

Transformation of Bimetal Carboxylate-MOF into Phosphonate-based MOF as Multifunctional Photocatalyst

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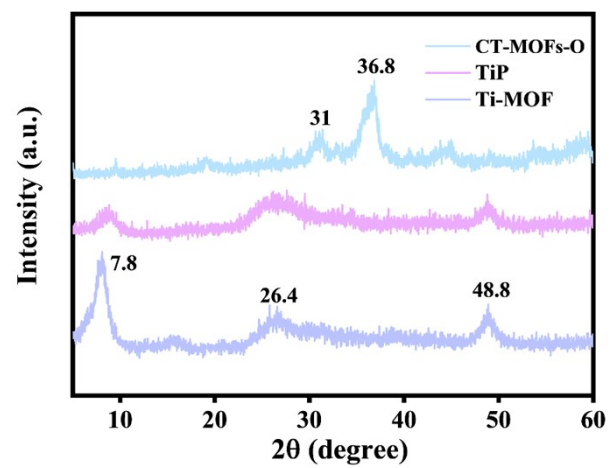


Figure S1. XRD pattern of Ti-MOF, TiP and CT-MOFs-O.

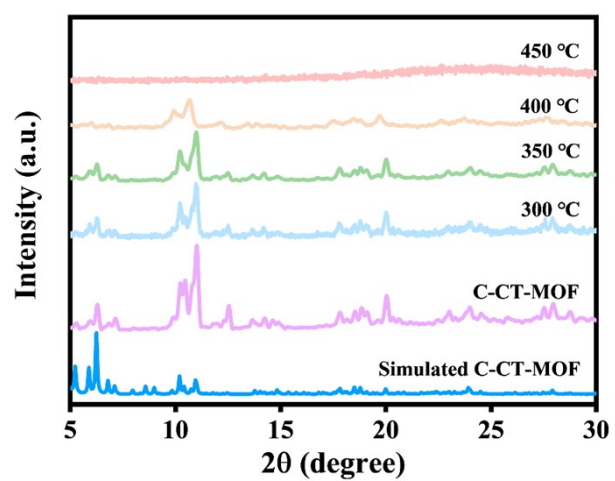


Figure S2. XRD pattern of C-CT-MOF with different temperature.

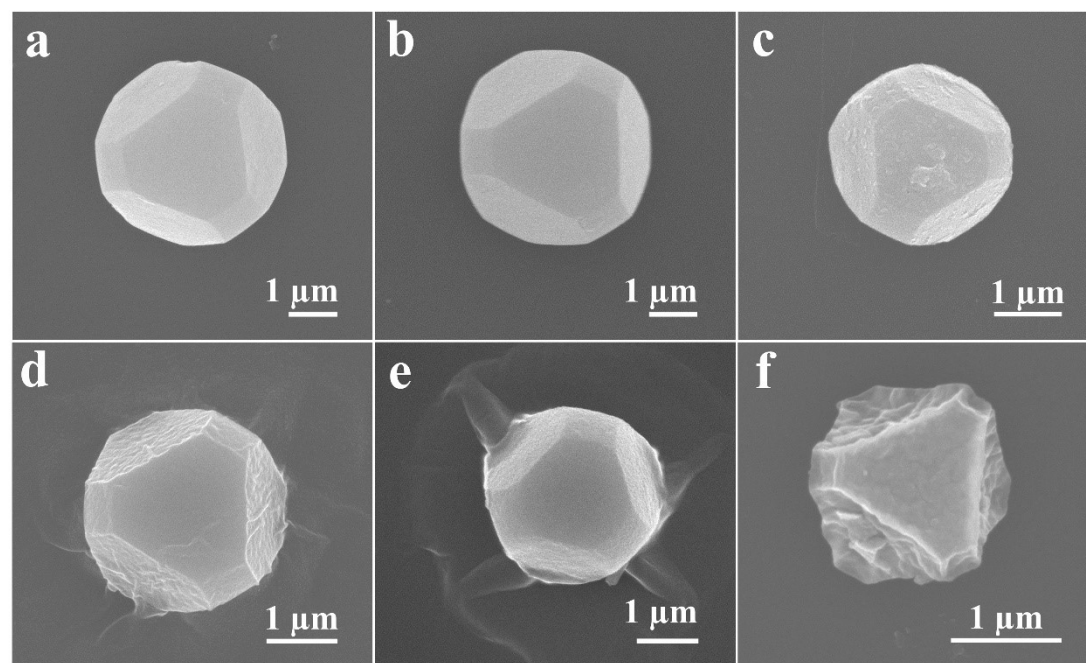


Figure S3. SEM images of (a) C-CT-MOF with different calcined temperature, (b) 300°C, (c) 350°C, (d) 400°C, (e) 450°C and (f) 350°C with O₂ atmosphere.

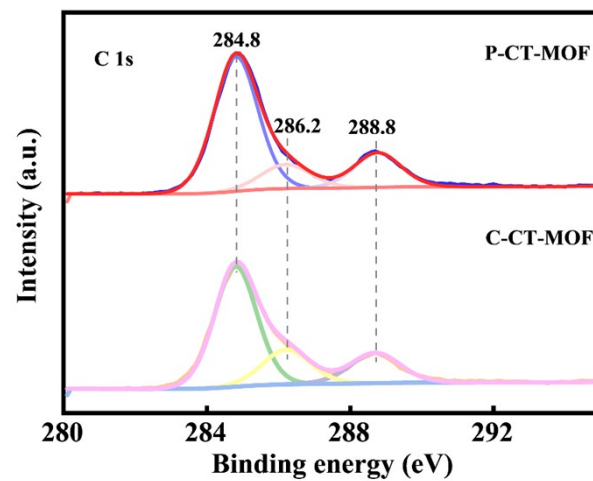


Figure S4. High resolution XPS spectra of C 1s of C-CT-MOF and P-CT-MOF.

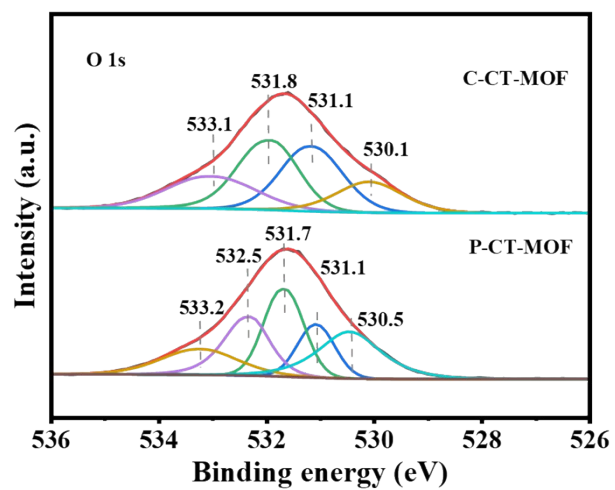


Figure S5. High resolution XPS spectra of O 1s of C-CT-MOF and P-CT-MOF.

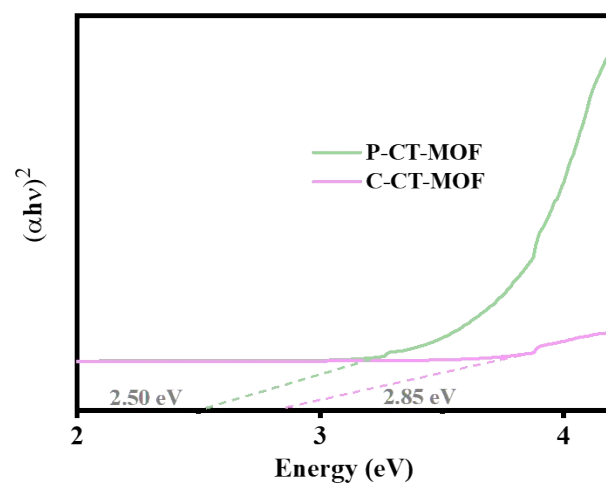


Figure S6. The band gap of C-CT-MOF and P-CT-MOF.

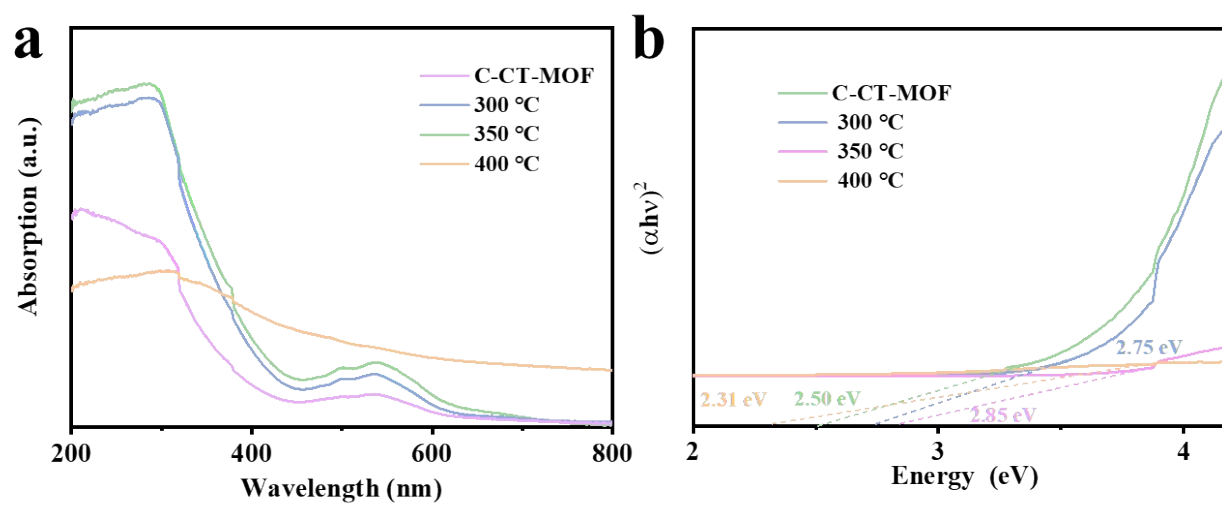


Figure S7. The UV-vis DRS and the band gap of C-CT-MOF and P-CT-MOF at different calcination temperature.

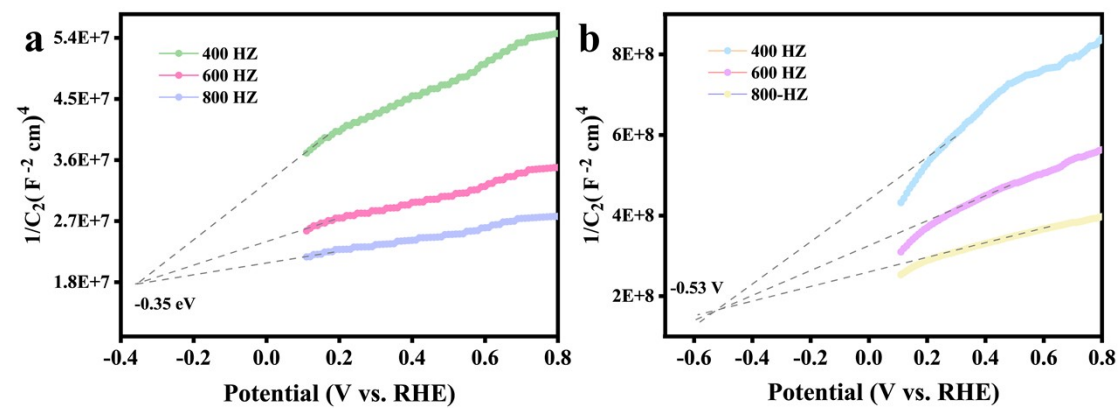


Figure S8. M-S curve plot of C-CT-MOF and P-CT-MOF.

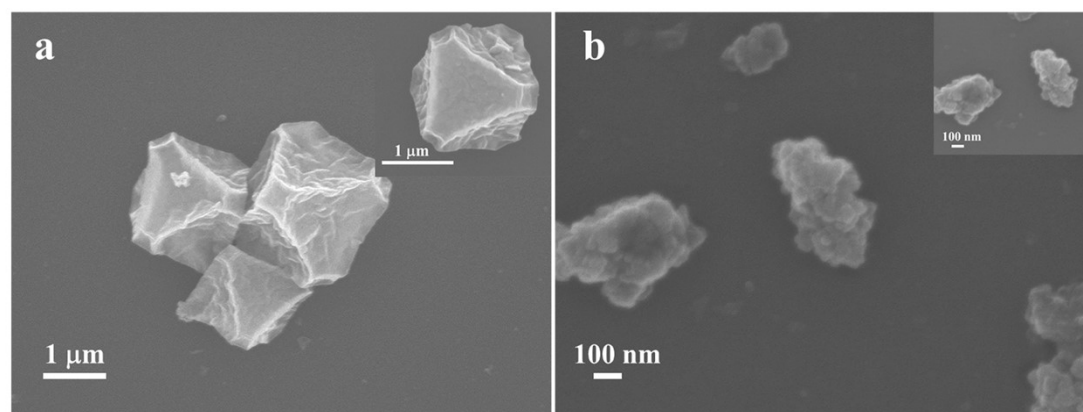


Figure S9. The SEM image of CT-MOFs-O and Ti-MOF.

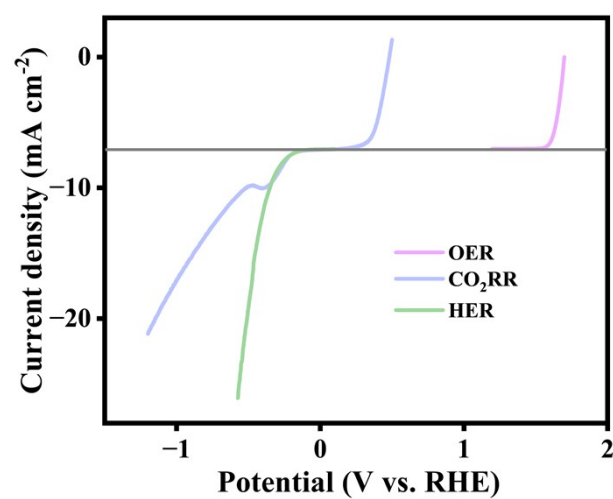


Figure S10. LSV curves in Electrocatalytic OER, HER and CO₂RR of P-CT-MOF.

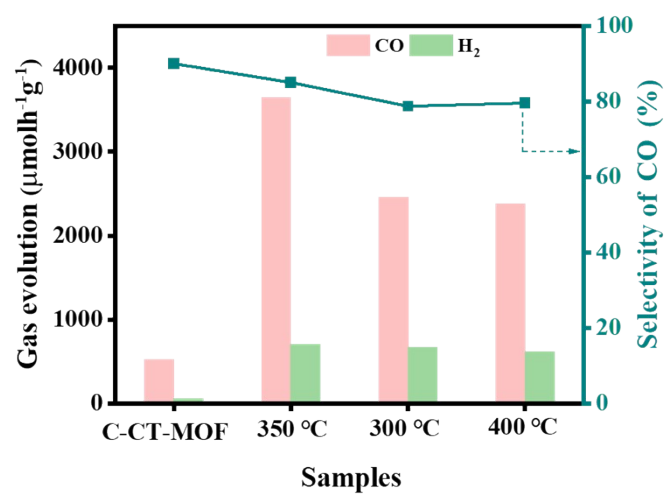


Figure S11. The Photocatalytic CO₂RR performance of samples at different calcination temperature.

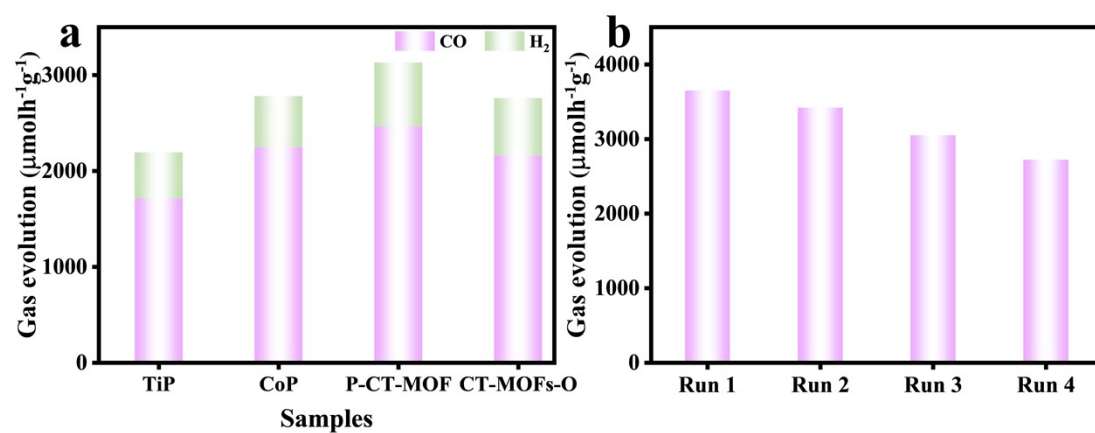


Figure S12. The Photocatalytic CO₂RR performance of samples at (a) different

catalyst and (b) run stability.

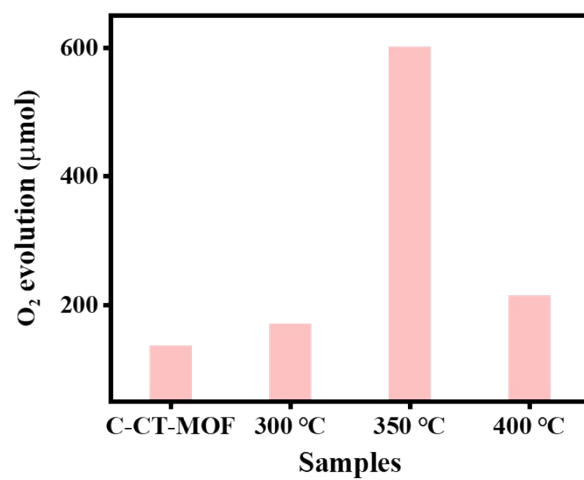


Figure S13. The Photocatalytic OER performance of samples at different calcination temperature.

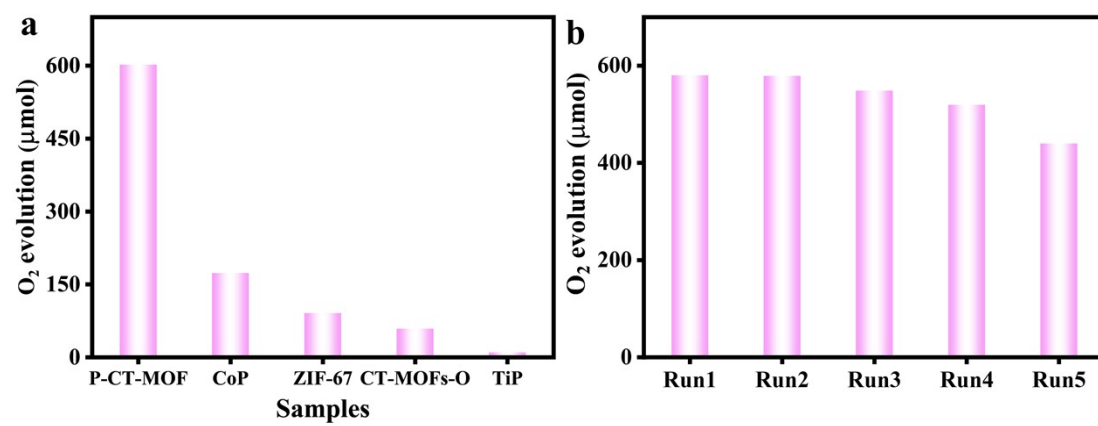


Figure S14. The photocatalytic OER performance of (a) different samples and (b) run stability.

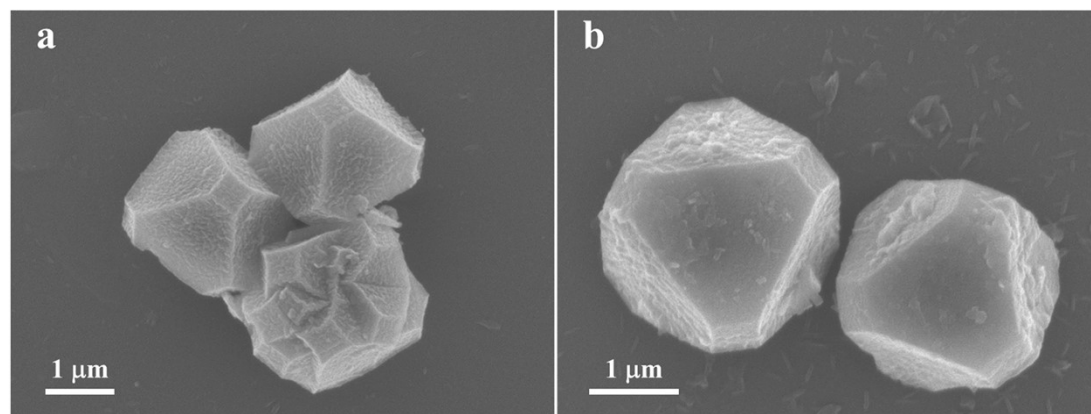


Figure S15. The SEM image of P-CT-MOF after photocatalytic CO₂RR.

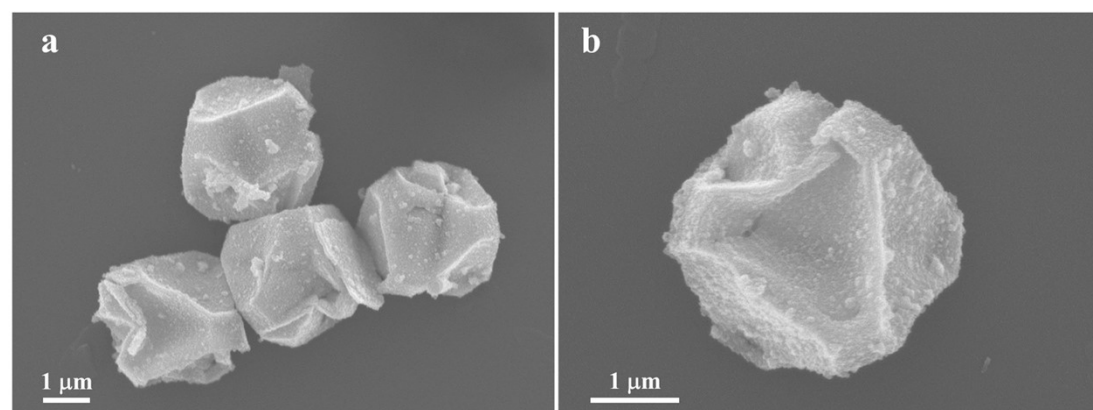


Figure S16. (a, b) The SEM image of P-CT-MOF after photocatalytic OER.

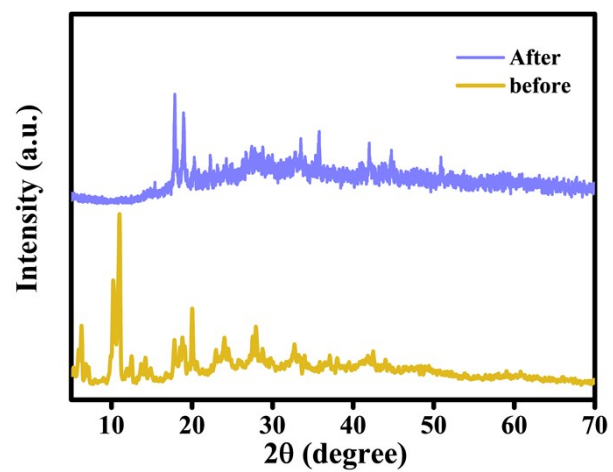


Figure S17. The XRD pattern of P-CT-MOF before and after photocatalytic CO₂RR.

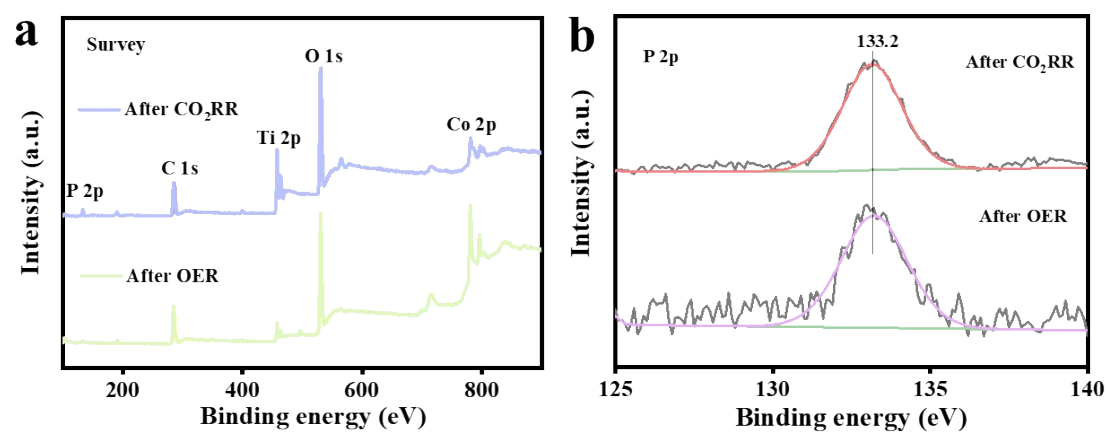


Figure S18. (a) The XPS full survey spectra and (b) P 2p spectra of after photocatalytic CO₂RR and OER of P-CT-MOF samples.

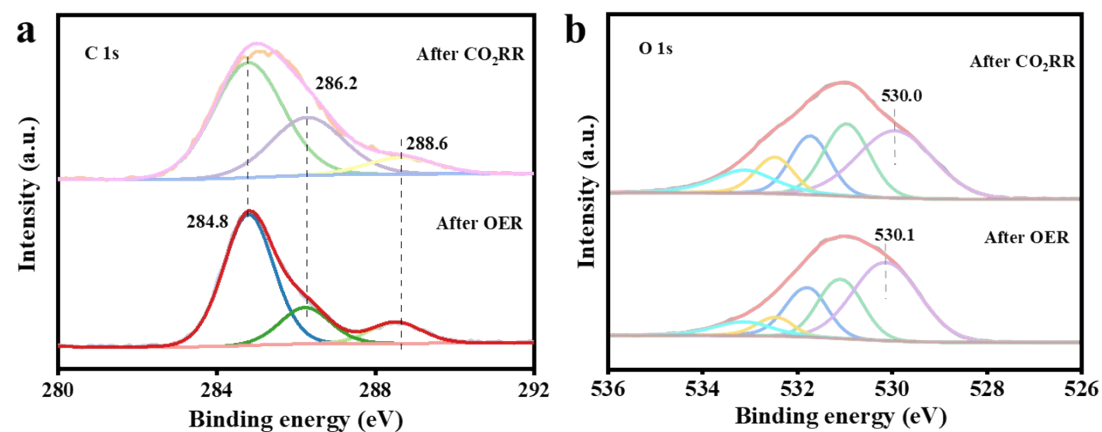


Figure S19. The XPS spectra of after photocatalytic CO₂RR and OER of P-CT-MOF samples; (a) C 1s, (b) O 1s.

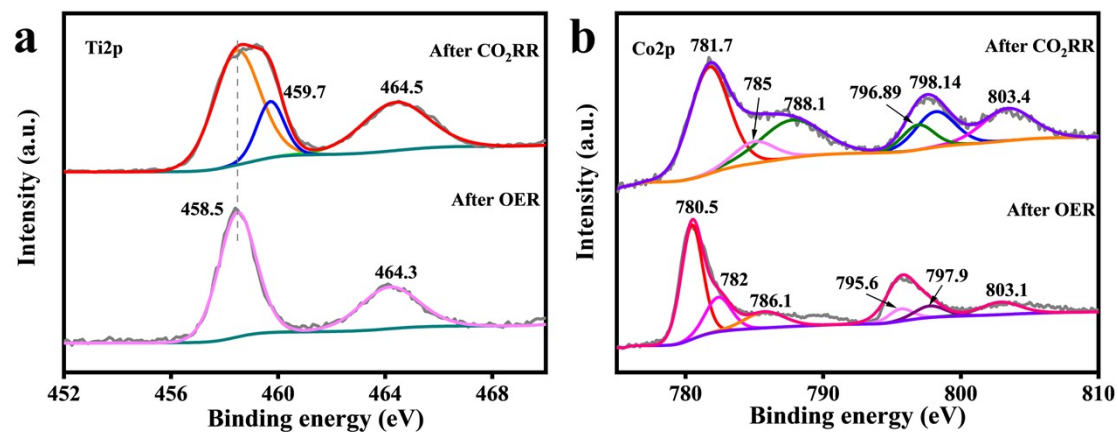


Figure S20. The SEM (a, b) and XPS (c, d) spectra after photocatalytic CO₂RR and OER of P-CT-MOF samples (c) Ti 2p and (d) Co 2p.

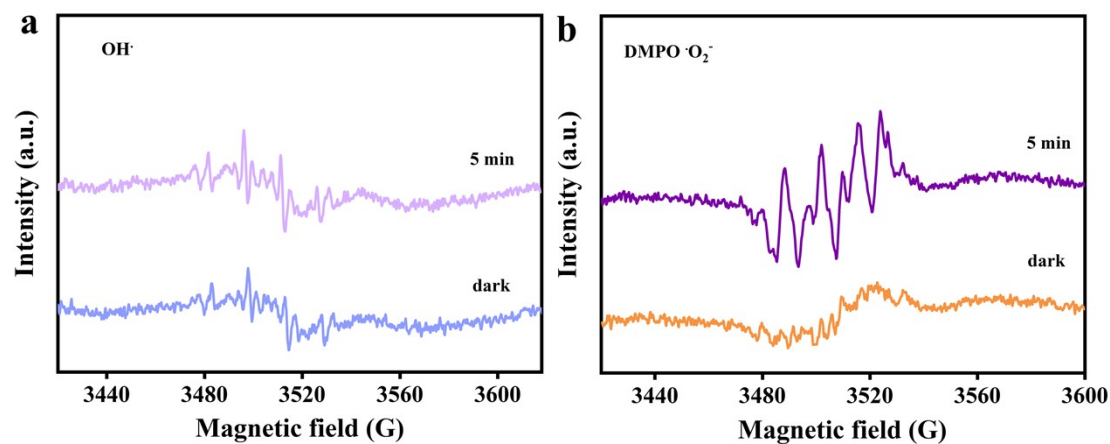


Figure S21. EPR of C-CT-MOF in aqueous solution and methanol solution.

Table S1. The composition of P-CT-MOF by ICP-OES before and after CO₂RR

reaction.

P-CT-MOF	before	after
Co:Ti (%)	14.37:8.80	11.40: 6.90

Table S2. Comparison table of photocatalytic activity of semiconductor catalysts for CO₂RR.

Materials	λ	CO rate($\mu\text{molh}^{-1}\text{g}^{-1}$)	AQY	literature
P-CT-MOF	$\lambda \geq 420\text{ nm}$	3650.3	no	Our work
NiMOF/g-C ₃ N ₄	$\lambda \geq 420\text{ nm}$	54.5	no	[1]
FeOOH/CdS	$\lambda \geq 420\text{ nm}$	12.55	4.6	[2]
Re(I)MOF	$\lambda \geq 420\text{ nm}$	1070	1.97	[3]
Au-NC@UiO-68NH ₄	$\lambda \geq 400\text{ nm}$	57.6	no	[4]
Re-UiO-67	$\lambda \geq 400\text{ nm}$	412.5	no	[5]
MnMOF	$\lambda \geq 420\text{ nm}$	21	no	[6]
NH ₂ -TiMOF	$\lambda \geq 420\text{ nm}$	100.46	no	[7]
MIL-101(Fe)	$\lambda \geq 400\text{ nm}$	11.5	0.943	[8]
UiO-67-Re	$\lambda \geq 420\text{ nm}$	60.6	no	[9]
ZrMOF	$\lambda \geq 420\text{ nm}$	1920	no	[10]
TiO ₂ /NiIP NSs	$\lambda \geq 420\text{ nm}$	54.9 mmol g ⁻¹ h ⁻¹	no	[11]
InTCPP	$\lambda \geq 420\text{ nm}$	61.2	no	[12]
CoIn-LDH/MOF	$\lambda \geq 420\text{ nm}$	2320	no	[13]
Zr-NDI@Ru-tpy	$\lambda \geq 420\text{ nm}$	2449	no	[14]
CoTiPa	$\lambda \geq 420\text{ nm}$	49.92	2.05	[15]

Literature

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