Electronic Supplementary Information for

# A ruthenium carbolong skeleton with ten carbon atoms

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#### 1. General methods

All syntheses were carried out under an inert atmosphere (nitrogen or argon) using standard Schlenk techniques unless otherwise stated. The organic compound **L** was synthesized according to the published literature<sup>1</sup>. The other reagents and solvents were used as purchased from commercial sources without further purification. Column chromatography was performed on silica gel (200–300 mesh) in air. NMR spectra was collected on the Brucker AVIII-400 (400 MHz) or Brucker AVIII-600 spectrometer (600 MHz). <sup>1</sup>H and <sup>13</sup>C{<sup>1</sup>H} NMR chemical shifts ( $\delta$ ) are relative to tetramethylsilane, and <sup>31</sup>P{<sup>1</sup>H} NMR chemical shifts are relative to 85% H<sub>3</sub>PO<sub>4</sub>. Two-dimensional and one-dimensional NMR spectra are abbreviated as HSQC (heteronuclear single quantum coherence), HMBC (heteronuclear multiple bond coherence). The absolute values of the coupling constants are given in hertz (Hz). Multiplicities are abbreviated as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet) and br (broad). High resolution mass spectra (HRMS) experiments were recorded on a Thermo Scientific Q Exactive instrument. UV-Vis absorption spectra were recorded on a UV3600 spectrometer (Shimadzu, Japan).

## 2. NMR and HRMS spectra



Figure S1 The <sup>31</sup>P{<sup>1</sup>H} NMR (242.9 MHz, CDCl<sub>3</sub>) spectrum for compound 3a



Figure S2 The <sup>1</sup>H NMR (600.1 MHz, CDCl<sub>3</sub>) spectrum for compound 3a



Figure S4 The HRMS spectrum for compound 3a



Figure S5 The <sup>31</sup>P{<sup>1</sup>H} NMR (242.9 MHz, CDCl<sub>3</sub>) spectrum for compound 3b



Figure S16 The <sup>1</sup>H NMR (600.1 MHz, CDCl<sub>3</sub>) spectrum for compound 3b



Figure S7 The <sup>13</sup>C{<sup>1</sup>H} NMR (150.9 MHz, CDCl<sub>3</sub>) spectrum for compound 3b





Figure S9 The  ${}^{31}P{}^{1}H$  NMR (242.9 MHz, CD<sub>2</sub>Cl<sub>2</sub>) spectrum for compound 3c



Figure S10 The <sup>1</sup>H NMR (600.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>) spectrum for compound 3c



Figure S11 The <sup>13</sup>C{<sup>1</sup>H} NMR (150.9 MHz, CD<sub>2</sub>Cl<sub>2</sub>) spectrum for compound 3c



Figure S12 The HRMS spectrum for compound 3c



Figure S13 The <sup>31</sup>P{<sup>1</sup>H} NMR (242.9 MHz, CDCl<sub>3</sub>) spectrum for compound 3d



Figure S14 The <sup>1</sup>H NMR (600.1 MHz, CDCl<sub>3</sub>) spectrum for compound 3d



Figure S15 The  ${}^{13}C{}^{1}H$  NMR (150.9 MHz, CDCl<sub>3</sub>) spectrum for compound 3d



#### 3. Crystallographic details

The single-crystals suitable for X-ray diffraction were grown from dichloromethane solution layered with hexane. Single-Crystal X-ray diffraction data were collected on an Oxford Gemini S Ultra CCD area detector for **3a** with a Cu K $\alpha$  radiation ( $\lambda = 1.54178$  Å). Multi-scan absorption corrections were applied for **3a**. Using Olex2<sup>2</sup>, the structures were solved with ShelXT<sup>3</sup> structure solution program using Intrinsic Phasing and refined with the ShelXL<sup>4</sup> refinement package using Least Squares minimisation. All non-hydrogen atoms were refined anisotropically unless otherwise stated. Hydrogen atoms were placed at idealized positions and assumed the riding model. CCDC-2447172 (**3a**) contain supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via http:// www.ccdc.cam.ac.uk/structures/.

		<u> </u>	0	1		
Bond Distances (Å)						
Ru(1)-C(1)	2.069(3)	C(4)-C(5)	1.428(4)	C(11)-C(12)	1.494(5)	
Ru(1)-C(4)	2.134(3)	C(5)-C(6)	1.348(4)	C(5)-C(13)	1.503(4)	
Ru(1)-C(7)	2.145(3)	C(6)-C(7)	1.421(4)	C(13)-C(14)	1.558(4)	
Ru(1)-C(10)	2.079(3)	C(7)-C(8)	1.386(4)	C(14)-C(15)	1.567(4)	
Ru(1)-O(1)	2.249(2)	C(8)-C(9)	1.398(5)	C(6)-C(15)	1.504(4)	
C(1)-C(2)	1.367(4)	C(9)-C(10)	1.366(5)	Ru(1)-P(1)	2.3842(8)	
C(2)-C(3)	1.433(4)	C(10)-C(11)	1.387(5)	Ru(1)-P(2)	2.3827(8)	
C(3)-C(4)	1.358(4)	C(11)-O(1)	1.285(4)	C(2)-P(3)	1.789(3)	
		Bond Ang	gels (°)			
C(1)-Ru(1)-C	C(4)	73.41(12)	C(5)-C(	6)-C(7)	116.7(3)	
C(4)-Ru(1)-C	C(7)	74.91(11)	C(6)-C(7	C(6)-C(7)-Ru(1)		
C(7)-Ru(1)-C(10)		69.94(12)	C(8)-C(7	C(8)-C(7)-Ru(1)		
C(10)-Ru(1)-O(1)		60.61(10)	C(7)-C(8)-C(9)		112.8(3)	
C(2)-C(1)-Ru(1)		121.0(2)	C(8)-C(9)-C(10)		110.9(3)	
C(1)-C(2)-C(3)		113.3(3)	C(9)-C(1	C(9)-C(10)-Ru(1)		
C(2)-C(3)-C(4)		113.4(3)	C(11)-C(1	C(11)-C(10)-Ru(1)		
C(3)-C(4)-Ru(1)		118.9(2)	O(1)-C(11)-C(10)		110.0(3)	
C(5)-C(4)-Ru(1)		116.8(2)	C(11)-O(1)-Ru(1)		92.36(18)	
C(4)-C(5)-C(6)		115.5(3)				

Table S1	Selected	Bond I	Lengths	and Bo	nd Ang	les for	Com	olex	3a
			()						

Identification code	<b>3a</b> 1.5CH <sub>2</sub> Cl <sub>2</sub>		
Empirical formula	C75.5H67Cl3F3O9RuP3S		
Formula weight	1507.68		
Temperature/K	173.00(14)		
$\lambda$ [Å] (CuK $\alpha$ radiation)	1.54178		
crystal system	Triclinic		
space group	P-1		
<i>a</i> [Å]	12.7589(4)		
<i>b</i> [Å]	16.0325(4)		
<i>c</i> [Å]	17.4998(6)		
α [°]	91.986(2)		
β[°]	94.645(3)		
γ [°]	103.761(2)		
Volume/[Å <sup>3</sup> ]	3460.16		
Z	2		
$\rho_{calc} [gcm^{-3}]$	1.447		
$\mu$ [mm <sup>-1</sup> ]	4.387		
<i>F</i> (000)	1550		
crystal size [mm <sup>3</sup> ]	0.2×0.2×0.2		
$2\theta$ range [°]	7.164 to 124.278		
reflns collected	29700		
independent reflns	10882		
data/restraints/params	10882/2/868		
GOF on $F^2$	1.044		
$R_1/wR_2 \ [I \ge 2\sigma(I)]$	0.0439/0.1137		
$R_1/wR_2$ (all data)	0.0470/0.1168		
largest peak/hole [e Å <sup>-3</sup> ]	1.28 /-1.05		

Table S2 Crystal Data and Structure Refinement for 3a

#### 4. Theoretical calculations



Figure S17 ACID plot of 3a'

**Computational Details** All structures were optimized at the B3LYP level of density functional theory<sup>4-6</sup>. Frequency calculations were performed to identify all the stationary points as minima. All these structures evaluated were optimized by B3LYP-D3BJ/Def2-TZVP<sup>7</sup>. NICS values were calculated at the B3LYP-GIAO/Def2-TZVP level and the ACID calculations were carried out at the same level with NICS with the ACID program<sup>8,9</sup>. All calculations were done in Gaussian 16 program package, Revision A.03<sup>10</sup>.

# 5. Optimized Cartesian coordinates for all stationary points



 $[Ru]' = Ru(PH_3)_2$ 

3a'

E= -1320.06997518 a.u.

Ru	0.000000000	0.000000000	0.000000000
Р	-0.047085569	-0.149536174	2.299079686
Р	0.059174914	-0.029968691	-2.300212179
0	0.965912702	2.018697679	0.069515108
С	1.995308620	-0.573462914	0.000000000
Н	2.745251351	0.214433838	-0.002221470
С	2.338309086	-1.886738590	0.004677567
С	1.217659104	-2.763808369	0.003808301
Н	1.308698190	-3.847751070	-0.002280938
С	0.000000000	-2.136043843	0.000000000
С	-1.246794806	-2.794997873	-0.016663508
С	-2.337696259	-1.974546592	-0.052405303
С	-2.049997696	-0.590457472	-0.057058795
С	-3.049367609	0.387591218	-0.091895012
Η	-4.105315114	0.136337200	-0.129224965
С	-2.565552671	1.693185409	-0.068011985
Η	-3.202227835	2.574062874	-0.089134613
С	-1.191344584	1.729551869	-0.014028691
С	-0.159797625	2.646623528	0.047178358
С	-0.195980001	4.139560768	0.096151657
Η	0.280317150	4.493903544	1.013530168
Η	-1.219300306	4.511514852	0.058979030
Η	0.367093123	4.553849411	-0.743551765
Η	0.099823732	1.026340810	3.065517778
Η	-1.206179666	-0.660574937	2.913189956
Η	0.917389333	-0.948883870	2.941993245
Η	0.707733322	-1.102548879	-2.940057663
Η	-1.148278533	-0.035023701	-3.025909019
Η	0.699749735	1.039729789	-2.959495542
Н	-3.359329310	-2.342080525	-0.074344227
Н	-1.317187724	-3.879432188	-0.007874501
Н	3.362445520	-2.248833839	0.009821280



E= -1666.69702794 a.u.

Ru	-0.34116400	-0.29934500	-0.09951200
Р	-0.42368200	-0.83799400	2.13649100
Р	-0.58999500	0.06084800	-2.37181500
0	-0.87574800	-2.35935100	-0.60217800
С	-2.45733500	-0.31458500	0.15344900
С	-3.13644300	0.96978200	0.41188700
С	-2.24398700	2.08925600	0.39988100
Н	-2.58508000	3.10891700	0.55924300
С	-0.94926600	1.78974300	0.16391500
С	0.10968900	2.76374900	0.02482800
С	1.43811900	2.15774100	-0.02828700
С	1.44332800	0.72662900	0.04864600
С	2.66464000	-0.00956800	0.25437100
С	2.52717900	-1.46515800	0.11594100
С	1.17979200	-1.79037200	-0.17600500
С	0.37004500	-2.80368500	-0.53458900
С	0.65906400	-4.22866500	-0.86907400
Н	0.16604700	-4.89624200	-0.15692900
Н	1.73159500	-4.42241400	-0.84897100
Н	0.27119300	-4.46834500	-1.86263800
Н	-0.88558600	-2.14000300	2.39388000
Н	0.74865100	-0.83819600	2.92113600
Н	-1.27042800	-0.10276800	2.98614500
Н	-1.89993200	0.12412400	-2.88023000
Н	-0.06405500	1.22868200	-2.95897300
Н	-0.04719300	-0.89729800	-3.25219700
С	-3.38108900	-1.29318000	0.17043500
Н	-3.20715200	-2.34518700	0.00705100
С	2.37597800	3.12319900	-0.22728300
Н	3.43790200	2.99385100	-0.36437300
С	3.94746300	0.30445400	0.58992500
Н	4.35891100	1.27712800	0.80829600
С	-4.74412200	-0.69615100	0.43888600
Н	-5.45117100	-0.90720300	-0.37427800
Н	-5.19854800	-1.11714500	1.34573000
С	1.67924600	4.44976200	-0.29962100
Н	2.05146900	5.13554900	0.47428100
Н	1.87889500	4.95608200	-1.25389500
С	4.73132400	-0.96930400	0.66291500
Н	5.57550200	-0.94226700	-0.04144300
Н	5.19104300	-1.08979200	1.65407200
С	-4.46429300	0.77523300	0.57836100
Н	-5.21887400	1.52342300	0.77416200
С	3.73278200	-2.04996200	0.34968400
Н	3.97319800	-3.10131000	0.34088400
С	0.22397500	4.10283900	-0.12731500
Н	-0.57056800	4.83304100	-0.15085500



E= -1666.76268999 a.u.

Ru	0.37451200	-0.37371800	0.01614000
Р	0.36686900	-0.39295500	-2.27966900
Р	0.37293700	-0.16815100	2.30882300
0	0.70448000	-2.58356000	0.11749400
С	2.45265900	-0.44249100	-0.02940800
С	3.17875100	0.73584400	-0.04491600
С	2.41175700	1.90879700	-0.03015000
Н	2.83766700	2.90889000	-0.04980100
С	1.04878100	1.68940400	-0.00206700
С	0.06075000	2.67064400	0.00741500
С	-1.25572300	2.22753500	-0.00798500
С	-1.43380400	0.83571000	-0.02157600
С	-2.66467400	0.16401300	-0.05210600
С	-2.57304600	-1.23344400	-0.02717100
С	-1.27807500	-1.69058200	0.02156900
С	-0.55925500	-2.85994000	0.10389100
С	-1.02149200	-4.27804500	0.18813300
Н	-0.60858700	-4.85705900	-0.64185600
Н	-2.10882400	-4.34004900	0.15950800
Н	-0.66187800	-4.73510600	1.11362600
Н	0.65495400	-1.60822200	-2.93621600
Н	-0.81423100	-0.05453100	-2.96913800
Н	1.26889400	0.43991800	-2.96932600
Н	1.57949800	0.13304200	2.97202900
Н	-0.44347100	0.82808600	2.87660400
Н	-0.03621700	-1.25872700	3.10707600
С	3.40244000	-1.54451400	-0.05757700
Н	3.11868200	-2.58547600	-0.05191500
С	4.66689800	-1.06469200	-0.08406000
Н	5.57353800	-1.65324400	-0.10428400
С	-2.13828900	3.39858500	-0.00596500
Н	-3.21705100	3.36907100	-0.01409800
С	-1.39450500	4.51838900	0.01033800
Н	-1.77062300	5.53138900	0.01623400
С	-4.08330700	0.52495400	-0.09928800
Н	-4.47536600	1.53027100	-0.12853400
С	-4.83343600	-0.59195800	-0.10432900
Н	-5.91267900	-0.63241200	-0.13612800
С	4.65293000	0.44102000	-0.07755200
Н	5.18413300	0.84849200	0.79033100
Н	5.14788200	0.85652600	-0.96288000
C	0.07937000	4.17923200	0.02089300
Н	0.60260500	4.59516400	-0.84647300
Н	0.58598200	4.57903300	0.90549100
C	-3.95909300	-1.82929300	-0.05933400
H	-4.16343700	-2.44746300	0.82106000
Н	-4.11074600	-2.47392600	-0.93122600

#### 6. References

1. Q. Zhuo, J. Lin, Y. Hua, X. Zhou, Y. Shao, S. Chen, Z. Chen, J. Zhu, H. Zhang and H. Xia, Multiyne Chains Chelating Osmium via Three Metalcarbon  $\sigma$  Bonds, *Nat. Commun.*, 2017, **8**, 1912.

2. O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard and H. Puschmann, Olex2: a Complete Structure Solution, Refinement and Analysis Program, *J. Appl. Cryst.*, 2009, **42**, 339-341.

3. G. M. Sheldrick, *SHELXT* – Integrated Space-group and Crystal-structure Determination, *Acta Cryst. A*, 2015, **71**, 3-8.

4. G. M. Sheldrick, Crystal Structure Refinement with SHELXL, Acta Cryst. C, 2015, 71, 3-8.

5. C. Lee, W. Yang and R. G. Parr, Development of the Colle-Salvetti Correlation-Energy Formula into a Functional of the Electron Density, *Phys. Rev. B*, 1988, **37**, 785-789.

6. B. Miehlich, A. Savin, H. Stoll and H. Preuss, Results Obtained with the Correlation Energy Density Functionals of Becke and Lee, Yang and Parr, *Chemical Phys. Lett.*, 1989, **157**, 200-206.

7. A. D. Becke, Density-Functional Thermochemistry. III. The Role of Exact Exchange, *J. Chem. Phys.*, 1993, **98**, 5648-5652.

 R. Herges and D. Geuenich, Delocalization of Electrons in Molecules. J. Phys. Chem. A, 2001, 105, 3214-3220.

9. D. Geuenich, K. Hess, F. Köhler and R. Herges, Anisotropy of the Induced Current Density (ACID), a General Method To Quantify and Visualize Electronic Delocalization, *Chem. Rev.*, 2005, **105**, 3758-3772.

Gaussian 16, Revision A.03. M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A.
 Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato,
 A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz,
 A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B.
 Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W.
 Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda,
 O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. J.
 Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J.
 Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M.

Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas,

J. B. Foresman and D. J. Fox, Gaussian, Inc., Wallingford CT, 2016.