Total Synthesis and ¹³C NMR Revision of Nagelamide C

Guanghu Tong,^{*ab} Long V. Nguyen^a and Timothy F. Jamison^a

^{*a*}Department of Chemistry, Massachusetts Institute of Technology Cambridge, 77 Massachusetts 02139, United States.

^bMedicinal Chemistry, Gilead Sciences, Foster City, California 94403, United States.

*Correspondence to: Guanghu.Tong1@gilead.com

Table of Contents

1. General methods	S3
2. Nagelamide alkaloids	S4
3. Attempted approaches to construct imidazo[1,2-a]pyrimidine derivatives	S6
4. Experimental procedures and characterization data	S7
4.1 Sonogashira coupling	S7
4.2 trans-Hydrostannylation	S9
4.3 cis-Hydrostannylation	S 11
4.4 Wittig salt formation	S11
4.5 Imidazole benzylic Wittig olefination	S12
4.6 NBS bromination	S14
4.7 Stille coupling of diene 11b'/11d' and bromide 12	S14
4.8 Synthesis of nagelamide C	S15
4.9 Stille coupling of diene 11a' and bromide 12	S21
4.10 Synthesis of nagelamide C <i>E</i> -isomer	S21
5. NMR Spectra	S25
6. References	S 90

1. General methods

Unless otherwise noted, all reactions were performed using flame-dried glassware under an atmosphere of argon. Air- and moisture-sensitive liquids and solutions were transferred via syringe or stainless steel cannula. Organic solutions were concentrated under reduced pressure by rotary evaporation at 35 °C. Reactions were monitored by thin-layer chromatography (TLC) using SiliCycle glass plates pre-coated with silica gel (0.25-mm thickness; 60-Å pore size) impregnated with a fluorescent indicator (254 nm). TLC plates were visualized by exposure to ultraviolet light (UV) and staining with a solution of ninhydrin, ethanolic solution of phosphomolybdic acid (PMA), acidic solution of anisoaldehyde, or aqueous solution of potassium permanganate followed by heating. Solid reagents were weighed out in air unless otherwise noted. Reaction work-ups and chromatographic purifications were performed on the bench top in air.

Materials

Dichloromethane (DCM, HPLC grade), methanol (MeOH, reagent grade), hexanes (HPLC grade), ethyl acetate (EtOAc, HPLC grade), chloroform (HPLC grade), acetone (HPLC grade) were purchased from Sigma and used as received. Anhydrous solvents (tetrahydrofuran (THF), dimethylformamide (DMF), triethylamine (Et₃N) were first degassed by sparging with nitrogen and dried by passing through a column of activated alumna on an SG Water solvent purification system. Deionized water was obtained from an in-house system. Silica gel was purchased from SiliCycle (40 μ m) and used as received. Unless otherwise noted, all chemicals were purchased from commercial sources at the highest available purity and used as received.

Instrumentation

¹H, ¹³C, ¹⁹F and ³¹P NMR spectra were recorded at ambient temperature on 400, 500, 600 MHz Bruker or 500 MHz JEOL NMR spectrometers. Hydrogen chemical shifts are expressed in parts per million (ppm) relative to the residual protio-solvent resonance: CDCl₃ δ 7.26, CD₃OD δ 3.31, DMSO-d₆ δ 2.50. For ¹³C spectra, the centerline of the solvent signal was used as internal reference: CDCl₃ δ 77.16, CD₃OD δ 49.00, DMSO-d₆ δ 39.50. Data are reported as follows: chemical shifts, multiplicity (s = singlet, d = doublet, t = triplet, dd = doublet of doublets, dt = doublet of triplets, dq = doublet of quartets, ddd = doublet of doublet of doublets, and m = multiplet, br = broad), coupling constant (J) in hertz (Hz), and integration. High resolution mass spectrometry (HRMS) were acquired in on JEOL AccuTOF (Direct Analysis in Real Time, DART) and Agilent 6545 LC/QTOF MS in the positive mode. HPLC analysis was performed on an Agilent 1260 Infinity II using an Agilent reverse-phase C18 column with a mobile phase of acetonitrile/water/0.1%

2. Nagelamide alkaloids



Nagelamide L

Nagelamide M

Nagelamide N

continued





Nagelamide Z

Figure S1. Nagelamide alkaloids¹⁻¹⁰



3. Attempted approaches to construct imidazo[1,2-a]pyrimidine derivatives a. Wittig reaction approach

Scheme S1. a. Wittig olefination approach, b. Julia-Kocienski olefination approach, c. Crossmetathesis approach, d. Suzuki approach

4. Experimental procedures and characterization data

4.1 Sonogashira coupling



Mono-Boc protected **18** was prepared from **13** and **14** according to the previous literature.¹¹ However, we observed that starting material **13** showed less reactivity with **14'** in the presence of $Pd(PPh_3)_2Cl_2$. A modified procedure was used: to a mixture of imidazo[1,2-a]pyrimidine **13** (2.3 g, 11.7 mmol, 1.0 equiv) and $Pd(PPh_3)_4$ (675 mg, 0.58 mmol, 5 mol %), CuI (222 mg, 1.17 mmol, 10 mol %), Et₃N (8.2 mL, 58.4 mmol, 5.0 equiv) in DMF (47 mL) was slowly added a solution of alkyne **14'** (4.47 g, 17.5 mmol, 1.5 equiv) in DMF (10 mL) at room temperature. The resulting solution was warmed to 50 °C and stirred for 2.5 h. The reaction mixture was concentrated in *vacuo*, and purified by column chromatography (silica gel, EtOAc:MeOH 20:1) to afford alkyne **18'** (3.3 g, 9 mmol, 77% yield) as yellow solid. **SI-1** (10% yield) was isolated as a yellow solid.

R_f 0.45 (silica gel, EtOAc:MeOH 20:1), UV active, stains brown in ninhydrin.

18':

¹**H NMR (400 MHz, CDCl₃):** δ 8.61 – 8.56 (m, 2H), 7.96 (s, 1H), 6.98 (dd, *J* = 6.8, 4.1 Hz, 1H), 4.71 (s, 2H), 1.54 (s, 18H).

¹³C NMR (151 MHz, CDCl₃): δ 151.96, 150.89, 148.50, 139.64, 132.97, 109.46, 107.23, 96.27, 83.56, 69.24, 36.88, 28.23.

The ¹H and ¹³C NMR data of **18**' are consistent with the reported data in literature.¹¹

Table S1. Optimization of Sonogashira coupling^a



	14' (equiv)	Conditions	Result (13: 18': SI-1)
1	1.1	Pd(PPh ₃) ₂ Cl ₂ (5 mol%), CuI (10 mol%), Et ₃ N (5 equiv), THF, 50 ^o C, 16 h	84: 16: 0
2	1.5	Pd(PPh ₃) ₂ Cl ₂ (5 mol%), CuI (10 mol%), Et ₃ N (5 equiv), THF, 50 ^o C, 16 h	73: 22: 5
3	1.1	Pd(PPh ₃) ₂ Cl ₂ (5 mol%), CuI (10 mol%), Et ₃ N (5 equiv), DMF, 50 ^o C, 16 h	69: 31: 0
4	1.1	Pd(PPh ₃) ₂ Cl ₂ (5 mol%), CuI (10 mol%), Et ₃ N (5 equiv), DMF, rt, 16 h	87: 13: 0
5	1.1	Pd(PPh ₃) ₄ (5 mol%), CuI (10 mol%), Et ₃ N (5 equiv), THF, 50 ^o C, 16 h	42: 46: 12
6	1.1	Pd(PPh ₃) ₄ (5 mol%), CuI (10 mol%), Et ₃ N (5 equiv), DMF, 50 ^o C, 16 h	57: 43: 0
7 ^b	1.5	Pd(PPh₃)₄ (5 mol%), CuI (10 mol%), Et₃N (5 equiv), DMF, 50 ^o C, 16 h	9: 77: 10
8	2.0	Pd(PPh ₃) ₄ (5 mol%), Cul (10 mol%), Et ₃ N (5 equiv), DMF, 50 ^o C, 16 h	10: 78: 12

^a0.1 mmol scale, ratio was based on crude ¹H NMR, 1,3,5-trimethoxylbenzene as internal standard. ^b2 g scale, isolated yield

4.2 trans-Hydrostannylation



18' То а (745 2.0 1.0 solution of alkyne mg, mmol, equiv) and dichloro(pentamethylcyclopentadienyl)ruthenium(III) polymer (61.4 mg, 0.2 mmol, 10 mol %) in DCM (20 mL) was slowly added *n*Bu₃SnH (0.65 mL, 2.4 mmol, 1.2 equiv) at room temperature. The brown solution quickly turned to black. The resulting solution was stirring for 1 h. At this point, a second portion of *n*Bu₃SnH (0.65 mL, 2.4 mmol, 1.2 equiv) was added and stirred for another 1 h. TLC indicated the starting material consumption. The reaction mixture was concentrated in *vacuo* and purified by column chromatography (silica gel, DCM to DCM:EtOAc = 1:1) to give 11b'/11d' as yellow oil (1.33 g, 1.8 mmol, 89% yield, 2:1 inseparable mixture).

R_f 0.54 (silica gel, EtOAc:DCM 1:1), UV active, stains brown in ninhydrin.

11b':

¹**H NMR (600 MHz, CDCl₃):** δ 8.47 (dd, J = 4.0, 2.0 Hz, 1H), 8.37 (dd, J = 6.9, 2.0 Hz, 1H), 7.46 (s, 1H), 6.79 (dd, J = 6.9, 4.0 Hz, 1H), 6.47 (t, J = 5.6 Hz, 1H), 4.40 (d, J = 5.6 Hz, 2H), 1.51 (s, 18H), 1.43 - 1.38 (m, 6H), 1.27 - 1.21 (m, 6H), 1.01 - 0.95 (m, 6H), 0.83 (t, J = 7.3 Hz, 9H).

¹³C NMR (151 MHz, CDCl₃): δ 152.79, 148.60, 148.15, 146.49, 132.27, 131.99, 131.27, 127.89, 108.21, 82.95, 49.24, 29.13, 28.25, 27.38, 13.72, 11.21.

HRMS (ESI): m/z calc'd for C₃₁H₅₃N₄O₄Sn [M+H]⁺: 665.3089, found: 665.3100.

11d':

¹**H NMR (600 MHz, CDCl₃):** δ 8.55 (dd, J = 4.1, 2.0 Hz, 1H), 8.18 (dd, J = 6.8, 2.0 Hz, 1H), 7.66 (d, J = 1.0 Hz, 1H), 6.89 (q, J = 1.8 Hz, 1H), 6.87 (dd, J = 6.8, 4.1 Hz, 1H), 4.55 (d, J = 2.1 Hz, 2H), 1.51 (s, 18H), 1.34 – 1.28 (m, 6H), 1.20 – 1.14 (m, 6H), 0.93 – 0.87 (m, 6H), 0.79 (t, J = 7.3 Hz, 9H).

¹³C NMR (151 MHz, CDCl₃): δ 152.62, 152.59, 149.43, 148.44, 133.27, 131.01, 123.36, 120.11, 108.68, 82.64, 53.82, 29.08, 28.22, 27.36, 13.67, 10.64.

Table S2. Optimization of *trans*-hydrostannylation^a

_

22

Boc, Boc

$-NR^1R^2$		-NR ¹ R ²						
	li.	Bu₃Sn	Bu ₃ Sn	NR'R ²	SnBu ₃ /	Γ	NR'R ²	
		<i>n</i> Bu ₃ SnH (1.2 equiv)	2 ²			SI	nBu ₃	
					N (
	18. R ¹ = H. R ² = I	Boc $11a, R^1 = H, R^2 = Boc$	11b	1	1c	11d		
	18', R ¹ = R ² = Bo	c $11a', R^1 = R^2 = Boc$	11b'	1	1c'	11d'		
	R ¹ , R ²	Conditions	Yield%	11a%	11b%	11c%	11d%	
1	H, Boc	Pd(PPh ₃) ₂ Cl ₂ (10 mol%), DCM, rt	92	100	0	0	0	
2	H, Boc	[Cp*RuCl]₄ tetramer (10 mol%), DCM, rt	91	5	23	22	50	
3	H, Boc	[Cp*RuCl ₂] _n polymer (10 mol%), DCM, rt	84	6	17	21	56	
4	H, Boc	Cp*Ru(cod)Cl (10 mol%), DCM, rt	88	39	12	18	31	
5	H, Boc	[Cp*Ru(MeCN)₃]PF ₆ (10 mol%), DCM, rt	65	64	0	36	0	
6	H, Boc	[CpRu(MeCN) ₃]PF ₆ (10 mol%), DCM, rt	69	51	12	37	0	
	R ¹ , R ²	Conditions	Yield%	11a'%	11b'%	11c'%	11d'%	
7	Boc, Boc	[Cp*RuCl] ₄ tetramer (10 mol%), DCM, rt	96	20	50	0	30	
8	Boc, Boc	[Cp*RuCl ₂] _n polymer (10 mol%), DCM, rt	89 ^c	5	63	0	32	
9	Boc, Boc	[Cp*Ru(MeCN)₃]PF₀ (10 mol%), DCM, rt	72	24	47	0	29	
10	Boc, Boc	[Cp*Ru(MeCN)₃]OTf (10 mol%), DCM, rt	NR	-	-	-	-	
11	Boc, Boc	[CpRu(MeCN) ₃]PF ₆ (10 mol%), DCM, rt	87	71	19	0	10	
12	Boc, Boc	Cp*Ru(cod)Cl (10 mol%), DCM, rt	82	55	34	0	11	
13	Boc, Boc	[Ru(benzene)Cl ₂] ₂ (10 mol%), DCM, rt	69	100	0	0	0	
14	Boc, Boc	[Ru(p-cymene)Cl ₂] ₂ (10 mol%), DCM, rt	80	100	0	0	0	
15	Boc, Boc	MgBu₂ (1.0 equiv), THF, 50 °C	decom.	-	-	-	-	
16	Boc, Boc	ZrCl₄ (1.0 equiv), THF, rt	52 ^d	-	-	-	-	
17	Boc, Boc	AIBN (10 mol%), THF, 70 °C	53	100	0	0	0	
18	Boc, Boc	[Cp*RuCl ₂] _n polymer (10 mol%), DCE, rt	82	6	62	0	32	
19	Boc, Boc	[Cp*RuCl ₂] _n polymer (10 mol%), CHCl ₃ , rt	34	12	50	0	38	
20	Boc, Boc	[Cp*RuCl ₂] _n polymer (10 mol%), THF, rt	91	10	60	0	30	
21	Boc, Boc	[Cp*RuCl ₂] _n polymer (10 mol%), MeCN, rt	21	19	50	0	31	

 23
 Boc, Boc
 Pd(PPh_3)_2Cl_2 (10 mol%), DCM, rt
 93e
 100
 0
 0
 0

 *0.1 mmol scale, ratio in the crude reaction mixture was determined by ¹H NMR spectroscopy, 1,3,5-trimethoxylbenzene as internal standard.
 18 and 18' were prepared from *N*-Boc-propargylamine and *N,N*-bisBoc propargylamine, respectively. "The product was isolated in gram scale, 2.0 equiv *n*-Bu₃SnH was used. ^d18 was isolated as the product, ^eisolated yield.

[Cp*RuCl₂]_n polymer (10 mol%), acetone, rt

80

8

61

0

31



4.3 cis-Hydrostannylation



To a mixture of alkyne **18'** (37 mg, 0.1 mmol, 1.0 equiv) and Pd(PPh₃)₂Cl₂ (7 mg, 0.01 mmol, 10 mol %) in DCM (1 mL) was slowly added *n*Bu₃SnH (32 μ L, 0.12 mmol, 1.2 equiv) at room temperature. The resulting solution was stirred for 1 h. The reaction mixture was concentrated in *vacuo*, and purified by column chromatography (silica gel, EtOAc) to afford **11a'** (62 mg, 0.093 mmol, 93% yield) as yellow oil.

R_f 0.54 (silica gel, EtOAc), UV active, stains brown in ninhydrin.

¹**H NMR (600 MHz, CDCl₃):** δ 8.56 (dd, J = 4.1, 2.0 Hz, 1H), 8.13 (dd, J = 6.8, 2.0 Hz, 1H), 7.50 (s, 1H), 6.91 (dd, J = 6.8, 4.1 Hz, 1H), 6.13 (t, J = 5.4 Hz, 1H), 4.04 (d, J = 5.4 Hz, 2H), 1.45 (s, 18H), 1.42 - 1.33 (m, 6H), 1.25 - 1.18 (m, 6H), 0.90 - 0.85 (m, 6H), 0.82 (t, J = 7.3 Hz, 9H).

¹³C NMR (151 MHz, CDCl₃): δ 152.37, 149.31, 147.72, 146.96, 131.58, 131.11, 124.69, 108.72, 82.86, 46.77, 28.97, 28.17, 27.32, 13.69, 10.57 (one carbon was not found).

HRMS (**DART**): calc'd for C₃₁H₅₃N₄O₄Sn [M+H]⁺: 665.3089, found 665.3085.

4.4 Wittig salt formation



A 250 mL round bottom flask was charged with a well-stirred magnetic stir bar. 1,3dichloroacetone **16** (10.0 g, 7.9 mmol, 1.5 equiv) was added dropwise to a suspension of 2aminopyrimidine **15** (5.0 g, 5.2 mmol, 1.0 equiv) in THF (50 mL). The resulting mixture was stirred under reflux for 16 h. The resulting slurry was cooled to room temperature and filtered. The yellow solid was washed thoroughly with THF (3×20 mL), and then dried in *vacuo* to provide salt **19** as yellow solid which was used for the next step.

To a 250 mL round bottom flask charged with above salt **19** in MeCN (30 mL) was added triphenylphosphine (20.7 g, 7.9 mmol, 1.5 equiv) at room temperature. The yellow suspension was heated at 85 °C for 12 h. A large amount of purple solid formed at which point (~15 h) the solvent was removed under reduced pressure. The resulting suspension was filtered, washed with acetone (3×50 mL), and dried under air to afford **20** (13.5 g, 2.9 mmol, 56%, two steps) as pink solid.

¹**H NMR (500 MHz, DMSO-d₆):** δ 9.02 (dd, J = 6.7, 2.0 Hz, 1H), 8.57 (dd, J = 4.1, 2.0 Hz, 1H), 7.90 – 7.68 (m, 16H), 7.14 (dd, J = 6.7, 4.1 Hz, 1H), 5.52 (d, J = 15.4 Hz, 2H).

¹³C NMR (126 MHz, DMSO-d₆): δ 152.08, 146.64, 135.74, 134.93 (d, *J* = 3.1 Hz), 134.02 (d, *J* = 10.2 Hz), 133.68 (d, *J* = 8.9 Hz), 129.99 (d, *J* = 12.6 Hz), 118.51 (d, *J* = 86.6 Hz), 112.57 (d, *J* = 9.2 Hz), 109.96, 22.99 (d, *J* = 50.7 Hz), 22.79.

³¹P NMR (162 MHz, DMSO-d₆): δ 22.71.

HRMS (ESI): m/z calc'd for C₂₅H₂₁N₃P [M]⁺: 394.1468, found: 394.1478.

4.5 Imidazole benzylic Wittig olefination



To a suspension of Wittig salt **20** (5.06 g, 10.9 mmol, 1.0 equiv) and aldehyde 17^{12} (2.82 g, 10.9 mmol, 1.0 equiv) in THF (110 mL) was slowly added DBU (8.3 mL, 54.5 mmol, 5.0 equiv) at room temperature. The resulting pink suspension was vigorously stirred for 16 h, at which point TLC detected reaction completion. The reaction mixture was filtered through Celite (5 cm), washed with THF (3 × 30 mL). The combined organic solution was concentrated and purified by column chromatography (silica gel, DCM:EtOAc 1:1) to afford *E*-isomer **21** (2.17 g, 5.8 mmol, 53% yield) as white solid and *Z*-isomer **SI-2** (693 mg, 1.85 mmol, 17% yield) as white solid.

E-isomer 21:

R_f 0.15 (silica gel, DCM:EtOAc 1:2), UV active, stains brown in ninhydrin.

¹**H NMR (500 MHz, CDCl₃):** δ 8.50 (dd, J = 4.1, 2.1 Hz, 1H), 8.33 (dd, J = 6.7, 2.1 Hz, 1H), 7.42 (s, 1H), 7.04 – 6.67 (m, 2H), 6.57 (dt, J = 15.7, 1.5 Hz, 1H), 4.39 (dd, J = 6.0, 1.5 Hz, 2H), 1.50 (s, 18H).

¹³C NMR (126 MHz, CDCl₃): δ 152.40, 150.19, 148.69, 145.41, 132.91, 129.73, 123.16, 108.63, 108.06, 82.60, 48.07, 28.24.

HRMS (DART): calc'd for C₁₉H₂₇N₄O₄ [M+H]⁺: 375.2032, found: 375.2039.

Z-isomer SI-2:

R_f 0.25 (silica gel, DCM:EtOAc 1:2), UV active, stains brown in ninhydrin.

¹**H NMR (500 MHz, CDCl₃):** δ 8.51 (dd, J = 4.1, 2.0 Hz, 1H), 8.38 (dd, J = 6.7, 2.0 Hz, 1H), 7.51 (s, 1H), 6.83 (dd, J = 6.7, 4.1 Hz, 1H), 6.48 (dt, J = 11.6, 2.1 Hz, 1H), 5.80 (dt, J = 11.6, 5.7 Hz, 1H), 4.99 (dd, J = 5.7, 2.1 Hz, 2H), 1.44 (s, 18H).

¹³C NMR (126 MHz, CDCl₃): δ 152.61, 149.90, 148.22, 145.15, 133.19, 132.89, 120.73, 109.69, 108.75, 82.40, 46.30, 28.16.

HRMS (ESI): m/z calc'd for $C_{19}H_{27}N_4O_4$ [M+H]⁺: 375.2032, found: 375.2035.

Table S3. Optimization of Wittig olefination^a



	20 (equiv)	17 (equiv)	Conditions	yield	E:Z
1	1	1	KOtBu (2 equiv), THF	40% (not scalable)	Ε
2	2	1	KOtBu (2 equiv), THF	29%	Ε
3	1	2	KOtBu (2 equiv), THF	43%	Ε
4	1	1	KOtBu (2 equiv), premixed substrate, THF	No prd	-
5	1	1	NaOMe (2 equiv), MeOH	No prd	-
6	1	1	NaH (2 equiv), THF	18%	1:1
7	1	1	NaH (2 equiv), THF, 60 °C	11%	1:1
8	1	1	NaH (2 equiv), DMF	No prd	-
9	1	1	KOtBu (2 equiv), dioxane	No prd	-
10	1	1	DBU (5 equiv), THF	70% ^b	3:1
11	1	1	DBU (5 equiv), MeCN	77%	1.5:1
12	1	1	DBU (5 equiv), dioxane	70%	2.7:1
13	1	1	DBU (5 equiv), DCM	71%	1:1
14	1	1	DIPEA (5 equiv), THF	No prd	-
15	1	1	Et₃N (5 equiv), THF	No prd	-
16	1	1	DBN (5 equiv), THF	68%	2.8:1
17	1	1	DBU (5 equiv), MeOH	No prd	-
18	1	1	DBU (5 equiv), THF/DCM (1:1)	61%	1.6:1
19	2	1	DBU (5 equiv), THF	68%	3:1
20	1.5	1	DBU (5 equiv), THF	73%	3:1
21	1	2	DBU (5 equiv), THF	68%	2.5:1
22	1	1	DBU (2 equiv), THF	64%	3.2:1
23	1	1	DBU (3 equiv), THF	56%	2.7:1
24	1	1	DBU (4 equiv), THF	64%	3:1

^a0.1 mmol scale, 2 mL solvent, rt, 16h, ratio was based on crude ¹H NMR, 1,3,5-trimethoxylbenzene as internal standard. ^bisolated yield, gram scale.

4.6 NBS bromination



To a solution of olefin **21** (6.2 g, 16.6 mmol, 1.0 equiv) in MeCN (400 mL) was slowly added a solution of NBS (178 g, 18.0 mmol, 1.1 equiv) in MeCN (20 mL) at room temperature. The orange solution was stirred in dark for 1 h at room temperature. The solvent was removed in *vacuo* to give an off-white solid, which was then triturated by a solution of 10% Na₂S₂O₃ in H₂O (500 mL). The solid was filtered, washed with H₂O (3×200 mL), and dried under air to afford bromide **12** (6.9 g, 15.2 mmol, 92% yield) as white solid.

R_f 0.43 (silica gel, DCM:EtOAc 1:1), UV active, stains orange in ninhydrin.

¹**H NMR (500 MHz, CDCl₃):** δ 8.54 (dd, J = 4.2, 2.0 Hz, 1H), 8.34 (dd, J = 6.8, 2.0 Hz, 1H), 7.08 – 6.81 (m, 2H), 6.60 (dt, J = 15.6, 1.5 Hz, 1H), 4.42 (dd, J = 6.0, 1.5 Hz, 2H), 1.51 (s, 18H).

¹³C NMR (126 MHz, CDCl₃): δ 152.25, 150.52, 148.45, 143.01, 131.79, 131.30, 120.74, 109.21, 92.63, 82.68, 48.08, 28.21.

HRMS (DART): calc'd for C₁₉H₂₆BrN₄O₄ [M+H]⁺: 453.1137, found: 453.1136.

4.7 Stille coupling of diene 11b'/11d' and bromide 12



A solution of stannane **11b'/11d'** (2:1 mixture, 231 mg, 0.35 mmol, 1.1 equiv) and bromide **12** (143 mg, 0.32 mmol, 1.0 equiv) in DMF (3 mL) was treated with Pd(PPh₃)₄ (18.2 mg, 0.016 mmol, 5 mol %), CuI (6 mg, 0.032 mmol, 10 mol %), CsF (96 mg, 0.63 mmol, 2.0 equiv) at 100 °C for 16 h. The resulting dark brown solution was concentrated in *vacuo*, and purified by column chromatography (silica gel, CHCl₃:MeOH = 20:1 to 15:1) to furnish diene **22** (150 mg, 0.2 mmol, 63% yield) as pale yellow oil and **23** (62 mg, 83 µmol, 26% yield) as pale yellow oil.

22:

R_f 0.40 (silica gel, CHCl₃:MeOH 15:1), UV active, stains red in ninhydrin.

¹**H** NMR (600 MHz, CDCl₃): δ 8.50 (dd, J = 4.1, 2.0 Hz, 1H), 8.48 (dd, J = 4.0, 2.0 Hz, 1H), 7.95 (s, 1H), 7.70 (dd, J = 6.9, 2.0 Hz, 1H), 7.67 (dd, J = 6.9, 2.0 Hz, 1H), 7.10 (dt, J = 15.5, 6.3 Hz, 1H), 6.66 (dd, J = 6.9, 4.0 Hz, 1H), 6.63 (dd, J = 6.9, 4.1 Hz, 1H), 6.58 (dt, J = 15.4, 1.4 Hz, 1H), 6.50 (dd, J = 9.3, 3.6 Hz, 1H), 4.49 – 4.34 (m, 3H), 4.19 (dd, J = 15.7, 9.3 Hz, 1H), 1.48 (s, 18H), 1.43 (s, 18H).

¹³C NMR (151 MHz, CDCl₃): δ 152.43, 152.31, 151.37, 150.23, 149.73, 149.27, 144.60, 135.55, 135.06, 132.68, 132.08, 131.70, 121.61, 121.49, 117.31, 114.03, 109.74, 108.84, 83.10, 82.64, 48.16, 44.93, 28.26, 28.17.

HRMS (ESI): m/z calc'd for C₃₈H₅₁N₈O₈ [M+H]⁺: 747.3830, found: 747.3836.

23:

Rf 0.37 (silica gel, CHCl₃:MeOH 15:1), UV active, stains orange in ninhydrin.

¹**H NMR (600 MHz, CDCl₃):** δ 8.55 (dd, J = 4.1, 2.0 Hz, 1H), 8.50 (dd, J = 4.1, 2.0 Hz, 1H), 8.41 (dd, J = 7.0, 2.0 Hz, 1H), 7.96 (dd, J = 6.8, 2.0 Hz, 1H), 7.11 (s, 1H), 6.97 (dd, J = 6.9, 4.1 Hz, 1H), 6.90 (dt, J = 15.5, 5.9 Hz, 1H), 6.70 (s, 1H), 6.68 (dd, J = 6.8, 4.1 Hz, 1H), 6.51 (dt, J = 15.5, 1.5 Hz, 1H), 4.70 (dd, J = 16.1, 1.5 Hz, 1H), 4.65 (dd, J = 16.1, 1.3 Hz, 1H), 4.37 (ddd, J = 15.9, 5.7, 1.6 Hz, 1H), 4.30 (ddd, J = 15.9, 6.9, 1.4 Hz, 1H), 1.40 (s, 18H), 1.38 (s, 18H).

¹³C NMR (151 MHz, CDCl₃): δ 152.79, 152.21, 150.57, 150.14, 149.00, 148.97, 142.76, 135.33, 131.41, 130.69, 124.07, 121.77, 118.98, 118.53, 115.93, 109.44, 108.66, 83.33, 82.48, 51.84, 48.22, 28.12, 28.06.

HRMS (ESI): m/z calc'd for C₃₈H₅₁N₈O₈ [M+H]⁺: 747.3830, found: 747.3836.

4.8 Synthesis of nagelamide C



To a solution of diene **22** (6.0 mg, 8.0 μ mol, 1.0 equiv) in DCM (1 mL) was slowly added TFA (0.1 mL) at room temperature. The reaction was stirred for 1 h at the same temperature. The solvent was removed under reduced pressure to afford a dark red residue. The residue was dissolved in DMF (0.5 mL). Na₂CO₃ (8.5 mg, 0.08 mmol, 10.0 equiv) was added at room temperature. After vigorous stirring for 30 min, 4,5-dibromo-2-trichloroacetylpyrrole **24** (17.8 mg, 0.048 mmol, 6.0 equiv) was added as solid then stirred for 12 h. The solvent was removed under vacuum, hydrazine hydrate (0.2 mL) was added at room temperature. The vial was gently warmed to 50 °C and stirred for 15 min, at which point a dark red solution was formed. Excess amount of hydrazine was quickly removed under high vacuum. The residue was dissolved in MeOH (2 mL), then filtered through a short pad of Celite (1 cm) and washed with MeOH (2 mL). The combined MeOH solution was added TFA (50 µL), then concentrated and purified on C18 silica gel column to furnish nagelamide C (4.1 mg, 4.1 µmol, 51% yield) as off white solid.

HPLC: Agilent ZORBAX Eclipse Plus C18 column ($4.6 \times 250 \text{ mm}$, 5 µm). Samples were eluted with a linear gradient of 0% acetonitrile–water containing 0.1% TFA \rightarrow 100% acetonitrile containing 0.1% TFA over 11 min (flow rate 1.0 mL/min, UV detection at 254 nm), t_R = 9.29 min.

Purification: C18 silica gel (25 g, 40 μ m), 100% H₂O/0.1% TFA to 30% MeCN/70% H₂O/0.1% TFA.

¹**H NMR (600 MHz, DMSO-d**₆) δ 13.10 (s, 1H), 12.93 (s, 1H), 12.77 (s, 1H), 12.72 (d, J = 2.9 Hz, 1H), 12.69 (d, J = 2.8 Hz, 1H), 12.49 (s, 1H), 8.49 (t, J = 5.7 Hz, 1H), 8.46 (t, J = 5.8 Hz, 1H), 7.86 (s, 2H), 7.72 (s, 2H), 6.93 (d, J = 2.7 Hz, 1H), 6.92 (d, J = 2.7 Hz, 1H), 6.79 (s, 1H), 6.25 (t, J = 6.7 Hz, 1H), 6.17 (dt, J = 16.1, 5.9 Hz, 1H), 6.05 (d, J = 16.0 Hz, 1H), 3.95 (t, J = 5.7 Hz, 2H), 3.86 (t, J = 6.2 Hz, 2H).

¹³C NMR (151 MHz, DMSO-d₆) δ 158.81, 158.73, 148.23, 148.07, 129.47, 127.94, 127.82 (two carbons), 125.45, 123.37, 116.81, 116.60, 115.91, 112.86, 112.72, 112.52, 104.84, 104.73, 97.90 (two carbons), 40.24, 37.47.

¹⁹F NMR (565 MHz, DMSO-d₆) δ -74.17.

Nagelamide C free base:

¹**H NMR (600 MHz, CD₃OD)** δ 6.81 (s, 1H), 6.79 (s, 1H), 6.23 (s, 1H), 6.15 (d, *J* = 16.0 Hz, 1H), 6.02 (t, *J* = 6.9 Hz, 1H), 5.90 (dt, *J* = 15.9, 6.1 Hz, 1H), 3.96 (t, *J* = 6.9 Hz, 4H).

HRMS (ESI): m/z calc'd for C₂₂H₂₁Br₄N₁₀O₂ [M+H]⁺: 772.8582, found: 772.8575.



Figure S2. HRMS of nagelamide C



Figure S3. In situ TFA salt formation of nagelamide C free base

- a. ¹H NMR of crude nagelamide C free base in DMSO-d₆ (600 MHz)
- b. ¹H NMR of crude nagelamide C TFA salt after addition of 1 µL TFA
- c. ¹H NMR of purified nagelamide C TFA salt in DMSO-d₆ (600 MHz)



Figure S4. ¹H NMR spectra comparison of the natural and synthetic nagelamide C



Figure S5. ¹³C NMR spectra comparison of the natural and synthetic nagelamide C

4.9 Stille coupling of diene 11a' and bromide 12



A mixture of *cis*-adduct **11a'** (130 mg, 0.2 mmol, 1.1 equiv), bromide **12** (81 mg, 0.18 mmol, 1.0 equiv), Pd(PPh₃)₄ (10 mg, 9 μ mol, 5 mol %), CuI (3.4 mg, 18 μ mol, 10 mol %), CsF (55 mg, 0.36 mmol, 2.0 equiv) in DMF (5 mL) was stirred at 100 °C for 16 h. The reaction mixture was concentrated in *vacuo*, and purified by column chromatography (silica gel, CHCl₃:MeOH = 20:1 to 15:1) to give diene **25** (106 mg, 0.14 mmol, 79% yield) as pale yellow oil.

R_f 0.40 (silica gel, CHCl₃:MeOH 15:1), UV active, stains red in ninhydrin.

¹**H NMR (600 MHz, CDCl₃):** δ 8.51 (dd, J = 4.1, 2.0 Hz, 1H), 8.46 (dd, J = 4.1, 2.0 Hz, 1H), 8.14 (s, 1H), 7.80 (dd, J = 6.9, 2.0 Hz, 1H), 7.56 (dd, J = 6.9, 2.0 Hz, 1H), 7.03 (dt, J = 15.4, 6.7 Hz, 1H), 6.69 (dd, J = 15.4, 1.4 Hz, 1H), 6.65 (dd, J = 6.9, 4.0 Hz, 1H), 6.61 (dd, J = 6.9, 4.1 Hz, 1H), 6.23 (t, J = 6.0 Hz, 1H), 4.66 (d, J = 6.0 Hz, 2H), 4.36 (dd, J = 6.7, 1.4 Hz, 2H), 1.50 (s, 18H), 1.49 (s, 18H).

¹³C NMR (151 MHz, CDCl₃): δ 152.61, 152.38, 151.14, 150.75, 149.51, 148.55, 143.61, 138.97, 137.50, 133.09, 132.20, 131.38, 121.18, 117.67, 117.03, 115.84, 109.68, 109.09, 83.39, 82.55, 48.41, 45.48, 28.29, 28.25.

HRMS (ESI): m/z calc'd for C₃₈H₅₁N₈O₈ [M+H]⁺: 747.3830, found: 747.3838.

4.10 Synthesis of nagelamide C E-isomer



To a solution of diene **25** (5.7 mg, 7.7 μ mol, 1.0 equiv) in DCM (1 mL) was slowly added TFA (0.1 mL) at room temperature. The reaction was stirred for 1 h at the same temperature. The solvent was removed under vacuum to afford a dark residue. The residue was dissolved in DMF (0.5 mL). Na₂CO₃ (8.2 mg, 0.077 mmol, 10.0 equiv) was added at room temperature. After vigorous stirring for 30 min, 4,5-dibromo-2-trichloroacetylpyrrole **24** (17.0 mg, 0.046 mmol, 6.0 equiv) was stirred for 12 h, the solvent was removed under vacuum. Hydrazine hydrate (0.2 mL) was added at room temperature. The vial was gently warmed to 50 °C and stirred for 15 min. A dark red solution was formed. Excess amount of hydrazine was quickly removed under high vacuum. The residue was

dissolved in MeOH (2 mL), then filtered through a short pad of Celite (1 cm) and washed with MeOH (2 mL). The combined MeOH solution was added TFA (50 μ L), then concentrated and purified on C18 silica gel column to furnish nagelamide C *E*-isomer **26** (4.3 mg, 4.3 μ mol, 56% yield) as off white solid.

HPLC: Agilent ZORBAX Eclipse Plus C18 column ($4.6 \times 250 \text{ mm}, 5 \mu \text{m}$). Samples were eluted with a linear gradient of 0% acetonitrile–water containing 0.1% TFA \rightarrow 100% acetonitrile containing 0.1% TFA over 11 min (flow rate 1.0 mL/min, UV detection at 254 nm), t_R = 9.38 min.

Purification: C18 silica gel (25 g, 40 μ m), 100% H₂O/0.1% TFA to 30% MeCN/70% H₂O/0.1% TFA.

¹**H NMR (600 MHz, DMSO-d₆)** δ 12.79 (s, 1H), 12.74 (d, J = 2.8 Hz, 1H), 12.64 (d, J = 2.8 Hz, 1H), 12.57 (s, 1H), 12.47 (s, 1H), 12.43 (s, 1H), 8.54 (t, J = 5.6 Hz, 1H), 8.41 (t, J = 5.8 Hz, 1H), 7.66 (s, 2H), 7.60 (s, 2H), 7.20 (s, 1H), 6.94 (d, J = 2.8 Hz, 1H), 6.92 (d, J = 2.8 Hz, 1H), 6.17 – 6.05 (m, 2H), 5.90 (t, J = 6.8 Hz, 1H), 4.11 (t, J = 6.2 Hz, 2H), 3.89 (t, J = 5.2 Hz, 2H).

¹³C NMR (151 MHz, DMSO-d₆) δ 159.10, 158.69, 147.59, 147.56, 133.05, 128.01, 127.89, 127.73, 122.47, 121.74, 120.70, 116.70, 116.28, 114.65, 112.91, 112.64, 105.04, 104.78, 97.96, 97.91, 40.58, 37.85.

¹⁹F NMR (565 MHz, DMSO-d₆) δ -74.10.

Nagelamide C *E*-isomer free base:

¹**H NMR (600 MHz, CD₃OD)** δ 6.82 (s, 1H), 6.80 (s, 1H), 6.33 (s, 1H), 6.13 (d, *J* = 15.8 Hz, 1H), 5.83 (dt, *J* = 15.7, 6.4 Hz, 1H), 5.66 (t, *J* = 7.1 Hz, 1H), 4.20 (d, *J* = 7.2 Hz, 2H), 3.91 (d, *J* = 6.4 Hz, 2H).

HRMS (ESI): m/z calc'd for C₂₂H₂₁Br₄N₁₀O₂ [M+H]⁺: 772.8582, found: 772.8570.



Figure S6. HRMS of nagelamide C E-isomer



Figure S7. In situ TFA salt formation of nagelamide C *E*-isomer free base

a. ¹H NMR of crude nagelamide C *E*-isomer free base in DMSO-d₆ (600 MHz) b. ¹H NMR of crude nagelamide C *E*-isomer TFA salt after addition of 1 μ L TFA c. ¹H NMR of purified nagelamide C *E*-isomer TFA salt in DMSO-d₆ (600 MHz)











HSQC, 600 MHz, CDCl₃




















140 120 100 80 60 40 20 0 -20 -40 -60 -80 -100 -120 -140 -160 -180 -200 -220 -240 f1 (ppm) ¹³P NMR, 162 MHz, DMSO-d₆




































































































6. References

- Endo, T.; Tsuda, M.; Okada, T.; Mitsuhashi, S.; Shima, H.; Kikuchi, K.; Mikami, Y.; Fromont, F.; Kobayashi, J. Nagelamides A-H, New Dimeric Bromopyrrole Alkaloids from Marine Sponge Agelas Species. J. Nat. Prod. 2004, 67, 1262-1267.
- Iwai, T.; Kubota, T.; Fromont, J.; Kobayashi, J. Nagelamide I and 2,2'-Didebromonagelamide B, New Dimeric Bromopyrrole–Imidazole Alkaloids from a Marine Sponge *Agelas* sp. *Chem. Pharm. Bull.* 2014, **62**, 213–216.
- Araki, A.; Tsuda, M.; Kubota, T.; Mikami, Y.; Fromont, J.; Kobayashi, J. Nagelamide J, a Novel Dimeric Bromopyrrole Alkaloid from a Sponge *Agelas* Species. *Org. Lett.* 2007, 9, 2369–2371.
- Araki, A.; Kubota, T.; Tsuda, M.; Mikami, Y.; Fromont, J.; Kobayashi, J. Nagelamides K and L, Dimeric Bromopyrrole Alkaloids from Sponge *Agelas* Species. *Org. Lett.* 2008, **10**, 2099– 2102.
- Kubota, T.; Araki, A.; Ito, J.; Mikami, Y.; Fromont, J.; Kobayashi, J. Nagelamides M and N, New Bromopyrrole Alkaloids from Sponge *Agelas* Species. *Tetrahedron* 2008, 64, 10810– 10813.
- 6. Yasuda, T.; Araki, A.; Kubota, T.; Ito, J.; Mikami, Y.; Fromont, J.; Kobayashi, J. Bromopyrrole Alkaloids from Marine Sponges of the Genus *Agelas. J. Nat. Prod.* 2009, **72**, 488–491.
- Araki, A.; Kubota, T.; Aoyama, K.; Mikami, Y.; Fromont, J.; Kobayashi, J. Nagelamides Q and R, Novel Dimeric Bromopyrrole Alkaloids from Sponges *Agelas* sp. *Org. Lett.* 2009, **11**, 1785–1788.
- Appenzeller, J.; Tilvi, S.; Martin, M.-T.; Gallard, J.-F.; El-bitar, H.; Dau, E. T. H.; Debitus, C.; Laurent, D.; Moriou, C.; Al-Mourabit, A. Benzosceptrins A and B with a Unique Benzocyclobutane Skeleton and Nagelamide S and T from Pacific Sponges. *Org. Lett.* 2009, 11, 4874–4877.
- Tanaka, N.; Kusama, T.; Takahashi-Nakaguchi, A.; Gonoi, T.; Fromont, J.; Kobayashi, J. Nagelamides U–W, Bromopyrrole Alkaloids from a Marine Sponge *Agelas* sp. *Tetrahedron Lett.* 2013, 54, 3794–3796.
- Tanaka, N.; Kusama, T.; Takahashi-Nakaguchi, A.; Gonoi, T.; Fromont, J.; Kobayashi, J. Nagelamides X–Z, Dimeric Bromopyrrole Alkaloids from a Marine Sponge *Agelas* sp. *Org. Lett.* 2013, 15, 3262–3265.
- Juillet, C.; Ermolenko, L.; Boyarskaya, D.; Baratte, B.; Josselin, B.; Nedev, H.; Bach, S.; Iorga, B.; Bignon, J.; Ruchaud, S.; Al-Mourabit, A. From Synthetic Simplified Marine Metabolite Analogues to New Selective Allosteric Inhibitor of Aurora B Kinase. *J. Med. Chem.* 2021, 64, 1197–1219.
- 12. Commandeur, M.; Commandeur, C.; Cossy, J. Synthesis of a Platform to Access Bistramides and their Analogues. *Org. Lett.* 2011, **13**, 6018–6021.