

## Supplementary information

### **Microwave-Assisted Hydrothermal Synthesis of Amino Acid-Loaded Cu<sub>2</sub>O Hybrid Particles for CO<sub>2</sub> Reduction Electrocatalysis**

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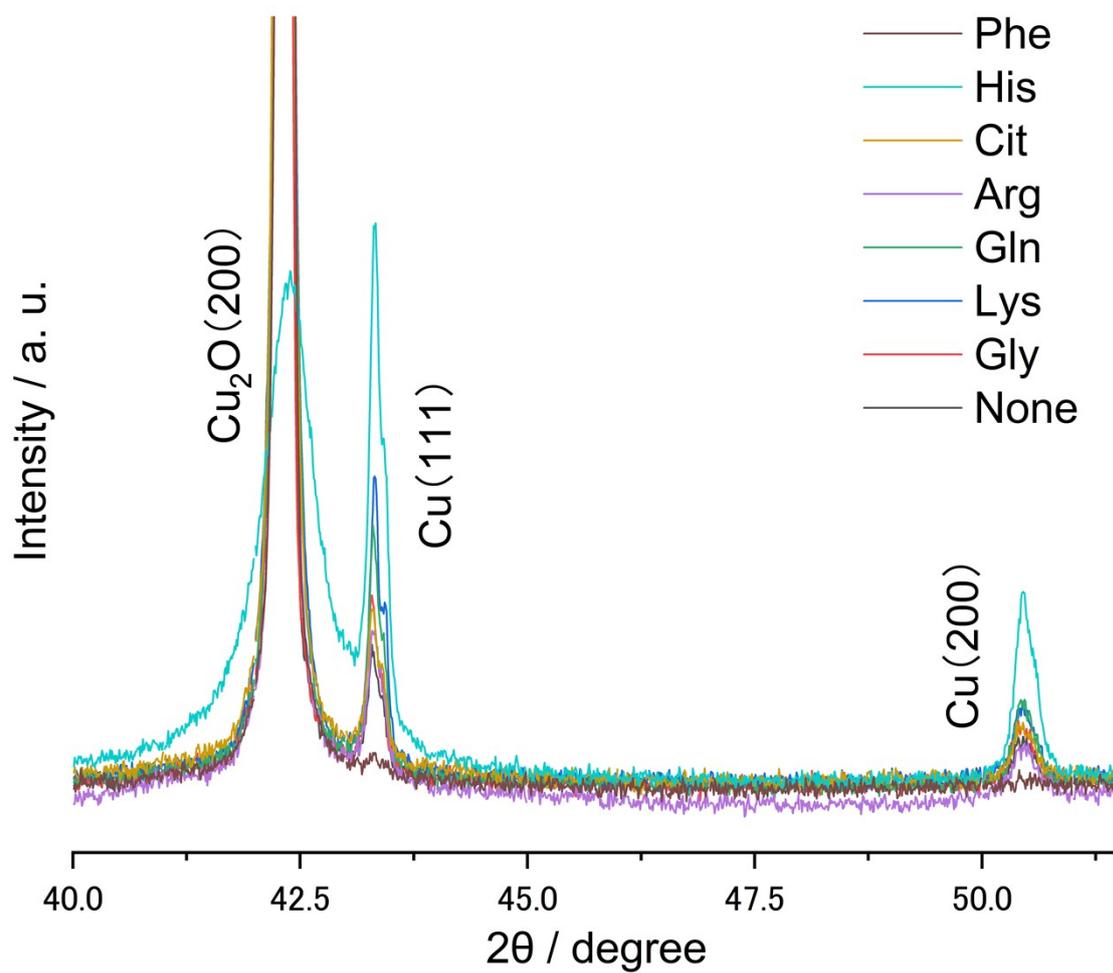


Fig. S1 Zoom-in XRD patterns in a  $2\theta$  range of 40.0–51.5°.

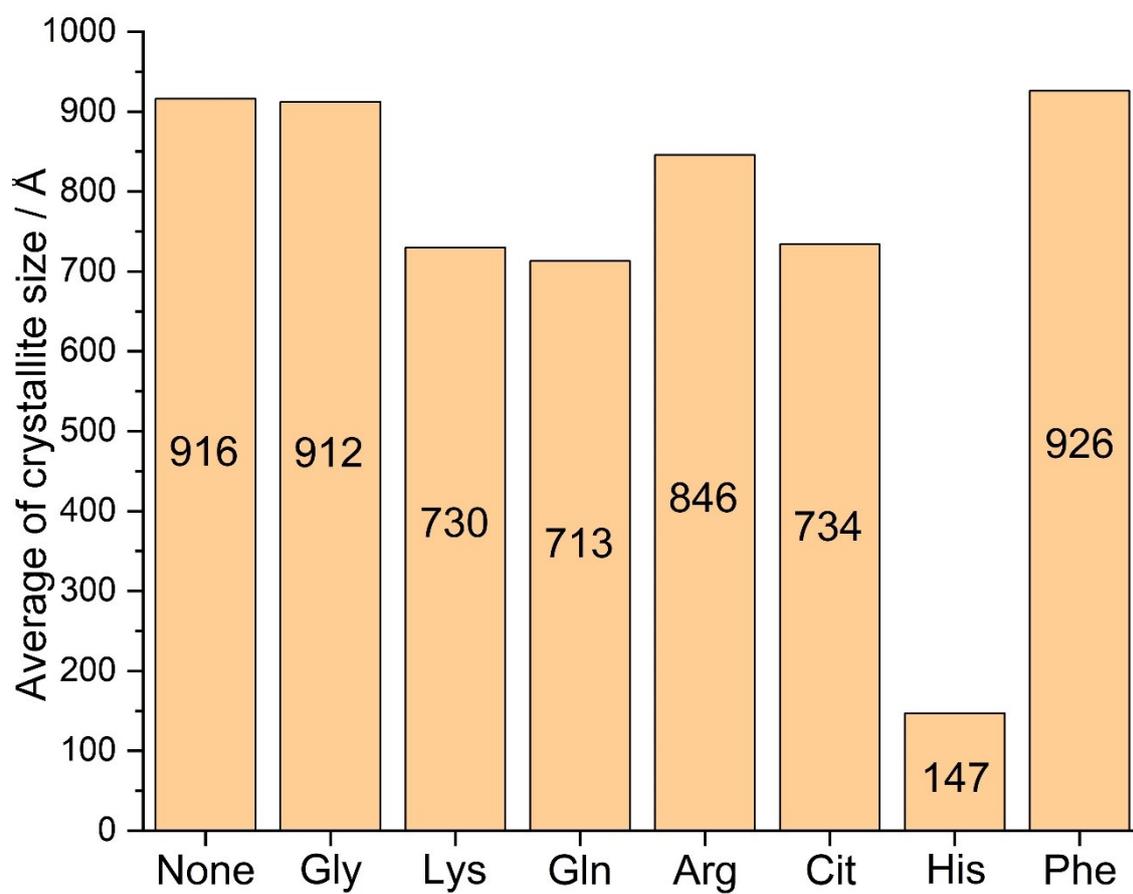


Fig. S2 Average crystallite size calculated by applying Scherrer equation for  $\text{Cu}_2\text{O}$  (111) diffraction peak of the synthesized  $\text{Cu}_2\text{O}$  with and without  $5.0 \text{ mmol dm}^{-3}$  amino acids.

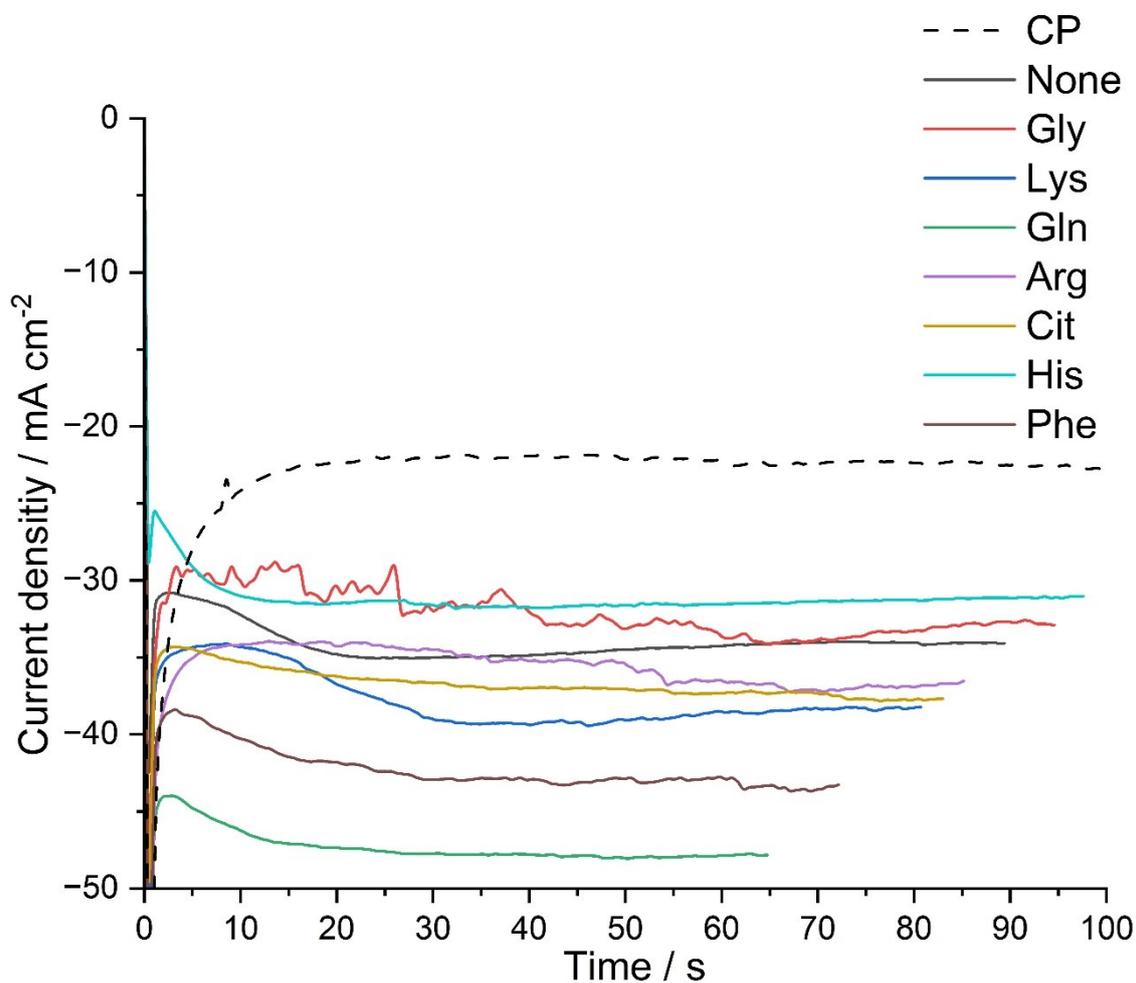


Fig. S3 Chronoamperogramas measured during CO<sub>2</sub> electrolysis under an applied potential of  $-1.27$  V vs. RHE at  $3.0$  C in a CO<sub>2</sub>-purged  $0.5$  mol dm<sup>-3</sup> aqueous KHCO<sub>3</sub> solution (pH  $\approx 8.75$ ) using unloaded and amino acid-loaded Cu<sub>2</sub>O electrodes synthesized with  $5.0$  mmol dm<sup>-3</sup> amino acids and CP with Nafion ionomer.

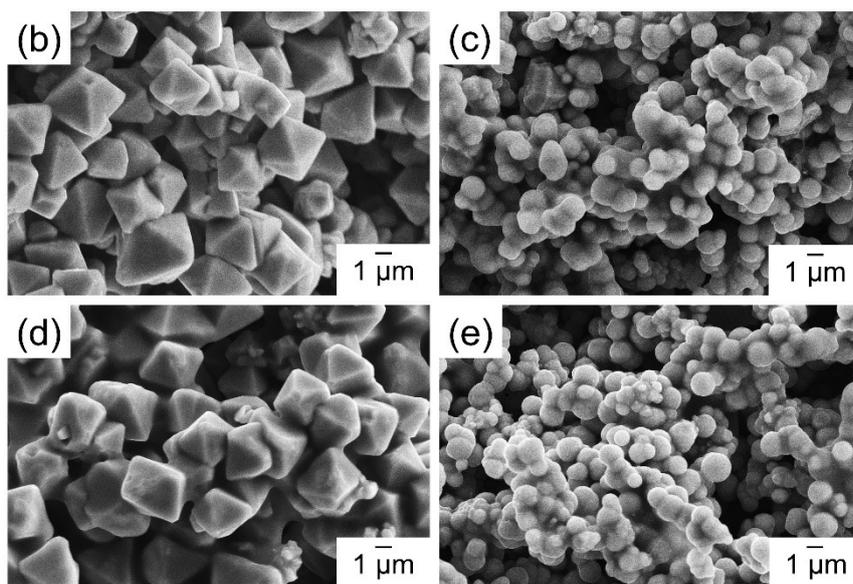
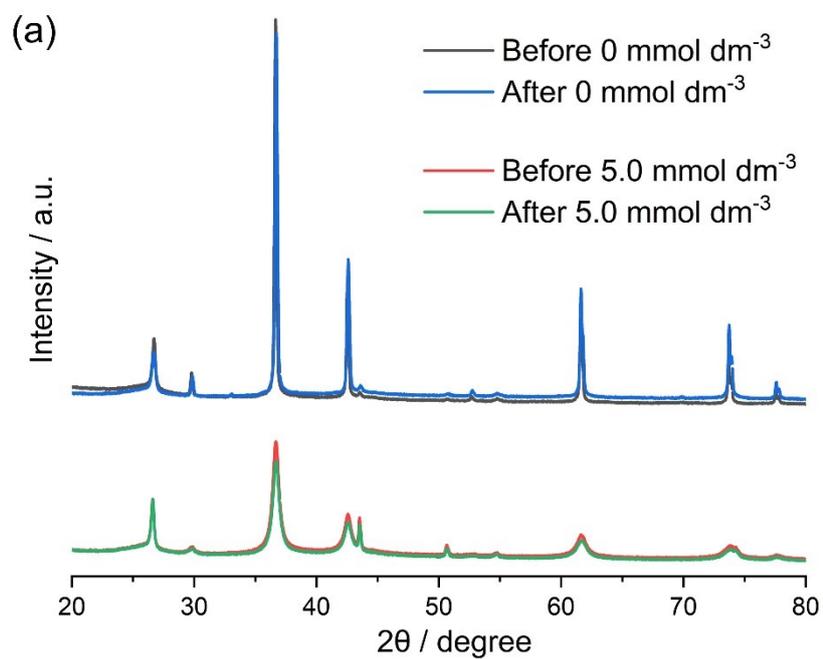


Fig. S4 XRD patterns (a) and SEM pictures of fabricated electrodes applied unloaded (b, d) and His-loaded Cu<sub>2</sub>O particles synthesized  $5.0 \text{ mmol dm}^{-3}$  His (c, e) before (b, c) and after (d, e) CO<sub>2</sub> electrolysis.

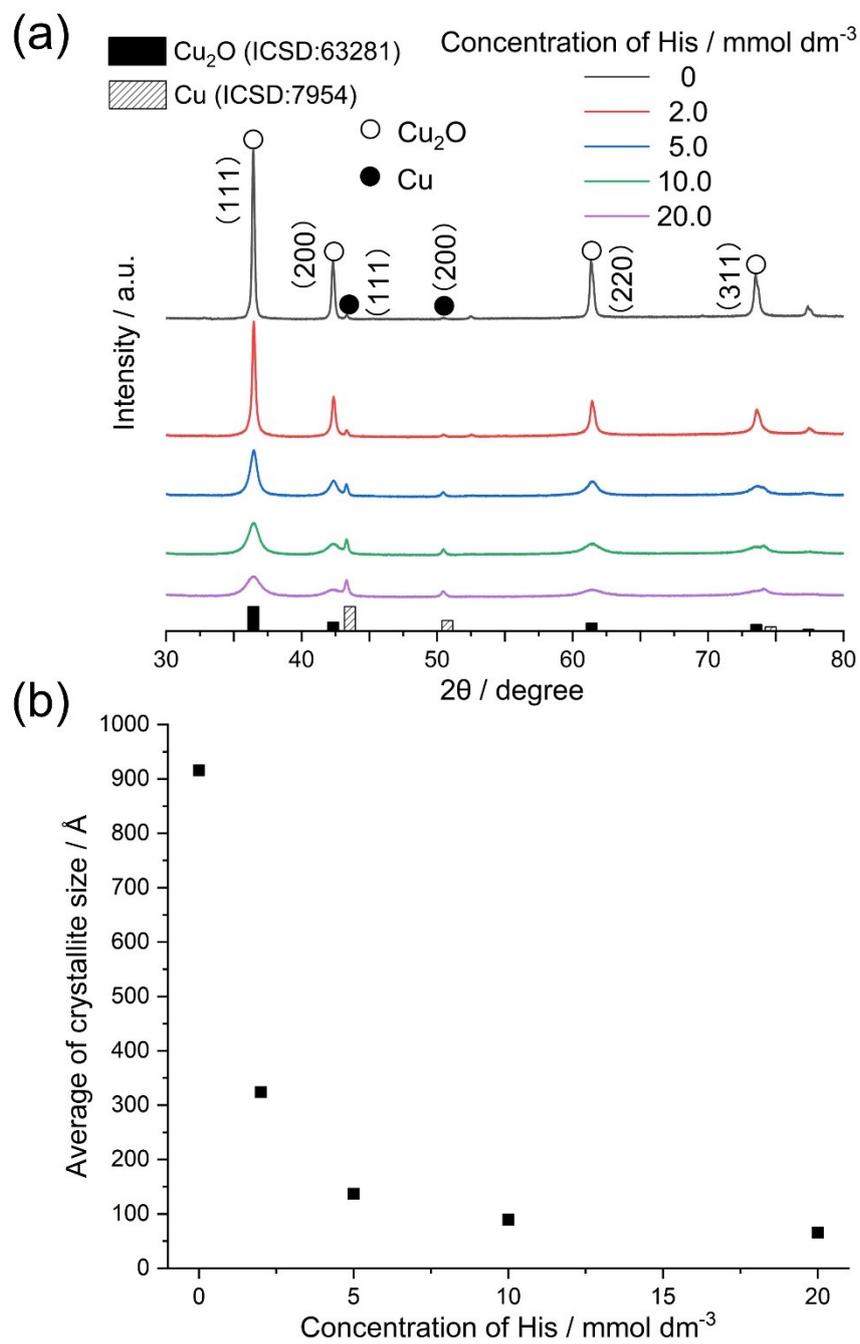


Fig. S5 XRD patterns and average of crystallite size of the His-loaded  $\text{Cu}_2\text{O}$  hybrid particles synthesized with 0, 2.0, 5.0, 10.0 and 20.0  $\text{mmol dm}^{-3}$  of His.



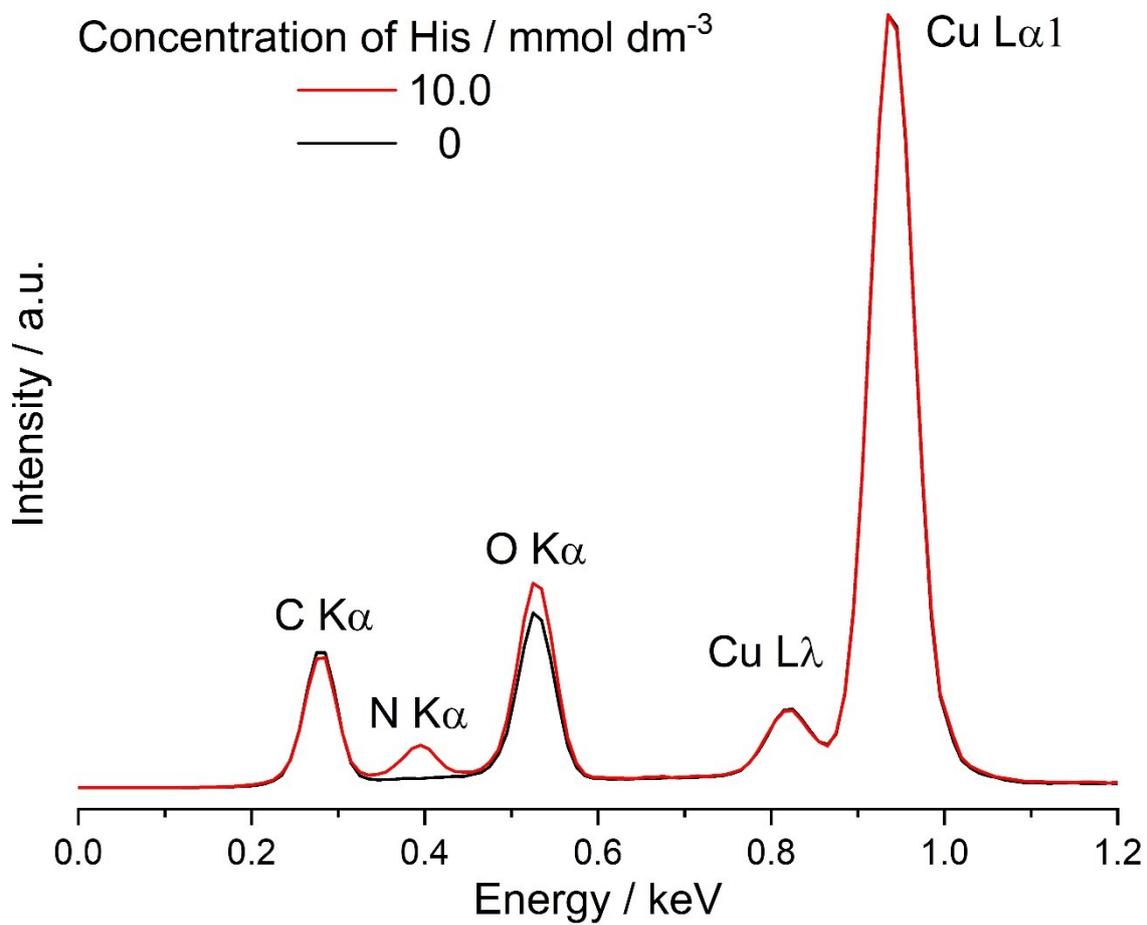


Fig. S6 EDS spectra of His-loaded Cu<sub>2</sub>O hybrid particles synthesized with 0 and 10 mmol dm<sup>-3</sup> L-His. Spectra were normalized to the intensity of the Cu L $\alpha$ 1 peak.

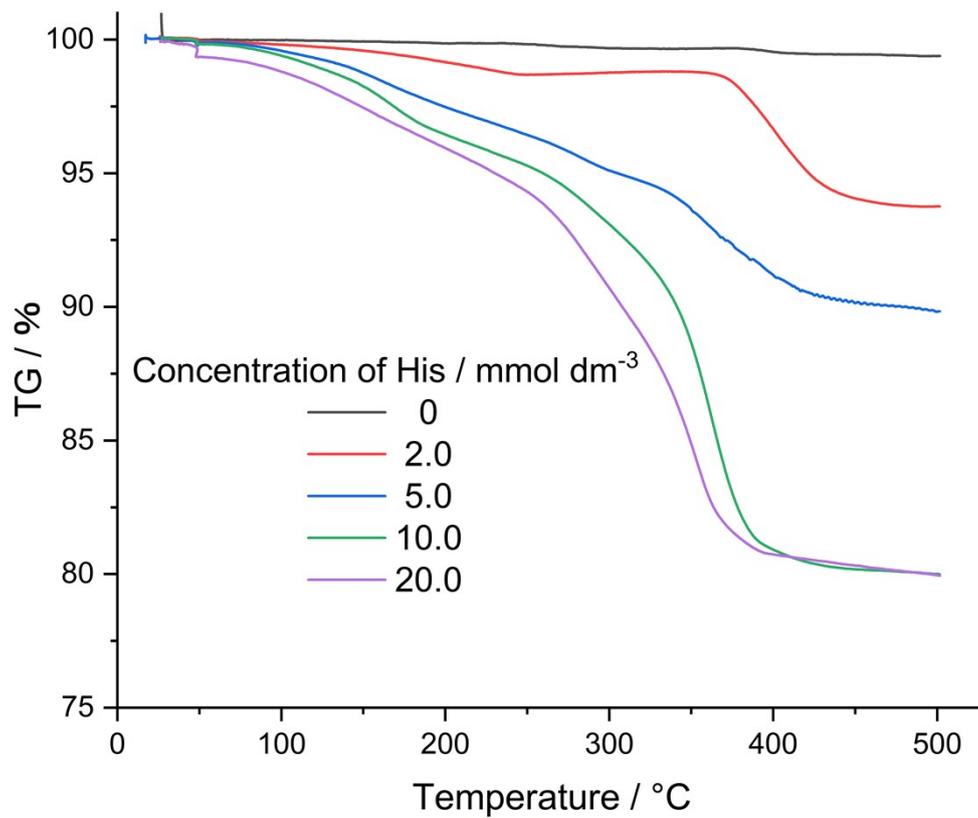


Fig. S7 TG curves of His-loaded Cu<sub>2</sub>O hybrid particles synthesized with 0, 2.0, 5.0, 10.0 and 20.0 mmol dm<sup>-3</sup>.

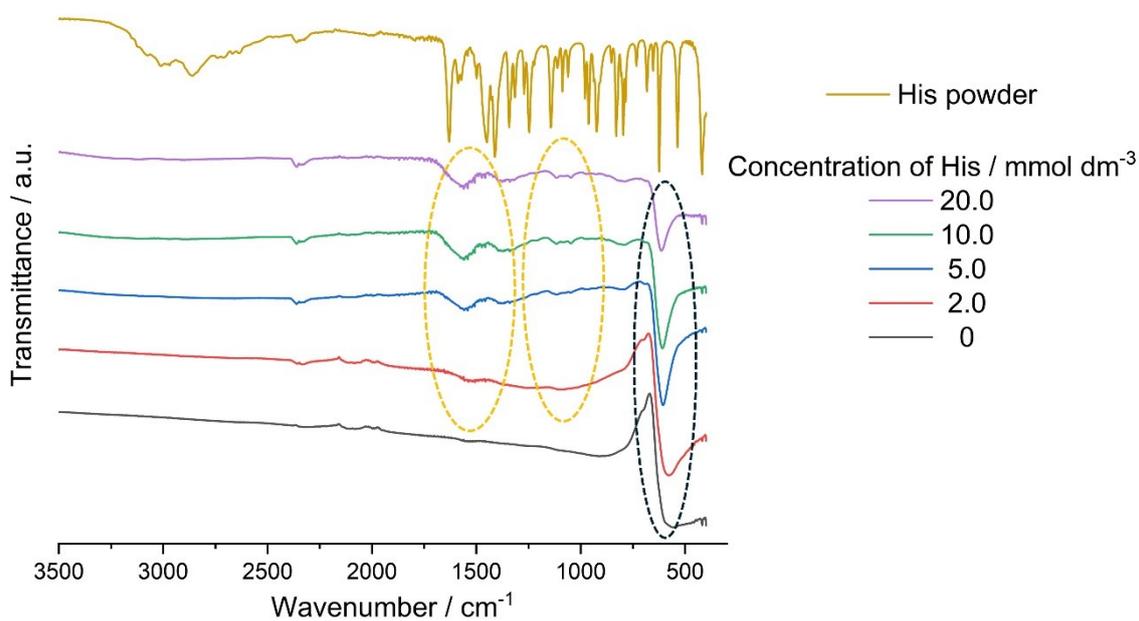


Fig. S8 IR spectra of His-loaded  $\text{Cu}_2\text{O}$  hybrid particles synthesized with 0, 2.0, 5.0, 10.0 and 20.0  $\text{mmol dm}^{-3}$  of His and commercializing His powder. Black dashed circle means originated to  $\text{Cu}_2\text{O}$  peaks and yellow ones are peaks appeared by loading His.

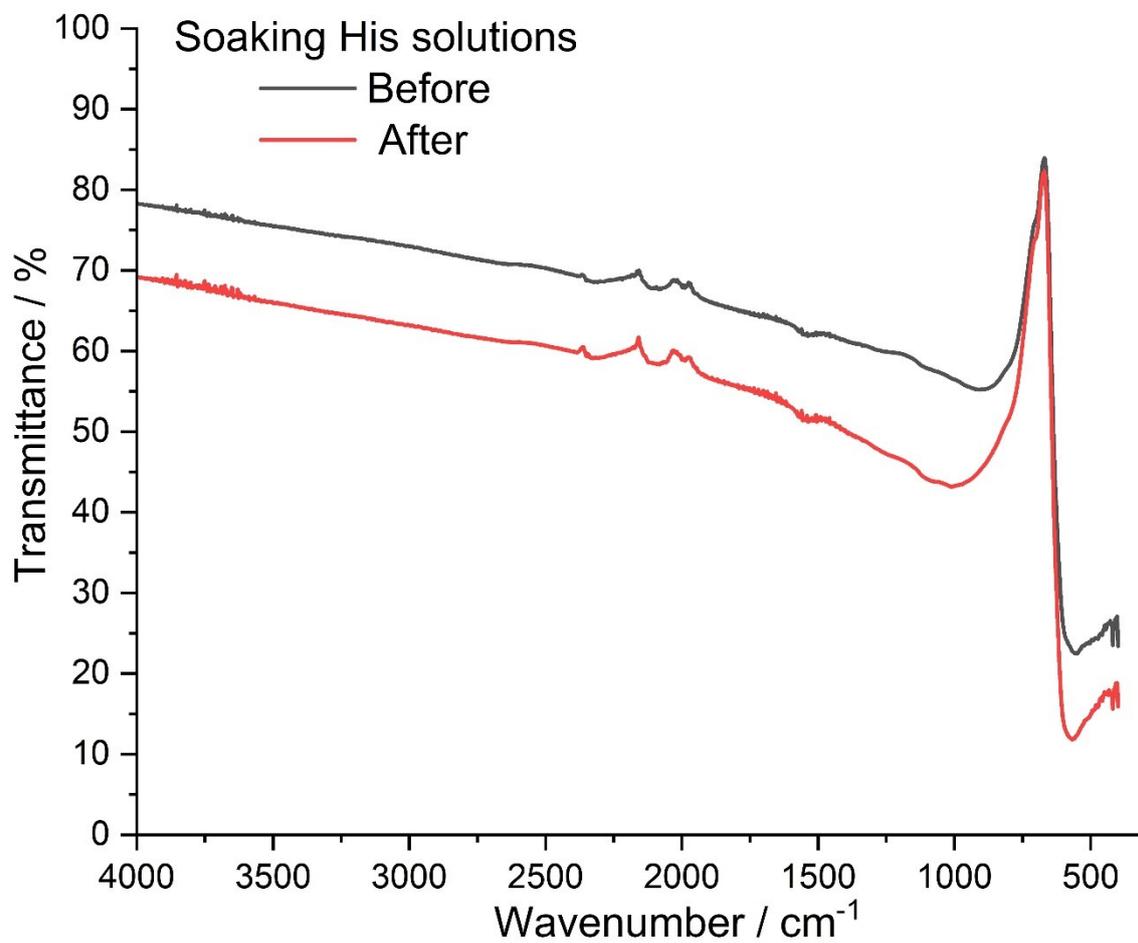


Fig. S9 FT-IR spectra of synthesized unloaded  $\text{Cu}_2\text{O}$  before and after soaking 60 min in the  $10.0 \text{ mmol dm}^{-3}$  His aqueous solutions.

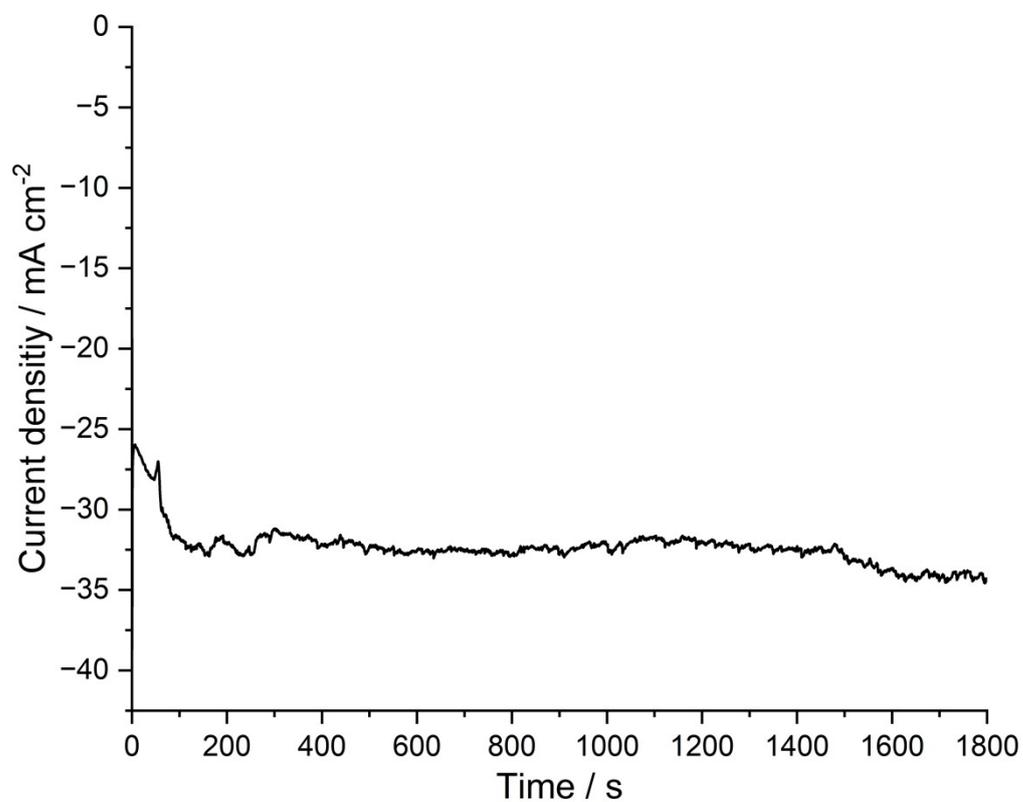


Fig. S10 Chronoamperogram during CO<sub>2</sub> electrolysis for 1800 seconds in a CO<sub>2</sub>-purged 0.5 mol dm<sup>-3</sup> aqueous KHCO<sub>3</sub> solution (pH ≈ 8.75) using His-loaded Cu<sub>2</sub>O electrocatalysis synthesized with 5.0 mmol dm<sup>-3</sup> His.