## Efficient visible-light-driven photocatalytic detoxification of a sulfur

## mustard simulant in air using Rose Bengal-functionalized MOFs

Jinfeng Zhou<sup>\*a,b,c</sup>, Xiaodong Zhang<sup>\*a,b</sup>, Kesheng Cao<sup>a,b</sup>, Qing Zhou<sup>d</sup>, Jinping Cao<sup>a</sup>, Renpeng Guan<sup>a</sup>, Chunjie Chu<sup>a</sup>

<sup>a</sup> College of Chemistry and Environmental Engineering, Pingdingshan University,

Pingdingshan 467000, P. R. China

<sup>b</sup> Yaoshan Laboratory, Pingdingshan 467000, P. R. China

<sup>c</sup> Henan Province Engineering Technology Research Center of Green Hydrogen & Electrochemical Energy Storage

<sup>d</sup> Fujian Engineering Research Center of Advanced Manufacturing Technology for
Fine Chemicals, College of Chemical Engineering, Fuzhou University, Fuzhou
350116, P. R. China

Corresponding author: zhoujf016@163.com



Fig. S1 The structure of RB.



Fig. S2 UV-Vis absorption spectrum of RB solution in  $H_2O$ .



Fig. S3 TGA curves of RB, ZIF-8 and RB@ZIF-8 composite.



Fig. S4 The zeta potential of ZIF-8 and the RB@ZIF-8 composite.



Fig. S5 The color of the mixture of ZIF-8+RB.





Fig. S7 The size distributions of (a) ZIF-8 and (b) RB@ZIF-8 composite.



Fig. S8 EPR signals of the TEMP- $^{1}O_{2}$  under different conditions.



Fig. S9 Solid-state UV-Vis diffused reflectance spectrum of the RB@ZIF-8 composite after catalysis.



Fig. S10 Conversion efficiency for the detoxification of CEES in the absence or presence of  ${}^{1}O_{2}$  scavenger (carotene) upon light irradiation (carotene: 1.0 mmol; RB@ZIF-8 composite: 20.0 mg; CEES: 20 µL; methanol: 1.5 mL; reaction time: 10 min).



Fig. S11 Conversion efficiency for the detoxification of CEES in the presence of RB@ZIF-8 and RB+ZIF-8 upon light irradiation (RB@ZIF-8 composite: 20.0 mg; RB+ZIF-8 composite: 20.0 mg; CEES: 20  $\mu$ L; methanol: 1.5 mL; reaction time: 6 min).



Fig. S12 Conversion efficiency for the detoxification of CEES in the presence of RB@ZIF-8 upon light irradiation (RB@ZIF-8 composite: 20.0 mg, 5 mg and 1 mg; CEES: 20  $\mu$ L; methanol: 1.5 mL; reaction time: 6 min).

Scheme S1. Proposed mechanism for the photochemical detoxification of CEES by the RB@ZIF-8 composite.

**Table S1** Comparison of the performances in the detoxification of of CEES usingvarious MOFs-based heterogeneous photosensitizers.

Catalyst <sup>a</sup>	Solvent	Effective Photosensitizer	Atmo. <sup>b</sup>	Light	Half-	
					life	Ref.
		Thotosensitizer			(min)	

TCPP@MIL-101(Cr)	МеОН	Porphyrin	O <sub>2</sub>	Blue LED	1	1
Au/TCPP@MIL-101(Cr)	MeOH	Porphyrin	O <sub>2</sub>	Blue LED	0.75	1
Ag <sub>12</sub> TPyP	CD <sub>3</sub> OD	Porphyrin	O <sub>2</sub>	White LED	1.5	2
Br-BDP@NU-1000	MeOH	BODIPY	O <sub>2</sub>	Green LED	2	3
Br-BDP@NU-1000	MeOH	BODIPY	O <sub>2</sub>	Green LED	2.5	3
UiO-68-TBTD	MeOH	TBTD	air	Blue LED	3	4
PCN-67-Se	MeOH	Benzoselenadia zole	O <sub>2</sub>	Purple LED	3.5	5
NU-1000-PCBA	MeOH	Pyrene and C <sub>60</sub>	O <sub>2</sub>	UV LED	3.5	6
UMCM-313	MeOH	Pyrene	O <sub>2</sub>	Blue LED	4	7
Al-PMOF on fiber	MeOH	Porphyrin	O <sub>2</sub>	Blue LED	4	8
I <sub>2</sub> -BODIPY@ZIF-8	МеОН	BODIPY	O <sub>2</sub>	Green LED	4.5	9
MOF/BA/textile	No solvent	Porphyrin	O <sub>2</sub>	Blue LED	4.4	10
MOF/BA/textile	No solvent	Porphyrin	air	Simulated sunlight	17.6	10
Ag <sub>12</sub> TPyP	CD <sub>3</sub> OD	Porphyrin	air	White LED	6	11
NU-1000	MeOH	Pyrene	O <sub>2</sub>	UV LED	6	12
PCN-222/MOF-545	MeOH	Porphyrin	O <sub>2</sub>	Blue LED	11	12
PCN-57-S	MeOH	Benzothiadiazo le	O <sub>2</sub>	UV LED	7.5	13
PCN-222/MOF	MeOH	Porphyrin	O <sub>2</sub>	Blue (325)	13	14
PCN-222/MOF	MeOH	Porphyrin	O <sub>2</sub>	White LED	26	14
PCN-222/MOF	МеОН	Porphyrin	O <sub>2</sub>	Red LED	33	14
ZnTTP@ZIF-8	МеОН	Porphyrin	air	Blue LED	1.5	15
MB@UiO-66-(COOH)2	МеОН	Methylene blue	air	Red LED	1.8	16
RB@ZIF-8	MeOH	Rose Bengal	air	Green LED	2.5	This work

<sup>*a*</sup>The catalysts from literature have been named as published; <sup>*b*</sup>Atmo.: atmosphere

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