Synthesis and Design of Terbium-Activated CaWO₄ Nanocrystals for Fluorescence Sensing of Alkaline Phosphatase and Bioimaging Applications

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Fig. S1 Optimization of synthesis parameters A) temperature: insets are corresponding aqueous dispersion of PEG-Tb-CaWO₄ NCs and B) time.



Fig. S2 Optimization of synthesis parameters A) %doping of Tb³⁺ and B) amount of PEG in the reaction medium

Characterization of the as-synthesized PEG-Tb-CaWO₄ NCs



Fig. S3 HR-TEM image showing poly-crystalline nature of the PEG-Tb-CaWO₄ NCs.



Fig. S4 Energy dispersive X-ray (EDS) spectrum of the PEG-Tb-CaWO₄ NCs.



Fig. S5 XRD patterns of the Tb-CaWO₄ NCs and the standard diffraction pattern of CaWO₄ NCs.



Fig. S6 Thermal gravimetric analysis (TGA) of bare and PEG coated Tb-CaWO₄ NCs.



Fig. S7 A) Aqueous dispersion of PEG-coated (left) and uncoated (right) Tb-CaWO₄ NCs.



Fig. S8 Fluorescence emission of CaWO₄ and Inset: corresponding solid-state luminescence under excitation of 254 nm UV-light.



Fig. S9 CIE chromaticity diagram of A) CaWO₄ and B) PEG-Tb-CaWO₄ NCs. Insets show corresponding fluorescence emission coordinates.

Stability of PEG-Tb-CaWO₄ NCs



Fig. S10 Stability of the fluorescence intensity of PEG-Tb-CaWO₄ NCs against A) various ionic concentrations, B) storage time at 4°C, and C) various pH solutions.



Designing PEG-Tb-CaWO₄ NCs based sensor for ALP Detection

Fig. S11 Fluorescence emission intensity of PEG-Tb-CaWO₄ NCs with various concentrations of Ce^{4+} ions.



Fig. S12 UV-Visible absorption spectrum of PEG-Tb-CaWO₄ NCs with and without the addition of Ce^{4+} ions.



Fig. S13 Fluorescence excitation spectrum of PEG-Tb-CaWO₄ NCs and UV-visible absorption spectrum of Ce^{4+} ions.



Fig. S14 Optimization of detection conditions A) Effect of pH on fluorescence quenching efficiency of Ce⁴⁺ ions in the absence of ALP. B) Effect of pH on the fluorescence recovery of the detection system in the presence of ALP. C) Concentration of AAP. D) Enzymatic reaction time. E) Amount of PEG-Tb-CaWO₄ NCs.

Optimization of detection conditions

Optimization of ratio of AA-to-Ce4+

Redox reaction between Ce4+ ions and AA. 1

$$2Ce^{4+} + C_6H_8O_6 \rightarrow 2Ce^{3+} + C_6H_6O_6 + 2H^+ \qquad (S1)$$



Fig. S15 UV-Vis absorption spectra showing optimization of the ratio (concentrations) of AA-to-Ce⁴⁺ ions.





Fig. S16 MTT assay of HeLa cell viability after incubation with PBS (control) and PEG-Tb-CaWO₄ NCs at the concentration range of 0–400 μ g/mL for 24 h.

Method	Sensing Material	LOD (U/L)	Linear range (U/L)	Reference
Colorimetry Fe/C NS		0.03	0.05 - 6.00	2
Colorimetry	orimetry Ag NPs		0.5 - 225	3
Electrochemistry	AuNPs/Ag ⁺	0.03	0.1 - 120	4
Electrochemistry	Luminol Si NPs	0.8	5 - 50	5
Fluorescence	Au/Ag NCs	0.193	0.5 - 10	6
Fluorescence	(PAA)-Ca ²⁺ (Ce ³⁺) NCs	0.62	5.0 - 200	7
Fluorescence	Au NCs-MnO ₂ NSs	1.5	5-800	8
Fluorescence	CQDs	5.5	18.2-1300	9
Fluorescence	E)-8-((4-methylbenzylidene) amino) napthalen-1-amine (L)	50	400-3000	10
Fluorescence	Allosteric probe (AP)	12	20-150	11
Fluorescence	Pyrococcus furiosus argonaute	2.7	10-1000	12
Fluorescence	PEG- Tb-CaWO₄ NCs	0.5	6 - 40	This Work

Table S1. Comparison of ALP detection method

Table 52 Determination of ALD in fatal basing communication using the proposed method (
Table SZ . Determination of ALP in retai bovine serum samples using the proposed method (n=3)

Sample	Added ALP (U/L)	Detected ALP (U/L)	Recovery (%)	RSD (%)		
1	-	-	-	-		
2	10	10.77 ± 0.02	107.68	1.61		
3	16	13.83 ± 0.03	86.47	1.52		
4	20	16.78 ± 0.05	83.89	2.11		

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