

Supporting Information for

**Catalytic Efficiency of Cu-MOFs: HKUST-1 and CuBDC for the
Protodeboronation of Aryl Boronic Acids**

Yudha P. Budiman,^{a,b,*} Muhamad Rashifari,^{a,b} Muhamad Diki Permana,^{b,c} Kansy
Haikal,^a Iis I. Widyowati,^a Yessi Permana,^d Ubed S. F. Arrozi,^e Wirawan Ciptonugraha,^f
Tri Mayanti,^a Allyn P. Sulaeman,^a Juliandri,^a Witri Wahyu Lestari^f

^aDepartment of Chemistry, Faculty of Mathematics and Natural Sciences, Universitas
Padjadjaran, 45363 Sumedang, Indonesia

^bSpecial Educational Program for Green Energy Conversion Science and Technology,
Integrated Graduate School of Medicine, Engineering, and Agricultural Sciences, University of
Yamanashi, Kofu 400-8511, Japan

^cCenter for Crystal Science and Technology, University of Yamanashi, Kofu 400-8511, Japan

^dInorganic and Physical Chemistry Research Division, Faculty of Mathematics and Natural
Sciences, Institut Teknologi Bandung, Bandung 40132, Indonesia

^eDepartment of Chemistry, Faculty of Mathematics and Natural Sciences, State University of
Malang, 65145, Malang, Indonesia

^fDepartment of Chemistry, Faculty of Mathematics and Natural Sciences, Universitas Sebelas
Maret, Surakarta, 57126, Indonesia,

*Corresponding author: y.p.budiman@unpad.ac.id

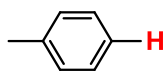
S1. General Information

Unless otherwise noted, all reagents were purchased from Sigma-Aldrich were checked for purity by GC-MS and used as received. Absolute ethanol (Sigma-Aldrich, 99.8%) was used as received. The removal of solvent was performed by an extraction method in a mixture of brine water (2 mL) and ethyl acetate (2 mL) three times.

High-angle X-ray diffraction patterns were obtained on a Rigaku Miniflex 600 D/teX Ultra diffractometer with Cu K α radiation. Infrared spectra were recorded on a Nicolet 380 FT-IR spectrometer as solids, using an ATR unit, and are reported in cm⁻¹. GC-MS analyses were performed using an Agilent 7890A gas chromatograph (column: DB-5MS 5% phenylmethylsiloxane, 30 m, Ø 0.25 mm, film 0.25 μ m; *injector: 200 °C; oven: 40 °C (2 min), 40 °C to 280 °C (20 °C min⁻¹) (5 min) *; carrier gas: He (1 mL min⁻¹) equipped with an Agilent 5977B GC/MSD operating in EI mode). Copper atomic absorption spectrometry were conducted on a Perkin Elmer AAnalyst 400 AA with Series N30501XX Copper (Cu) Lumina hollow cathode lamp.

All NMR spectra were recorded at 298 K using an Agilent DD2 (¹H-NMR, 500 MHz; ¹³C{¹H}-NMR, 125 MHz). ¹H-NMR chemical shifts are reported relative to TMS and were referenced via residual proton resonances of the corresponding deuterated solvent (CDCl₃: 7.24 ppm), ¹³C{¹H}-NMR spectra are reported relative to TMS via the carbon signals of the deuterated solvent (CDCl₃: 77.16 ppm

S1.1 Synthesis of toluene (2a) from *p*-tolylboronic acid (1a)

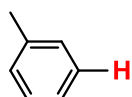


Compound **2a** was synthesized following the general procedure and using the following chemicals and conditions: *p*-tolylboronic acid (54.3 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 3.630 min] m/z: 91 [*M*]⁺.

This compound is known.¹

S1.2 Synthesis of toluene (2a) from *m*-tolylboronic acid (1b)

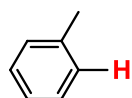


Compound **2a** was synthesized following the general procedure and using the following chemicals and conditions: *m*-tolylboronic acid (54.3 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 3.625 min] m/z: 91 [*M*]⁺.

This compound is known.¹

S1.3 Synthesis of toluene (2a) from *o*-tolylboronic acid (1c)



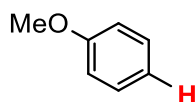
Compound **2a** was synthesized following the general procedure and using the following chemicals and conditions: *o*-tolylboronic acid (54.3 mg, 0.4 mmol), HKUST-1 (26.4 mg,

0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μ L, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 $^{\circ}$ C, and 1.5 h.

GCMS: [t = 3.633 min] m/z: 91 [M]⁺.

This compound is known.¹

S1.4 Synthesis of anisole (2d) from *p*-methoxyphenylboronic acid (1d)

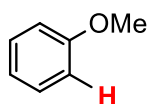


Compound **2d** was synthesized following the general procedure and using the following chemicals and conditions: *p*-methoxyphenylboronic acid (60.8 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μ L, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 $^{\circ}$ C, and 1.5 h.

GCMS: [t = 5.353 min] m/z: 109 [M]⁺.

This compound is known.¹

S1.5 Synthesis of anisole (2d) from *o*-methoxyphenylboronic acid (1e)

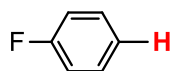


Compound **2d** was synthesized following the general procedure and using the following chemicals and conditions: *o*-methoxyphenylboronic acid (60.8 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μ L, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 $^{\circ}$ C, and 1.5 h.

GCMS: [t = 5.346 min] m/z: 109 [M]⁺.

This compound is known.¹

S1.6 Synthesis of fluorobenzene (2f)

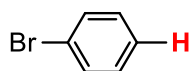


Compound **2f** was synthesized following the general procedure and using the following chemicals and conditions: *p*-fluorophenylboronic acid (56.0 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 40 °C, and 1.5 h.

GCMS: [t = 2.626 min] m/z: 96 [M]⁺.

This compound is known.²

S1.7 Synthesis of bromobenzene (2g)

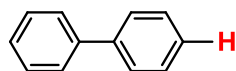


Compound **2g** was synthesized following the general procedure and using the following chemicals and conditions: *p*-bromophenylboronic acid (80.3 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 5.518 min] m/z: 156 [M]⁺.

This compound is known.²

S1.8 Synthesis of biphenyl (2h)



Compound **2h** was synthesized following the general procedure and using the following chemicals and conditions: biphenylboronic acid (79.2 mg, 0.4 mmol), HKUST-1 (26.4

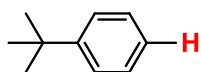
mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 9.287 min] m/z: 155 [M]⁺.

¹H NMR (500 MHz, CDCl₃) δ = 7.64–7.63 (*m*, 4H; Ar-H), 7.48 (*t*, 4H, ³J_{H,H} = 8 Hz; Ar-H), 7.39 (*t*, 2H, ³J_{H,H} = 7 Hz; Ar-H); **¹³C{¹H} NMR** (125 MHz, CDCl₃) δ = 141.4, 128.9, 127.4, 127.3

This compound is known.²

S1.9 Synthesis of *tert*-butylbenzene (2i)

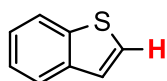


Compound **2i** was synthesized following the general procedure and using the following chemicals and conditions: *tert*-butylphenylboronic acid (71.2 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 6.091 min] m/z: 134 [M]⁺.

This compound is known.²

S1.10 Synthesis of benzo[*b*]thiophene (2j)



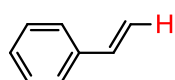
Compound **2j** was synthesized following the general procedure and using the following chemicals and conditions: benzo[*b*]thien-2-ylboronic acid (71.2 mg, 0.4 mmol), HKUST-1 (26.4 mg, 0.04 mmol), K₂CO₃ (55.3 mg, 0.4 mmol), mesitylene (56 μL, 0.4 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h. Due to strong interactions

between sulfur-containing heterocycles such as benzo[*b*]thiophene (**2j**) with silica gel during column chromatography, elution becomes challenging, resulting in a low isolated yield that is insufficient for NMR measurement.

GCMS: [t = 7.921 min] m/z: 134 [*M*]⁺.

This compound is known.³

S1.11 Synthesis of styrene (**2k**)

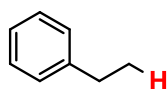


Compound **2k** was synthesized following the general procedure and using the following chemicals and conditions: trans-2-Phenylvinylboronic acid (29.6 mg, 0.2 mmol), HKUST-1 (13.2 mg, 0.02 mmol), K₂CO₃ (27.6 mg, 0.2 mmol), mesitylene (28 μL, 0.2 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 4.946 min] m/z: 104 [*M*]⁺.

This compound is known.⁴

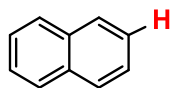
S1.12 Synthesis of ethylbenzene (**2l**)



Compound **2l** was synthesized following the general procedure and using the following chemicals and conditions: trans-2-Phenylvinylboronic acid (30 mg, 0.2 mmol), HKUST-1 (13.2 mg, 0.02 mmol), K₂CO₃ (27.6 mg, 0.2 mmol), mesitylene (28 μL, 0.2 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

This compound is known.⁵

S1.13 Synthesis of naphthalene (2m)

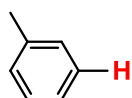


Compound **2m** was synthesized following the general procedure and using the following chemicals and conditions: 4,4,5,5-tetramethyl-2-(naphthalen-2-yl)-1,3,2-dioxaborolane (50.82 mg, 0.2 mmol), HKUST-1 (13.2 mg, 0.02 mmol), K₂CO₃ (27.6 mg, 0.2 mmol), mesitylene (28 μ L, 0.2 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

GCMS: [t = 7.759 min] m/z: 128 [M]⁺. **¹H NMR** (500 MHz, CDCl₃) δ = 3.10 (*q*, 4H, ³J_{H,H} = 3 Hz), 7.52 (*d*, 4H, ³J_{H,H} = 3 Hz); **¹³C NMR** (125 MHz, CDCl₃) δ = 133, 5, 128.0, 125.9; **GCMS:** [t = 7.935 min] m/z: 134 [M]⁺.

This compound is known.²

S1.14 Synthesis of toluene (2a) from 4,4,5,5-tetramethyl-2-(*m*-tolyl)-1,3,2-dioxaborolane (1n)



Compound **2n** was synthesized following the general procedure and using the following chemicals and conditions: 4,4,5,5-tetramethyl-2-(*m*-tolyl)-1,3,2-dioxaborolane (43.6 mg, 0.2 mmol), HKUST-1 (13.2 mg, 0.02 mmol), K₂CO₃ (27.6 mg, 0.2 mmol), mesitylene (28 μ L, 0.2 mmol), EtOH/H₂O (1 mL : 1 mL), O₂ atmosphere, 70 °C, and 1.5 h.

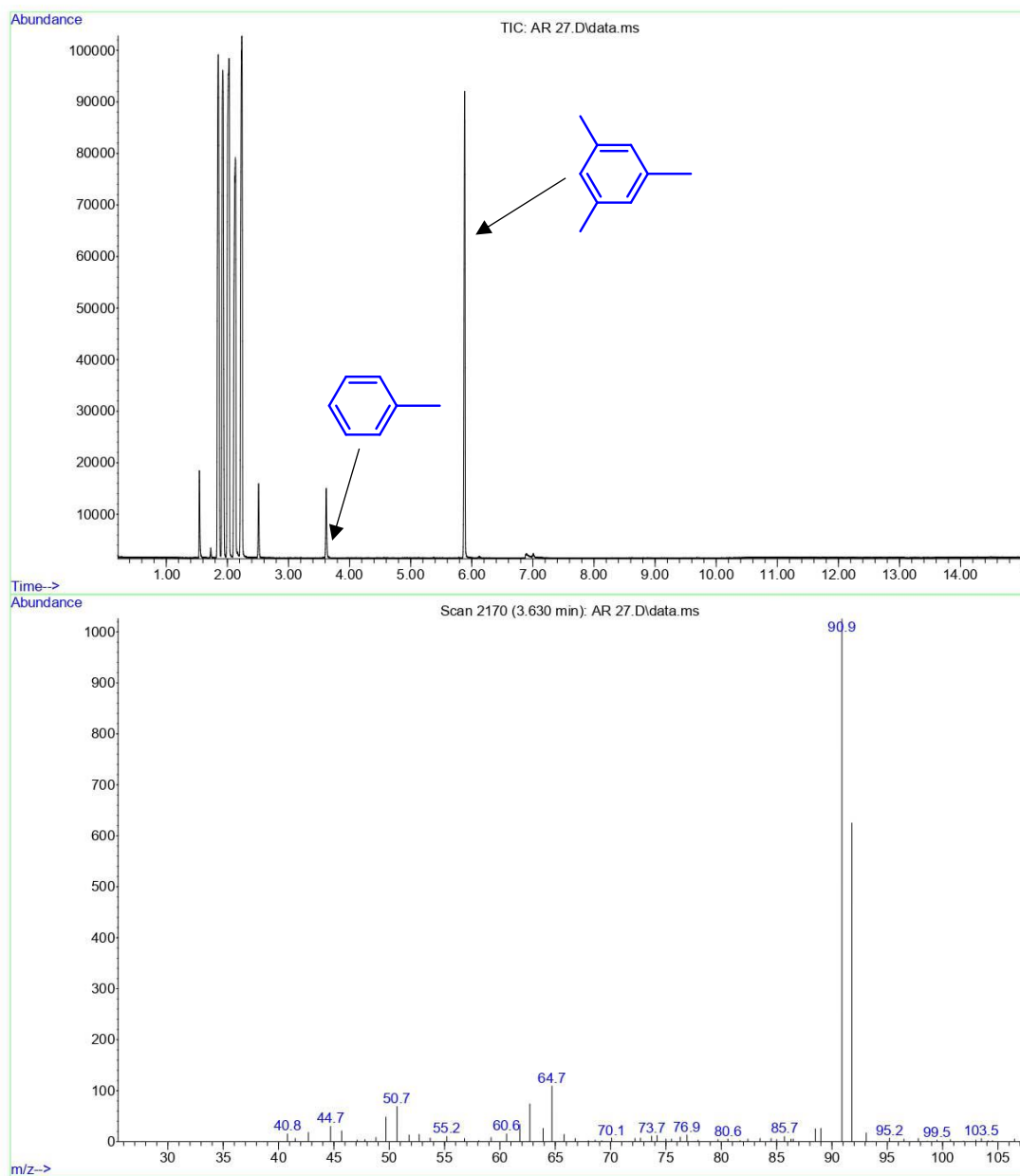
GCMS: [t = 3.633 min] m/z: 91 [M]⁺.

This compound is known.¹

S2. GC-MS and ^1H , $^{13}\text{C}\{^1\text{H}\}$ NMR Spectra

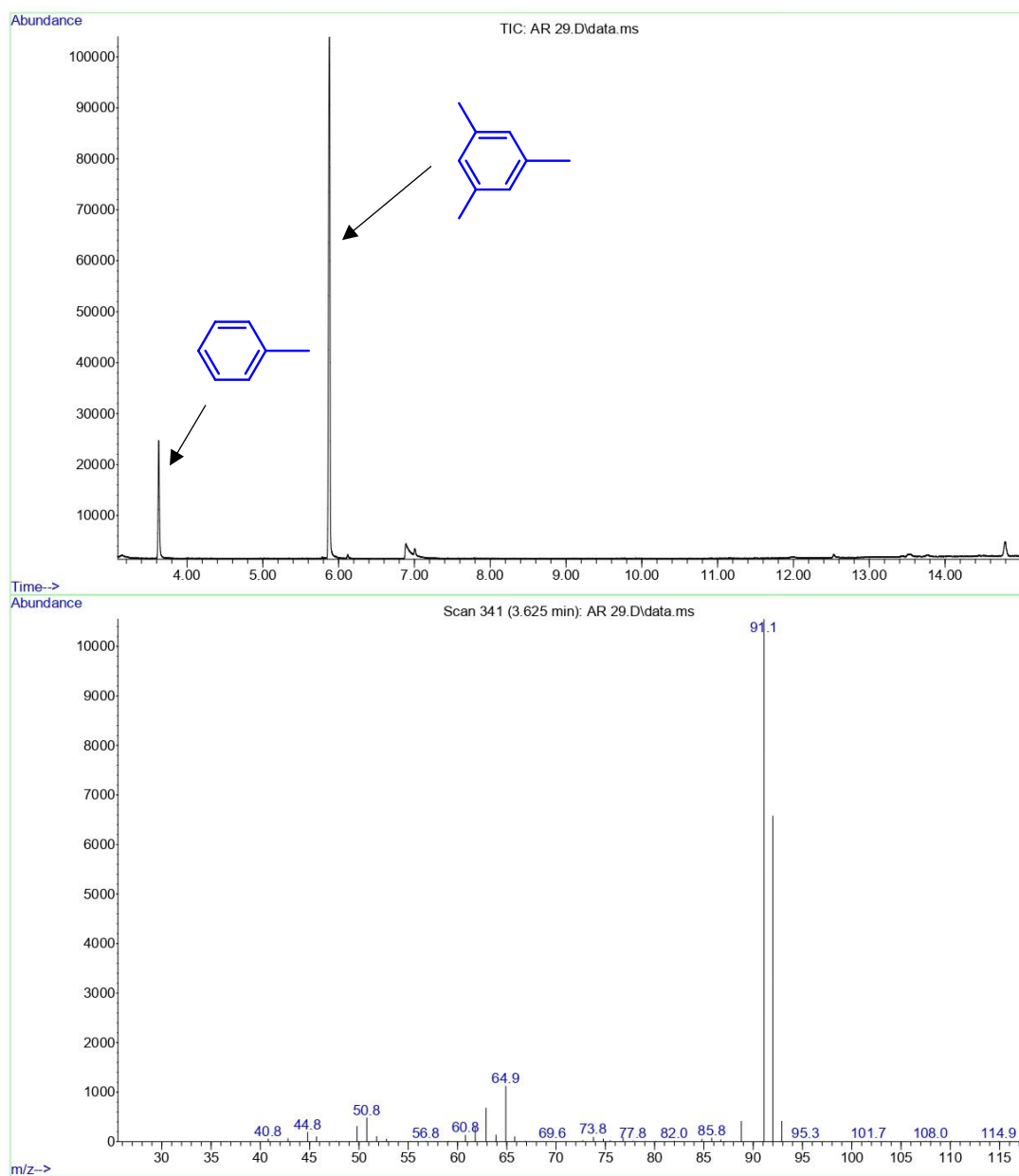
S2.1 GC-MS spectrum of the reaction described in Table 3, entry 1, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 27.D
Operator :
Acquired : 15 Nov 2024 17:45 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



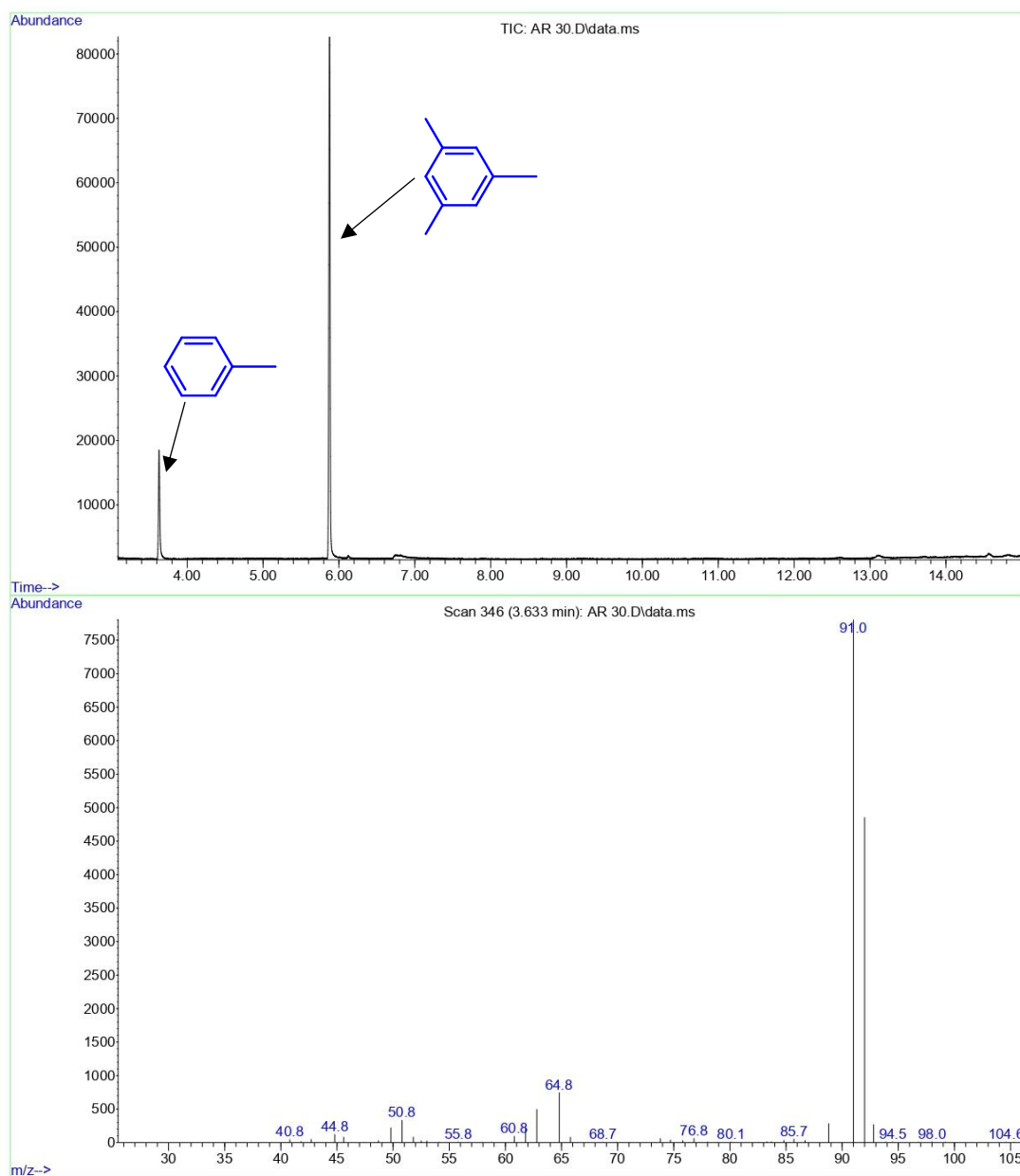
S2.2 GC–MS spectrum of the reaction described in Table 3, entry 2, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 29.D
Operator :
Acquired : 28 Nov 2024 07:46 using AcqMethod Umi.M
Instrument : GCMS
Sample Name :
Misc Info :
Vial Number: 0



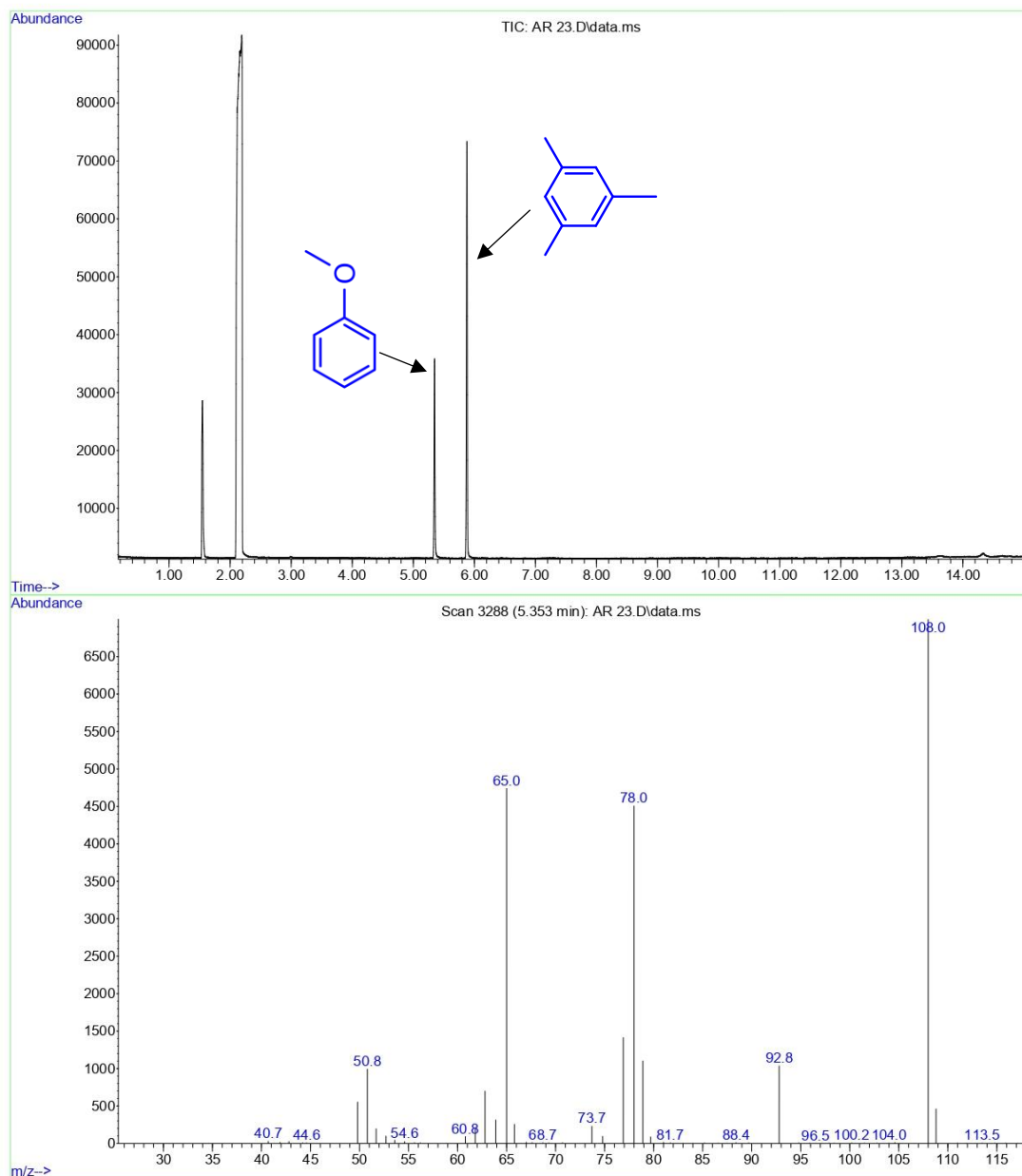
S2.3 GC–MS spectrum of the reaction described in Table 3, entry 3, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 30.D
Operator :
Acquired : 28 Nov 2024 08:25 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



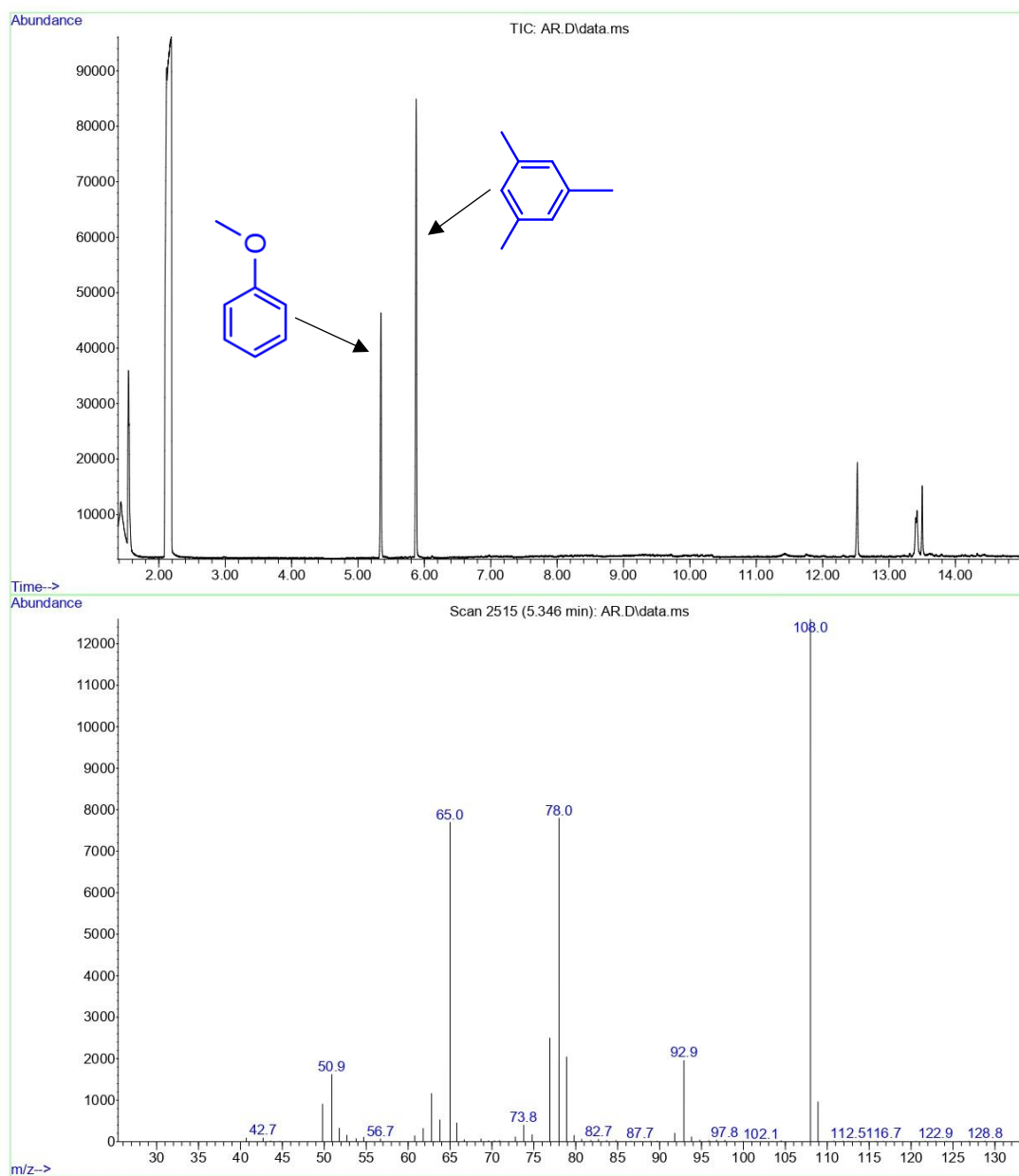
S2.4 GC–MS spectrum of the reaction described in Table 3, entry 4, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 23.D
Operator :
Acquired : 15 Nov 2024 16:09 using AcqMethod Umi.M
Instrument : GCMS
Sample Name :
Misc Info :
Vial Number: 0



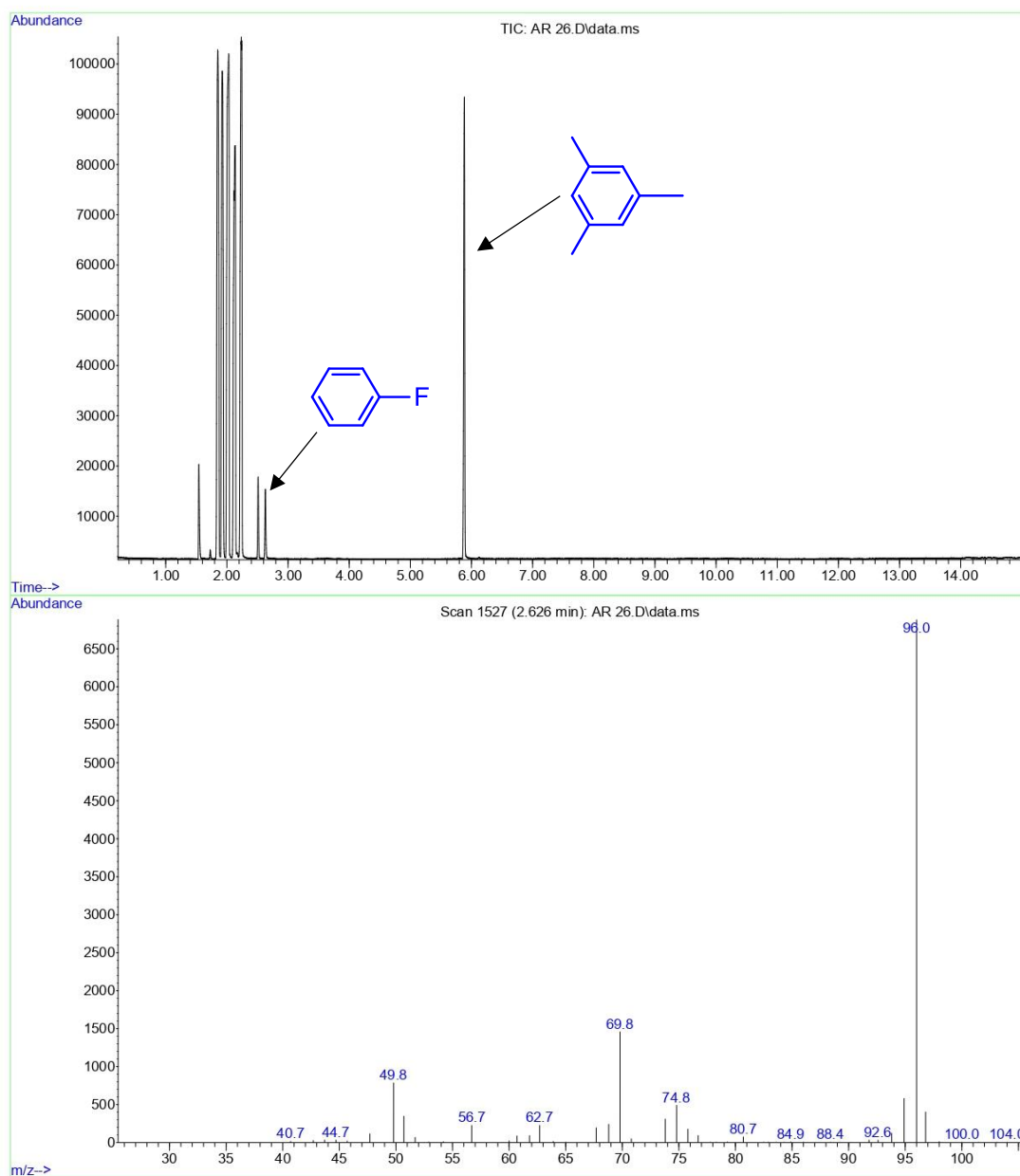
S2.5 GC–MS spectrum of the reaction described in Table 3, entry 5, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR.D
 Operator :
 Acquired : 15 Nov 2024 14:31 using AcqMethod Umi.M
 Instrument : GCMS
 Sample Name:
 Misc Info :
 Vial Number: 0



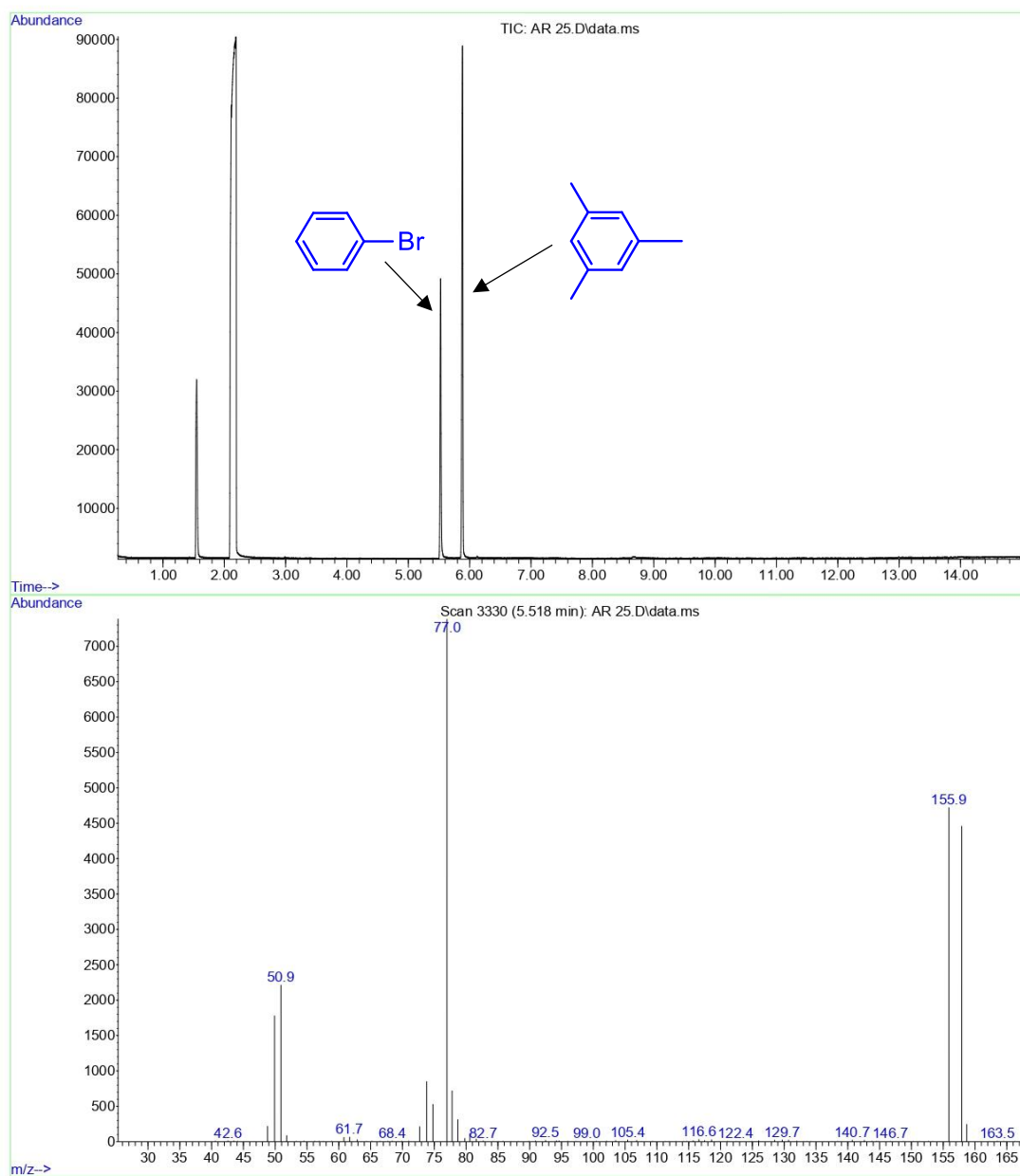
S2.6 GC–MS spectrum of the reaction described in Table 3, entry 6, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 26.D
Operator :
Acquired : 15 Nov 2024 16:58 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



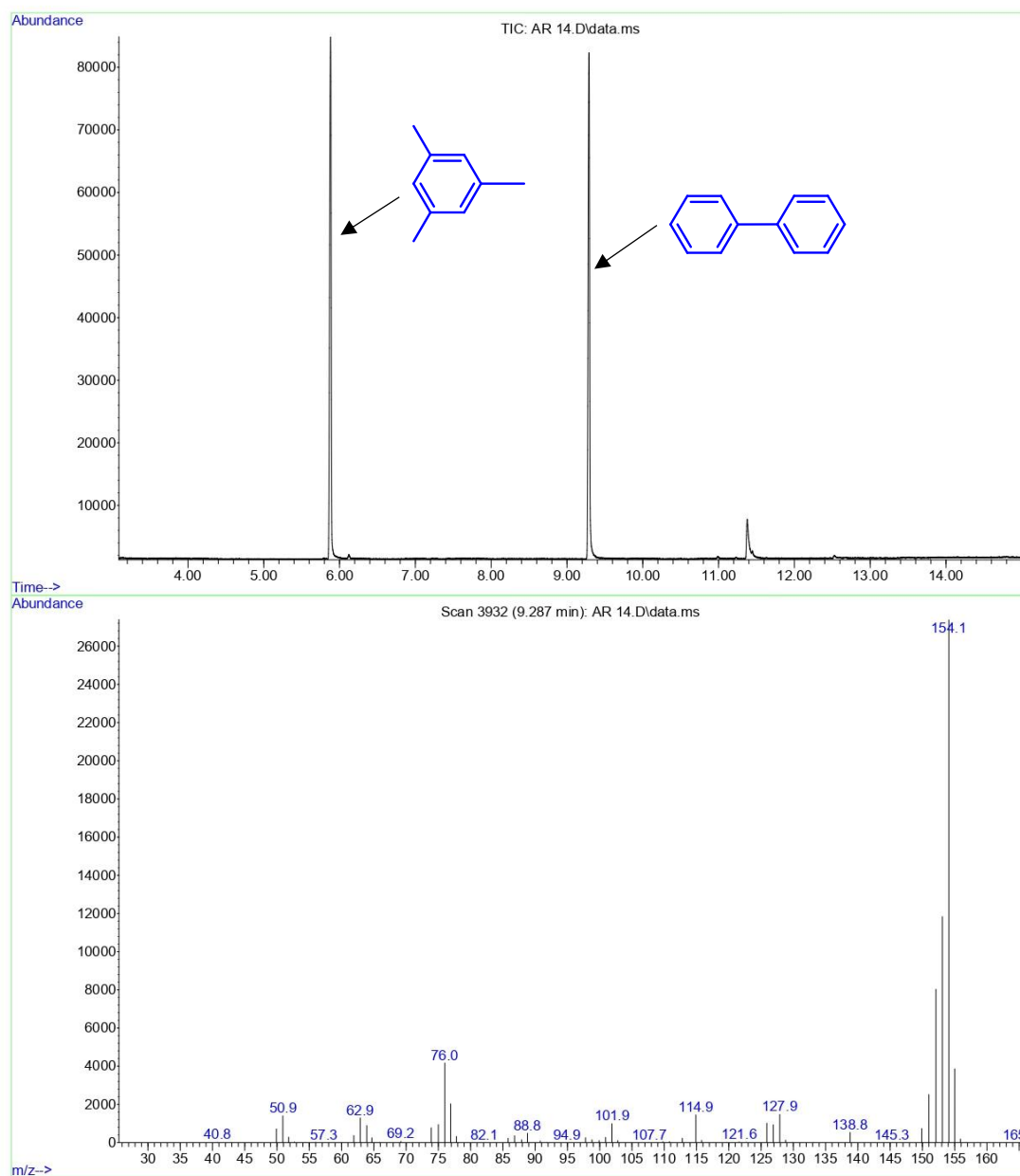
S2.7 GC–MS spectrum of the reaction described in Table 3, entry 7, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 25.D
Operator :
Acquired : 15 Nov 2024 16:33 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0

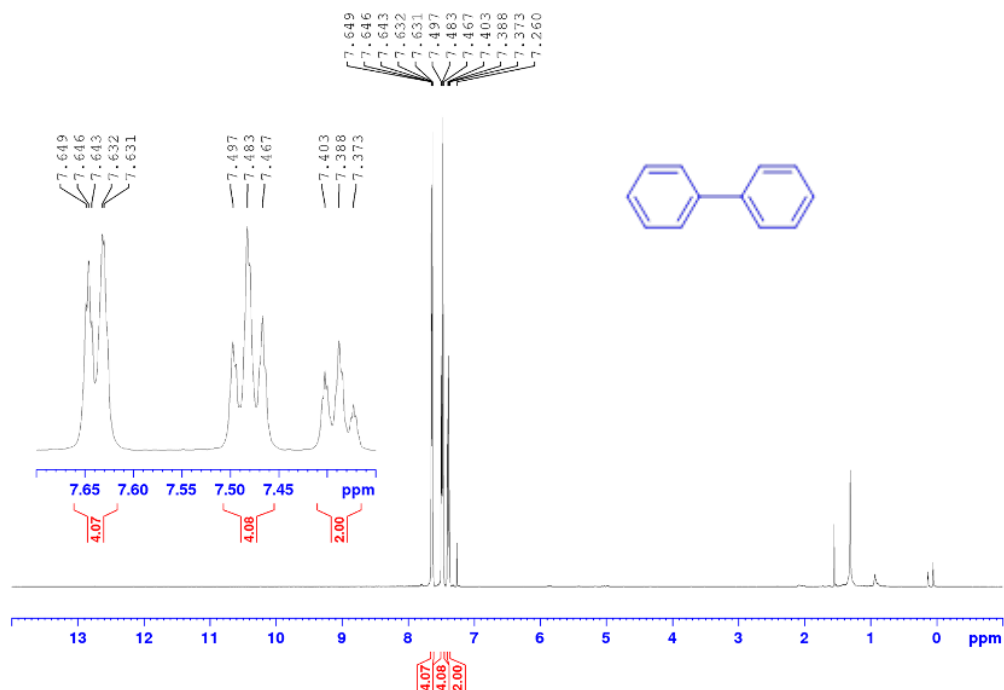


S2.8 GC–MS spectrum of the reaction described in Table 3, entry 8, with mesitylene as the internal standard.

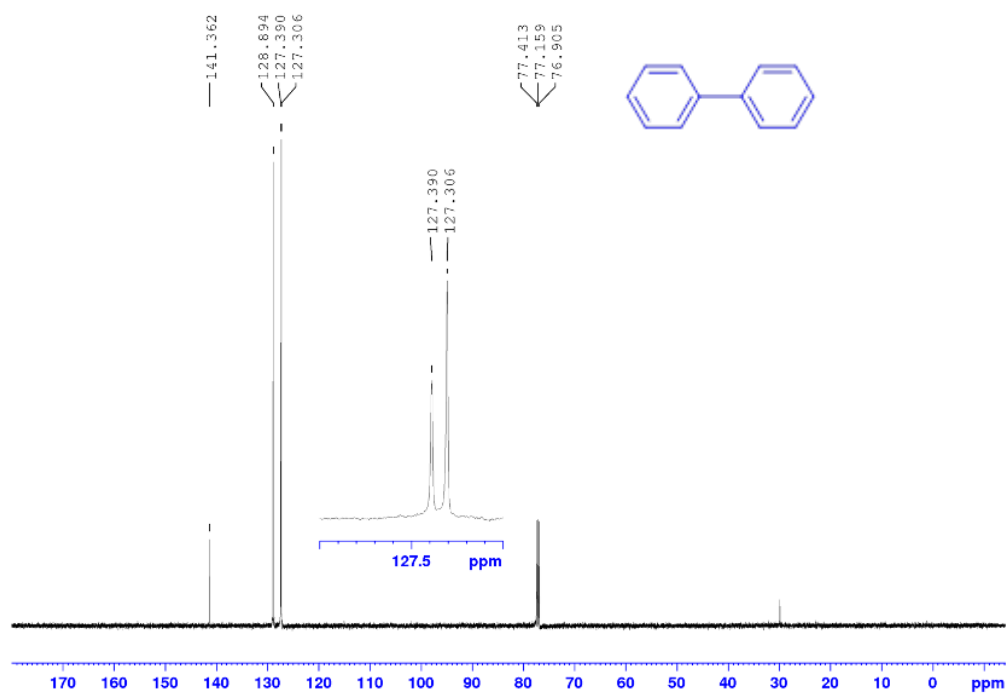
File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 14.D
Operator :
Acquired : 07 Nov 2024 13:56 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



S2.9 ^1H NMR spectrum of **2h** (500 MHz, CDCl_3)

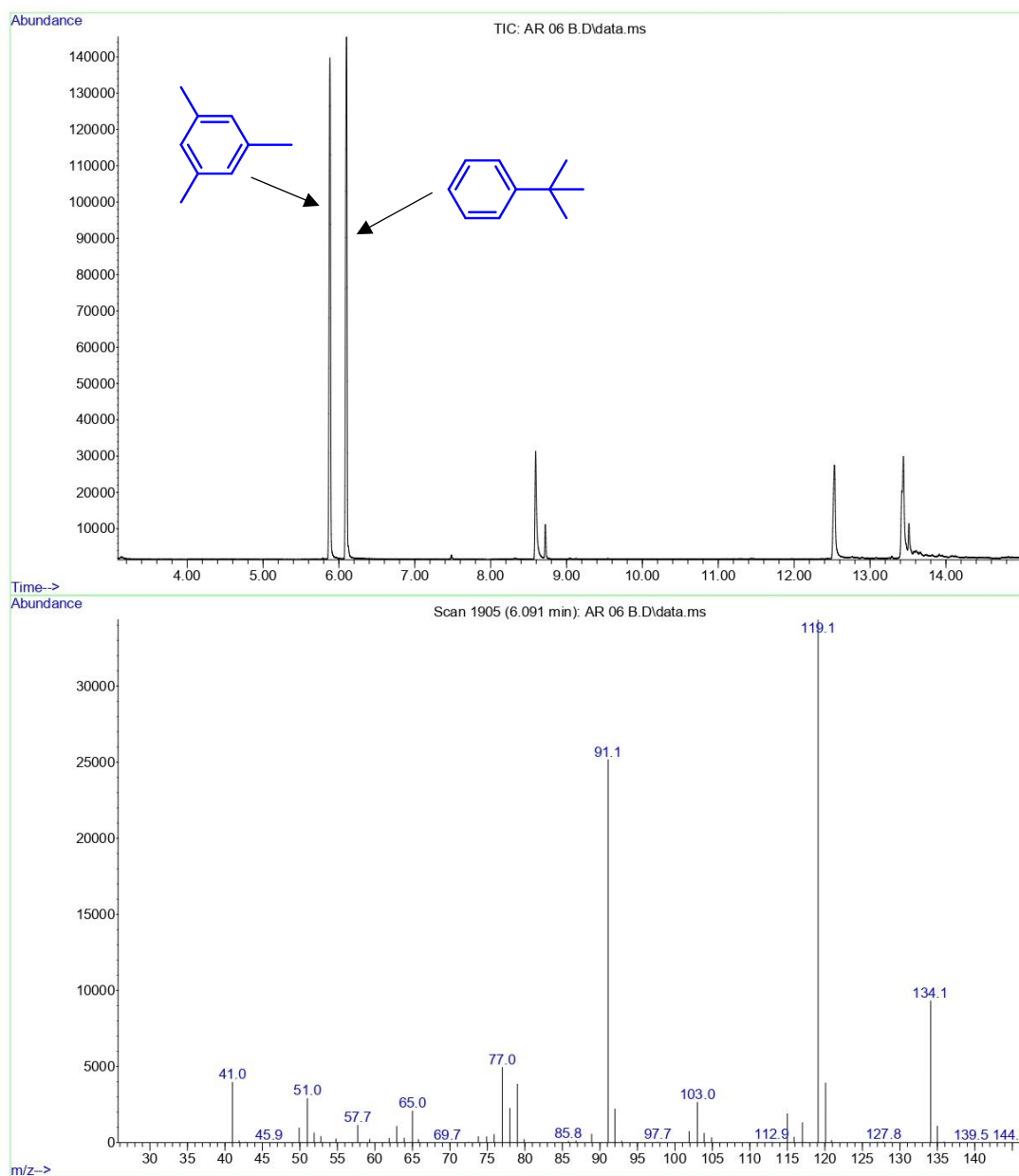


S2.10 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **2h** (125 MHz, CDCl_3)



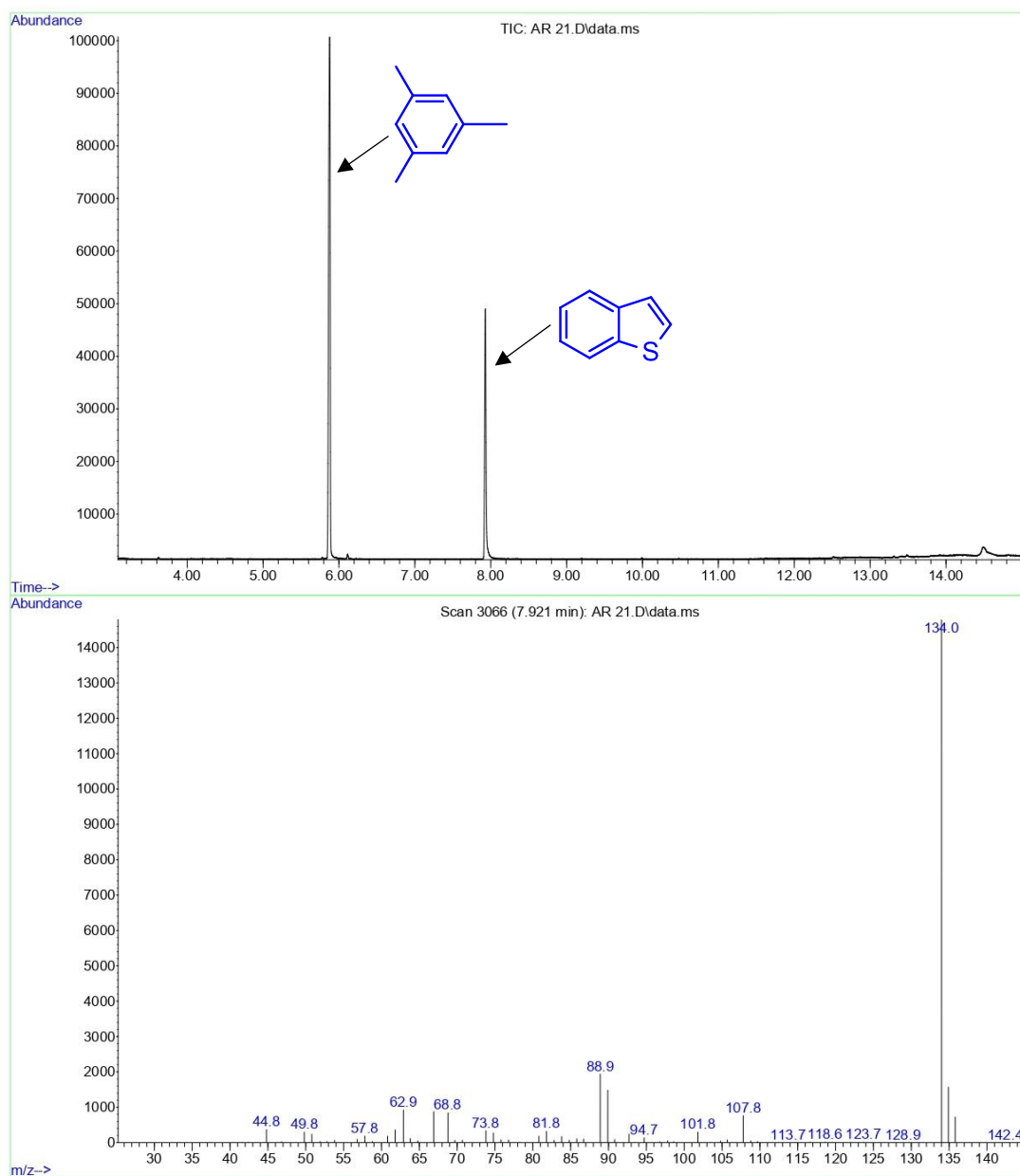
S2.11 GC–MS spectrum of the reaction described in Table 3, entry 9, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 06 B.D
Operator :
Acquired : 19 Aug 2024 08:40 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



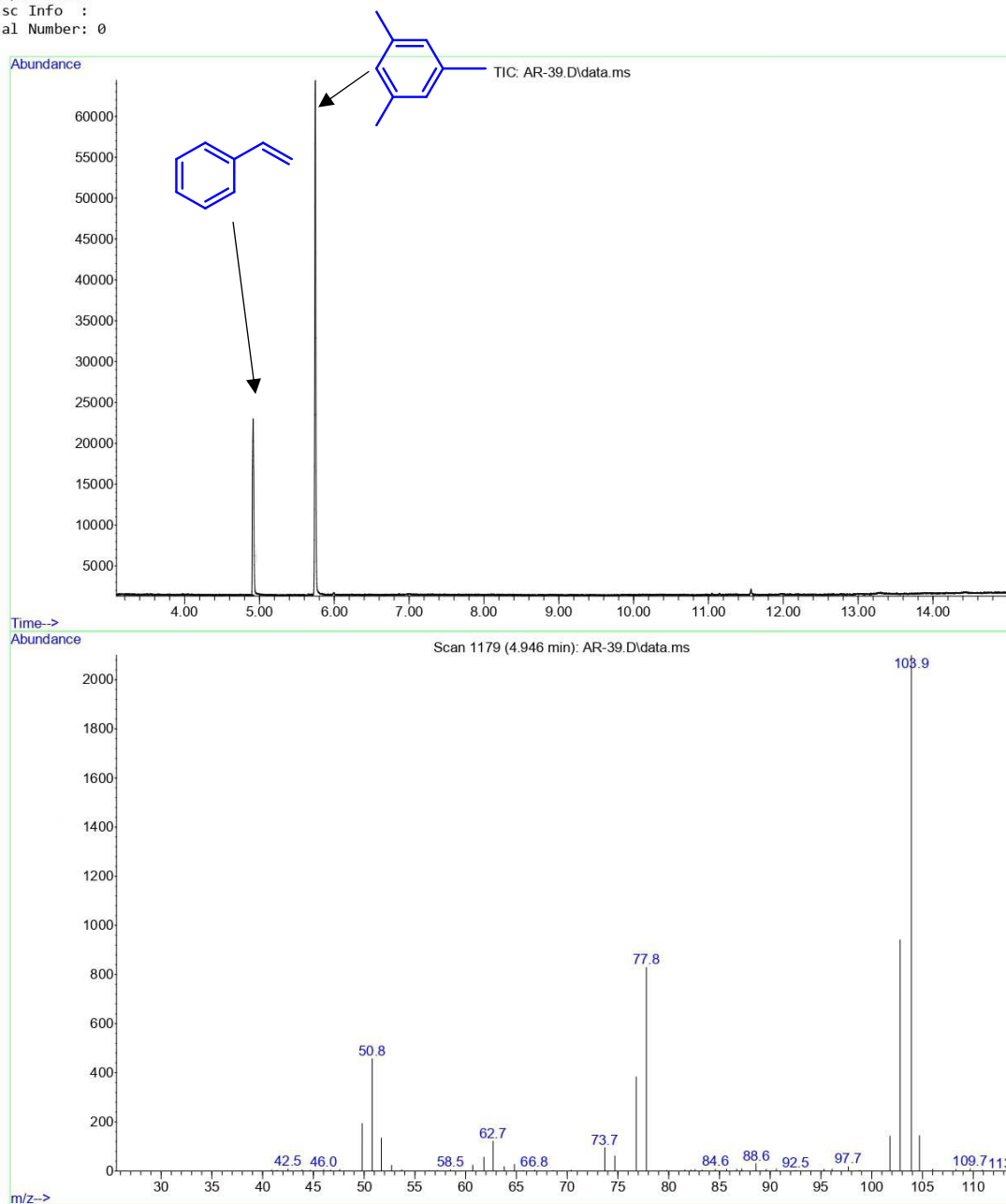
S2.12 GC–MS spectrum of the reaction described in Table 3, entry 10, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2024\AR 21.D
Operator :
Acquired : 12 Nov 2024 08:09 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



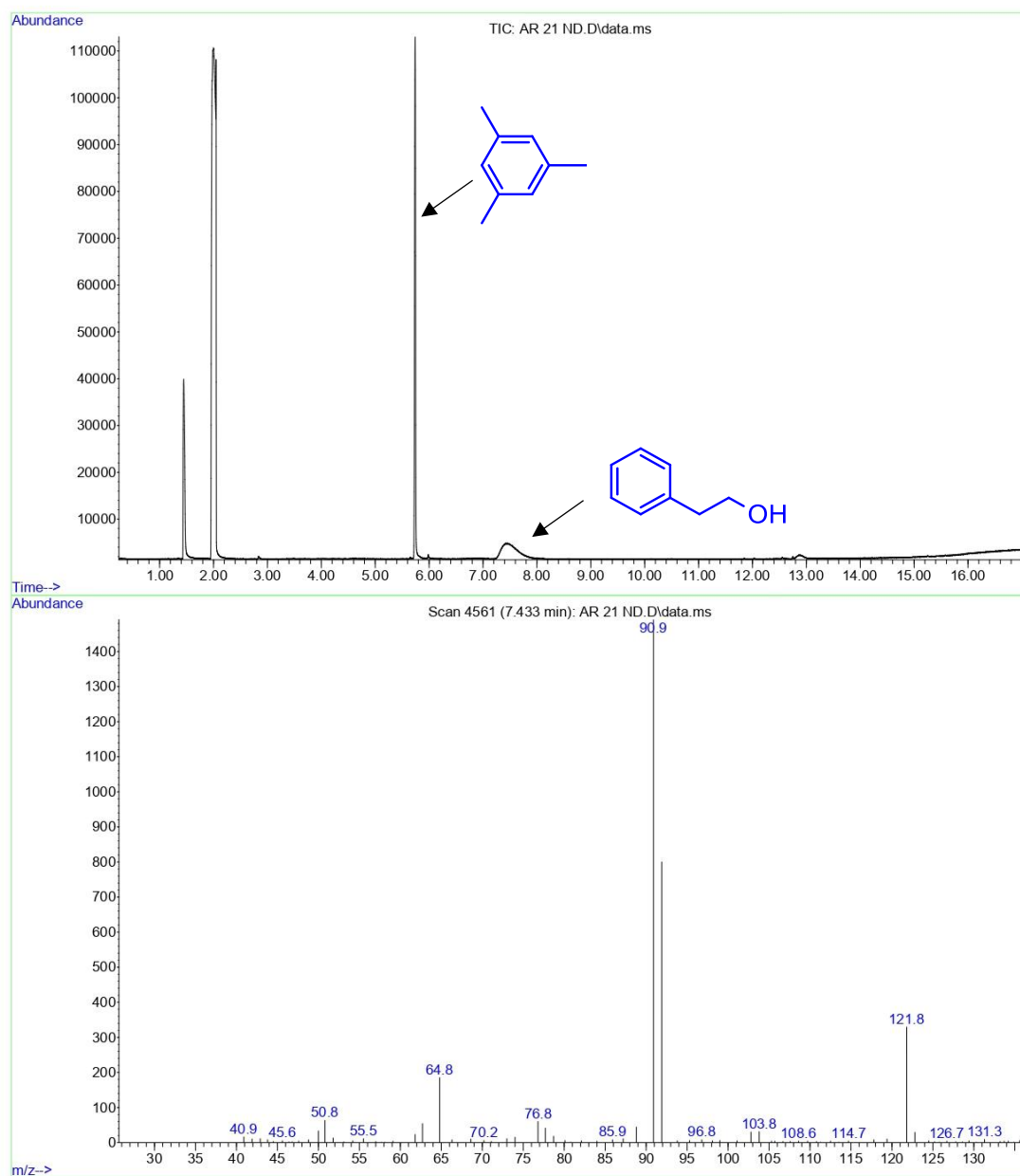
S2.13 GC–MS spectrum of the reaction described in Table 3, entry 11, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2025\AR-39.D
 Operator :
 Acquired : 31 Jan 2025 16:16 using AcqMethod Umi.M
 Instrument : GCMS
 Sample Name:
 Misc Info :
 Vial Number: 0



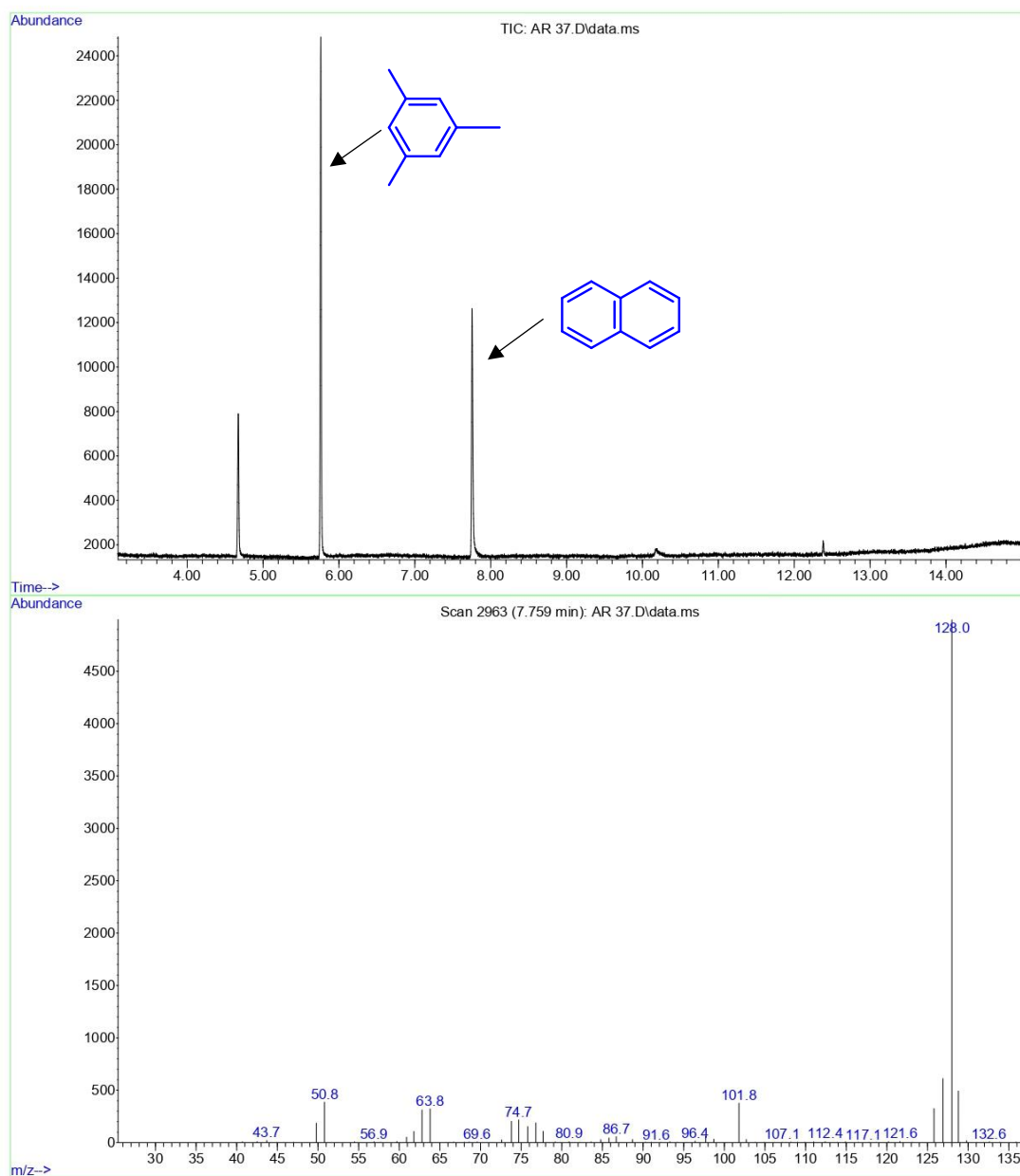
S2.14 GC–MS spectrum of the reaction described in Table 3, entry 12, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2025\AR 21 ND.D
Operator :
Acquired : 28 Feb 2025 14:28 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0

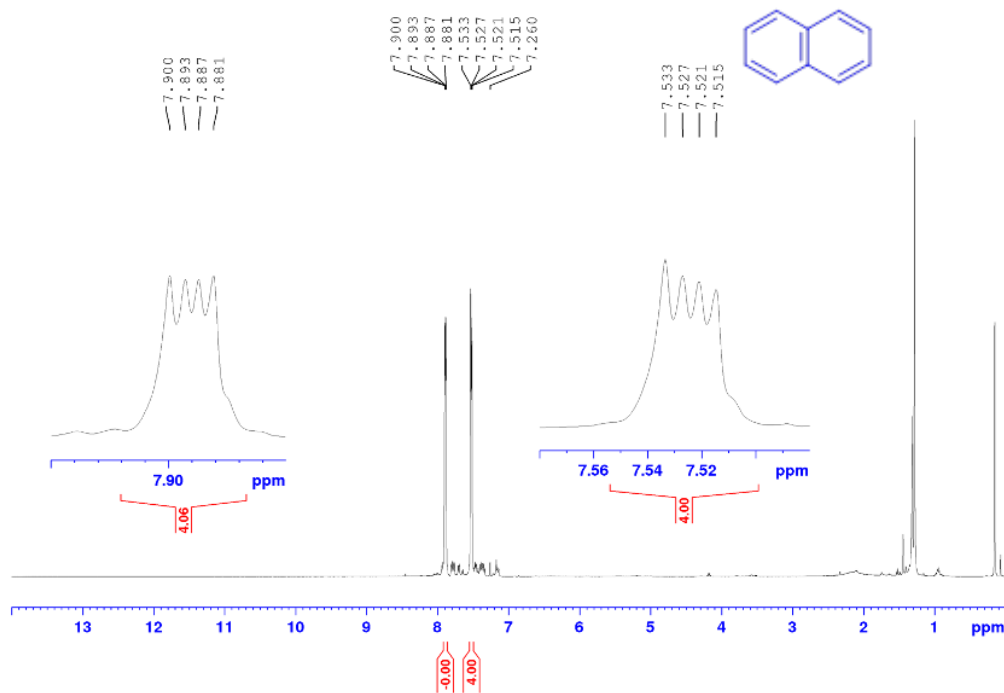


S2.15 GC–MS spectrum of the reaction described in Table 3, entry 13, with mesitylene as the internal standard.

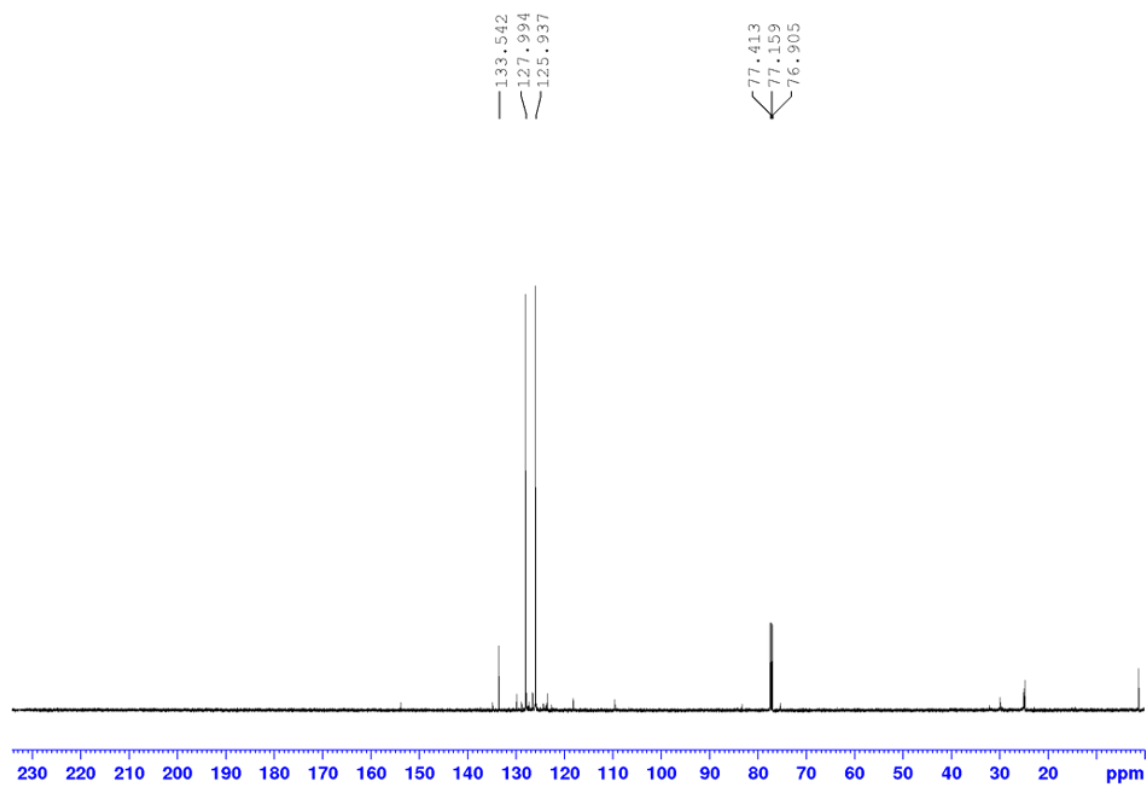
File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2025\AR 37.D
Operator :
Acquired : 10 Jan 2025 14:38 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



S2.16 ^1H NMR spectrum of **2m** (500 MHz, CDCl_3)

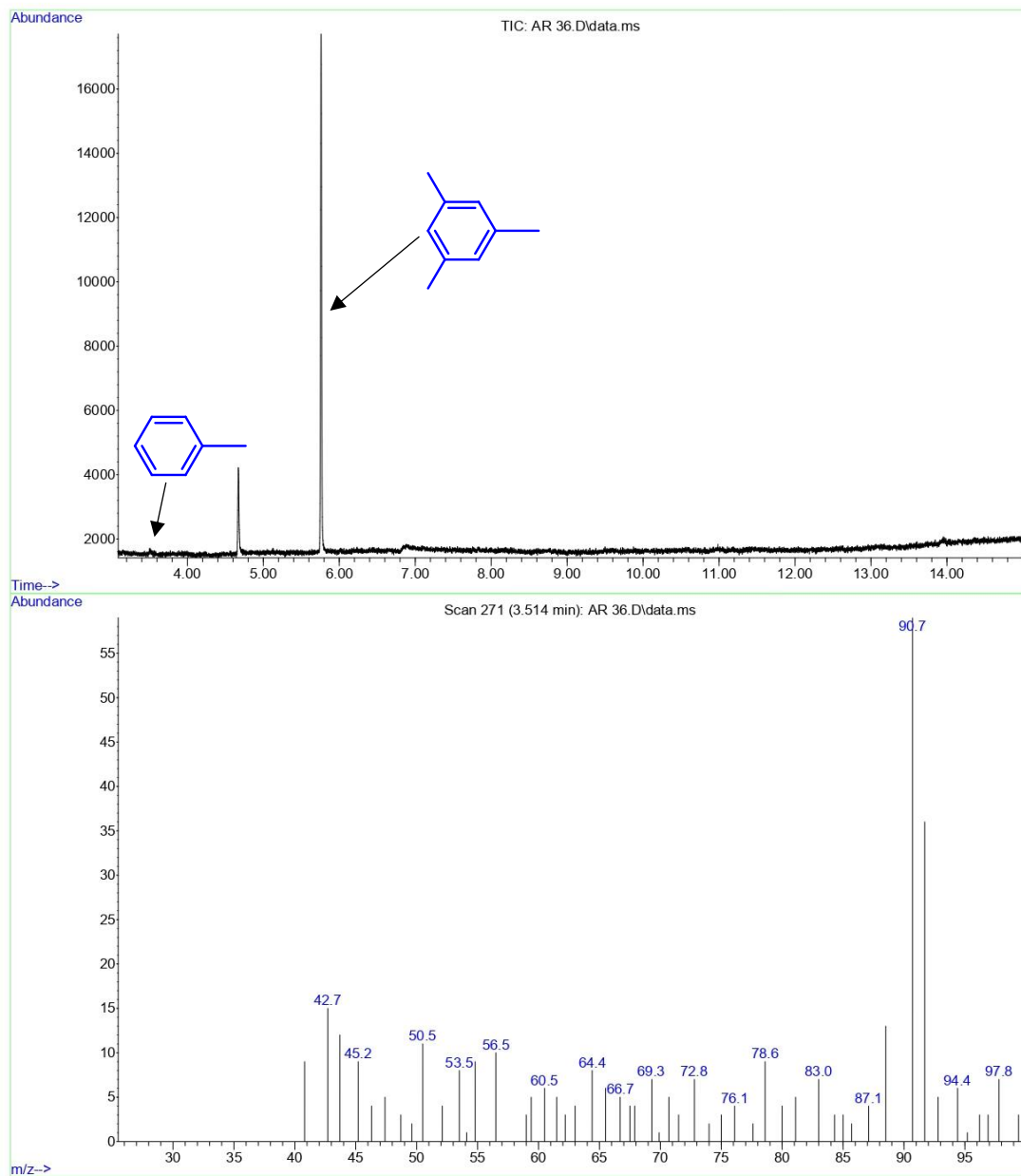


S2.17 $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **2m** (125 MHz, CDCl_3)



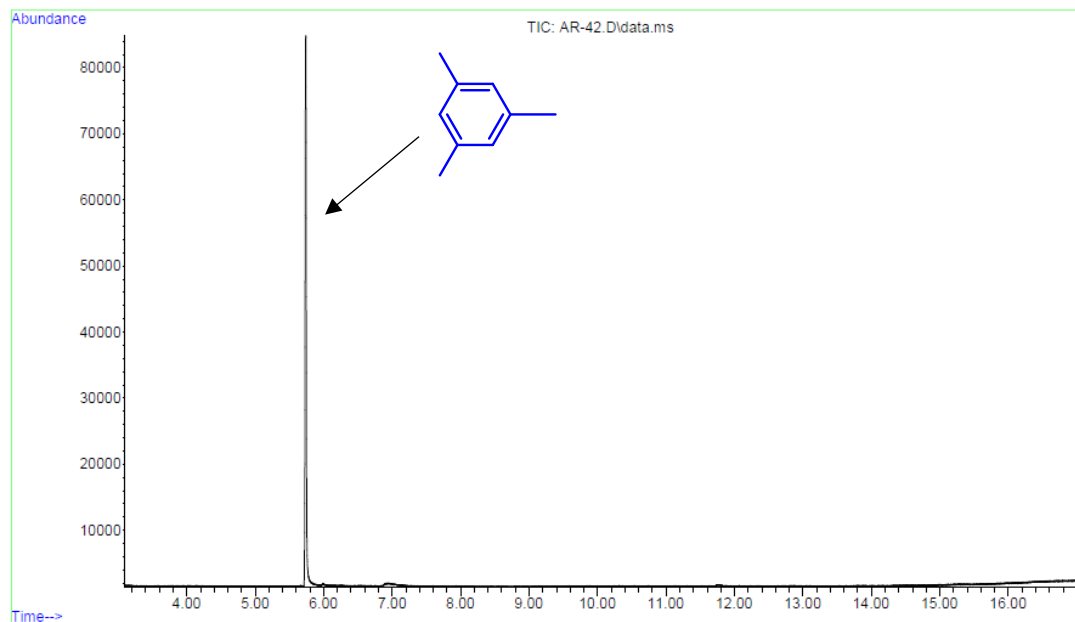
S2.18 GC–MS spectrum of the reaction described in Table 3, entry 13, with mesitylene as the internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2025\AR 36.D
Operator :
Acquired : 10 Jan 2025 14:12 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



S2.19 GC-MS spectrum of stock toluene adsorption on HKUST-1 with mesitylene as an internal standard.

File :D:\Lab Sentral\Data\2022\Yudha Reserch Grup\2025\AR-42.D
Operator :
Acquired : 24 Apr 2025 08:10 using AcqMethod Umi.M
Instrument : GCMS
Sample Name:
Misc Info :
Vial Number: 0



S3. XRD Data

S3.1 XRD Pattern of HKUST-1

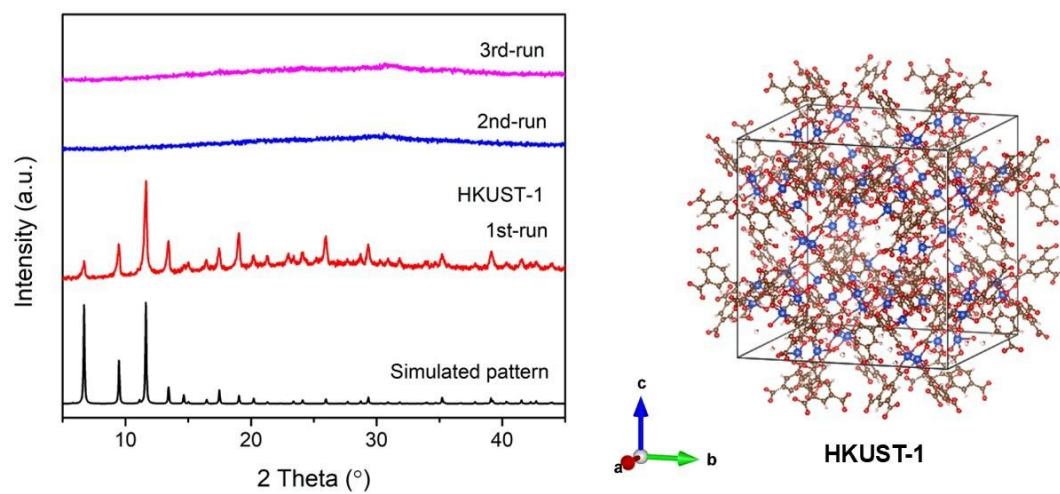


Fig. S1. Crystal structure and XRD pattern of HKUST-1 and after several uses.

S3.2 XRD Pattern of CuBDC

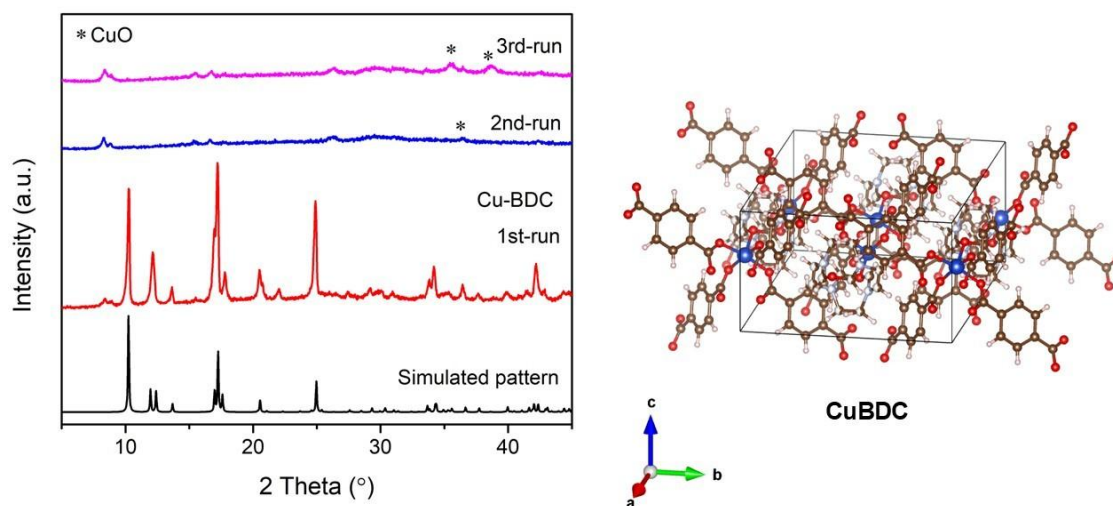
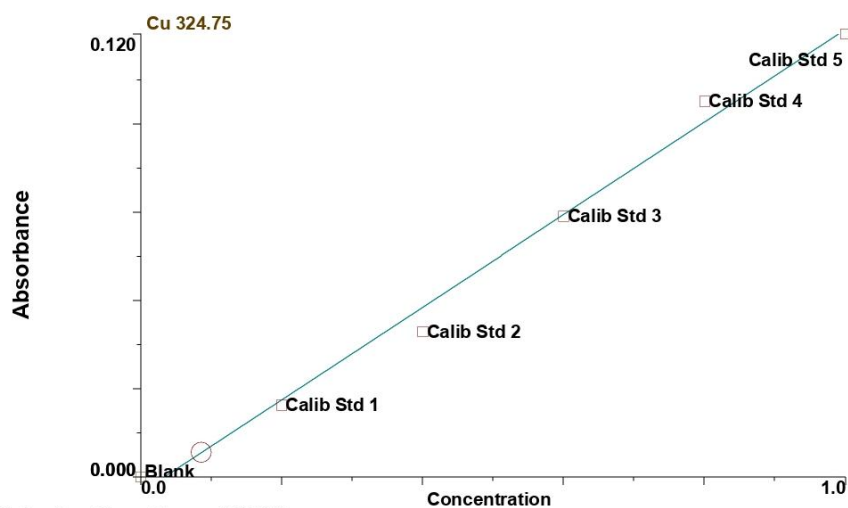


Fig. S2. Crystal structure and XRD pattern of CuBDC and after several uses.

S4. AAS Data



Calibration Curve Slope: 0.12555
 Calibration Curve Intercept: -0.00418
 Calibration Curve Correlation Coefficient: 0.995663
 Calibration Curve Type: Linear, Calculated Intercept
 Current Sample Concentration: 0.086 mg/L

Std #	Standard ID	Entered Conc.	Calculated Conc.	Action
Blank	Blank	0	0.033	Include
1	Calib Std 1	0.2	0.189	Include
2	Calib Std 2	0.4	0.347	Include
3	Calib Std 3	0.6	0.596	Include
4	Calib Std 4	0.8	0.845	Include
5	Calib Std 5	1.0	0.990	Include

The detected concentrations of nitric acid and ethyl acetate in the sample were 0.002 ppm and 0.003 ppm, respectively. Given a total sample concentration of 0.086 ppm, the measured copper concentration was 0.081 ppm.

Equation: Linear, Calculated Intercept

ID	Mean Signal (Abs)	Entered Conc. mg/L	Calculated Conc. mg/L	Standard Deviation	%RSD
Blank	0.0000	0	0.033	0.00	7.2
Calib Std 1	0.0195	0.2	0.189	0.00	1.0
Calib Std 2	0.0394	0.4	0.347	0.00	2.2
Calib Std 3	0.0706	0.6	0.596	0.00	1.0
Calib Std 4	0.1019	0.8	0.845	0.00	0.3
Calib Std 5	0.1201	1.0	0.990	0.00	0.4
Correlation Coef.: 0.995663 Slope: 0.12555 Intercept: -0.00418					

Autosampler Location:

Date Collected: 2/24/2025 10:31:17 AM

Data Type: Original

Repl	SampleConc	StdndConc	BlnkCorr	Time	Signal
#	mg/L	mg/L	Signal		Stored
1	0.026	0.026	-0.001	10:31:18	No
2	0.006	0.006	-0.003	10:31:22	No
3	0.005	0.005	-0.004	10:31:27	No
Mean:	0.012	0.012	-0.003		
SD:	0.012	0.012	0.0015		
%RSD:	99.58	99.58	58.26		

Autosampler Location:

Date Collected: 2/24/2025 10:31:34 AM

Data Type: Original

Rep#	SampleConc	StdConc	BlncCorr	Time	Signal
#	mg/L	mg/L	Signal		Stored
1	0.003	0.003	-0.004	10:31:34	No
2	0.002	0.002	-0.004	10:31:38	No
3	0.002	0.002	-0.004	10:31:42	No
Mean:	0.002	0.002	-0.004		
SD:	0.001	0.001	0.0001		
%RSD:	25.65	25.65	1.83		

Autosampler Location:

Date Collected: 2/24/2025 10:31:56 AM

Data Type: Original

Repl #	SampleConc mg/L	StdConc mg/L	BlnkCorr Signal	Time	Signal Stored
1	0.006	0.006	-0.003	10:31:57	No
2	0.004	0.004	-0.004	10:32:02	No
3	0.005	0.005	-0.004	10:32:06	No
Mean:	0.005	0.005	-0.004		
SD:	0.001	0.001	0.0001		
%RSD:	22.22	22.22	3.74		

Autosampler Location

Date Collected: 2/24/2025 10:32:20 AM

Data Type: Original

Repl #	SampleConc mg/L	StdConc mg/L	BlnkCorr Signal	Time	Signal Stored
1	0.087	0.087	0.007	10:32:20	No
2	0.086	0.086	0.007	10:32:25	No
3	0.086	0.086	0.007	10:32:29	No
Mean:	0.086	0.086	0.007		
SD:	0.001	0.001	0.0001		
%RSD:	1.067	1.067	1.74		

References

1. C. Liu, X. Li, Y. Wu and J. Qiu, *RSC Adv.*, 2014, **4**, 54307–54311.
2. Y.P. Budiman, M. Rashifari, S. Azid, I.Z. Ghafara, Y. Deawati, Y. Permana, U.S.F. Arrozi, W. Ciptonugroho, T. Mayanti and W.W. Lestari, *ChemistrySelect*, 2024, **9**, e202304913.
3. Z. Gai, B. Yu, X. Wang, Z. Deng and P. Xu, *Microbiology*, 2008, **154**, 3804–3812.
4. S. Abrantes, *J. High Resol. Chromatogr.*, 1993, **16**, 113–115.
5. B.M. El-Haj, A.M. Al-Amri, M.H. Hassan, R.K. Bin-Khadem and A.A. Al-Hadi, *J. Anal. Toxicol.*, 2000, **24**, 390–394.