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Supporting Information

Electrochemically driven reductive coupling of nitroarenes with alkyl bromides

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1. General Remarks

Unless otherwise noted, catalytic reactions were carried out in undivided electrochemical cells (10 mL) using pre-dried glassware. All reagents and solvents were purchased from commercial suppliers (Energy Chemical, Adamas-beta, Bidepharm, General Reagent) and used without further purification. Nickel foam (10 mm × 20 mm × 1.5 mm, 99.9%; obtained from Kunshan City Yushan town Wulife electronic material management department, China) and graphite felt electrodes (10 mm × 20 mm × 3.0 mm, Anhui Senrise Technologies, China) were connected using stainless steel adapters. The electrocatalytic reactions were conducted at constant current modes using a HSPY-600 as power supply. Thin-layer chromatography (TLC) was performed on $200 \pm 50 \,\mu m$ glass-backed silica gel plates purchased from Shanghai Titan Technology Co., Ltd., China. Column chromatography was carried out on silica gel (200–300 mesh) purchased from Yantai Xinnuo Chemical Co., Ltd., China. NMR spectra were recorded on Bruker Avance NEO 600 or Avance 400 instrument. Chemical shifts (δ) are given in ppm relative to the solvent residual peak, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), and coupling constants (J) are in Hertz (Hz). GC-MS analysis was performed on a Shimadzu GC-2030AM or a Hexin EI-Q 1000. High revolution mass spectra (HR-MS) were obtained on Waters Xevo G3 QTof UPLC/MS spectrometer. All IR spectra were recorded on Thermo Nicolet is 20 device in the range from 4000 to 400 cm⁻¹. Melting point ranges were taken with Melting Point Apparatus X-5 (Beijing Unitedvision Technology Co., Ltd., China).

2. General Procedure for Electrochemical Reduction.

Scheme S1. General procedure for electrochemical reduction.

General Procedure A: The electrocatalysis was carried out in an undivided cell under N₂ with a graphite felt (GF) anode (10 mm × 20 mm × 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm × 20 mm × 1.5 mm). The cell was charged with nitrobenzene (0.40 mmol), benzyl bromide (1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μL, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL). Electrocatalysis was performed at 50 °C with a constant current of 4.0 mA maintained for 12 h. The electrodes were washed with dichloromethane (3 × 4.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. The residue was purified by chromatography on silica gel to afford the the desired products.



Figure S1. Pictures of the setup of electrolysis.

General Procedure B (for Scale-up reaction)

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Scheme S2. General procedure for scale-up reaction.

The electrocatalysis was carried out in an undivided cell under N_2 with a GF anode (20 mm \times 50 mm \times 3.0 mm) and a Ni foam cathode (20 mm \times 50 mm \times 1.5 mm). The cell was charged with nitrobenzene **1a** (492 mg, 4.0 mmol), benzyl bromide **2a** (2.05 g, 12 mmol), TBABF₄ (1.98 g, 6.0 mmol), 2,6-lutidine (1.40 mL, 12 mmol), B_2Pin_2 (2.03 g, 8.0 mmol) and MeOH solvent (40.0 mL). Electrocatalysis was performed at 50 °C with a constant current of 10 mA maintained for 36 h. The electrodes were washed with dichloromethane (3 \times 8.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. The residue was purified by chromatography on silica gel (PE/DCM = 50:1) to afford product **3aa** (433 mg, 40%) as white solid.



Figure S2. Pictures of the scale-up reaction setups.

3. Optimization of the Reaction Conditions

Table S1. The effect of solvent.[a]

Scheme S3. The effect of solvent.

Entry	Solvent	Yield (%)
1	МеОН	92%, 85 ^[b]
2	HFIP	trace
3	DMA	27
4	CH ₃ CN	47
5	THF	trace
6	H_2O	NR
7	DMF	45

^[a] Reaction conditions: undivided cell, GF as anode, Ni Foam as cathode, constant current electrolysis (CCE) = 4.0 mA, **1a** (41 μL, 0.40 mmol), **2a** (143 μL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), B₂Pin₂ (203 mg, 0.80 mmol), 2,6-lutidine (140 μL, 1.20 mmol), solvent (4.0 mL), 50 °C, 12 h, under N₂ atmosphere. GC yields with 1,3,5-trimethoxybenzene as the internal standard. ^[b] Isolated yield of **3aa**.

Table S2. The effect of base.^[a]

Scheme S4. The effect of base.

Entry	Base	Yield (%)
1	2,6-lutidine	92%, 85 ^[b]
2	$\mathrm{Et}_{3}\mathrm{N}$	44
3	pyridine	trace
4	DMAP	trace
5	DBU	50
6	DBN	27
7	DIPEA	57
8	2,4,6-collidine	64
9	2,6-di-tert-butylpyridine	NR
10	K_2CO_3	NR

^[a] Reaction conditions: undivided cell, GF as anode, Ni Foam as cathode, constant current electrolysis (CCE) = 4.0 mA, **1a** (41 μL, 0.40 mmol), **2a** (143 μL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), B₂Pin₂ (203 mg, 0.80 mmol), base (1.20 mmol), MeOH (4.0 mL), 50 °C, 12 h, under N₂ atmosphere. GC yields with 1,3,5-trimethoxybenzene as the internal standard. ^[b] Isolated yield of **3aa**.

Table S3. The effect of boron ester.^[a]

Scheme S5. The effect of boron ester.

Entry	Boron Ester	Yield (%)
1	B ₂ Pin ₂	92%, 85 ^[b]
2	B_2Cat_2	64
3	Cy-Bpin	11

^[a] Reaction conditions: undivided cell, GF as anode, Ni Foam as cathode, constant current electrolysis (CCE) = 4.0 mA, **1a** (41 μL, 0.40 mmol), **2a** (143 μL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), boron ester (0.80 mmol), 2,6-lutidine (140 μL, 1.20 mmol), MeOH (4.0 mL), 50 °C, 12 h, under N_2 atmosphere. GC yields with 1,3,5-trimethoxybenzene as the internal standard. ^[b] Isolated yield of **3aa**.

Table S4. The effect of current density.

Scheme S6. The effect of current density.

Entry	CCE	Yield (%)
1	4.0 mA	85
2	2.0 mA	58
3	6.0 mA	74
4	8.0 mA	62

^[a]Reaction conditions: undivided cell, GF as anode, Ni Foam as cathode, **1a** (41 μL, 0.40 mmol), **2a** (143 μL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), B₂pin₂ (203 mg, 0.80 mmol), 2,6-lutidine (140 μL, 1.20 mmol), MeOH (4.0 mL). After that, the reaction was electrolyzed at a constant current of 2.0 mA (0.5 mA/cm²), 4.0 mA (1.0 mA/cm²), 6.0 mA (1.5 mA/cm²), 8.0 mA (4.0 mA/cm²), 50 °C, 12 h, under N₂ atmosphere. Isolated yield of **3aa**.

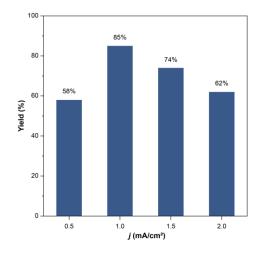


Figure S3. The effect of current density.

Table S5. The effect of applied charge.

Scheme S7. The effect of applied charge.

Entry	Time	Yield (%)
1	12 h	85
2	4 h	37
3	6 h	44
4	8 h	71
5	16 h	66
6	18 h	65

^[a] Reaction conditions: undivided cell, GF as anode, Ni Foam as cathode, **1a** (41 μL, 0.40 mmol), **2a** (143 μL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), B₂pin₂ (203 mg, 0.80 mmol), 2,6-lutidine (140 μL, 1.20 mmol), MeOH (4.0 mL). After that, the reaction mixture was electrolyzed under a constant current electrolysis at 4.0 mA for different time: 4 h (1.49 F/mol), 6 h (2.24 F/mol), 8 h (2.98 F/mol), 12 h (4.48 F/mol), 16 h (5.97 F/mol), 18 h (6.72 F/mol), 50 °C, under N₂ atmosphere. Isolated yield of **3aa**.

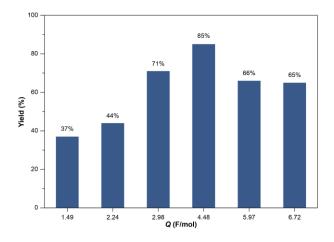


Figure S4. The effect of applied charge.

Table S6. The effect of pH.

Scheme S8. The effect of pH.

Entry	рН	Yield (%)
1	8.35	85
2	7.35	66
3	9.35	59

[a] Reaction conditions: undivided cell, GF as anode, Ni Foam as cathode, **1a** (41 μL, 0.40 mmol), **2a** (143 μL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), B₂pin₂ (203 mg, 0.80 mmol), 2,6-lutidine (140 μL, 1.20 mmol), MeOH (4.0 mL). We first measured the pH (8.35) of the reaction mixture. Subsequently, we adjusted the pH of the reaction mixture using hydrochloric acid methanol solution and sodium hydroxide methanol solution, respectively: 1.25 M methanolic hydrochloric acid methanol solution was added to adjust the pH to 7.35, while 1.0 M methanolic sodium hydroxide methanol solution was added to adjust the pH to 9.35, 50 °C, under N₂ atmosphere. Isolated yield of **3aa**.

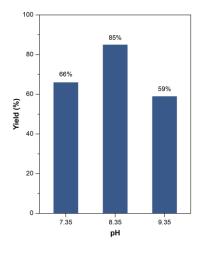
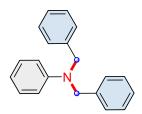


Figure S5. The effect of pH.

4. Characterization Data of Products



N,*N*-dibenzylaniline (3aa): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3aa (92.8 mg, 85%) as white solid. M.p. = 67–68 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.34 (t, J = 7.5 Hz, 4H), 7.28 (d, J = 7.4 Hz, 6H), 7.19 (t, J = 7.7 Hz, 2H), 6.77 (d, J = 8.1 Hz, 2H), 6.73 (t, J = 7.2 Hz, 1H), 4.68 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.3 (C_q), 138.7 (C_q), 129.3 (CH), 128.8 (CH), 127.0 (CH), 126.8 (CH), 116.9 (CH), 112.6 (CH), 54.3 (CH₂). IR (KBr): 3020, 1597, 1505, 1450, 1398, 1229, 1026, 957, 750, 731, 693 cm⁻¹. MS (ESI) m/z (relative intensity): 274 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₀H₁₉N [M+H]⁺ 274.1590, found 274.1593. The data are in agreement with those reported in literature. ¹

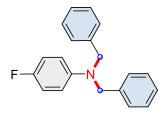
N,*N*-dibenzyl-4-methylaniline (3ba): The general procedure A was followed using 1b (54.8 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ba (72.6 mg, 63%) as white solid. M.p. = 68–69 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.35 (t, J = 7.5 Hz, 4H), 7.30–7.27 (m, 6H), 7.01 (d, J = 8.3 Hz, 2H), 6.69 (d, J = 8.3 Hz, 2H), 4.65 (s, 4H), 2.26 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ = 147.2 (C_q), 139.0 (C_q), 129.9 (C_q), 128.7 (CH), 126.9 (CH), 126.8 (CH), 126.0 (CH), 112.8 (CH), 54.5 (CH₂), 20.4 (CH₃). IR (KBr): 3022, 2920, 1618, 1521, 1362, 1238, 1074, 951, 800, 738, 695 cm⁻¹. MS (ESI) *m/z* (relative intensity): 288 (100) [M+H]⁺. HR-MS (ESI) *m/z* calcd for C₂₁H₂₁N [M+H]⁺ 288.1747, found 288.1747. The data are in agreement with those reported in literature.²

N,*N*-dibenzyl-4-methoxyaniline (3ca): The general procedure A was followed using 1c (61.2 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ca (91.2 mg, 75%) as white solid. **M.p.** = 64–65 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.35 (t, J = 7.5 Hz, 4H), 7.30–7.27 (m, 6H), 6.80 (d, J = 8.7 Hz, 2H), 6.73 (d, J = 8.7 Hz, 2H), 4.60 (s, 4H), 3.76 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ = 151.8 (C_q), 144.0 (C_q), 139.1 (C_q), 128.7 (CH), 127.1 (CH), 127.0 (CH), 114.9 (CH), 114.7 (CH), 55.8 (CH₃), 55.3 (CH₂). **IR** (KBr): 3027, 2924, 1602, 1512, 1355, 1232, 1081, 965, 805, 735, 694 cm⁻¹. **MS** (ESI) *m/z* (relative intensity): 304 (100) [M+H]⁺. **HR-MS** (ESI) *m/z* calcd for C₂₁H₂₁NO [M+H]⁺ 304.1696, found 304.1695. The data are in agreement with those reported in literature.³

N,N-dibenzyl-4-iodoaniline (3da): The general procedure A was followed using 1d (99.6 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3da (68.8 mg, 68%) as white solid. **M.p.** = 65–66 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.40 (d, J = 8.5 Hz, 2H), 7.34 (t, J = 7.5 Hz, 4H), 7.27 (t, J = 6.5 Hz, 2H), 7.22 (t, J = 7.5 Hz, 4H), 6.51 (d, J = 8.5 Hz, 2H), 4.64 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.8 (C_q), 138.1 (C_q), 137.9 (CH), 128.9 (CH), 127.2 (CH), 126.6 (CH), 114.9 (CH), 77.7 (C_q), 54.4 (CH₂). **IR** (KBr): 3022, 2924, 1589, 1493, 1350, 1242, 1070, 961, 799, 728, 692 cm⁻¹. **MS** (ESI) m/z (relative intensity): 400 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₀H₁₈IN [M+H]⁺ 400.0557, found 400.0562. The data are in agreement with those reported in literature.⁴

N,*N*-dibenzyl-4-bromoaniline (3ea): The general procedure A was followed using 1e (80.8 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ea (99.9 mg, 71%) as white solid. M.p. = 190–191 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.33 (t, J = 7.5 Hz, 4H), 7.26 (t, J = 7.1 Hz, 2H), 7.25–7.21 (m, 6H), 6.59 (d, J = 9.0 Hz, 2H), 4.63 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.2 (C_q), 138.1 (C_q), 132.0 (CH), 128.9 (CH), 127.2 (CH), 126.7 (CH), 114.3 (CH), 108.7 (C_q), 54.6 (CH₂). IR (KBr): 2924, 1591, 1497, 1380, 1241, 1082, 965, 801, 729, 691 cm⁻¹. MS (ESI) m/z (relative intensity): 352 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₀H₁₈⁷⁹BrN [M+H]⁺ 352.0695, found 352.0695. The data are in agreement with those reported in literature.⁵

N,N-dibenzyl-4-chloroaniline (3fa): The general procedure A was followed using 1f (63.2 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3fa (85.3 mg, 69%) as white solid. **M.p.** = 92–93 °C. ¹H NMR (600 MHz, CDCl₃) δ =7.34 (t, J = 7.7 Hz, 4H), 7.27 (t, J = 7.3 Hz, 2H), 7.23 (d, J = 7.1 Hz, 4H), 7.10 (d, J = 9.0 Hz, 2H), 6.64 (d, J = 9.0 Hz, 2H), 4.64 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 147.8 (C_q), 138.2 (C_q), 129.1 (CH), 128.9 (CH), 127.2 (CH), 126.7 (CH), 121.7 (C_q), 113.8 (CH), 54.6 (CH₂). **IR** (KBr): 2924, 1591, 1497, 1351, 1241, 1190, 1082, 965, 801, 729, 691 cm⁻¹. **MS** (ESI) m/z (relative intensity): 308 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₀H₁₈³⁵CIN [M+H]⁺ 308.1201, found 308.1202. The data are in agreement with those reported in literature.⁵



N,N-dibenzyl-4-fluoroaniline (3ga): The general procedure A was followed using 1g (56.4 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ga (71.9 mg, 62%) as brown oil. ¹H NMR (600 MHz, CDCl₃) δ = 7.32 (t, J = 7.5 Hz, 4H), 7.25–7.23 (m, 6H), 6.85 (t, J = 8.7 Hz, 2H), 6.64 (dd, J = 9.0, 4.3 Hz, 2H), 4.60 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 155.6 (d, ¹J_{C-F} = 235.6 Hz, C_q), 145.8 (d, ⁴J_{C-F} = 1.9 Hz, C_q), 138.6 (C_q), 128.8 (CH), 127.1 (CH), 126.8 (CH), 115.7 (d, ²J_{C-F} = 21.1 Hz, CH), 113.9 (d, ³J_{C-F} = 6.0 Hz, CH), 55.2 (CH₂). ¹⁹F NMR (565 MHz, CDCl₃) δ = -128.87. IR (KBr): 3027, 2924, 1602, 1512, 1355, 1232, 1081, 965, 805, 735, 694 cm⁻¹. MS (ESI) m/z (relative intensity): 292 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₀H₁₈FN [M+H]⁺ 292.1496, found 292.1494. The data are in agreement with those reported in literature.⁶

N,*N*-dibenzyl-4-(trifluoromethyl)aniline (3ha): The general procedure A was followed using **1h** (76.4 mg, 0.40 mmol) and **2a** (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded **3ha** (69.8 mg, 51%) as white solid. **M.p.** = 80–81 °C. ¹H **NMR** (600 MHz, CDCl₃) δ = 7.39 (d, J = 8.7 Hz, 2H), 7.35 (t, J = 7.5 Hz, 4H), 7.28 (t, J = 7.4 Hz, 2H), 7.22 (d, J = 8.7 Hz, 4H), 6.74 (d, J = 8.7 Hz, 2H), 4.71 (s, 4H). ¹³C **NMR** (151 MHz, CDCl₃) δ = 151.4 (C_q), 137.6 (C_q), 129.0 (CH), 127.4 (CH), 126.7 (q, ³J_{C-F} = 3.0 Hz, CH), 126.6 (CH), 125.2 (q, 1J _{C-F} = 270.3 Hz, C_q), 118.4 (q, 2J _{C-F} = 33.2 Hz, C_q), 111.7 (CH), 54.3 (CH₂). ¹°F **NMR** (565 MHz, CDCl₃) δ = -60.99. **IR** (KBr): 3026, 2923, 1617, 1493, 1390, 1296, 1069, 963, 813, 741, 694 cm⁻¹. **MS** (ESI) m/z (relative intensity): 342 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₁H₁₈F₃N [M+H]⁺ 342.1464, found 342.1468. The data are in agreement with those

reported in literature.⁷

N,N-dibenzyl-4-(trifluoromethoxy)aniline (3ia): The general procedure A was followed using 1i (82.8 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ia (88.7 mg, 62%) as white solid. **M.p.** = 71–72 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.36 (t, J = 7.4 Hz, 4H), 7.30–7.26 (m, 3H), 7.26–7.23 (m, 3H), 7.02 (d, J = 8.5 Hz, 2H), 6.68 (d, J = 8.5 Hz, 2H), 4.67 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 147.9 (C_q), 140.3 (C_q), 138.1 (C_q), 128.9 (CH), 127.3 (CH), 126.7 (CH), 122.5 (q, ${}^{1}J_{C-F}$ = 255.2 Hz, C_q), 118.3 (CH), 113.0 (CH), 54.8 (CH₂). ¹⁹F NMR (565 MHz, CDCl₃) δ = -58.37. IR (KBr): 3026, 2922, 1605, 1517, 1362, 1234, 1150, 953, 808, 732, 695 cm⁻¹. MS (ESI) m/z (relative intensity): 358 (100) [M+H]⁺.HR-MS (ESI) m/z calcd for C₂₁H₁₈F₃NO [M+H]⁺ 358.1413, found 358.1414. The data are in agreement with those reported in literature.⁸

4-(Dibenzylamino)phenyl trifluoromethanesulfonate (3ja): The general procedure A was followed using **1j** (108 mg, 0.40 mmol) and **2a** (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded **3ja** (108 mg, 64%) as pale yellow oil. ¹**H NMR** (600 MHz, CDCl₃) δ = 7.34 (t, J = 7.5 Hz, 4H), 7.27 (t, J = 7.3 Hz, 2H), 7.22 (d, J = 7.4 Hz, 4H), 7.02 (d, J = 9.2 Hz, 2H), 6.66 (d, J = 9.2 Hz, 2H), 4.66 (s, 4H). ¹³**C NMR** (151 MHz, CDCl₃) δ = 148.9 (C_q), 140.6 (C_q), 137.7 (C_q), 129.0 (CH), 127.3 (CH), 126.5 (CH), 122.1 (CH), 118.9 (q, ${}^{1}J_{C-F}$ = 321.0 Hz, C_q), 112.9 (CH), 54.7 (CH₂). ¹⁹**F NMR** (565 MHz, CDCl₃) δ = -72.85. **IR** (KBr): 3055, 1509, 1418, 1249, 1211, 1141, 883, 818, 736, 697, 608 cm⁻¹. **MS** (ESI)

m/z (relative intensity): 422 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₁H₁₈F₃NO₃S [M+H]⁺ 422.1032, found 422.1030.

2-(4-(Dibenzylamino)phenyl)ethan-1-ol (3ka): The general procedure A was followed using **1k** (66.8 mg, 0.40 mmol) and **2a** (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded **3ka** (63.5 mg, 50%) as pale yellow oil. ¹**H NMR** (600 MHz, CDCl₃) δ = 7.33 (t, J = 7.5 Hz, 4H), 7.27–7.24 (m, 6H), 7.02 (d, J = 8.2 Hz, 2H), 6.70 (d, J = 8.2 Hz, 2H), 4.64 (s, 4H), 3.80 (t, J = 6.5 Hz, 2H), 2.75 (t, J = 6.5 Hz, 2H). ¹³**C NMR** (151 MHz, CDCl₃) δ = 147.9 (C_q), 138.7 (C_q), 129.9 (CH), 128.7 (CH), 127.0 (CH), 126.7 (CH), 126.2 (C_q), 112.7 (CH), 64.0 (CH₂), 54.4 (CH₂), 38.2 (CH₂). **IR** (KBr): 3027, 2926, 1615, 1520, 1452, 1360, 1231, 1045, 809, 731, 697 cm⁻¹. **MS** (ESI) m/z (relative intensity): 318 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₂H₂₃NO [M+H]⁺ 318.1852, found 318.1850. The data are in agreement with those reported in literature.⁹

2-(4-(Dibenzylamino)phenyl)acetonitrile (3la): The general procedure A was followed using **1l** (64.8 mg, 0.40 mmol) and **2a** (143 µL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded **3la** (56.6 mg, 45%) as pale yellow solid. **M.p.** = 75–76 °C. ¹H **NMR** (600 MHz, CDCl₃) δ = 7.35 (t, J = 7.4 Hz, 4H), 7.30–7.25 (m, 6H), 7.12 (d, J = 8.2 Hz, 2H), 6.74 (d, J = 8.2 Hz, 2H), 4.68 (s, 4H), 3.63 (s, 2H). ¹³C **NMR** (151 MHz, CDCl₃) δ = 148.7 (C_q), 138.1 (C_q), 129.0 (CH), 128.9 (CH), 127.2 (CH), 126.7 (CH), 118.7 (C_q), 117.7 (C_q), 113.1 (CH), 54.6 (CH₂), 22.8 (CH₂). **IR** (KBr): 3026, 2921, 2246, 1615, 1523, 1449, 1237, 958, 802,

730, 695 cm⁻¹. **MS** (ESI) m/z (relative intensity): 318 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for $C_{22}H_{23}NO$ [M+H]⁺ 318.1852, found 318.1850. The data are in agreement with those reported in literature.¹

Methyl 2-(4-(dibenzylamino)phenyl)acetate (3ma): The general procedure A was followed using 1m (78.0 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/EA = 20:1) yielded 3ma (58 mg, 42%) as brown solid. M.p. = 82–83 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.32 (t, J = 7.2 Hz, 4H), 7.26–7.24 (m, 6H), 7.07 (d, J = 8.2 Hz, 2H), 6.70 (d, J = 8.2 Hz, 2H), 4.64 (s, 4H), 3.67 (s, 3H), 3.50 (s, 2H). ¹³C NMR (151 MHz, CDCl₃) δ = 172.8 (C_q), 148.3 (C_q), 138.6 (C_q), 130.1 (CH), 128.8 (CH), 127.0 (CH), 126.8 (CH), 122.0 (C_q), 112.7 (CH), 54.4 (CH₂), 52.0 (CH₃), 40.3 (CH₂). IR (KBr): 3014, 2924, 1732, 1615, 1523, 1400, 1230, 958, 813, 737, 696 cm⁻¹. MS (ESI) m/z (relative intensity): 346 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₃H₂₃NO₂ [M+H]⁺ 346.1802, found 346.1803. The data are in agreement with those reported in literature. ¹

Ethyl 2-(4-(dibenzylamino)phenyl)acetate (3na): The general procedure A was followed using **1n** (83.6 mg, 0.40 mmol) and **2a** (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded **3na** (76.5 mg, 53%) as pale yellow oil. ¹**H NMR** (600 MHz, CDCl₃) δ = 7.32 (t, J = 6.7 Hz, 4H), 7.26–7.24 (m, 6H), 7.08 (d, J = 8.4 Hz, 2H), 6.69 (d, J = 8.4 Hz, 2H), 4.64 (s, 4H), 4.13 (q, J = 7.1 Hz, 2H), 3.48 (s, 2H), 1.25 (t, J = 7.1 Hz, 3H). ¹³**C NMR** (151 MHz, CDCl₃) δ = 172.4 (C_q), 148.3 (C_q), 138.7 (C_q), 130.1 (CH), 128.8

(CH), 127.0 (CH), 126.8 (CH), 122.2 (C_q), 112.7 (CH), 60.8 (CH₂), 54.4 (CH₂), 40.5 (CH₂), 14.3 (CH₃). **IR** (KBr): 2982, 1734, 1615, 1521, 1393, 1150, 1030, 957, 810, 733, 696 cm⁻¹. **MS** (ESI) m/z (relative intensity): 360 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₄H₂₅NO₂ [M+H]⁺ 360.1958, found 360.1958. The data are in agreement with those reported in literature.¹⁰

N,*N*-dibenzyl-4-(1*H*-pyrrol-1-yl)aniline (3oa): The general procedure A was followed using 1o (75.2 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 20:1) yielded 3oa (56.8 mg, 42%) as brown solid. **M.p.** = 66–67 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.32 (t, J = 7.8 Hz, 4H), 7.26–7.24 (m, 6H), 7.14 (d, J = 9.3 Hz, 2H), 6.92 (t, J = 2.2 Hz, 2H), 6.74 (d, J = 9.3 Hz, 2H), 6.26 (t, J = 2.2 Hz, 2H), 4.65 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 147.5 (C_q), 138.4 (C_q), 131.5 (C_q), 128.8 (CH), 127.2 (CH), 126.7 (CH), 122.4 (CH), 119.8 (CH), 113.2 (CH), 109.4 (CH), 54.7 (CH₂). IR (KBr): 3026, 2918, 1617, 1522, 1390, 1243, 1075, 955, 812, 728, 694 cm⁻¹. MS (ESI) m/z (relative intensity): 339 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₄H₂₂N₂ [M+H]⁺ 339.1856, found 339.1856.

N,N-dibenzyl-3-methylaniline (3pa): The general procedure A was followed using 1p (54.8 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3pa (70.7 mg, 62%) as white solid. **M.p.** = 71–72 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.33 (t, J = 7.5 Hz, 4H), 7.26–7.25 (m, 6H), 7.07 (t, J = 7.8 Hz, 1H), 6.60 (s, 1H), 6.56 (t, J = 8.6 Hz, 2H), 4.64 (s, 4H), 2.25 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.5

(C_q), 139.1 (C_q), 138.8 (C_q), 129.2 (CH), 128.7 (CH), 127.0 (CH), 126.8 (CH), 117.9 (CH), 113.1 (CH), 109.8 (CH), 54.1 (CH₂), 22.1 (CH₃). **IR** (KBr): 3020, 2857, 1603, 1493, 1381, 1253, 1072, 959, 838, 770, 692 cm⁻¹. **MS** (ESI) m/z (relative intensity): 288 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₁H₂₁N [M+H]⁺ 288.1747, found 288.1749. The data are in agreement with those reported in literature.¹¹

N,N-dibenzyl-2-methylaniline (3qa): The general procedure A was followed using 1q (121 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3qa (48.3 mg, 42%) as colorless oil. ¹H NMR (600 MHz, CDCl₃) δ = 7.27–7.25 (m, 8H), 7.22–7.16 (m, 3H), 7.05 (q, J = 6.4 Hz, 1H), 6.94 (dd, J = 9.7, 4.1 Hz, 2H), 4.06 (dd, J = 5.8, 2.6 Hz, 4H), 2.45–2.44 (m, 3H). ¹³C NMR (151 MHz, CDCl₃) δ = 150.0 (C_q), 138.7 (C_q), 133.9 (C_q), 131.2 (CH), 128.8 (CH), 128.3 (CH), 127.0 (CH), 126.2 (CH), 123.6 (CH), 122.6 (CH), 57.0 (CH₂), 18.7 (CH₃). IR (KBr): 3061, 2923, 1597, 1493, 1361, 1212, 1005, 1029, 766, 722, 697 cm⁻¹. MS (ESI) m/z (relative intensity): 288 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₁H₂₁N [M+H]⁺ 288.1747, found 288.1748. The data are in agreement with those reported in literature.¹¹

N,N-dibenzylbenzo[*d*][1,3]dioxol-5-amine (3ra): The general procedure A was followed using 1r (66.8 mg, 0.40 mmol) and 2a (143 μ L, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ra (72.3 mg, 57%) as colorless oil. ¹H NMR (600 MHz, CDCl₃) δ = 7.35 (t, J = 8.5 Hz, 4H), 7.28 (d, J = 7.2 Hz, 6H), 6.66 (d, J = 8.6 Hz, 1H),

6.43 (d, J = 2.6 Hz, 1H), 6.18 (dd, J = 8.6, 2.6 Hz, 1H), 5.82 (s, 2H), 4.58 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) $\delta = 148.5$ (C_q), 145.4 (C_q), 139.3 (C_q), 138.8 (C_q), 128.7 (CH), 127.0 (CH), 126.9 (CH), 108.5 (CH), 105.2 (CH), 100.7 (CH₂), 96.3 (CH), 55.4 (CH₂). **IR** (KBr): 3060, 2924, 1608, 1503, 1343, 1275, 1037, 919, 870, 724, 686 cm⁻¹. **MS** (ESI) m/z (relative intensity): 318 (100) [M+H]⁺.**HR-MS** (ESI) m/z calcd for C₂₁H₁₉NO₂ [M+H]⁺ 318.1489, found 318.1488.

N,*N*-dibenzylnaphthalen-1-amine (3sa): The general procedure A was followed using 1s (69.2 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/EA = 50:1) yielded 3sa (46.6 mg, 36%) as brown oil. ¹H NMR (600 MHz, CDCl₃) δ = 8.52 (d, J = 8.4 Hz, 1H), 7.82 (d, J = 8.0 Hz, 1H), 7.52 (t, J = 8.2 Hz, 2H), 7.47 (t, J = 7.4 Hz, 1H), 7.28–7.26 (m, 4H), 7.26–7.17 (m, 7H), 6.91 (d, J = 7.4 Hz, 1H), 4.29 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 147.6 (C_q), 138.3 (C_q), 135.1 (C_q), 129.8 (C_q), 128.7 (CH), 128.6 (CH), 128.3 (CH), 127.1 (CH), 125.9 (CH), 125.7 (CH), 125.6 (CH), 123.9 (CH), 123.7 (CH), 118.6 (CH), 57.3 (CH₂). IR (KBr): 3059, 2923, 1575, 1493, 1452, 1400, 1223, 1028, 774, 751, 698 cm⁻¹. MS (ESI) m/z (relative intensity): 324 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₄H₂₁N [M+H]⁺ 324.1747, found 324.1747. The data are in agreement with those reported in literature. ¹²

N,*N*-Dibenzyl-1,2-dihydroacenaphthylen-5-amine (3ta): The general procedure A was followed using 1t (80.5 mg, 0.40 mmol) and 2a (143 μ L, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ta (55.0 mg, 39%) as colorless oli. ¹H NMR (600

MHz, CDCl₃) δ = 8.10 (d, J = 8.8 Hz, 1H), 7.52 (d, J = 6.7 Hz, 1H), 7.39 (d, J = 8.1 Hz, 4H), 7.36–7.30 (m, 5H), 7.29–7.24 (m, 2H), 7.12 (d, J = 8.7 Hz, 1H), 6.92 (d, J = 5.6 Hz, 1H), 3.47–3.42 (m, 2H), 3.37–3.32 (m, 2H). ¹³C **NMR** (151 MHz, CDCl₃) δ = 146.6 (C_q), 143.9 (C_q), 140.8(C_q), 140.7 (C_q), 138.7 (C_q), 128.5 (CH), 128.3 (CH), 128.0 (C_q), 127.3 (CH), 127.0 (CH), 119.4 (CH), 119.3 (CH), 119.2 (CH), 118.9 (CH), 57.1 (CH₂), 31.0 (CH₂), 29.78 (CH₂). **IR** (KBr): 3026, 2921, 1735, 1590, 1430, 1216, 1173, 1028, 958, 832, 697 cm⁻¹. **MS** (ESI) m/z (relative intensity): 350 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₆H₂₃N [M+H]⁺ 350.1903, found 349.1883.

5-(Dibenzylamino)-2-methoxyphenyl acetate (3ua): The general procedure A was followed using **1u** (84.5 mg, 0.40 mmol) and **2a** (143 μL, 1.20 mmol). Purification by column chromatography (PE/EA = 20:1) yielded **3ua** (93.3 mg, 65%) as yellow oli. ¹**H NMR** (400 MHz, CDCl₃) δ = 7.30–7.26 (m, 3H), 7.26–7.19 (m, 8H), 6.82–6.75 (m, 2H), 4.53 (s, 4H), 3.79 (s, 3H), 3.75 (s, 3H). ¹³**C NMR** (101 MHz, CDCl₃) δ = 167.1 (C_q), 151.2 (C_q), 143.2 (C_q), 138.4 (C_q), 138.4 (C_q), 128.6 (CH), 127.0 (CH), 120.7 (CH), 118.5 (CH), 116.2 (CH), 114.1 (CH), 56.8 (CH₃), 54.9 (CH₂), 29.7 (CH₃). **IR** (KBr): 2924, 1730, 1506, 1435, 1245, 1082, 1026, 807, 742, 697 cm⁻¹. **MS** (ESI) *m/z* (relative intensity): 362 (100) [M+H]⁺, 384 (21) [M+Na]⁺. **HR-MS** (ESI) *m/z* calcd for C₂₃H₂₃NO₃ [M+H]⁺ 362.1751, found 362.1751. The data are in agreement with those reported in literature. ¹³

N,N-dibenzyl-4-(2,4-dichlorophenoxy) aniline (3va): The general procedure A was followed

using **1v** (114 mg, 0.40 mmol) and **2a** (143 μ L, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded **3va** (128 mg, 74%) as white solid. **M.p.** = 64–65 °C. **1H NMR** (400 MHz, CDCl₃) δ = 7.40 (d, J = 2.5 Hz, 1H), 7.34 (t, J = 7.6 Hz, 4H), 7.28–7.26 (m, 6H), 7.09 (dd, J = 8.8, 2.5 Hz, 1H), 6.83 (d, J = 9.0 Hz, 2H), 6.76 (d, J = 8.8 Hz, 1H), 6.70 (d, J = 5.4 Hz, 2H), 4.64 (s, 4H). ¹³**C NMR** (151 MHz, CDCl₃) δ = 153.4 (C_q), 146.9 (C_q), 146.5 (C_q), 138.5 (C_q), 130.3 (CH), 128.8 (CH), 127.8 (CH), 127.6 (C_q), 127.2 (CH), 126.9 (CH), 125.0 (C_q), 120.5 (CH), 119.1 (CH), 113.9 (CH), 55.0 (CH₂). **IR** (KBr): 2922, 1603, 1511, 1470, 1231, 1053, 954, 812, 728, 696 cm⁻¹. **MS** (ESI) m/z (relative intensity): 434 (100) [M+H]⁺, 472 (3) [M+K]⁺. **HR-MS** (ESI) m/z calcd for C₂₆H₂₁Cl₂NO [M+H]⁺ 434.1070, found 434.1073.

N-(4-(dibenzylamino)-2-phenoxyphenyl) methanesulfonamide (3wa): The general procedure A was followed using 3w (123 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/EA = 10:1) yielded 3wa (97.9 mg, 53%) as white solid. **M.p.** = 100–101 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.35 (d, J = 9.0 Hz, 1H), 7.30 (t, J = 7.4 Hz, 4H), 7.25 (t, J = 7.1 Hz, 2H), 7.22 (t, J = 7.9 Hz, 2H), 7.16 (d, J = 6.9 Hz, 4H), 7.07 (t, J = 7.4 Hz, 1H), 6.81 (d, J = 7.9 Hz, 2H), 6.53 (dd, J = 9.0, 2.8 Hz, 1H), 6.22 (s, 1H), 6.21 (d, J = 2.8 Hz, 1H), 4.56 (s, 4H), 2.88 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ = 155.9 (C_q), 150.2 (C_q), 148.6 (C_q), 137.8 (C_q), 130.1 (CH), 128.9 (CH), 127.5 (CH), 127.3 (CH), 126.8 (CH), 123.9 (CH), 118.2 (CH), 116.5 (C_q), 108.7 (CH), 103.2 (CH), 55.1 (CH₂), 39.2 (CH₃). IR (KBr): 3061, 1615, 1515, 1489, 1393, 1155, 961, 749, 702, 691 cm⁻¹. MS (ESI) m/z (relative intensity): 459 (100) [M+H]⁺, 481 (38) [M+Na]⁺, 497 (1) [M+K]⁺. HR-MS (ESI) m/z calcd for C₂₇H₂₆N₂O₃S [M+H]⁺ 459.1737, found 459.1739.

N-(4-(benzylamino)-3-(trifluoromethyl)phenyl)isobutyramide (3xa): The general procedure A was followed using 1x (111 mg, 0.40 mmol) and 2a (143 μL, 1.20 mmol). Purification by column chromatography (PE/EA = 10:1) yielded 3xa (74.3 mg, 44%) as white solid. **M.p.** = 126–127 °C. ¹H NMR (600 MHz, d_6 -DMSO) δ = 9.66 (s, 1H), 7.82 (d, J = 2.5 Hz, 1H), 7.40 (dd, J = 8.9, 2.5 Hz, 1H), 7.33–7.29 (m, 4H), 7.20 (m, 1H), 6.58 (d, J = 9.0 Hz, 1H), 6.07 (t, J = 6.1 Hz, 1H), 4.42 (d, J = 6.0 Hz, 2H), 3.37 (s, 1H), 1.06 (d, J = 6.8 Hz, 6H). ¹³C NMR (151 MHz, d_6 -DMSO) δ = 174.7 (C_q), 141.2 (C_q), 139.8 (C_q), 128.4 (CH), 128.2 (C_q), 126.7 (CH), 126.7 (CH), 125.4 (q, ${}^1J_{C-F}$ = 271.8 Hz, C_q), 124.9 (CH), 117.60 (q, ${}^3J_{C-F}$ = 6.0 Hz, CH), 112.8 (CH), 111.2 (q, ${}^2J_{C-F}$ = 28.7 Hz, C_q), 45.9 (CH₂), 34.8 (CH), 19.5 (CH₃). ¹⁹F NMR (565 MHz, d_6 -DMSO) δ = -61.38. IR (KBr): 3267, 2970, 1658, 1549, 1426, 1337, 1251, 1107, 1048, 701 cm⁻¹. MS (ESI) m/z (relative intensity): 337 (100) [M+H]⁺, 359 (90) [M+Na]⁺. HR-MS (ESI) m/z calcd for C₁₈H₁₉F₃N₂O [M+H]⁺ 337.1522, found 337.1526.

N,*N*-bis(4-methylbenzyl)aniline (3ab): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2b (222 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 5a (68.3 mg, 56%) as white solid. M.p. = 92–93 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.20–7.16 (m, 3H), 7.16–7.14 (m, 7H), 6.76 (d, *J* = 8.1 Hz, 2H), 6.71 (t, *J* = 7.2 Hz, 1H), 4.62 (s, 4H), 2.35 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.4 (C_q), 136.6 (C_q), 135.7 (C_q), 129.4 (CH), 129.3 (CH), 126.8 (CH), 116.7 (CH), 112.6 (CH), 54.0 (CH₂), 21.2 (CH₃). IR (KBr): 3017, 2921, 1597, 1506, 1430, 1387, 1237, 1174, 792, 745, 688 cm⁻¹. MS (ESI) *m/z* (relative intensity): 302 (100) [M+H]⁺. HR-MS (ESI) *m/z* calcd for C₂₂H₂₃N [M+H]⁺

302.1903, found 302.1903. The data are in agreement with those reported in literature.²

N,*N*-bis(4-(*tert*-butyl)benzyl)aniline (3ac): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2c (272 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ac (55.4 mg, 36%) as white solid. **M.p.** = 105–106 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.27 (d, J = 7.8 Hz, 4H), 7.13–7.07 (m, 6H), 6.68 (d, J = 8.1 Hz, 2H), 6.61 (t, J = 7.3 Hz, 1H), 4.55 (s, 4H), 1.25 (s, 18H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.8 (C_q), 149.5 (C_q), 135.7 (C_q), 129.3 (CH), 126.5 (CH), 125.6 (CH), 116.6 (CH), 112.5 (CH), 53.9 (CH₂), 34.6 (C_q), 31.6 (CH₃). **IR** (KBr): 3022, 2956, 1599, 1505, 1412, 1268, 1016, 937, 819, 748, 690 cm⁻¹. **MS** (ESI) *m/z* (relative intensity): 386 (100) [M+H]⁺. **HR-MS** (ESI) *m/z* calcd for C₂₈H₃₅N [M+H]⁺ 386.2842, found 386.2842. The data are in agreement with those reported in literature. ¹⁴

N,*N*-bis(4-iodobenzyl)aniline (3ad): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2d (356 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ad (140.8 mg, 67%) as white solid. M.p. = 98–99 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.64 (d, J = 8.1 Hz, 4H), 7.19 (t, J = 7.7 Hz, 2H), 6.99 (d, J = 8.1 Hz, 4H), 6.75 (t, J = 7.4 Hz, 1H), 6.69 (d, J = 8.2 Hz, 2H), 4.55 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.7 (C_q), 138.2 (C_q), 137.8 (CH), 129.5 (CH), 128.8 (CH), 117.7 (CH), 112.9 (CH), 92.3 (C_q), 54.1 (CH₂). IR (KBr): 3036, 2923, 1597, 1504, 1390, 1234, 1006, 960, 800, 744, 689 cm⁻¹. MS (ESI) m/z (relative intensity): 525 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for

C₂₀H₁₇I₂N [M+H]⁺ 525.9523, found 525.9537. The data are in agreement with those reported in literature.¹⁵

N,*N*-bis(4-chlorobenzyl)aniline (3ae): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2e (246 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ae (89.7 mg, 66%) as white solid. M.p. = 120–121 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.29 (d, J = 8.2 Hz, 4H), 7.19–7.15 (m, 6H), 6.74 (t, J = 7.3 Hz, 1H), 6.70 (d, J = 8.2 Hz, 2H), 4.57 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.8 (C_q), 137.0 (C_q), 132.8 (C_q), 129.5 (CH), 129.0 (CH), 128.1 (CH), 117.5 (CH), 112.9 (CH), 53.9 (CH₂). IR (KBr): 2924, 1597, 1505, 1487, 1384, 1237, 1090, 1014, 812, 746, 687 cm⁻¹. MS (ESI) m/z (relative intensity): 342 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₀H₁₇³⁵Cl₂N [M+H]⁺ 342.0811, found 342.0814. The data are in agreement with those reported in literature. ¹

N,N-bis(4-fluorobenzyl)aniline (3af): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2f (226 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3af (71.7 mg, 58%) as brown oil. ¹H NMR (600 MHz, CDCl₃) δ = 7.22–7.19 (m, 6H), 7.05–6.98 (m, 4H), 6.75 (t, J = 8.7 Hz, 3H), 4.59 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 162.0 (d, ¹J_{C-F} = 244.6 Hz, C_q), 149.0 (C_q), 134.1 (C_q), 129.4 (CH), 128.4 (d, ³J_{C-F} = 8.2 Hz, CH), 117.4 (CH), 115.6 (d, ²J_{C-F} = 21.3 Hz, CH), 112.9 (CH), 53.7 (CH₂). ¹⁹F NMR (565 MHz, DMSO-d₆) δ = -116.39. IR (KBr): 3039, 2922, 1599, 1507, 1354, 1222, 1154, 1094, 822, 748, 692 cm⁻¹. MS (ESI) m/z (relative intensity): 310 (100) [M+H]⁺. HR-MS (ESI)

m/z calcd for $C_{20}H_{17}F_2N$ [M+H]⁺ 310.1402, found 310.1407. The data are in agreement with those reported in literature.²

N,N-bis(4-(trifluoromethyl)benzyl)aniline (3ag): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2g (286 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ag (108 mg, 66%) as white solid. **M.p.** = 74–75 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.60 (d, J = 8.1 Hz, 4H), 7.38 (d, J = 8.1 Hz, 4H), 7.21 (t, J = 7.8 Hz, 2H), 6.79 (t, J = 7.3 Hz, 1H), 6.72 (d, J = 8.2 Hz, 2H), 4.71 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.6 (C_q), 142.7 (C_q), 129.7 (q, ${}^2J_{\text{C-F}}$ = 33.2 Hz, C_q), 129.6 (CH), 127.1 (CH), 125.8 (q, ${}^3J_{\text{C-F}}$ = 3.7 Hz, CH), 124.3 (q, ${}^1J_{\text{C-F}}$ = 271.8 Hz, C_q), 118.0 (CH), 112.9 (CH), 54.4 (CH₂). ¹⁹F NMR (565 MHz, CDCl₃) δ = -62.44. IR (KBr): 2924, 1601, 1507, 1418, 1325, 1131, 1065, 1016, 825, 747, 690 cm⁻¹. MS (ESI) m/z (relative intensity): 410 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₂H₁₇F₆N [M+H]⁺ 410.1338, found 410.1339. The data are in agreement with those reported in literature. ¹⁵

Dimethyl 4,4'-((phenylazanediyl)bis(methylene))dibenzoate (3ah): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2h (275 mg, 1.20 mmol). Purification by column chromatography (PE/EA = 20:1) yielded 3ah (79.4 mg, 58%) as white solid. M.p. = 96–97 °C. ¹H NMR (600 MHz, CDCl₃) δ = 8.00 (d, J = 8.1 Hz, 4H), 7.32 (d, J = 8.1 Hz, 4H), 7.18 (t, J = 8.2 Hz, 2H), 6.75 (t, J = 7.3 Hz, 1H), 6.70 (d, J = 8.1 Hz, 2H), 4.69 (s, 4H), 3.91 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ = 167.0 (C_q), 148.6 (C_q), 144.0 (C_q), 130.2 (CH), 129.5

(CH), 129.2 (C_q), 126.8 (CH), 117.7 (CH), 112.9 (CH), 54.6 (CH₂), 52.2 (CH₃). **IR** (KBr): 2949, 2923, 1708, 1598, 1506, 1430, 1283, 1105, 885, 753, 694 cm⁻¹. **MS** (ESI) m/z (relative intensity): 390 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for C₂₄H₂₃NO₄ [M+H]⁺ 390.1700, found 390.1699. The data are in agreement with those reported in literature.¹⁶

N,*N*-bis(4-(trifluoromethoxy)benzyl)aniline (3ai): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2i (306 mg, 1.20 mmol). Purification by column chromatography (PE/EA = 50:1) yielded 3ai (128 mg, 73%) as colorless liquid. ¹H NMR (600 MHz, CDCl₃) δ = 7.28 (d, J = 8.2 Hz, 4H), 7.23–7.20 (m, 2H), 7.19 (d, J = 8.2 Hz, 4H), 6.77 (t, J = 7.3 Hz, 1H), 6.74 (d, J = 8.2 Hz, 2H), 4.64 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.8 (C_q), 148.3 (C_q), 137.2 (C_q), 129.6 (CH), 128.1 (CH), 120.6 (q, ${}^{1}J_{C-F}$ = 256.7 Hz, C_q), 121.4 (CH), 117.6 (CH), 112.8 (CH), 53.81 (CH₂). ¹⁹F NMR (565 MHz, CDCl₃) δ = -57.89. IR (KBr): 3444, 2920, 2849, 1643, 1507, 1384, 1265, 1018, 746, 706 cm⁻¹. MS (ESI) m/z (relative intensity): 442 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₂₂H₁₇F₆NO₂ [M+H]⁺ 442.1236, found 442.1238.

N,N-bis(2-methylbenzyl)aniline (3aj): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2j (222 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3aj (60.0 mg, 50%) as white solid. **M.p.** = 141–142 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.34–7.30 (m, 6H), 7.30–7.27 (m, 4H), 6.81 (t, *J* = 7.2 Hz, 1H), 6.73 (d, *J* = 8.1 Hz, 2H), 4.70 (s, 4H), 2.39 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.1 (C_q), 135.7

(C_q), 135.1 (C_q), 130.5 (CH), 129.3 (CH), 126.9 (CH), 126.3 (CH), 125.9 (CH), 116.6 (CH), 112.2 (CH), 52.5 (CH₂), 19.0 (CH₃). **IR** (KBr): 3025, 2923, 1597, 1507, 1352, 1265, 1049, 989, 859, 743, 690 cm⁻¹. **MS** (ESI) m/z (relative intensity): 302 (100) [M+H]⁺. **HR-MS** (ESI) m/z calcd for $C_{22}H_{23}N$ [M+H]⁺ 302.1903, found 302.1902. The data are in agreement with those reported in literature.¹⁵

N,*N*-bis(2-chlorobenzyl)aniline (3ak): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2k (246 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ak (86.3 mg, 63%) as white solid. **M.p.** = 128–129 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.46–7.43 (m, 2H), 7.31–7.26 (m, 6H), 7.22 (t, *J* = 8.5 Hz, 2H), 6.77 (t, *J* = 7.3 Hz, 1H), 6.64 (d, *J* = 8.4 Hz, 2H), 4.76 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.3 (C_q), 135.2 (C_q), 133.1 (C_q), 129.8 (CH), 129.5 (CH), 128.4 (CH), 127.8 (CH), 127.2 (CH), 117.3 (CH), 112.2 (CH), 52.9 (CH₂). **IR** (KBr): 2926, 1598, 1505, 1441, 1356, 1235, 1047, 960, 755, 743, 699 cm⁻¹. **MS** (ESI) *m/z* (relative intensity): 342 (100) [M+H]⁺. **HR-MS** (ESI) *m/z* calcd for C₂₀H₁₇³⁵Cl₂N [M+H]⁺ 342.0811, found 342.0812. The data are in agreement with those reported in literature.²

N,N-bis(3-methylbenzyl)aniline (3al): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2l (222 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3al (66.5 mg, 55%) as yellow oil. ¹H NMR (600 MHz, CDCl₃) δ = 7.26–7.20 (m, 4H), 7.13–7.08 (m, 6H), 6.83 (d, J = 8.2 Hz, 2H), 6.75 (t, J = 7.3 Hz, 1H), 4.66 (s, 4H), 2.38 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.6 (C_q), 138.9 (C_q), 138.4 (C_q), 129.3

(CH), 128.6 (CH), 127.8 (CH), 127.6 (CH), 124.1 (CH), 116.9 (CH), 112.9 (CH), 54.5 (CH₂), 21.6 (CH₃). **IR** (KBr): 3024, 2918, 1598, 1506, 1349, 1227, 1091, 989, 776, 747, 692 cm⁻¹. **MS** (ESI) *m/z* (relative intensity): 302 (100) [M+H]⁺. **HR-MS** (ESI) *m/z* calcd for C₂₂H₂₃N [M+H]⁺ 302.1903, found 302.1906. The data are in agreement with those reported in literature.¹⁵

N,*N*-bis(3-chlorobenzyl)aniline (3am): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2m (246 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3am (83.2 mg, 61%) as colorless oil. ¹H NMR (600 MHz, CDCl₃) $\delta = 7.25-7.21$ (m, 6H), 7.19 (t, J = 8.1 Hz, 2H), 7.12 (d, J = 6.7 Hz, 2H), 6.75 (t, J = 7.3 Hz, 1H), 6.70 (d, J = 7.8 Hz, 2H), 4.60 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) $\delta = 148.6$ (C_q), 140.8 (C_q), 134.8 (C_q), 130.1 (CH), 129.5 (CH), 127.4 (CH), 126.8 (CH), 124.9 (CH), 117.6 (CH), 112.7 (CH), 54.0 (CH₂). IR (KBr): 3060, 2924, 1596, 1505, 1348, 1198, 1076, 958, 779, 748, 682 cm⁻¹. MS (ESI) *m/z* (relative intensity): 342 (100) [M+H]⁺. HR-MS (ESI) *m/z* calcd for C₂₀H₁₇³⁵Cl₂N [M+H]⁺ 342.0811, found 342.0811.

N,N-bis(naphthalen-2-ylmethyl)aniline (3an): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2n (265 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3an (82.4 mg, 55%) as white solid. M.p. = 154–155 °C. ¹H NMR (600 MHz, CDCl₃) δ = 7.85–7.81 (m, 4H), 7.78 (d, J = 7.6 Hz, 2H), 7.72 (s, 2H), 7.49–7.44 (m, 4H), 7.41 (d, J = 8.4 Hz, 2H), 7.20 (t, J = 7.7 Hz, 2H), 6.87 (d, J = 8.1 Hz, 2H), 6.75 (t, J = 7.4 Hz, 1H), 4.88 (s, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.4 (C_q), 136.2 (C_q), 133.7 (C_q), 132.8 (C_q), 129.5 (CH), 128.6 (CH), 127.9 (CH), 127.8 (CH), 126.3 (CH), 125.8 (CH), 125.3

(CH), 125.2 (CH), 117.1 (CH), 112.9 (CH), 54.6 (CH₂). **IR** (KBr): 3043, 2923, 1599, 1505, 1372, 1189, 963, 875, 807, 748, 692 cm⁻¹. **MS** (ESI) *m/z* (relative intensity): 374 (100) [M+H]⁺. **HR-MS** (ESI) *m/z* calcd for C₂₈H₂₃N [M+H]⁺ 374.1903, found 374.1909.

N,*N*-diallylaniline (3ao): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2o (145 mg, 1.20 mmol). Purification by column chromatography (PE/DCM = 50:1) yielded 3ao (24.5 mg, 35%) as yellow oil. ¹H NMR (600 MHz, CDCl₃) δ = 7.24 (t, J = 7.9 Hz, 2H), 6.76–6.71 (m, 3H), 5.93–5.87 (m, 2H), 5.24–5.19 (m, 4H), 3.97 (d, J = 4.9 Hz, 4H). ¹³C NMR (151 MHz, CDCl₃) δ = 148.8 (C_q), 134.2 (CH), 129.2 (CH), 116.4 (CH), 116.1 (CH₂), 112.5 (CH), 52.9 (CH₂). IR (KBr): 3061, 2924, 1599, 1505, 1388, 1233, 1182, 989, 919, 747, 692 cm⁻¹. MS (ESI) m/z (relative intensity): 174 (100) [M+H]⁺. HR-MS (ESI) m/z calcd for C₁₂H₁₅N [M+H]⁺ 174.1277, found 174.1272. The data are in agreement with those reported in literature.³

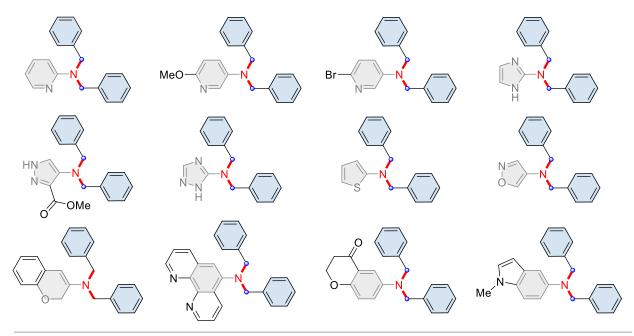
N,*N*-bis(2-methylallyl)aniline (3ap): The general procedure A was followed using 1a (41 μL, 0.40 mmol) and 2p (162 mg, 1.20 mmol). Purification by column chromatography (PE/EA = 50:1) yielded 3ap (34.3 mg, 43%) as colorless oli. ¹H NMR (600 MHz, CDCl₃) δ = 7.18 (t, J = 8.3 Hz, 2H), 6.66 (t, J = 7.2 Hz, 1H), 6.62 (d, J = 8.2 Hz, 2H), 4.85 (s, 2H), 4.79 (s, 2H), 3.81 (s, 4H), 1.74 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ = 149.0 (C_q), 140.7 (C_q), 129.0 (CH), 116.1 (CH), 112.0 (CH), 110.3 (CH₂), 56.4 (CH₂), 20.2 (CH₃). IR (KBr): 2924, 1599, 1505, 1389, 1229, 1186, 894, 746, 690, 512 cm⁻¹. MS (ESI) *m/z* (relative intensity): 202 (100) [M+H]⁺. HR-MS (ESI) *m/z* calcd for C₁₄H₁₉N [M+H]⁺ 202.1590, found 202.1586. The data are

in agreement with those reported in literature. 17

Unsuccessful Substrates:

Representative non-reactive examples

Substrates of nitroaromatics:



Substrates of alkyl bromide:

Scheme S9. Unsuccessful substrates.

5. Mechanistic Studies

5.1 Control Experiments

Scheme S10. Control experiments for potential intermediate.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with potential intermediate (0.4 mmol), benzyl bromide **2a** (143 μ L, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μ L, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol, if used) and MeOH (4.0 mL), the reaction cells were performed at 50 °C with a constant current of 4.0 mA maintained for 12 h. The electrodes were washed with dichloromethane (3 \times 4.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. The residue was purified by chromatography on silica gel (PE/DCM) to afford the corresponding products **3aa**.

5.2 Investigation on the Role of 2,6-Lutidine

Scheme S11. Investigation on the role of 2,6-lutidine.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with nitrobenzene **1a** (41 μ L, 0.40 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μ L, 1.20 mmol, if used), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL). Electrocatalysis was performed at 50 °C with a constant current of 4.0 mA maintained for 12 h. The yields of aniline **Int-III** were determined by GC-MS analysis using 1,3,5-trimethoxybenzene as the internal standard.

Scheme S12. GC-MS monitors compounds in the reaction system.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with nitrobenzene **1a** (41 μ L, 0.4 mmol), benzyl bromide **2a** (143 μ L, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μ L, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL), the reaction cells were performed at 50 °C with a constant current of 4.0 mA maintained for 12 h. The yields of corresponding products were determined by GC-MS analysis using 1,3,5-trimethoxybenzene as the internal standard.

Scheme S13. Investigation of the products without 2,6-lutidine.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with nitrobenzene **1a** (41 μ L, 0.4 mmol), benzyl bromide **2a** (143 μ L, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL), the reaction cells were performed at 50 °C with a constant current of 4.0 mA maintained for 12 h. The yields of corresponding products were determined by GC-MS analysis using 1,3,5-trimethoxybenzene as the internal standard.

5.3 Cyclic Voltammetry Experiments

The cyclic voltammetry measurements were conducted with a Lanlike LK98BII electrochemical workstation and following analysis was performed with LK98BII software. A glassy-carbon (GC) electrode (3.0 mm-diameter, disc-electrode) was used as the working electrode, polished with 0.3 and 0.05 μm aluminum oxide, and then washed with distilled water and acetone before air drying. A platinum plate (1.0 cm × 1.0 cm × 0.1 cm) was used as the auxiliary electrode and an Ag/AgCl electrode was used as the reference. The voltammograms were recorded at room temperature under N₂ atmosphere. The solution was degassed by nitrogen gas bubbling for 5 mins before each measurement. The scan rate was 100 mVs⁻¹. Details of measurements are indicated in the respective figures and descriptions.

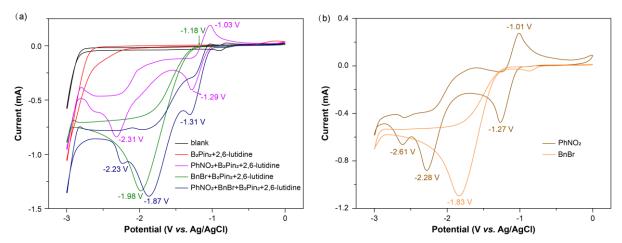


Figure S6. Cyclic voltammograms at 100 mVs⁻¹ using MeCN (4.0 mL) as solvent and TBABF₄ (0.10 M) as the electrolyte; blank (black), B₂Pin₂ (40 mM) + 2,6-lutidine (60 mM) (red line), PhNO₂ (20 mM) + B₂Pin₂ (40 mM) + 2,6-lutidine (60 mM) (purple line), BnBr (60 mM) + B₂Pin₂ (40 mM) + 2,6-lutidine (60 mM) (green line), PhNO₂ (20 mM) + BnBr (60 mM) + B₂Pin₂ (40 mM) + 2,6-lutidine (60 mM) (blue line), PhNO₂ (20 mM) (brown line), BnBr (60 mM) (orange line). The scan rate is 100 mVs⁻¹, ranging from 0.0 V to -3.0 V. Initial scan direction: negative.

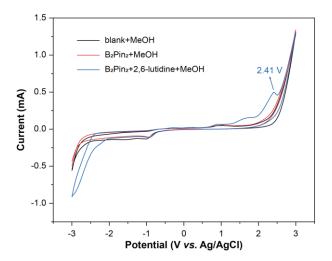


Figure S7. Cyclic voltammograms at 100 mVs^{-1} using MeCN (4.0 mL) as solvent and TBABF₄ (0.10 M) as the electrolyte; blank + MeOH (0.8 mL) (black), B₂Pin₂ (40 mM) + MeOH (0.8 mL) (red), B₂Pin₂ (40 mM) + 2,6-lutidine (60 mM) + MeOH (0.8 mL) (blue line). The scan rate is 100 mVs^{-1} , with scanning direction from -3.0 V to 3.0 V.

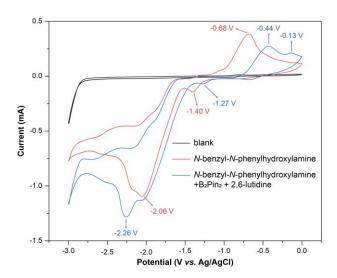
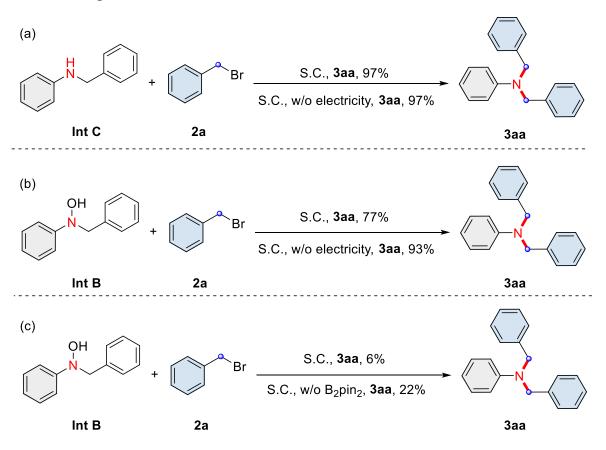


Figure S8. Cyclic voltammograms at 100 mVs⁻¹ using MeCN (4.0 mL) as solvent and TBABF₄ (0.10 M) as the electrolyte; blank (black), *N*-benzyl-*N*-phenylhydroxylamine (20 mM) (red line), *N*-benzyl-*N*-phenylhydroxylamine (20 mM) + B₂Pin₂ (40 mM) + 2,6-lutidine (60 mM) (blue line).

5.4 Control Experiments



Scheme S14. Studies on *N*-benzylaniline and *N*-benzylhydroxylamine as substrates.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with potential intermediate (0.4 mmol), benzyl bromide **2a** (143 μ L, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μ L, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol, if used) and MeOH (4.0 mL), the reaction cells were performed at 50 °C with a constant current of 4.0 mA maintained for 12 h, or without electricity for control experiments. The electrodes were washed with dichloromethane (3 \times 4.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. The residue was purified by chromatography on silica gel (PE/DCM) to afford the corresponding products **3aa**.

5.5 11 B NMR Analysis on the Reaction Components

Scheme S15. ¹¹B NMR analysis on the reaction components.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with nitrobenzene **1a** (41µL, 0.4 mmol), benzyl bromide **2a** (143 µL, 1.20 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 µL, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL), the reaction cells were performed at 50 °C with a constant current of 4.0 mA maintained for 12 h. When the electrolysis was done, the resultant reaction mixture was cooled to room temperature. Transfer the reaction to a nitrogen-filled glovebox to open the reaction. A 200 µL aliquot of the solution was diluted with 400 µL CDCl₃. Then 600 µL of the resultant solution was added to an NMR sample tube, and the sample was analyzed by ¹¹B NMR spectroscopy¹⁸⁻¹⁹.

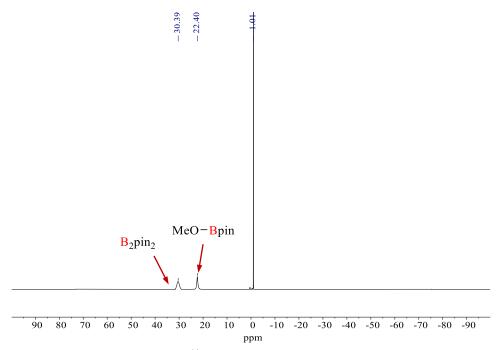


Figure S9. The ¹¹B NMR spectrum of the reaction.

5.6 Radical Trapping Experiments

Scheme S16. Radical trapping experiments.

In order to confirm whether the reaction undergoes a radical mechanism, radical trapping experiments were performed. The yields of corresponding products were determined by GC-MS analysis using 1,3,5-trimethoxybenzene as the internal standard. When the radical scavenger 1,1-diphenylethylene was added, the yield of product 3aa slightly decreased to 70%. The radical generated from 2a in the reaction could be captured by 1,1-diphenylethylene (detected by GC-MS). This result suggested that the reaction mainly proceed through two-electron process.

5.7 Competition Experiments

Scheme S17. Competition experiments electron-donating/withdrawing benzyl bromide

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm × 20 mm × 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm × 20 mm × 1.5 mm). The cell was charged with nitrobenzene **1a** (41 μL, 0.40 mmol), 1-(bromomethyl)-4-methylbenzene **2b** (111 mg, 0.60 mmol), 1-(bromomethyl)-4-(trifluoromethyl)benzene **2g** (143 mg, 0.60 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μL, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL). Electrocatalysis was performed at 50 °C with a constant current of 4.0 mA maintained for 2 h. The electrodes were washed with dichloromethane (3 × 4.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. Upon GC analysis, the residue gave the corresponding products **3aa**, **3abg** in yields of 17% and 10%, respectively. The yields of corresponding products were determined by GC-MS analysis using 1,3,5-trimethoxybenzene as the internal standard.

Scheme S18. Comparative reactions between: aliphatic bromides and aromatic bromides.

The electrocatalysis was carried out in an undivided cell with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm) and a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The cell was charged with nitrobenzene **1a** (41 μ L, 0.40 mmol), benzyl bromide **2a** (71 μ L, 0.60 mmol), 3-bromo-2-methylprop-1-ene **2p** (61 μ L, 0.60 mmol), TBABF₄ (198 mg, 0.60 mmol), 2,6-lutidine (140 μ L, 1.20 mmol), B₂Pin₂ (203 mg, 0.80 mmol) and MeOH (4.0 mL). Electrocatalysis was performed at 50 °C with a constant current of 4.0 mA maintained for 2 h. The electrodes were washed with dichloromethane (3 \times 4.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. Upon GC analysis, the residue gave the corresponding products **3aa**, **3abg** in yields of 11% and 5%, respectively. The yields of corresponding products were determined by GC-MS analysis using 1,3,5-trimethoxybenzene as the internal standard.

5.8 Divided Cell Experiments

Scheme S19. Divided cell experiments with B₂Pin₂ at the cathode.

In a N₂-filled glovebox, a pre-assembled Cation Exchange Membrane $(1.5\times1.5 \text{ cm}^2)$ divided H-type cell was equipped with a magnetic stir bar in each 10 mL chamber. The anodic chamber was charged with TBABF₄ (198 mg, 0.15 M), 2,6-lutidine (140 μ L, 1.20 mmol), and MeOH (4.0 mL). The anodic chamber was sealed by rubber septum cap with a graphite felt (GF) anode (10 mm \times 20 mm \times 3.0 mm). The cathodic chamber was charged with nitrobenzene 1a (49.2 mg, 0.4 mmol), benzyl bromide 2a (143 μ L, 1.20 mmol), B₂Pin₂ (203 mg, 0.8 mmol) TBABF₄ (198 mg, 0.15 M) and MeOH (4.0 mL). The cathodic chamber was sealed by rubber septum cap with a nickel foam (Ni foam) cathode (10 mm \times 20 mm \times 1.5 mm). The distance between the anode and cathode was almost 5.5 cm. The cell was transferred out from the glovebox. After that, constant current of 4.0 mA was conducted for 12 h at 50 °C with

stirring. Upon completion of the electrolysis, the solution in the cathodic chamber of the cell was transferred to a 100 mL round-bottomed flask. The Ni foam cathode and cathodic chamber were rinsed with dichloromethane ($3 \times 4.0 \text{ mL}$) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. The residue was purified by chromatography on silica gel (PE/DCM = 50:1) to afford the corresponding products 3aa.

Scheme S20. Divided cell experiments with B₂Pin₂ at the anode.

In a N₂-filled glovebox, a pre-assembled Cation Exchange Membrane $(1.5\times1.5~\text{cm}^2)$ divided H-type cell was equipped with a magnetic stir bar in each 10 mL chamber. The anodic chamber was charged with TBABF₄(198 mg, 0.15 M), B₂Pin₂(203 mg, 0.8 mmol), 2,6-lutidine (140 μ L, 1.20 mmol), and MeOH (4.0 mL). The anodic chamber was sealed by rubber septum cap with a graphite felt (GF) anode (10 mm × 20 mm × 3.0 mm). The cathodic chamber was charged with nitrobenzene **1a** (49.2 mg, 0.4 mmol), benzyl bromide **2a** (143 μ L, 1.20 mmol), TBABF₄ (198 mg, 0.15 M) and MeOH (4.0 mL). The cathodic chamber was sealed by rubber septum cap with a nickel foam (Ni foam) cathode (10 mm × 20 mm × 1.5 mm). The distance between the anode and cathode was almost 5.5 cm. The cell was transferred out from the glovebox. After that, constant current of 4.0 mA was conducted for 12 h at 50 °C with stirring. Upon completion of the electrolysis, the solution in the cathodic chamber of the cell was transferred to a 100 mL round-bottomed flask. The Ni foam cathode and cathodic chamber were rinsed with dichloromethane (3 × 4.0 mL) in an ultrasonic bath. The combined mixtures were filtered and concentrated under reduced pressure at 50 °C. The residue was purified by chromatography on silica gel (PE/DCM = 50:1) to afford the corresponding products **3aa**.



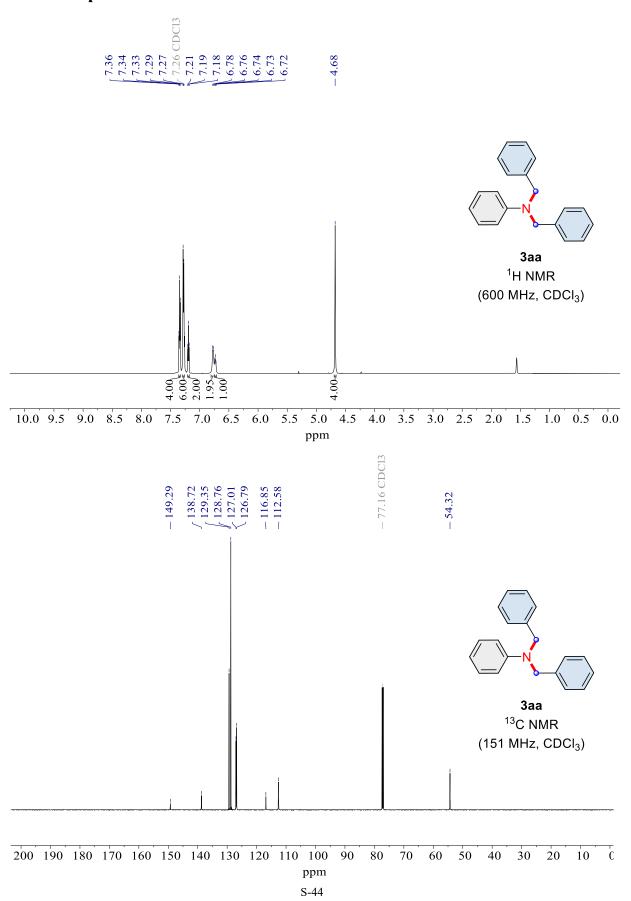
Figure S10. Pictures of the divided cell reaction setups.

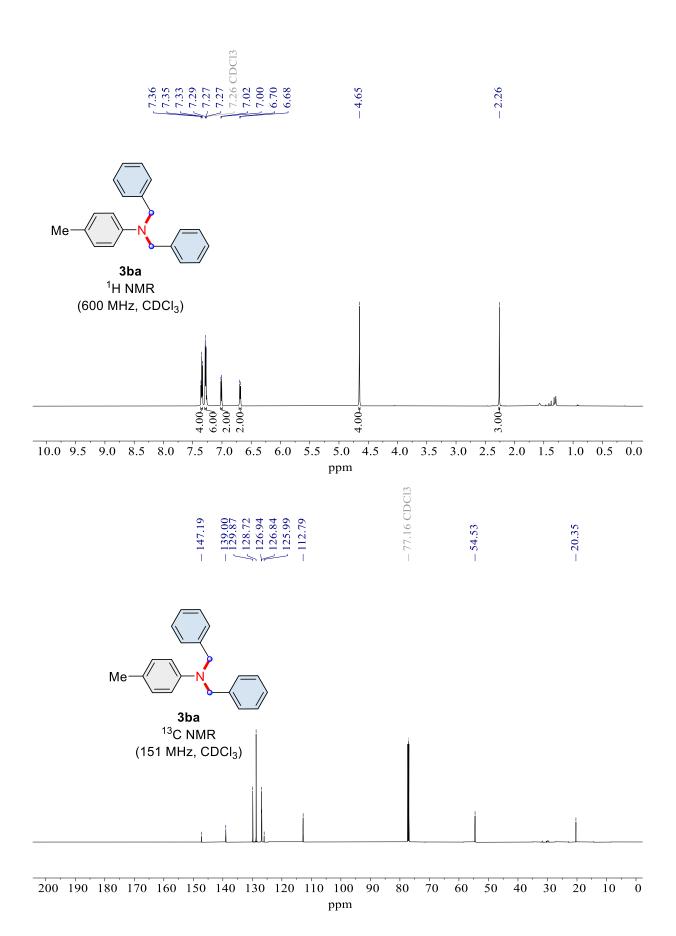
6. References

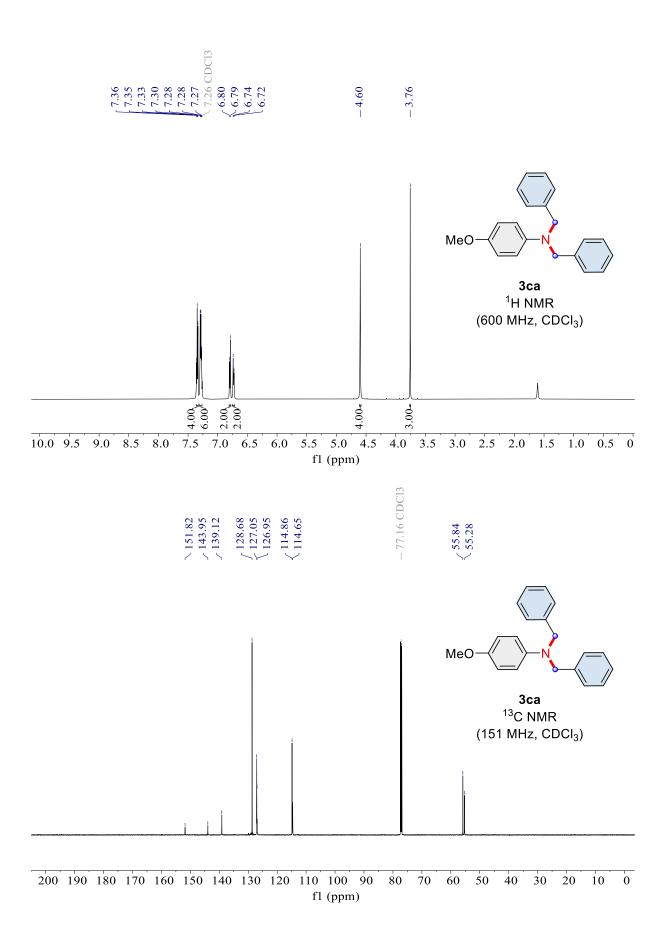
- 1. S. Wang, T. Li, C. Gu, J. Han, C.-G. Zhao, C. Zhu, H. Tan, J. Xie, Decarboxylative Tandem C–N Coupling with Nitroarenes *via* SH₂ Mechanism, *Nat. Commun.* **2022**, *13*, 2432–2441.
- 2. C. Feng, Y. Liu, S. Peng, Q. Shuai, G. Deng, C.-J. Li, Ruthenium-Catalyzed Tertiary Amine Formation from Nitroarenes and Alcohols, *Org. Lett.* **2010**, *12*, 4888–4891.
- M. H. Nguyen, A. B. Smith, III, Copper-Catalyzed Electrophilic Amination of Organolithiums Mediated by Recoverable Siloxane Transfer Agents, *Org. Lett.* 2013, *15*, 4872–4875.
- 4. Y.-M. Zhang, X. Wang, W. Li, W. Zhang, M. Li, S. X.-A. Zhang, Bio-Inspired Enol-Degradation for Multipurpose Oxygen Sensing, *Chem. Commun.* **2014**, *50*, 13477–13480.
- 5. N. Matsuda, K. Hirano, T. Satoh, M. Miura, Copper-Catalyzed Amination of Arylboronates with N,N-Dialkylhydroxylamines, *Angew. Chem. Int. Ed.* **2012**, *51*, 3642–3645.
- J. Li, X. Chen, S. Xie, H. Wang, J. Mo, H. Huang, Photoredox/Bismuth Relay Catalysis Enabling Reductive Alkylation of Nitroarenes with Aldehydes, *Chem. Eur. J.* 2024, 30, e202401456.
- M. B. Johansen, A. T. Lindhardt, Copper-Catalyzed and Additive Free Decarboxylative Trifluoromethylation of Aromatic and Heteroaromatic Iodides, *Org. Biomol. Chem.* 2020, 18, 1417–1425.
- 8. B. A. Dar, V. Shrivastava, A. Bowmik, M. A. Wagay, B. Singh, An Expeditious *N,N*-Dibenzylation of Anilines under Ultrasonic Irradiation Conditions Using Low Loading Cu(II)-Clay Heterogeneous Catalyst, *Tetrahedron Lett.* **2015**, *56*, 136–141.
- A. Ashimori, T. Ono, Y. Inoue, S. Morimoto, M. Eda, T. Uchida, Y. Ohtaki, Y. Fujino, H. Kido, Y. Ogura, C. Fukaya, M. Watanabe, K. Yokoyama, Novel 1, 4-Dihydropyridine Calcium Antagonists. II. Synthesis and Antihypertensive Activity of 3-[4-(Substituted Amino)phenylalkyl]ester Derivatives, *Chem. Pharm. Bull.* 1991, 39, 91–99.
- 10. Y.-S. Choi, Y.-J. Kim, L.-L. Shen, Y. S. Lee, J.-H. Jeong, Direct N-Alkylation of Aromatic Amines Using a Microflow Reactor: Enhancement of Selectivity and Reactivity, Synlett 2015, 26, 970–974.

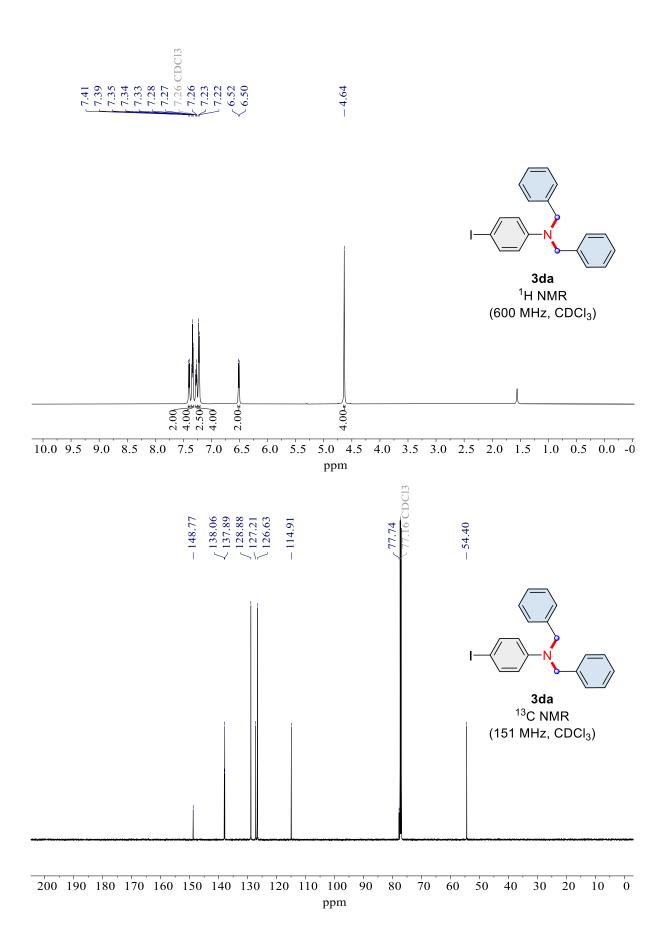
- 11. Y. Fukami, T. Wada, T. Meguro, N. Chida, T. Sato, Copper-Catalyzed Electrophilic Amination Using *N*-Methoxyamines, *Org. Biomol. Chem.* **2016**, *14*, 5486–5489.
- 12. Q. Xiao, L. Tian, R. Tan, Y. Xia, D. Qiu, Y. Zhang, J. Wang, Transition-Metal-Free Electrophilic Amination of Arylboroxines, *Org. Lett.* **2012**, *14*, 4230–4233.
- 13. S. A. Shaw, A. J. Clarke, T. J. Friends, A. Mathur, M. C. Myers, J. Li, K. B. Pabbisetty, S. Su, G. O. Tora, B. P. Vokits, N. R. Wurtz, D. J. P. Pinto, L. Pasunoori, Norbornyl Benzamide Derivatives as RXFP1 Agonists and Their Preparation, WO 2023077040 Al, 2023.
- 14. S.-Y. Hsueh, C.-C. Lai, S.-H. Chiu, Squaraine-Based [2]Rotaxanes that Function as Visibly Active Molecular Switches, *Chem. Eur. J.* **2010**, *16*, 2997–3000.
- 15. Y. Du, S. Oishi, S. Saito, Selective *N*-Alkylation of Amines with Alcohols by Using Non-Metal-Based Acid–Base Cooperative Catalysis, *Chem. Eur. J.* **2011**, *17*, 12262–12267.
- J. R. Johnson, N. Fu, E. Arunkumar, W. M. Leevy, S. T. Gammon, D. Piwnica-Worms, B.
 D. Smith, Squaraine Rotaxanes: Superior Substitutes for Cy-5 in Molecular Probes for Near-Infrared Fluorescence Cell Imaging, *Angew. Chem. Int. Ed.* 2007, 46, 5528–5531.
- 17. A. Gansäuer, D. von Laufenberg, C. Kube, T. Dahmen, A. Michelmann, M. Behlendorf, R. Sure, M. Seddiqzai, S. Grimme, D. V. Sadasivam, G. D. Fianu, R. A. Flowers Ii, Mechanistic Study of the Titanocene(III)-Catalyzed Radical Arylation of Epoxides, *Chem. Eur. J.* 2015, 21, 280–289.
- 18. S. Kobayashi, P. Xu, T. Endo, M. Ueno, T. Kitanosono, Chiral Copper(II)-Catalyzed Enantioselective Boron Conjugate Additions to α,β-Unsaturated Carbonyl Compounds in Water, *Angew. Chem. Int. Ed.* **2012**, *51*, 12763–12766.
- 19. X. Cao, W. Wang, K. Lu, W. Yao, F. Xue, M. Ma, Magnesium-Catalyzed Hydroboration of Organic Carbonates, Carbon Dioxide and Esters, *Dalton Trans.* **2020**, *49*, 2776–2780.

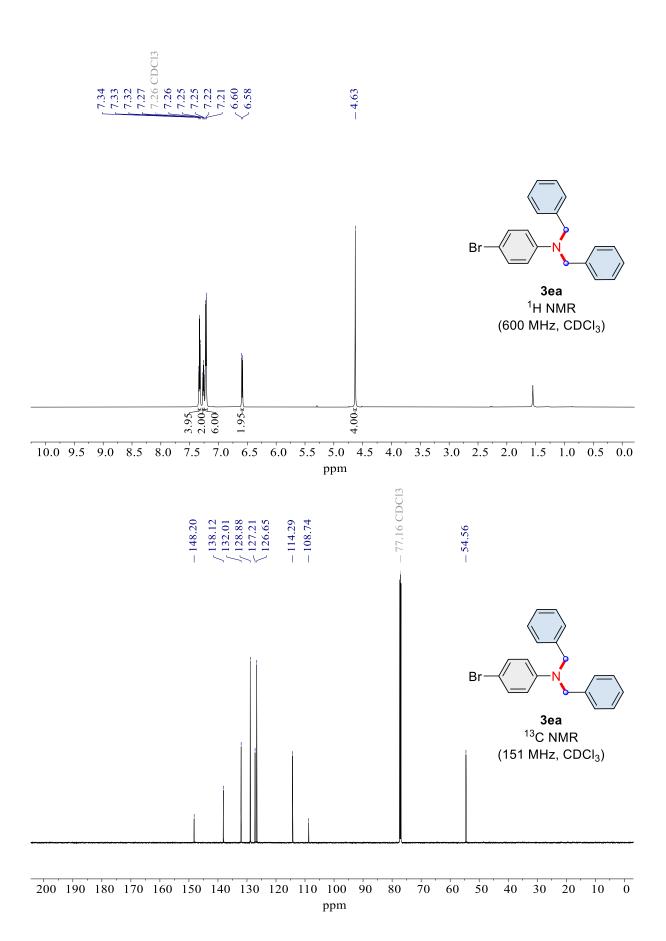
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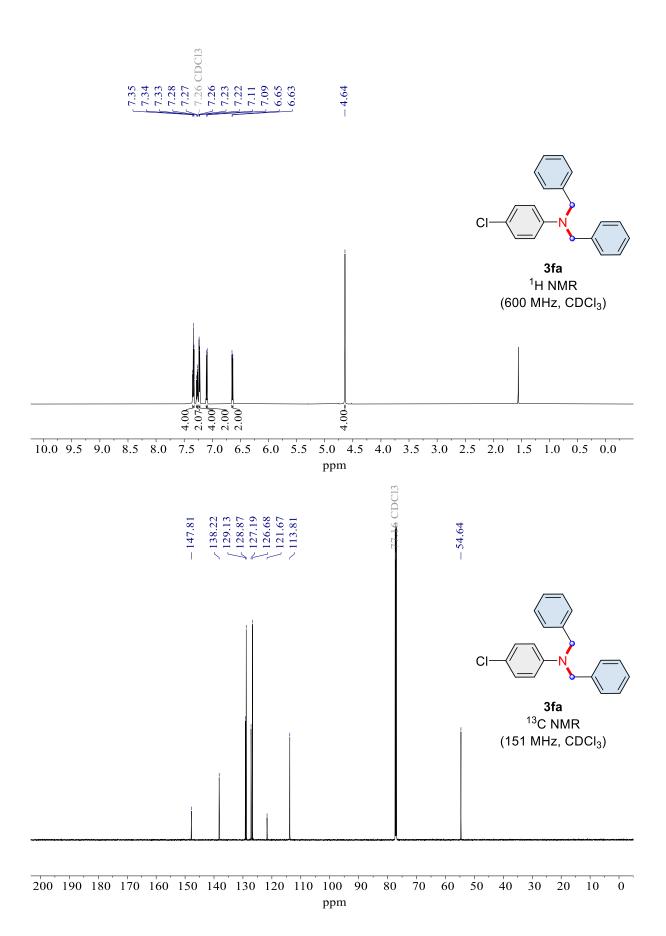


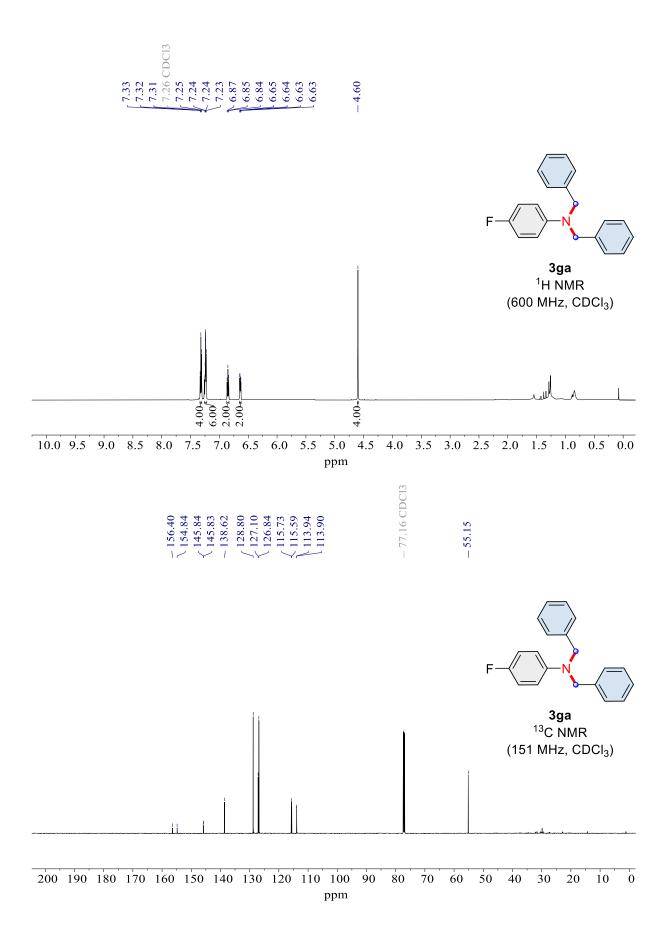




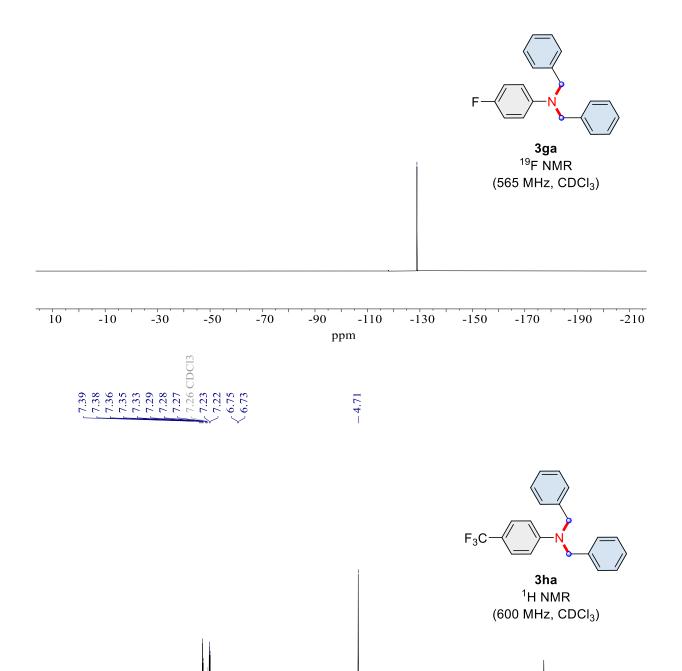










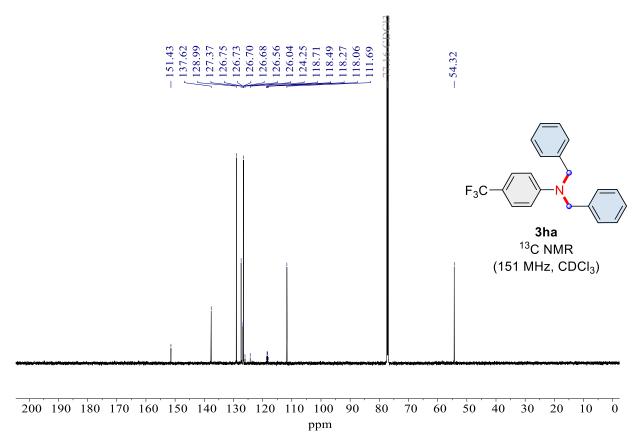


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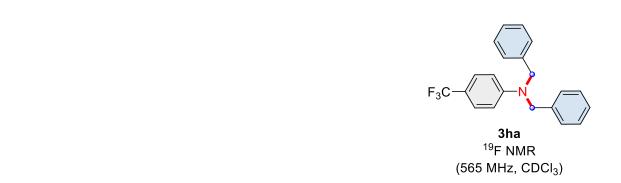
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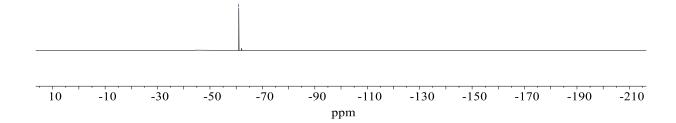
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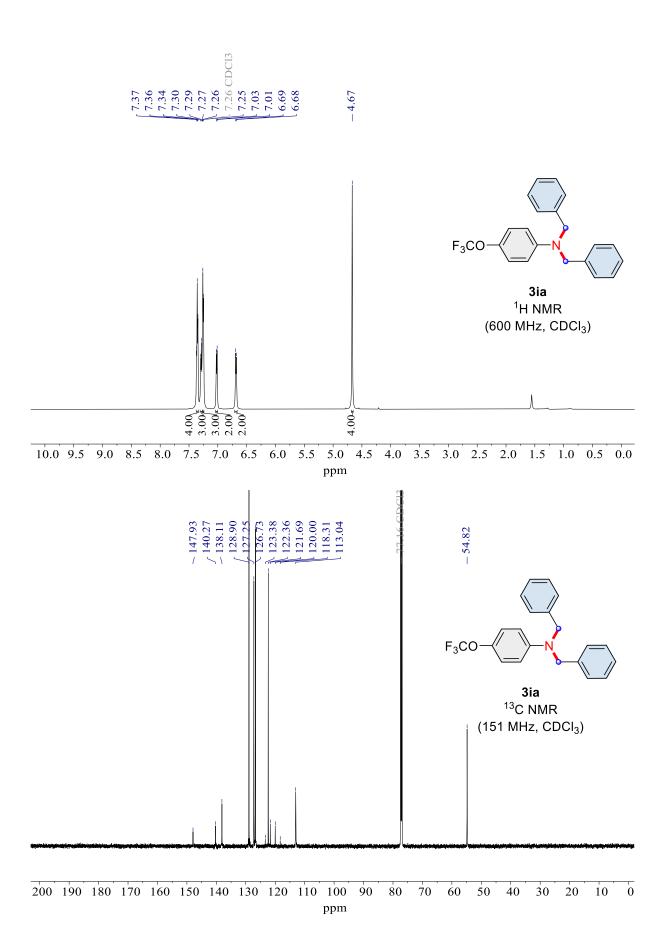
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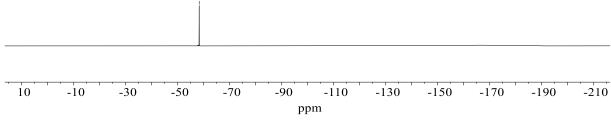


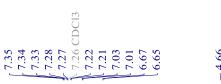


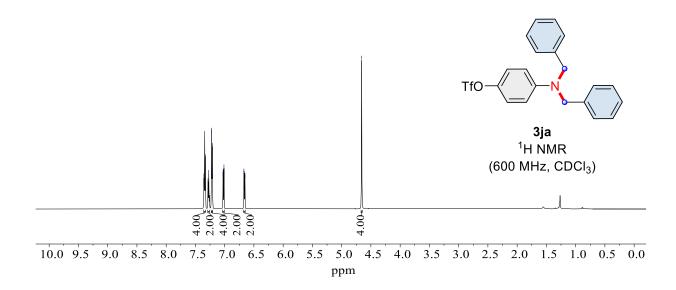


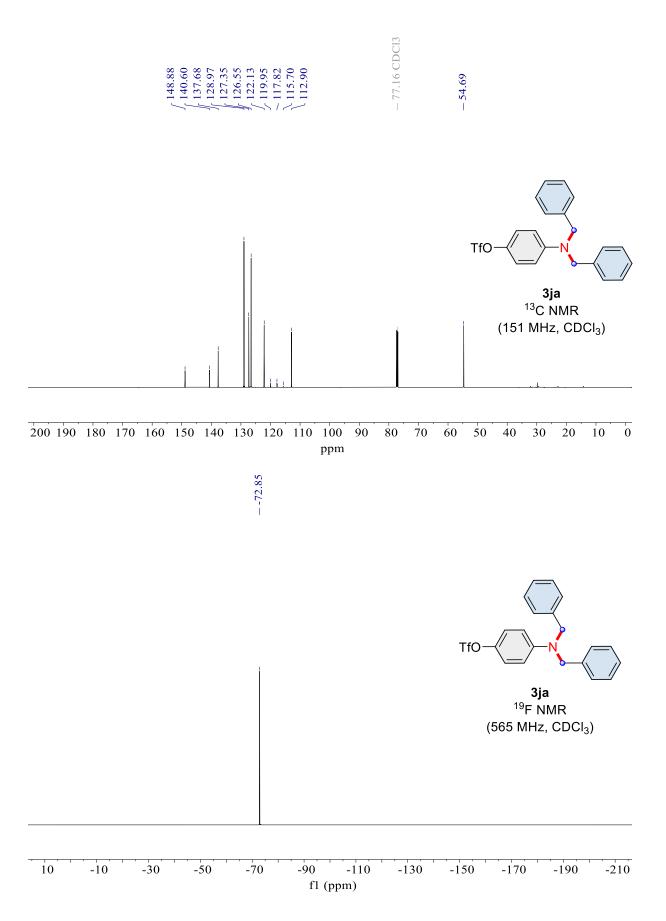


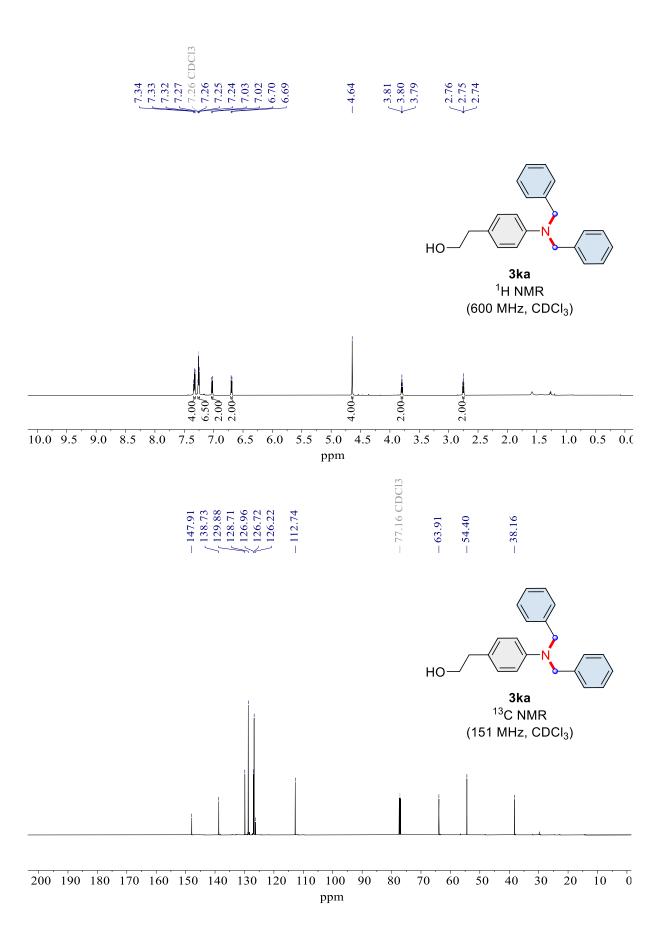
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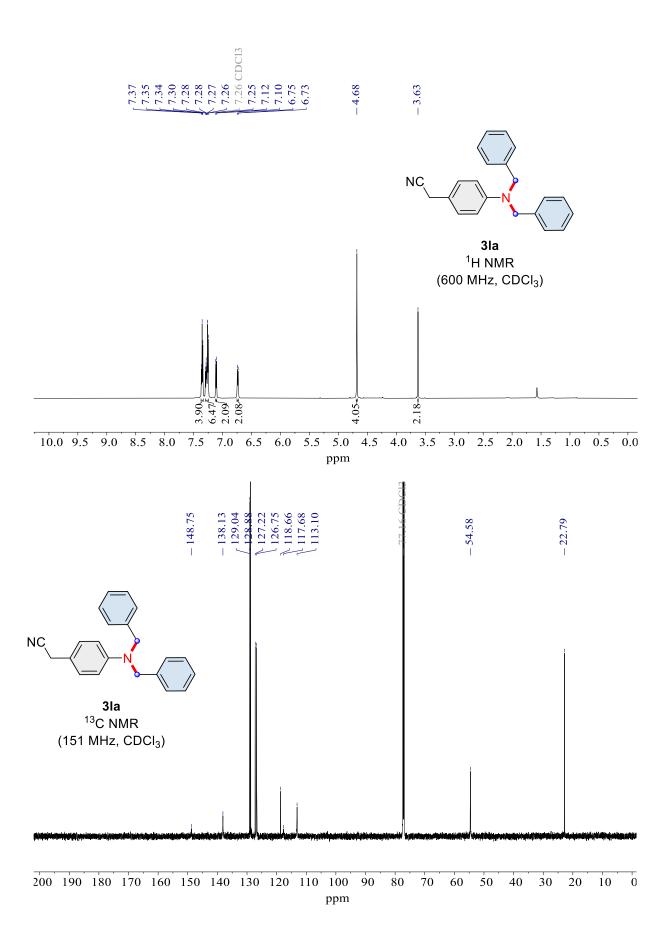


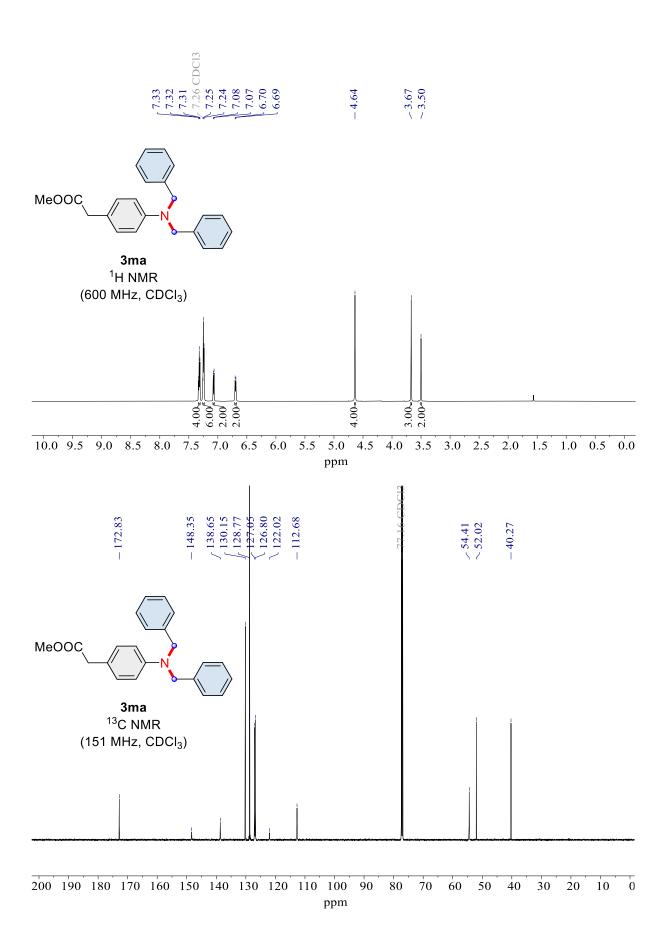


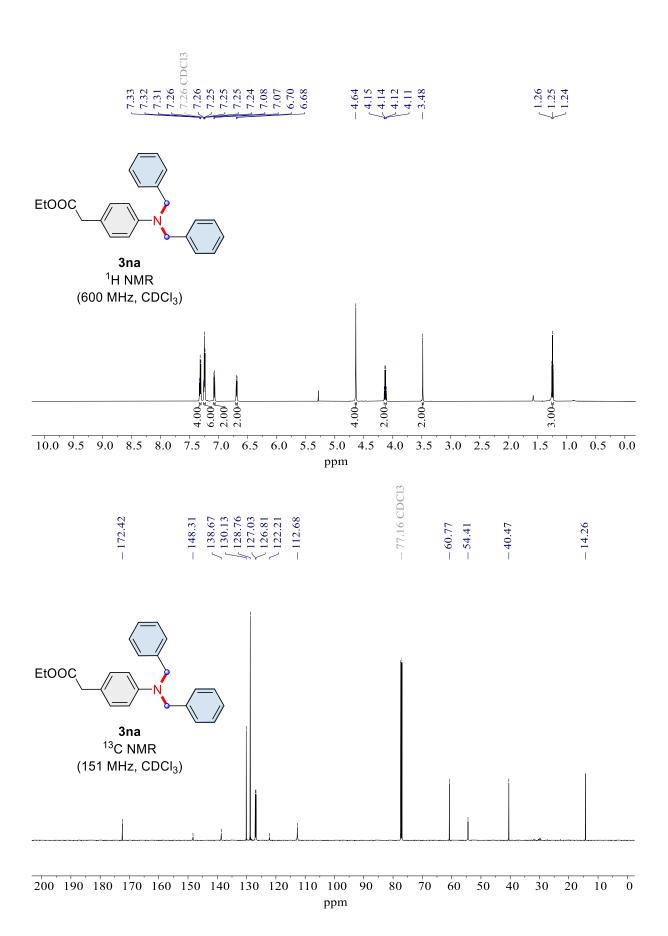




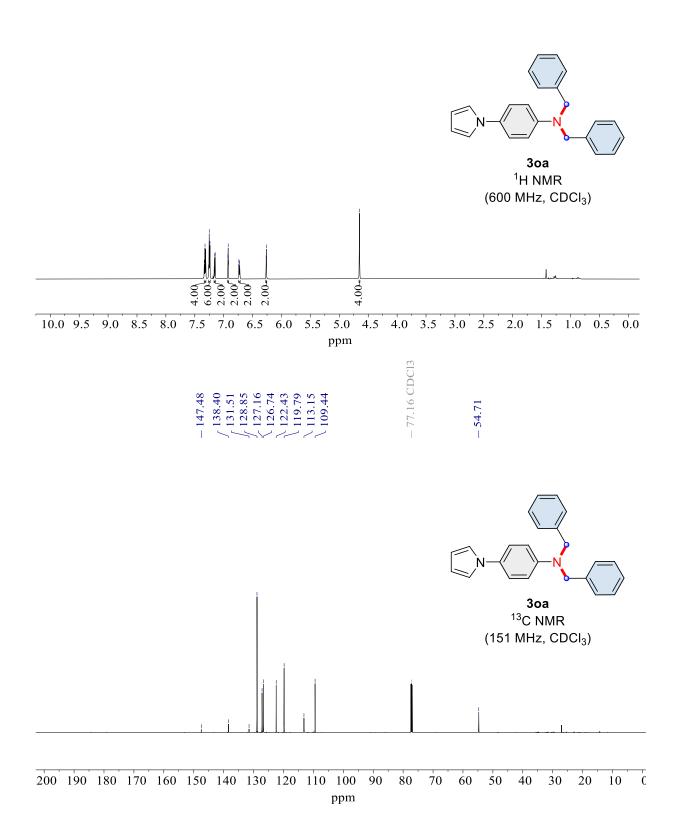


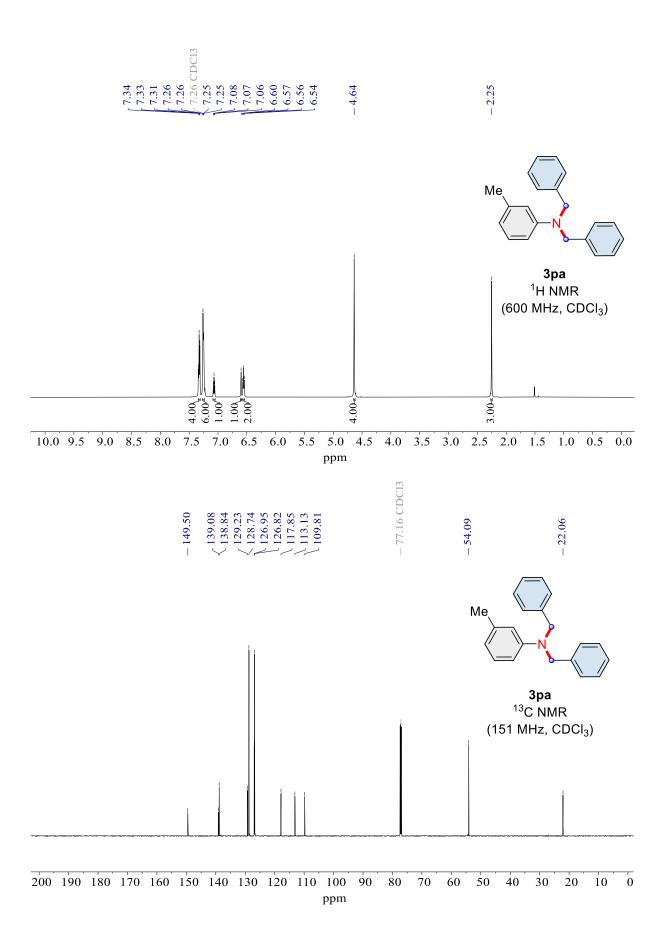


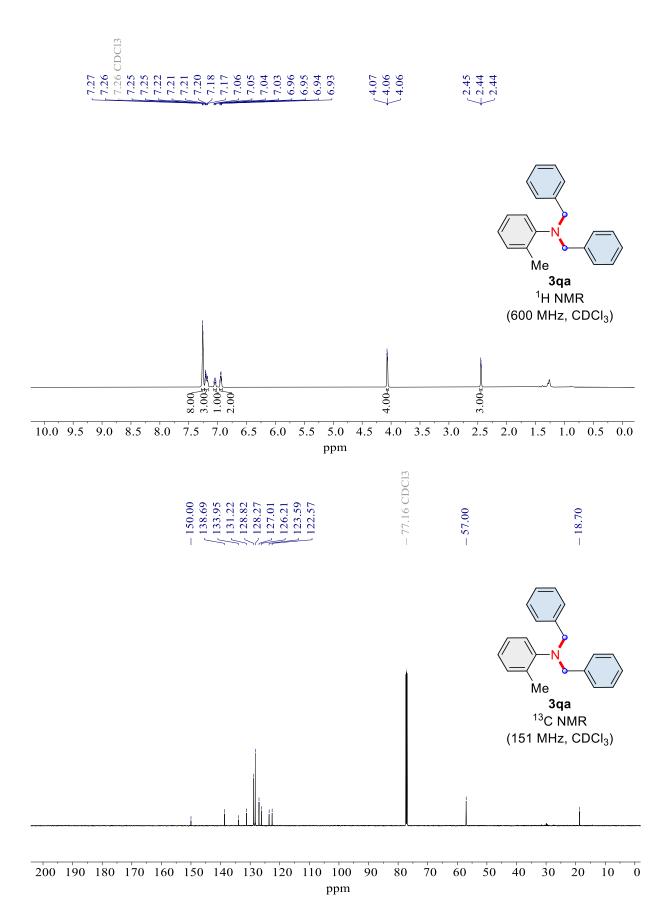


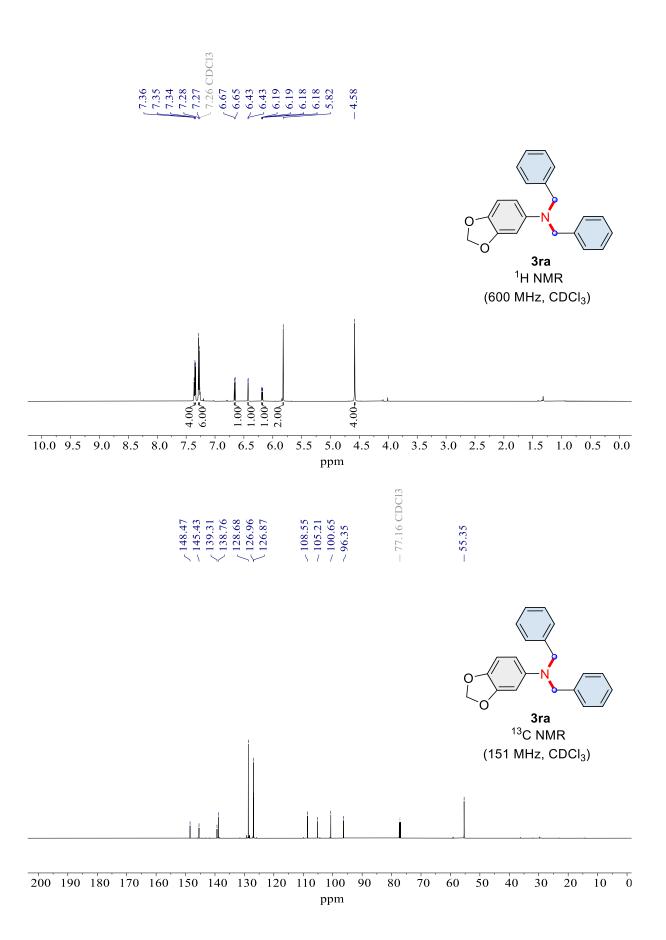




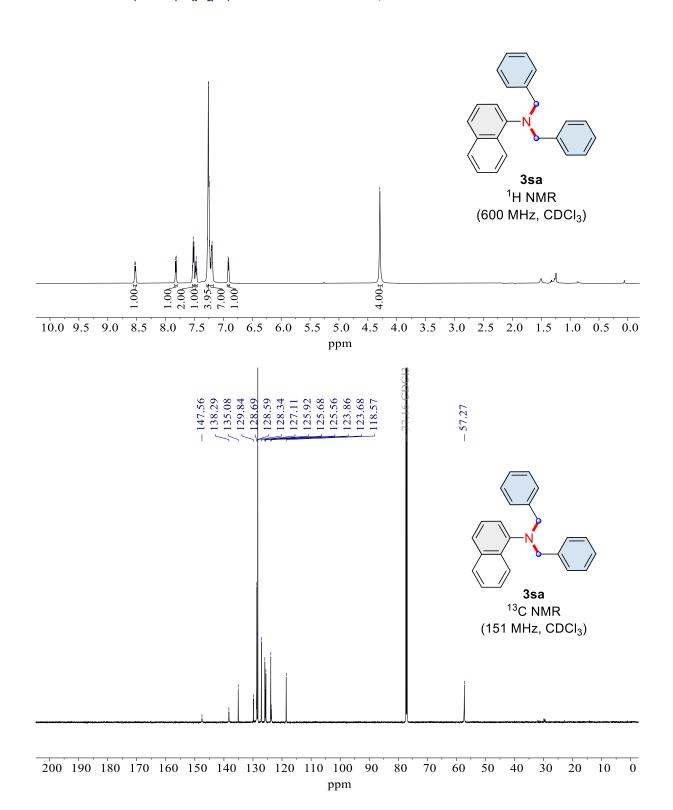


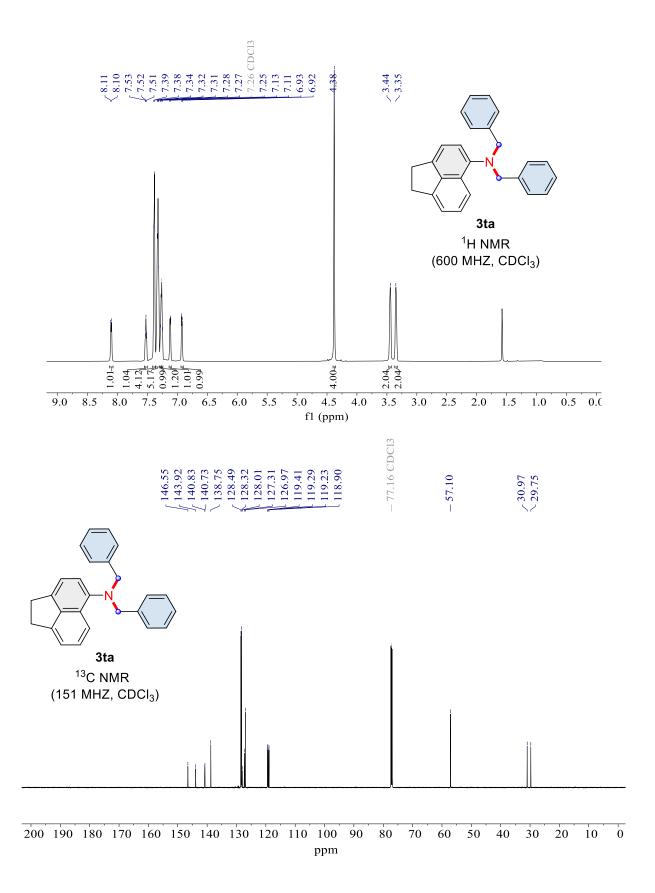


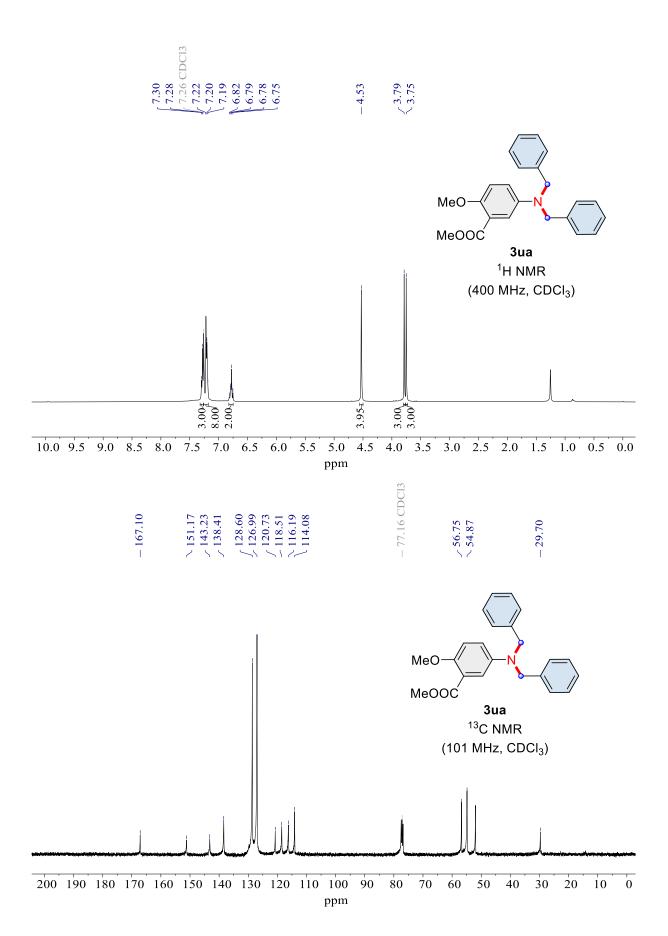


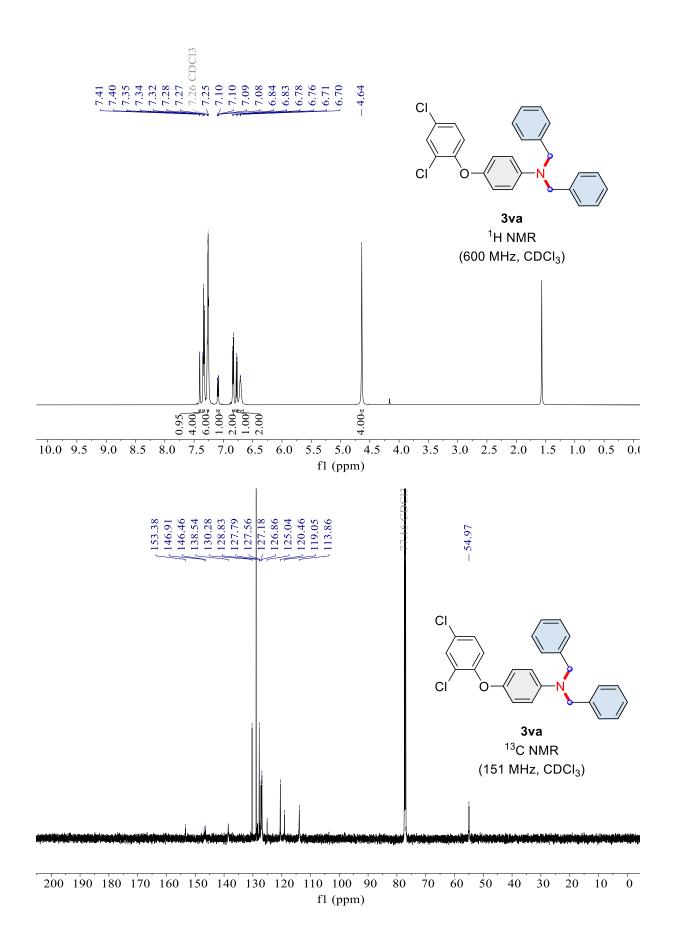


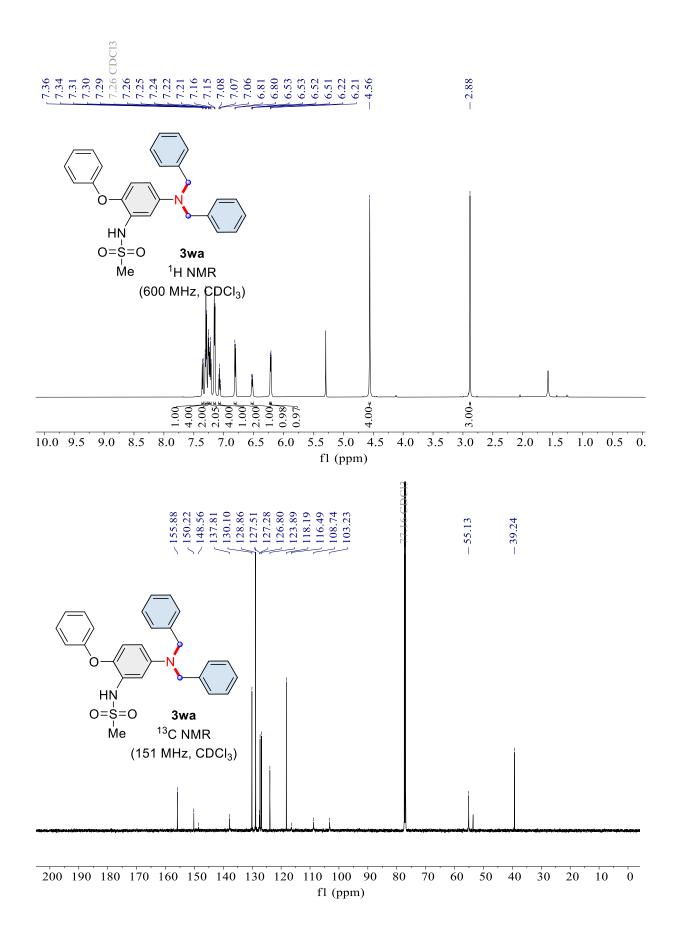


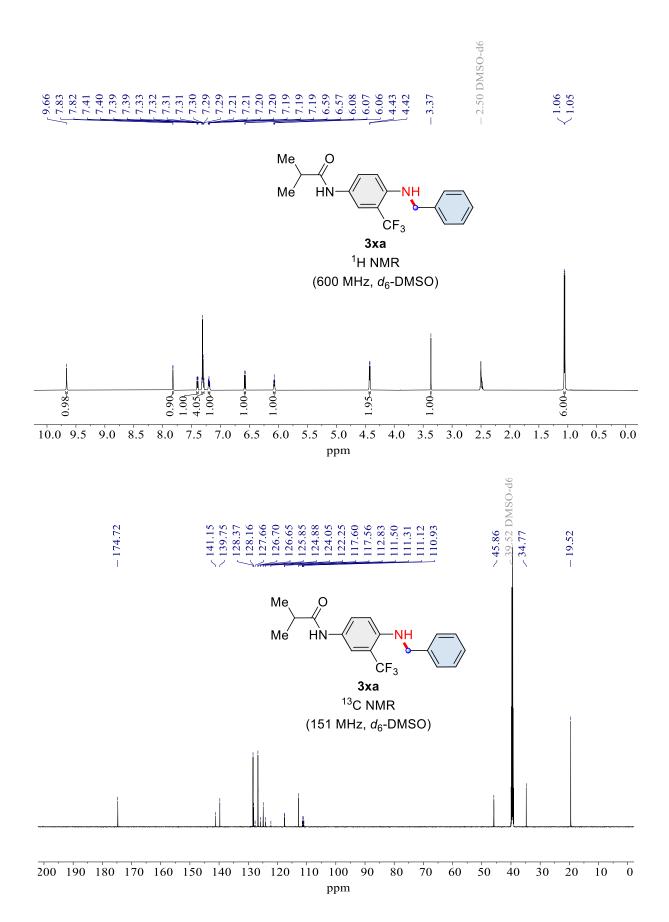




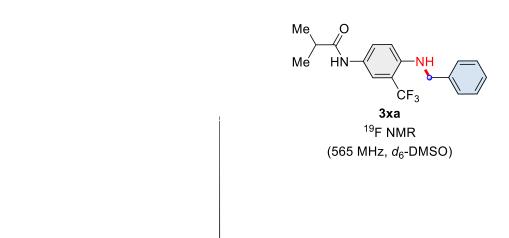


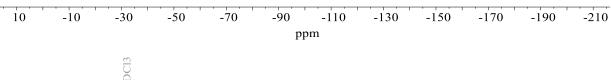




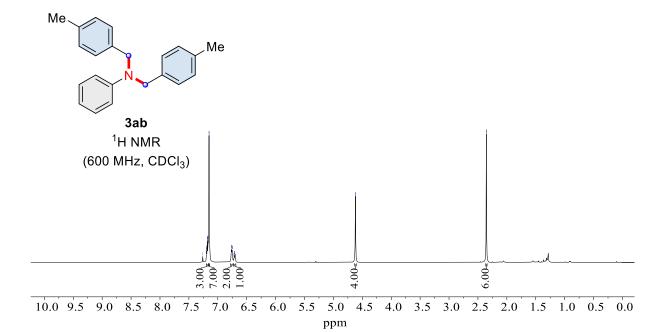


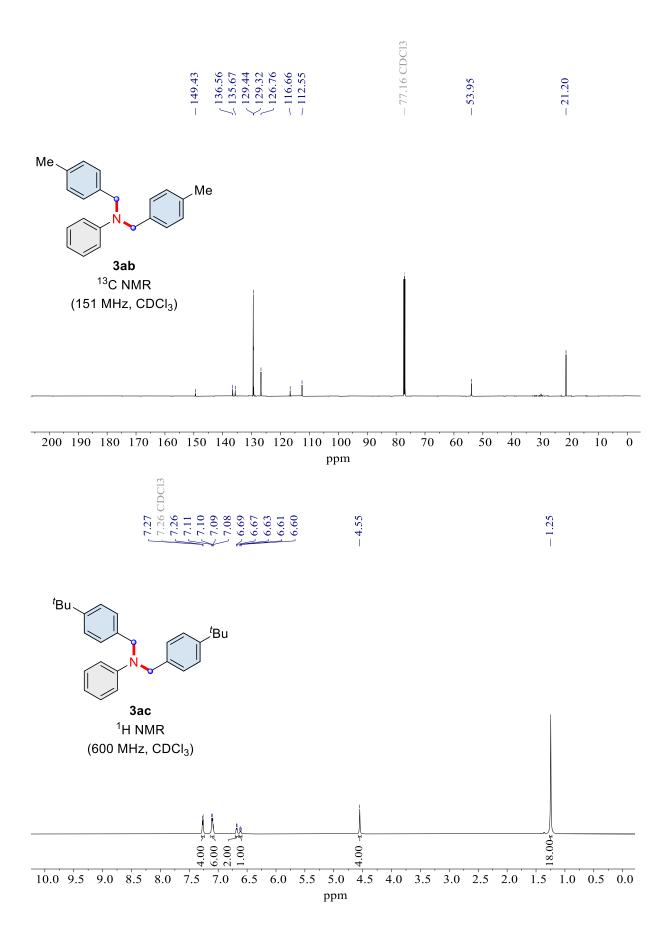


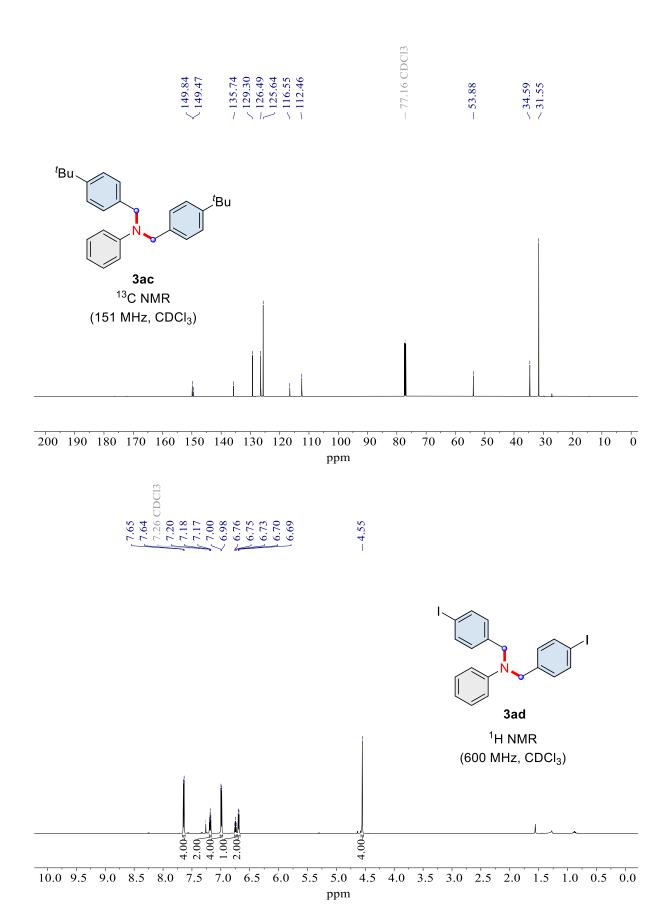


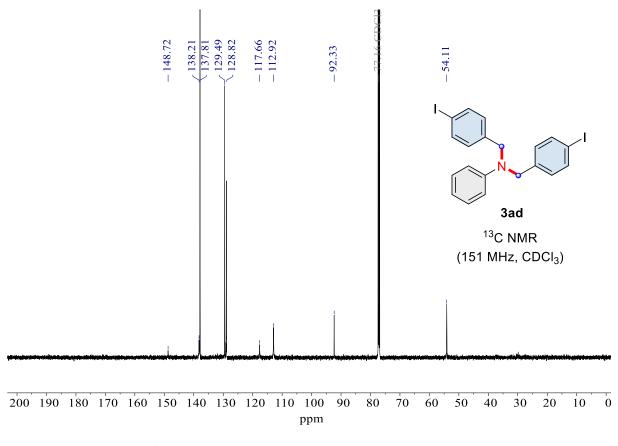


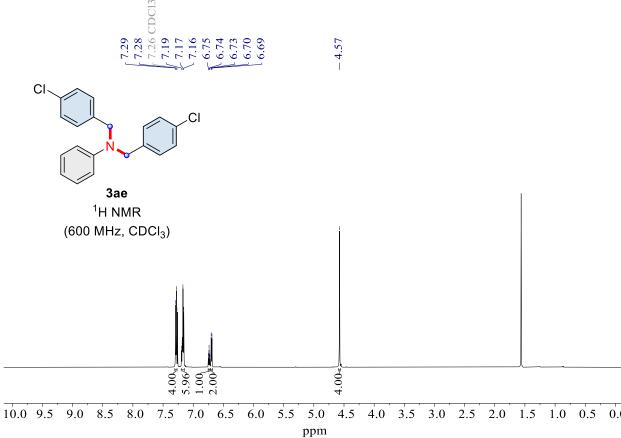


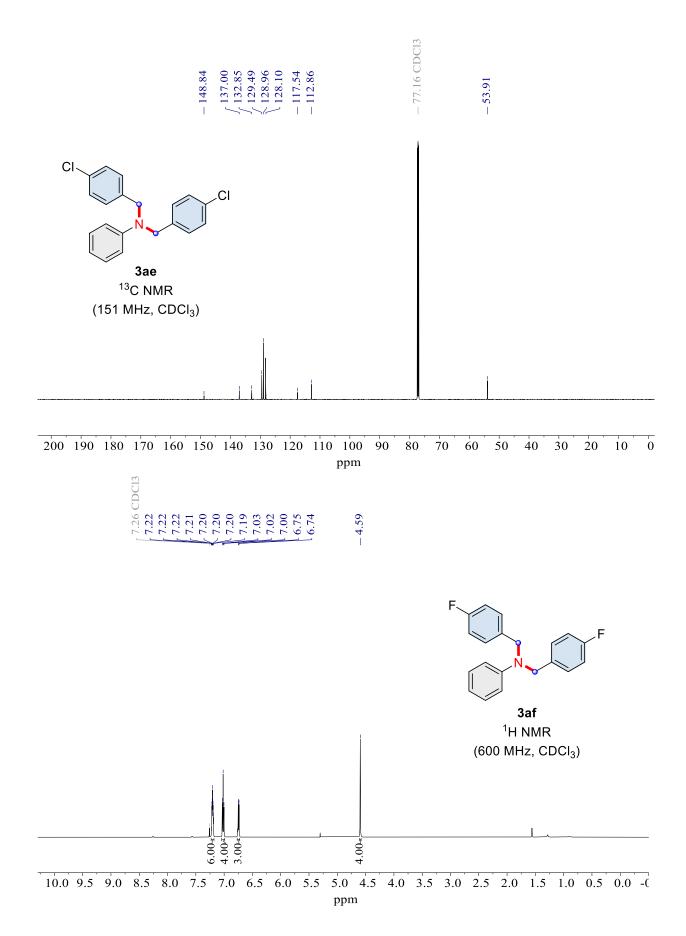


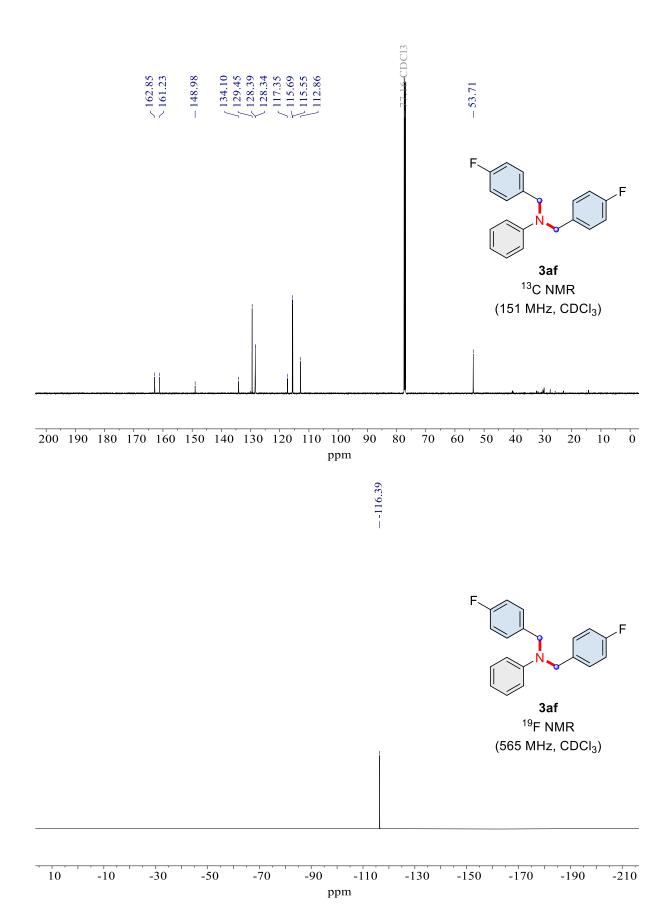




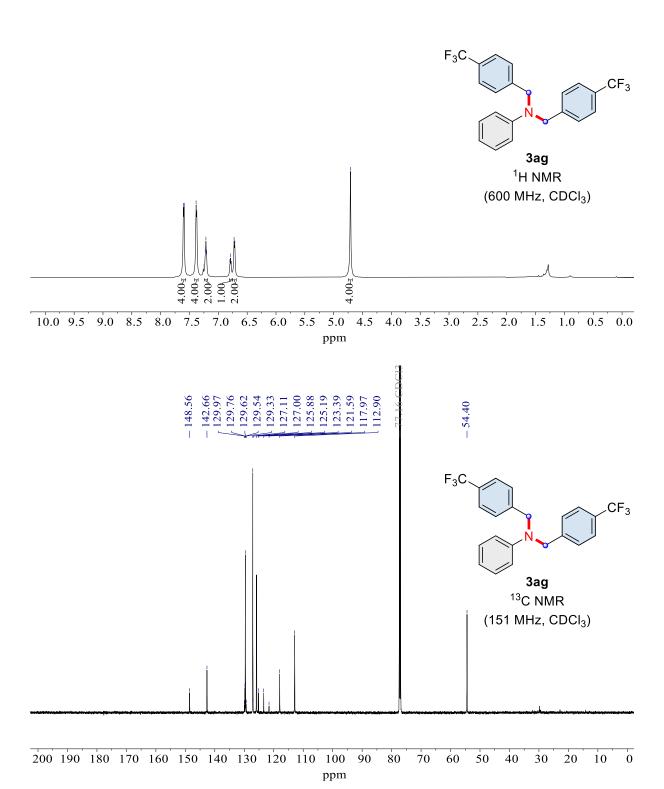




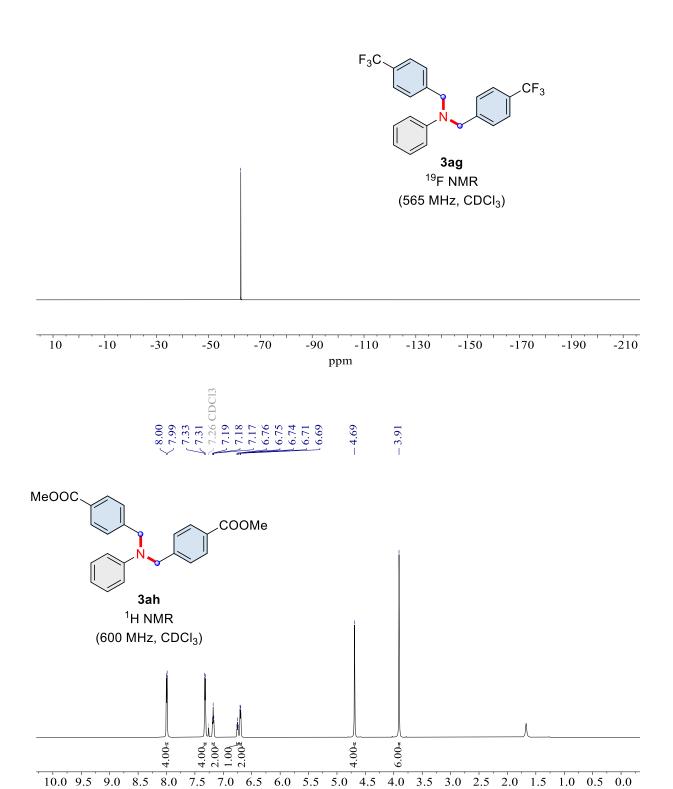












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