Photoelectrochemical upcycling of PVC plastic waste for the synthesis of

chlorinated quinolinone derivatives

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1. General Information

All the chemicals were purchased from Aladdin reagent and used without any prior purification. All products were isolated by short chromatography on a silica gel (200-300 mesh) column using petroleum ether (60-90°C) and ethyl acetate unless otherwise noted. ¹H, and ¹³C spectras were recorded on a Bruker Advance 500 and 400 spectrometer at ambient temperature with Chloroform-*d* or DMSO-*d*₆ solvent and tetramethylsilane (TMS) as the internal standard. Analytical thin layer chromatography (TLC) was performed on Merk precoated TLC (silica gel 60 F254) plates. Compounds for HRMS were analyzed by positive mode electrospray ionization (ESI) using Agilent 6530 QTOF mass spectrometer.

2. Experimental Section

2.1 General procedures for photoelectricalchemical reaction properties



To a 25 mL tube was added quinolinone (1) (0.2 mmol), PVC (3.0 equiv. (repeat unit)), nBu_4NBF_4 (0.1 M), N, N-dimethylformamide/THF (6 mL) were added, respectively. Under blue (405 nm) LED irradiation, the reaction mixture was subjected to constant current electrolysis (platinum plate (1 cm × 1 cm) as anode and cathode). After the completion (as indicated by TLC), the water 30 mL was added. Then extracted with ethyl acetate and the collected organic layer was washed with brine, and dried with MgSO₄. The solvent was removed under reduced pressure and further purified by flash chromatography (neutral alumina, petroleum ether/ethyl acetate = 5:1-3:1) to give the desired product.

2.2 Radical inhibition experiment



To a 25 mL tube was added quinolin-2(1*H*)-one (**1**) (0.2 mmol), PVC (3.0 equiv. (repeat unit)), nBu_4NBF_4 (0.1 M), *N*,*N*-dimethylformamide/THF (6 mL) were added, respectively. Under blue (405 nm) LED irradiation, the reaction mixture was subjected to constant current electrolysis (platinum plate (1 cm × 1 cm) as anode and cathode). No target product **2a** was obtained in the presence of the radical scavengers TEMPO, DPE, or BHT, which showed that a radical pathway should be

involved.

2.3 Gram-scale experiment



To a 250 mL tube was added quinolin-2(1*H*)-one (**1**) (8.0 mmol), PVC (3.0 equiv. (repeat unit)), nBu_4NBF_4 (0.1 M), *N*, *N*-dimethylformamide/THF (30 mL) were added, respectively. Under blue (405 nm) LED irradiation, the reaction mixture was subjected to constant current electrolysis (platinum plate (8 cm × 6 cm) as anode and cathode). After the completion (as indicated by TLC), the water 150 mL was added. Then extracted with ethyl acetate and the collected organic layer was washed with brine, and dried with MgSO₄. The solvent was removed under reduced pressure and further purified by flash chromatography (neutral alumina, petroleum ether/ethyl acetate = 5:1-3:1) to give the desired product.

2.4 Isolation of dechlorinated PVC (dPVC) under extended conditions

After electrolysis, dPVC was recovered using the following protocol. The reaction mixture was poured into 200 mL H₂O while stirring to precipitate the polymer. The polymer was collected by filtration and then redissolved in 5 mL THF. The solution was poured into 200 mL methanol while stirring to precipitate the polymer, again. The polymer was collected by filtration, and the dissolution and precipitation processes were repeated one more time. The polymer was collected by filtration and dried under vacuum overnight.



Figure S1 IR spectra of PVC (black), dPVC recovered from reactions (blue)

After the chlorination reaction, an amount of dechlorinated PVC (dPVC) remains in the reaction solution. The FTIR spectra of PVC and dPVC are shown in Figure S1. A new peak was observed at 1725 cm⁻¹, which was due to C=O stretching vibration. According to literature reports,¹ the carbonyl group was formed during the photodegradation of PVC under air atmosphere. Fourier transform infrared spectroscopy could prove this result.



Figure S2 HRMS analysis of DPE adduct



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Entry	Chlorine source	Yield [%] ^b
1	PVC	71
2 ^c	NaCl	84
3 ^c	NH ₄ Cl	81
4 ^c	DCE	78
5 ^c	DCM	69
6 ^d	NCS	0

^a Reaction conditions: **1a** (0.2 mmol), PVC (3.0 equiv.), *n*Bu₄NBF₄ (0.1 M), DMF/THF (3 mL), rt, in air, undivided cell, Pt (+)/Pt (-), constant current = 10 mA, 405 nm LEDs, 8 h. ^b Isolated yields. ^c **1a** (0.2 mmol), chlorine source (3.0 equiv.), *n*Bu₄NBF₄ (0.1 M), DMF (3 mL), rt, in air, undivided cell, Pt (+)/Pt (-), constant current = 10 mA. ^d **1a** (0.2 mmol), NCS (1.5 equiv.), DMF (3 mL), rt, in air.



Scheme S2 Ineffective arenes for the photoelectricalchemical reaction

To expand the substrate scope, we tested some other arenes, such as quinoline (**1ae**), quinoxaline (**1af**), indole (**1ah**), naphthyl amine (**1ai**), and naphthol (**1aj**).Unfortunately, the substrates above were not compatible with this method.

3. Characterization of Products

3-Chloro-1,4-dimethylquinolin-2(1*H*)-one (2a)



Obtained as a yellow solid in 71% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.75 (dd, *J* = 8.1, 1.2 Hz, 1H), 7.58 (ddd, *J* = 8.5, 7.2, 1.4 Hz, 1H), 7.37 (d, *J* = 8.3 Hz, 1H), 7.31 – 7.28 (m, 1H), 3.77 (s, 3H), 2.63 (s, 3H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 158.05, 142.58, 137.97, 130.53, 126.34, 125.51, 122.69, 120.66, 114.59, 30.81, 16.53. HRMS (ESI+): Calculated for C₁₁H₁₁ClNO: [M+H]⁺ 208.0529, Found 208.0523.

3-Chloro-1-ethyl-4-methylquinolin-2(1H)-one (2b)



Obtained as a yellow solid in 70% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.76 (dd, *J* = 8.1, 1.3 Hz, 1H), 7.56 (ddd, *J* = 8.6, 7.2, 1.4 Hz, 1H), 7.39 (d, *J* = 8.5 Hz, 1H), 7.29 – 7.25 (m, 1H), 4.41 (q, *J* = 7.2 Hz, 2H), 2.63 (s, 3H), 1.36 (t, *J* = 7.2 Hz, 3H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 158.17, 143.09, 137.58, 131.07, 127.01, 126.37, 123.08, 121.63, 115.06, 39.32, 17.16, 13.35. HRMS (ESI+): Calculated for C₁₂H₁₃CINO: [M+H]⁺ 222.0686, Found 222.0681.

3-Chloro-4-methyl-1-propylquinolin-2(1H)-one (2c)



Obtained as a yellow solid in 69% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.72 (d, *J* = 8.1 Hz, 1H), 7.53 (t, *J* = 7.8 Hz, 1H), 7.33 (d, *J* = 8.5 Hz, 1H), 7.24 (t, *J* = 7.7 Hz, 1H), 4.30 – 4.23 (m, 2H), 2.59 (s, 3H), 1.74 (dt, *J* = 15.0, 7.5 Hz, 2H), 1.02 (t, *J* = 7.4 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 158.22, 142.96, 137.65, 130.90, 126.79, 126.17, 122.95, 121.36, 115.10, 45.62, 21.34, 17.03, 11.88. HRMS (ESI+): Calculated for C₁₃H₁₅ClNO: [M+H]⁺ 236.0842, Found 236.0836.

3-Chloro-1-isobutyl-4-methylquinolin-2(1H)-one (2d)



Obtained as a yellow solid in 71% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.77 (d, *J* = 8.1 Hz, 1H), 7.59 – 7.53 (m, 1H), 7.39 – 7.35 (m, 1H), 7.28 (t, *J* = 7.6 Hz, 1H), 4.25 (d, *J* = 7.0 Hz, 2H), 2.65 (s, 3H), 2.26 (dp, *J* = 13.5, 6.7 Hz, 1H), 1.01 (d, *J* = 6.7 Hz, 6H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 158.77, 143.00, 138.00, 130.76, 126.91, 126.19, 122.97, 121.39, 115.57, 50.62, 27.92, 20.75, 17.14. HRMS (ESI+): Calculated for C₁₄H₁₇ClNNaO: [M+H]⁺ 250.0999, Found 250.0992.

3-Chloro-4-methyl-1-(prop-2-yn-1-yl) quinolin-2(1H)-one (2e)



Obtained as a yellow solid in 60% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.83 – 7.75 (m, 1H), 7.69 – 7.62 (m, 1H), 7.57 (d, *J* = 8.4 Hz, 1H), 7.36 (t, *J* = 7.6 Hz, 1H), 5.20 (d, *J* = 2.3 Hz, 2H), 2.67 (s, 3H), 2.30 – 2.26 (t, 1H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 157.24, 143.43, 136.56, 130.64, 125.98, 125.68, 123.10, 120.94, 115.07, 77.69, 72.90, 32.96, 16.62. HRMS (ESI+): Calculated for C₁₃H₁₁CINO: [M+H]⁺ 232.0529, Found 232.0522.

1-Benzyl-3-chloro-4-methylquinolin-2(1H)-one (2f)



Obtained as a yellow solid in 57% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.79 (d, *J* = 8.1 Hz, 1H), 7.46 (t, *J* = 7.8 Hz, 1H), 7.34 – 7.25 (m, 7H), 5.64 (s, 2H), 2.71 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 158.94, 143.83, 138.04, 136.75, 131.11, 129.50, 128.08, 127.38, 126.91, 126.20, 123.37, 121.57, 116.09, 47.95, 17.30. HRMS (ESI+): Calculated for C₁₇H₁₅ClNO: [M+H]⁺ 284.0842, Found 284.0837.

3-Chloro-1-(3-chlorobenzyl)-4-methylquinolin-2(1H)-one (2g)



Cl Obtained as a yellow solid in 60% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.73 (s, 1H), 7.40 (d, *J* = 9.0 Hz, 1H), 7.29 – 7.11 (m, 5H), 7.09 (d, *J* = 5.5 Hz, 1H), 5.55 (s, 2H), 2.67 – 2.65 (m, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 157.90, 142.50, 137.82, 135.73, 135.02, 130.61, 130.33, 128.70, 128.01, 127.50, 126.78, 125.23, 124.90, 122.16, 116.63, 47.01, 16.78. HRMS (ESI+): Calculated for C₁₇H₁₄Cl₂NO: [M+H]⁺ 318.0452, Found 318.0446.

3-Chloro-4-methyl-1-(4-methylbenzyl) quinolin-2(1H)-one (2h)



Obtained as a yellow solid in 58% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.71 (d, *J* = 2.2 Hz, 1H), 7.38 (dd, *J* = 9.0, 2.1 Hz, 1H), 7.29 (d, *J* = 3.8 Hz, 1H), 7.12 (s, 5H), 5.55 (s, 2H), 2.65 (t, *J* = 2.0 Hz, 3H), 2.31 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 157.77, 142.02, 137.23, 135.77, 132.60, 130.28, 129.52, 128.23, 127.40, 126.64, 124.82, 121.91, 116.84, 77.48, 76.84, 47.13, 21.03, 16.57. HRMS (ESI+): Calculated for C₁₈H₁₇ClNO: [M+H]⁺ 298.0999, Found 298.0993.

3-Chloro-4-methyl-1-(4-(trifluoromethyl)benzyl)quinolin-2(1H)-one (2i)



 $^{\sf CF_3}$ Obtained as a yellow solid in 56% yield. ¹H NMR (400 MHz, Chloroform-*d*)

δ 7.75 – 7.73 (m, 1H), 7.55 (d, *J* = 6.0 Hz, 2H), 7.42 – 7.37 (m, 1H), 7.28 (dd, *J* = 19.9, 5.1 Hz, 3H), 7.13 (dd, *J* = 8.9, 3.3 Hz, 1H), 5.62 (s, 2H), 2.67 – 2.65 (m, 3H). ¹³**C NMR** (101 MHz, Chloroform-*d*) δ 158.52, 143.17, 140.36, 136.23, 131.24, 129.38, 128.07, 127.62, 126.81, 126.63, 126.60, 125.90, 122.76, 117.09, 47.70, 17.37. **HRMS** (ESI+): Calculated for C₁₈H₁₄ClF₃NO: [M+H]⁺ 352.0716, Found 352.0711. 3-Chloro-1-(4-isopropylbenzyl)-4-methylquinolin-2(1H)-one (2j)



Obtained as a yellow solid in 55% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.82 (d, *J* = 2.3 Hz, 1H), 7.52 – 7.47 (m, 1H), 7.39 (d, *J* = 9.0 Hz, 1H), 7.26 (s, 5H), 5.89 (d, *J* = 185.8 Hz, 2H), 2.97 (p, *J* = 6.8 Hz, 1H), 2.75 (s, 3H), 1.32 (d, *J* = 6.9 Hz, 6H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 157.94, 148.37, 142.10, 135.99, 133.01, 130.44, 128.38, 127.05, 126.86, 126.79, 124.97, 122.08, 116.99, 47.31, 33.82, 24.00, 16.72. HRMS (ESI+): Calculated for C₂₀H₂₁ClNO: [M+H]⁺ 326.1312, Found 326.1307.

3-Chloro-1-(4-fluorobenzyl)-4-methylquinolin-2(1H)-one (2k)



F Obtained as a yellow solid in 56% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.83 – 7.73 (m, 1H), 7.50 – 7.42 (m, 1H), 7.39 – 7.09 (m, 4H), 7.05 (t, *J* = 8.2 Hz, 2H), 5.60 (s, 2H), 2.71 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 163.47, 161.02, 157.93, 142.36, 135.77, 131.48, 130.52, 128.60, 128.52, 127.55, 125.17, 122.15, 116.69, 116.06, 115.84, 46.85, 16.74. HRMS (ESI+): Calculated for C₁₇H₁₄CIFNO: [M+H]⁺ 302.0748, Found 302.0743.

3-Chloro-1-(4-chlorobenzyl)-4-methylquinolin-2(1H)-one (2I)



C^IObtained as a yellow solid in 61% yield. ¹H NMR (400 MHz, Chloroform-*d*)
 δ 7.77 – 7.69 (m, 1H), 7.43 (dd, J = 9.0, 2.0 Hz, 1H), 7.35 – 7.14 (m, 6H), 5.57 (s, 2H), 2.69 (s, 3H).
 ¹³C NMR (101 MHz, Chloroform-*d*) δ 157.95, 142.44, 135.76, 134.28, 133.62, 130.58, 129.23,

128.69, 128.22, 127.57, 125.24, 122.19, 116.67, 46.95, 16.79. HRMS (ESI+): Calculated for $C_{17}H_{14}Cl_2NO$: [M+H]⁺ 318.0452, Found 318.0446.

1-(4-Bromobenzyl)-3-chloro-4-methylquinolin-2(1H)-one (2m)



^{Br} Obtained as a yellow solid in 62% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.77 (s, 1H), 7.56 – 7.26 (m, 4H), 7.21 (d, *J* = 8.9 Hz, 1H), 7.13 (d, *J* = 7.6 Hz, 2H), 5.56 (s, 2H), 2.69 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 157.92, 142.46, 135.67, 134.77, 132.13, 130.57, 128.66, 128.52, 127.48, 125.21, 122.13, 121.62, 116.65, 46.95, 16.78. HRMS (ESI+): Calculated for C₁₇H₁₄BrClNO: [M+H]⁺ 361.9947, Found 361.9941.

3-Chloro-1-(cyclopropylmethyl)-4-methylquinolin-2(1H)-one (2n)



Obtained as a yellow solid in 45% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.65 (d, *J* = 8.0 Hz, 1H), 7.49 – 7.39 (m, 2H), 7.17 (t, *J* = 7.5 Hz, 1H), 4.19 (d, *J* = 6.9 Hz, 2H), 2.52 (s, 3H), 0.79 – 0.73 (m, 1H), 0.51 – 0.39 (m, 4H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 158.59, 143.07, 137.95, 130.90, 127.00, 126.18, 122.99, 121.38, 115.41, 78.02, 17.11, 10.39, 4.67. HRMS (ESI+): Calculated for C₁₄H₁₅CINO: [M+H]⁺ 248.0842, Found 248.0836.

3-Chloro-1,6-dimethylquinolin-2(1H)-one (2o)



Obtained as a yellow solid in 58% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 8.37 (s, 1H), 7.41 (d, *J* = 8.7 Hz, 1H), 7.40 – 7.17 (m, 1H), 7.16 (d, *J* = 8.7 Hz, 1H), 3.75 (d, *J* = 2.9 Hz, 3H), 2.44 (s, 3H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 158.56, 138.87, 133.63, 133.23, 132.41, 131.59, 128.68, 118.58, 113.31, 31.85, 20.68. HRMS (ESI+): Calculated for C₁₁H₁₁ClNO: [M+H]⁺ 208.0529, Found 208.0523. 3-Chloro-1,7-dimethylquinolin-2(1H)-one (2p)



Obtained as a yellow solid in 56% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 8.41 (s, 1H), 7.45 (d, *J* = 8.7 Hz, 1H), 7.43 – 7.22 (m, 1H), 7.21 (d, *J* = 8.7 Hz, 1H), 3.80 (s, 3H), 2.34 (s, 3H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 157.89, 138.20, 132.96, 132.55, 131.73, 130.92, 128.01, 117.91, 112.64, 31.18, 20.13. HRMS (ESI+): Calculated for C₁₁H₁₁ClNO: [M+H]⁺ 208.0529, Found 208.0521.

6-Bromo-3-chloro-1-methylquinolin-2(1H)-one (2q)



H Obtained as a yellow solid in 43% yield. ¹H NMR (400 MHz, DMSO- d_6) δ 12.40 (s, 1H), 8.23 (s, 1H), 7.88 (d, J = 2.0 Hz, 1H), 7.64 (dd, J = 8.8, 2.2 Hz, 1H), 7.26 (d, J = 8.8 Hz, 1H). ¹³C NMR (101 MHz, DMSO- d_6) δ 157.28, 136.66, 136.51, 132.99, 129.25, 127.01, 120.38, 117.29, 113.92. HRMS (ESI+): Calculated for C₉H₆BrCINO: [M+H]⁺ 257.9321, Found 257.9315.

3-Chloro-4-methylquinolin-2(1H)-one (2r)



H Obtained as a yellow solid in 41% yield. ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.18 (s, 1H), 7.81 (d, *J* = 8.1 Hz, 1H), 7.54 (t, *J* = 7.7 Hz, 1H), 7.35 (d, *J* = 8.1 Hz, 1H), 7.26 (t, *J* = 7.6 Hz, 1H), 2.59 (s, 3H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 156.95, 143.98, 136.62, 130.42, 125.17, 125.05, 122.30, 119.04, 115.51, 16.09. HRMS (ESI+): Calculated for C₁₀H₉CINO: [M+H]⁺ 194.0373, Found 194.0365.

3-Chloro-1-methylquinoxalin-2(1*H*)-one (2s)²



Obtained as a white solid in 51% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.83 (dd, *J* = 8.0, 1.3 Hz, 1H), 7.62 (ddd, *J* = 8.6, 7.4, 1.5 Hz, 1H), 7.40 (td, *J* = 8.0, 7.5, 1.2 Hz, 1H), 7.35 (dd, *J* = 8.4 Hz, 0.8Hz, 1H), 3.78 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 151.72, 148.79, 133.26, 131.68, 131.07, 129.66, 124.45, 113.99, 30.57. HRMS (ESI+): Calculated for C₉H₈ClN₂O: [M+H]⁺ 195.0320, Found 195.0322.

3-Chloro-1-ethyquinoxalin-2(1H)-one (2t)²



Obtained as a white solid in 47% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.83 (d, *J* = 8.0 Hz, 1H), 7.62 (td, *J* = 7.8, 1.4 Hz, 1H), 7.42 – 7.34 (m, 2H), 4.38 (q, *J* = 7.2 Hz, 2H), 1.42 (t, *J* = 7.2 Hz, 3H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 151.19, 148.83, 132.22, 132.01, 131.05, 129.93, 124.26, 113.83, 38.92, 12.34. HRMS (ESI+): Calculated for C₁₀H₁₀ClN₂O: [M+H]⁺ 209.0476, Found 209.0471.

3-Chloro-1-vinylquinoxalin-2(1H)-one (2u)³



Obtained as a brown solid in 41% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.81 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.60 – 7.54 (m, 1H), 7.40 – 7.34 (m, 1H), 7.32 (d, *J* = 8.5 Hz, 1H), 5.92 (ddt, *J* = 17.1, 10.4, 5.2 Hz, 1H), 5.30 (d, *J* = 10.2 Hz, 1H), 5.21 (d, *J* = 17.2 Hz, 1H), 4.94 (d, *J* = 5.2 Hz, 2H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 151.29, 148.76, 132.51, 131.84, 130.97, 129.93, 129.73, 124.41, 118.82, 114.54, 45.93. HRMS (ESI+): Calculated for C₁₁H₉ClN₂O: [M+H]⁺ 221.0476, Found 221.0472.

3-Chloro-1-ethynylquinoxalin-2(1H)-one (2v)³



Obtained as a white solid in 49% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.82 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.68 – 7.62 (m, 1H), 7.50 (dd, *J* = 8.4, 0.8 Hz, 1H), 7.44 – 7.39 (m, 1H), 5.09 (d, *J* = 2.5 Hz, 2H), 2.33 (s, 1H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 150.81, 148.44, 131.81, 131.16, 129.78, 124.80, 114.53, 76.02, 73.99, 32.97. HRMS (ESI+): Calculated for C₁₁H₇ClN₂O: [M+H]⁺ 219.0320, Found 219.0315.

3-Chloro-6-fluoro-1-methylquinoxalin-2(1*H*)-one (2w)²



Obtained as a white solid in 46% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.53 (dd, *J* = 8.4, 2.8 Hz, 1H), 7.40 – 7.35 (m, 1H), 7.33 – 7.29 (m, 1H), 3.78 (s, 3H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 158.98, 151.32, 150.51, 132.16, 129.99, 118.90, 115.31, 115.17, 30.85. HRMS (ESI+): Calculated for C₉H₇CIFN₂O: [M+H]⁺ 213.0225, Found 213.21.

3,6-Dichloro-1-methylquinoxalin-2(1*H*)-one (2x)²



Obtained as a white solid in 45% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.81 (d, *J* = 2.3 Hz 1H), 7.58 (dd, *J* = 9.0, 2.4 Hz, 1H), 7.29 (d, *J* = 9.0 Hz, 1H), 3.77 (s, 3H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 151.36, 150.28, 150.51, 132.09, 131.95, 131.07, 129.87, 128.94, 115.10, 30.75. HRMS (ESI+): Calculated for C₉H₆Cl₂N₂NaO: [M+Na]⁺ 250.9749, Found 250.9762.

6-Bromo-3-chloro-1-methylquinoxalin-2(1*H*)-one (2y)²



Obtained as a white solid in 43% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.78 (d, *J* = 2.2 Hz, 1H), 7.54 (dd, *J* = 8.9, 2.2 Hz, 1H), 7.28 (d, *J* = 8.9 Hz, 1H), 3.73 (s, 3H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 151.36, 150.29, 132.09, 131.95, 131.07, 129.87, 128.95, 115.09, 30.75. HRMS (ESI+): Calculated for C₉H₆BrClN₂NaO: [M+Na]⁺ 294.9244, Found 294.9237.

3-Chloro-6,7-difluoro-1-methylquinoxalin-2(1*H*)-one (2z)²



Obtained as a white solid in 42% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.67 – 7.63 (m, 1H), 7.18 - 7.13 (m, 1H), 3.74 (s, 3H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 151.92, 151.29, 149.41, 147.11, 130.61, 127.77, 117.37, 102.80, 31.11. HRMS (ESI+): Calculated for C₉H₅ClF₂N₂NaO: [M+Na]⁺ 252.9951, Found 252.9976.

3,6,7-Trichloro-1-methylquinoxalin-2(1H)-one (2aa)²



Obtained as a white solid in 41% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ
 7.90 (s, 1H), 7.44 (s, 1H), 3.73 (s, 3H). ¹³C NMR (100 MHz, Chloroform-*d*) δ 151.12, 150.30, 135.39,
 132.57, 130.56, 130.36, 128.47, 115.50, 30.83. HRMS (ESI+): Calculated for C₉H₅Cl₃N₂NaO: [M+Na]⁺
 284.9360, Found 284.9399.

3-Chloro-1,6,7-trimethylquinoxalin-2(1H)-one (2ab)



Obtained as a yellow solid in 47% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.53 (s, 1H), 7.09 (s, 1H), 3.74 (d, *J* = 2.42 Hz, 3H), 2.44 (s, 3H), 2.36 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) 151.77, 147.41, 141.25, 133.62, 131.27, 130.04, 129.52, 114.48, 30.48, 20.75, 19.27. HRMS (ESI+): Calculated for C₁₁H₁₁ClN₂O: [M+Na]⁺ 223.0633, Found 223.0637.

3-Chloro-1-methyl-5,6-diphenylpyrazin-2(1H)-one (2ac)



Obtained as a yellow solid in 56% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.44 – 7.35 (m, 3H), 7.24 – 7.18 (m, 2H), 7.15 – 7.07 (m, 5H), 3.37 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 152.88, 145.84, 138.53, 136.37, 132.14, 131.77, 130.02, 129.98, 129.37, 129.27, 127.98, 127.53, 35.54. HRMS (ESI+): Calculated for C₁₇H₁₄ClN₂O: [M+H]⁺ 297.0795, Found 297.0791.

3-Chlorocinnolin-4(1H)-one (2ad)



H Obtained as a white solid in 40% yield. ¹H NMR (400 MHz, DMSO- d_6) δ 8.05 (q, J = 8.95 Hz), 7.92 – 7.75 (m), 7.60 (q, J = 9.42 Hz), 7.46 (t, J = 9.51 Hz). ¹³C NMR (101 MHz, DMSO- d_6) δ 165.86, 141.68, 141.21, 134.83, 126.12, 124.97, 117.24. HRMS (ESI+): Calculated for C₈H₆ClN₂O:

[M+H]⁺ 181.0169, Found 181.0167.

2,6-Dichloro-8-cyclopentyl-5-methylpyrido[2,3-d]pyrimidin-7(8H)-one (2ag)



Obtained as a white solid in 51% yield.¹H NMR (400 MHz, Chloroform-*d*) δ 8.87 (s, 1H), 6.23 – 5.69 (m, 1H), 2.65 (s, 3H), 2.33 – 2.12 (m, 4H), 1.97 (m, *J* = 10.68, 6.68 Hz, 2H), 1.82 – 1.60 (m, 2H). HRMS (ESI+): Calculated for C₁₃H₁₄Cl₂N₃O: [M+H]⁺ 299.1750, Found 299.1746.

1,4-Dimethyl-3-phenylquinolin-2(1H)-one (3a)



Obtained as a yellow solid in 91% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.94 (d, *J* = 2.0 Hz, 1H), 7.65 – 7.57 (m, 3H), 7.47 – 7.37 (m, 3H), 7.34 – 7.25 (m, 3H), 3.76 (s, 3H), 2.64 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 161.67, 140.94, 138.29, 138.05, 137.28, 133.10, 129.30, 129.12, 128.77, 128.70, 127.46, 124.11, 123.51, 116.51, 115.80, 30.54, 16.82. HRMS (ESI+): Calculated for C₁₇H₁₆NO: [M+H]⁺ 250.1232, Found 250.1238.

(E)-1,4-Dimethyl-3-styrylquinolin-2(1H)-one (4a)



Obtained as a yellow solid in 85% yield. ¹H NMR (400 MHz, Chloroform-*d*) δ 8.02 – 7.98 (m, 1H), 7.82 (dd, *J* = 8.7, 2.0 Hz, 1H), 7.66 (d, *J* = 7.7 Hz, 2H), 7.48 (dt, *J* = 10.4, 7.2 Hz, 5H), 7.39 (d, *J* = 1.9 Hz, 1H), 7.32 (d, *J* = 1.2 Hz, 1H), 3.80 (s, 3H), 2.39 (s, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 162.20, 142.98, 140.90, 139.07, 137.39, 135.65, 133.46, 130.66, 129.68, 129.58, 128.82, 128.10, 127.99, 127.64, 124.57, 122.45, 115.34, 30.52, 17.56. HRMS (ESI+): Calculated for C₁₉H₁₈NO: [M+H]⁺ 276.1388, Found 276.1384.

Palbociclib



Obtained as a white solid in 75% yield. ¹H NMR (500

MHz, Chloroform-*d*) δ 9.14 (s, 1H), 8.91 (s, 1H), 8.17 (d, *J* = 9.1 Hz, 1H), 8.11 (d, *J* = 2.8 Hz, 1H), 7.33 (dd, *J* = 9.1, 3.0 Hz, 1H), 5.93 – 5.86 (m, 1H), 3.16 (dd, J = 6.2, 3.6 Hz, 4H), 3.08 (dd, *J* = 6.2, 3.7 Hz, 4H), 2.55 (s, 3H), 2.39 (s, 3H), 2.36 (dd, *J* = 14.0, 6.0 Hz, 2H), 2.07 (dq, *J* = 13.2, 7.4 Hz, 2H), 1.92 – 1.86 (m, 2H), 1.79 (s, 1H), 1.70 (dt, *J* = 10.3, 4.7 Hz, 2H). **HRMS** (ESI+): Calculated for C₂₄H₃₀N₇O₂: [M+H]⁺ 448.2461, Found 448.2453.

9,10-Diphenyl-9,10-dihydro-9,10-epidioxyanthracene (5a)⁴



 $\dot{P}h$ Obtained as a white solid in 18% yield. ¹H NMR (500 MHz, Chloroform-*d*) δ 7.70 (d, *J* = 7.3 Hz, 4H), 7.63 (t, *J* = 7.6 Hz, 4H), 7.19 (ddt, *J* = 9.1, 6.2, 3.2 Hz, 8H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 140.27, 133.03, 128.38, 128.30, 127.68, 127.56, 123.54, 84.12. 4. ¹H and ¹³C NMR Spectra of These Compounds



¹³C NMR (126 MHz, Chloroform-d) Spectra of compound 2a



¹³C NMR (126 MHz, Chloroform-*d*) Spectra of compound 2b



¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 2c



¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 2d



¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 2e



¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2f



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¹³C NMR (101 MHz, DMSO-*d*₆) Spectra of compound 2g



¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2h



¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2i



¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 2j



¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 2k



¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2I

¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2m

¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2n

¹³C NMR (126 MHz, Chloroform-d) Spectra of compound 20

¹³C NMR (126 MHz, Chloroform-*d*) Spectra of compound 2p

¹³C NMR (101 MHz, DMSO-*d*₆) Spectra of compound **2q**

¹³C NMR (101 MHz, DMSO-d₆) Spectra of compound 2r

¹³C NMR (101 MHz, Chloroform-*d*) Spectra of compound 2ab

¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 2ac

¹³C NMR (101 MHz, DMSO-*d*₆) Spectra of compound 2ad

¹H NMR (400 MHz, Chloroform-d) Spectra of compound 3a

¹H NMR (400 MHz, Chloroform-*d*) Spectra of compound 4a

¹³C NMR (101 MHz, Chloroform-d) Spectra of compound 4a

¹H NMR (500 MHz, Chloroform-d) Spectra of compound Palbociclib

5. References

- (1) (a) J. Li, D. Zhou and D. Zhao, The photo-degradation of PVC: part II—structural changes in PVC chains. *Polym. Degrad. Stabil.*, 1991, **31**, 1-7; (b) M. Balandier and C. Decker, Photodegradation and photo-oxidation of poly(vinyl chloride) in solution. *Eur. Polym. J.*, 1978, **14**, 995-1000; (c) D. Jin, S. Khanal, C. Zhang and S. Xu, Photodegradation of polybenzimidazole/polyvinyl chloride composites and polybenzimidazole: density functional theory and experimental study. *J. Appl. Polym. Sci.*, 2021, **138**, 46693; (d) V. Najafi, E. Ahmadi, F. Ziaee, H. Omidian and H. Sedaghat, Polyaniline-modified TiO₂, a highly effective photo-catalyst for solid phase photocatalytic degradation of PVC. *J. Polym. Environ.*, 2019, **27**, 784-793.
- (2) M. -C. Wu, M. -Z. Li, J. -Y. Chen, J. -A. Xiao, H. -Y. Xiang. K. Chen and H. Yang, Photoredox-catalysed chlorination of quinoxalin-2(1*H*)-ones enabled by using CHCl₃ as a chlorine source. *Chem. Commun.*, 2022, **58**, 11591-11594.
- (3) D. Yu, R. Ji, Z. Sun, W. Li and Z. -Q. Liu, Electrochemical chlorination and bromination of electron-deficient C-H Bonds in quinones, coumarins, quinoxalines and 1,3-diketones. *Tetrahedron Lett.*, 2021, **86**, 153514.
- (4) X. Shi, Y. Cao, Y. Liu, K. Niu, H. Song, J. Zhang and Q. Wang, Catalyst-free visiblelight-Induced decarbonylative C–H alkylation of quinoxalin-2(1*H*)-ones. *Org. Chem. Front.*, 2023, **10**, 1296-1300.