Supporting Information

High-Performance Green Emission from Mn²⁺-Doped 0D OIHMH Crystals for

White LEDs and Anti-Counterfeiting Applications

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Fig. S1 High-resolution XPS full spectra of Zn 2p, Mn 2p, Cl 2p for C₆H₁₄N₂ZnCl4:Mn²⁺.



Fig. S2 The EDS pattern of C₆H₁₄N₂ZnCl4:Mn²⁺.



Fig. S3 The XRD patterns of $C_6H_{14}N_2Zn_{1-x}Mn_xCl_4$ with various Mn^{2+} ions concentrations.



Fig. S4 PL spectra of C₆H₁₄N₂Zn_{1-x}Mn_xCl₄ with various x values under 455 nm

excitation.



Fig. S5 Thermal gravimetric analysis (TGA) curves measured from room temperature to 900 °C at the heating rate of 10 °C min⁻¹ on a synchronous thermal analyzer (TGA/DSC, Mettler, Switzerland).



Fig. S6 XRD of C₆H₁₄N₂Zn_{0.6}Mn_{0.4}Cl₄ after being exposed to air for different months.



Fig. S7 EL spectra of $C_6H_{14}N_2Zn_{0.6}Mn_{0.4}Cl_4$ -based LED under different applied current.



Fig. S8 CCT and CRI of C₆H₁₄N₂Zn_{0.6}Mn_{0.4}Cl₄-based LED under different applied current.

Table S1 Comparison of efficiency and thermal decomposition temperature with

Materials	Emission peak (nm)	PLQY (%)	Thermal decomposition (°C)	Ref	
(C8H20N)2MnCl4	520	87%	300	[1]	
$\{TETA[Pb_2Cl_6]_n:Mn^{2+}$	551	25%	450	[2]	

reported relevant materials.

$Cs_2ZnCl_4:30\%Mn^{2+},10\%Sb^{3+}$	530	64.43%	600	[3]
$M_2CdCl_4{:}Mn^{2+}$	605	87%	297	[4]
$[(CH_3)_4N]_2MnX_4$	523	51%	300	[5]
$C_6H_{14}N_2ZnCl_4:Mn^{2+}$	535	70%	340	This work

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