

Supplementary Information

Mixed spinel $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ on nickel foam for electrocatalytic oxygen evolution reaction: revealing the influence of temperature and promoting performance with zero voltage switching inductive heating

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Table S1 The linearity (R^2) of anodic-cathodic current density (ΔI) and the scan rate (ν) for ECSA measurements.

Table S2 OER activity comparisons.

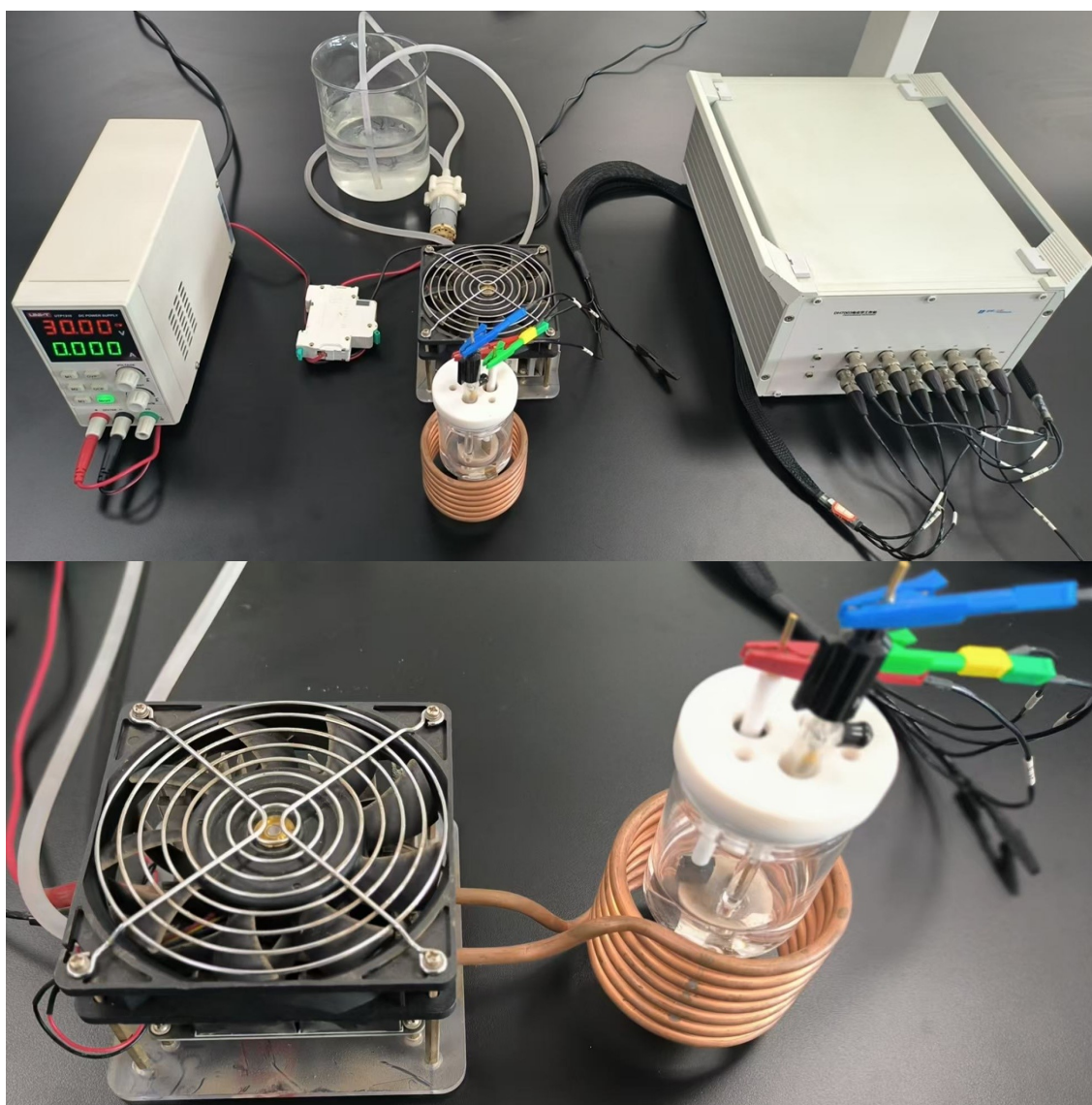


Fig. S1 The photograph of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ cell with ZVS induction heater for electrochemical measurements. The ZVS inductive heater was composed of a DC power supply and a ZVS inductive device that was purchased from Shenzhen Huali Electronic Technology Co., Ltd, China.

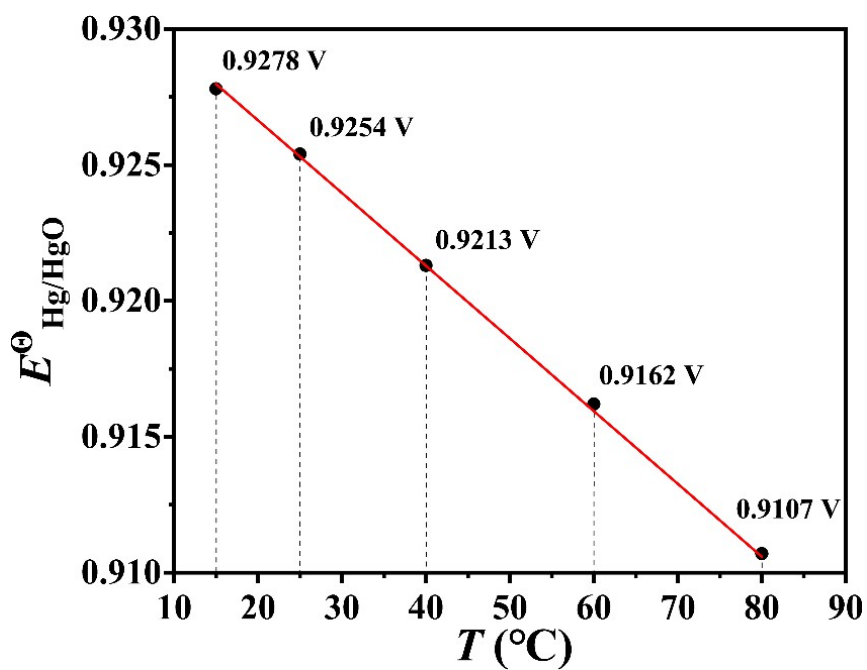


Fig. S2 The standard potential of Hg/HgO reference electrode ($E^{\ominus}_{\text{Hg/HgO}}$) at different temperatures.

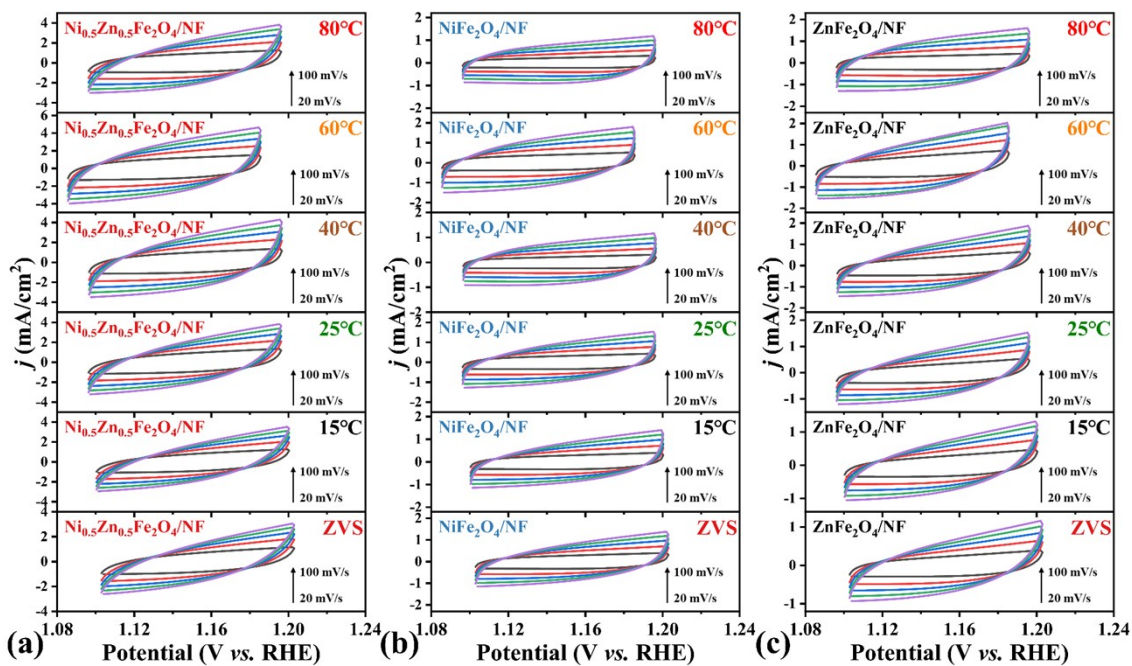


Fig. S3 CV data of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$, $\text{NiFe}_2\text{O}_4/\text{NF}$ and $\text{ZnFe}_2\text{O}_4/\text{NF}$ for their ECSA calculation.

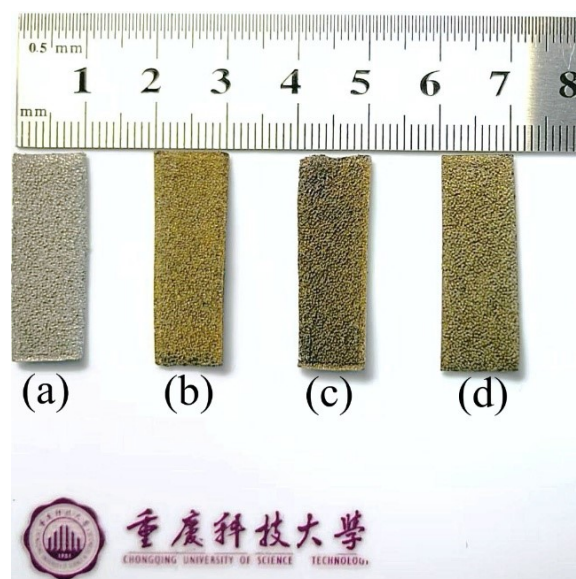


Fig. S4 The photographs of (a) Ni foam, (b) $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$, (c) $\text{NiFe}_2\text{O}_4/\text{NF}$ and (d) $\text{ZnFe}_2\text{O}_4/\text{NF}$ sample.

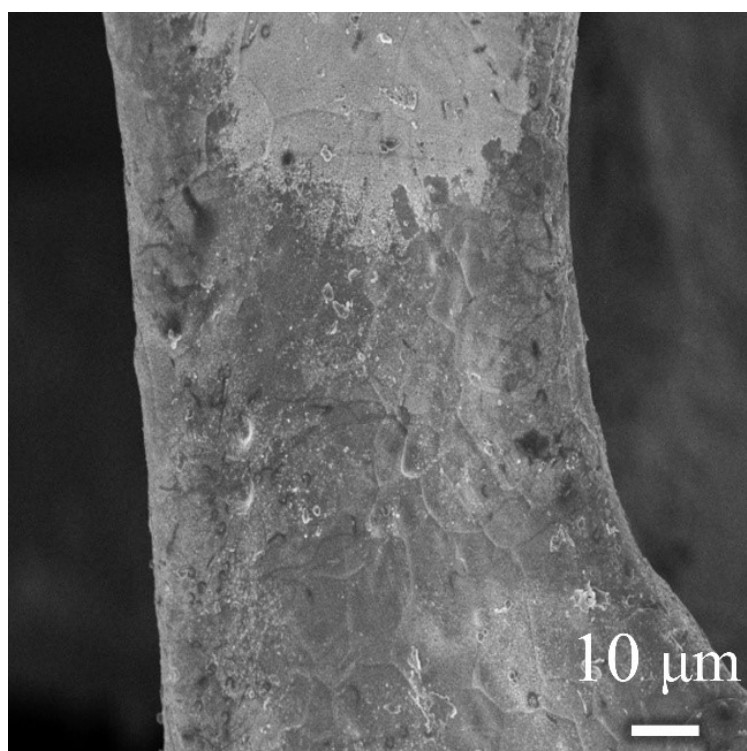


Fig. S5 Low-magnification SEM image of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$.

Fig. S6 Raman spectra of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$, ZnFe_2O_4 and NiFe_2O_4 powders.

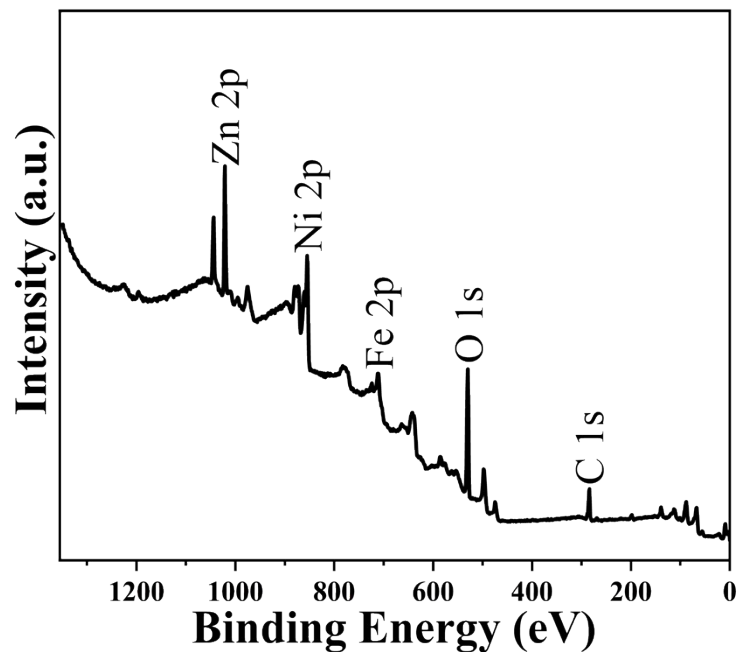


Fig. S7 Survey XPS spectrum of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$.

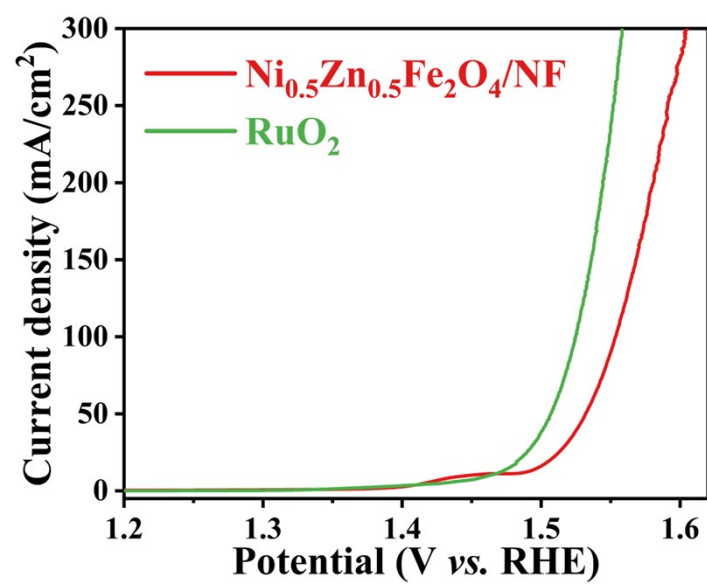


Fig. S8 LSV curves of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ and commercial RuO_2 in 25°C 1.0 M KOH.

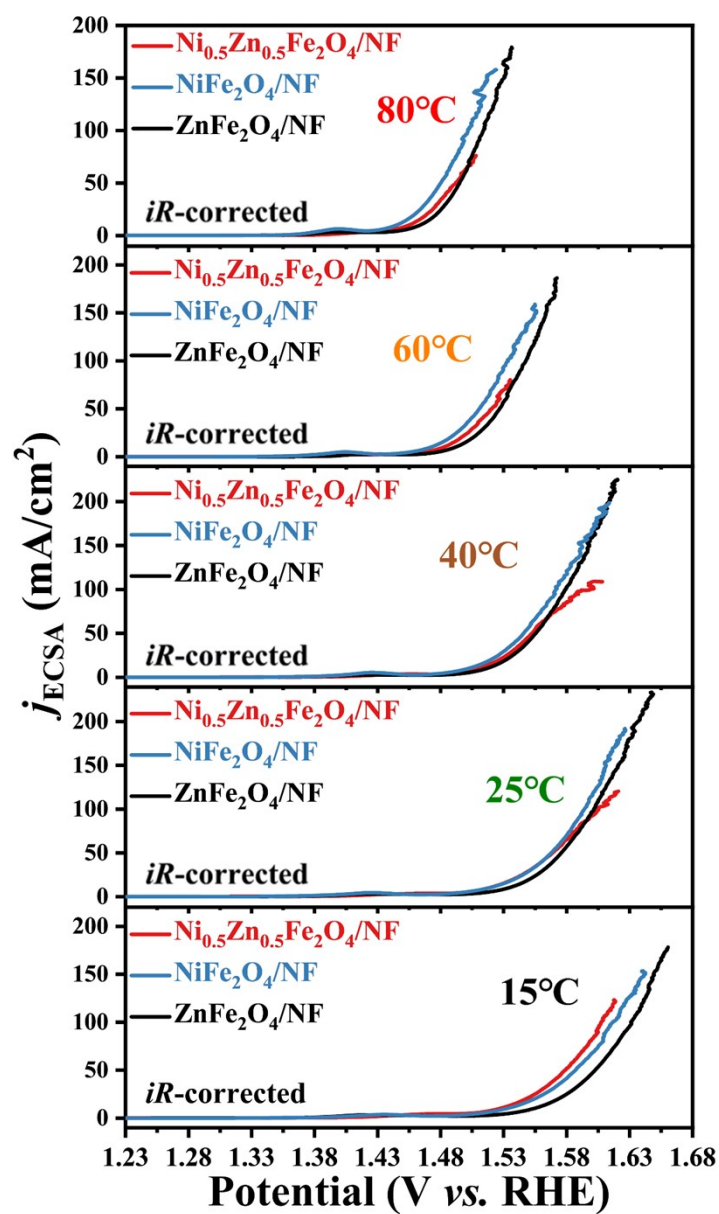


Fig. S9 ESCA-normalized current density (j_{ECSA}) curves of $Ni_{0.5}Zn_{0.5}Fe_2O_4/NF$, $NiFe_2O_4/NF$ and $ZnFe_2O_4/NF$ in 1.0 M KOH at different temperatures.

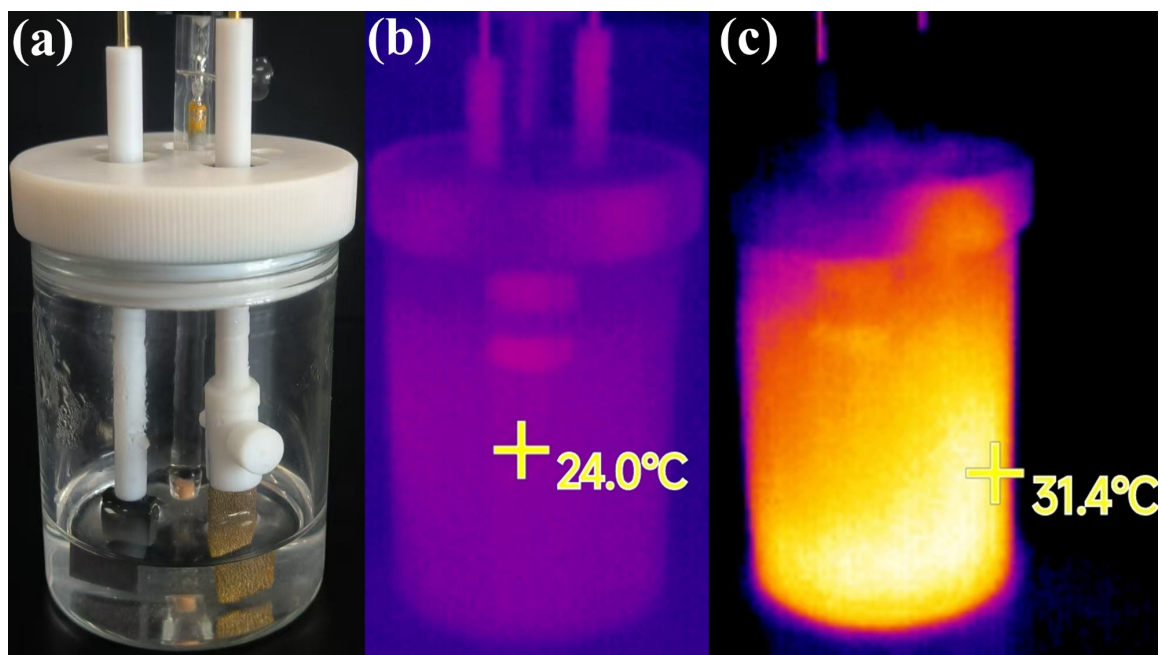


Fig. S10 (a) Photographs of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ -based cell for water electrolysis, thermal image of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ -based cell (b) before and (c) after ZVS induction heating.

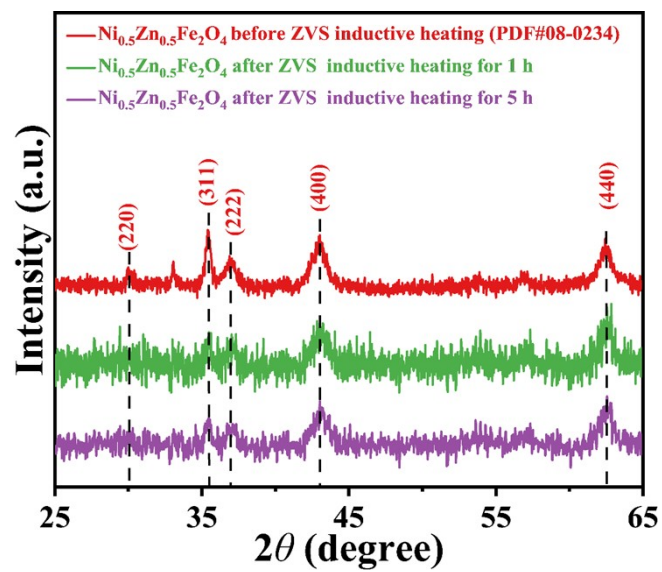


Fig. S11 XRD patterns of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ powders that collected from $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ before and after ZVS inductive heating for 1h and 5 h.

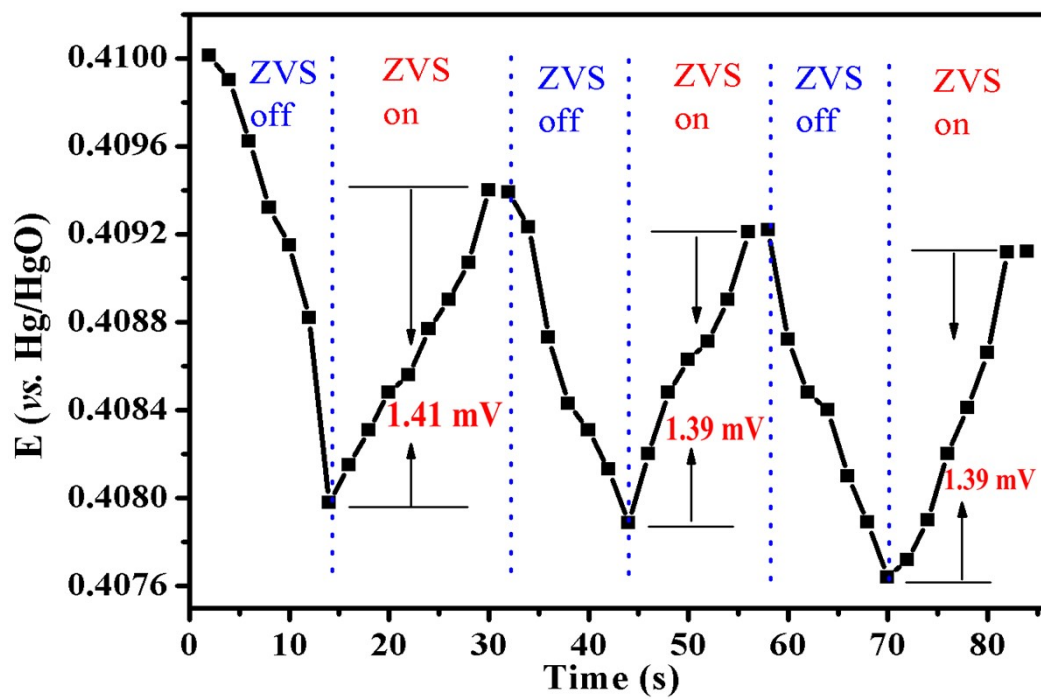


Fig. S12 OCP curve of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ in 1.0 M KOH with ZVS inductive heating on and off.

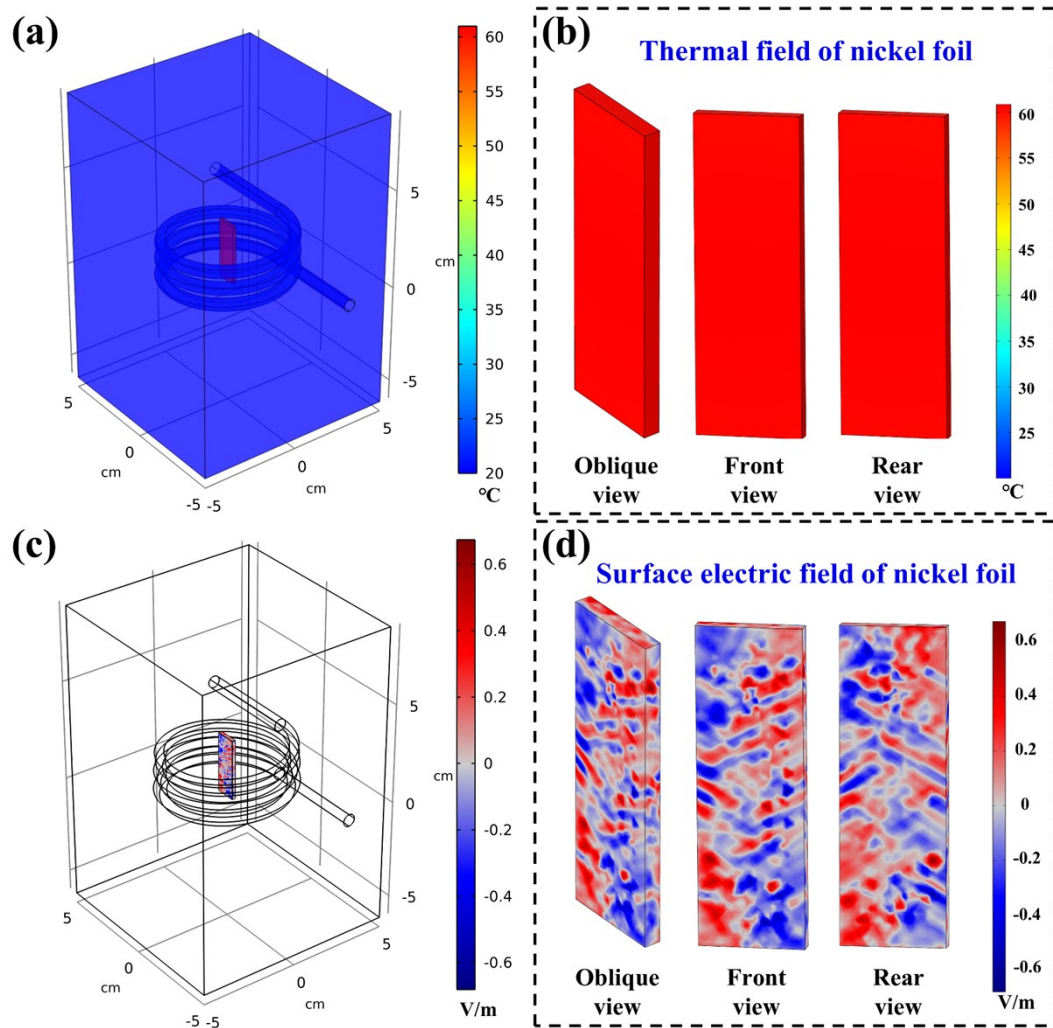


Fig. S13 COMSOL Multiphysics software simulation for the formation of (a, b) thermal field, (c, d) electric field on nickel foil under ZVS inductive heating.

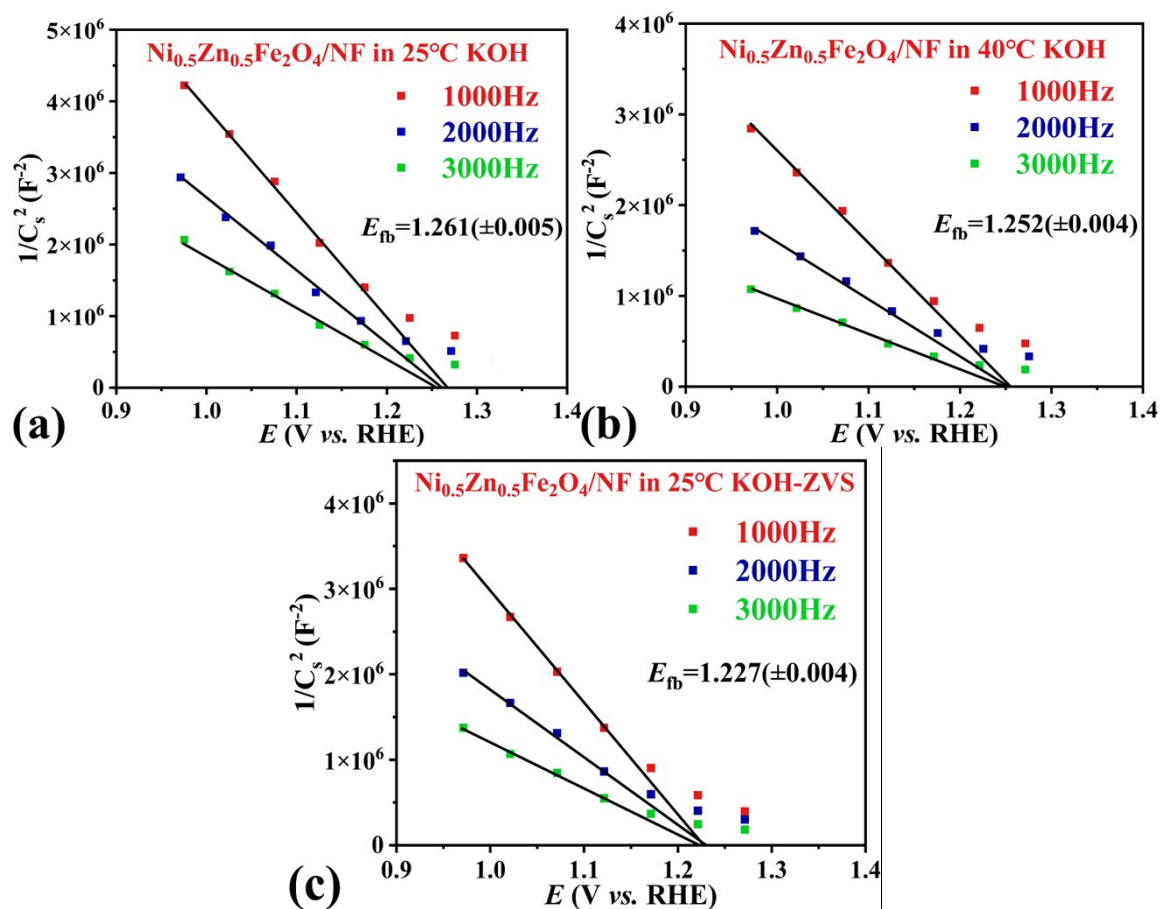


Fig. S14 Mott-Schottky plots of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ in (a) 25°C and (b) 40°C 1.0 M KOH without ZVS inductive heating, and in (c) 25°C 1.0 M KOH with ZVS inductive heating.

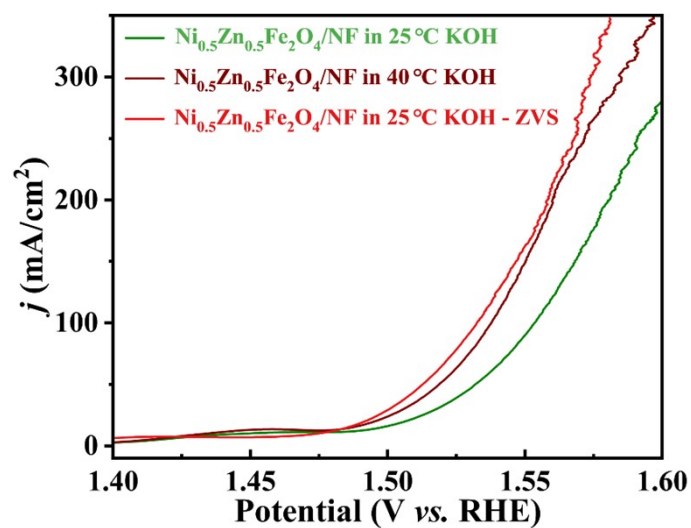


Fig. S15 LSV curve of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ in 25°C and 40°C 1.0 M KOH without ZVS inductive heating, and in 25°C under ZVS inductive heating.

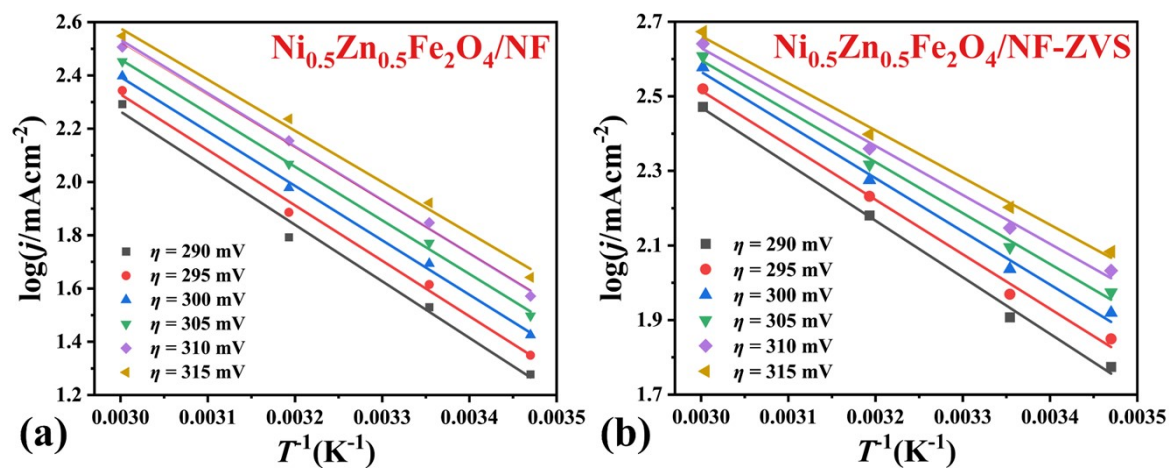


Fig. S16 $\log j$ - T^{-1} curves of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ in OER Tafel range with and without ZVS inductive heating.

Table S1 The linearity (R^2) of anodic-cathodic current density (ΔI) and the scan rate (v) for ECSA measurements.

Sample \ R^2	15°C	25°C	40°C	60°C	80°C	ZVS
Ni _{0.5} Zn _{0.5} Fe ₂ O ₄ /NF	0.97	0.97	0.98	0.98	0.98	0.97
NiFe ₂ O ₄ /NF	0.98	0.98	0.98	0.99	0.98	0.99
ZnFe ₂ O ₄ /NF	0.98	0.97	0.98	0.99	0.98	0.99

Table S2 OER activity comparison of the resultant $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$ to the previously reported spinel ferrite electrocatalysts.

Electrocatalyst	Overpotential	Reference
$\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{NF}$	276 mV@20 mA cm ⁻² 293 mV@36 mA cm ⁻²	This work
$\text{Co}_{1.2}\text{Fe}_{1.8}\text{O}_4$	320 mV@10 mA cm ⁻²	1
ZCF-50($\text{Cu}_{0.5}\text{ZnFe}_2\text{O}_4$)	280 mV@10 mA cm ⁻²	2
$\text{NiDy}_{0.05}\text{Fe}_{1.95}\text{O}_4$	360 mV@36 mA cm ⁻²	3
$\text{NiCo}_{0.03}\text{Fe}_{1.97}\text{O}_4$	450 mV@10 mA cm ⁻²	4
$\text{Co}_{0.8}\text{Mn}_{0.2}\text{Gd}_{0.1}\text{Fe}_{1.9}\text{O}_4$	342 mV@10 mA cm ⁻²	5
$(\text{Mn}_{0.2}\text{Co}_{0.2}\text{Ni}_{0.2}\text{Cu}_{0.2}\text{Fe}_{0.2})\text{Fe}_2\text{O}_4$	295 mV@10 mA cm ⁻²	6
ZnFe_2O_4	330 mV@10 mA cm ⁻²	7
$\text{Co}_{2.25}\text{Fe}_{0.75}\text{O}_4$	350 mV@10 mA cm ⁻²	8
$(\text{CoNiCuZnMg})\text{Fe}_2\text{O}_4$	420 mV@10 mA cm ⁻²	9
NiFe_2O_4	278 mV@10 mA cm ⁻²	10
CoFe_2O_4	350 mV@10 mA cm ⁻²	11
P-Mo- Co_3O_4 @CC	276 mV@10 mA cm ⁻²	12
SFO/Sm-NFO	228 mV@10 mA cm ⁻²	13
Ta-NiFe LDH	260 mV@50 mA cm ⁻²	14
$\text{CoNiFeO}_x\text{-NC}$	265 mV@50 mA cm ⁻²	15

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