

Supplement File

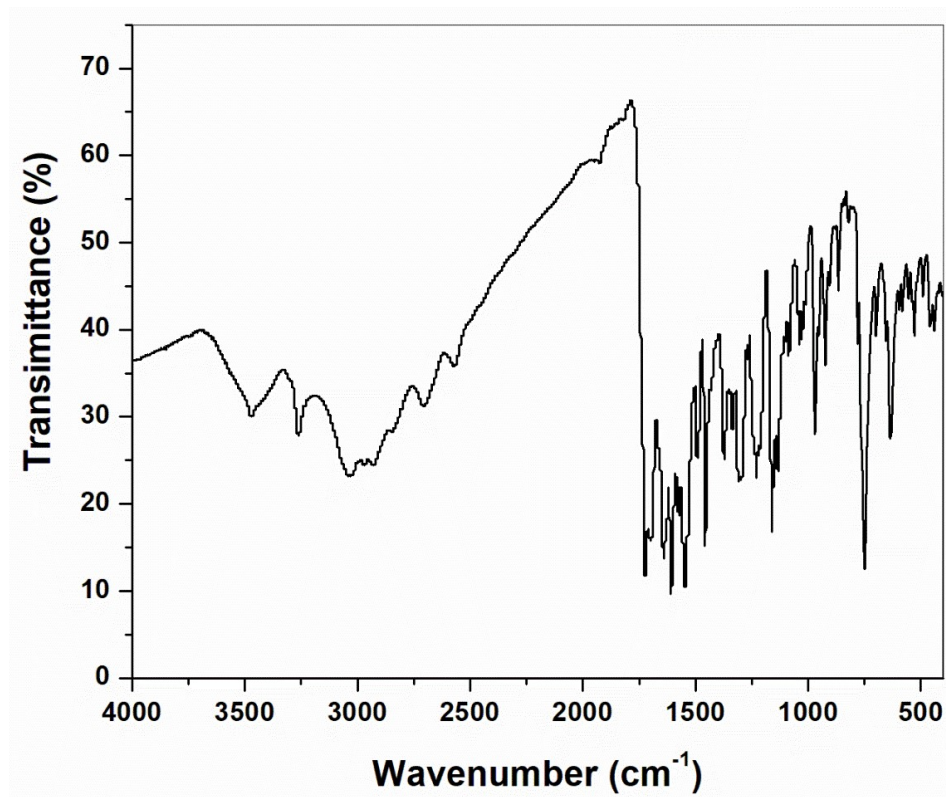


Figure S1. FT-IR spectrum of **HCBH**.

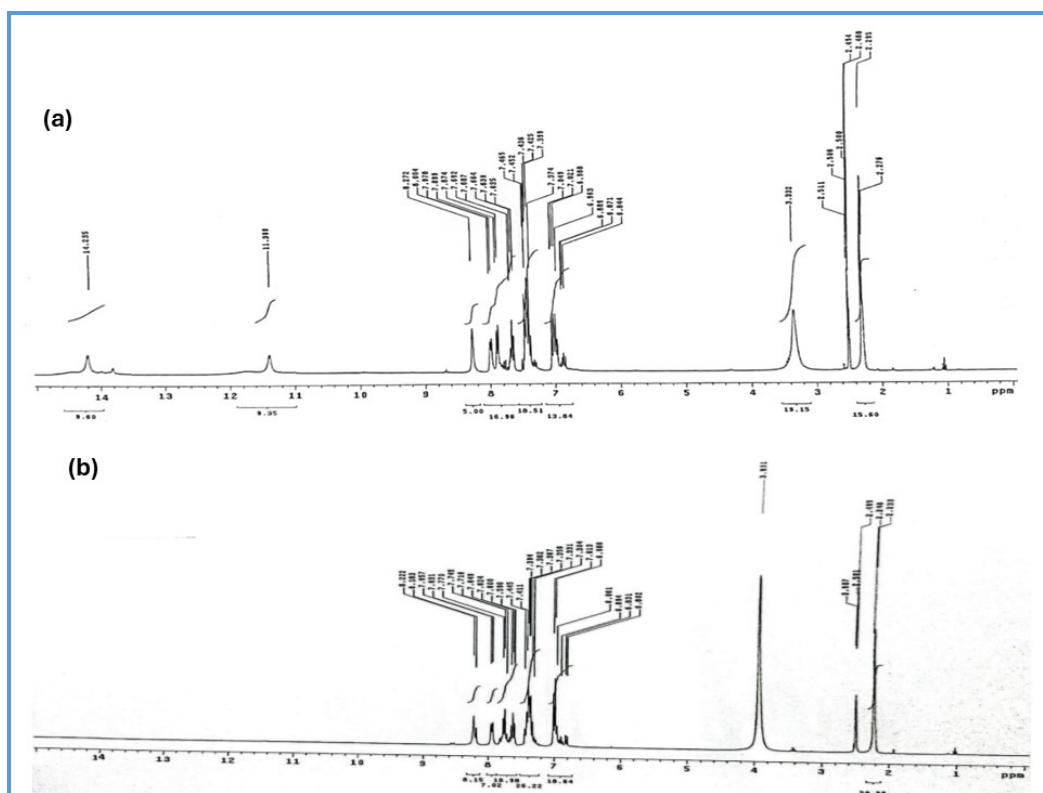


Figure S2. ^1H NMR spectrum of **HCBH** (a) with DMSO (b) with DMSO/ D_2O .

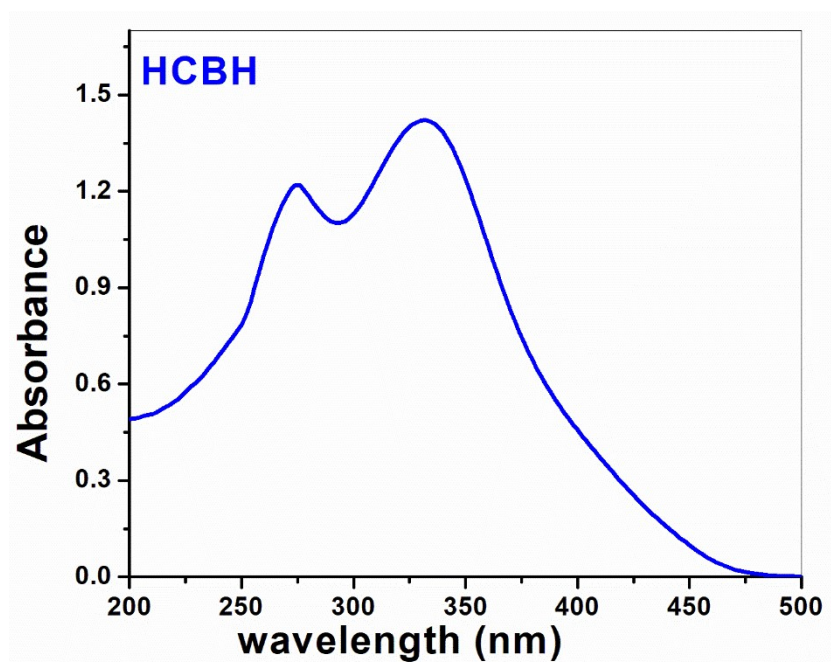


Figure S3. The electronic spectrum of **HCBH**

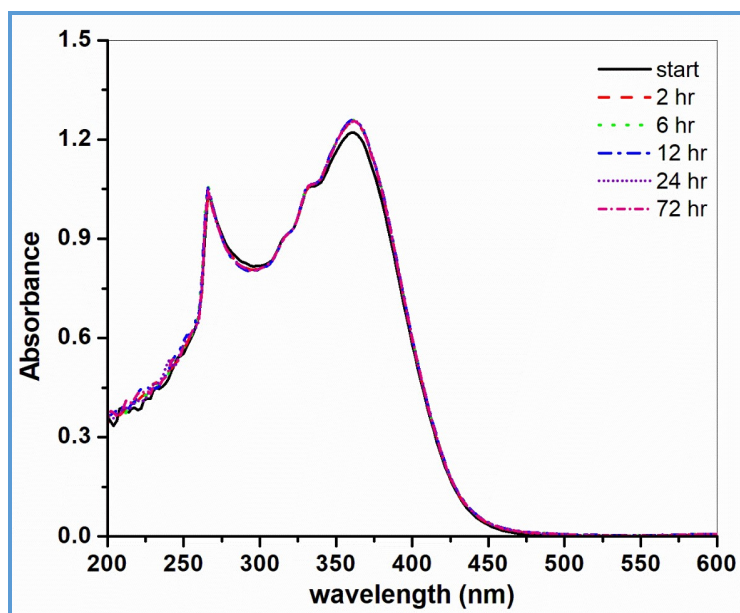


Figure S4. Representative time-dependent UV-vis spectra of **CuH** nano-chelate at 25 °C.

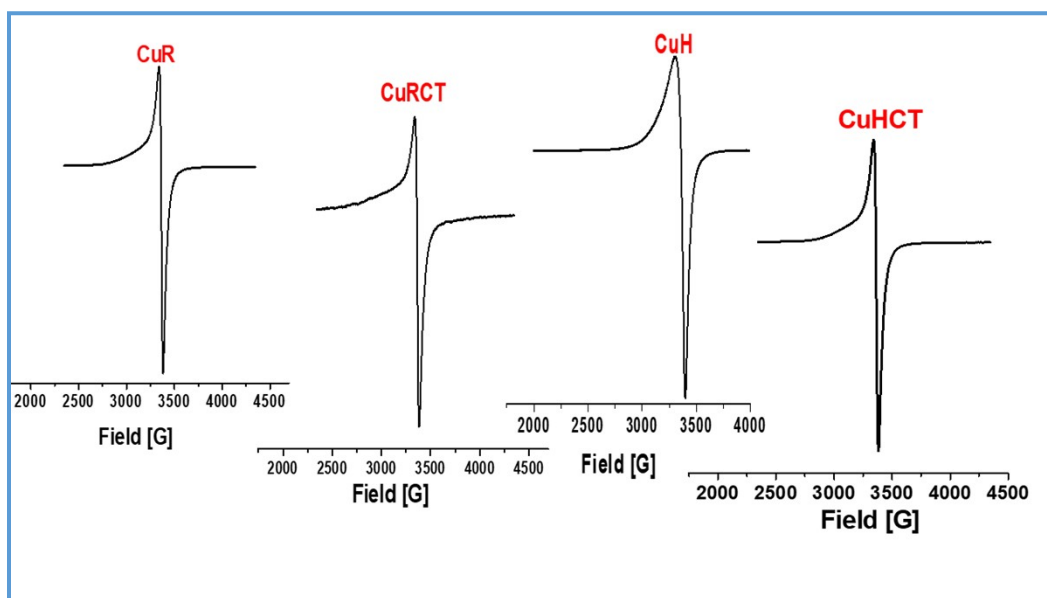


Figure S5. ESR spectra of **CuR**, **CuRCT**, **CuH** and **CuHCT** nano-chelates.

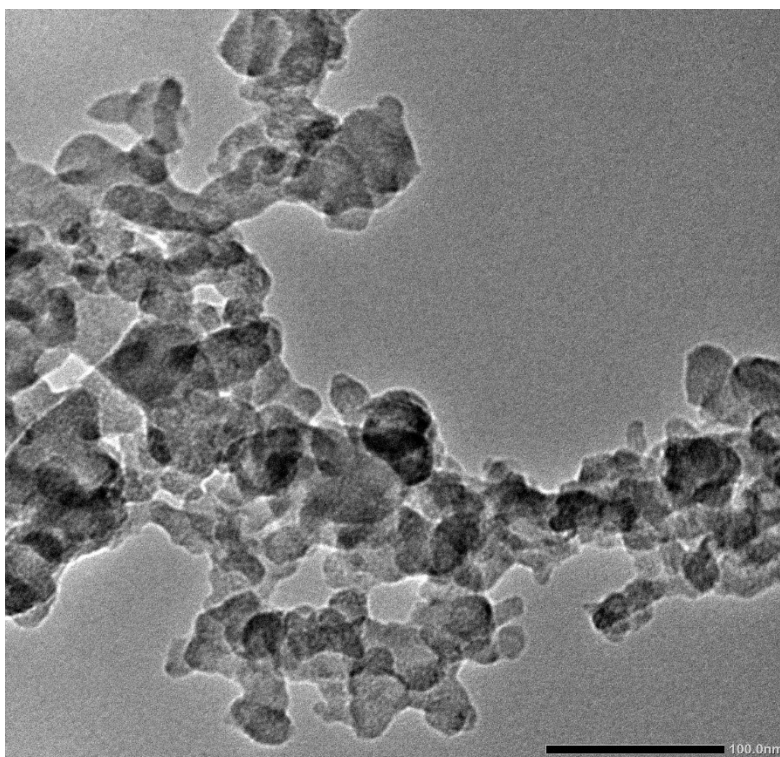


Figure S6. TEM image of **CuH** dispersed in silica

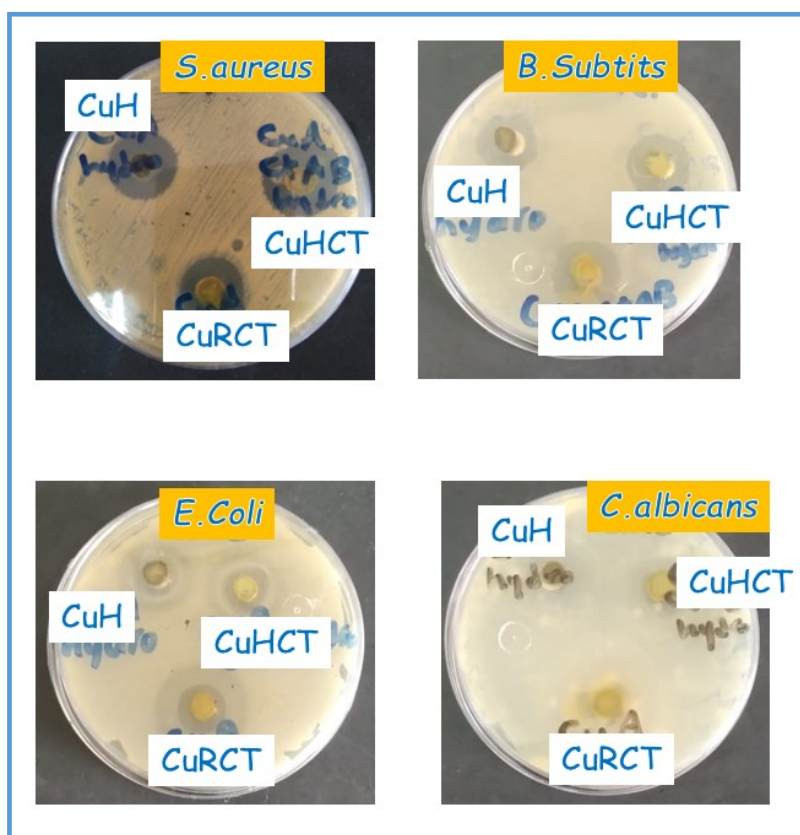


Figure S7. The Inhibition zone diameter of copper nano-chelates against *S. aureus*, *B. subtilis*, *E. coli*. (bacteria) and *C. albicans* (fungi).

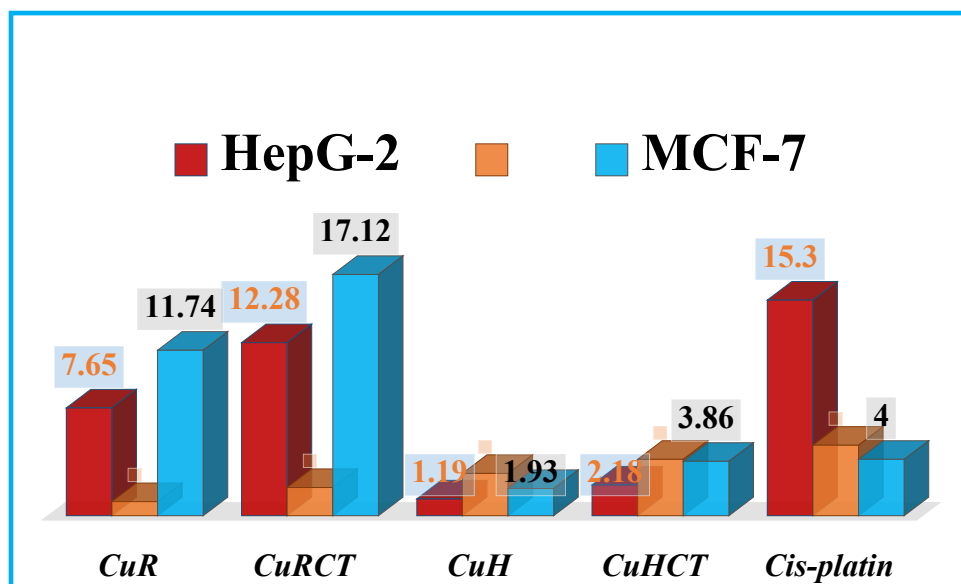


Figure S8. Comparison between IC₅₀ of HCBH, CuR, CuRCT, CuH and CuHCT nano-chelates and standard *cis-platin*.

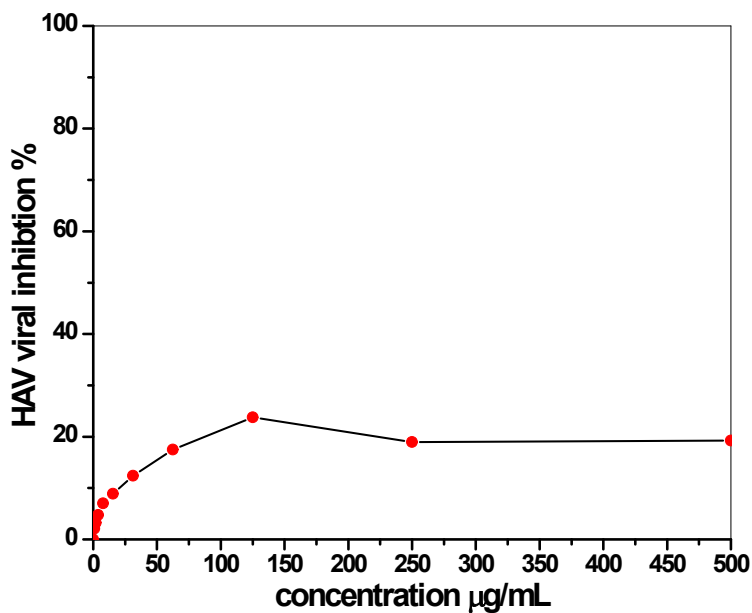


Figure S9. The antiviral activity of CuH nano-chelate against Hepatitis A virus (HAV).

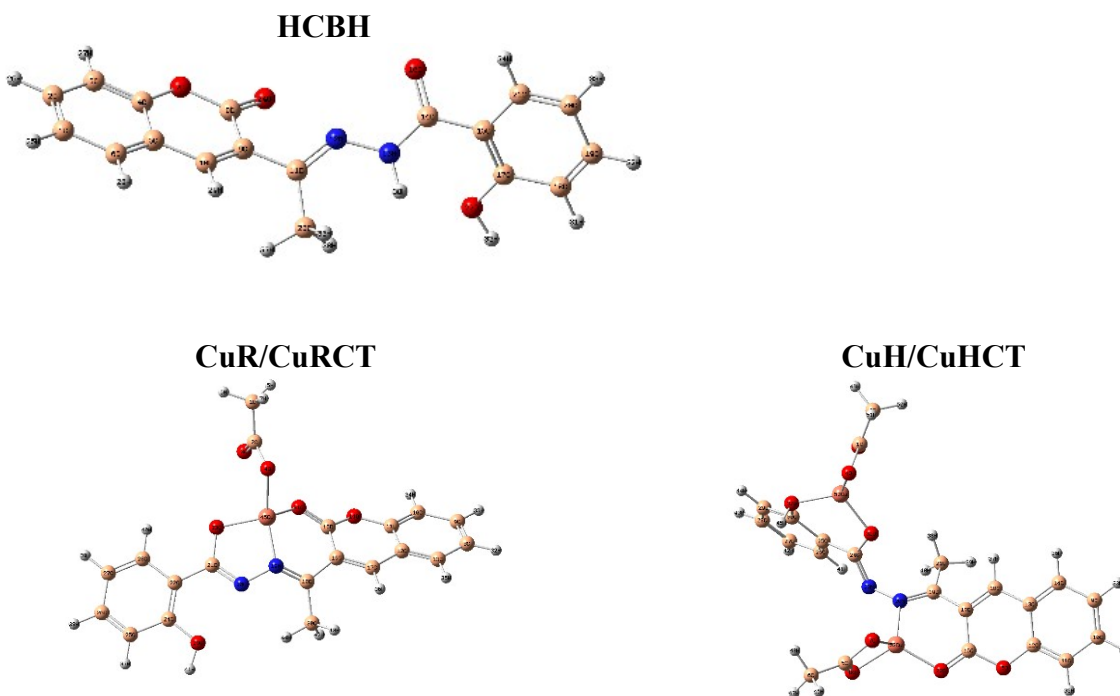


Figure S10. Optimized 3D structures of the **HCBH** and its copper chelates.

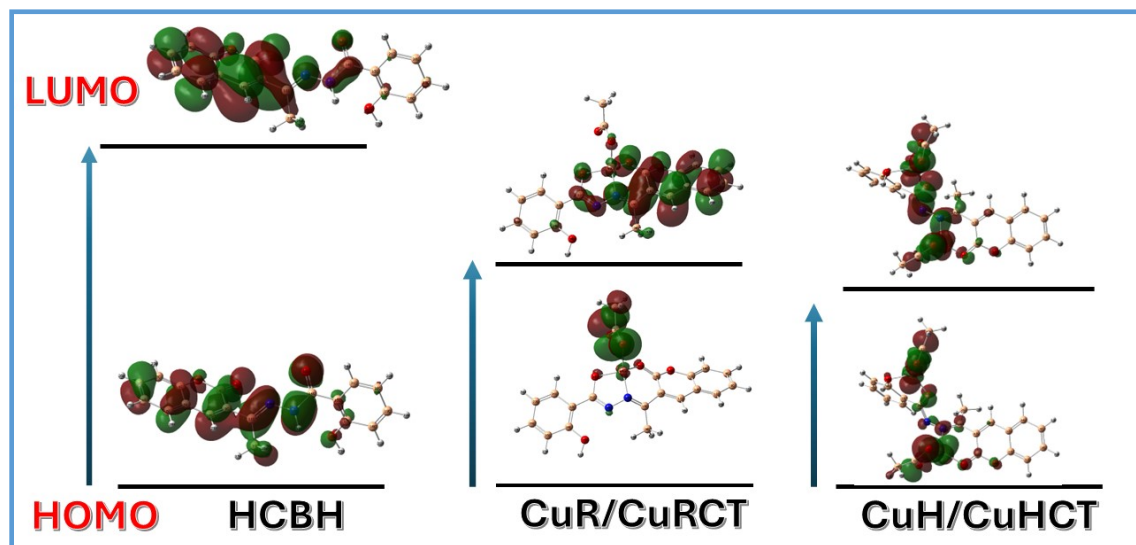


Figure S11. Graphical representation of HOMO-LUMO distribution of the HCBH and its copper nano-chelates.

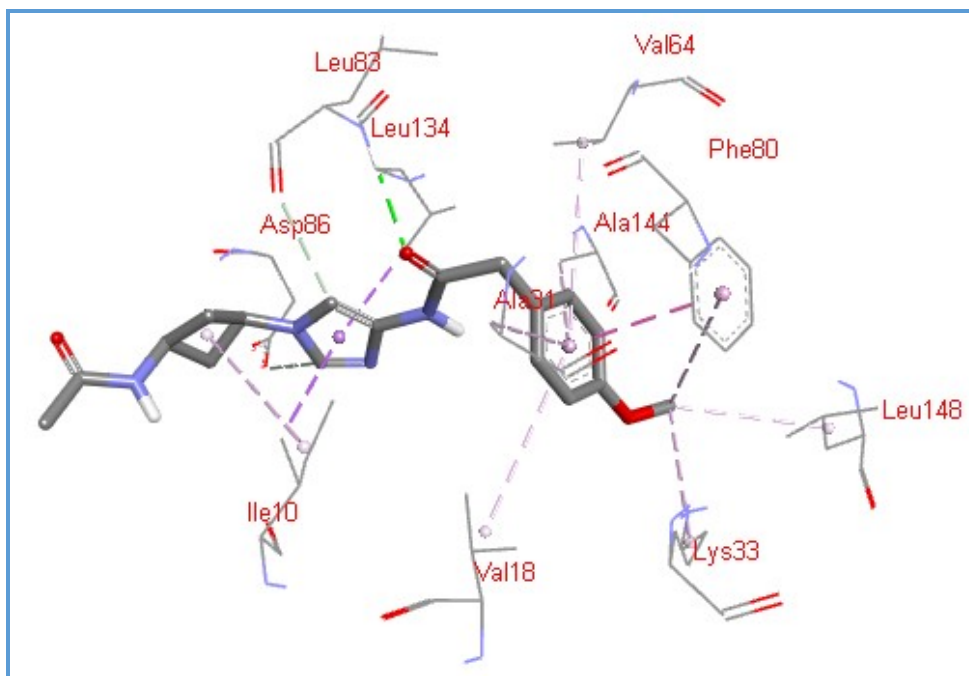


Figure S12. 3D representation of the (N-(1-[cis-3-(acetylamino)cyclobutyl]-1H-imidazol-4-yl)-2-(4-methoxyphenyl)acetamide ligand in target protein (pdb ID: 3IG7).

Table S1. Analytical, physical, and molar conductance data of the (**HCBH**) and its copper nano-chelates.

Compound	Molecular Formula, [F.Wt]	Color	Yield (%)	M.P. (°C)	Elemental analysis, found % (calcd %)				Ω_m (ohm ⁻¹ cm ² mol ⁻¹)
					% C	% H	% N	% M	
HCBH	C ₁₈ H ₁₄ N ₂ O ₄ [322.32]	White	92	229	67.00 (67.07)	4.30 (4.37)	8.55 (8.69)	-	-
CuR	C ₂₀ H ₁₆ N ₂ O ₆ Cu, [443.90]	brown	79	>300	54.43 (54.11)	3.87 (3.63)	6.16 (6.31)	14.05 (14.31)	12
CuRCT	C ₂₀ H ₁₇ N ₂ O _{6.5} Cu, [452.90]	Brown	72	>300	52.86 (53.04)	3.77 (3.78)	6.31 (6.18)	14.26 (14.03)	11
CuH	C ₂₄ H ₂₂ N ₂ O ₁₀ Cu ₂ , [625.53]	Brown	72	>300	46.00 (46.08)	3.46 (3.54)	4.29 (4.47)	20.51 (20.31)	75
CuHCT	C ₂₄ H ₂₂ N ₂ O ₁₀ Cu ₂ , [625.53]	Brown	79	>300	46.02 (46.08)	3.77 (3.54)	4.19 (4.47)	20.05 (20.31)	70

Table S2. Temperatures of decomposition and the kinetic parameters of complexes.

Chelates	Step	n order	T (K)	A (S ⁻¹)	Δ E (kJ mol ⁻¹)	ΔH (kJ mol ⁻¹)	ΔS (kJ mol ⁻¹ K ⁻¹)	ΔG (kJ mol ⁻¹)
CuR	First	1	540	1.34x10 ⁹	97.51	93.02	-0.835	138.15
	Second	0.33	672	2.55x10 ⁸	29.43	23.84	-0.099	90.39
CuH	First	0.33	567	1.16x10 ¹¹	13.88	9.17	-0.046	35.25
	Second	0.5	635	2.95x10 ⁵	69.20	63.92	-0.155	162.16
CuHCT	First	0	450	5.72x10 ⁹	14.43	10.68	-0.069	42.12
	Second	1	562	7.06x10 ⁶	42.69	38.02	-0.127	109.60
CuRCT	First	1	345	8.72 x10 ⁹	64.85	61.98	-0.064	84.06
	Second	0	590	3.81 x10 ⁷	34.83	29.92	-0.113	96.59
	Third	0.66	598	2.88x10 ⁶	77.82	72.85	-0.135	153.58

Table S3. Characteristic IR spectral data of the **HCBH** and its copper nano-chelates.

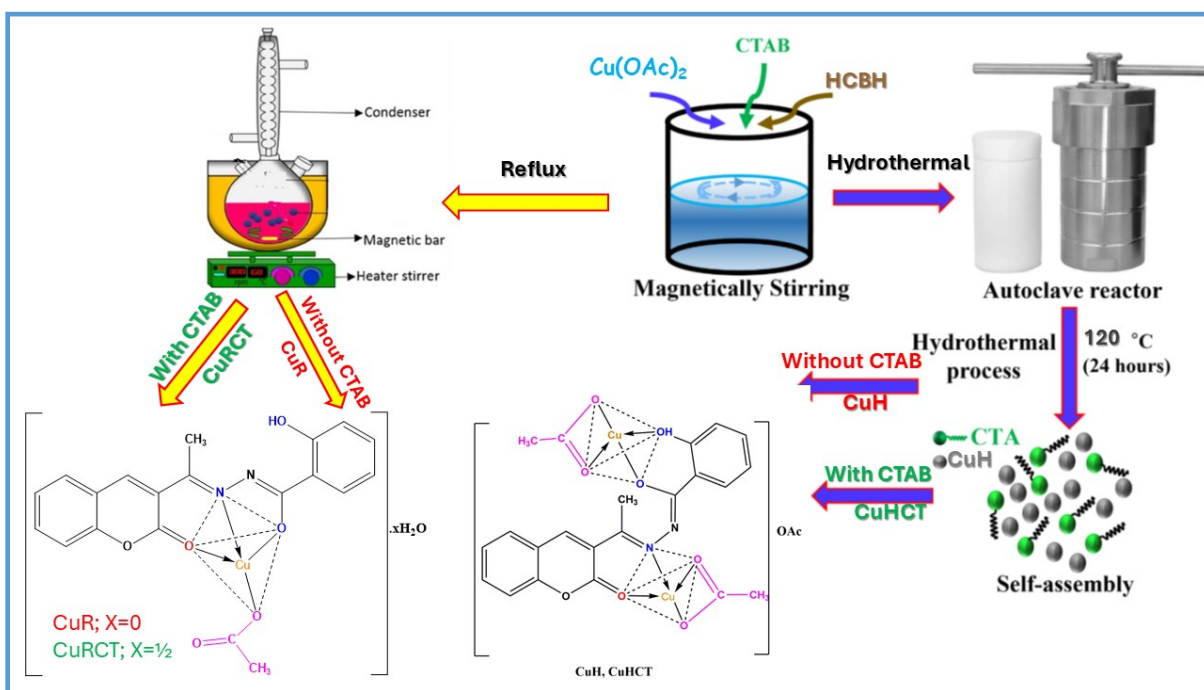
Compound	IR spectral bands (cm ⁻¹)						Additional bands
	ν(OH /H ₂ O)	ν(C-H)	ν(C=O) lactone	ν(C=N)	ν(M-O)	ν(M-N)	
HCBH	3469	3041	1724	1572	-	-	-
CuR	3419	3048	1694	1560, 1510	555, 514	454	1487, ν _{as} (COO ⁻)1256, ν _s (COO ⁻) (Monodentate oAc ⁻)
CuRCT	3411	3054	1694	1560, 1512	585, 539	451	1486, ν _{as} (COO ⁻)1255, ν _s (COO ⁻) (Monodentate oAc ⁻)
CuH	3377	3046	1693	1550, 1520	575, 539	433	1478, ν _{as} (COO ⁻)1301, ν _s (COO ⁻) (bidentate oAc ⁻)
CuHCT	3377	3046	1693	1555, 1510	583, 540	450	1434, ν _{as} (COO ⁻)1313, ν _s (COO ⁻) (bidentate oAc ⁻)

Table S4. Quantum chemical parameters of **HCBH** and its copper chelates

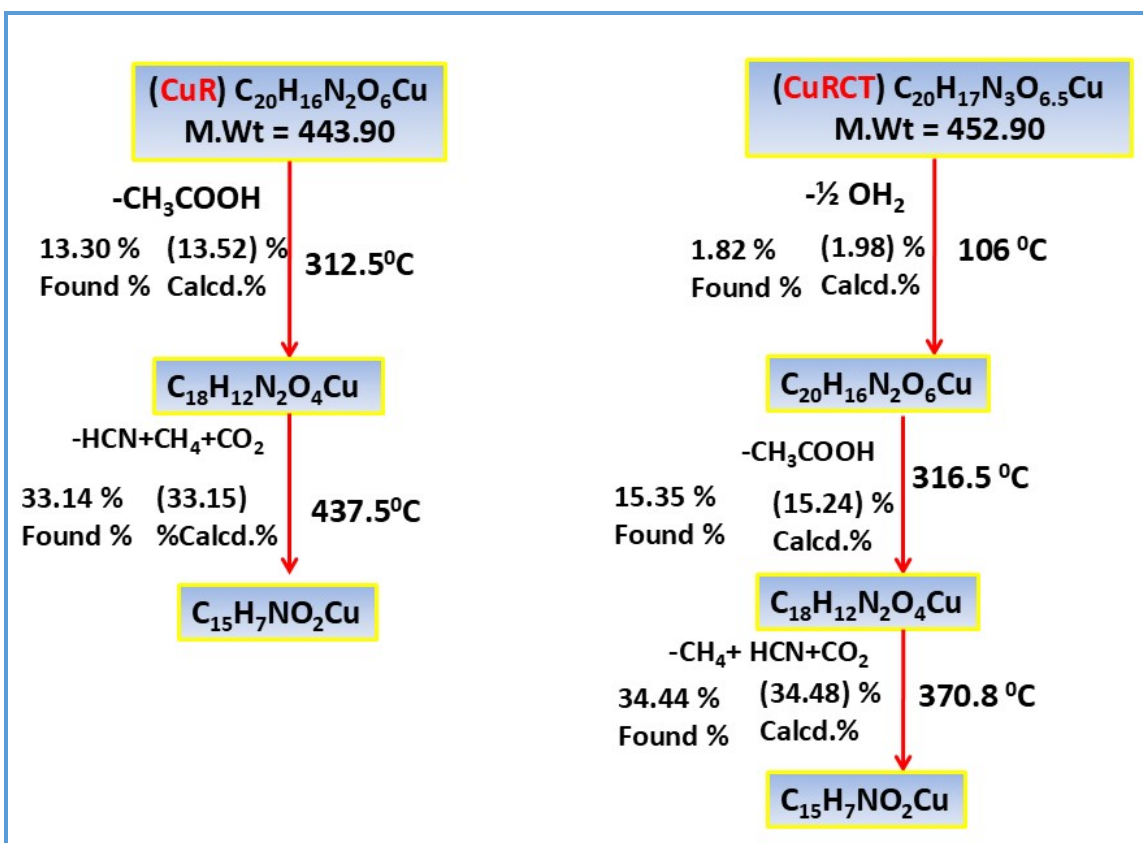
Compound	E _{HOMO}	E _{LUMO}	ΔE	IP	EA	χ	μ	η	σ	ω
HCBH	-6.26	-2.24	4.02	6.26	2.24	4.25	-4.25	2.01	0.25	4.49
CuR/CuRCT	-8.83	-5.98	2.84	8.83	5.98	7.40	-7.40	1.42	0.35	19.27
CuH/CuHCT	-8.22	-5.60	2.62	8.22	5.60	6.91	-6.91	1.31	0.38	18.23

Table S5. Hydrogen bonds between the studied compounds and the amino acids residues of the target protein (pdb ID: 3IG7).

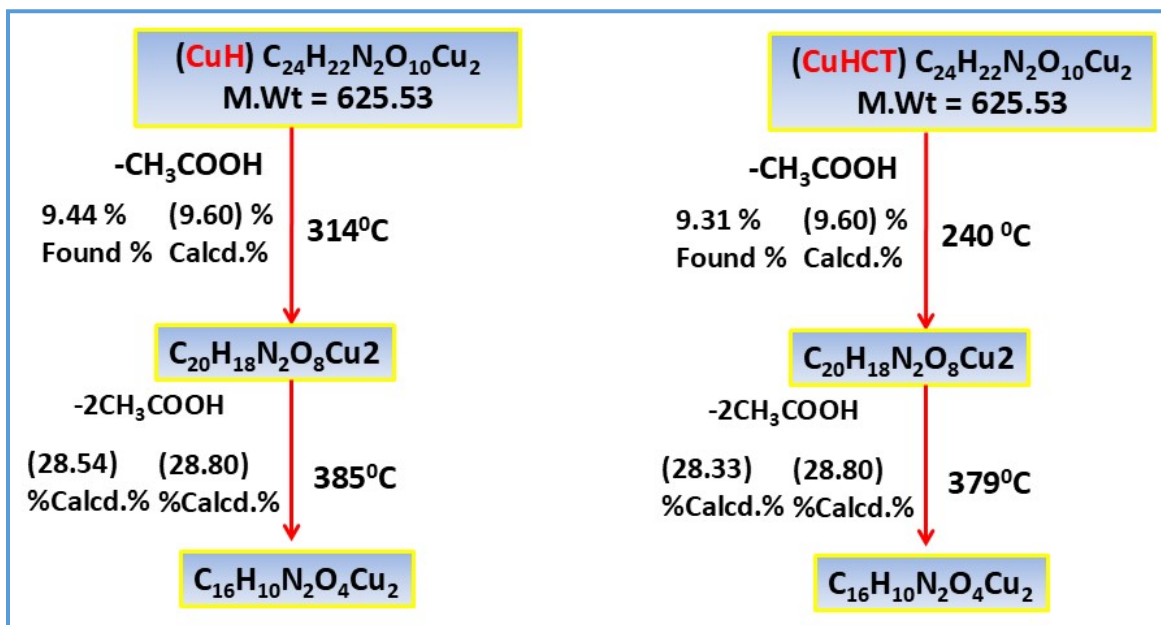
	Residual	Distance	Interaction	Binding energy (kcal/mol)
HCBH	HIS84	2.75	Hydrogen Bond	-6.70
	ASP145	4.78	Pi-Anion	
	LYS89	4.20	Pi-Alkyl	
CuR/CuRCT	LYS33	2.87	Hydrogen Bond	-8.20
	ASN132	2.86	Hydrogen Bond	
	ASP145	3.01	Hydrogen Bond	
	ASP86	2.55	Pi-Anion	
	LYS88	4.60	Pi-Alkyl	
	LYS33	4.32	Pi-Alkyl	
	LEU32	2.97	Hydrogen Bond	
CuH/CuHCT	GLY11	2.83	Hydrogen Bond	-8.40
	PHE80	4.08	Pi-Pi	
	VAL18	4.06	Pi-Alkyl	
	LYS33	4.94	Pi-Alkyl	
	VAL18	5.45	Pi-Alkyl	
	VAL18	4.51	Pi-Alkyl	
	ALA31	4.00	Pi-Alkyl	



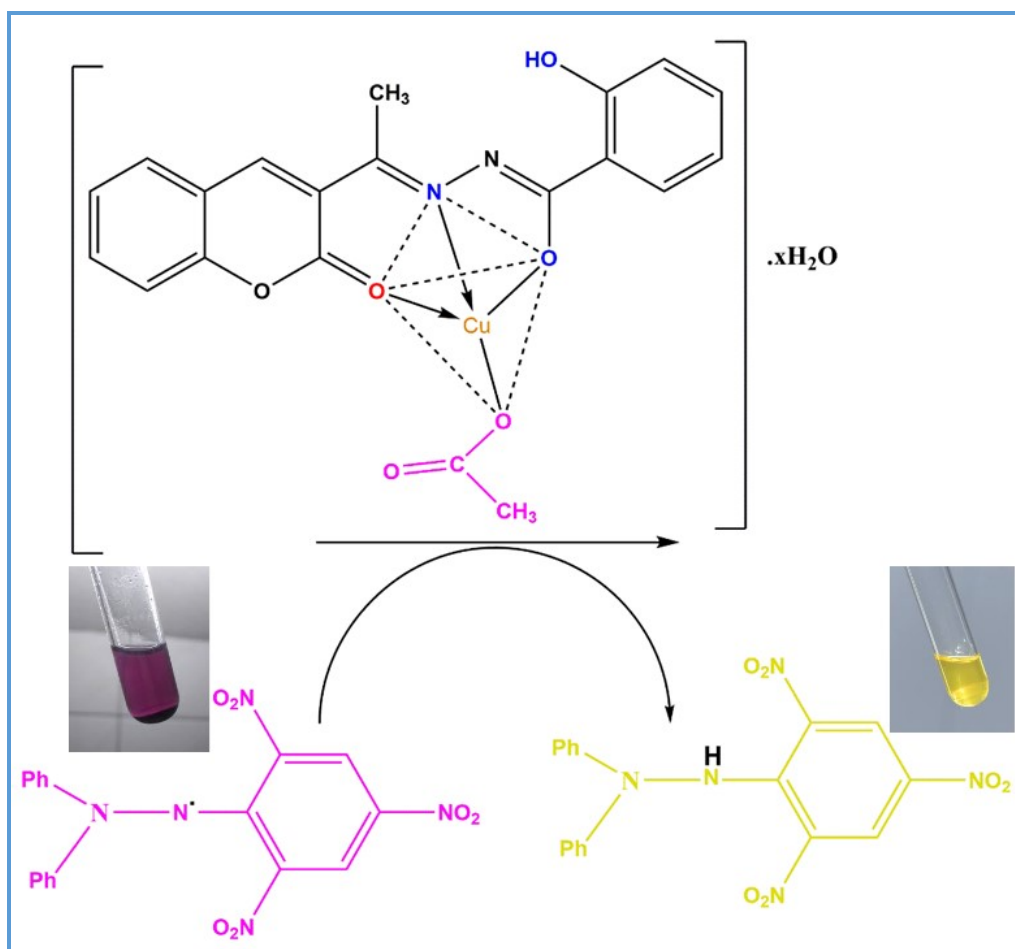
Scheme S1. Schematic representation of copper nano-chelate.



Scheme S2. Schematic representation of thermal degradation of **CuR**, and **CuRCT** chelates.



Scheme S3. Schematic representation of thermal degradation of **CuH**, and **CuHCT** chelates.



Scheme S4. The schematic representation of the color change of DPPH from purple to yellow after mixing with copper nano-chelate.