Supplementary Information

Cumulative 2c-2e vs 3c-2e π -Back Donations: Conformation

Preferences in Germylene Gold(I) Complexes

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1. Experimental Procedures

General Procedures: All reactions were performed under an atmosphere of nitrogen by using standard Schlenk or dry box techniques. Solvents are used as anhydrous solvents. ¹H, ¹³C, and ²⁹Si NMR spectra were obtained with a Bruker AV 400 instrument at 400 MHz (¹H NMR), and 101 MHz (¹³C NMR), as well Bruker AV 500 instrument at 500 MHz (¹H NMR), 126 MHz (¹³C NMR), 99 MHz (²⁹Si NMR) and 471 MHz (¹⁹F NMR) at 298 K. The ¹H and ¹³C NMR chemical shifts were referenced to residual ¹H and ¹³C of the solvents: C₆D₆ (¹H δ 7.16 and ¹³C δ 128.0) and CDCl₃ (¹H δ 7.26 and ¹³C δ 77.16). NMR multiplicities are abbreviated as follows: s = singlet, brs = broad singlet, d = doublet, dt = doublet of triplets, t = triplet, and m = multiplet. Coupling constants *J* are given in Hz. **1** was synthesized according to literature procedures.⁶¹ **2a** was synthesized according to literature procedures.⁶⁰

Synthesis of 2b

NHCAuCl (312 mg, 0.50 mmol) was added to a 25 mL tetrahydrofuran solution of dialkylgermylene **1** (200 mg, 0.48 mmol) at room temperature. After stirring for 10 hours, the solvent was removed in vacuo, and the residue was washed with *n*-hexane to afford the germylgold **2b** (449 mg) as a white powder in 90% yield.: ¹H NMR (400 MHz, C₆D₆): δ = 7.28-7.24 (t, 2H, Ar-*H*), 7.11-7.13 (d, 4H, Ar-*H*), 6.33 (s, 2H, C*H*), 2.61 (m, 4H, C*H*(CH₃)₂), 2.35 (m, 2H, C*H*₂), 2.04 (m, 2H, C*H*₂), 1.50 (d, *J* = 1.50 Hz, 12H, CH(CH₃)₂), 1.05 (d, *J* = 1.05 Hz, 12H, CH(CH₃)₂), 0.48 (s, 18H, Si(CH₃)₃), 0.15 (s, 18H, Si(CH₃)₃); ¹³C NMR (126 MHz, C₆D₆): δ = 200.54 (*C*^{carbene}), 145.50 (*C*H), 135.05 (*C*H), 130.81 (*C*H), 124.67 (*C*H), 123.29 (*C*H), 35.57 (*C*^q), 29.15 (*C*H₂), 25.03 (*C*H), 24.73 (*C*H₃), 24.49 (*C*H₃), 5.41 (Si(*C*H₃)₃),4.93 (Si(*C*H₃)₃); ²⁹Si NMR (99 MHz, C₆D₆): δ = 3.62, 1.52. HRMS (ESI) m/z: Calcd for C₄₃H₇₆AuGeN₂Si₄: 1003.3963 [(M-Cl)]⁺; Found: 1003.3958.

Synthesis of 3

2a (200 mg, 0.27 mmol) was reacted with an equimolar amount of AgAl(OC(CF₃)₃)₄ (290 mg, 0.27 mmol) in chloroform at room temperature for 30 minutes. The reaction mixture gradually changed from colorless to brownish-yellow, forming a brown precipitate. After filtration, concentration, and recrystallization at -35° C, white solid **3** was obtained (160 mg, 31% yield): ¹H NMR (500 MHz, C₆D₆): δ = 2.55 (s, 8H, CH₂), 0.31(s, 72H, Si(CH₃)₃); ¹³C NMR (126 MHz, C₆D₆): δ = 62.22 (C^q), 36.12 (CH₂), 2.83 (Si(CH₃)₃); ²⁹Si NMR (99 MHz, C₆D₆) δ = 0.62; ¹⁹F NMR (471 MHz, C₆D₆) δ = -75.38. HRMS (ESI) m/z: Calcd for C₃₂H₈₀AuGe₂Si₈: 1033.2503 [(M+H)]⁺; Found: 1034.2472.

Synthesis of 4

AgAl(OC(CF₃)₃)₄ (207 mg, 0.19 mmol) was added to a solution of **3b** (200 mg, 0.19 mmol) in CHCl₃. The mixture was stirred at room temperature for 30 minutes, turning brownish-yellow with the formation of a gray silver chloride precipitate. After filtration, the filtrate was recrystallized at -35° C to yield white solid **5** (330 mg, 87% yield): ¹H NMR (400 MHz, THF-*d*₈): δ = 7.62-7.58 (m, 4H, Ar), 7.45 (d, 2H, *J* = 8 Hz, Ar), 3.39 (d, 2H, *J* = 4 Hz, C*H*), 2.67-2.63 (m, 4H, C*H*), 2.40 (s, 4H, C*H*₂), 1.36 (d, *J* = 1.31 Hz, 12H, CH(C*H*₃)₂), 1.26 (d, *J* = 1.31 Hz, 12H, CH(C*H*₃)₂), 0.05 (s, 36H, Si(C*H*₃)₃); ¹³C NMR (126 MHz, THF-*d*₈): δ = 189.55 (*C*^{carbene}), 144.44 (*C*^q), 130.09 (*C*H), 124.85 (*C*^q), 123.58 (*C*^q), 121.76 (*C*H) ,118.85 (*C*H), 34.67 (*C*^q), 27.91 (*C*H₂), 22.25 (*C*H₃), 0.90 (Si(*C*H₃)₃); ²⁹Si NMR (99 MHz, THF-*d*₈): δ = 0.04. HRMS (ESI) m/z: Calcd for C₄₃H₇₆AuGeN₂Si₄: 1003.3963 [(M-Al(OR)₄]⁺; Found: 1003.3958.

Synthesis of 5

To a solution of NaHMDS (770 mg, 4.1 mmol) in toluene (10 mL) at low temperature, $GeCl_2$ ·dioxane (500 mg, 2.1 mmol) was added. The reaction mixture was stirred for 2 hours, turning gradually yellow.⁶⁵ After filtration and solvent removal

under vacuum, **5** was isolated (793 mg, 96% yield) as ayellow oil; ¹H NMR (400 MHz, C₆D₆): $\delta = 0.32$ (s, 36H, Si (CH₃)₃).

Synthesis of 6a

PMe₃AuCl (157 mg, 0.50 mmol) was added to a solution of **5** (200 mg, 0.50 mmol) in toluene and stirred at room temperature for 12 hours. The resulting solution was filtered, concentrated under reduced pressure, and recrystallized at room temperature to afford colorless solid **6a** (339 mg, 95% yeild); ¹H NMR (400 MHz, C₆D₆): $\delta = 0.68$ (s, 36H, Si(CH₃)₃), 0.43-0.45 (d, J = 0.44 Hz, 9H, P(CH₃)₃); ¹³C NMR (126 MHz, C₆D₆): $\delta = 6.90$ (Si(CH₃)₃), 14.76 (d, J = 37.8 Hz, P(CH₃)₃); ³¹P NMR (162 MHz, C₆D₆): $\delta =$ 7.99. HRMS (ESI) m/z: Calcd for C₁₅H₄₅AuGeN₂PSi₄Cl: [M]⁺ 702.0963; Found: [C15H45N2AuGeSi4]⁺ 636.1716.

Synthesis of 6b

NHCAuCl (316 mg, 0.50 mmol) was slowly added to a toluene solution of **5** (200 mg, 0.50 mmol) and stirred at room temperature for 12 hours. The resulting mixture was filtered, concentrated, and recrystallized at -35 °C to afford colorless solid **6b** (480 mg, 93% yeild): ¹H NMR (400 MHz, C₆D₆): $\delta = 6.26$ (s, 2H, CH), 2.62 (m, 4H, CH(CH₃)₂), 1.45 (d, J = 1.44 Hz, 12H, CH(CH₃)₂), 1.02 (d, J = 1.01 Hz, 12H, CH(CH₃)₂), 0.44 (s, 36H, Si(CH₃)₃); ¹³C NMR (126 MHz, C₆D₆): $\delta = 197.12$ (C^{carbene}), 145.34 (CH), 134.97 (C^q), 131.05 (CH), 124.73 (CH), 123.59 (C^q), 29.02 (CH), 24.70 (CH₃), 24.26 (CH₃), 6.93 (Si(CH₃)₃); ²⁹Si NMR (99 MHz, CD₃CN): $\delta = 1.30$. HRMS (ESI) m/z: Calcd for C₁₂H₃₆AuClGeN₂Si₄: 626.0522 [(M-NHC)]⁺; Found: 626.2795.

Synthesis of 7

An chloroform solution of **6b** (200 mg, 0.19 mmol) was treated with one equivalent of $AgAl(OC(CF_3)_3)_4$ (212 mg, 0.19 mmol) at room temperature for 30 minutes. The reaction mixture turned brown-yellow, accompanied by the formation of a gray silver chloride precipitate. Filtration of the reaction mixture, followed by

recrystallization at -35 °C, afforded colorless crystals of **8** (330 mg, 78% yeild): ¹H NMR (400 MHz, CDCl₃): δ = 7.22 (s, 2H, CH), 2.52 (m, 4H, CH(CH₃)₂), 1.32 (d, J = 1.32 Hz, 12H CH(CH₃)₂), 1.22 (d, J = 1.21 Hz, 12H, CH(CH₃)₂), -0.44 (s, 18H, Si(CH₃)₃); ¹³C NMR (126MHz, CDCl₃): δ = 187.22 (*C*^{carbene}), 145.36 (*C*H), 133.53 (*C*^q), 131.73 (*C*H), 125.10 (*C*H), 122.61 (*C*^q), 120.28 (*C*H), 117.95 (*C*H), 31.78 (*C*^q), 29.05 (*C*H), 24.73 (*C*H₃), 24.11 (*C*H₃), 5.25 (Si(*C*H₃)₃); ²⁹Si NMR (99 MHz, CDCl₃): δ = 6.77; ¹⁹F NMR (471 MHz, CDCl₃): δ = 75.92. HRMS (ESI) m/z: Calcd for C₃₉H₇₂AuGeN₄Si₄: 979.3712 [(M-Al(OR)₄)]⁺; Found: 973.5372; Calcd for C₁₆O₄AlF₃₆: 966.9037 [(M-Al(OR)₄)]⁻; Found: 966.9017

Synthesis of 8

AgBF₄ (72 mg, 0.38 mmol) and **6b** (400 mg, 0.38 mmol) were mixed and stirred in chloroform at room temperature for 30 minutes. The solution turned from colorless to brown-yellow with the formation of a brown precipitate. After filtration to remove AgCl, the filtrate was cooled to -35 °C to afford colorless crystals of **7** (142mg, 42% yeild); ¹H NMR (400 MHz, CDCl₃): δ = 7.52-7.48 (m, 2H, CH), 7.30-7.26 (m, 4H, CH), 7.22 (s, 2H, CH), 2.53 (m, 4H, CH(CH₃)₂), 1.33 (d, *J* = 1.32 Hz, 12H, CH(CH₃)₂), 1.23 (d, *J* = 1.22 Hz, 12H, CH(CH₃)₂), -0.04 (s, 36H, Si(CH₃)₃); ¹³C NMR (126 MHz, CDCl₃): δ = 195.91 (*C*^{carbene}), 145.64 (CH), 133.75 (*C*^q), 131.02 (CH), 124.48 (CH), 123.72 (*C*^q), 28.98 (CH), 24.80 (CH₃), 24.23 (CH₃), 4.89 (Si(CH₃)₃); ²⁹Si NMR (99 MHz, CDCl₃): δ = 6.77; ¹⁹F NMR (471 MHz, CDCl₃): δ = -109.66. HRMS (ESI) m/z: Calcd for C₃₃H₅₄AuFGeN₃Si₂: 837.2718 [(M-F)]⁺; Found: 837.2608.

2. Crystallographic Details

All crystallographic intensity data was collected using a Rigaku Oxford Diffraction XtaLAB Synergy-S diffractometer equipped with a HyPix-6000HE Hybrid Photon Counting (HPC) detector, and PhotonJet-S microfocus sealed tube X-ray sources for generating Cu K α radiation ($\lambda = 1.54184$ Å). A suitable single crystal identified by microscopy was mounted on a Nylon loop with paratone oil, and then quickly placed onto the instrument. The crystal temperature was held at 173 K using an Oxford Cryosystems CryostreamPlus 800 open-flow N₂ cryostat. Reflections were recorded, indexed and corrected for absorption with the CrysAlis^{pro} software suit.⁶⁸ All structures were solved by intrinsic phasing (ShelXT-2015),^{S1, S2} and refined to convergence by full-matrix least squares methods based on F^2 (SHELX-2018) ^{S3} embedded in the Olex2.^{S4} All non-hydrogen atoms were subjected to anisotropic refinement. The hydrogen atoms were located at calculated positions or found in the Fmap. CCDC: 2401588, 2401589, 2401590, 2401591, 2401592, 2401593 and 2401594 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from the Cambridge Crystallography Data Center via www.ccdc.cam.ac.uk/data_reCquest/cif.

Parameters	2b
Empirical formula	C43H76AuClGeN2Si4
Formula weight	1038.36
Temperature/K	170.00
Crystal system	monoclinic
space group	P2 ₁ /c
a/Å	28.7462(15)
b/Å	17.8390(8)
c/Å	24.1570(11)
$\alpha/^{\circ}$	90
β/°	101.624(2)
$\gamma/^{\circ}$	90
Volume /Å ³	12133.7(10)
Z	4
$\rho_{calc}g/cm^3$	1.464
μ/mm^{-1}	6.429
F(000)	5416.0
Crystal size/mm ³	$0.08 \times 0.07 \times 0.03$
Radiation	$GaK\alpha \ (\lambda = 1.34139)$
2Θ range for data collection/°	5.398 to 121.714
Index ranges	$-37 \le h \le 37, -19 \le k \le 23, -30 \le l \le 31$
Reflections collected	152207
Independent reflections	27806 [R _{int} = 0.0666, R _{sigma} = 0.0528]
Data/restraints/parameters	27806/76/1199
Goodness-of-fit on F ²	1.054
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0578, wR_2 = 0.1547$
Final R indexes [all data]	$R_1 = 0.0795, wR_2 = 0.1683$
Largest diff. peak/hole / e Å ⁻³	2.03/-1.31

Table S1. X-ray Data for 2b.

Parameters	3
Empirical formula	$C_{48}H_{80}AlAuF_{36}Ge_2O_4Si_8$
Formula weight	1998.96
Temperature/K	170.00
Crystal system	Monoclinic
space group	P21/c
a/Å	20.9302(13)
b/Å	33.697(2)
c/Å	22.7129(13)
α/°	90
β/°	94.980(2)
$\gamma^{/\circ}$	90
Volume /Å ³	15958.6(16)
Z	8
$\rho_{calc}g/cm^3$	1.664
μ/mm^{-1}	4.459
F(000)	7936.0
Crystal size/mm ³	0.08 imes 0.06 imes 0.04
Radiation	$GaK\alpha \ (\lambda = 1.34139)$
2Θ range for data collection/°	4.336 to 121.686
Index ranges	$-27 \le h \le 27, -43 \le k \le 43, -29 \le l \le 29$
Reflections collected	280324
Independent reflections	$36620 [R_{int} = 0.0514, R_{sigma} = 0.0333]$
Data/restraints/parameters	36620/1643/2230
Goodness-of-fit on F ²	1.052
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0449, wR_2 = 0.1166$
Final R indexes [all data]	$R_1 = 0.0573, wR_2 = 0.1245$
Largest diff. peak/hole / e Å ⁻³	1.89/-1.15

 Table S2. X-ray Data for 3.

Parameters	4
Empirical formula	C59H76AlAuF36GeN2O4Si4
Formula weight	1970.11
Temperature/K	170.00
Crystal system	Monoclinic
space group	$P2_1/c$
a/Å	10.6882(6)
b/Å	20.2536(11)
c/Å	36.451(2)
$\alpha/^{\circ}$	90
β/°	90.609(2)
$\gamma/^{\circ}$	90
Volume /Å ³	7890.3(8)
Z	4
$\rho_{calc}g/cm^3$	1.658
μ/mm^{-1}	3.881
F(000)	3920.0
Crystal size/mm ³	0.04 imes 0.02 imes 0.01
Radiation	GaKa ($\lambda = 1.34139$)
2Θ range for data collection/°	4.218 to 107.806
Index ranges	$-12 \le h \le 12, -24 \le k \le 24, -43 \le l \le 43$
Reflections collected	63724
Independent reflections	14393 [R _{int} = 0.1210, R _{sigma} = 0.1367]
Data/restraints/parameters	14393/344/993
Goodness-of-fit on F ²	1.068
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0782, wR_2 = 0.1826$
Final R indexes [all data]	$R_1 = 0.1184, wR_2 = 0.2068$
Largest diff. peak/hole / e Å $^{-3}$	1.54/-0.75

 Table S3. X-ray Data for 4.

Parameters	ба
Empirical formula	C ₁₅ H ₄₅ AuClGeN ₂ PSi ₄
Formula weight	701.86
Temperature/K	170.00
Crystal system	triclinic
space group	P-1
a/Å	10.9157(5)
b/Å	11.3209(5)
c/Å	13.9382(7)
α/°	84.0760(10)
β/°	70.022(2)
$\gamma/^{\circ}$	65.5240(10)
Volume /Å ³	1471.82(12)
Z	2
$\rho_{calc}g/cm^3$	1.584
μ/mm^{-1}	9.118
F(000)	696.0
Crystal size/mm ³	$0.07 \times 0.06 \times 0.04$
Radiation	$GaK\alpha \ (\lambda = 1.34139)$
2Θ range for data collection/°	5.876 to 121.306
Index ranges	-14 \leq h \leq 14, -14 \leq k \leq 14, -18 \leq 1 \leq
	18
Reflections collected	21905
Independent reflections	6687 [Rint = 0.0318, Rsigma = 0.0319]
Data/restraints/parameters	6687/65/282
Goodness-of-fit on F ²	1.067
Final R indexes [I>= 2σ (I)]	R1 = 0.0233, $wR2 = 0.0567$
Final R indexes [all data]	R1 = 0.0258, wR2 = 0.0575
Largest diff. peak/hole / e Å ⁻³	1.22/-0.98

Table S4. X-ray Data for 6a.

Parameters	6b
Empirical formula	C53H88AuClGeN4Si4
Formula weight	1198.64
Temperature/K	170.00
Crystal system	triclinic
space group	P-1
a/Å	11.4734(4)
b/Å	13.2785(4)
c/Å	19.7989(6)
α / $_{\circ}$	86.7530(10)
β/°	81.1060(10)
γ/°	84.3230(10)
Volume /Å ³	2962.84(16)
Z	2
$\rho_{calc}g/cm^3$	1.344
μ/mm^{-1}	3.142
F(000)	1236.0
Crystal size/mm ³	$0.29 \times 0.16 \times 0.06$
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/°	4.43 to 50.698
Index ranges	$-13 \le h \le 13, -15 \le k \le 15, -23 \le l \le 22$
Reflections collected	64553
Independent reflections	10641 [$R_{int} = 0.0325$, $R_{sigma} = 0.0222$]
Data/restraints/parameters	10641/1/599
Goodness-of-fit on F ²	1.016
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0225, wR_2 = 0.0545$
Final R indexes [all data]	$R_1 = 0.0237, wR_2 = 0.0553$
Largest diff. peak/hole / e Å ⁻³	2.53/-0.88

Table S5. X-ray Data for 6b.

Parameters	7
Empirical formula	C55H72AlAuF36GeN4O4Si4
Formula weight	1946.06
Temperature/K	250.00
Crystal system	monoclinic
space group	$P2_1/n$
a/Å	17.6176(10)
b/Å	27.1229(15)
c/Å	17.9460(10)
α/°	90
β/°	109.893(2)
γ/°	90
Volume /Å ³	8063.6(8)
Z	4
$\rho_{calc}g/cm^3$	1.603
μ/mm^{-1}	3.797
F(000)	3864.0
Crystal size/mm ³	$0.26 \times 0.15 \times 0.13$
Radiation	GaK α (λ = 1.34139)
2Θ range for data collection/°	5.282 to 121.58
Index ranges	$-22 \le h \le 21, -35 \le k \le 35, -23 \le l \le 23$
Reflections collected	118975
Independent reflections	18565 [$R_{int} = 0.0874$, $R_{sigma} = 0.0730$]
Data/restraints/parameters	18565/702/975
Goodness-of-fit on F ²	1.058
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0651, wR_2 = 0.1773$
Final R indexes [all data]	$R_1 = 0.0883, wR_2 = 0.1955$
Largest diff. peak/hole / e Å ⁻³	1.81/-1.02

 Table S6. X-ray Data for 7.

Parameters	8
Empirical formula	$C_{33}H_{54}AuF_2GeN_3Si_2$
Formula weight	856.55
Temperature/K	100.0(2)
Crystal system	triclinic
space group	P-1
a/Å	10.6937(12)
b/Å	17.289(2)
c/Å	24.936(3)
α/°	70.706(4)
β/°	78.905(4)
γ/°	81.874(5)
Volume /Å ³	4254.9(9)
Z	4
$\rho_{calc}g/cm^3$	1.523
μ/mm^{-1}	6.634
F(000)	1952.0
Crystal size/mm ³	$0.18 \times 0.16 \times 0.08$
Radiation	GaK α ($\lambda = 1.34138$)
2Θ range for data collection/°	3.306 to 113.964
Index ranges	-13 \leq h \leq 13, -21 \leq k \leq 21, -31 \leq l \leq
	30
Reflections collected	85057
Independent reflections	17305 [R _{int} = 0.0887, R _{sigma} = 0.0568]
Data/restraints/parameters	17305/58/871
Goodness-of-fit on F ²	1.058
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0490, wR_2 = 0.1275$
Final R indexes [all data]	$R_1 = 0.0599, wR_2 = 0.1326$
Largest diff. peak/hole / e Å $^{-3}$	2.66/-2.37

 Table S7. X-ray Data for 8.



Fig. S1. Solid structures of 2b. Thermal ellipsoids are shown at 30% probability level. Hydrogen atoms are omitted for clarity.



Fig. S2. Solid structures of **6a**. Thermal ellipsoids are shown at 30% probability level. Hydrogen atoms are omitted for clarity.



Fig. S3. Solid structures of **6b**. Thermal ellipsoids are shown at 30% probability level. Hydrogen atoms are omitted for clarity.



Fig. S4. Solid structures of 3. Thermal ellipsoids are shown at 30% probability level. Hydrogen atoms are omitted for clarity.



Fig. S5. Solid structures of 4. Thermal ellipsoids are shown at 30% probability level. Hydrogen atoms are omitted for clarity.



Fig. S6. Solid structures of 7. Thermal ellipsoids are shown at 30% probability level. Hydrogen atoms are omitted for clarity.

3. Computational Details

All the calculations were performed with the Gaussian 16 program package.^{S7} Geometry optimizations and frequency calculations were performed at the level of MN15-D3BJ/def2svp.^{S8-S11} Based on the optimized structures, we further carried out the single point calculations to obtain better wavefunctions for the principal interacting orbital (PIO) analysis.^{70, 71} In these single-point calculations, all atoms were calculated with def2tzvp basis sets.^{S12-13} To compared with the gas phase results, we performed additional PIO analysis in the presence of SMD solvation models.⁷² The plots of structures and orbitals were displayed by GaussView 6.0.^{S14}

Workflow for PIO analysis

First, we carried out single point calculation followed by natural population analysis to obtain the natural atomic orbital (NAO) coefficients and the NAO-based density matrix. ^{S15}

Second, we divided the complexes into two fragments.

Third, we performed the PIO program to generate a set of orbitals for each fragment that interacts with the other fragment.^{70, 71}

Fragmentation for PIO analysis

For complex **3**, one of the germylene moieties was treated as one fragment and the remaining part as the other fragment to study the bonding interactions between them.

For complexes **4** and **7**, the Au center was treated as one fragment and the remaining part as the other fragment to study the bonding interactions between them.

Selected bond lengths for crystal structures and optimized structures



Fig. S7 Optimized structure of complex 3. For clarity, the hydrogen atoms are omitted, and the trimethylsilyl moieties are displayed in wireframes.

Table S9. Selected bond lengths (Å) for crystal structure and optimized structure (in parentheses, on average) of complex **3**.

Au1-Ge1	2.41 (2.39)	Au1-Ge2	2.41 (2.39)
Gel-Cl	1.92 (1.93)	C1-C2	1.56 (1.57)
C2-C3	1.54 (1.54)	C3-C4	1.56 (1.57)
C4-Ge1	1.92 (1.93)	Ge2-C5	1.92 (1.94)
C5-C6	1.56 (1.56)	C6-C7	1.54 (1.54)
C7-C8	1.56 (1.56)	C8-Ge2	1.92 (1.93)



Fig. S8 Optimized structure of complex 4. For clarity, the hydrogen atoms are omitted, and the trimethylsilyl moieties as well as the 2,6-diisopropylphenyl moieties are displayed in wireframes.

Table S10. Selected bond lengths (Å) for crystal structure and optimized structure (in parentheses) of complex **4**.

Au1-Ge1	2.40 (2.38)	Au1-C5	2.07 (2.07)
Ge1-C1	1.93 (1.95)	C1-C2	1.56 (1.59)
C2-C3	1.54 (1.53)	C3-C4	1.56 (1.57)
C4-Ge1	1.93 (1.93)	C5-N1	1.35 (1.36)
N1-C6	1.38 (1.39)	C6-C7	1.36 (1.33)
C7-N2	1.38 (1.38)	N2-C5	1.35 (1.33)



Fig. S9 Optimized structure of complex 7. For clarity, the hydrogen atoms are omitted, and the trimethylsilyl moieties as well as the 2,6-diisopropylphenyl moieties are displayed in wireframes.

Table S10. Selected bond lengths (Å) for crystal structure and optimized structure (in parentheses) of complex **4**.

Au1-Ge1	2.40 (2.38)	Au1-C1	2.06 (2.05)
Ge1-N1	1.80 (1.81)	Ge1-N2	1.80 (1.81)
C1-N3	1.35 (1.35)	N3-C2	1.39 (1.38)
C2-C3	1.36 (1.33)	C3-N4	1.39 (1.38)
N4-C1	1.35 (1.35)		

Fragmentation for PIO analysis

For complex **3**, one of the germylene moieties were treated as one fragment and the remaining part as the other fragment to study the bonding interactions between them.

For complexes **4** and **7**, the Au center was treated as one fragment and the remaining part as the other fragment to study the bonding interactions between them.



Fig. S10 Selected principal interacting with orbital (PIO) plots (isovalue =0.05). The results were obtained in the presence of SMD solvation models. (a) PIO pairs representing cumulative 2c-2e π -back donations in complex **3**, with corresponding populations (Pop) and PIO-based bond indices (PBI). (b) PIO pairs representing 3c-2e π -back donations in complexes **4** and **7**, with corresponding populations (Pop) and PIO-based bond indices (PBI).

4. ¹H, ¹³C, ¹⁹F and ²⁹Si NMR spectra



Fig. S12 13 C NMR spectrum of 2b in C₆D₆.





Fig. S14 ¹H NMR spectrum of 3 in CDCl₃.



Fig. S15¹³C NMR spectrum of 3 in CDCl₃.



Fig. S16²⁹Si NMR spectrum of 3 in CDCl₃.



20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -130 -190 -200 -210 -22 fi (ppm)





Fig. S18 ¹H NMR spectrum of 4 in THF- d_8 .



Fig. S19 13 C NMR spectrum of **4** in THF- d_8 .



Fig. S20 29 Si NMR spectrum of 4 in THF- d_8 .







Fig. S22 13 C NMR spectrum of **6a** in C₆D₆.



Fig. S23 31 P NMR spectrum of **6a** in C₆D₆.



Fig. S24 ¹H NMR spectrum of **6b** in C_6D_6 .



Fig. S25 13 C NMR spectrum of 6b in C₆D₆.



Fig. S26 29 Si NMR spectrum of **6b** in C₆D₆.

52	20	48	30	28	26	22
2.	r'	N'	N'	5	2	2



Fig. S28 ¹³C NMR spectrum of 7 in CDCl₃.



-6.77



Fig. S29²⁹Si NMR spectrum of 7 in CDCl₃.



Fig. S30 ¹⁹F NMR spectrum of 7 in CDCl₃.



Fig. S32 ¹³C NMR spectrum of 8 in CDCl₃



Fig. S34 ¹⁹F NMR spectrum of 8 in CDCl₃.

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6. XYZ Coordinates

3+

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