153–15161.

7. P. F. Hao, H. H. Zhu and Y. Pang, et al. Positional isomerism controlled electronic and photochromic properties of naphthalene diimide-based chlorozincate hybrids, *Cryst. Growth Des.*, 2020, **20**, 345–351.

8. Y. X. Xie, W. N. Zhao and G. C. Li, et al. A naphthalenediimide-based metal–organic framework and thin film exhibiting photochromic and electrochromic properties, *Inorg. Chem.*, 2016, **55**, 549–551.

9. M. H. You, M. H. Li and H. H. Li, et al. The impact of metal cations on the photochemical properties of hybrid heterostructures with infinite alkaline-earth metal oxide clusters, *Dalton Trans.*, 2019, **48**, 17381–17387.

10. S. M. Zhang, P. F. Hao and Y. F. Zhang, et al. The simultaneous modulation effect of *N*-substituents on the photochromic and electrochromic properties of naphthalenediimide-based coordination polymers, *Inorg. Chem. Front.*, 2024, **11**, 1226–1237.

11. L. Qin, W. N. Zhao and G. J. Yu, et al. A zinc-organic coordination polymer of glycine-functionalized naphthalenediimide ligand, *Inorg. Chem. Commun.*, 2013, **34**, 47–50.