

Supplementary Information

Tailoring Janus In₂SeTe Monolayers with Al Doping: A Computational Study for NO_x Sensing Applications

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Table S I The lattice constant *a*; bond lengths of In-In (d1,d4), In-Se(d2,d5) and In-Te (d3,d6); bond angles of In-In-Se (θ1), In-In-Te (θ2), In-Se-In(θ3) and In-Te-In (θ4); the thickness layer(*t*); the band gap (*E_g*); VBM/CBM positions.

Type of Material	<i>a</i> (Å)	d1(Å) d2(Å) d3(Å)	<i>t</i> (Å)	θ1(°) θ2(°) θ3(°) θ4(°)	<i>E_g</i> (eV)	Stability	VBM/CBM positions
In ₂ SeTe	4.038	2.768, 2.684, 2.815	5.4	117.938°, 118.175°, 98.630°, 92.546°	1.192	Dynamically Stable	Γ - Γ
Al-Doped In ₂ SeTe	<i>a</i> (Å)	d4(Å) d5(Å) d6(Å)	<i>t</i> (Å)	θ5(°) θ6(°) θ7(°) θ8(°)	<i>E_g</i> (eV)	Stability	VBM/CBM positions
	4.054	2.767, 2.712, 2.809	5.34	116.296°, 122.713°, 98.540°, 95.217°	1.109	Dynamically Stable	Γ - Γ

Table S II Comparison of lattice constant, thickness, In-Se and In-Te bond lengths, and band gap(PBE) of Pristine In₂SeTe and Al-Doped In₂SeTe with other theoretical studies of In₂SeTe

Reference / System	Lattice constant a (Å)	Thickness l (Å)	In-Se bond length d_1 (Å)	In-In bond length d_l (Å)	In-Te bond length d_2 (Å)	Bond angle In-In-Se (°)	Bond angle In-In- Te (°)	Band gap PBE (eV)
This work (pristine In₂SeTe)	4.038	5.4	2.684	2.768	2.815	117.938	122.797	1.192
This work (Al-doped In₂SeTe)	4.054	5.3	2.712	2.767	2.809	116.296	122.713	1.109
In₂SeTe [31]	4.234	5.466	2.709	2.827	2.854	115.509	121.070	1.075
In₂SeTe [30]	4.23	6.02	—	2.83	—	—	—	1.38
In₂SeTe [29]	4.238	5.532	2.731	2.838	2.861	—	—	1.137
In₂SeTe [18]	4.18	5.55	2.71	2.82	2.84	—	—	1.07

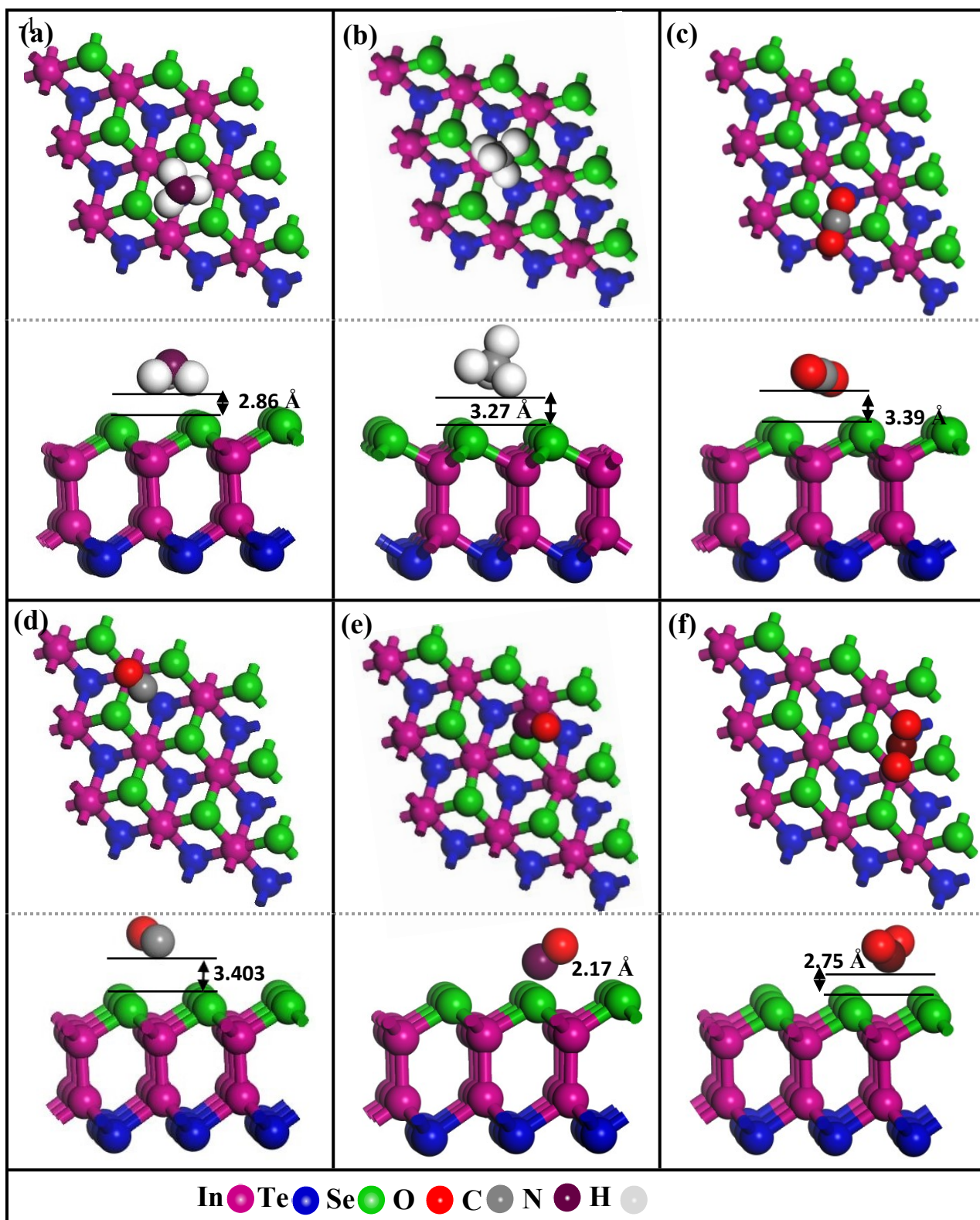


Fig. S1 Top View and Side View of In_2SeTe Monolayer after (a) NH_3 (b) CH_4 , (c) CO_2 , (d) CO , (e) NO and (f) NO_2 gas adsorption.

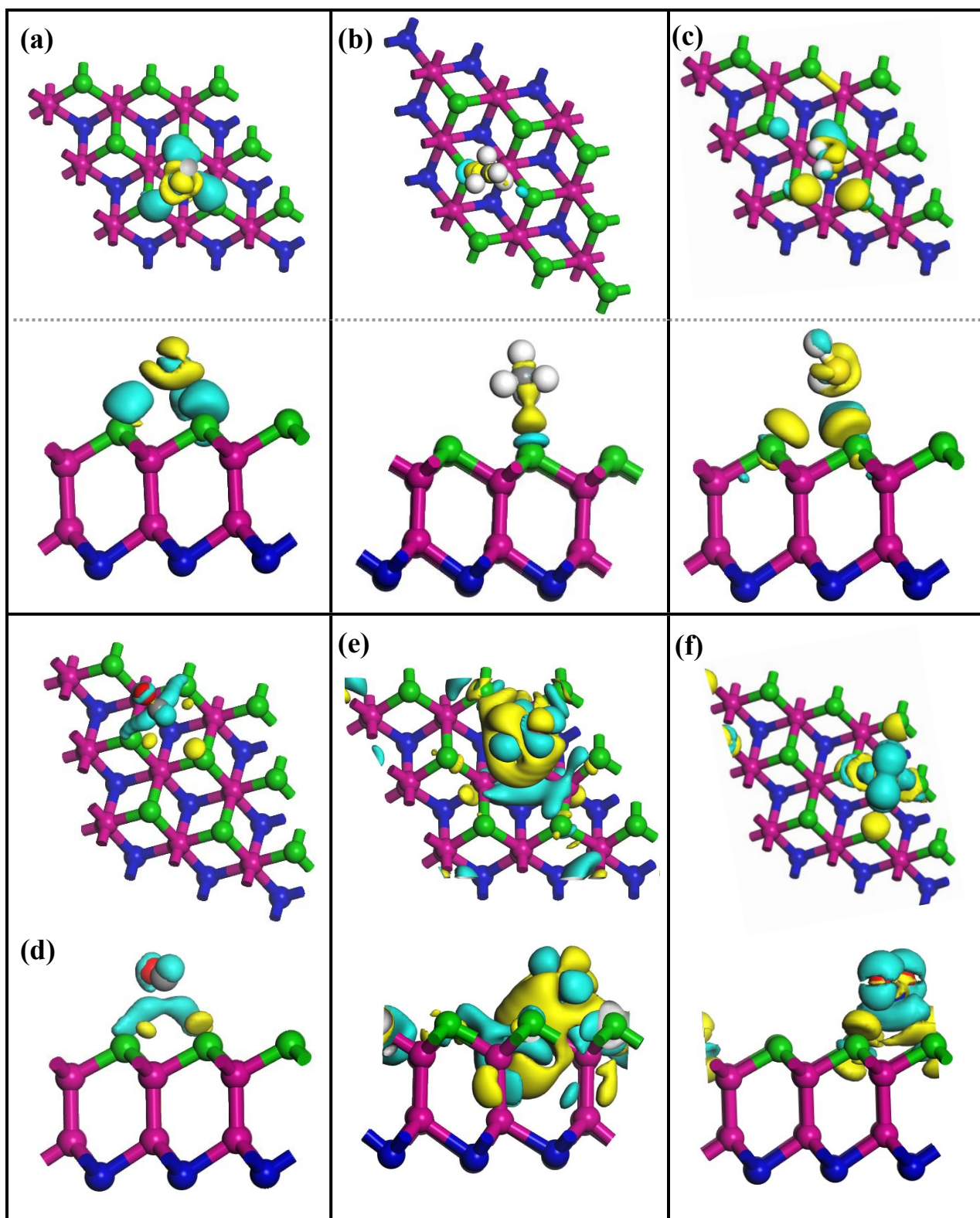


Fig. S2 Top View and Side View of Charge Density of In₂SeTe Monolayer after (a) NH₃ (b) CH₄, (c) CO₂, (d) CO, (e) NO and (f) NO₂ gas adsorption.

Table S III Variation of Bandgap and Sensitivity with Doping Percentage

%Doping	Bandgap (Without gas Adsorption) (eV)	Bandgap		Change in Bandgap(ΔE_g)		Log ₁₀ (%Sensitivity)	
		After NO ₂ Adsorption	After NO Adsorption	NO ₂	NO	NO ₂	NO
2.77%	1.109	0.497	0.041	-0.612	-1.068	7.135	10.962
5.5%	0.965	0.282	0.054	-0.683	-0.911	7.731	9.645
8.33%	0.861	1.024	0.096	0.163	-0.765	1.98	8.420
11.11%	0.819	0.992	0.097	0.173	-0.722	1.99	8.059

Table S IV (In-N/O) bond lengths for NO and NO₂ for Pristine and Al-doped In₂SeTe

Structure	Gas Molecules/Adsorption Configuration	Adsorption Height(h), Å	In-N/O
Pristine In ₂ SeTe	NO ₂	2.75	3.377/3.253
	NO	2.17	3.018/3.082
Al-Doped In ₂ SeTe	NO ₂	2.72	3.753/3.658
	NO	2.67	3.676/3.620

Table S V (In-Se/Te) bond lengths and bond distortion for NO and NO₂ for Pristine and Al-Doped In₂SeTe

Structure/Ga s Molecules	NH ₃	CH ₄	CO	CO ₂	NO	NO ₂
Pristine In ₂ SeTe	2.678/2.80 7	2.643/2.80 0	2.683/2.81 6	2.679/2.80 5	2.681/2.81 2	2.672/2.82 4
Al-Doped In ₂ SeTe	2.709/2.79 9	2.709/2.79 9	2.716/2.79 9	2.709/2.79 9	2.709/2.81 2	2.716/2.80 9

Bond Distortion	0.031/0.008	0.066/0.001	0,033/0.017	0.030/0.006	0.028/0	0.044/0.015
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Table S VI Sensitivity of different gas molecules adsorbed on pristine and Al-doped In₂SeTe monolayer, respectively

Gas Molecule	Pristine In₂SeTe		Al-doped In₂SeTe	
	Sensitivity (%)	Log₁₀(Sensitivity%)	Sensitivity (%)	Log₁₀(Sensitivity%)
NH₃	18.995	1.278	28.55	1.4556
CH₄	59.005	1.7708	10.14	1.0060
CO₂	3.4278	0.5350	7.43	0.8709
CO	5.6322	0.7506	12.29	1.0895
NO	7.249958×10 ⁵	5.8603	9.17×10 ¹⁰	10.9623
NO₂	2.653623×10 ⁵	3.4238	1.367×10 ⁷	7.1357

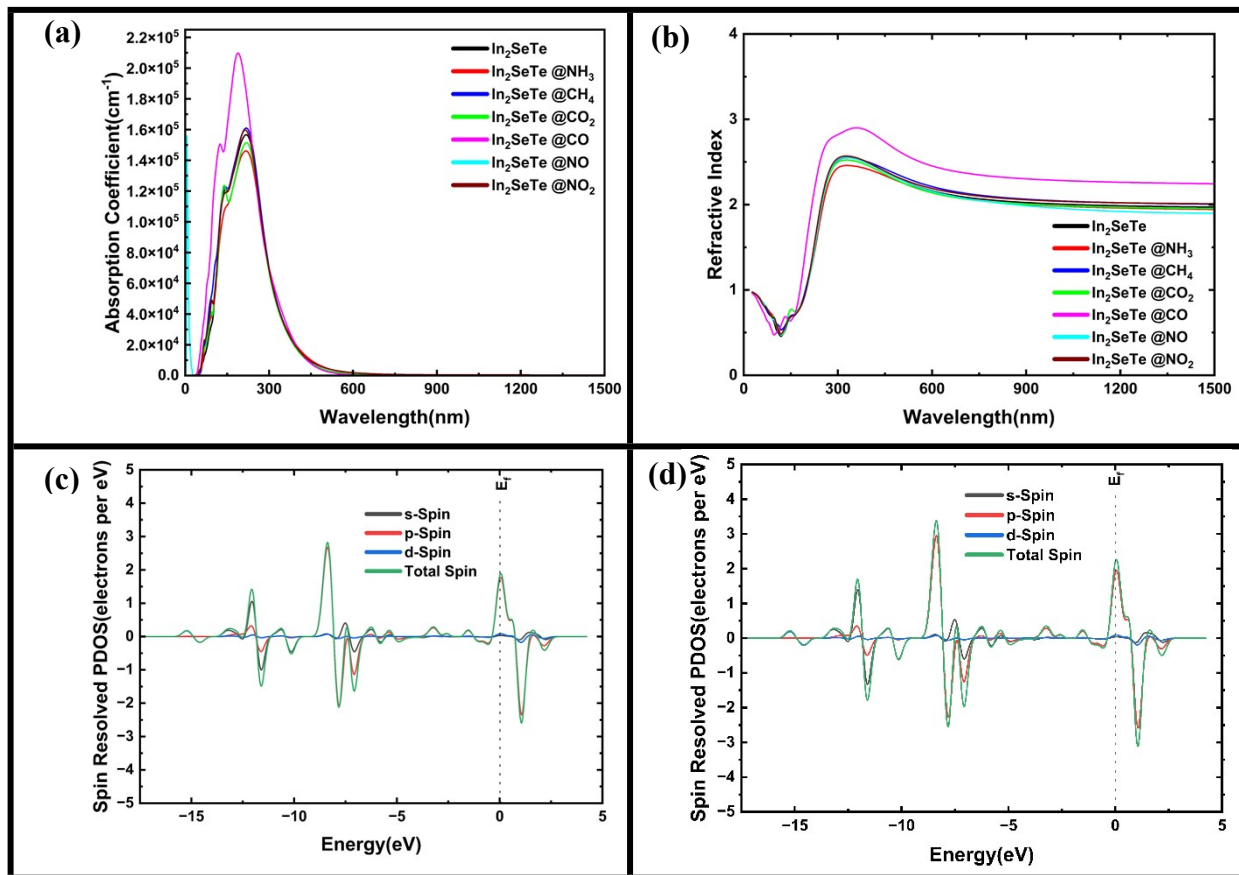


Fig. S3 (a) Absorption Coefficient, (b) Refractive Index of pristine In_2SeTe Monolayer, Spin Resolved PDOS of pristine In_2SeTe Monolayer for (c) NO, (d) NO_2