

# Chitosan functionalized nitrogen-doped carbon dots nanocomposites: A turn-on sensor for fluoride detection

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**Table S1.** List of chemicals and their make required for the synthesis.

Order	Chemicals Name	Make
1.	Ethylene diaminetetracetic acid	Sigma-Aldrich
2.	Urea	Loba Chemie
3.	Chitosan	Loba Chemie
4.	Tetrabutylammonium fluoride hydrate	Sigma-Aldrich

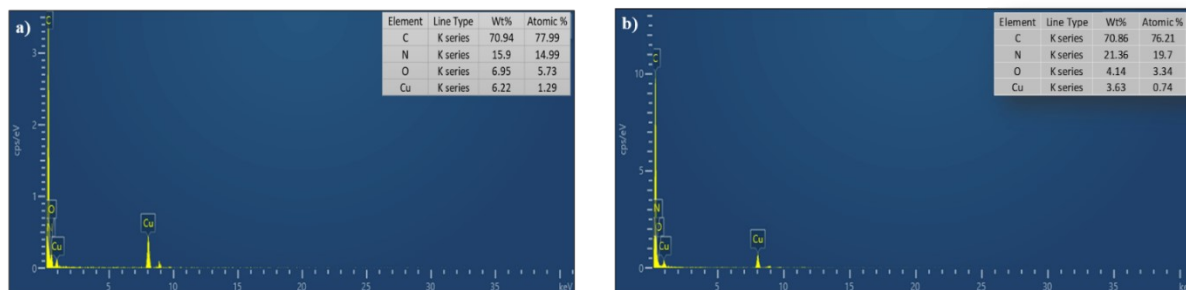
**Table S2.** Details of the instruments applied for the characterization and analysis of N-CDs and N-CDs@C

Order	Technique	Model
1.	Fourier transform infrared (FTIR) spectra	PerkinElmer Spectrum Two FTIR Spectrometer
2.	X-ray photoelectron spectroscopy	Thermofisher scientific Nexus base

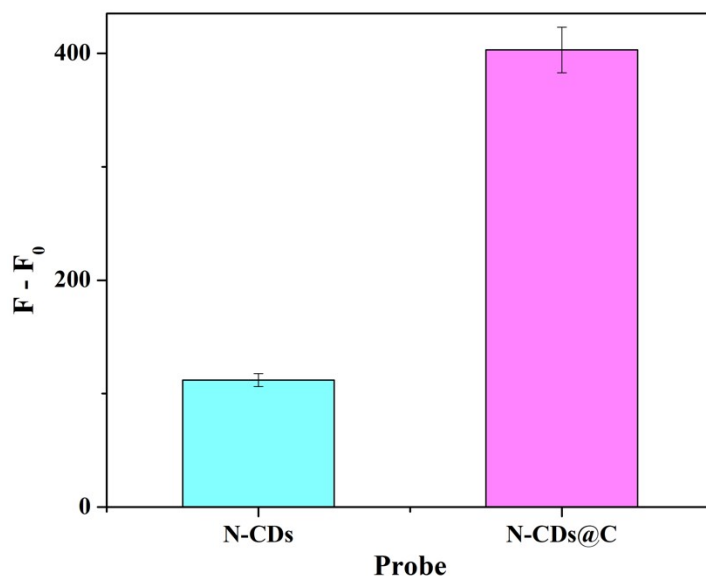
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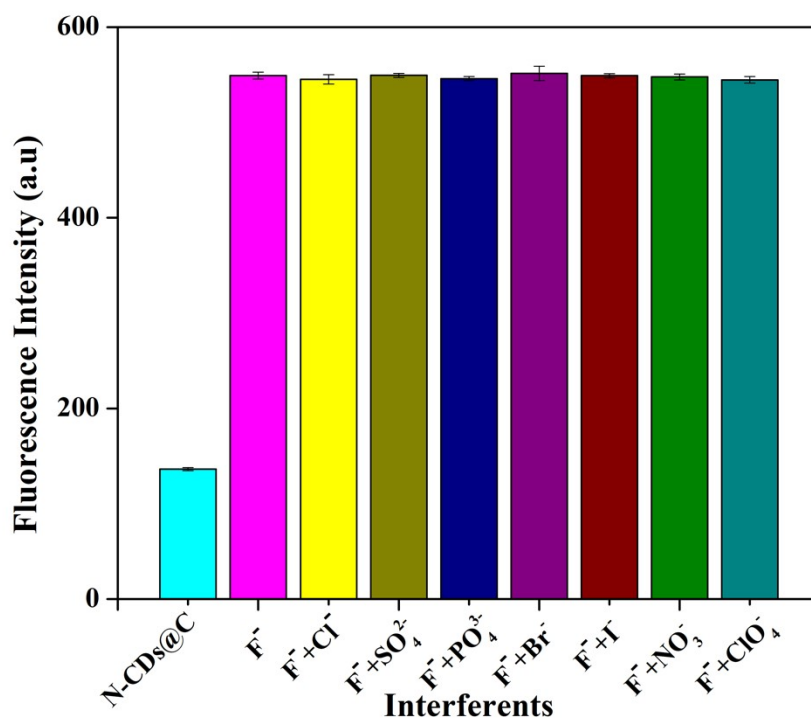
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|----|-----------------------------------------------------------|------------------------------------------------------------|
| 3. | Brunauer-Emmett-Teller (BET) surface area analyzer        | Autosorb iQ Station 2 system, Quntachrome Instruments, USA |
| 4. | High- resolution transmission electron microscopy (HRTEM) | JEOL JEM 2100 Plus, JEOL, Japan                            |
| 5. | Energy dispersive X-ray spectra (EDS)                     | JEM 2100 plus, JEOL, Japan                                 |
| 6. | Fluorescence Spectra                                      | Shimadzu Fluorescence RF-6000 equipped with a quartz cell  |
| 7. | Life time measuring system                                | Delta Flex TCSPC system (Horiba)                           |
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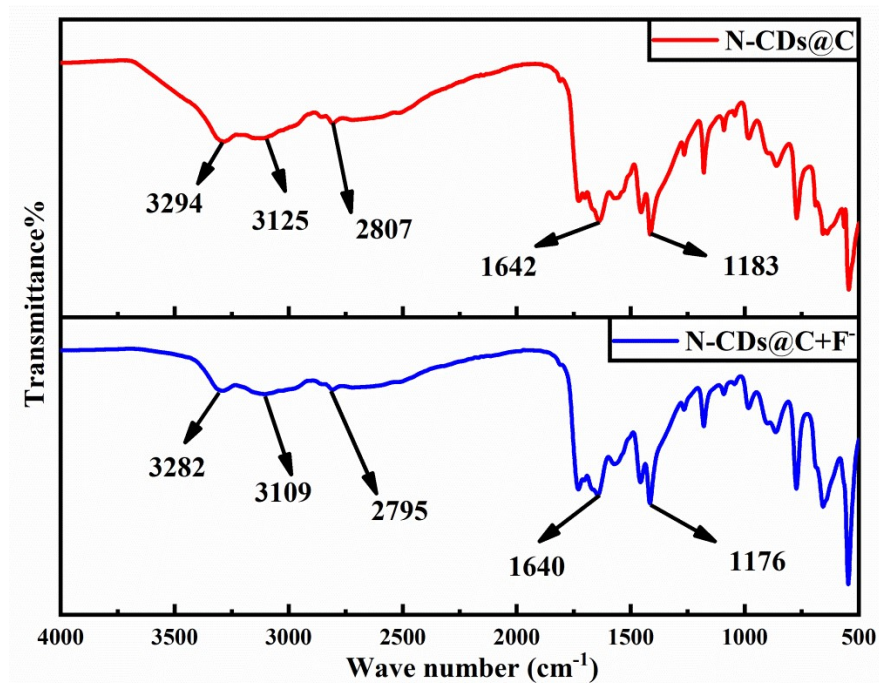
**Figure S1. EDS spectra of the synthesized materials: a) N-CDs and b) N-CDs@C**



**Figure S2. Comparative analysis of the sensing performance of N-CDs and N-CDs@C towards  $F^-$  ion**



**Figure S3. Fluorescence emission intensity of N-CDs@C in the presence of  $F^-$  and other interfering anions at a 1:1 ratio concentration**



**Figure S4. Comparison of FTIR spectra between N-CDs@C before and after binding with  $\text{F}^-$**