Supporting Information

Z-Scheme BAIO/C₃N₄ Heterojunction Enabling Efficient Visible-Light

Photocatalytic H₂O₂ Production via Direct One-Step Two-Electron O₂ Reduction

Reaction

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Figure S1 SEM images of (a, b) BAIO, (c, d) C₃N₄, and (e, f) BAIO/C₃N₄.



Figure S2 (a) N_2 adsorption-desorption isotherms and (b) corresponding BJH pore size distribution calculated from the desorption branch of the isotherms of BAIO, C_3N_4 and BAIO/ C_3N_4 .



Figure S3 XPS spectra of survey scan in BAIO, C₃N₄ and BAIO/C₃N₄.



Figure S4 XPS spectra of (a) Ag 3d of BAIO and BAIO/C₃N₄. (b) O 1s XPS spectra of BAIO, C₃N₄ and BAIO/C₃N₄.



Figure S5 (a) Determination of UV-vis absorption intensity of different H_2O_2 concentrations by iodimetry. (b) The linear fitting formula of standard H_2O_2 concentration.



Figure S6 Photocatalytic degradation of H_2O_2 (2 mM) on BAIO/C₃N₄ in the air pure water.



Figure S7 (a, b) TEM images and (c) XRD patterns of $BAIO/C_3N_4$ before and after photocatalytic H_2O_2 evolution.



Figure S8 XPS spectra of BAIO/C₃N₄ before and after the photocatalytic H₂O₂ evolution: (a) C 1s, (b) N 1s, (c) Ba 3d and (d) I 3d



Figure S9 The transient photocurrent density of BAIO, C₃N₄ and BAIO/C₃N₄.

The built-in electric field (IEF) strength is calculated using the model formula proposed by Kanata et al., which is given by:

$$F_S = (-\frac{2\rho V_S}{\varepsilon \varepsilon_0})^{1/2}$$

Here, ${}^{F_{S}}$ represents the intensity of the built-in electric field (in V/cm), ρ is the surface charge density (in cm⁻²), and V_{S} is the surface potential (in volts). ${}^{\varepsilon_{0}}$ and ${}^{\varepsilon}$ denote the vacuum permittivity and low-frequency permittivity, respectively, with ${}^{\varepsilon_{0}}$ being a constant. For materials within the same class that exhibit similar structures and compositions, ${}^{\varepsilon}$ can be considered approximately constant. Therefore, only the surface charge density and surface potential need to be measured. The surface potential is determined by measuring the open-circuit potential of the material, while the surface charge density is derived from the transient photocurrent spectrum.

Consequently, the IEF values were calculated to be 1, 1.28, and 3.34 for BAIO,

 C_3N_4 , and BAIO/ C_3N_4 , respectively, with the intensity of BAIO normalized to "1" as a reference.



Figure S10 Water contact angle tests on the surface of (a) BAIO, (b) C_3N_4 and (c) BAIO/ C_3N_4 .



Figure S11 In situ DRIFTS spectra of BAIO and C_3N_4 for H_2O_2 production.

		H ₂ O ₂ yield	
Materials	Reaction conditions	-	Ref.
		(µmol g ⁻¹ h ⁻¹)	
BAIO/C ₃ N ₄	300 W Xe lamp ($\lambda > 420$ nm)	535.9	This work
Reduced g-C ₃ N ₄	300 W Xe lamp ($\lambda > 420$ nm)	460	[1]
$Bi_4O_5Br_2/g$ - C_3N_4	300 W Xe lamp ($\lambda > 420$ nm)	124	[2]
$Cd_{0.6}Zn_{0.4}S/g$ - $C_{3}N_{4}$	300 W Xe lamp ($\lambda > 420$ nm)	1098.5	[3]
C ₃ N ₄ /Mn/AB-CN	40 W LED lamp ($\lambda = 427$ nm)	410	[4]
NiS@g-C ₃ N ₄	$\lambda > 400 \text{ nm}$	400	[5]
B,Cs co-doped g-C ₃ N ₄	$\lambda > 420 \text{ nm}$	113	[6]
BP@CN	$\lambda > 420 \text{ nm}$	540	[7]

Table S1 Currently reported H_2O_2 yields for C_3N_4 -based and heterojunction photocatalysts.

[8]

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